

A .HANDBOOK

OF

Petroleum, Asphalt and Natural Gas .

Methods of Analysis, Specifications, Properties, Refining Processes, Statistics, Tables and Bibliography

Roy Cross Black Stub p. 495-5

BY

Member American Chemical Society, American Society for Testing Materials, American Association for Advancement of Science, American Society for Municipal Improvements, Kansas City Engineers Club

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PREFACE

81

The purpose of this publication is to set forth in concise form for the petroleum producer, seller, refiner and technologist, scientific information and statistics on the production, properties, handling, refining and methods of valuation of petroleum and related products.

All matter formerly published in Bulletin No. 14 has been revised and included in this publication. In addition there has been added, fifty-five new illustrations, complete temperature-Baumé correction tables, extensive tank gauging tables, refinery engineering formulae, complete specifications for petroleum products, much additional data on oil cracking, geology, lubricants and asphalt, a complete set of methods of analysis of petroleum, asphalt and natural gas and a fairly complete bibliography.

The sources of original information have been from the research, commercial and engineering departments of the Kansas City Testing Laboratory and from the bibliography published at the end of the book.

November 1, 1919. Kansas City, Missouri.

Publications of Kansas City Testing Laboratory on Petroleum and Related Products.

- Bulletin No. 4. Asphalt and Asphalt Pavements. (Out of print.)
- Bulletin No. 9. Petroleum and its Products. (Out of print.)
- Bulletin No. 14. Petroleum, Asphalt and Natural Gas. 200 page book of Tables, Data and Statistics. \$2.00 per copy, postpaid.

Bulletin No. 15. Handbook of Petroleum, Asphalt and Natural Gas. \$5.00 per copy, postpaid.

CONTENTS.

(General outline only—see index for detailed subject matter.)

Economics				1- 3
Geology				4-11
Production and Refining Stati	stics ·			12-81
Storage, Measurement, Gaugi	ing, Tra	nsporta	tion -	82-119
Chemical and Physical proper crude oil	rties and	d distill:	ation o	of 120-137
Gasoline, Naphtha, Benzines,	, Minera	al spirit	s -	138-148
Kerosene, Illuminating oils, A	bsorptio	on oils		149-159
Lubricating oils, greases, wa	axes			160-180
Fuel oil and fuels				181-190
Asphalt and Road oil				191-208
Cracking and Engineering -				209-232
Oil shales and shale oil prod	ucts			233-244
Natural gas		- 1 - 2		245-269
Methods of analysis	-	- 0-		269-363
Tables				. 364-437
Patents				438-458
Bibliography				458-466
Index				466-500

Value of Petroleum as a Mineral Product

On page 32 is a statement showing the value and amount of production of the most important marketed mineral products of the United States in 1918. An examination of this table, as well as other tables on this page, shows that petroleum in the United States in 1918 exceeded in value any of the metals except iron which it equalled in value and was greater than the combined value of gold, silver, copper, lead and zinc. Coal was the only mined product exceeding it in value.

The chief change in the demand for petroleum products has been its relative limitation as fuel for steam or stationary power plant purposes and its increase in use for lubrication and for automobile engines. However, nearly one-half of the consumption of petroleum is still due to its use as fuel. More than 100 million barrels of petroleum could, should and probably will eventually be replaced by coal. The U. S. Navy normally may be expected to consume at least six million barrels of fuel oil per year and there are some industries which require the flexibility of fuel oil and its low sulphur content with absence of ash. The price of coal must, in the long run, very largely govern the price of petroleum products as the demand for gasoline increases and gasoline must remain at the present or higher price. The governing factor in this situation has been the gasoline automobile. It is quite apparent that the point of saturation for automobiles has not been reached as is indicated by the following table showing the demand for gasoline.

		Per Cent
Automobiles	Gasoline	from Crude Oil
85,000	7,900,000	5.91
400,000	14,750,000	7.04
1,253,000	34,900,000	13.14
2,225,000	49,020,000	19.85
3,250,000	64,290,000	21.15
4,500,000	86,561,150	26.07
	85,000 400,000 1,253,000 2,225,000 3,250,000	$\begin{array}{rrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrrr$

The increase in automobiles must diminish or the increase in the percentage of gasoline obtainable from crude oil must continue.

During 1918 practically the entire increase in gasoline production was due to an additional production of cracked or artificial gasoline almost entirely from the Standard Oil Company. It is probable that the limit has been reached for the quality of gasoline as there has been no change in the past year. It does not seem probable that a satisfactory automobile engine will be worked out soon which will be capable of handling distillate mixed with gasoline as it seems that the increased efficiency by reason of cracking the heavy oil in the cracking plant or refinery easily offsets the lower price of distillate that might be used by cracking it in the automobile cylinder.

Wax and lubricants are the most valuable products of the refining of petroleum, both of which have shown very great increases in amount produced during the past year. For this purpose, however, the highest grades of petroleum are necessary and very elaborate and expensive refinery equipments are required for their production. It seems that Mexican and other high sulphur oils must, to a large extent, in the very near future be the source of fuel oils. When, however, natural petroleum has passed as a fuel a very abundant potential source of synthetic petroleum exists in the oil shales and cannel coal. The destructive distillation of oil shales yields fuel oil, lubricating oil, wax and illuminating oil. The very substantial yield of wax and lubricants particularly may stimulate an earlier development of the oil shale industry than might otherwise be expected.

The following outlines some of the uses of petroleum products: Gasoline and Naphtha—Gas lighting, laboratory solvents, cleansing, gasoline stoves, automobiles, extraction of seed oils, metal polishes, gasoline engines, paint vehicles, asphalt paint and road binder solvent.

Kerosene and Illuminating Oils-Lamps, distillate engines, signal lights, gas washing and absorbents, portable stoves.

- Gas Oil—Pintsch gas, Blaugas, town gas, straw oil, heating, cracking, anti-corrosives.
- Heavy Distillates—Lubricants, spindle oil, auto oil, machine oil, engine oil, cylinder oil, greases, vaseline, wax, medicinal oil, waterproofing for fabrics, candles, soap filler, paints, polishes.

Liquid Residua—Steam production, heating, concrete waterproofing, road and macadam oils, dust prevention, cracking.

Semi-Solid Residua—Asphalt pavement, waterproofing, brick filler, roofing, rubber filler or substitute.

Crude Oils-Diesel engines, dust prevention, waterproofing.

Geological Occurrence of Petroleum and Natural Gas

The following summarizes the geological conditions under which petroleum and natural gas occur:

1. They occur in sedimentary rocks of all geologic ages from Silurian upward. The most productive areas are the Paleozoic in North America and the Miocene in Russia.

2. There is no relation of the occurrence of petroleum to volcanic or igneous action. There seems to be some relation particularly in the carboniferous and the Mississippian to the deposits of coal.

3. The most productive areas for oil in great quantity are where the strata are comparatively undisturbed. Oil frequently occurs where the strata are highly contorted and disturbed but in less abundance, and gas is usually absent.

4. In comparatively undisturbed as well as in disturbed areas a folded or dome structure often favors the accumulation of oil and gas in the domes or anticlines.

5. Important requisites for a productive oil or gas field are an impervious cap rock or cover and a porous reservoir.

6. Salt water almost universally accompanies oil and gas in the same sand.

In the United States, oil is found most abundantly in the Tertiary rocks in California and the Gulf Coast, in upper cretaceous in Wyoming, in carboniferous locally known as the Cherokee Shales in the Mid-Continent field, in the sub-carboniferous or Mississippian and the Upper Devonian in the Appalachian field and in Illinois, and in the Ordovician in Ohio and Indiana. The oils from the Tertiary are heavy and of low grade, those from the cretaceous, carboniferous and subcarboniferous are light, high grade oils. The Mississippian in the Mid-Continent field is not supposed to carry any oil and very little is known of it or deeper strata in this territory. It is assumed that the deeper strata have vanished west of the Ozark uplift.

The accumulation of petroleum occurs in a pervious reservoir which usually consists of a loose sand though it may be a coarse gravel or a disrupted shale or limestone. It is merely necessary that the rock should contain a considerable amount of voids. The ordinary sand will have from 15 per cent to 35 per cent of voids and the amount of oil contained and the ease with which it is discharged into a well vary greatly. As a general rule, one gallon of oil may be obtained from one cubic foot of oil sand. It is probable that never over 75 per cent of the oil surrounding a well is discharged into it even with the lighter oils, and the per cent abstracted is much lower with the heavier and more viscous oils. Porous sand and gravel and heavy gas pressure are conducive to rapid expulsion of oil. Fine sand and low pressure give steadily producing wells of great longevity. The ultimate production of a well would be determined by the depth and extent of the sand, the physical character of the sand, the physical character of the oil and the pressure. Water is a very important element in the actual production of a well. It frequently causes very extensive subterranean oil movements destroying one productive structure and making new productive structures.

In nearly every oil sand there occur together, gas, oil and salt water. The gas invariably occupies the uppermost portion of the sand, the salt water, the bottom with the oil intermediate. The sand usually lies at the same angle or dip as the stratum in which it is contained, so that this fact forms the basis, to a great extent, of the geologist's work. It is to be noted that the surface topography has no relation to the probable location of oil or the dip or "strike" of the formation beneath the surface. Asphalt exposures are not good indications of oil in the immediate vicinity but indicate that oil may be found of good quality where this same geological structure is capped by an impervious cover. Anticlines bear no definite relation to surface topography, though the anticline is more likely to be found corresponding in a general way to the bottom of an old river or stream bed than corresponding to the divide between two streams.

Oil of good quality is usually found at sufficient depth that the lighter fractions have not evaporated, though some good wells are found at depths as shallow as 250 feet. The best wells of the Mid-Continent field vary from 1,000 to 3,500 feet in depth. The deepest well in the United States is the Lake Well in Harrison County, West Virginia, and is 7579 feet deep. Wells at Ranger, Texas, are about 3,400 feet deep. A well in Banner County, Nebraska, is 5,600 feet deep. Named in order of depth, the four deepest wells in the world are the Lake; the Goff, West Virginia, 7,386 feet, and a well at Czuchow, Germany, 7,348. In comparison with these great depths, other depths reached by wells or mines sunk in the crust of the earth are rather insignificant. The deepest mine in the world is Shaft No. 3 of the Tamarack mine, in Houghton County, Michigan, which has reached a depth of 5,200 feet.

The preponderance of evidence points to the theory that the greater part of petroleum has been produced from organic matter of any kind undergoing decomposition, followed by its segregation by the action of water and accumulation in pervious rocks of the oil produced. Other theories are that oil originated from animal matter and also that it came from the reaction of metallic carbides at high pressure with water.

A demonstration as to the origin of petroleum hydrocarbons is very readily made by the use of the cracking' test described on page 319. By heating corn oil, cottonseed oil or other vegetable or animal oil a product is made which is identical in boiling point range with that of ordinary crude oil though it contains a rather large amount of volatile fatty acids. An almost exact duplication of crude petroleum oil can be produced with this apparatus by placing lime in the receptacle with the vegetable oil. In this case the light distillate is almost entirely composed of paraffin hydrocarbons.

TEMPERATURE IN WELLS (WEST VIRGINIA)

100	feet
1,000	feet
2,000	feet
3,000	feet
5,000	feet114.2°
	feet
7,000	feet153.2°
7,310	feet

The rate of temperature increase varies continuously from 1 degree Fahr. in 97.5 feet at the surface to 1 degree Fahr. in 46.5 feet over the interval 6,000 to 7,000 feet. In the Texas and Oklahoma oil fields temperatures at a given depth differ widely from those found in Pennsylvania and West Virginia. The temperature of the oil in two wells near Mannington, W. Va., is 83.2 degrees Fahr. at a depth of about 2,900 feet. In the Ranger field, Texas, the temperature of the oil at 3,400 feet is estimated, from measurements at higher levels, to be about 135 degrees. The average rate of temperature increase at the surface for thirteen wells in Texas and Oklahoma is about 1 degree Fahr. in 51 feet, as compared with 1 degree in 91.5 feet for twelve wells in Pennsylvania and West Virginia.

Field	Structure	Geologic Age	Kind of Rock	Kind of Petroleum
Appalachian or Eastern	Geo-Syncline with subordi- nate anti- clines	Ordovician to Carboniferous	Sandstone	Paraffin base
Ohio-Indiana	Anticlines	Ordovician	Mostly lime- stone	Paraffin base
Illinois	Low anticlines	Carboniferous	Sandstone	Paraffin and semi-paraffin base
Mid-Continent	Anticlines	Carboniferous	Sandstone	Paraffin semi-paraffin base
Wyoming	Folds	Carboniferous to Tertiary	Mostly sand- stone	Paraffin and asphalt base
Gulf Coast	Domes	Tertiary and Cretaceous	Dolomite and - sandstone	Asphalt base
California	Folds and Faults	Tertiary	Sandstone, shales and conglomer- ates	Asphalt base
Mexico		Tertiary		Asphalt base

SUMMARIZED TABLE OF OIL OCCURRENCES IN THE UNITED STATES

TYPICAL COMPOSITION OF "MISSISSIPPI LIME" AT TOP

(From Wilson County, Kansas)

Carbon dioxide	32.0%
Silica + Insoluble	20.5%
Iron and Alumina (R ₂ O ₃)	3.3%
Lime (CaO).	23.4%
Magnesia (MgO)	11.8%

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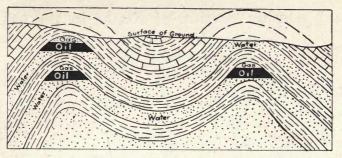


Diagram representing the accumulation of oil and **qas** in anticlines.

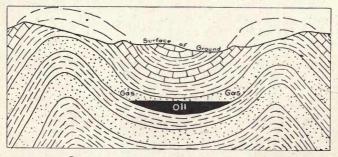
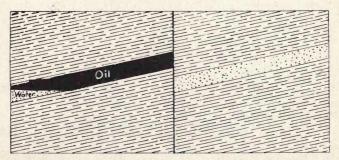


Diagram showing the accumulation of oil in a syncline in the absence of water



Ideal section showing an oil sand faulted in such a manner that an accumulation of oil will result

Stratigraphic Section of Rocks in the Oil-Bearing Region of Kansas

(Kansas Geological Survey) PERMIAN SERIES

Thickness

Wellington Formation-	
Sandstone, limestone, shale, salt and gypsum4	00-600
Marion Formation-	
Abilene Conglomerate.	25 - 50
Pearl Shale	55-75
Herington Limestone	12-15
Enterprise Shale	30-50
Luta Limestone	20-40
Chase Formation—	
Winfield Limestone	20-25
Doyle Shale.	50-70
Fort Riley Limestone	40-45
Florence Flint.	15-25
Matfield Shale	60-70
Wreford Limestone	35-50
Council Grove Formation—	
Garrison Limestone and Shale1	35-150
Cottonwood Limestone	5-7

PENNSYLVANIAN SERIES

Wabaunsee Formation-		
Eskridge Shale		. 30-40
Neva Limestone		
Elmdale Shale		
Americus Limestone		
Admire Shale, possibly contains shallow oil sand		
Eldorado.		
Emporia Limestone.		
Willard Shale		
Burlingame Limestone		
Shawnee Formation-		
Scranton Shale.		160-200
Howard Limestone.		
Severy Shale.		
Topeka Limestone.		
Calhoun Shale.		
Deer Creek Limestone		
Tecumseh Shale.		
Lecompton Limestone.		
Kanawha Shale.		
Douglas Formation—		
Oread Limestone.		. 50-70
Lawrence Shale, including Chautauqua Sandstone me		
probably 1,500 feet sand at Augusta and Eldora	do	150-300
Iatan Limestone.		
Weston Shale		

PENNSYLVANIAN SERIES—Continued

Lansing Formation—	
Stanton Limestone	20-40
Vilas Shale	5-125
Plattsburg Limestone	5-80
Lane Shale	50-150
Kansas City Formation—	
Iola Limestone	2-40
Chanute Shale.	
Drum Limestone	0-80
Cherryvale Shale-possibly horizon of oil sand at 2,400	
feet at Augusta and Eldorado	25-125
Winterset Limestone	
Galesburg Shale	
Bethany Falls Limestone	4-25
Ladore Shale.	3-50
Hertha Limestone	
Marmaton Formation	
Pleasanton Shale.	100-150
Coffevville Limestone	8-10
Walnut Shale.	60-80
Altamont Limestone	3-10
Bandera Shale	60-120
Pawnee Limestone (Big Lime)	
Labette Shale (Horizon Peru Oil Sand)	0-60
Fort Scott (Oswego) Limestone	
Cherokee Formation—	
Cherokee Shale-includes the main oil sands outside	
Augusta and Eldorado and Peru-contains Bartles-	
ville and Burgess sands	400-500
Mississippian Limestone-	
Limestone, calcareous shale and chert shown in Neosho	
well-Boone Formation.	320
Probably Older than Mississippian-	
1-Dolomitic limestone, sandstone and chert shown in	
Neosho well.	77
2-Conglomerate and shale in Neosho well	23
3-Sandstone, conglomerate with pables up to three-	
quarters inch diameter; shown in Neosho well	1823

Thickness

Stratigraphic Section in Main Oil and Gas District of Northern Oklahoma

(Oklahoma Geological Survey and Other Sources)

PERMIAN SERIES

Thickness Feet

1-Red and gray sandstone, clay-iron conglomerates, red and vari-colored shale, thin beds of concretionary limestone near base, beds of gypsum and salt in the upper portion. Quartermaster, Greer, Wood- ward, Blaine and upper portions of Enid forma- tions.	1200-2000
2—Beds of thin limestone, sandstone and shale. Con- tains near base the shallow gas sands at Black- well, Billings and Garber	500-600
PENNSYLVANIAN SERIES Ralston Group— Consists of red and gray sandstone, red shale and beds of thin limestone. Contains the Hoy oil sand at	
Garber. 1—Upper division down to Pawhuska limestone, inclusive 2—Lower division down to Elgin sandstone	650 140
 4—Avant Limestone	$\begin{array}{c} 20-150\\ 0-20\\ \end{array}\\ 700-1000\\ 0-10\\ \end{array}\\ 300-400\\ 15-25\\ 350-400\\ 10-20\\ \end{array}$
Tulsa Group-	
1—Nowata Shale—includes the Wayside oil sand and its correlations (local coal bed, Dawson coal)	75-150

2-Oolagan Limeston 3-Labette Shale. abette Shale. Sandstone, shale and beds of thin limestone. Includes the Cleveland and Peru oil sands. 250-300 4-Oswego Limestone (Fort Scott) ... 25-100

Thickness Feet

Muskogee Group— Beds of shale, sandstone and thin limestone correlating with the Cherokee shale (Boggy and Winslow for- mations at Muskogee). Includes the main oil sands of Oklahoma, the Red Fork, Bartlesville, (Glenn), Tucker, Taneha, Booch, Morris and Mus- kogee sands, the latter lying at the unconformable	
base of the Pennsylvania series	400-1000
MISSISSIPPIAN SERIES 1—Morrow Limestone	100-200
2—Pitkin Limestone	40-60
stone contains the Mounds oil sand and a deep sand near Sapulpa	20-200
4—Boone formation. Massive white limestone and mas- sive beds of chert	200-400
DEVONIAN SYSTEM 1—Chattanooga formation. Black fissile shale 2—Sylamore sandstone, clear quartz sandstone (Unconformity)	30-50 0-25
ORDIVICIAN SYSTEM 1—Turner formation. Thin sandstone and limestone in shale	60-100
stone	5-100
CAMBRIAN SYSTEM Massive limestone beds shown in Harrington well at Joplin, Mo	1165

	World	l's Pro	World's Production of Petroleum	of P	etroleun	r		
	Total Production	uction						
	1857 to 1917	1917	Production 1916	1916	Production 1917	1917	Production 1918	1918
Country	Barrels	Per Cent	Barrels	Per Cent	Barrels	Per Cent	Barrels	Per Cent
United States	4,252,644,003	60.89	300,767,158	65.29	335,315,601	. 66.98	345,500,000	66.94
Russia	1,832,583,017	26.24	72,801,110	15.81	69,000,000	13.78	65,000,000	12.59
Mexico	222,082,472	3.18	39,817,402	8.64	55,292,770	11.04	64,605,422	12.52
Dutch East Indies	175,103,267	2.51	13,174,399	2.86	12,928,955	2.58	13,000,000	2.52
Roumania	142,992,465	2.05	10,298,208	2.24	2,681,870	.54	2,900,000	0.56
India	98,583,522	2.41	8,228,571	1.79	8,500,000	1.70	8,500,000	1.65
Galicia	148,459,653	2.13	6,461,706	1.40	5,965,447	1.19	6,000,000	1.16
Japan and Formosa	36,065,454	.52	2,997,178	0.65	2,898,654	.58	2,750,000	0.53
Peru	21,878,285	.31	2,550,645	0.55	2,533,417	.51	2,500,000	0.48
Trinidad	5,418,885	.08	1,000,000	0.22	1,599,455	.32	1,600,000	0.31
Germany	15,952,861	2.30	995,764	0.22	995,764	.20	1,000,000	0.19
Argentina	3,047,858	0.04	870,000	0.19	1,144,737	.23	1,000,000	0.19
Egypt	2,768,686	.04	411,000	0.09	1,008,750	.20	1,000,000	0.19
Canada	24,112,529	3.50	198,123	0.04	205,332	.04	300,000	0.06
Italy	947,289	.01	43,143	0.01	50,334	.01	50,000	0.01
Other countries	927,000	.01	25,000	0.01	530,000	.10	500,000	0.10
	6,983,567,246	100.00	460,639,407	100.00	500,651,086	100.00	516,205,422	100.00

BULLETIN NUMBER FIFTEEN OF

PETROLEUM PRODUCTION BY STATES.

State	1915	1916	1917	1918
State				
Oklahoma	97,915,243	111,000,000	97,600,000	84,950,300
California	86,591,535	92,000,000	97,000,000	101,493,000
Texas	17,467,598	26,000,000	30,000,000	42,000,000
Illinois	19,041,695	16,500,000	11,000,000	11,000,000
Louisiana	18,191,539	17,000,000	15,000,000	15,900,000
West Virginia	9,264,798	8,500,000	8,000,000	8,000,000
Pennsylvania	7,838,705	8,000,000	8,000,000	8,000,000
Ohio	7,825,325	7,400,000	7,000,000	8,000,000
Kansas	2,823,487	11,500,000	38,000,000	43,253,470
Wyoming-Montana.	4,245,525	6,300,000	10,000,000	13,815,000
Kentucky	437,274	1,200,000	4,000,000	7,000,000
Indiana	875,758	1,000,000	1,000,000	1,000,000
New York	887,778	900,000	900,000	900,000
Colorado	208,475	190,000	200,000	200,000
Other States	14,262	10,000	10,000	10,000

281,104,104 307,500,000 327,610,000 345,521,770

PRODUCTION OF PETROLEUM	BY DISTRIC	TS
Field	1917	1918
Appalachian	24,932,205	25,300,000
Lima-Indiana	3,670,293	3,100,000
Illinois	15,776,360	13,300,000
Oklahoma-Kansas	155,043,596	139,600,000
Central and Northern Texas	10,900,646	15,600,000
North Louisiana	8,561,963	13,000,000
Gulf Coast	26,087,587	21,700,000
Rocky Mountain	9,199,310	12,600,000
California.	93,877,549	101,300,000
Alaska and Michigan	10,300	

Totals.	335,315,601	345,500,000

SOURCES OF CRUDE OIL IN THE UNITED STAT	TES IN 1918
	Barrels
Produced in the United States	339,400,000
Drawn from stocks	27,000,000
Imported (Mexico)	37,735,000
Total	404,135,000

DISPOSITION OF CRUDE OIL IN U.S. IN 1918

	Barrels
Total amount refined	326.024.630
Exported	
The first state of the state of	
Unrefined sold as fuel and road oil	73,210,370

404,135,000

PETROLEUM MARKETED IN THE UNITED

Year	Pennsyl- vania and New York	Ohio	West Virginia	California	Kentucky and Tennessee	Colorado	Indiana	Illinois
	Barrels	Barrels	Barrels	Barrels	Barrels	Barrels	Barrels	Barrels
1859	2,000							<mark></mark>
1830							· · · · · · · · · · · · · ·	
1861								• • • • • • • • • • •
1862	3,056,690					• • • • • • • • • • •		• • • • • • • • • • •
1863								
1864								
1865								
1863	3,597,700				•••••		• • • • • • • • • • • • •	• • • • • • • • • • • • • • • • • • •
1867	3,347,300				•••••	• • • • • • • • • • • •	• • • • • • • • • • • •	• • • • • • • • • • •
1868								
1869	4,215,000				•••••		· · · · · · · · · · · · · · ·	••••
1870	5,200,745		• • • • • • • • • • • • •			•••••	• • • • • • • • • • • •	• • • • • • • • • • • •
1871	5.200,234				•••••	•••••		• • • • • • • • • • •
1872	0,293.194					• • • • • • • • • • • •	•••••	•••••
1873 1874						• • • • • • • • • • • • •	• • • • • • • • • • • •	
1875	0 007 514			•••••				
1875	8,181,914	01 700	100.000					
1877	8,968.906	31,763	120,000					
	13,135,475	29,888	172,000				• • • • • • • • • • • • • • • • • • • •	
1878	15,163.462	38,179	180,000					
1879	19,685,176	29,112	180,000					
1880	26,027,631	38,940	179,000					
1881	27.376,509	33,867	151,000					
1882	30,053,500	39,761	128,000			• • • • • • • • • • • •		
1883	23,128,389	47,632	126,000	142,857				
1884	23,772,209	90,081	90,000					
1885	20.776,401	661,580	91,000		5,164	• • • • • • • • • • • •	······	• • • • • • • • • • •
1886	25,798,000	1,782,970	102,000	377,145				
1887	22,356,193	5,022,632	145,000	678,572	4.791			
1888	16,488,668	10,010.868	119,448		5 096	297,612		
1889	21,487,435	12,471,466	544,113	303,220	5,400			1,
1890	28,458,208	16,124,656	• 492,578	307,360	6,000	368,842	63,496	
1891	33.009,236	17,740,301	2,406,218		9,000	665,482	133,634	6,
1892	28,422,377	16,362,921	3,810,086		6,500	824,000	698 068	52
1893	20.314,513	16,249,769	8,445,412	470,179	3,000	594,390	2,335,293	40
1894	19,019,990	16,792,154	8,577,624	705,969	1,500	515,746	3,688,666	30
1895 1896	19,144,390	19,545,233	8,120,125	1,208,482	1,500	438,232	4,386,132	20
1890	20,584,421 19,262,066	23,941,169	10,019,770	1,252,777	1,680	361,450	4,680,732	25
1897		21,560,515	13,090,045	1,903,411	322	384,934	4,122,356	50
1899	15,948,464 14,374,512	18,738,708	13,615,101 13,910,630	2,257,207 2,642,095	5,568 18,280	444,383	3,730,907	39
1900	14,574,512	21,142,108				390,278	3,848,182	36
1900	13,831,996	22,362,730 21,648,083	16,195,675 14,177,126	4,324,484 8,786,330	62,259	317,385	4,874.392	20
1902	13,183,610				137,259	460,520	5,757,086	25
1902	12,518,134	21,014,231 20.480,286	13,513,345 12,899,395	13,984,268 24,382,472	185,331 554,286	396,901	7,480,896	
1903	12,239,026	10,900,200				483,925	9,186,411	•••••
1904	12,239,020	18,876,631 16,346,600	12,644,686 11,578,110	29,649,434 33,427,473	998,284 1,217,337	501,763	11,339,124	101.00
1905	11,500,410	14,787,763	10,120,935			376,238	10,964,247	181,08
1905	11,211,606	12,207,448	9,095,296	33,098,598 39,748,375	1,213,548 820.844	327,582	7,673,477	4,397,05
1907	10.584,453	12,207,448	9,095,290 9,523,176			331,851	5.128,037	24,281,97
1909	10,381,493	10,858,797	9,523,176	44,854,737 55,471,601	f727,767 f639,016	379,653	3,283,629	33,686,23
1910	9.848,500	9,916,370	11.753.071	73.010.560		310,861	2,296,086	30,898,33
1910	9,200,673	8.817,112	9,795,464	81,134,391	f468,774 f472,458	239,794 226,926	2,159,725	33,143,36
1911	8,712,076	a8,969,007	12,128,932		1472,458 f484,368		1,695,289	31,317,03
1912	8,865,493	8,781,468	11,567.299	g87,272,593 97,788,525	1484,308	206,052	970,009	28,601,30
1913	9,109,309	8,536,352	9,680,033		1524,508	188,799 222,773	956,095	23.893,89
1915	8,726,483	7,825,326	9,080,033		f437,274	208.475	1,335,456	21,919,74
1010	0,120,400	1,020,320	0,201,190	00,001,000	1401,214	208,479	875,758	19,041,69
			-		and the second s			

a Includes the production of Michigan. b Includes the production of Oklahoma. c Included with Kansas.

d Estimated.

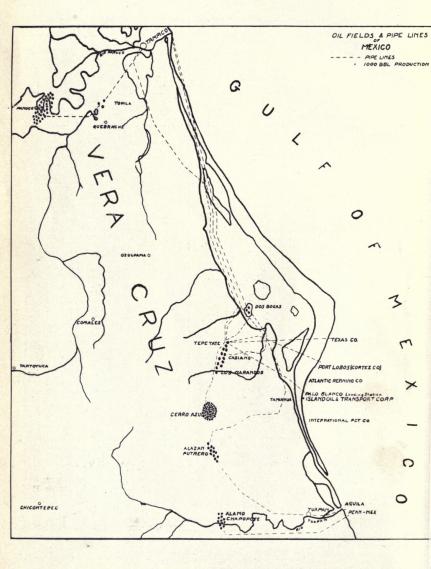
e Includes production of Utah.

KANSAS CITY TESTING LABORATORY

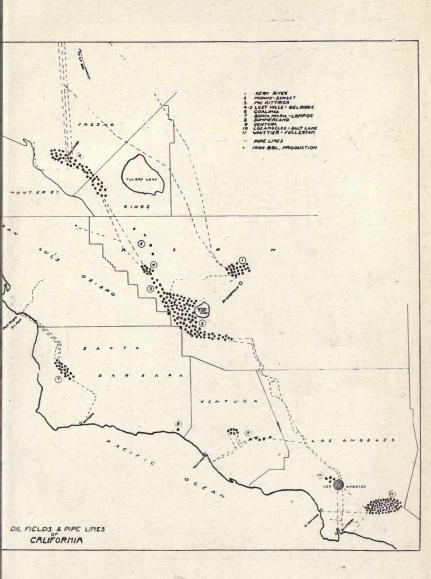
STATES, 1859-1915 (in 42-Gal. Bbls.)

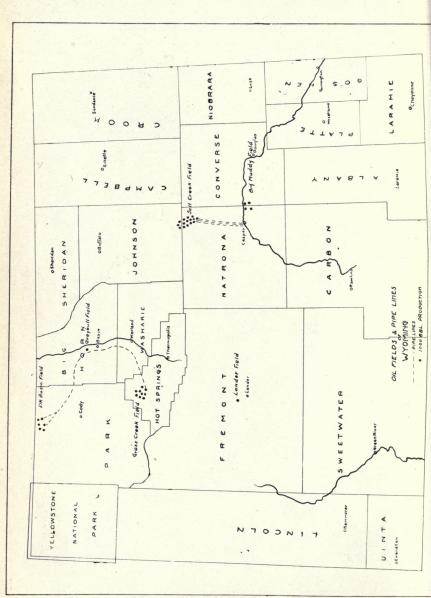
					1	1			
				For Ser Lar	1			101.2	
Kar	isas	Texas	Missouri	Oklahoma	Wyoming	Louisiana	United States	Total Value	Year
Bar	rels	Barrels	Barrels	Barrels	Barre's	Barrels	Barrels		16.5
							2,000	\$32,000	1859
							500,000		1860
							2,113,609		1861
							3,056,690	3,209,525	1862
							2,611,309	8,225,663	1863
							2,116,109	20,893,576	1864
							2,497,700 3,597,700	16,459,853	1865
							3,597,700	13,455,398	1866
							3,347,300	8,066,993	
							3,646,117	13,217,174	1868
							4,215,000	23,730,450	
							5,260,745	20,503,754	1870
					••••		5,205.234	22,591,180	
						•••••	6,293,194	21,440,503	1872
							9,893,786	18,100,464	1873
					•••••		10,926,945	12,647,527	1874
						• • • • • • • • • • • • • • • • • • • •	8,787,514	7,338,133	
			• • • • • • • • • • • • •				9,132,669	22,982,822	
			• • • • • • • • • • • •				13,350,363	31,788.566	1877
			• • • • • • • • • • • • •				15.396,868	18,044,520	1878
							19,914.146	17,210,708	1879
							26,286,123	24,600,638	1880
							27.661,238	25,448,339	1881
							30,349,897	23,631,165	1882
							23.449,633 24,218,438	25.790,252	1883
							24,210,400	20,595,966 19,198,243	1884 1885
							28.064.841	19,196,245	1886
							28,283.483	19,990,313	1887
							27,612,025	17,947,620	1888
	500	48					35.163.513	26,963,340	1889
	1.200	54					45.823.572	35,365,105	1890
100	1,400	54	25				54,292,655	30 526,553	1891
nd -	5,000	45					50,514,657	25,906,463	1892
	18,000	50					48,431,066	28,950,326	1893
	40,000	60			2,369		49,344,516	35,522,095	1894
	44.430	50	10	37			52,892,276	57,632,296	1895
	13.571	1,450		170			60,960,361	58,518,709	1896
	81,098	65,975		625	3,650		60,475,516	40,874,072	1897
	71 980	546,070	10		5,475		55,364.233	44,193.359	1898
	69,700	669.013	132		5 560		57,070,850	64 603,904	1899
	74,714	836,039	a1,602	6,472			63,620,529	75,989,313	
	79,151	4,393,658	a2,335	10,000			69,389,194	66,417.335	1901
	31,749	18.083,658	a757	37,100		548 617	88,766,916	71,178,910	1902
	32,214	17,955,572	a3,000	138,911			100,461,337	94,694,050	1903
	50,779	22,241,413	a2,572	1,366,748		2,958,958	117,080,960	101,175,455	1904
	13,495	28,136,189	a3,100	(c)	8,454	8,910,416	134,717,580	84,157,399	1905
	18.648	12,567,897	a 3,500	(c)	d7,000		126,493 936	92,444,735	1906
	09,521	12,332,696		43,524,128		5 000.221	166,095,335	120,106,749	1907
	01 781	11,206,464	a15,246	45,798,765		5,788 874	178,527,355	129,079,184	1908
	63,764	9,534,467	a5,750	47,859,218		3 059,531	183,170,874	128,328,487	1909
	28,669	8,899,266	a3,615	52,028,718	e115,430 e186,695	6,841,395	209,557,248		1910
1,2	78,819 92,796	9,526,474 11,735,057	a7,995 (h)	56,069,637 51,427,071	e180,699 1,572,306	10,720,420	220,449,391	134,044.752	1911
1,0	92.790 75,029	11,735,057 15,009,478	(II) i10,843	51,427,071 63,579,384	2,406,522	9,263.439	222,935 044	164,213,247	1912
	03,585	20,068,184	j7,792	03,579,384 73.631,724	2,400,522 3,560,375	12.498 828	248,446,230		1913
	23,487	24,942,701	j14,265	97,915,243		14 309,435	265,762,535 281,104,104		1914
									1915
57,7	25 079	228,742,082	86,977	533,394,201	12,210,469	108,086,972	3616561,244	2971388,126	_

f No production in Tennessee recorded. g Includes small production of Alaska. h No production in Missouri; Michigan included in Ohio. i Includes production of Alaska, Michigan and New Mexico. j Includes production of Alaska and Michigan.

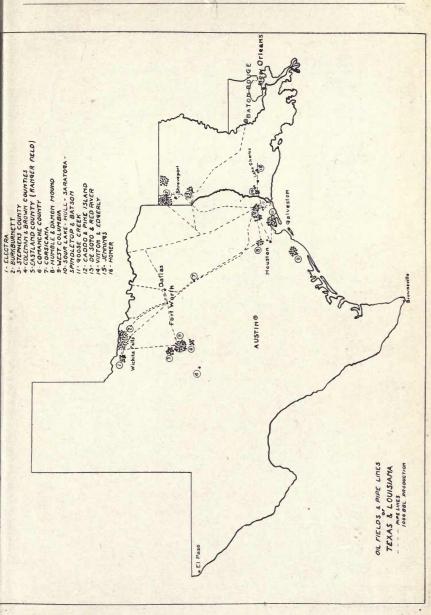


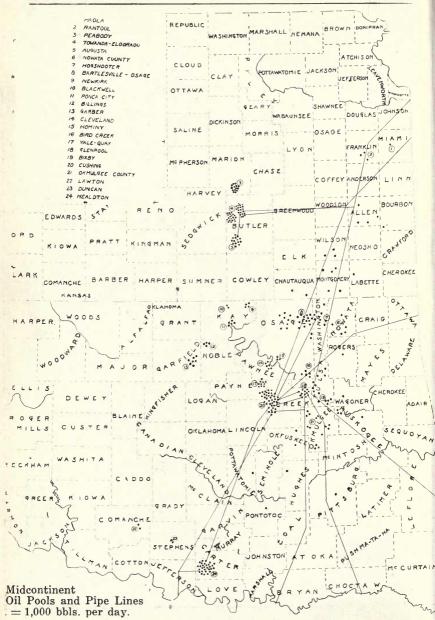
KANSAS CITY TESTING LABORATORY



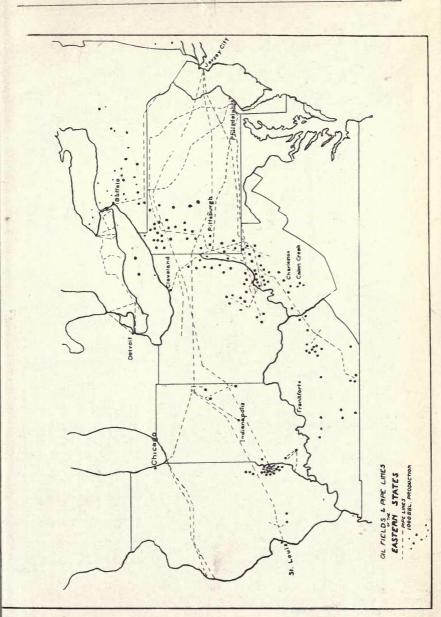


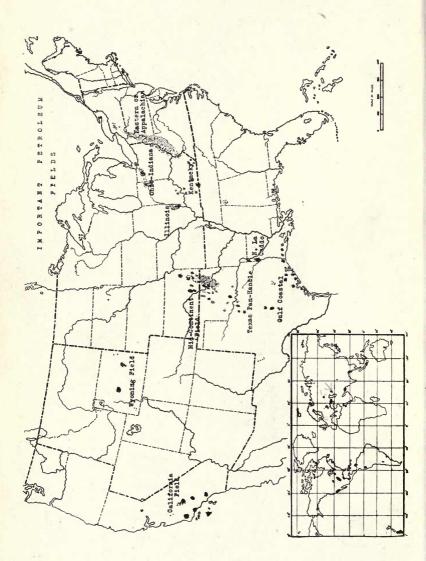
KANSAS CITY TESTING LABORATORY





KANSAS CITY TESTING LABORATORY





Production of Natural Gas-Gasoline in 1917

By Compression and by Vacuum Pumps

		Gas Used.	Avg. Yield
	Gasoline	Estimated	of Gasoline
	Produced,	Volume	per M Cu. Ft.
State No.	Quantity, Gallons	M Cu. Ft.	Gallons
Oklahoma	108,728,213	36,399,280	2.987
California 40	23,478,521	27,477,443	0.854
West Virginia	12,276,784	4,845,648	2.534
Pennsylvania	9,011,199	3,572,356	.2.522
Louisiana	4,459,920	1,468,346	3.037
Illinois 54	4,268,158	2,020,044	2.113
Texas 9	3,997,337	2,685,316	1.489
Ohio 54	2,331,498	836,639	2.788
New York 5			
Kansas 1			
Kentucky 3	369,925	150,784	2.453
Colorado 1			
Totals	168,921,555	79,455,856	2.126

By Absorption

	Gasoline	Estimated	of Gasoline
	Produced.	Volume	per M Cu. Ft.
State No.	Quantity, Gallons	M Cu. Ft.	Gallons
West Virginia 29	20,391,863	163,925,703	0.125
Oklahoma 27	6,395,211	48,320,661	0.132
California 9	5,339,083	17,873,804	0.299
Pennsylvania 17	4,815,051	45,914,700	0.105
Kentucky 2	3,725,893	24,871,590	0.150
Ohio , 7	3,108,062	29,225,502	0.106
Texas 3	2,978,068	10,010,233	0.298
Kansas 5	1,071,633	9,274,289	- 0.116
Illinois 1	665,851	665,851	1.000
Louisiana 2	519,834	675,165	0.770
New York	7,000	2,776	
Colorado			
Totals	49,017,549	349,760,274	0.140

TOTAL GASOLINE FROM NATURAL GAS MARKETED IN THE UNITED STATES IN 1917

					Average Yield of
		Daily		Price per	
	No.	Capacity	Quantity	Gallon	M Cu. Ft.
State	Plants	Gallons	Gallons	Cents	Gas Gals.
Oklahoma	234	492,436	115,123,424	18.71	1.359
West Virginia	188	135,663	32,668,647	19.93	0.195
California	49	99,761	28,817,604	15.40	0.635
Pennsylvania	251	59,164	13,826,250	20.01	0.279
Texas		32,550	6,920,405	16.61	-0.546
Ohio	61	25,137	5,489,560	19.38	0.181
Louisiana	20	20,118	4,979,754	- 16.36	2.323
Illinois	55	17,302	4,934,009	17.55	1.837
Kentucky	5	13,400	3,818,209	19.99	0.153
Kansas		4.642	1.174.980	20.53	0.126
New York					
Colorado	6	2,122	181,262	18.27	2.659
Totals	886	902,385	217,884,104	18.45	0.508

23

m Wield

Gas Used, Avg. Yield

CASINGHEAD GASOLINE INDUSTRY

The growth of the casinghead gasoline industry since 1911 is shown by the following table:

	Plants	Production
1911	. 8	338,058
1912	. 13	1,575,644
1913	. 40	6,462,968
1914	. 58	17,277,555
1915	. 63	31,665,991
1916	. 116	48,359,602

Daily Production of Crude Oil—Various Fields

C liferrie Deile Destation January	1010		Barrels
California Daily Production, January,			275,596
		Production	1
	Producing	per Day	
Kern River		20,460	
McKittrick		7,806	
Midway-Sunset		87,871	
Lost Hills-Belridge	535	13,374	
Coalinga		43,805	
Santa Maria-Lompoc		17,520	
Ventura County-Newhall		4,503	
Los Angeles-Salt Lake		3,979	
Whittier-Fullerton		76,056	
Summerland		147	
Watsonville	5	75	
	0.000		
Totals.	8,606	275,596	
Average value per barrel, \$1.23			
Kentucky Daily Production, January,	1919		21,020
Big Sinking.			,
Pilot.			
Ross Creek			
Ravenna			
Fitchburg		. 1,420	
Zachariah			
Fallsburg		. 250	
Steubenville			
Ragland		. 150	
Parmleyville		. 150	
Cooper		. 150	
Busseyville, Beaver Creek, Cam			
Cannel City, Stillwater, Wage	ersville	. 300	
Louisiana Daily Production, January,	1919		53,200
North Louisiana.			00,200
Caddo and Pine Island			
De Soto and Red River			
South Louisiana.			
Vinton.			
Edgerly.			
Jennings			
		0.00	25 506
Wyoming Daily Production, 1918	• • • • • • • • • • • • • •	. 15 000	35,500
Salt Creek Field			
Grass Creek			
Elk Basin			
Lander.	• • • • • • • • • • • • • •	. 1,000	
Greybull and Basin	• • • • • • • • • • • • • •	. 1,000	
Greyoun and Dasm			

Texas Daily Production, 1919	310,265
High Gravity Crude Oil (North Texas) (Oct. 1).242,890 Burkburnett. 86,000 Eastland (Ranger). 66,100 Electra. 11,000 Stephens County. 44,800 Comanche County and Miscellaneous. 31,350 Petrolia. 750 Holliday. 175 Thrall. 90 Strawn. 500 Moran. 150 Coleman and Brown Counties. 1,000 Northeast Texas. 400 Somerset and Bexar Counties. 300 Piedras Pintas. 100 Iowa Park. 100 Cameron County. 75	
Low Gravity Crude Oil (South Texas) (Oct. 1). 67,375 Goose Creek. 22,000 West Columbia. 15,000 Humble. 10,000 Sour Lake. 8,000 Hull. 5,000 Spindletop. 1,500 Batson. 1,450 Damon Mound. 1200 Corsicana. 900 Markham. 150 Dayton. 25 Miscellaneous. 50	
Mexico Daily Production (Average for 1918)South Fields.Panuco.35,000Ebano.Topila.2,240	177,000
Oklahoma-Kansas (Mid-Continent) Average Daily Production, January, 1919 Washington County— Bartlesville. Hogshooter. Copan-Wann. 481	295,693
Nowata-Rogers Counties— Nowata. 3,487 Delaware. 1,300 Chelsea. 1,200 Inola. 297 6,284 Osage County 31 888 31 888	

Tulsa County—		
Bird Creek	6,807	
Lost City and Red Fork	752	
Broken Arrow and Jenks	2,151	
Bixby and Leonard	3,872	13,582
Okmulgee County—		
	9 500	
Mounds, Beggs and Youngstown Hamilton Switch	2,569 3,541	
Bald Hill.	5,393	
Morris.	2,481	
Tiger Flats.	2,805	
Schulter.	198	
Henryetta	191	17,178
Muskogee and Wagoner Counties-		
Coweta	1,280	
Haskell and Stone Bluff	1,528	
Boynton-Cole.	1,986	
Muskogee	500	5,294
Pawnee County-		
Cleveland.	6,948	6,948
	0,340	0,340
Creek County-		
	41,807	
Cushing-Shamrock	16,801	
Kellvville-Bristow	1,183	
Mannford and Olive	832	60,623
Payne County-	11 000	11 000
Yale, Quay, etc	11,800	11,800
Kay County		
Blackwell.	4,980	
Newkirk and Mervine	4,328	
Ponca City.	581	9,889
Garfield and Noble Counties-		
Billings.	4,029	
Garber	6,400	10,429
Carter County-		
Healdton and Fox	38 803	38 803
Miscellaneous	00,000	2,105
Total Oklahoma		221,214
Kansas—		
El Dorado	46,281	
Augusta		
Outside	14,798	74,479

PRODUCTION AND DECLINE OF IN'DIVIDUAL OIL WELLS Mid-Continent Field, 1916

Total number of wells drilled during year	11,240
Total number of dry holes (including gas)	1,970
Total number of the north man	
Total number with gas	475
Total producing at end of year	9,270
Per cent producing at end of year	92.5%
Average production of this year's producing wells arilled auring	
the year	26 Bbls.
Average production of all this year's producing wells, including	
dry holes	21.5 Bbls.
Total number of wells drilled up to end of this year	81,150
Total number of wells drilled and producing at end of this year	43,420
Per cent of wells drilled now productive	53.2%
Average production of all producing wells in field per day, includ-	
ing this year	8 Bbls.
Average production of all producing wells drilled excluding this	
year	3 Bbls.

OIL WELLS DRILLED IN UNITED STATES IN 1917-1918

	Con	npleted	Di	ry
District	1917	1918	1917	1918
Pennsylvania.	5,435	4,400	985	738
Lima-Indiana	800	793	140	140
Central Ohio	582	605	139	159
Kentucky-Tennessee	1,651	2,191	411	360
Illinois	647	396	151	108
Kansas	3,469	4,671	547	925
Oklahoma-Arkansas,	6,717	8,381	1,334	2,116
Texas Panhandle	1,020	1,140	262	625
North Louisiana	472	534	110	105
Gulf Coast	1,562	1,597	639	625
Total	22,355	24,708	4,718	5,901

OIL WELLS IN MEXICO, 1919

The total number of wells is 1,03	
Wells located	
Wells being driven	
Wells in production	• • • • • • • • • • • • • • • • • • • •
Wells not profitable	
Wells exhausted	
Wells not producing	• • • • • • • • • • • • • • • • • • • •
Total	

The largest number of productive wells belong to the following companies:

Aguila Company (Lord Cowdray) 55
Mexican Petroleum Company of California
The Corona Company 10
Union Petroleum Company Hispano-Americana 17
The Texas Company of Mexico
Mexican Gulf Oil Company
Chicholes Oil Company, Ltd 7
Mexican Combustible Company
Penn. Mex. Fuel Oil Company
Transcontinental Petroleum Company
Oil Fields of Mexico. 12
Un a long of mediatori i i i i i i i i i i i i i i i i i i

RELATIVE ACTIVITY OF OIL FIELDS IN 1918

The rigs and drilling wells, at the close of December, in these fields were as follows:

Field	Rigs	Drillings	Total
Pennsylvania ,	69	143	212
West Virginia.	94	187	281
Southeastern Ohio.	33	104	137
Central Ohio.	31	104	135
Northwestern Ohio	11	49	50
Indiana	2	56	58
Illinois ,	3	51	54
Kentucky	2	494	496
Tennessee		16	16
Arkansas	1	3	+
Kansas	132	481	613
Oklahoma	381	1,408	1,782
Wyoming. ,	106	165	271
Panhandle-Texas.	346	1,097	1,443
Gulf Coast	109	275	384
Louisiana	146	278	424
Totals.	1,456	4,904	6,360

These data show that the chief decline in the amount of oil produced occurs in the first year of the life of the oil well. This decline occurs suddenly after the first gushing due to the sudden local relief of pressure. After this, there is a decline due to the gradual exhaustion of the sand. Every reservoir of oil is limited in capacity by the depth of the sand and the degree of impregnation with oil.

As a general rule, 500 barrels of oil is all that may be expected from each acre for each one foot depth of oil-hearing sand though this varies with the porosity and degree of saturation of the sand.

While the chief general cause for decline of oil wells is exhaustion of the sand, there are many causes that account for a decline in individual wells or localities.

29

Oil Gushers

The largest oil well in the world is one which came in near Tampico, Mexico, February 10, 1916. It was known as Cero Azul No. 4 and was drilled by the Pan-American Petroleum and Transport Company. The first twenty-four hours of oil flow yielded 260,000 barrels. In two years it is said to have produced approximately 60 million barrels of oil or about one-half of the total production of oil from Mexico. Its initial pressure was 1,035 pounds per square inch and the gravity of the oil is 21° Baume' and without sediment or water. This well continued to produce at its usual rate during 1918.

It was in September, 1910, that the Mexican Petroleum Company brought in a well in the Juan Casiano field. It showed on a test that it was capable of giving a daily yield of something more than 100,000 barrels of oil. Pipeline connection was made, however, but not until more than 1,500,000 barrels of the inflammable product had been burned in order to prevent it from flowing into Lake Tamaihua, thus endangering boats and other property. It was throttled down to a flow of 20,000 barrels a day and for more than eight years it has been giving this yield. It has yielded, up to this time, more than 65,000,000 barrels of crude petroleum. Accompanying the oil is a gas pressure of 265 pounds per square inch. This natural gas is piped to the top of a hill a mile and a half distant from the well and is there burned in twelve great flares day and night, lighting up the country for a long way around.

On account of the lack of transportation facilities, it has not been allowed to flow at its maximum, being restrained to one million barrels per month at this time.

A number of wells in the Saboontchy-Romany oil fields of Russia have given daily yields of from 75,000 to 120,000 barrels per day for weeks and as much as 7,500,000 barrels in a year.

Another Mexican well at Dos Bocas, south of Tampico, yielded approximately five million barrels within two months.

A well in the Jennings pool in Louisiana in 1904 is reputed to be the largest gusher in the United States and gave 1,275,000 barrels of oil in four months.

Wells in Texas, California and Roumania have yielded 60,000 to 75,000 barrels of oil per day on the initial production.

The largest wells in the Mid-Continent field were in Butler County, Kansas, where, in the Towanda pool, gushers as large as 25,000 barrels per day, initial production, were struck in 1917.

TABLE SHOWING PRICE PER FOOT FOR DRILLING OIL AND GAS WELLS IN VARIOUS FIELDS

(Oklahoma Geological Survey)

	Feb. 22, 1916	June 23, 1917	July 27, 1917
To shallow sand in Bartles-			
ville, Nowata and Tulsa			
districts	30.80 to \$1.00	\$1.00 to \$1.25	\$1.25
To Layton sand in Cushing			
field	\$1.35	\$1.50	2.50
To Bartlesville sand in Cush-			
ing field, northwest	1.50	2.00	3.50
To Bartlesville sand in Cush-			
ing field, southeast	2.00	2.25	3.50 to \$4.00
To shallow sand in Newkirk,			
Ponca City and Garber			
fields	1.50	1.50	1.50
To deeper sands in Newkirk			
and Ponca City fields (over		E.	
2,500 ft.).	2.50	3.50	3.50 to 4.00
Healdton field	1.40 to 1.50	1.75	1.75
Electra and Burkburnett to			
1,200 ft. depth	2.00		
Electra and Burkburnett to			e for rotary
2,100 ft. depth	8.50		2,000 feet is
Electra and Burkburnett to		\$3.00.	
more than 2,500 ft. depth	5.00		

The regular charge for work by the day, February 22, 1917, was \$50 for a double shift. This held good throughout the above fields. All wildcat propositions some distance (50 miles or more) from any of the above mentioned fields demanded \$3.00 per foot. Contracts were let in 1918-1919, in Pine Island, La., at \$11,000-\$15,000 per well.

Refinery Operations on Crude Oil

		19	17	1918					
Quantity	%	Quantity	%	Quantity	%				
Crude Oil treated,					10				
bbls. ,		315,131,681	100.00	326,024,630	100.00				
Gasoline bbls 49,020,000	19.85	67,990,000	21.58	85,000,000	26.07				
Kerosene, bbls 34,655,000	14.03	41,120,000	13.05	43,450,000	13.33				
Gas and Fuel Oil and									
Loss, bbls134,290,000	54,37	148,900,000	47.25	138,600,000	42.51				
Lubricating Oil, bbls. 14,870,000	6.02	17,945,000	5.69	20,038,000	6.15				
Wax, bbls	0.55	481,200,081	0.48	505,144,357	0.49				
Coke, tons 405,319	1.04	539,366	1.05	559,663	1.05				
Asphalt, tons 716,490	1.83	739,425	1.44	607,968	1.14				
Miscellaneous, bbls 5,696,000	2.31	16,720,000	5.31	15,640,000	4.80				

Miscellanecus includes binder, flux oil, medicinal oil, petroleum, road oil, roofing wax, tar, acid oil, foots oil, motor spirits, pitch, residuum, slops, tar oil, wax oil, wax tailings, straw oil.

REFINERY OPERATIONS BY DISTRICTS FOR FIRST SIX MONTHS OF 1917

	Crude Handled	Per Cent	Per Ce
District		Gasoline	Kerose
Atlantic Coast	. 23,454,900	22.20	22.16
Pennsylvania.	. 8,659,200	24.69	21.43
Illinois ,		35.92	14.5%
Mid-Continent.		26.95	14.45
Gulf Coast	. 27,543,470	12.65	11.3
Wyoming.	4,035,800	37.43	· 17.0
California		11.14	4.2
January-July, 1917	. 143,189,374	20.35	13.04

INCREASE IN PRODUCTION OF CRACKED GASOLINE (Estimated from Still Capacity)

	11	10	01	111	a	50	30	1	т	τ,	U1	11	N	5	11	1	0	10	ŀ	a	i	1	υ.	У.		
Year																							1	1		Barrels
1913	 																									1,000,000
1914	 												•			•										3,000,000
1915	 																									4,000,000
1916	 																									6,000,000
																										9,000,000
1018																									1	8 000 000

KINDS OF GASOLINE AND AMOUNT PRODUCED IN 1917

Natural or Straight Run	54,000,000	bbls.
Artificial or Cracked	9,000,000	bbls.
From Natural and Casinghead Gas-		
By Compression	4,024,000	bbls.
By Absorption	1,167,000	bbls.
Total	68.191.000	bbls.

MARKETED MINERAL PRODUCTS IN THE UNITED STATES IN 1918

	Quantity		Value	World
Refined Petroleum			1,300,000,000	65
Pig Iron (\$33.50 per long ton)	38,820,000	tons	1,304,000,000	30
Copper (24.628c per pound)	1,869,949,686	lbs.	460,500,000	60
Zinc (8.159c per pound)	525,122	tons	85,710,000	30
Lead (6.777c per pound)	550,729	tons	74,640,000	35
Silver (.97875c per ounce)	67,879,206	oz.	66,430,000	50
Gold (\$20.67 per ounce)	3,314,000	oz.	68,493,500	20
Coal.	651,402,374	tons		

PRICES OF CRUDE OIL AT THE WELLS.

Eastern Fields

Mid-Continent Field

Janua 1918 Pennsylvania. \$3.75 Cabell. 2.72 Wooster. 2.38 Corning. 2.80 North Lima. 2.08 South Lima. 2.08 South Lima. 1.98 Princeton. 2.12 Somerset. 2.55	ary 1 1919 \$4.00 2.77 2.58 2.85 2.38 3.38 2.28 2.28 2.60 2.42	January 1918 19 Kansas-Oklahoma. \$2.00 \$2.0	19 25 45 75 75 80 80 80 80
Ragland. 1.20 Illinois. 2.12 Plymouth. 2.03	2.32 2.42 2.33	Batson 1.00 1.5 Saratoga 1.00 1.5	
Canada, Petrolia 2.48	2.78	Louisiana Field Caddo, above 38° 2.00 2.5 De Soto, above 38° 1.90 2.1 Caddo, 35° 1.90 2.1	15
Wyoming Field Elk Basin. 1.85 Grass Creek. 1.75 Big Muddy. 1.40 Salt Creek. 1.40	$1.85 \\ 1.85 \\ 1.50 \\ 1.50 \\ 1.50 $	Caddo, 32°. 1.85 2.1 Caddo, below 32°. 1.00 1.4 Crichton. 1.50 1.5	10 55

California Field

North Texas Field		Kern River, etc. 14-17.9°	.98	1.23
tra 2.00	2.25	18-18.9°	.99	1.24
rietta 2.00	2.25	Ventura County 25-25.9°	1.07	1.32
orsicana, light 2.00	2.25	Fullerton-Whittier		
· icana, heavy 1.05	1.30	16-17.9°		1.23
vn 2.00	2.25	18-18.9° 25-25.9°		1.24
ger 2.00	2.25	37-37.9°		1.57

REFINERY PRODUCTS (1919)

ILEF IN LIVE I ROA			
	Gasoline	Kerosene	Fuel Oil
	Gallon	Gallon	Barrel
At Refinery-Oklahoma	17.2c	8.0c	\$0.75
Kansas City	22.3c	10.8c	1.05
Tulsa		12.0c	1.00
Topeka	22.7c	11.2c	1.05
New York City		14.5c	4.00
Boston		10.7c	4.00
Chicago.		12.0c	1.60
San Francisco and Los Angeles		10.5c	1.60
Seattle. ,		11.5c	1.62
New Orleans.		12.00	2.00
Paraffin Waxn			7½c lb.
	acture point	120	8 1/2 c lb.
		125	9c lb.
		128	11c lb.
		133	13c lb.
		140	17c lb.
Lubricating Oil-		110	110 10.
Natural.		00-	
Black.	• • • • • • • • • • • • • • •	200	per ganon
Onlinden Dole	• • • • • • • • • • • • • • •		per ganon
Cylinder, Pale.		400	per gallon
Cylinder (low cold test)	• • • • • • • • • • • • • •	600	per gallon
Paraffin High Viscosity	• • • • • • • • • • • • • •	···· 40C	per gallon
Asphalt (at market)-	mallam		
50 per cent Asphalt Roal Oil, 7c per	ganon		so per ton
70 per cent Asphalt Road Oil, 8c per	gallon	20.	ou per ton
Texaco Asphalt (Dallas)	• • • • • • • • • • • • • • •	30.	uo per ton
California (San Francisco)			
Mexican (Houston).			
Trinidad (Kansas City)			
Stanolind (Kansas City)			
Stanolind (New York)			
Natural Gas			6c-60c

HIGHEST AND LOWEST PRICES OF CRUDE PETROLEUM OF PENNSYLVANIA GRADE, 1859-1918, PER BARREL

			-	
Year Month	Highest Price	Month	Lowest	Price
Year Month 1859September.	\$20.00			
1860January	20.00	December.		2.00
1861January	1.75			.10
1862December.	2.50			.10
1863December.				2.00
1864July 1865January	14.00			$3.75 \\ 4.00$
1866January.	5.50	December		1.35
1867October	4.00	June.	· · · · · · · · · · · · · · ·	1.50
1868July	5.75	January		1.70
1869January		December.		4.25
1870January		August		2.75
1871June				3.25
1873January.				
1874February.				
1875 February.	1.82	1/2 January.		
1876 December.				1.47 1/2
1877January			<mark></mark>	1.53 3/4
1878February 1879December.	1.87 1.28	¹ / ₂ September.		
1879June			· · · · · · · · · · · · · · · ·	
1881September.				.72 1/2
1882November.	1.37	July		
1883June				.83 1/4
1884January	1.15			
1885October	$\begin{array}{cccccccccccccccccccccccccccccccccccc$			
1886January 1887December.			• • • • • • • • • • • • • • • • • • •	.59 3/4
1888March				
1889November.				
1890January	1.07	% December.		.60 34
1891February.	81	38 August		.50
1892January 1893December.		1/8 October.	• • • • • • • • • • • • • •	.50
1893December.				
1895April.	2.60			
1896January	1.50	December.		.90
1897March				
1898December.	1.19		• • • • • • • • • • • • •	
1899December. 1900January.	1.68		• • • • • • • • • • • •	
1901January, S				
1902December.	1.54	Jan., Feb.,	March	1.15
1903 December.	1.90	Jan., Feb.,	March, April	,
	1.05	May, Ju	ne, July	1.50
1904January		July, Dece	mber	$1.50 \\ 1.27$
1905 October 1906 April, May		May Jan, Feb	March, April	1.27
1900 ipin, may	, ound, oury 1101	Aug., S	Sept., Oct.	
		Nov., De	ec	1.58
1907 March to 1	Dec., incl 1.78			
1908No change		No change		1.78
1909Jan., Feb., 1910January.		June to D	ec., incl	1.43
1911December.	1.35	lanuary fo	December	1 30
1912 December.	2.00	January	· · · · · · · · · · · · · · · · · · ·	1.35
1913 March to	Dec., incl 2.50	January.		2.00
1914January to		September	to Dec., incl.	1.45
1915December. 1916December.			ugust, incl	1.35 2.25
1917August 22			to 5, incl	
1918Feb. 8 to I			Feb. 8, incl.	3.75
CI NEL 2	•			

PRICE SCHEDULE	FOR CALIFORNIA CRUDE OIL-1919
Gravity	Price Gravity Price
14 to 17.9	\$1.23 35 to 35.9\$1.57
18 to 18.9	1.24 36 to 36.9 1.59
19 to 19.9	1.25 37 to 37.9 1.62
20 to 20.9	
21 to 21.9	
22 to 22.9	
23 to 23.9	
24 to 24.9	
25 to 25.9	
26 to 26.9	
27 to 27.9	
28 to 28.9	
29 to 29.9	
30 to 30.9	
31 to 31.9	
32 to 32.9	
33 to 33.9	
34 to 34.9	

PRICE CHANGES OF CRUDE OIL, MID-CONTINENT FIELD, SINCE 1905

1906	\$0.44	April 15	.85
1907	.40	April 27	.80
1908	.39	April 29	.75
1909	.36	September 22	.65
1910	.38	October 1	.55
1911—		1915—	
January 1	.44	February 8	.45
April	.46	February 15	.40
June	.48	August 2	.50
September	.50	August 4	.55
1912—		August 11	.60
January 2	.53	August 19	.65
January 15	.55	August 21	.75
January 25	.57	September 11	.80
February 5.	.60	November 13	.90
April 10	.62	November 15	1.00
April 16	.64	December 13	1.10
May 7	.66	December 14	1.20
May 17	.68	1916—	
July 16	.70	January 20	1.25
November 3	.73	January 26	1.30
November 27	.76	March 4	1.40
December 11	.78	March 11	1.45
December 16	.80	March 14	1.55
December 24	.83	July 24	1.45
1913—		July 29	1.35
January 27	.86	August 1	1.25
January 29	.88	'August 7	1.15
July 7	.93	August 12	1.05
July 21	.98	August 15	.95
August 19	1.03	August 26	.90
1914		November 29	1.00
February 2	1.05	December 12	1.10
April 8	1.00	December 18	1.20
April 10	.95	December 23	1.30
April 13	.90	December 28	1.40

PRICE CHANGES OF CRUDE OIL, MID-CONTINENT FIELD, SINCE 1905—Concluded

1917—		August 16	1.90
January 3	1.50	August 18	2.00
January 6	1.60	1918—	
January 12	1.70	March 18	2.25
August 3		1919	2.25

ACTUAL PRODUCTION BY COMPANIES IN MEXICO

	1918	1917
Companies	Barrels	Barrels
Cia. Pet. La Victoria		1,574
Topila Petroleum Company		2,000
Cia. Mex. Pet. del Golfo		29,993
National Oil Company		753,589
Panuco Petro. Maat. (Royal Dutch)	2,748	
Cia. Exp. de Pet. La Universal	3,075	
Hispano Mexicana (Tex. Mex. Fuel)	4,226	873
Mexico y Espana	5,459	29,625
Mexican Oil Company	3,490	288,770
Cia. Pet. Monterrey	25,021	24,958
Chijoles Oil Ltd. (R. Dutch)	25,266	1,515
Oil Fields of Mexico	29,906	34,689
Veracruz Mexico (S. O. N. J.)	51,716	360,258
La Petrolera Poblana	91,311	32,871
Cia. Mex. de Combustible (Pierce Oil)	300,064	60,852
La Corona (Royal Dutch)	337.603	740,576
	001,000	140,010
Transcontinental de Petroleo (Standard Oil	200.000	110 915
N. J.). Panuco Bost. Oil (Atlan. Ref.)	382,029	119,315
Panuco Bost. Oli (Atlan. Rei.)	531,511	828,067
Tampascas Oil Company	578,478	174,924
Internat. Pet. (J. H. Hamm'd)	609,733	619,828
Cia. Pet. Tal Vez. (So. O. & T.)	1,152,063	989,561
Tex. Co. of Mex. (Texas Co.)	1,279,746	2,315,433
Cia. Mex. de Petroleo (Mex. Pet. of Calif.).	1,445,976	1,125,702
Cia. Mex. de Pet. La Libertad (Island Ó.		
& T.).	1,550,869	
Mex. Gulf Oil (Gulf Oil Co.)	1,728,190	1,160,794
Cortez Oil Corp. (Port Lobos Pet. Corp.)	2,161,775	
East Coast Oil (So. Pac. Co.)	3,457,235	3,143,220
Freeport & Mex. F. O. Corp. (Sinclair Gulf).	4,119,654	4,076,982
Penn Mex. Fuel Co. (South Penn Oil)	6,854,080	4,129,296
Cia. Mex. de Pet. El Aguila (Mexican Eagle	1 1	
Oil).	16,910,646	16,922,322
Oil)	20,186,459	17,325,171
Totals.	63,828,326	55,292,770

Record of All Mexican Operations to Date-1919

Prepared by Mexican Petroleum Department, Secretary of Industry 1 Cubic Meter = 6.29 Barrels

				Potential		
	1 - E	Drillin		Daily		Total
Drilled by	Loca- tions	Feb. 28 1919	ducing	Prod. in	Aban- doned	No. of Wells
La Universal.		1919	1	Cub. Met. 511.00		vvens 2
México y Espana	• ••	-	1	626.00	••	1
T T 11 1 1		• • •	1			1
		• • •	1	8,000.00	• • •	2
Cántabros en Pánuco	• • •	1	••		1	
La Nacional.	• • •	1	• •		••	1
Pánuco Tamesí	. 1	• •	• • •			1
Alamo de Pánuco	. 1	•••	••		1	2
Tux. Ozuluama		2				2
Pet. Maritima					1	1
Freeport & Mex		4	7	5,794.90	2	14
Esfuerzo Tampiqueno					1	1
El Caimán			· · ·		1	1
Pánuco Valley	. 2		1	66.77		3
Southern Co			1	800.00		1
Expl. Topila			1	160.00		1
La Trasatlántica	. 1					1
Pánuco Mahuaves					1	1
Lluvia de Oro		1				1
Esfuerza Nacional					1	2
Vado Oil Fields					1	1
La Victoria			1	6.00		1
Transcontinental.	0	3	12	15,804.04	.7	24
R. A. Mestres				10,001.04		3
English Oil Co		2	4	1,444.00	4	10
El Espino.		1		1,111.00		1
Pedro Irisari.			'i	8.00		1
Tampascas Oil.		1	5	713.00	·i	7
National Pet.		1				i
Gulf Coast Corp	: 'i	1	4	22.69	·i	6
Guil Coast Corp	· 1	••	2	319.00	1	2
Los Perforadores			1		2	3
Hispano Mexicana.			2	1,600.00		3
Tal Vez, S. A		• •		1,155.00		
Monterrey, S. A		• :	1	16.00	•••	1
International Pet		4	3	6,661.22	8	17
Orbananos et al		:	• •		•••	1
Márgenes del Pám		1				1
Pánuco Topila			1	80.00		1
El Fénix, S. A					1	1
Las Dos Estrellas			• • •			1
Productora de Pet		1	1	238.50	1	3
National Oil Co			4	598.90	1	6
Mex. National Oil					2	3
Zaleta Mar Oil Co			• • •		1	1

37

RECORD OF ALL MEXICAN OPERATIONS TO DATE, 1919-Continued

				Potential		
		Drilling		Daily		Total
Drilled by		Feb. 28 1919 (Prod. in Cub. Met.	Aban- doned	
	tions	1919 (lucing			1
La Herradura.			14	1 500 00	1	
Continental Mex	• ••	• • •	1	1,500.00	1	2
El Indio.	• • •	1	• •		• :	1
Lá Oaxaquena	• • •	• •	::		1	1
Oil Fields of Méx		1	12	60.37	23	37
New England Fuel			4	3,900.02	• •	4
La Oriental Méx	. 1				• •	1
La Esperanza		1				1
Abastecedora		1			1	3
Pánuco Excelsior			1	190.00		1
Adrian Petroleum		2	1	5,000.00		4
Cortez Oil Corp	2		2	804.38	1	5
Inglesa Explot.		1		001.00	ī	2
Tantoyuca y Anexas		$\hat{2}$			-	$\overline{2}$
A. P. Wiechers			•••		• •	5
		•••	i	05.40	'i	2
Mex. Pet. del Golfo				95.40		
La Corona, S. A	• ••	4	10	8,095.42	12	26
Byrd, et al		2	• •		• • •	2
Oro Mexicano		•••	• •		1	1
La Bonanza		• •	1	16.00	• •	1
Am. Fuel Oil			2	802.95		2
Topila Petroleum			1	63.60		1
Mexican Gulf	. 2	2	8	22,370.50	8	20
Tampico Pánuco	. 3	2			3	8
Chijoles Oil			7	154.33		7
American Inter			1	4.77	7	8
Hispano Amer.			-			ĭ
East Coast Oil		1	17	4,561.06		27
Soria y Socios		1				1
Texas Co. of Méx	2	3	10	17 079 10	2	17
Marrison Oil Co	1	-	3	17,072.19		
Mexican Oil Co	1	••		639.98	• •	4
Smith's Oil Co		• • •	•••	075.00	1	1
Pan American Oil		1	2	875.00	• •	3
Orillas de Pánuco		1	• :		• •	1
Nuevo León		1	1	15.90	• •	2
Mex. de Combust	. 1		9	5,051.62	6	16
Hispano Cubana			1	397.00		1
M. C. Anderson			2	22.25		2
Piedras Devel. Co					1	1
Lot Seventeen Co			2	6.40		3
Punta Arena y Anex					1	ĩ
Comercio de Puebla					î	ī
La Argentina.					î	$\frac{1}{2}$
México Fuel Oil		1	5	367.13	2	9
		1	9	001.13	4	9 1
Hidalgo Oil Co	• • •	1	1	0.000.00	• • •	
El Nayarit.		• •	1	2,000.00	•••	1
Financiera de Pet		• •	••• =		• :	1
Mex. Development		• :	••		1	1
El Azadón, S. A	• • •	1	••		1	2

RECORD OF ALL MEXICAN OPERATIONS TO DATE, 1919-Continued

	CO	innueu				
				Potential		
		Drilling		Daily		Total
	Loca-	Feb. 28	Pro-	Prod. in	Aban-	No. of
Drilled by	tions	1919	ducing	Cub. Met.	doned	
I G I						-
La Concordia			1			1
Nueva Bonanza					1	1
El Aguila, S. A	. 32	18	55	20,590.18	284	389
Tamiahua Dat			00	20,000.10		
Tamiahua Pet	. 2	1			4	7
Mex. Pet. Co. Cal	. 21	1	33	2,497.65	36	91
Huasteca Pet. Co	. 3	11	4	48,553.70	19	36
Tuxpam Pet. Co		1				1
Man last lis Cl. A	• ••			• • • • • •	••	
Mundacádiz, S. A	• ••	1				1
Juan Casiano Tux	. 1					1
Harry Hummel					2	2
La Tolteca	1					ī
The protocol i T t l			• ;			9
Tampico Oil Ltd	. 1	• •	4	47.00	4	
Tampico Oil Co					1	- 1
Penn Mex. Fuel	. 4	22	7	13,969.35	13	26
La Equidad		1				1
		T	••		••	
Espana, S. A		• •	•••		• :	1
Pet. de Tepetate	. 6		2	21,462.86	1	9
Consolidada de Pet		1				1
Eugenio F. Ruiz		1				1
Bugenio F. Ituiz	• • •				•••	
Seguranza, S. A		2	•••		1	3
La Giralda			2	160.05		2
La Meridional	. 1		1	494.52		2
Tampiquena-San Javier		1	1000			.1
			i		•••	
Tex. Mex. Fuel Oil		••	T	400.00		1
Nacional de Petr		1				1
Mexican Premier		1				1
Eureka			1	1.072.00		2
Diames Transme			1			ī
Pánuco Tuxpam	• • •		_	223.00		
Sun Oil Co	. 1		1	127.20		2
Petrolera Poblana			1	2,400.00		1
La Comercial.			• 1	5.00		3
			_			2
Pánuco Boston			2	1,113.00		
Regiones Pet. Mex			4	3,465.10		4
Puebla en Pánuco	. 1	2			1	4
Allison W. Smith						1
						î
Rodolfo H. Rader	• 1	• :			• :	
Capuchinas Oil		1			1	2
Fomento de Chapala	. 1					1
Mexican Sinclair		5	4	2,951.00	1	11
		1				
Pet. Agríc. Mex. San José	• 1		••		•••	2
Scottish Mex. Oil					5	5
Los Brujos					2	2
Catopico Oil Co						1
		i				i
Dos Banderas Oil			••		••	
Clipton & Smith	. 1		• •			1
Freggs Oil Co					1	1
Hidalgo Pet. Co		1	-			1
W H Milikon			i	3.18		î
W. H. Miliken.	• ••	••			••	1
Ohio Mex. Oil	• ••		1	795.00	• •	1

RECORD OF ALL MEXICAN OPERATIONS TO DATE, 1919-Concluded

		Drilling	2	Potential Daily		Total
	Loca-	Feb. 28		Prod. in	Aban-	
	tions	1919	ducing	Cub. Met.	doned	Wells
Producers Oil Co	. 1	1	2	1,224.30		4
Río Vista					1	1
Sims & Bowser			1	79.50	1	2
Spanish Mex. Oil		1				1
J. W. Sloan		1				1
J. R. Sharp			1	39.75		1
Tampico Banking			2	2.24	· · ·	2
Tampico Fuel Oil			1	127.20		1
Boston Mex. Leasing			1	12,720.00		1
H. McKeever		1				1
Mex. Tex. Pet					1	1
Tamesí Pet. & Asph					2	2
Gobiorno de la Fed			4	3.86	5	9
Fom. del Sureste		1	· · ·		• •	• • •
		-	Transmitting and the			

Name Carter Oil Company Carter and S. W. Oil Co Magnolia Petroleum Company Mid-Kansas Oil Company Prairie Oil & Gas Co Tidal Oil Company		Eldorado Barrels 6,799 9,445 47 1,073	Outside Barrels	Total Barrels 6,953 9,445 3,126 2.108 794 1,073 1,562
Prairie Oil & Gas Co	747	47		794
	1,562			
Empire Gas & Fuel Co I Gypsy Oil Company		$31,376 \\ 18,812$		43,417 18,812
Monitor Oil & Gas Co				$1,539 \\ 251$
Producers' Oil Company C. B. Shaffer	83	1,502		83 1,502
Sinclair Oil & Gas Co		1,940		1,940
Totals. 2 All other companies. 2		$71,025 \\ 14,643$	13,000	92,605 29,256
	23,193	85.668	13,000	121.861

PRODUCTION IN MEXICO TO 1919

Year	Barrels			Barrels
1901	10,345	1910		3,634,080
1902	40,200			12,552,798
1903	75.375	1912		16,558,215
1904	125,625	1913		25,696,291
1905	251,250	1914		26,235,403
1906	502,500	1915		32,910.508
1907	1,005,000			40,545,712
1908	3,932,900	1917		55,292,770
1909	2,713,500			63,828,327
LARGE	PRODUCEE	RS IN CALIF	ORNIA	
L'AROLL A	RODUCE	Per Cent of	Proved Land	Number
Operator		Total Oil	Acres	Wells
Associated Oil Company.		9.1	7,347	1,048
Doheny (various compani	es)	7.3	4,286	379
General Petroleum Corpo	ration	4.3	2,584	400
Honolulu Consolidated Oi			2,701	35
A. T. & S. F. Rv. (oil sub			3,097	412
Shell Company of Califor			2,442	. 236
blieff Company of Camor				

40

LARGE PRODUCERS IN CALIFORNIA—Concluded.
Per Cent of Proved Land Number
Operator Total Oil Acres Wells
So. Pacific Co. (fuel oil department) 8.5 18,267 681
Standard Oil Company
So. Pacific Co. (fuel oil department) 8.5 18,267 621 Standard Oil Company
All others
Total
IMPORTANT OIL COMPANIES OPERATING IN OKLAHOMA,
IMPORTANT OIL COMPANIES OF ERATING IN ORLAHOMA,
CALIFORNIA, WYOMING, KANSAS AND TEXAS
Company Affiliations
Amalgamated Oil Co The Amalgamated Oil Co., the Arcturus Oil Co and the Salt Lake Oil Co. are affiliated and
and the Salt Lake Oil Co. are affiliated and
controlled by the Associated Oil Co. which in
controlled by the Associated Oil Co. which in turn is controlled by the Kern Trading & Oi Co., the producing company of the Southerr
Co., the producing company of the Southern
Pacific Railroad.
Associated Oil CoControlled by the Kern Trading & Oil Co.
Carter Oil CoOwned by Standard Oil Co. of New Jersey.
Cosden Oil & Gas Co Presumably independent. Some of its affiliated
companies are Cosden & Co., Cosden Pipeline
Co., Glenn Pool Pipeline Co., Union Petroleun
Co., Pen-Mar Oil Co.
Pacific Railroad. Pacific Railroad. Controlled by the Kern Trading & Oil Co. Carter Oil CoOwned by Standard Oil Co. of New Jersey. Cosden Oil & Gas CoPresumably independent. Some of its affiliated companies are Cosden & Co., Cosden Pipeline Co., Glenn Pool Pipeline Co., Union Petroleum Co., Pen-Mar Oil Co. Empire Gas & Fuel CoAffiliated with the Empire Refineries, Inc. In an independent concern.
an independent concern.
General Petroleum Corn An independent company.
Gulf Production CoOwned by the Gulf Oil Corporation which is
considered an independent.
Gypsy Oil Co
Humble Oil & Ref. CoAn independent organization.
Invincible Oil Co
Kern Trading & Oil Co A producing company of the Southern Pacific
Guif Production CoOwned by the Guif Oli Corporation which is considered an independent. Gypsy Oli Co
McMan Oil CoSold a controlling interest to the Magnolia
Magnolia Petroleum CoCommonly known as a Standard Oil Co.
Monitor Oil & Gas Co An independent company so far as generally
known.
Ohio Cities Gas Co An independent organization. Has a number o
subsidiaries, some of which are the Ardmore
Refining Co., International Refining Co., Pur-
Oil Co., Complanter Reining Co. and Quake
Ohio Cities Gas CoAn independent organization. Has a number o subsidiaries, some of which are the Ardmor- Refining Co., International Refining Co., Pur- Oil Co., Complanter Refining Co. and Quake Oil & Gas Co.
Ohio Oil CoOne of the Standard Oil group. Pan-American Petroleum &
Pan-American Petroleum &
Transport Co One of the Doneny interest, presumably with
no Standard Oil relations.
Prairle Oil & Gas Co One of the Standard Oil group and was a sub- sidiary of Standard Oil of New Jersey until i
sidiary of Standard Off of New Jersey until I
Was separated theretrom by dissolution decree
of the U. S. Supreme Court in 1911. Producers Oil CoControlled by the Texas Co., 20 per cent of the stock of which the Federal Trade Commission
Froucers on co Controlled by the lexas co., 20 per cent of the
stock of which the rederal Trade Commission
states is owned by the stockholders of different Standard Oil Co.
Ouches Oil & Ges Ge
Quaker on & Gas co
Trolled by Onlo Cities Gas Co.
Republic Production Co A newly organized company in Texas and in
Persona Detroloum do
Shell Co of California A subsidiary of the Royal Dutch Shell group.
Silurian Oil Co. An independent or constantion do for on known
Singlar Oil & Cas Co. An independent organization so har as known.
Sincial on & Gas co An independent company which has acquired in
ent Standard Oil Co. Quaker Oil & Gas CoOriginally controlled by Pure Oil Co. Now con- trolled by Ohio Cities Gas Co. Republic Production CoA newly organized company in Texas and in believed to be independent. Roxana Petroleum CoA subsidiary of the Royal Dutch Shell group. Shell Co. of CaliforniaA subsidiary of the Royal Dutch Shell group. Silurian Oil CoAn independent organization so far as known. Sinclair Oil & Gas CoAn independent company which has acquired a large number of smaller producers. The Sin- clair Oil and Sinclair Gulf are co-interests. Standard Oil Co. (Cal)One of the Standard Oil group.
Standard Oil Co. (Cal.)One of the Standard Oil group.
Sun Co
comparise is not generally known
companies is not generally known. Tidal Oil CoPrincipally owned by Tidewater Oil Co., some
of the stock of which is held by stockholders
in the Standard Oil Co., though presumably
independent.
Wyoming Oil Fields CoSupposedly independent.

Petroleum Refineries of North America

			Approx- imate	Ap. Barrels
Company	Location	Year Built	Invest- ment	Crude Daily
	ALABAMA			
Alabama Oil & Development O		(Bldg.)		
	ARKANSAS			
Ozark Oil & Refining Co	Fort Smith	1914	125,000	300
Control of the second	CALIFORNIA			
Beckett Refining Co	Arroyo Grando	e 1912	1 500 000	
Associated Oil Co Union Oil Co. of Calif	Avilla	1895	1,500,000 9,250,000*	22,000 17,000
Chion on co. of canter.			Union Oil H	Plants)
Phoenix Refining Co	Bakersfield	1902	300,000	1,200
Richfield Oil Co	Bakersfield		200,000	3,500
Slager Refining Co	Bakersfield	1914	200,000	1,200
Standard Oil Co. of Calif Union Oil Co. of Calif	Bakersfield	1895		20,000
Vulcan Oil Co	Bakersfield	1901		400
Canital Refining Co	Berkeley	1900		600
Monarch Oil Refining Co	Berkeley	1910		Idle
Pinal Dome Refining Co	Betteravis	$ 1911 \\ 1895 $	560,000	1,950
Monarch Oil Refining Co Pinal Dome Refining Co Union Oil Co. of Calif Columbian Oil, Asphalt & Ref.	Co. Carnanteria	1895	100,000	10,000
O'Neal Renning Co	Casmalia	1001	100,000	
Puente Oil Co	····Chino	1892	200,000	1,000
American Petroleum Co	····Coalinga	1912	1,250,000	10,000
Shell Co. of Calif	Coalinga		225,000	2,000
Shell Co. of Calif Standard Oil Co. of Calif Paraffin Paint Co	Emoryville	$ 1913 \\ 1895 $	100,000	40,000 300
Wilshire Renning Co	· · · · · Fellows	1912	150,000	10,000
Ventura Refining Co. (L)	Fillmore	1915	650,000	6,000
California-Fresno Oil Co	Fresno	1901	50,000	500
Pacific States Refining Co Anaheim Union Water Co	Fruitvale	1904	50,000	
St. Helens Petroleum Co	Fullerton	• • • •		500 600
Associated Oil Co	Gaviota	1899	530,570	10,000
Associated Oil Co Moore Refining Co California Liquid Asphalt Co	Goleta			
California Liquid Asphalt Co.	Hadley	1909		
Ensign Baker Refining Co Hanford Oil Refining Co	Hadley	1910	43,000	1,000
King Refining Co	Korn Rivor	$1913 \\ 1901$	45,000 175,000	$250 \\ 250$
King Refining Co Producers Oil Refining Co	Kern River	1904	65,000	Idle
Standard Oil Co	Kern River	1914	98,750,000*	65,000
Puelcone Defining C.			O. Plants in	a Calif.)
Buckeye Refining Co	Korn River	$ 1901 \\ 1914 $	100.000	1 500
Warren Bros. General Petroleum Co	Kerto	1914	$100,000 \\ 10,000$	1,500
Amalgamated Oil Co	Los Angeles	1905		10,000
Asphaltum & Oil Refining Co.	Los Angeles			
Atlas Refining Co California Oil & Asphalt Co	Los Angeles	1892	75,000	600
California Oil & Asphalt Co	Los Angeles	1900 1911		450
Continental Oil Co Densmore-Stabler Refining Co.	Los Angeles	1911	100,000	600
Golden State Oil Co	Los Angeles	1902	75,000	650
Golden State Oil Co Fairchild Gilmore Wilton Oil C	oLos Angeles	1912	40,000	700
Guaranty Oil Co Huasteca Petroleum Co	Los Angeles	1900		700
Iordan Oil Co	Los Angeles			1,000
Jordan Oil Co Pioneer Roll Paper Co	Los Angeles	1904	80,000	500
Richfield Oil Co	Los Angeles	1898	200,000	900
Richfield Oil Co Service Oil & Asphalt Co	Los Angeles	1892	100,000	800
L = Lubricating or Wax 1				

PETROLEUM REFINERIES OF NORTH	AMERI	ICA-Cont	inued
TERRODEOM REFINERIES OF NORTH	A FILL FREE	Approx-	Ap.
	Year	imate Invest-	Barrels Crude
Company Location	Built	ment	Daily
	d.		
CALIFORNIA—Concluder Shell Co. (Trumbull Process)Los Angeles Southern Refining CoLos Angeles Turner Oil CoLos Angeles Western Oil Co. of CalifLos Angeles Western Oil Co. (old Atlas)Los Angeles Wilshire Oil Co. (old Atlas)Los Angeles Vernon Oll Co. (old Atlas)Los Angeles Vosemite Oil Refining CoLos Angeles Yosemite Oil Refining CoLos Angeles Union Oil Co. of CalifMaltha Adeline Con. Road Oil CoMaricopa Sunset Monarch Oil CoMaricopa American Oriental Co. (Shell) (L) Martinez Dutch Shell Co. of CalifMartinez General Petroleum CoMojave Union Oil Co. of CalifOleum Richfield Oil CoOilinda Union Oil Co. of CalifOleum Milchfif Refining CoCostend Producers & Refiners Oil CoOil Port Standard Oil CoRodeo Warren BrosRodeo San Diego A-1 Refining CoSan Francisco West Coast Refining CoSan Francisco West Coast Refining CoSant Maria Union Oil Co. of CalifSant Maria Barta CoSant Araia Union Oil Co. of CalifSant Pedro Capital Crude Oil CoSanta Paula Marchus BrosSanta Paula Marchus BrosSanta Paula Marchus BrosSanta Paula Marchus BrosSanta Paula Marchus BrosSanta Paula Marchus BrosSanta Paula			5,000
Southern Refining Co Los Angeles	1900	175 000	700
Union Oil Co. of CalifLos Angeles	1895	175,000	1,100
Western Oil CoLos Angeles		81,000	
Vernon Oil Co. (old Atlas)Los Angeles	1912	21,000	15,000
Yosemite Oil Refining CoLos Angeles	1898	30,000	600
Union Oil Co. of Calif	1012	52 000	3,000
Sunset Monarch Oil CoMaricopa	1907		1,000
American Oriental Co. (Shell) (L). Martinez	1901	265,000	6,000
General Petroleum Co	1915	2,500,000	22,500
Union Oil Co. of CalifOleum			22,000
Richfield Oil CoOlinda	1005		800
Sunset Oil & Refining CoOstend	1903		2,000
Producers & Refiners Oil CoOil Port	1906		5,000
Milriff Refining Co. Rodeo	d 1902		60,000
Warren Bros	1903	80,000	800
San Diego A-1 Refining CoSan Diego	1911	30,000	300
Prutzman Refining Co			300
West Coast Refining CoSan Francisco			
Union Oil Co. of Calif. San Padro	1995		1,000
Capital Crude Oil CoSanta Paula			160
El Merito Refining CoSanta Paula			Idle
A F Gilmore Sherman		150 000	1.000
Tulara Refining CoTulara			
Amalgamated Oil CoVernon	1996	75,000	3,500
British-California Oil CoVernon	1000	50,000	6,000
California Oil & Asphalt CoVernon	1911	125,000	885
General Petroleum Co	1913	1.500.000	20.000
Hercules Oil Refining CoVernon	1900	250,000	1,000
Jordan Oil CoVernon	1907	175,000	700
Pioneer Paper CoVernon			400
Richfield Oil CoVernon	1907	175,000	2,000
National Oil Refining Co	1906	85.000	150
Marchus Bros			1,000
001 001 00			
COLORADO	1010	100.000	1 000
The Inland Refinery	1918	125.000	1,000
Florence Oil Co. (L)	1889	350,000	1,000
Apex Refining & Drilling CoBoulder The Inland RefineryBoulder Florence Oil Co. (L)Florence United Oil Co. (Standard)Florence Urado Oil CoUnitah Basin	1887	1,000,000	3,000
charte on contraint franching basin	1011	10,000	100
FLORIDA			
Jackson E. R. & CoJacksonville	(Bldg.)	150,000	1,500
GEORGIA			
Atlantic Refining CoBrunswick	1919	8,000,000	10,000
IDAHO			
Idaho Oil & Refining CoPocatello	(Bldg)	50 000	
L = Lubricating or Wax Plants.	(Diug.)	50,000	

	AMILIN		An
		Approx-	
	Voon	imate	Crude
Company Location	Year Built	Invest- ment	
ILLINOIS			
Midland Oil & Ref. Co. Allendale	1917		
Barnett Oil & Gas Co. Blue Island	1913	225.000	2.400
Erie Oil & Gas CoBridgeport	1912	35,000	500
Leader Refining CoCasey		250,000	500
Oil Jobbers Prod. & Ref. Co Chicago	1917		
Johnson Oil & Ref. Co Chicago Heights	1916	175,000	1,000
Republic Oil & Ref. Co East Moline	1917	500,000	2,500
Anderson & GustafsonEast St. Louis	1916	5,000	200
Consol. Oil Ref. Co East St. Louis	1915	35,000	300
Indiahoma Refining Co East St. Louis	1907	1,000,000	4,500
St. Clair Gas & Electr. CoEast St. Louis	1914	40,000	Idle
Lubrite Refining Co East St. Louis	1918	50,000	300
Great Northern Ref. Co. (Great			
Lakes Refineries, Inc.)Joliet	1917	300,000	1,500
Central Refining CoLawrenceville	1908 -	9 3,000,000	3,000
Indian Refining CoLawrenceville	1910	1,320,000	11,000
The Texas' CompanyLockport	1911	1,225,000	4,000
Inter Ocean Ref. CoMcCook	1918	250,000	2,000
Wabash Refining Co. No. 1 and 2. Robinson	1907	250,000	800
Smith Oil & Refining CoRockford	1909	75,000	300
Roxana Petroleum CorpWood River	1917	11,500,000	6,000
Standard Oil CoWood River	1912	5,000,000	25,000
ILLINOIS Midland Oil & Ref. Co. Allendale Barnett Oil & Gas Co. Blue Island Erie Oil & Gas Co. Bridgeport Leader Refining Co. Casey Johnson Oil & Ref. Co. Chicago Johnson Oil & Ref. Co. East Moline Anderson & Gustafson. East St. Louis Indiahoma Refining Co. East St. Louis St. Clair Gas & Electr. Co. East St. Louis Great Northern Ref. Co. East St. Louis Great Northern Ref. Co. Lawrenceville Indian Refining Co. Lawrenceville Inter Ocean Ref. Co. McCook Wabash Refining Co. No. 1 and 2 Robinson Smith Oil & Refining Co. Smith Oil & Refining Co. Rockford Roxana Petroleum Corp. Wood River Standard Oil Co. Wood River			
INDIANA			
	(Bldg)	10 000 000	6,500
Sinclair Oil & Ref. CoWhiting Standard Oil Co. of IndWhiting	(Diug.)	25 750 000	60,000
Standard On Co. of Ind whiting	• • • •	20,100,000	00,000
IOWA			
Washington Refining CoCedar Rapids	(Bldg.)	90,000	
		• • • • • • • • • • • • • • • • • • •	
KANSAS			
KANSAS Sinclair Refining CoArgentine	1917		4,500
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City	1917 1906	700,000	3,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City	1917 1906 1914	700,000 300,000	4,500 3,000 2,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City	1917 1906 1914 1917	700,000 300,000 1,150,000	3,000 2,000 6,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta	1917 1906 1914 1917 1917	700,000 300,000 1,150,000 200,000	3,000 2,000 6,000 3,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta	1917 1906 1914 1917 1917 1917	700,000 300,000 1,150,000 200,000 175,000	3,000 2,000 6,000 3,000 1,200
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta	1917 1906 1914 1917 1917 1917 1916	700,000 300,000 1,150,000 200,000 175,000 125,000	3,000 2,000 6,000 3,000 1,200 1,500
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta White Eagle Refining CoAugusta	1917 1906 1914 1917 1917 1917 1917 1916 1917	700,000 300,000 1,150,000 200,000 175,000 125,000 2,000,000	3,000 2,000 6,000 3,000 1,200 1,500 5,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta White Eagle Refining CoAugusta Good Eagle Refining CoBaxter Springs	1917 1906 1914 1917 1917 1917 1916 1917 1917	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 5,000
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta White Eagle Refining CoAugusta Good Eagle Refining CoBaxter Springs Chanute Refining CoChanute	1917 1906 1914 1917 1917 1917 1916 1917 1917 1917 1907	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ \end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 600 1,600
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta Walnut River Ref. CoBaxter Springs Chanute Refining CoChanute Kansas Cooperative Ref. Co. (L).Chanute	1917 1906 1914 1917 1917 1917 1916 1917 1917 1907 1906	700,000 300,000 1,150,000 200,000 175,000 2,000,000 50,000 250,000	3,000 2,000 6,000 3,000 1,200 1,500 5,000 600 1,600
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta Wuite Eagle Refining CoAugusta White Eagle Refining CoAugusta Good Eagle Refining CoAugusta Chanute Refining CoChanute Kansas Cooperative Ref. Co. (L). Chanute Sinclair Refining CoChanute	1917 1906 1914 1917 1917 1917 1916 1917 1917 1907 1906 1907	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ 50,000\\ 250,000\end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 600 1,600 1,000 2,200
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Mulliken Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta White Eagle Refining CoBaxter Springs Chanute Refining CoChanute Kansas Cooperative Ref. Co. (L). Chanute Sinclair Refining CoChanute Sinclair Refining CoChanute Sinclair Chanute Sinclair CoChanute Sinclair CoChanute Sinclair CoChanute Sinclair CoChanute Sinclair Refining CoChanute Sinclair Co	1917 1906 1914 1917 1917 1917 1916 1917 1917 1907 1906	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 10,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ 0\\ 250,000\\ 350,000\\ \end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 1,600 1,000 2,200 1,200
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArgentine Kanotex Refining Co. (National)Arkansas City Lesh Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta White Eagle Refining CoAugusta Good Eagle Refining CoChanute Kansas Cooperative Ref. Co. (L). Chanute Sinclair Refining CoChanute Uncle Sam Oil Co. (L)Chanute Wright Prod. & Ref. CoCharyvale	1917 1906 1914 1917 1917 1917 1916 1917 1907 1906 1907 1906 1917	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ \dots\\ 250,000\\ \dots\\ 350,000\\ 100,000\end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 600 1,600 1,000 2,200 1,200 1,200 1,000
KANSASSinclair Refining Co	1917 1906 1914 1917 1917 1917 1916 1917 1906 1907 1906 1907 1906	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 1,75,000\\ 125,000\\ 2,000,000\\ 50,000\\ 2,000,000\\ 50,000\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\ 0\\$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 1,600 1,600 1,000 2,200 1,200 1,200 1,800
KANSAS Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta Walnut River Ref. CoBaxter Springs Chanute Refining CoChanute Kansas Cooperative Ref. Co. (L). Chanute Sinclair Refining CoChanute Sinclair Refining CoChanute Wright Prod. & Ref. CoCherryvale Waight Prod. & Ref. CoCherryvale Kansas Oil Refining CoCoffeyville	1917 1906 1914 1917 1917 1917 1917 1917 1907 1906 1907 1906 1917 1906	$\begin{array}{c} 100,000\\ 300,000\\ 1,150,000\\ 100,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ \hline \\ 150,000\\ \hline \\ 1500,000\\ 1,550,000\\ 1,550,000\\ \end{array}$	3,000 2,000 3,000 1,200 1,500 5,000 600 1,600 1,000 2,200 1,200 1,200 1,200 1,800 1,800 4,600
KANSAS Sinclair Refining Co	1917 1906 1914 1917 1917 1916 1917 1916 1917 1906 1907 1906 1917 1906 1917 1906 1917 1906	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\\\ 250,000\\\\ 350,000\\ 1,500,000\\ 1,550,000\\ 1,550,000\\ \end{array}$	3,000 2,000 6,000 3,000 1,200 0,5000 6000 1,6000 1,0000 1,2000 1,2000 1,0000 1,2000 1,2000 1,2000 1,2000 1,0000 1,2000 1,2000 1,0000 1,2000 1,0000 1,2000 1,0000 1,2000 1,0000 1,2000 1,0000 1,2000 1,0000 1,2000 1,00000 1,00000 1,0000 1,0000 1,00000 1,000000 1,0000
KANSASSinclair Refining Co	1917 1906 1914 1917 1917 1917 1917 1917 1906 1907 1906 1917 1906 1917 1906 1917	$\begin{array}{c} & & & & & & \\ & & & & & & \\ & & & & & $	3,000 2,000 6,000 3,000 1,200 1,500 5,000 600 1,600 1,200 2,200 1,2
KANSAS Sinclair Refining Co	1917 1906 1914 1917 1917 1916 1917 1907 1906 1907 1906 1917 1906 1917 1909 1916 1907	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\\\ 250,000\\ 100,000\\ 1,500,000\\ 1,550,000\\\\ 250,000\\ 30,000\\ \end{array}$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 1,600 1,600 1,200 1,200 1,000 1,200 1,000 4,600 4,500 2,000 2,000
KANSASSinclair Refining Co	1917 1906 1914 1917 1917 1917 1917 1907 1907 1907 1906 1917 1906 1917 1906 1916 1917	$\begin{array}{c} & & & & & & & \\ & & & & & & & \\ & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & &$	3,000 2,000 6,000 3,000 1,200 1,500 5,000 1,600 1,600 1,000 1,200 1,200 1,200 1,200 2,200 1,800 4,600 2,000 2
KANSAS Sinclair Refining Co	1917 1906 1914 1917 1917 1917 1907 1906 1907 1906 1907 1906 1907 1906 1907 1906 1907 1908 1918	$\begin{array}{c} & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & & & & \\ & & & &$	3,000 2,000 6,000 3,000 1,200 1,500 6000 1,600 1,600 1,200 1,200 1,200 1,200 1,200 2,200 2,200 2,500 4,600 4,500 2,500 4,500 2,500 4,500 2,500 4,500 2,500 4,500 2,500 4,500 2,500 4,500 2,500 4,500 2,500 2,500 2,500 3,500 2,500 3,
KANSAS Sinclair Refining Co	1917 1904 1914 1917 1917 1917 1917 1907 1906 1907 1906 1917 1909 1918 1918 1917 1918 1917	$\begin{array}{c} 700,000\\ 300,000\\ 1,150,000\\ 1,250,000\\ 1,250,000\\ 2,000,000\\ 50,000\\ 2,000,000\\ 50,000\\ 1,250,000\\ 1,500,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 30,000\\ 250,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ 100,000\\ 1,50,000\\ $	3,000 2,000 6,000 3,000 1,200 1,500 5,000 1,600 1,600 1,200 1,000 1,200 1,200 1,000 1,800 4,600 4,600 4,500 2,500 4,000 1,500 1,500 1,000 1,800 1,900 1
KANSAS Sinclair Refining Co	1917 1904 1914 1917 1917 1917 1917 1906 1907 1906 1907 1906 1907 1906 1917 1906 1917 1908 1917 1918 1917	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSAS Sinclair Refining Co	1917 1904 1914 1917 1917 1917 1917 1906 1907 1906 1917 1906 1917 1909 1918 1918 1918 1918 1918 1915	$\begin{array}{c} 100,000\\ 300,000\\ 1,150,000\\ 200,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 250,000\\ 250,000\\ 250,000\\ 250,000\\ 100,000\\ 750,000\\ 125,000\\ 100,000\\ 1$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSASSinclair Refining Co.ArgentineKanotex Refining Co.Arkansas CityJustice ParticleArkansas CityMilliken Refining Co.Arkansas CityMilliken Refining Co.Arkansas CityMugusta Refining Co.AugustaBilss Oil & Ref. Corp.AugustaBilss Oil & Ref. Corp.AugustaWalnut River Ref. Co.AugustaWolte Eagle Refining Co.AugustaGood Eagle Refining Co.Baxter SpringsChanute Refning Co.ChanuteSinclair Refning Co.ChanuteSinclair Refning Co.ChanuteWright Prod. & Ref. Co.CherryvaleWright Ref. Co.CoffeyvilleSinclair Ref.Co.Uncle Sam Oil Co.CoffeyvilleSinclair Ref.Co.Co.El DoradoFidelity Refining Co.El DoradoFidelity Refining Co.El DoradoFidelity Refining Co.El DoradoRailroad Men's Refining Co.El DoradoRailroad Men's Refining Co.El DoradoRailroad Men's Refining Co.El DoradoRailroad Men's Refining Co.HunboldtHutchinson Refining Co.HumboldtHutchinson Refining Co.Independence	1917 1906 1914 1917 1917 1917 1917 1907 1906 1907 1906 1907 1906 1907 1906 1917 1908 1907 1908 1917 1908 1917 1918 1917 1908 1917	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
Sinclair Refining CoArgentine Kanotex Refining CoArkansas City Lesh Refining Co. (National)Arkansas City Milliken Refining Co. (L)Arkansas City Augusta Refining CoAugusta Bliss Oil & Ref. CorpAugusta Walnut River Ref. CoAugusta Walnut River Ref. CoAugusta Good Eagle Refining CoBaxter Springs Chanute Refining CoChanute Sinclair Refining CoChanute Sinclair Refining CoChanute Wright Prod. & Ref. Co. (L)Chanute Kansas Oil Refining CoCherryvale Wright Prod. & Ref. Co. (L)Coffeyville National Refining CoEl Dorado Fidelity Refining CoEl Dorado Fidelity Refining CoEl Dorado Great Western Pet. Corp. (L)Erie Miller Petroleum Refining CoErie	$\begin{array}{c} 1917\\ 1904\\ 1914\\ 1917\\ 1917\\ 1917\\ 1917\\ 1907\\ 1906\\ 1907\\ 1906\\ 1917\\ 1906\\ 1917\\ 1908\\ 1907\\ 1908\\ 1918\\ 1918\\ 1918\\ 1918\\ 1905\\ 1918\\ 1905\\ 1915\\ 1906\\ 1915\\ 1909\end{array}$	$\begin{array}{c} 100,000\\ 300,000\\ 1,150,000\\ 100,000\\ 175,000\\ 125,000\\ 2,000,000\\ 50,000\\ 100,000\\ 1,500,000\\ 1,500,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 1,550,000\\ 250,000\\ 100,000\\ 750,000\\ 100,000\\ 750,000\\ 2,750,000\\ 2,750,000\\ 100,000\\ 2,750,000\\ 100,000$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSASSinclair Refining Co	1917 1906 1914 1917 1917 1917 1907 1906 1917 1906 1907 1906 1906 1907 1906 1918 1909 1918 1915 1909 1905 1909 1905	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSASSinclair Refining Co	1917 1904 1914 1917 1917 1917 1917 1906 1907 1906 1907 1906 1907 1906 1917 1908 1917 1918 1905 1905 1909 1909 1909	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSASSinclair Refining Co	1917 1906 1914 1917 1917 1917 1907 1907 1907 1907 1906 1917 1909 1916 1918 1918 1915 1909 1905 1909 1915 1909 1906	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSASSinclair Refining Co.ArgentineKanotex Refining Co.Arkansas CityLesh Refining Co. (National).Arkansas CityMilliken Refining Co. (L)Arkansas CityMultiken Refining Co.AugustaBilss Oil & Ref. Corp.AugustaBilss Oil & Ref. Corp.AugustaWalnut River Ref. Co.AugustaWolte Eagle Refining Co.AugustaGood Eagle Refining Co.AugustaGood Eagle Refining Co.Baxter SpringsChanute Refning Co.ChanuteSinclair Refning Co.ChanuteSinclair Refning Co.ChanuteWright Prod. & Ref. Co.CherryvaleWright Prod. & Ref. Co.CherryvaleWational Refining Co.El DoradoFidelity Refining Co.El DoradoFidelity Refining Co.El DoradoFidelity Refining Co.El DoradoFidelity Refining Co.El DoradoRefining Co.El DoradoRailroad Men's Refining Co.El DoradoRailroad Men's Refining Co.HumboldtHutchinson Refining Co.HumboldtHutchinson Refining Co.Kansas CityKansas City Refining Co.Kansas CityStandard Asph. & Ref. Co.IndependenceGeneral Refining Co.Kansas CitySinclair Refining Co.Kansas CityStandard Asph. & Ref. Co.IndependenceGeneral Refining Co.Kansas CityStandard Asph. & Ref. Co.MoranaStandard Oil Co. of Kansas.Neodesha </td <td>1917 1904 1914 1917 1917 1917 1907 1906 1907 1906 1907 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1907 1906 1917 1907 1906 1917 1907 1906 1917 1907 1906 1907 1906 1917 1907 1906 1907 1905 1905 1907 1905 1905 1905 1909 1906 1907 1905 1905 1905 1905 1909 1905 1909 1906 1909 1905 1909 1905 1909 1905 1909 1909</td> <td>$\begin{array}{c} & &$</td> <td>$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$</td>	1917 1904 1914 1917 1917 1917 1907 1906 1907 1906 1907 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1906 1917 1907 1906 1917 1907 1906 1917 1907 1906 1917 1907 1906 1907 1906 1917 1907 1906 1907 1905 1905 1907 1905 1905 1905 1909 1906 1907 1905 1905 1905 1905 1909 1905 1909 1906 1909 1905 1909 1905 1909 1905 1909 1909	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSAS Sinclair Refining Co	1917 1904 1914 1917 1917 1917 1917 1906 1907 1906 1917 1906 1917 1906 1917 1909 1916 1918 1905 1919 1918 1905 1909 1906 1917 1905 1909 1906 1917 1905 1905 1905 1905 1905 1905 1905 1905	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
Kansas City Refining Co	$\begin{array}{c} 1917\\ 1906\\ 1914\\ 1917\\ 1917\\ 1917\\ 1917\\ 1907\\ 1906\\ 1907\\ 1906\\ 1917\\ 1906\\ 1918\\ 1908\\ 1918\\ 1918\\ 1905\\ 1909\\ 1918\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1909\\ 1905\\ 1915\\ 1909\\ 1905\\ 1915\\ 1909\\ 1905\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1909\\ 1915\\ 1915\\ 1909\\ 1915\\$	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$
KANSAS Sinclair Refining Co. Argentine Kanotex Refining Co. Arkansas City Milliken Refining Co. (L) Arkansas City Mulliken Refining Co. Augusta Bliss Oil & Ref. Corp. Augusta Biliss Oil & Ref. Corp. Augusta Walnut River Ref. Co. Augusta Wolte Eagle Refining Co. Augusta Good Eagle Refining Co. Baxter Springs Chanute Refning Co. Chanute Sinclair Refning Co. Chanute Sinclair Refning Co. Chanute Vinte Eagle Refining Co. Chanute Sinclair Refning Co. Chanute Vight Prod. & Ref. Co. Chanute Wright Prod. & Ref. Co. Cherryvale Wright Prod. & Refning Co. El Dorado Fidelity Refining Co. El Dorado Fidelity Refining Co. El Dorado Fidelity Refining Co. El Dorado Great Western Pet. Corp. (L) Erie Miller Petroleum Refining Co. Hutchinson Standard Asph. & Ref. Co. Independence General Refining Co. Kansas City Si	$\begin{array}{c} 1917\\ 1906\\ 1914\\ 1917\\ 1917\\ 1917\\ 1906\\ 1907\\ 1906\\ 1907\\ 1906\\ 1907\\ 1906\\ 1917\\ 1908\\ 1907\\ 1908\\ 1918\\ 1917\\ 1918\\ 1906\\ 1919\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1909\\ 1919\\ \end{array}$	$\begin{array}{c} & & & & & & & & & & & & & & & & & & &$	$\begin{array}{c} 3,000\\ 2,000\\ 6,000\\ 3,000\\ 1,200\\ 1,500\\ 600\\ 1,600\\ 1,600\\ 1,000\\ 2,200\\ 1,000\\ 1,200\\ 1,000\\ 1,200\\ 1,000\\ 4,600\\ 4,500\\ 2,500\\ 2,500\\ 1,000\\ 1,500\\ 1,000\\ 1,000\\ \end{array}$

PETROLEUM REFINERIES OF NORTH	AMERI		
		Approx- imate	Barrels
	Year	Invest-	Crude
Company Location	Built	ment	Daily
KANSAS-Concluded			
Red Ball Oll & Ref CoOttawa	1917	75,000	
Red Ball Oil & Ref CoOttawa North American Ref. Co. (operated by Smiley Petrol. Co.)Rosedale Cumberland Refining Co. (Quaker Oil Co.)Wichita Golden Rule Refining CoWichita Sterling Oil & Refining CoWichita Western Refining CoWichita Wichita Indep. Oil & Ref. CoWichita	1915	75,000	1,000
Oil Co.)	1919	400,000	1,000
Golden Rule Refining Co	1917	35,000	1,000
Western Refining Co	1917	35,000	1,000 1,000 5,000 1,200
Wichita Indep. Oil & Ref. CoWichita	1914	200,000	4,000
KENTUCKY			
Standard Oil CoBarbourville	1916	2,500,000	10,000
Standard Oil CoBarbourville Neha Refining CoCompton Jct.	1916 1917	90,000	500
Indian Refining CoGeorgetown	1017	1,500,000	Idle 30,000
Southern Oil Ref. CoLexington	1917	1,500,000	
Melick Refining CoLexington	1917	100,000	1 500
Aetna Refining CoLouisville	1917	1,100,000 1,000,000	3,000 3,500
Standard Oil Co. of KentuckyLouisville	(Bldg.)	1,000,000	3,000
Victor Refining CoLouisville	1917		6 000
Oleum Refining CoPryse	1917	125,000	1,000
Neha Refining Co	1918	125,000 100,000 100,000	1,000
LOUISIANA			
Federal Oll & Def. Ge. Alexandela	1915	150.000	1,000
Federal Oli & Ker. Co	1910	6,000,000	
Pelican Oil & Ref. CoChalmette	1915	225,000	1,200
Red River Refining CoCrichton	1916	200,000 2,000,000	1,000 2,000
Tar Island Oil & Ref. Co	1918	60.000	2,000
Roxana Petroleum CoNew Orleans	1918	1,500,000	5,000
Grops Oil Co. (Dutch Shell Co.) New Orleans	1917	1,500,000 2,000,000	10,000
Freeport & Tampico Fuel Oil Corp. New Orleans	Prop.	2,000,000	10,000
Liberty Oil Co., LtdNew Orleans	1915	40,000	700
Union Refining Co. (Dutch Shell) New Orleans	1917	50.000	300
Southern Oil Co., Inc	1917	20,000	500
Louisiana Oil Refining CoShreveport	1912	1,350,000	2,500
Great Southern Prod & Ref Co. Shrevenort) 1919 1919	100,000	1,000. 1,500
Pine Island Refining CoShreveport	1916	50,000 500,000 300,000	300
Caddo Oil RefineryShreveport	1913	500,000	2,000
Shreveport Oil Ref. Co	1918	300,000 50,000	1,000 1,300
Tar Island Oil & Ref. Co.MooringsportRoxana Petroleum Co.New OrleansFreeport & Mexican Fuel Oil Corp. MerauxCorona Oil Co. (Dutch Shell Co.). New OrleansLiberty Oil Co., Ltd.New Orleans Ref. Co. (Dutch Shell) New OrleansNew Orleans Ref. Co. (Dutch Shell) New OrleansUnion Refining Co.Outhern Oil Co., Inc.PlaquemineLouisiana Oil Refining Co.Superior Oil Works.Malvern (LewisGreat Southern Prod. & Ref. Co.ShreveportShreveportShreveportShreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.Shreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.ShreveportShreveport Oil Ref. Co.ShreveportRio Bravo Oil Co.Welsh	1907	50,000	200
MARYLAND			
Prudential Oil Corp. (L)	1915	3,750,000	10,000
Standard Oil Co. of N. JBaltimore		3,750,000	10,000
Gasoline CorporationCurtis Bay	1917	3,750,000 100,000 250,000	Idle
U. S. Asphalt Ref. Co	1913	250,000 1,000,000	5 000
Prudential Oil Corp. (L)Baltimore Standard Oil Co. of N. JBaltimore Gasoline CorporationCurtis Bay Inter-Ocean Oil Co. (L)E. Brooklyn U. S. Asphalt Ref. CoE. Brooklyn Red "C" Oil Mfg. Co. (L)Highland Town	n	350,000	- 725
MASSACHUSETTS	S Years		
Galena-Signal Oil CoBoston			300
MICHIGAN			12 42
White Star Oil CoDetroit	1917	175,000	400
L = Lubricating or Wax Plants.			

FEIROLEUM REFINER	ILS OF NORTH A	men		
			Approx-	Ap.
		Year	imate Invest-	Barrels Crude
Company	Location	Built	ment	Daily
		2-4-1-C	mone	Duity
The second se	MINNESOTA			
Pure Oil Co	Minneapolis	1917	60,000	400
	MISSOURI			
Wilhoit Refining Co	Joplin	1914	150,000	1,000
Evans-Thwing Renning Co	Kansas City	1917	500,000	4,000
St Los Viscosity Oil & Ref	Co St Joseph	1917	150,000	4,000 500
Wilhoit Refining Co Evans-Thwing Refining Co North American Ref. Co St. Jos. Viscosity Oil & Ref. Standard Oil Co. of Indiana	Sugar Creek	1917	25,000 3,000,000	
	and a second		0,000,000	10,000
	MONTANA			
Dillon Oil Co	Dette		50.000	
Dillon Oli Co	····Butte		50,000	250
	NEBRASKA			
Omaha Oll Refining Co	Omaha		150,000	1,000
	NEW JERSEY			
Columbia Oil Co. of N. Y Standard Oil Co. of N. J. (L) Tidewater Oil Co. (L) Standard Oil Co. of N. J	Bayonne		600,000	1,000
Standard Oil Co. of N. J. (L)	Bayonne	1873	37,000,000	78,000
Tidewater Oil Co. (L)	Bayonne	1879	33,000,000	13,000
Standard Oil Co. of N. J	Bayway	1914	15,000,000	
Vacuum Oil Co., Paving	Bramwell's Pt.	1917	200,000	2,000
Valvoline Oil Co. (L)	Edgewater	1910	25,000	1,000
Galena-Signal Oil Co. (L)	Elizabeth	1901	500,000 500,000	1,000 1,500
Columbia Refining Co	Jersev City	1010	50,000	100
Standard Oil Co. of N. J. (L)	Jersey City	1871	10,000,000	15,000
Barber Asphalt Co	Maurer			
Warner-Quinlan Co	Maurer	1916	25,000	1,000
Vacuum Oil Co., Paving Warner-Quinlan Asphalt Co. Valvoline Oil Co. (L) Galena-Signal Oil Co. (L) Columbia Refining Co Standard Oil Co. of N. J. (L) Barber Asphalt Co Warner-Quinlan Co Vacuum Oil Co. (L)	Paulsboro	1916	3,000,000	10,000
	NEW YORK			
Glandard Old Grand N. W. (I.)			1 959 999	0 500
Standard Oil Co. of N. Y. (L). Mexican Petroleum Co			1,250,000	3,500
		1915	350,000	3,000
Standard Oil Co. of N. Y. (L) Vacuum Oil Co. (L)	New York City	1882	55 000 000	22 000
Vacuum Oil Co. (L)	Olean	1882	5,000,000	12,00
Vacuum Oil Co. (L) Vacuum Oil Co. (L) Wellsville Refining Co. (L).	Rochester		750,000	
Wellsville Refining Co. (L).	Wellsville	1901	664,000	1,000
	OHIO			
Canfield Oil Co	Cleveland	1907	150,000	300
Clarke, Fred G., Co	Cleveland		200,000	1,500
Great Western Oll Co	Cleveland		200,000 100,000	400
Industrial Oil & Ref. Co	Cleveland			
Standard Oil Co. of Ohio	Cleveland	1970	75,000	8 400
Middle West Refining Co	Columbus	1918	125 000	8,400 1,000
National Refining Co	Findlay		250,000	1,200
Craig Oil Co	Ironville	1891	250,000	1,200
Solar Refining Co	Lima	1886	2,575,000	10,000
National Refining Co	Marietta	• • • •	150,000	500
Sterling Oil Works	Marietta			400
fineries)	Newark	1919	300,000	3.000
Rajah Oil & Ref. Co	New Middletown			80
Paragon Refining Co	Toledo	1888	4,500,000 350,000	3,000
Canfield Oll Co Clarke, Fred G., Co Great Western Oll Co Industrial Oll & Ref. Co Standard Oll Co. of Ohio Middle West Refining Co National Refining Co Solar Refining Co Starling Oll Works Ohio Cities Gas Co. (Heath fineries). Rajah Oll & Ref. Co Paragon Refining Co Sun Oll Co Oli Co Oli Refining & Devel, Co Oli Refining & Devel, Co Ohio Valley Ref. Co L = Lubricating or Wax	Toledo	:::::	350,000	4,000
Oil Refining & Devel. Co	Urbana St Marv's	1917	450 000	1.000
Unio Valley Ref. Co	Dianta	1919	300,000	1,000
L = Lubricating or Wax	Plants.			

			Approx-	Ap.
		Year	imate Invest-	Barrels Crude
Company	Location	Built	ment	
	OKLAHOMA			ALC: NO
Crystal White Ref. Co Ardmore Ref. Co. (Ohio Cities) Cameron Refining Co	Allen	1915	$\begin{array}{r} 25,000\\ 1,000,000\\ 250,000\\ 250,000\\ 250,000\\ 250,000\end{array}$	1,000
Cameron Refining Co	Ardmore	1914	1,000,000	6,000
Chickasha Refining Co	Ardmore	1917	250,000	3,000
Chickasha Refining Co Imperial Refining Co Bigheart Petroleum Ref. Co	Ardmore	1917 1908	250,000	2,600
Bigheart Petroleum Ref. Co		1908		
Bixby Oli & Ref. Co Economy Oli & Ref. Co Globe Oli & Ref. Co Modern Refining Co Producers & Refiners Corp Bownton Refining Co	Blackwell	1917	$\begin{array}{c} 100,000\\ 200,000\\ 120,000\\ 175,000\\ 250,000\\ 850,000\\ \end{array}$	2,000 1,500
Globe Oil & Ref. Co	Blackwell	1916 1917 1918	175.000	1.500
Modern Refining Co	Blackwell	1918	250,000	1,500 2,000
Producers & Refiners Corp	Blackwell	1916	850,000	1,850
Boynton Refining Co Continental Refining Co	Bristow	1916 1914	250,000 275,000	2,500 2,000
Oklamade Ref. Co	Chelsea	1918	50,000 500,000 750,000 85,000	300
Great Central Ref. Co	Claremore	1917	500,000	500
American Oil & Tank Line Co	DCleveland	1913	750,000	1,200
Webster Refining Co	Coalton	1913	85,000	650
Superior Oil Ref. Co	Covington	1917	150,000	1,000
Anderson & Gustafson (Hillm	an			
Ref. Co.).	Cushing	1914	27,000	600
Consumers Refining Co. (L)	Cushing	1917	20,000 1,250,000	450 5,000
Cosden & Co	Cushing	1911		2,000
Cushing Acid Works	Cushing			450
Cushing Petroleum Prod. Co.	Cushing	1917	30,000	450
Empire Refineries (Cushing)	Cushing	1910	25,000	Idle 4,000
Federal Refining Co	Cushing	1917	85,000	2,000
Illinois Oil Co	Cushing	1914	175,000 25,000	2,000
Indian Chief Ref	Cushing	1918	25,000	3,000
International Ref Co. (Obio (Cushing	1917	350,000 300,000	5,000
Gas Co.)	····	1010	000,000	
Occident Oil & Ref. Co	Cushing	1916	50,000 651,000	1,000
Peerless Ref. Co. (Empire)	Cushing	1914	651,000	3,000
Kay County Refining Co	Dilworth	1914	95,000	6,000 700
Central Refining Co	Drumright	1917	95,000 15,000 25,000	300
Interstate Oil Refining Co	Drumright	1917	25,000	
Bu-Co Oll & Refining Co	Enid	1917	10,000	1,500
Globe Oil & Ref. Co.	Enid	1917	500.000	5,000
Oil State Refining Co	Enid	1918	250,000	1,200
Southwestern Oil Corp	Enid	1917	75,000 500,000 250,000 85,000	1,500
Garber Rennery	Garber	1918	100,000	600
Carbo Oil Refining Co	Guthrie	1918	100.000	600 100 1,500 2,000
Forty-Sixth Star Ref. Co	Healdton	1917	150,000.	2,000
Terminal Refining Co	Healdton	1917	400,000	2,000
Osage Refining Co	Hominy	1917	10,000 30,000	1,000
Wabash Refining Co	Hominy	1917	100,000	1,500
Southern Refining Co	Haskell	1919	100,000	1,000
Great American Refining Co	Jennings	1917	600,000	3,000
Odessa Oil & Befining Co.	Jennings	1917	250,000 100,000	
Republic Refining Co	Jennings	1918	85,000	
Comanche Oil & Ref. Co	Lawton	1917	125,000	500
Producers & Refiners Corp Boynton Refining Co. Continental Refining Co. Oklamade Ref. Co. American Oil & Tank Line Co. Consolidated Ref. Co. Superior Oil Ref. Co. Anderson & Gustafson (Hillm Ref. Co.). Consumers Refining Co. Consumers Refining Co. Consumers Refining Co. Cushing Petroleum Prod. Co. Dean Oil Co. Empire Refineries (Cushing) Federal Refining Co. Indian Chief Ref. Inland Refining Co. Cocident Oil & Ref. Co. Peerless Ref. Co. May County Refining Co. Cocident Oil & Ref. Co. Constructional Ref. Co. Cortral Refining Co. International Ref. Co. Superless Ref. Co. Central Refining Co. Interstate Oil & Ref. Co. Suchwestern Oil Co. Cocident Oil & Ref. Co. Suchwestern Oil Co. Contral Refining Co. Consouthwestern Oil Corp. Garber Refining Co. Consage Refining Co. Southern Refining Co. Souther	Lawton	1916	365,000	1,550
Birmingham Oil & Gas Co	Muskoree	1917	50,000 1,000,000	
Haskell Refining Co	Muskogee	1917	150,000	
Muskogee Refining Co. (L)	Muskogee	1905	1,350,000	2.100
Haskell Refining Co Muskogee Refining Co. (L) Nupro Refining Co Oklahoma Prod. & Ref. Corp L = Lubricating or Wax	Muskogee	1917	50,000	800
L = Lubricating or Way	Plants	1916	2,000,000	2,000

I MARCONDONA IONA	III IIIIIIII	OI MOINT	LA TRITELA	Approx-	
				imate	Barrels
			Year	Invest-	Crude
Company		Location	Built		Daily
Company	OKLAH	OMA-Contin	Dunc	ment	Dany
Simpleir Oil & Def Ge					0.0.0
Sinclair Oil & Ref. Co.	(Cudany).	.Muskogee	1905		800
Crescent Refining Co.		.Newkirk	1917	200,000	3,000
Dilworth Oil & Refinin	lg Co	. Newkirk	1917		
Ardmore Producing & R Triangle Oil Refining	lefining Co.	.New Wilson	1917	350,000	2,500
Triangle Oil Refining	Co	.New Wilson	1917	35,000	
Carter Oil Co		Norfolk	1916	3,500,000	18,000
Oilton Refining Co		Oilton	1917	15,000	500
Riverside Refining Co.		Oilton	1918	300,000	
Riverside Refining Co., Atwood Refining Co.,		Oklahoma C	ity 1915	350,000	1,250
Capital Refining Co. of	Olzla	Oklahoma C	ity 1915	20,000	300
Empire Pofinaniag (Okl	Dat (a)	Oklahoma C	ity 1915		
Empire Refineries (Okla	t. Rel. (0.)	.Oklahoma C	11y 1900	250,000	2,200
Golden Belt Refining C Home Petroleum Co. (0	.Oklanoma C	ity 1918	200,000	
Home Petroleum Co. (L)	.Oklahoma C	ity 1918	1,500,000	2,500
Naphth-oil Mfg. Co		.Oklahoma C	ity 1918	80,000	300
Sterling Refining Co		.Oklahoma C		200,000	1,000
Allied Refining Co Empire Ref. (American		.Okmulgee	1917	250,000	1,000
Empire Ref. (American	Ref.) (L)	.Okmulgee	1907		4,000
Indiahoma Refining C	0	Okmulgee	1910	1,258,000	3,750
Lake Park Refining C	0	Okmulgee	1915	750,000	2,000
Okmulgee Prod. & Ref	Co	Okmulgee	1916	2,125,000	2,500
Oneta Refining Co		Oneta	1917	40,000	1,300
Limbocker Oil & Ref.	Co	Paul's Valley	y Prop.	150,000	1,000
Osage Mutual Refining	Co	Pawhuska	1917	30,000	
North American Refinir	e Co	Pemeta	1915	200,000	2,500
Empire Ref. (Ponca Re	f. Co.) (T.)	Ponca City	1912		3,504
Lake Park Refining Co		Ponca City	1917	150,000	2,0
Marrland Refining Co.		Ponca City	1918	2,500,000	2,0
Bison Refining Co		Quay	1918	125,000	1,4
Peoples Refining Co		Ringling	1917	100,000	
Empire Ref. (Americar Indiahoma Refining C Lake Park Refining C Okmulgee Prod. & Ref. Oneta Refining Co Limbocker Oil & Ref. Osage Mutual Refining North American Refinit Empire Ref. (Ponca Ref Lake Park Refining Co Bison Refining Co Peoples Refining Co Phonenix Refining Co Phoenix Refining Co Pierce Oil Corporation Wabash Refining Co Golden Glow Refining Co Refining Co.).		Sand Spring	s 1917		~
Phoenix Refining Co.,		Sand Spring	s 1913	350,000	5,0
Pierce Oil Corporation	(L)	Sand Spring	s 1913	2,750,000	9,5
Wabash Refining Co		Sand Spring	s 1917	250,000	5,0
Golden Glow Refining C	o. (Duluth				
Refining Co.)		.Sapulpa	1917	175,000	3,000
Sapulpa Refining Co		Sapulpa	1908	2,000,000	7,500
Victor Refining Co		. Sapulpa	1917	100,000	1,000
Shawnee Refining Co		.Shawnee	1917	100,000	
Mayfield Oil & Ref. Co		Terlton	1918	25,000	1,500
Bliss Oil & Refining Co		Tulsa	1917	3,000,000	
Brazilian Oil & Refinin	g Co	Tulsa	1917	100,000	
Constantin Refining Co		West Tulsa	1911	1,350,000	8,000
Consumers Oil & Refin	ing Co	west Tulsa	1917	340,000	1,200
Cosden & Co. (L)		West Tulsa	1913	47,000,000	15,000
 Golden Glow Refining Co Golden Glow Refining Co Sapulpa Refining Co Victor Refining Co Shawnee Refining Co Shawnee Refining Co Brazilian Oil & Refining Co. Brazilian Oil & Refining Co. Brazilian Oil & Refining Co. Constantin Refining Co Federal Refining Co Federal Refining Co Pan-American Refining Co Pan-American Refining The Texas Company Uncle Sam Oil Co. (V Valley Refining Co Western Products & R White Star Refining Co. 		Tulsa	1917	50,000	
Jaynawker Renning Co	0	Tuisa	1917	100,000	
Mid-Continent Gasoline	e Co	west Tuisa	1916	250,000	4,000
Phoenix Relining Co	••••••••••	Tuisa Weath Walso	1916	300,000	
Pan-American Renning	; Co	West Tulsa	1910	2,000,000	6,500
Uncle Same Oil Co. (M.		West Tuisa	1910	2,350,000	8,500
Volley Defining Co. (Vi	aney)	West Tuisa	1906 1906	150,000	600
Wastown Droducts & D	·····	West Tuisa	1906	150,000 1,000	1,000
White Star Polining Co	er. Co	Wost Tulo	1917	100,000	1,500
Willikon Dof Co. (Sincl	(T)	West Iula	1917	100,000	10,000
Western Products & R White Star Refining Co Canfield Refining Co Home Oll Refining Co Liberty Refining Co Pawnee Bill Oll & Ref Star Refining Co	(L)	Wilson	1910	20,000	1,000
Canfield Refining Co	• • • • • • • • • • • •	Vale	1917	250,000	1,000
Home Oil Refining Co.	••••••	Vale	1916	40,000	2,000
Liberty Refining Co.		Vale	1917	30,000	2,000
Pawnee Bill Oil & Rot	Co	Vale	1916	125,000	1,000
Southern Oil Cornorati	on	Vale	1915	1,000.000	5,000
Star Refining Co		Yale	1916	16,000	600
Sun Company		Yale	1915	600,000	3,000
Star Refining Co Sun Company Superior Refining Co L Ubricating or		Yale	1916	21,000	190
L . ubricating or	Wax Plant	ts.		,050	20
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KANSAS CITY TESTING LABO	ORAT	ORY	49
PETROLEUM REFINERIES OF NORTH A	MEE	RICA Cont	inuad
THIRDELOW HEITHERIED OF NORTH A	The Fat	Approx-	
		imate	
	Year	Invest-	Crude
Company Location	Built		Daily
OKLAHOMA—Concluded			
Victor Refining CoYale Webster Oil & Gasoline CoYale	1916-	17 100 000	1 000
Webster Oil & Gasoline Co Yale	1915	80,000	800
Worth Oil & Refining CoYale	1918	125,000	250
Yale Oil Refining CoYale	1916	-17 100,000 80,000 125,000 30,000	1.000
			-,
PENNSYLVANIA			
Donecker-Hiller Oil Ref. Co. (L)Allentown	1917	50,000	150
Emery Mfg. Co. (L)Bradford	1888	610 500	1.200
Kendall Refining Co. (L)Bradford	1882	495 000	500
Chippewa Refining CoW. Bridgewater	1919	(Bldg.)	
Butler County Oil Ref. Co. (L)Bruin	1911	600.000	800
Valvoline Oil CoButler			1.000
Kendall Refining Co. (L)Bradford Chippewa Refining CoW.Bridgewater Butler County Oil Ref. Co. (L)Bruin Valvoline Oil CoButler East Welbourne Oil Co. (L)Butler	1896	500,000	1,000
Manufacturer's Paraffin CoChester Clarendon Refining Co. (L)Clarendon		$\begin{array}{c} & & \\ & 220,000 \\ & 150,000 \\ & 326,000 \\ & 50,000 \\ & 50,000 \\ & 180,000 \\ & 225,000 \\ & 225,000 \end{array}$	
Clarendon Refining Co. (L)Clarendon	1885	220,000	1,300
Levi Smith, LtdClarendon	1890	150,000	1,050
Levi Smith, LtdClarendon Tiona Refining Co. (L)Clarendon Amber Oll & Realty CoClarendon	1886	326,000	400
Amber Oil & Realty CoClarendon	1915	50,000	150
Canfield Oil Co. (L)Coraopolis Glenshaw Development CoCoraopolis Pittsburgh Oil Refining Co. (L)Coraopolis	1897	180,000	370
Giensnaw Development CoCoraopolis	1000	225,000	600
Robinson Oil CorporationCoraopolis	1892	225,000	1,000
^v ulcan Oil Refining Co. (L)Coraopolis	1000	22,037,000	
ansylvania Oil Prod. Ref. Co. (L)Eldred	1012	200,000	850 500
mlenton Refining Co. (L) Emlenton	1801	500,000	500
Robinson Oil CorporationCoraopolis 'ulcan Oil Refining Co. (L)Coraopolis innsylvania Oil Prod. Ref. Co. (L)Eldred mlenton Refining Co. (L)Emlenton verson Oil WorksErie	1001	000,000	000
ted Oil Manufacturing Co Erie			
ntic Ref. Co. (Eclipse) (L), Franklin	1872		8,000
Oil Co. (L)	1917	175.000	200
Llin Quality Ref. Co. (L)Franklin	1918	100.000	100
ena-Signal Oil Co. (L)Franklin	1869	4,500,000	2,000
nklin Oil Works (L)Franklin	1877	20,000	300
edom Oil Refining Co. (L)Freedom	1889	300,000	1,500
Refining CoGibson's Point			5,000
ansylvania Refining Co. (L)Karnes City	1901	90,000	100
 Ferson Oil WorksErie ted Oil Manufacturing CoErie ntic Ref. Co. (Eclipse) (L)Franklin Oil Co. (L)Franklin Lin Quality Ref. Co. (L)Franklin na-Signal Oil Co. (L)Franklin nklin Oil Works (L)Franklin edom Oil Refining Co. (L)Freedom Refining CoGibson's Point ansylvania Refining Co. (L)Karnes City tarlight Refining CoKarnes City 	1893	60,000	100
Co.) (L)	1890	2,500,000 3,500,000	4,500
Sun Oil Co (L)	1912		6,000 1,000
Island Petroleum Co. (L)Neville Island Advance Oil CoOil City	1912	75,000	300
Ias Berry's Sons (L) Oil City	1911	550,000	2,200
Continental Refining Co. (L) Oil City	1985	275,000	2,200
Jas. Berry's Sons (L)Oil City Continental Refining Co. (L)Oil City Crystal Oil WorksOil City	1886	275,000 250,000 350,000	750 800
Independent Refining Co. (L)Oil City	1882	350,000	1,000
Penn-American Oil Co. (L)Oil City	1892	2,000,000	2.500
Sunrise Oil Co	1917	100.000	
Crew Levick Co Petty's Island			
W. H. Daugherty & Son Ref. Co. Petrolia	1880	125,000	200
Independent Refining Co. (L)Oil City Penn-American Oil Co. (L)Oil City Sunrise Oil CoOil City Crew Levick CoPetty's Island W. H. Daugherty & Son Ref. CoPetrolia Petrolia Refining CoPetrolia	1890	20,000	50
Crew Levick Co. Seaboard			
(Doherty) (L)		500,000	800

Crew Levick Co. Seaboard Philadelphia (Doherty) (L). Philadelphia Sunlight Oil & Gasoline Wks. Philadelphia Atlantic Refining Co. (L). Pittsburgh Chippena Refining Co. (L). Pittsburgh A. D. Millers' Sons Co. (L). Pittsburgh Waverly Oil Works (L). Point Breeze Coldwater Refining Co. Raymilton Empire Oil Works (L). Rouseville Pan-American Refining Co. Rouseville Mutual Sales Co. (L). Russell Amber Oil & Realty Co. Stoneham L = Lubricating or Wax Plants. 500,000 1862 57,000,000 1917 1862 1,125,000 1880 650,000 1866 350,000 250,000 1886 1886 1892 2,000,000 1918 75,000

1 -

130

3,500

1,000

650 42,000

650

800

200 782

3,000

40

Company	Location	Year Built	Approx- imate Invest- ment	Ap. Barrels Crude Daily
	DENNEYL VANIA Const		ment	Dany
Malashine Oll G.	PENNSTEVANIA-CONCI	uueu		
Natural Carolina Co	Strutners		(D14a)	• • • • •
Interior Oil & Cas Co.	monstion	1919	(Bldg.) 50,000	250
American Oil Works (T.) Titusville	1888	350,000	800
Crew Levick (Messim	er plant)Titusville	1912	300,000	780
Crew Levick (Pa. Par	Wks.) (L). Titusville	1905	400,000	800
Muir Oil Works	Titusville			
Titusville Oil Works.	Titusville	1876	210,000	1,000
Fred G. Clarke Co	Warren		500,000	
Conewango Renning	Struthers Tidioute rporationTiona L)Titusville er plant)Titusville Titusville Titusville Titusville Warren CoWarren CoWarren Warren	1895	400,000	450
Cition Con Co	Co. (Onio	1888	1,150,000	2,000
Mutual Refining Co. (T.) Warren	1909	166,800	500
Cities Gas Co.) Mutual Refining Co. (Ohio Cities Gas Co Seneca Oil Works (L) Crew Levick Co. (Gla	Warren	1000	200,000	1,500
Seneca Oil Works (L)	Warren	1893	350,000	560
Crew Levick Co. (Gla	ade Oil			
Wks.) (L)	aue on Warren b. Warren b. Warren c. Warren c. Warren c. Warren c. Warren c. Warren L) Washington	1885	350,000	680
United Oil Refining Co	o. (L)Warren	1902	425,000	500
Superior Oil Works (1)Warren	1901	275,000	400
Wilburing Oil Works	(L) Warren	1890 1897	1,000,000	$1,700 \\ 500$
Beaver Befining Co (L) Washington	1890	115,000	200
Beaver menning co. (L)	1000	110,000	200
	RHODE ISLAND			
Standard Oil Co. of N.	YProvidence	1919	(Bldg.)	
	TENNESSEE			
Lookout Oil & Refining	g CoChattanooga	1917	100,000	1,500
Dixie Refining Co	Memphis	1917	10,000	200
Dixie Refining Co General Ref. & Produc	ing CoNashville	1915	30,000	400
Dix et al		1917	75,000	300
	TEXAS			
Magnolia Petroleum C	o. (L)Beaumont	1902	6,800,000	55,000
United Oil & Rof Co	Decumont	1903	200,000	2,000
Brown Ard Renning C	CoBrownwood CoBrownwood CoBrownwood CoBrownwood CoBrownwood CoBrownwood	1918		600
Carson Oil & Refining	CoBrownwood	$1918 \\ 1918$	40,000 25,000	600 600
Gotebo Oil & Refining	CoBrownwood	1918	50,000	Idle
Hall-Mountain Refinir	CoBrownwood	1918	75,000	1,000
Burkburnett Refining	CoBurkburnett	1918	300.000	2,000
Dilman & Wright	Burkburnett	1918		1,500
Federal Oil & Refining	CoBurkburnett Burkburnett CoBurkburnett ef. CoCisco Cisco CoColeman Corsicana o. (1)Corsicana oWest Dallas	1918	• • • • <mark>• •</mark>	2,000
Burkburnett-Victor Re	ef. CoBurkburnett	1918		2,500
Beaver Valley Oil & Re	ef. CoCisco	1918	100,000	1,000
Long Star Oil & Pof	Colorean Colorean	1918	50,000 50,000	$1,350 \\ 500$
Central Oil Co	Concionan	$1918 \\ 1903$	75,000	Idle
Magnolia Petroleum Co	0. (1). Corsicana	1898	600,000	5,000
Hercules Petroleum C	oWest Dallas	1918	215,000	3,600
Oriental Oil Co. (1) (L)Dallas	1912	500,000	3,500
State Refining Co	Dallas	1919	150,000	
Hercules Petroleum C Oriental Oil Co. (1) (L State Refining Co The Texas Co Dallas Refining Co Eastland Oil & Refinin Great Southern Oil & Beaver-Electra Refining Co.	Dallas	1908		16,000
Dallas Refining Co	DeLeon	1918	300,000	3,500
Creat Southern Oil &	ng Co Eastland	1918		600
Begyer-Flootro Dofini	Ref. CoEastland	$ 1918 \\ 1918 $	250,000	1,200 2,000
Electra Refining Co	Electra	1918	100,000	2,000
Hercules Refining Co.	Electra	1918	125,000	2,000
Robert Lignon	El Paso	1917	25,000	
Hercules Refining Co. Robert Lignon	land Refin-			
mg C0.)	······································	1918		5,000
L = Lubricating of	r Wax Plants.			

FEIROLEUM REFINERIES OF IN	Onthe America		inucu
		Approx-	Ap.
	Term	imate	Barrels
	Year	Invest-	Crude
	tion Built	ment	Daily
TEXAS-Con	cluded		
El Dorado Refining Co	Vorth (s)		5.000
El Dorado Refining CoFort Y Evans-Thwing Refining CoFort Y Federal Refining CoFort Y Gulf Refining CoFort Y Home Oil & Refining CoFort Y Magnolia Petroleum CoFort Y Panther City Oil & Ref. CoFort Y Pierce Oil Corporation	Vorth 1918	(Prop.) 1,500,000	5,000
Federal Refining CoFort V	Vorth 1918	(Prop.)	2,000
Gulf Refining Co Fort V	Vorth 1911	1,500,000	6,000
Home Oil & Refining Co Fort V	Vorth 1919		
Magnolia Petroleum CoFort V	Vorth 1914	1,000,000	15,000
Panther City Oil & Ref. CoFort V	Vorth (s)		1.000
Partner City Oil & Ref. CoFort V Foirce Coll CorporationFort V Southern Oil & Refining Assn. (L).Fort V Texas Producing & Ref. CoFort V Star Refining CoFort V Producers Refining Co. (Empire). Gaines Inland Refining Co. (Empire). Gaines	Vorth 1912	5,600,000	15,000
Southern Oil & Refining Assn. (L).Fort V	Vorth 1918	100,000	500
Texas Producing & Ref. CoFort V	Vorth		4,000
Star Refining Co Fort V	Vorth 1919		1,000
Producers Refining Co. (Empire)Gaines	ville 1915	1,500,000	13,000
Inland Refining Co	m		3,000
Empire Refineries Housto	on		2,000
Hoffman Oil & Ref. Co. (L) Houst	on 1916	150,000	1,000
Petroleum Kenning Co. (L) Housto	JU 1910	1,000,000	1,000
Sinclair-Gulf Corp. (L)Houst Trans-Atlantic Pet. Co. (L)Houst	on 1918	2,500,000	40,000
Trans-Atlantic Pet. Co. (L)Housto	on 1918	150,000	Idle 150
Globe Refining CoHumbl Humble Oil & Refining CoHumbl Mary Owens Oil CoHumbl	le 1916 le 1916	10,000 15,000	600
Many Oweng Oil Co	le 1916	15,000	Idle
Wighita Valley Defining Co. Lowe 1	Park 1914		
Avis Pafining Co	oro 1915	150,000	
Fureka Pafining Co. (tar) I.o. Do	rte 1917	10,000	100
Oriental Oil Co. (tal)	al 1911	10,000	600
Seaboard Oil & Refining Co. (L) · Orange	al e 1917	150,000	
Gulf Refining Co.	Arthur 1901	25,000,000	60,000
The Texas Co (L)	Arthur 1902	50,000,000	32,000
The Texas Co. (L)	Veches 1906		13,000
Odessa Refining Co. (L)	r 1918	100,000	3,600
Mary Owens Oil Co	r		1,000
Ranger Refining CoRange	r 1919	350,000	1,000
Ranger Refining CoRange Dixie Oil & Refining CoSan A	ntonio 1913	240,000	700
Eggleston & ToddSan A	ntonio 1918		1 000
Humble Oil & Ref. Co. (L)San A	ntonio 1913	350,000	1,800
Slump Oil CoSomer	set 1915	25,000	300
Pierce Oil Corp. (L)Texas	City 1911	2,000,000 500,000	5,000
Black Diamond Oil CoThrall	1916	500,000	600
Riverside Oil & Refining CoWaco			1,500
Dixie Oil & Refining CoSan A Eggleston & ToddSan A Humble Oil & Ref. Co. (L)San A Slump Oil CoSomer Pierce Oil Corp. (L)Texas Black Diamond Oil CoTrall Riverside Oil & Refining CoWaco South Bosque Refining CoWaco American Refining CoWaco American Refining CoWichit Banker Petroleum & Ref. CoWichit Gilliland & Fisher (L)Wichit			3,000
American Renning CoWichit	a Falls 1919		0,000
Banker Petroleum & Kel. Co Wichli	a Falls 1918	100.000	1,500 1,200
Gilliand & Fisner (L)	ta Falls 1918 ta Falls 1918	100,000 250,000	2,500
Banhandla Bafning Co	a Falls 1915		5,000
Power Oil & Posning Co Wichit	a Falls 1918		1,000
Banker Petroleum & Ref. Co Wichil Gilliland & Fisher (L) Wichil Lone Star Refining Co Wichil Panhandle Refining Co Wichil Power Oil & Refining Co Wichil Ranger Wichita Oil & Ref. Co Wichil Sunshine State Oil & Ref. Co Wichil Texas Gulf Ref. & Pipeline Co Wichil Victory Refining Co Wichil	a Falls 1918		
Red River Refining Co Wichit	a Falls		500
Sunshine State Oil & Ref Co. Wichit	a Falls 1918	335,000	1,250
Texas Gulf Ref. & Pipeline Co Wichit	a Falls		
Victory Refining CoWichit	a Falls (s)		2,700
	(0)		
UTAH			
Basin Oil Refining CoBasin	1917		
Utah Refining Co	ake City 1907	250,000	800
Utah Oil & Refining CoSalt L	ake City 1916	250,000 1,200,000	1,500
Utah Refining Co	ake City		
Urado Oil CoUintah	Basin		
Dixie Oil Refining CoVirgin	ia City 1918	10,000	100
VIRGIN	A		
Maxican Patroleum Corn Norfol	k 1917		
Gulf Refining CoNorfol Mexican Petroleum CorpNorfol Louisiana Oil Refining CoRichm	ond 1917		
L Lubricating or Way Dlants	1911		

L = Lubricating or Wax Plants.

PETROLEUM REFINERIES OF NORTH	AMER	ICA-Com	inuea
		Approx-	Ap.
	***	imate	Barrels
	Year	Invest- ment	Crude Daily
Company Location	Built	ment	Daily
WEST VIRGINIA			100
Warner-Quinlan CoCairo		1 500 000	400
Ohio Cities Gas CoCabin Creek Jo	et. 1917	1,500,000	4,000 800
Elk Refining Co Falling Rock	1913	100,000	200
Petroleum Products Co	1896	650,000	2,000
Galena-Signal Oll Co. (L) Parkersburg	1893	1,500,000	
Standard Oll Co. of N. J Parkersburg	1913	900,000	1,000
Indiana Defining Co. Staunton	1916		
Company Location WEST VIRGINIA Warner-Quinlan CoCairo Ohio Cities Gas CoCabin Creek Jo Elk Refining CoFalling Rock Petroleum Products CoJacksonburg Galena-Signal Oil Co. (L)Parkersburg Standard Oil Co. of N. JParkersburg Ohio Valley Ref. Co. (L)St. Mary's Indiana Refining CoStaunton			
			05 000
Mid-West Refining CoCasper	1912	25,250,000	35,000
Natrona Pipeline & Ref. CoCasper	(\mathbf{D})	687,767	••••
Northwestern Refining CoCasper	(Bldg.)	3,750,000	40,000
Wyowing Mid-West Refining CoCasper Natrona Pipeline & Ref. CoCasper Standard Oil Co. of IndianaCasper Utah-Wyoming Oil Ref. CoCasper Kinney Oil & Refining CoCosper Northwestern Oil Ref. CoCowley Wyatt Oil & Refining CoDouglas Colorado-Wyoming Ref. CoDouglas	1914		
Utah-Wyoming Oll Ref. CoCasper	(Bldg.)	• • • • • • •	
Northwortown Oil Bof Co. Cowley	1909		20,000
Wratt Oil & Poining Co. Douglas	1918	175,000	500
Colorado-Wyoming Ref Co			
Idaho-Wyoming Oil CoFossil			
Consumers Oil & Refining Co Grevbull	1918	30,000	3,000
Grevbull Refining CoGrevbull	1915	1,500,000	12,000
Standard Oil CoGreybull	1916	1,500,000	5,000
Mid-West Refining CoGreybull	1915	25,000,000	10,000
Glenrock Refining CoGlenrock	(Bldg.)		2,000
Mutual Producing & Ref. CoGlenrock	1918	500,000	800
Wyoming Refining CoGreybull		800,000	
Western Exploration CoLander	1918	250.000	1,000
Wind River Renning Co Lander	1918		5,000
Standard Pergrue Oil Co		2,221,629	0,000
Wyatt Oil & Refining Co.DouglasColorado-Wyoming Ref. Co.DouglasIdaho-Wyoming Oil Co.FossilConsumers Oil & Refining Co.GreybullGreybull Refining Co.GreybullStandard Oil Co.GreybullMid-West Refining Co.GlenrockMutual Producing & Ref. Co.GlenrockWyoming Refining Co.GlenrockWyoming Refining Co.GlenrockMutual Producing & Ref. Co.GlenrockWyoming Refining Co.LanderWind River Refining Co.LaramieStandard Reserve Oil Co.LaramieStandard Reserve Oil Co.RivertonWyoming Refining Co.ThermopolisSouthwest Oil Co.Thermopolis	1918	150,000	1,000
Wyoming Refining Co			
Southwest Oil CoThornton	1918	25,000	100
CANADA	~		
Imperial Oil Co. (L)Dartmouth, N.	S. 1918	2,500,000	
Imperial OII Co. (L)	1914	4,000,000	3,500
Colgory Detroloum Products Itd Okotoka Alt	e. 1917 1915	2,000,000	2,500
Cargary Petroleum Products, Litu. Okotoks, Alt.	1915	$\frac{45,000}{35,000}$	$\frac{30}{25}$
Southern Alberta Ref Ltd Okotoks, Alt	1916	45,000	
Canadian Oil Companies, Ltd. (L), Petrolia	1909	1,500,000	
Canadian Oil Prod. & Ref. Co. (L). Petrolia	1919	200,000	
British Columbia Ref. CoMoody, B. C.	1902	1,000,000	
Continental Oil CoRegina, Sask.		1,000,000	
Imperial Oil Co. (L)Regina, Sask.	1916	2,000,000	2,500
Imperial Oil Co. (L)Sarnia	1898	25,000,000	20,000
British-American Oil Co. (L) Toronto	1906	1,500,000	800
Imperial Oil Co. (L). Dartmouth,N. Imperial Oil Co. (L). Doco, B.C. Imperial Oil Co. (L). Montreal, Que Calgary Petroleum Products, Ltd. Okotoks, Alt. Canada Southern Oil & Ref. Co. Okotoks, Alt. Canadian Oil Companies, Ltd. (L).Petrolia Canadian Oil Prod. & Ref. Co. (L).Petrolia British Columbia Ref. Co. Moody, B. C. Continental Oil Co. (L). Regina, Sask. Imperial Oil Co. (L). Sashia British-American Oil Co. (L). Toronto Great Lakes Oil & Ref. Co. (L). Wallaceburg	1910	200,000	250
MEXICO			
MEXICO Atlantic Refining CoPort Lobos (Cia. Refinadores y Productori de Petroleo J Port Lobos			10,000
(Cia. Refinadores y Productori de Petroleo I	La Atlan	tica)	10,000
Texas Co			
Texas Co Port Lobos Mexican Eagle Co., Ltd Puerto Minatitian			
willatitall	1908		15,000
(Isthmus' of	f Tehaun	tepec)	
La Corona Petroleum CoTampico (plan	ns)		6,000
Dience Oil Composition	1914		12,500
Hussters Petroleum Co	1896		10,000
La Corona Petroleum Co	1915	• • • • • •	
Texas Co.	1919		6,000 6,000
Texas Co	1010	• • • • • • • •	
Pierce Oil CorporationVera Cruz	1914 1918		2,500
L = Lubricating or Wax Plants.		1	

Texas Oil Companies With Production in April, May and June, 1919

F	roduction,	F	Production,
Company and Address.	barrels.	Company and Addres's.	barrels.
Abner Davis, Wichita		Broome Oil Co., Brown-	
Falls	2,350.78	wood	2,077.29
Falls American Reclaim Oil		wood Bullington, Orville, Wich-	
Co., South Houston	2,796.74	Ita Falls	40,886.02
Arlington Oil Company,	FOF OIL	Bishon Evans Oil Co	
Arlington	595.04	Wichita Falls	11,244.43
top Okla	3,080.58	Brock-Lunday Oil Co.,	736.00
Arlington Anderson Oil Co., Law- ton, Okla. A No. 1 Oil Co., Law-	0,000.00	Bowie Big Four Oil Co., Sour	130.00
ton, Okla.	1,528.49	Lake	21,933.91
Adams, Brown & McAlis-	-,	Lake B. O. O. G. Oil Co.,	22,000.01
Adams, Brown & McAlis- ter, Wichita Falls	37,783.83	Iowa Park Brazos River Oil Corp.,	861.07
Abilene-Brownwood Oil		Brazos' River Oil Corp.,	
Co., Abilene Anna Zip Oil Association, Brownwood	251.04	Fort Worth Buchanan, S. R., Batson Barkley, T. G., Sour Lake	92,263.55
Anna Zip Oil Association,		Buchanan, S. R., Batson	20,455.99
Brownwood	1,994.59	Barkley, T. G., Sour Lake	1,536.29
Allday Oil Co., Wichita	1 701 70	Dig Flow On Co., wichita	9 994 50
Falls	1,761.72	Falls Big Burk Oil & Gas Co.,	3,334.50
Art Oil Co., Wichita Falls	1,383,13	Wichita Falls	4,841.47
Arcade Oil Co., Beaumont	763.46	Bradley, E. L., Beaumont	1,613.00
Aikin, L. H., San An-		Burkburnett - Van Cleve	.,
tonio	1,533.22	Oil Co., Wichita Falls	1,622.19
Adams Oil Co Wichita		Burnett Petroleum Co.,	
Falls	12,868.68	Wichita Falls	12,104.35
Amalgamated Oil Co.,		B. C. Oil Co., Wichita	
Wichita Falls Ada Bell Oil Co., Inde-	3,856.28	Falls Big Seven Oil Company, Wichita Falls	246.12
Ada Bell Oll Co., Inde-	00 154 05	Big Seven Oll Company,	0 000 00
pendence, Kan.	20,154.65 6,823.58	Rig Three Oil Company	6,622.08
Acorn Oil Co., Beaumont Annox Oil Co., Beaumont Apple, Dunlap & Sykes,	14.18	Big Three Oil Company, Wichita Falls Brown Oil Co. No. 1, Wichita Falls	2,286.18
Apple, Dunlan & Sykes.	11.10	Brown Oil Co. No. 1.	4,400.10
Aramore, Ukla	7,805.83	Wichita Falls	4,150.65
Abernathy Oil & Gas Co.,	.,	Bowman, S. M., Brown-	
Abernathy Oil & Gas Co., Wichita Falls	1,833.15	wood	258.81
Burk-Star Oil Co., Wich-		Bowman & Williams,	
ita Falls Bartles & Jones, Ranger	4,773.30	wood Bowman & Williams, Brownwood Burkburnett Oil & Gas	41.76
Bartles & Jones, Ranger	520.00	Burkburnett Oil & Gas	150.09
Butler-Harper Oil Asso- ciation, Lawton, Okla	1,402.76	Co., Custer City, Okla. Baker Oil Co., Houston.	150.23 2,404.00
Block Six Oil Co Fred	1,102.10	Burnett-Mann Oil Co.	2,101.00
erick. Okla.	19,103.80		192.50
erick, Okla. Big Pool Oil Co., Wichita Falls Block Twenty Oil Co., Wighter Follo		No. 2, Wichita Falls Block Thirty-Six Oil Co.,	
Falls	11,823.49	Wichita Falls	3,411.98
Block Twenty Oil Co.,		Burgess, Burgess &	
withing rais	5,627.21	Chrestman, Dallas	2,368.25
Brundage - Hancock Oil		Burkburnett Production	05 670 00
Co. No. 2, Wichita	7 01 1 00	Diaga Oil & Coa Co	25,679.20
Falls Bradley Bros. Oil Co.,	7,614.30	Burkburnett Production Co., Dallas Biggs Oil & Gas Co., McKinney B. M. C. Oil Co., Electra.	1,115.36
Houston	5,560.00	B M C Oil Co Electra	151.72
Houston Burkburnett - O'Nail Oil	0,000.00	Burkburnett Southern Oil	101.10
Co., Wichita Falls	1,540.15	Co., Wichita Falls	2,378.41
Bowers & Witherspoon, Palestine		Bi-State Oil and Gas Co.	
Palestine	987.40	Granfield, Okla	381.83
Brown & Jones Wichita		Granfield, Okla. Big Burk Oil and Gas Co. No. 1, Wichita Falls	
Falls Oil C	160,987.50	Co. No. 1, Wichita Falls	10,424.73
Falls	0.014.01	Big Jahn Oll Co., Beau-	100 45
Burkdell Oil Co Odell	2,614.61 971.88	mont Burk-Electra Petroleum	106.45
Bernstein, Eli, Dallas	24.64	Co., Dallas	3,769.72
Broome Bros., Brown-	41.04	Central Producing Co.	0,100.15
Broome Bros., Brown- wood	191.25	Central Producing Co., Chickasha, Okla	5,531.31

AN	D JUNE, 19	919—Continued	
1	Production,	F	roduction
Company and Addres's.	barrels.	Company and Addres's.	barrels
		Double Stondard Oil Go	
Chenault, N. B., Wichita	8,660.59	Double Standard Oil Co., Wichita Falls	111.50
Falls	8,000.55	East Batson Oil Co., Bat-	
Wichita Falls	22,965.00	son	18,843.45
Wichita Falls Colony School Well Co.,		son Electra-Burk Oil Co.,	
Eastland Cotton Oil Co., Saratoga	66,777.50		3,910.75
Cotton Oil Co., Saratoga	2,980.18	Engel, Hendrickson &	100.00
Cullinan Oil Association, Ardmore, Okla.	1.366.52	Haron, Wichita Falls Eastland Oil & Ref. Co.,	420.00
Cooknon Collie' Oil Co	1,300.94	Dallas	21,252.56
Wichita Falls	448.71	Eddy Oil Co., Guffey	738.32
Crosbie, J. E., Tulsa	9,762.92	Dallas Eddy Oil Co., Guffey Ellett Oil Co., Wichita	
Wichita Falls Crosbie, J. E., Tulsa Caldwell Oil Co., Okla-			1,180.62
	1,579.74	Elm Hill Oil Co. Corsi-	000.00
	3,649.23	cana. Farabee Oil Co, Wichita	662.00
Okla. Conner, W. E., Wichita	0,040.20	Falls	3,249.68
Falls	10,379.57	Findley-Leach Oil Asso-	0,010000
Crown Oil & Ref. Co.,		Findley-Leach Oil Asso- ciation, Wichita Falls.	1,442.89
Houston	302,065.71	Freedman, Alex, Corsi-	
Cain-Marvin Oil Co., Dal-	1 000 05	cana Fisher - Parker Oil Co.,	187.04
las	1,889.05 1,929.11	Wichita Falls	1,740.14
Church Oil Co., Corsicana Capital Oil & Gas Com-	1,929.11	Frederick Oil Co., Fred-	1,710.11
pany, Hereford	134.56	erick, Okla.	2,467.31
Canada Oil Co., Wichita		Fisher - Parker Oll Co., Wichita Falls Frederick Oil Co., Fred- erick, Okla Four and Four Oil Co., Dallas Farish & Ireland, Hous- ton	
Falls	5,297.03	Dallas	1,269.17
Cozy Oil Corporation,	0 000 00	Farish & Ireland, Hous-	00 120 00
Wichita Falls	2,622.86	ton Farqueharson, C. B., Wichita Falls	22,130.26
Cline, W. D. & Co., Wichita Falls C. Y. T. Oil Co., Beau-	1,334.61	Wichita Falls	6,198.93
C. Y. T. Oil Co., Beau-	-,	Wichita Falls Federal Oil Co. of Texas,	-,
mont	2,198.27	Cleveland, Ohio	1,637.00
Crowell & Gant, Dallas.	19,972.72	Findley-Leach, Wichita	1 110 00
Coalson Bros. & Af- fleck, Brownwood	139.44	Falls Fisher, Gates & Co.,	1,442.89
Castell Oil Co., Houston.	5,708.15	Wichita Falls	185.75
Cass Oil Co. Wichita	0,100.10	Fisher & Gilliland,	100.10
Falls	2,778.08	Wichita Falls	7,618.68
Falls Centerfield Oil Company, Wichita Falls Crescent Oil Co., Wichita		Fowler Farm Oil Co.,	
Wichita Falls	1,611.70	Wichita Falls	30,922.52
Crescent Oll Co., Wichita	1,815,19	Findley-Minnick Oil and Gas Co., Benjamine	6,458.00
Falls Clay, J. D., Houston Couch Winfrey Oil Com-	4,477.36	Foster-Sander Oil Co.,	0,400.00
Couch Winfrey Oil Com-	.,		1,005.16
pany, Wichita Falls Cadillac Oil & Gas Co.,	3,334.00	Floydada Oil Co., Wichita	
Cadillac Oil & Gas Co.,		H'alls	4,872.50
Denton	$406.50 \\ 2,023.92$	Forest Oil Co., Wichita Falls	5,338.73
Castles Oil Co., Corsicana Crowell, L. R., Dallas Castro, M., Brownwood Dale - Knott Oil Co.,	27,126.73	Floyd Oil Co. Electra	520.00
Castro, M., Brownwood.	10.00	Floyd Oil Co., Electra Fowler, M., Wichita Falls Fritz, D. L., Wichita	961.91
Dale - Knott Oil Co.,		Fritz, D. L., Wichita	
Wichita Falls	240.00	Falls	1,498.73
Diplomat Oil Co., Waco.	4,528.27	Fritz, L. W., Wichita	100.05
Diebel Oil Co., Thrall E. Z. Mark Oil Co., Elec-	265.00	Falls Jates, F. M., Wichita	462.35
tra	351.38	Falls	6,432.56
Eclipse Oil Co., Ft. Worth	5,749.02	Gusher Oil Co., Wichita	.,
Eclipse Oil Co., Ft. Worth Excelsior Oil Co., Wichita		Falls	345.88
Falls Davis, L. R., Tulsa	1,634.99	Gladstone Oil and Ref.	
Davis, L. R., Tuisa	4,118.32	Co., Oklahoma City Gulf Coast Oil Corpora-	31,274.14
Davis-Coggins Oil Co., Wichita Falls	1,020.55	tion, Houston	110,734.97
Duggan Oil Co., Dallas.	480.69	Tinsite Oil Co. Frederick	
Developers Oil & Gas Co., Wichita Falls	1.52	Okla. Gatlin, Mrs. M. W., San	17,577.51
Wichita Falls	800.49	Gatlin, Mrs. M. W., San	
Drillers Oil & Gas Co., Wichita Falls	5,842.66	Antonio Gilbert Co., Beaumont	54.12
Wienita Falls	0,042.00	ander Co., Deaumont	7,672.58

The second se	Production,	,	Production,
Company and Address.	barrels.	Company and Address.	barrels.
Guaranty Oil Co., Electra Granite Oil & Gas Co.,	851.08	Itex Oil Co Wichita Falls Invincible Oil Co., Hous-	
Electra	589.34	Independent Oil Co.	60,005.81
Houston1	,999,294.73	Jones, Cham, Waurika,	08.04
Great Dome Oil Co., Wichita Falls	3,290.50	Okla. Julia Oll Co., Sour Lake. Jones, Roy B., Trustee, Wichita Falls Junior Oil and Pipeline Co., Corsicana Jackson, J. S., Trustee, Sour Lake	379.00 4,296.67
Grayburg Oil Co., San Antonio	8,840.01	Jones, Roy B., Trustee, Wichita Falls	4,195.68
Antonio	450,980.14	Junior Oil and Pipeline Co., Corsicana	432.47
Galconda Oil Co., Wichita Falls	47,231.00	Jackson, J. S., Trustee, Sour Lake Janellen Oil Co. Tulsa,	1,980.81
Falls Goodloe-Kennedy Oil Co., Wichita Falls Gem Oil & Gas Co., Iola,	560.00	Janellen Oil Co. Tulsa, Okla.	2,634.06
Kans.	573.91	John and Jeff Oil Co., Wichita Falls	5,245.38
Harvester Oil Co., Wichita Falls	20,630.33	Jones - Light Petroleum Co., Pilot Point	1,399.90
Falls Healdton Oil and Gas Co., Wichita Falls Hoffman Oil & Ref. Co.,	11,511.32	Jacks, A. L. & Co., Bena- vides Kirby Oil Association,	824.00
Houston's Texas Petro-	26,469.44	Keever & Gordon Oil Co	0,000.02
Houston Houston's Texas Petro- leum Co., Houston Hartzell Oil Co., Corsi-	794.64	Beaumont Kribs Oil Co., Wichita	387.00
cana Harvey, R. O., lease, Wichita Falls	357.89	Knotts, F. F., Wichita	4,012.20
Hiawatha Uli Co., Hop-	26,925.43	Falls Kemp, E. R., Tulsa, Okla. K. A. P. Oil Co., Wichita	449.82 12,358.31
Hereford Oil Co., Hereford	$117.50 \\ 2,276.35$	K. A. P. Oil Co., Wichita Falls	1,673.66
Houston Ull Co of Texas	11,535.12	Falls Keim, F. D., Wichita Falls	1,388.00
Houston Hunt, J. C., Wichita Falls Humble Oil and Ref. Co., Houston Heydrick, J. C., Wichita	1,273.34	Knauth Oll Co., Wichita	1,610.24
Heydrick, J. C., Wichita	,490,503.96	Falls Kurz Oil Co., Von Ormy. Kerr, T. P., Corsicana Kemp & Farris, Chilli-	2,190.00 343.66
Falls	74.53 6,328.75	cothe Lone Star Oil Co., Burk-	2,682.40
Falls	7,713.90	burnett Lawton Oil Co., Burkbur-	8,788.22
Falls Hicks, E. P., Wichita Falls	12,325.06	Long, R. A., Association, Wichita Falls	59,030.00
Helen-Elizabeth Oil Co., Wichita Falls	7,446.40	Ligon. Blair & Rowe.	
Harvey Oil Co. Wichita	4,383.62	Alvarado Lake View Oil Co., Sour	4,516.05
Falls Hodge Oil Co., Burkbur- nett Hardin, Willis, Fowler	482.53	Leon Oil Co., Wichita	0,042.00
Hardin, Willis, Fowler & Staley, Burkburnett	194.68	Falls Lone Star Gas Co., Dallas Lyle Oil Co., Mineral	7,302.28 8,856.28
& Staley, Burkburnett Holiday & Gaffall, Beau- mont Haile Oil Co., Wichita	2,775.49	Lyle Oil Co., Mineral Wells	900.00
Halle Oil Co., Wichita Falls High Land Oil & Gas Co.,	336.82	Wells Lake Oil Co., Beaumont. Logan Oil Co., Humble.	33,147.16 1,305.90
Electra	908.34	Lucky Seven Oil Co., Wichita Falls Liberty Oil Association	3,764.97
Hollingsworth, W. E., Brownwood Hall Bros. Oil Co., Brownwood	302.00	Wichita Falls Lord, C. A. & Co., Beau-	3,908.16
Imperial Petroleum Co., Wichita Falls	1,051.00	mont Lee-Graham Oil Co., Sour	1,688.75
Wichita Falls Ilsong-Worth Oil Co	27,493.18	mont Lee-Graham Oil Co., Sour Lake Lone Acre Oil Co., Beau-	5,612.72
Ilsong-Worth Oil Co., Wichita Falls	13,471.39	mont	214.83

AND JUN.	E, 1919—Continued
Production Company and Address.	els. Company and Address. barrels.
Lucky Six Oil Co., Bangs. 782 L. N. Lockridge, Wichita Falls 122	Falls 3,870.68
Leon Valley Oil Co., De	Wichita Falls 1,006.55 Mayfield Adams & Co
Leon 3,398 Liberty Oil & Gas Co., Tulsa, Okla 37,278	Fort Worth 212.15 Mitchel Petroleum Co.,
Minta Oil Co., Saratoga. 979	.80 Mid-Texas Oil Co., Wichi-
Munger-Verchoyle Oil Co., Dallas	40 Matador Oil and Gas
Wichita Falls 3,029 Merrimac Oil Co., Beau-	.83 Minn-Texas Oil Co., Elec-
mont 868 Lary Lou Gile Oil Co.,	Wichita Falls 3,114.02
Wichita Falls	Co., Wichita Falls 3,421.82
Mackeckney Oil Co., Wichita Falls 1,660	Memphis 518.48
Markenin Col, Winner 10,583 Mackeckney Oil Co., 10,660 Wichita Falls 1,660 Minneapolis Oil and Development Co., Minneapolis Oil 1460	Mayflower Oil Co., Ard-
apolis, Minn 154 Magnolia Petroleum Co., Dallas	Nacona-Burk Oil Co.
George A. Martin, Hum- ble	Burkburnett 12,306.00 Northern Oil and Gas 00 Co., Humble 1,542.00
Levely-Maxwell Oil Co., Wichita Falls 322	Nutt, Horace, Wichita Falls 2,330.30
Mills & Garrity, Corsi-	
Morrisey, Shaw & Hey- drick, Wichita Falls, 1.419	sha, Okla 5,767.22
drick Wichita Falls 553	Co., Wichita Falls 3,655.58 Nineteen Oil Co., Beau-
Martin Oil Co., Beaumont Minor Oil Co., Beaumont. McGoldrick, E. W., Bat-	31 Norton Lester L. Indiana
Mann-Isleng Oil Co	apolis, Ind. 1,000.00 .63 Ozark Trail Oil Co., Electra 4,566.61
Maer, W. Newton, Wichita	2.70 O'Neil, John, Wichita
Falls1,278Marnet Oil Co., Corsicana6,748Morris Oil Co., Wichita6,748	Falls 537.00 Odell Oil Co., Wichita S.12 Falls 4,570.00 Oktaba Oil Co., Tulsa, Oktaba Oil Co., Tulsa,
Falls 2,799 Minchew Oil Co., Wichita	2.74 Okla 2,792.51 Okla
Minchew and Street,	Osage Oil and Gas Co.
Wichita Falls 6,288 McNamara Oil Co., Beau- mont 4,220	Old Dominion Oil Co.,
mont	Oriental Oil Co., Dallas. 3,537.83
COMPTODIS, UNIO	ton
Michael Murphy Estate.	Patterson Oil Co., Brown-
Mid-Kansas' Oil and Gas	Palo Pinto Oil Co.
Mennis & Horn, Beau-	Purcell Oil Co., Wichita
mont 1,010 Mann - Power Oil Co., Wichita Falls 6,000	b.19 Falls 4,783.91 Plainview Oil and Gas Co., 6,215.34 8.67 Wichita Falls 6,215.34
	1

Company and Address.	Production, barrels.	Company and Address.	Production, barrels.
Peerless Oil Co., Saratoga	2,934.13	Reliance Oil Co., Beau-	
Petroleum Ref. Co., Hous- ton	103.426.65	mont Reynolds Oil Co., Wichita	11,846.99
Pippin Oil Co., Brown-	6,888,52	Falls Russell-Mann-Frank Oil Co., Wichita Falls	421.44
wood Possum Hill Oil Co.,		Rio Bravo Oil Co., Hous-	4,377.43
San Antonio Perkins, J. J., Wichita	800.00	Red River Oil Co.,	60,426.41
Falls Panhandle Oil Co., Wichi-	8,492.69	Republic Production Co.,	18,124.50
Prime Oil Co., St. Jo	2,580.85 1,282.34	Ryan Petroleum Co.,	244,015.27
Plains Oil and Gas Co	36,675.00	Silurian Oil Co., St. Louis,	59,187.78
Ardmore, Okla Paraffine Oil Co., Beau- mont	11,104.66	Mo. Slaughter & Hutchinson,	12,924.49
mont Paggi Bros. Oil Co., Beaumont	17,216.43	Sinclair Gulf Oil Co.,	100.20
Beaumont Palmer Oil Co., Henri-	602.00	Tulsa, Okla. Sinclair Gulf Oil Co. No.	22,786.53
etta Peerless Oil Co., Dallas. Prather, Ad. "Special,"	5,666.79	Simoloin Cult Oil Co No	43,736.95
Houston Parker-Ezzell Oil Co.,	7,117.81	3, Tulsa, Okla Skelly, W. G., Tulsa,	73,275.77
	1,941.12 5,749.31	Shelar Gur Okla 3, Tulsa, Okla Skelly, W. G., Tulsa, Okla Okla	17,079.72
Primrose Oil Co., Houston Pilot Point Oil and Gas Co., Pilot Point Powhatan Oil Co., Hous-	438.88	Okla. Southwestern Petroleum	417.55
Powhatan Oil Co., Hous-	1,028.05	Co., Tulsa, Okla Shelby Oil and Gas Co.	2,632.12
Phillip Bros. Oil Co., Guf-	3,893.22	(J. E. Crosbie), Tulsa.	2,532.41
fey Panther Oil Co., Wichita	12,577.79	Okla. Swastika Oil Co., Beau- mont	1,418.04
Falls Pinto Oil Co., Wichita	1.275.79	Stratton Oil Co., Wichita Falls Sheldon & Woodruff, Electra Shallow Oil Co., Wichita Falls	7,327.93
Falls Powell, J. L., Wichita Falls	719.71	Sheldon & Woodruff, Electra	318.20
Plainview-Littlefield Oil Co., Littlefield	8,685.93	Shallow Oil Co., Wichita	186.60
Pivote, M. V., Sour Lake Pilant Lake Oil and Gas	4,191.67	Sextette Oil Co., Lawton,	4,671.05
Co., Houston Quanah Oil Co., Quanah. Red River Oil and Gas	1,045.31 5.325.85	Okla. 6666 Oil Co., Wichita Falls	10.985.70
Red River Oil and Gas	303.96	Falls States Oil Corporation, Eastland Spencer Oil Co., Wichita Falls	1,094.73
Richardson Oil Co.,	1,535.76	Spencer Oil Co., Wichita	7,664.50
Co., Bowie Richardson Oil Co., Brownwood Robertson, N. A., Law- ton, Okla.	2.097.62	South Bosque Petroleum	852.04
Robertson Petroleum Co.		Sixty-Six Oil Co., Wichi-	1,201.90
Lawton, Okla Regna Oil Co., Saratoga. Robertson & Knotts,	39,605.22 12,408.32	South Bosque Petroleum Co., Waco Sixty-Six Oil Co., Wichi- ta Falls Sinks, Joel Co., Corsicana Storge Oil Co. Wichita	254.00
Wichita Falls	562.88		2,433.40
Rowe, M. D., Wichita	608.57	Falls Sheperd-Conrey Oil Co., Wichita Falls	1,222.66
Falls Russell-Sanderson Oil Co.,	8,754.79	Sun Co. (North Texas Division), Dallas Somerset Oil Co., San An-	354,295.06
Russell-Sanderson Oil Co., Wichita Falls Ream Oil Co., Wichita	4,536.55	tonio	2,380.87
Ream Oil Co., Wichita Falls Roberts & Hill, Wichita	341.88	Burkburnett	7,744.89
Falls	739.30	tonio School Block Oil Co., Burkburnett Simms, E. F. & Co., Houston Superior Oil Co., Superior,	260,364.10
Falls Ruyle Farm Oil Co., Wichita Falls	39,010.25	Wis.	854.98

	Production,	Production,
Company and Address.	barrels.	Company and Address. barrels.
Shamrock Oil Co., Wichita	21,284.28	Tex-Penn Oil Co., Pitts- burgh, Pa 149,051.36
Falls Sun Co., Beaumont	127,997.27	Texas-Eastern Oil Co.,
Schucher On Co., Sour	1,434.05	Texas-Eastern Oil Co., Buffalo, N. Y 153.48
Lake Staley Mashburn Oil Co.,		Texas Company Produc- ing Department, Hous-
Wichita Falls Sam Oil Co., Wichita	4,211.88	ton
Falls Snider, C. W., Wichita	2,354.36	Texas Pacific Coal & Oil Co., Thrall1,521,379.67 Tip Top Oil and Mineral
Falls	7,884.89	Tip Top Oil and Mineral
Speed, C. D., Corsicana Stella Oil Co., Beaumont	$156.00 \\ 7.928.98$	Co., San Antonio 80.20 Tatum & Cunningham,
Falls	816.92	Corsicana
Stephens Oil Co., Sour Lake Sykes, C. E., Ardmore, Okla.		Taylor 8.081.88
Okla. Sanders-Taylor Oil Co.,	402.68	Texas Dividend Co., Wichita Falls 203.71
wichita Fails	8,495.28	Texas-Electra Co., Dal-
Sammies Oil Corporation, Ranger	313.00	las
Ranger Surenuff Oil Co., Wichita	4,063.95	fining Co., Wichita
Falls	1,000100	Triangle Oil Co., Wichita
(Damon Mound), Hous- tor.	67,142.79	Falls 3,461.08 T. H. Y. Oil Co., Sour 1,278.95
Swangondala Oil Co Val-	146,002.95	Lake 1,278.95 Tri-Mutual_Oil Co., Rapid
ley Mills Sunshine Surety Oil Co., Wichita Falls Silver Lake Oil Co.,		City, S. D 1,886.94
Wichita Falls Silver Lake Oil Co	475.00	Linited Petroleum (Co
Abilene Schultz-Britain Oil Co.,	495.60	Houston
Seymour	257.52	Wichita Falls 15,292.01 United Oil and Fuel Co.,
Sutherland, W. C., Wichi- ta Falls	893.58	United Oil and Fuel Co., Philadelphia, Pa 3,294.02
Southern Petroleum Co.,	23,393.69	Unity Oil Co., Beaumont 18,208.79
Houston		Denver, Colo 1,445.84
Co., Alice	932.00	Valley Oil Co., Petrolia 157.76 Vat Oil Co., Byers 873.33
Henrietta	471.29	Victor Oil Co., Freder-
Henrietta Steelsmith, C. A., Electra San Bernard Oil Co., Begument	703.88	Vindicator Oil Co.
Beaumont	4,304.49	Wichita Falls
Lake	3,631.37	Worth 57,641.36
snowden, Geo. M., Hum- ble	542.00	Vertate Oil Co., Dallas 1,141.15 Virginia Oil Association,
ble	4,583.96	Houston 3,158.48 Victory Petroleum Co.,
ton Thirty-Nine Oil Co.,		Wichita Falls 5.605.02
Wichita Falls Turner & Sheegog, Wichi-	2,620.00	vernon Oll Co., wichita
ta Falls Thompson Oil Co., Elec-	1,189.50	Valley View Oil Co
tra Thrity-One Oil Co., Law-	3,910.75	Willis Oil Co., Wichita
ton, Okla.	386.54	Wichita Burk Oil Co.
ton, Okla. Trojan Oil Co., Wichita Falls	6,952,62	Wichita Falls 5,423.77
Town Line Oil Co.		Wichita Southern Oil Co., Wichita Falls 5,230.06
Wichita Falls Thirty-Two Oil Associa- tion, Wichita Falls	8,367.00	West Production Co.
tion, Wichita Falls	1,344.15	Woods, G. C., Wichita
Tarver Drilling Co., Dal- las	557.69	Wichita Valley Oil and
las	433,46	Gas Co., Wichita Falls 120.00

	JUNIL, I	Ji June Concluded	
P	roduction.	P	roduction,
Company and Address.		Company and Address.	barrels.
Wilson-Broach Oil Co., Beaumont Weowna Oil Co., Wichita	21,943.43	Watson - Lee Oil Co., Brownwood Worth Oil Co., Tulsa,	1,307.49
Falls Wichita Falls Gas Co.,	28,415.61	Okla	1,475.92
Wichita Falls Walker-Caldwell Produc-	71.56	Freeport	2,028.14
ing Co., Dallas Witherspoon Oil Co., San	6,095.69 3,746.58	Wichita Oil and Gas' Co., Wichita Falls	2,160.00
Antonio Walker-Smith Oil Co., Brownwood	39.14	Wichita Falls Petroleum Co., Wichita Falls	11,093.06
West Texas Oil Co.,		Waggoner, J. J., Hamlin	766.69
Wichita Falls Woodrow-Lee Oil Co., Wichita Falls	5,435.75 26,073.22	Woods Oil Co., Beaumont Webb Oil Co., Humble Whale Oil Co., Durant,	4,014.79, 4,092.52
Wichita-Clay Oil Co., Wichita Falls	780.48	Okla. Yount-Lee Oil Co., Sour	2,512.95
Weiss-Martin Oil Co., Dallas	677.41	Lake Ramming, R. W., Wichita	56,114.67
Williams, J. L., Brown- wood	622.00	Falls Ramming, Staley & Co.,	3,880.00
Welden Oil Co., Saratoga Willis, W. T., Wichita	13,400.00	Wichita Falls South Side Oil Co.,	760.00
Falls	3,251.67	Wichita Falls Staley, Langford & Co.,	2,460.00
McKinney	179.00	Wichita Falls Staley, J. I. & Co.,	98,377.59
Mineral Co., Dallas Wood-Dale Oil Co., Hen-	34,192.49	Wichita Falls Staley, J. A., Wichita	92,111.00
rietta	4,913.32	Falls	8,841.19

STANDARD OIL GROUP

Refiners and Marketers

Reinfela	inu marketera		
Company	Capitalization	Mkt. Price	Mkt. Value
Anglo-American	\$15.000.000	25	\$ 75,000,000
Atlantic Refining		1350	67,500,000
Borne-Scrymser		500	1,000,000
Chesebrough Mfg		310	4,650,000
Continental Can		655	19,650,000
Galena Signal, 2d pfd		107	6,420,000
Galena Signal Oil, 1st pfd		125	2,500,000
Galena Signal, common	16,000,000	138	22.080,000
International Pet		31	38,844,000
Solar Refining		370	7,400,000
S. O. of California		282	280,232,706
S. O. of Indiana		800	240,000,000
S. O. of Kansas		600	12,000,000
S. O. of Kentucky		400	24,000,000
S. O. of Nebraska		550	5,500,000
S. O. of New Jersey	98,338,300	710	698,201,939
S. O. of New York	75,000,000	382	286,500,000
S. O. of Ohio		525	36,750,000
Swan & Finch		100	1,450,000
Vacuum Oil		440	66,000,000
Provide the second s	- 0		
Producin	g Companies		
Ohio Oil Company	15,000,000	386	231,000,000
Prairie Oil & Gas Company		750	135,000,000
South West Penn		313	62,600,000
Washington Oil		40	400,000
Carter Oil Co			

STANDARD OIL GROUP—Concluded Pipe Lines and Carriers

Buckeye Pipe Line 10,000,000 Crescent Pipe Line 3,000,000 Cumberland Pipe Line 1,488,851 Eureka Pipe Line 5,000,000 Illinois Pipe Line 20,000,000 Indiana Pipe Line 5,000,000 National Transit 6,862,500 New York Transit Company 5,000,000 Prairie Pipe Line 4,000,000 Prairie Pipe Line 10,000,000 Southern Pipe Line 10,000,000 South West Penn 3,500,000 Union Tank Line 12,000,000	$\begin{array}{ccccccc} 100 & 20,000,000\\ 36 & 2,160,000\\ 200 & 2,977,600\\ 167 & 8,320,000\\ 184 & 36,800,000\\ 105 & 10,500,000\\ 22 & 11,198,000\\ 185 & 9,250,000\\ 112 & 4,480,000\\ 300 & 81,000,000\\ 165 & 10,500,000\\ 165 & 15,600,000\\ 130 & 15,600,000 \end{array}$
Total market values, all companies Market value refining and marketing companies Market value producing companies Market value pipe line and carrying companies	
PRINCIPAL AMERICAN AFFILIATIONS OF SHELL PETROLEUM COMBINE	
Shell Transport & Trading Co., Ltd.—London Royal Dutch Roxana Petroleum Co.—N. J. Roxana Petroleum Co.—Okla. Puora Oil Co.—Okla. Turner Oil Co.—Cal. New Orleans Refining Co. Shell Co. of California. Simplex Refining Co.—Cal. Valley Pipeline Co.—Cal. General Asphalt Co. Caribbean Petrol. Syndicate—Venezuela Petrol. Development Co.—Trinidad	60,750,000 60,000,000 8,000,000 500,000 400,000 45,000,000 3,000,000 10,000,000
Trinidad Lake Petroleum Co.—Trinidad Bermudez Co., Ltd.—Venezuela Anglo Saxon Petroleum Co. Ltd.—London Mexican Eagle Oil Co., Ltd.—Mexico Eagle Oil & Transp. Co. Anglo Mexican Petrol. Co., Ltd. Oil Fields of Mexico, Ltd. Alliance Co. of Mexico. La Corona Petroleum Co.—Mexico (Tampico-Pan Tampico-Panuco Petrol. Co.—(Holland), Mexico Val. Ry. Co.) British-American Oil Co.—Canada (Chjoles Oil Co Shell Co. of Canada United British and West Indies Petrol. Synd., L	90,000,000 uco Oil Flds., Ltd.) o (Tampico-Panuco Co.)

1....

Casinghead Gasoline Plants (1917)

CALIFORNIA

Capacity, Gallons

Fellows Gasoline Co......Fellows, Calif.

ILLINOIS

Vacuum Gasoline Co.....Bridgeport, Ill. Central Refining Co....Lawrenceville, Ill. Warner-Caldwell Oil Co....Robinson, Ill. Roxana Petroleum Co. of Oklahoma.....Wood River, Ill.

KANSAS

Paul F. Dahlgren	Elgin, Kan.
Rhode Island Oil Co	Independence Kan.
S. C. Redd	
Hygrade Petroleum & Gasoline Co	
Sedan Gasoline Co	Sedan, Kan.

LOUISIANA

De Soto Gasoline CoGoss, La.	
Bayou Gasoline CoOil City, La.	
Standard Oil Co Trees City, La.	
Central Oil & Gasoline Co., IncVivian, La.	

OHIO

Kinkade Oil & Gas CoBremen, Ohlo
Marietta Oil CoMarietta, Ohio
Jefferson County Oil CoRayland, Ohio
Jefferson Gasoline CoRayland, Ohio
Summerfield Gas CoSummerfield, Ohio
Dinsmore & Co Washington, Ohio
John Mildren Sons & CoWinton, Ohio

OKLAHOMA

Mid-Co. Gasoline CoAdair Okla.	
T. B. Gasoline Co Alluwe, Okla.	4,000
Hygrade Petroleum & Gasoline CoAvant, Okla.	1,200
Brighton Gasoline Co Bald Hill, Okla.	1,000
Crystal Gasoline CoBald Hill, Okla.	1,500
Mileage Gasoline CoBald Hill, Okla.	1,000
Producers Oil CoBald Hill, Okla.	
Sinclair Oil & Gasoline CoBald Hill, Okla.	
Twin Hill Gasoline CoBald Hill, Okla.	600
Akin Gasoline Co	000
Mid-Co. Petroleum & Gasoline CoBartlesville, Okla.	
Moon Gasoline Co. Davide Contraction of the Contrac	
Moon Gasoline Co Bartlesville, Okla.	9 500
Frank Phillips	2,500
Wolverine Oil CoBartlesville, Okla.	5,000
Corlis Oil CoBartlesville, Okla.	-
Mileage Gasoline CoBartlett, Okla.	
Smith & Swan Gasoline CoBartlett, Okla.	600
Chestnut & SmithBeggs, Okla.	
H. F. WilcoxBeggs, Okla.	
Paul F. DahlgrenBig Heart, Okla.	
Whitehall, Donavan, Hayden & WhitehallBird Creek, Okla.	
Aiken Gasoline CoBixby, Okla.	
Livingston Oil CorporationBixby, Okla.	
Okla. Petroleum & Gasoline CoBixby, Okla.	
The Three Gasoline CoBixby, Okla.	
S. C. ReddBixby, Okla.	
H. F. WilcoxBixby, Okla.	
Boynton Gasoline CoBoynton, Okla.	4,000
Carter Oil CoBoynton, Okla.	3,000

CASINGHEAD GASOLINE PLANTS—Continued

CASINGHEAD GASOLINE PLANIS—Continued	Capacity,
OKLAHOMA—Continued	Gallons
OKLAHOMA—Continued Hays Gasoline Co. Boynton, Okla. Sterling Gasoline Co. Boynton, Okla. Arrow Gasoline Co. Broken Arrow, Okla. Arrow Gasoline Co. Broken Arrow, Okla. Misener Gasoline Co. Broken Arrow, Okla. Misener Gasoline Co. Broken Arrow, Okla. Okla. Petroleum & Gasoline Co. Broken Arrow, Okla. Okla. Petroleum & Gasoline Co. Broken Arrow, Okla. Altena Oil Co. Chelsea, Okla. Liquefield Petroleum & Gasoline Co. Chelsea, Okla. Liquefield Petroleum & Gasoline Co. Chelsea, Okla. Una Gasoline Co. Chelsea, Okla. Whitehall, Donavan, Hayden & Whitehall. Childers, Okla. Whitehall, Donavan, Hayden & Whitehall. Childers, Okla. Okla. Petroleum & Gasoline Co. Cleveland, Okla. Okla. Petroleum & Gasoline Co. Cleveland, Okla. Mitehall, Donavan, Hayden & Whitehall. Childers, Okla. Okla. Petroleum & Gasoline Co. Cleveland, Okla.	1,100
Sterling Gasoline CoBoynton, Okla.	
Arrow Gasoline Co Broken Arrow, Okla.	500
Mischar Cagoline Co. Broken Arrow Okla	500
Okla Petroleum & Gasoline Co	000
Piedmont Petroleum & Gasoline CoBroken Arrow, Okla.	1,100
Altena Oil CoChelsea, Okla.	2,500
Cinco Oil CoChelsea, Okla.	500
Liquefield Petroleum Co Chelsea, Okla.	5,000
Una Gasoline Co	1,200
Henderson Gasoline Co	16,000
Whitehall, Donavan, Hayden & Whitehall, Childers, Okla.	,
Gypsy Oil CoCleveland, Okla.	
National Products Co Cleveland, Okla.	
Okla. Petroleum & Gasoline Co Cleveland, Okla.	
B T Curley Coolton Okla	200
Tidal Gasoline Co	200
Chestnut & SmithCushing, Okla.	
Hillman Refining CoCushing, Okla.	500
Magnolia Petroleum CoCushing, Okla.	
S. C. ReddCushing, Okla.	
Standard Oil Co. of Indiana	600
Roxana Petroleum Co. of Okla.	
Diamond Gasoline CoDelaware, Okla,	8,000
Aikin Gasoline Co Dewey, Okla.	2,000
Paul F. Danigren	
Mid-Co Gasoline Co	600
Barmont Oil Co	250
Chestnut & SmithDrumright Okla	200
Consumers Refining Co Drumright, Okla.	
Gypsy Oil CoDrumright, Okla.	
Imperial Gasoline CoDrumright, Okla.	
McMan Gasoline CoDrumright, Okla.	2,000 600
Mid-Co. Petroleum & Gasoline Co Drumright, Okla	000
Ohio Cities Gasoline CoDrumright, Okla.	3,000
Producers Oil CoDrumright, Okla.	
Sinclair Oll & Gasoline CoDrumright, Okla.	
Tidal Gasoline Co.	
Okla. Petroleum & Gasoline Co	
Producers Oil CoGlenn Pool, Okla.	
Sun Gasoline CoGlenn Pool, Okla.	
Viotor Gasoline CoGlenn Pool, Okla.	600
Watking Oil Co	
National Products Co. Cleveland, Okla. Sinclair Oil & Gasoline Co. Cleveland, Okla. Sinclair Oil & Gasoline Co. Cleveland, Okla. B. T. Curley. Coalton, Okla. Tidal Gasoline Co. Coalton, Okla. Chestnut & Smith. Cushing, Okla. Magnolia Petroleum Co. Cushing, Okla. S. C. Redd. Cushing, Okla. S. C. Redd. Cushing, Okla. Standard Oil Co. of Indiana. Cushing, Okla. Bamond Gasoline Co. Delaware, Okla. Paul F. Dahlgren. Dewey, Okla. Paul F. Dahlgren. Dewey, Okla. Dewey Portland Cement Co. Dewey, Okla. Mid-Co. Gasoline Co. Drumright, Okla. Consumers Refining Co. Drumright, Okla. Gypsy Oil Co. Drumright, Okla. McMan Gasoline Co. Drumright, Okla. Mid-Co. Petroleum & Gasoline Co. Drumright, Okla. McMan Gasoline Co. Drumright, Okla. Mid-Co. Petroleum & Gasoline Co. Drumright, Okla. McMan Gasoline Co. Drumright, Okla. McMan Gasoline Co. Drumright, Okla. McMan Gasoline Co. Dru	
Magnolia Petroleum Co	3,000
Superior Oil & Gas Co Healdton, Okla.	
Mileage Gasoline CoHaskell, Okla.	
Gupsy Oil Co	
Oil State Gasoline Co	2,500
Okla. Petroleum & Gasoline Co	2,000
Totem Gasoline CoJenks, Okla.	
Atlas Petroleum CoJennings, Okla.	
Chestnut & Smith	9,000
D. W. Franchot & Co	1,000
Glenn Gas Co	1,100
Gypsy Oil CoKiefer, Okla.	-,
Victor Gasoline Co	
Lawton Befining Co	
Watkins Oil Co	1,000
our cas compressing commence and compressing commence and compressing commence and compressing commence and commence and commence and commence and	1,000

CASINGHEAD GASOLINE PLANTS—Continued

	Gallons
OKLAHOMA—Concluded Mileage Gasoline CoLost City, Okla.	Capacity,
Marland Befining Co. Mervin Field Okla	3,000
Marland Refining Co	0,000
National Products CoMounds, Okla,	
Nine Oil & Gas CoMaud, Okla,	
Chestnut & SmithMorris, Okla.	
Braðstreet & Co	250
De Soto Gasoline Co Muskogee, Okla.	250
Motor Gasoline Co	1,100
Persian Oil Co. Muskopee Okla	250
Persian Oil CoMuskogee, Okla. Red Demon Gasoline CoMuskogee, Okla.	800
Sun Gasoline Co	000
Victor Gasoline Co Muskogee, Okia	
Whitfield Sears Oil CoMuskogee, Okla. Childers Gasoline CoNowata, Okla. Tidal Gasoline CoNowata, Okla.	250
Childers Gasoline CoNowata, Okla.	500
Tidal Gasoline Co Nowata, Okla.	
Usage Gasoline Co	2,750
Tidal Gasoline CoOchelata, Okla.	0.000
A. C. F. Gasoline CoOilton, Okla. Chieftain Gasoline CoOilton, Okla.	2,000
B B Jones Oilton Okla	500
Mid-Co Gasoline Co Oilton, Okla	000
B. B. Jones. Oilton, Okla. Mid-Co. Gasoline Co. Oilton, Okla. Mid-Co. Petroleum & Gasoline Co. Oilton, Okla. National Products Co. Oilton, Okla. Southland Gas Co. Oilton, Okla. Standard Oil Co. Oilton, Okla.	
National Products Co	
Southland Gas CoOilton, Okla.	600
Standard Oli Co. of Indiana	
Kingwood Oil CoOkmulgee, Okla.	
Magnolia Petroleum CoOkmulgee, Okia.	
O. K. Refining CoOkmulgee, Okla.	
Pine Pool Gasoline Co	600
Anagoolia Petroleum Co	1 000
Mac Batty Casoline Co	1,000
Mac Betty Gasoline Co	
Victor Gasoline Co. Peru. Okla.	
Victor Gasoline Co	
Marland Chemical CoPonca City, Okla.	
Marland Gasoline CoPonca City, Okla.	
Whitehall, Donavan, Hayden & Whitehall. Pumpkin Center, Okla.	
Mileage Gasoline Co	
Arthur Oil CoSapulpa, Okla. Bluff Gasoline CoSapulpa, Okla.	500
Biult Gasoline Co	200
Max Rhas Gasoline Co	1,000 600
Richards Gasoline Co	600
Commerce Gasoline Co	000
W. G. SkellySapulpa, Okla	
Cosden Oil & Gas Co Shamrock, Okla.	8,000
Magnolia Gasoline Co	
Sinclair Oil & Gasoline CoShamrock, Okla.	
Union Skiatook Gasoline CoSkiatook, Okla.	
Rotary Gasoline Co	
Black Hawk Petroleum Co Stone Bluff, Okla.	1 000
Union Skiatook Gasoline Co	1,200
Okla Petroleum & Gasoline Co. Standard Spur Okla	
O. G. Bantley,	-11.00
The Dallas Co Tulsa, Okla.	
Pulaski Refining Co Turkey Mountain, Okla	. 700
Silver Gasoline CoVega, Okla.	
De Soto Gasoline Co Wann Okla.	3,000
Mid-Co. Gasoline Co	
Chostnut & Fitzgemild	200
Eagle Gasoline Co. Watkins, Okla.	600 1,100
Okla. Petroleum & Gasoline Co	1,100
· · · · · · · · · · · · · · · · · · ·	1,100

CASINGHEAD GASOLINE PLANTS-Concluded

PENNSYLVANIA

Bradford Oil & Gasoline CoBell's Camp, Pa.	
Pennsylvania Gasoline CoBradford Pa.	
B. B. Stroud CoBradford Pa.	
W. H. MillerChicora, Pa.	
Clarendon Gasoline CoClarendon, Pa.	
Clarendon Refining CoClarendon, Pa.	
D. and C. P. McKeeClintonville, Pa.	
Jane Oil Co Emlenton, Pa.	
Gilmore Gasoline CoGilmore, Pa.	
Kane Gasoline CoKane, Pa.	
C. J. Ritzert Co	
Henry Farm Oil CoWarren, Pa.	
Gilmore Gasoline Co Wafferty Hollow, H	Pa.
Wayne Naptha Co Waynesburg, Pa.	

TEXAS

Humble Oil	& Refining Co	Burkburnett, Tex.
Schulz Gase	oline Co	Burkburnett, Tex.
Forest Oil	Co	Electra, Tex.
Forest Oil	Co	Iowa Park, Tex.

WEST VIRGINIA

Imperial Oil & Gas	Products Co	.Hannahdale W. Va.
Jas. B. Berry's Sons	Co	.Sisterville, W. Va.
Consumers Refining	Co	. Waverly, W. Va.
Laughner & Flemin	g	.Wellsburgh, W. Va.

American Gas Syndicates and Their Holdings (Gas Record 1919)

Company	City	State
SOUTHERN CALIFORNIA GAS CO Operating at Los Angeles, Glendale, Sa Bernardino, Gardena, Riverside, Colto Arlington, Rialto, Beverley Hills, Va Nuys, Tropico, Lankersheim, San Fe nando, Eagle Rock and Burbank.		California
SOUTHERN COUNTIES GAS CO. of Cal A consolidation of Southern Counti Gas Co., Long Beach (Calif.) Gas C. and gas properties of the Southe Calif. Edison Co. Serves natural g to 42 cities of Los Angeles, Orange an San Bernardino counties.	ifLos Angeles es 724 S. Spring St. o., rn as ad	California
W. F. BOARDMAN CO Operates Oregon Gas & Electric Co. Grant's Pass, Medford, Ashland ar Roseburg; Ukiah Gas Co., Ukiah, Calif Guadalajara Gas Co., Guadalajar Jalisco, Mexico.	San Francisco at 718 Mission St. d .; a,	California
CALIFORNIA LIGHT & FUEL CO Engineers: Palo Alto (Calif.) Gas Co Nevada Gas Co., Tonopah, Nev.	San Francisco 0.; 626 Pacific Bldg.	California
OAST COUNTIES GAS & ELECTRIC C Operates at Santa Cruz, Watsonvill Hollister and the Gilroy (Calif.) G Works, Contra Costa Gas Co., at Mart Mart, Pittsburgh, Antioch, Concor Crockett, Calif.	le, 454 California St. as i-	California
AST VALLEYS GAS & ELECTRIC CO Operates at Monterey, Pacific Grov Carmel-by-the-Sea, Salinas, King Cit Soledad, Gonzales, Chular.	e. 58 Sutter St.	California
NORTHERN CALIFORNIA POWER CO Operates Northern Calif. Power Co Keswick Electric Power Co., Batt Creek Power Co., Redding Water C and Sacramento Valley Power Co. Own and operates gas plants at Reddin Red Bluff and Willows.	o., 995 Market St. le o. os	California
PACIFIC GAS & ELECTRIC CO Supplies gas to over 50 Californ towns and cities.	San Francisco ia 445 Sutter St.	California
NORTHERN COLORADO POWER CO. Operates: Cheyenne (Wyo.) Ligh Fuel & Power Co., Boulder (Colo Elec. Lt. & Pr. Co., Western Lt. & P Co., Lafayette, Colo.	Boulder t, .) r.	Colorado
SOUTHERN UTILITIES CO	Jacksonville	Florida
COPLEY GAS & ELEC. SYNDICATE Owns: Western United Gas & Ele Co., operating gas plants at Auror Elgin and Joliet; Murpheysboro (III Water Wks. & Elec. & Gas Light Co Southern III Gas Co Marion III	Aurora c. a, .)	Illinois

AMERICAN GAS SYNDICATES AND THEIR HOLDINGS -Continued City State Company TRACTION SYSTEM (Mc-ILLINOIS Illinois Operate by product plants for Dover Operate by Charle Canal Dover, O.Chicago Illinois 608 S. Dearborn St. By-Products Coke Co., Canal Dover, O., Union By-Products Coke Co., Buffalo, N. Y., Geo. W. Niedringhaus and as-sociates, Granite City, Ill. K. L. AMES SYNDICATE..... Owns and operates Jacksonville (Fla.) Woma Gas Co. Illinois Woman's Temple Bldg. Gas Co. GAS & ELECTRIC IMPROVEMENT CO....Chicago Operates Austin (Tex.) Gas Lt. Co., 39 S. Benton Harbor & St. Joseph (Mich.) Gas & Fuel Co.; Ft. Madison (Ia.) Gas Light Co.; Palestine (Tex.) Lt., Ht. & Illinois 39 S. LaSalle St. Chicago Illinois Cont. & Com'l Bk. Bldg. WESTERN STATES GAS & ELEC. CO, Stockton, Richmond and Eureka, Calif.; ARKANSAS VALLEY RY. LT. & PR. CO., Pueblo, Victor, Cripple Creek, Rocky Ford, La Junta and Canon City; LOUISVILLE GAS & ELECTRIC CO. METROPOLITAN GAS & ELECTRIC CO...Chicago Owns and operates: Southwestern Gas & Elec Co. Shravenort La and Texa Illinois Harris Trust Bldg. & Elec. Co., Shreveport, La., and Tex-arkana, Tex.; Mobile (Ala.) Gas Co.; Central Indiana Gas Co. of Muncie (hdqrs.), Anderson, Marion; Alex-(hdqrs.), Anderson, Marion; Alex-andria, Elwood, Fairmount and Hart-ford City, Ind.; Jackson Co. Lt., Ht. & Pr. Co. of Independence, Mo.; Beau-mont (Tex.) Gas Lt. Co.; Seattle

AMERICAN	GAS	SYNDICATES	AND	THEIR	HOLDINGS
		-Continu	ed		

Company City State (Wash) Lighting Co.; Mt. Clemens Mindan State Mindan	Continued			
L. E. MYERS CO	- Company		State	
L. E. MYERS CO	(Wash.) Lighting Co.; Mt. Clemens (Mich.) Gas Light Co.; Gainesville (Tex.) Gas & Elec. Co.			
BIDDLE WEST UTILITIES COChicago Illinois Controls and operates following gas 72 W. Adams St. properties: —Illinois Northern Utilities CoDixon Illinois Belvidere, DeKalb, Dixon, Geneseo, Mendota, Morrison, Rock Falls, Sterling Illinois and Sycamore, Ill. —Central Illinois Public Service CoMattoon Illinois Beardstown, Charleston, Macomb, Mat- toon, Pana, Paris and Taylorville, Ill. —Indiana polis Indiana Bedford, Greenfield, New Castle, Sey- mour and Shelbyville, Ind.; Franklin Indiana Indiana —Central Indiana Lighting Co	L. E. MYERS CO Owns Ashland (Wis.) Lt., Pr. & St. By Co.	.Chicago Monadnock Blk.	Illinois	
BIDDLE WEST UTILITIES COChicago Illinois Controls and operates following gas 72 W. Adams St. properties: —Illinois Northern Utilities CoDixon Illinois Belvidere, DeKalb, Dixon, Geneseo, Mendota, Morrison, Rock Falls, Sterling Illinois and Sycamore, Ill. —Central Illinois Public Service CoMattoon Illinois Beardstown, Charleston, Macomb, Mat- toon, Pana, Paris and Taylorville, Ill. —Indiana polis Indiana Bedford, Greenfield, New Castle, Sey- mour and Shelbyville, Ind.; Franklin Indiana Indiana —Central Indiana Lighting Co	UNION UTILITIES CO Control The Indiana Gas Light Co., operating plants at Noblesville and Tipton, Ind. Also, Lewanee County Gas & Electric Co., Vicksburg, Miss. & St. Charles Lighting Co., St. Charles, Mo.; Dubuque (Ia.) Elec. Co.; Northern Ia. Gas & Elec. Co., Hdqrs, Humboldt, Ia., serving 20 towns in Northern Iowa; rese at Facele Groupe	. Chicago 39 S. LaSalle St.	Illinois	
Illinois Northern Utilities CoDixon Illinois Belvidere, DeKalb, Dixon, Geneseo, Mendota, Morrison, Rock Falls, Sterling and Sycamore, Ill. Central Illinois Public Service CoMattoon Illinois Beardstown, Charleston, Macomb, Mat- Illinois Beardstown, Charleston, Macomb, Mat- toon, Pana, Paris and Taylorville, Ill.	MIDDLE WEST UTILITIES CO Controls and operates following gas		Illinois	
—Central Illinois Public Service CoMattoon Illinois Beardstown, Charleston, Macomb, Mat- toon, Pana, Paris and Taylorville, Ill. —Hoopeston (Ill.) Gas & Elec. Co. —Hoopeston (Ill.) Gas & Elec. Co.	—Illinois Northern Utilities Co Belvidere, DeKalb, Dixon, Geneseo, Mendota, Morrison, Rock Falls, Sterling and Sycamore, Ill.		Illinois	
Indury and Shebyline, Hair, Franklin, (Ind.) Water, Light & Power Co.; Central Indiana Lighting Co., Bloomington, Ind. —Central Indiana Lighting Co., Bloomington, Ind. —Central Indiana Lighting Co., Bloomington, Ind. —Central Indiana Lighting Co., Status, —Twinkin (Ind.) Water, Lt. & Pr. Co. —United Gas & Electric CoNew Albany Indiana Jeffersonville and New Albany, Ind. —Twin State Gas & Electric CoDover New Hampshire Bennington, Ya.; Brattleboro, Va. —Michigan Gas & Electric Co. Ishpeming, Negaunee, Houghton and Hancock, Mich. —Michigan Gas & Electric Co. —Kington, Marshall, Mo. —Kenucky Utilities Co. Shelbyville, Ky. —Chickasha Gas & Electric Co	-Central Illinois Public Service Co Beardstown, Charleston, Macomb, Mat- toon, Pana, Paris and Taylorville, Ill.	. Mattoon	Illinois	
—Central Indiana Lighting CoIndianapolis Indiana —Franklin (Ind.) Water, Lt. & Pr. Co. …United Gas & Electric CoNew Albany Indiana Jeffersonville and New Albany, Ind.	(Ind.) Water, Light & Power Co.; Cen- tral Indiana Lighting Co., Bloomington,			
Missouri Gas & Electric Service Co. Lexington, Marshall, Mo. Kentucky Utilities Co. Shelbyville, Ky. Chickasha Gas & Electric CoChickasha Oklahoma Michigan Gas & Electric CoThree Rivers Michigan Nebraska City (Neb.) Utilities Co. Citizens Gas Light CoJackson Tennessee PUBLIC SERVICE CO. OF NORTHERN ILLINOISChicago Illinois Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.	-Central Indiana Lighting Co	.Indianapolis	Indiana	
Missouri Gas & Electric Service Co. Lexington, Marshall, Mo. Kentucky Utilities Co. Shelbyville, Ky. Chickasha Gas & Electric CoChickasha Oklahoma Michigan Gas & Electric CoThree Rivers Michigan Nebraska City (Neb.) Utilities Co. Citizens Gas Light CoJackson Tennessee PUBLIC SERVICE CO. OF NORTHERN ILLINOISChicago Illinois Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.	Franklin (Ind.) Water, Lt. & Pr. Co. United Gas & Electric Co	New Albany	Indiana	
Missouri Gas & Electric Service Co. Lexington, Marshall, Mo. Kentucky Utilities Co. Shelbyville, Ky. Chickasha Gas & Electric CoChickasha Oklahoma Michigan Gas & Electric CoThree Rivers Michigan Nebraska City (Neb.) Utilities Co. Citizens Gas Light CoJackson Tennessee PUBLIC SERVICE CO. OF NORTHERN ILLINOISChicago Illinois Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.	Jeffersonville and New Albany, Ind. —Twin State Gas & Electric Co Bennington, Va.; Brattleboro, Va. —Michigan Gas & Electric Co. Ishpeming, Negaunee, Houghton and	Dover New	Hampshire	
PUBLIC SERVICE CO. OF NORTHERN ILLINOISChicago Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.	Migroupi Cog & Electric Comice Co			
PUBLIC SERVICE CO. OF NORTHERN ILLINOISChicago Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.	-Chickasha Gas & Electric Co -Michigan Gas & Electric Co -Nebraska City (Neb.) Utilities Co.	Chickasha Three Rivers	Michigan	
Gas plants: Evanston, Blue Island, 72 W. Adams St. Weber, Morris, Ottawa, Strator, Pon- tiac and Kankakee.		Jackson	Tennessee	
NORTH AMERICAN LIGHT & POWER CO	Gas plants: Evanston, Blue Island, Weber, Morris, Ottawa, Strator, Pon-	Chicago 72 W. Adams St.	Illinois	
 Hold and operate: Adair County Lt., 2013 Peoples Gas Bldg. Pr. & Ice Co. and Mo. Ht., Lt. & Pr. Co., of Kirksville, Mo.; Moberly (Mo.) Lt. & Pr. Co.; Huntsville (Mo.) Lt. & Pr. Co.; Boonville (Mo.) Lt., Ht. & Pr Co.; Ardmore (Okla.) City Gas Co.; Durant (Okla.) Consumers Lt. & Pr. Co.; Washington C. H. (O.) G. & E. Co.; Pocatello (Ida.) Gas & Pr. Co.; Waurika (Okla.) Consumers Lt. & Pr 	NORTH AMERICAN LIGHT & POWER		T 111	
	CO	Chicago 2013 Peoples Gas Blo		

AMERICAN GAS SYNDICATES AND THEIR HOLD	INGS
Company City	State
Co.; Southern Okla, Pipe Line Co., Ard- more, Okla; Citizens Elec. Co., Higbee, Mo.; La Plata (Mo.) Lt. & Pr. Co.	· · ·
UNITED LT. & RYS. CO. (FINANCIAL OFFICE)Chicago UTILITIES DEVELOPMENT COChicago	Illinois Illinois
327 S. LaSalle St. WISCONSIN PR. & LT. & HT. COChicago Owns' Baraboo, Beaver Dam, Berlin. 72 W. Adams St.	Illinois
E. A. POTTERChicago Controls Madison, Ind., Creston, Ia., Rector Bldg. Junction City, Abilene, Great Bend, and Manhattan, Kansas.	Illinois
J. J. FREY SYNDICATEHillsboro Owns Southern Illinois Lt. & Pr. Co. and Citizens Gas, Elec. & Htg. Co. Gas plants at Mt. Vernon, Litchfield, and Hillsboro.	Illinois
NORTHERN INDIANA GAS & ELEC- TRIC CO	Indiana
Bluffton, Degatisport, Frankfort, Craw- fordsville, and Wabash; Obio Division operates Ling. St. Marris	
ville, and Coldwater.	
INTERSTATE PUBLIC SERVICE COIndianapolis (Listed above.)	Indiana
W. A. MARTIN GAS SYNDICATELaporte Operates Greencastle (Ind.) Gas & Elec. Co., Rochester Gas & Fuel Co.	Indiana
CONSOLIDATED GAS & OIL CORidgeville Owns plants at Ridgeville, Red Key and Dunkirk, Ind.	Indiana
IOWA ELECTRIC COCedar Rapids Operates gas plants at Fairfield, Iowa Falls and Perry; also various electric and railway plants. Under name of Iowa Ry. & Lt. Co. also operates Mar-	Iowa
shantown la.	. 14
R. K. RUNNERCharles City Interested in Austin, Minn.; Charles City, Ia., and Cherokee, Ia.	Iowa
IOWA GAS & ELEC. COIowa City Owns gas plants at Mt. Pleasant and Washington, Ia.	Iowa
AMERICAN GAS CONSTRUCTION CONewton Interested in Ia. Pub. Serv. Co., Ames, Ia.; Citizens Gas Co., Carroll, Ia.; Belle Plaine (Ia.) Gas Co.	Iowa
AMERICAN CITIES CO	Louisiana
GENERAL UTILITIES & OPERATING	Monulau
CO	Maryland

Company	City	State
COMMONWEALTH GAS & ELECTRIC COS Owns: Marlboro-Hudson Gas Co., Marlboro, Mass., and Athol (Mass.) Gas & Elec. Co.	.Boston 78 Devonshire	Massachusetts
MASSACHUSETTS GAS COS Controls Boston Con. Gas Co., E. Bos- ton Gas Co., Citizens Gas Lt. Co., of Quincy, Mass.; Newton & Watertown Gas Lt. Co., of Newton, Mass.; New England Coal & Coke Co., of Boston; New Eng. Fuel & Trans. Co.	Boston 111 Devonshire	Massachusetts
MASSACHUSETTS LIGHTING COS Operating companies— Adams Gas Light Co., Arlington Gas Light Co., Clinton Gas Light Co., Glou- cester Light Co., Leominster Gas Light Co., Lexington Gas Co., Milford Gas Light Co., Northampton Gas Light Co., North Adams Gas Light Co., Spencer Gas Co., Williamstown Gas Co., Wor- cester County Gas Co. Gas & Electric Improvement Co., Bos- ton; The Light, Heat & Power Corpora- tion, Boston; Daytona (Fla.) Public Service Co. and New Smyrna (Fla.)	. Boston 77 Franklin St.	Massachusetts
STONE & WEBSTER	. Boston 147 Milk St.	Massachusetts
Represents: Suburban Gas & Elec. Co., Revere, Mass.; Peoples Gas & Elec. Co., Oswego, N. Y.; Springfield (Mass.) Gas Lt. Co., Nyack, N. Y.; Malden & Mel- rose (Mass.) Gas Lt. Co.; Fitchburg (Mass.) Gas & Elec. Lt. Co.; No. Bos- ton Lighting Properties; Bristol & Plainville Tramway Co., Bristol, Conn., Montpelier & Barre Lt. & Pr. Co., Montpelier Vt	201 Devonshire	Massachusetts
W. E. MOSS & CO Operates: Coldwater (Mich.) Gas. Lt. & Fuel Co.; Columbus (Ind.) Gas Lt. Co.; Fulton (N. Y.) Fuel & Lt. Co.; Grand Haven (Mich.) Gas Co.; Citizens Gas Co. of Hannibal, Mo.; Winston- Salem (N. C.) Gas Co.; Monroe (Mich.) Gas, Light & Fuel Co.; Hillsdale (Mich.) Gas Light Co.	.Detroit 710 Union Trust	
AMERICAN PUBLIC UTILITIES CO (Kelsey-Brewer interests.) Holland (Mich.) City Gas Co., Albion (Mich.) Gas Light Co., Valparaiso (Ind.) Lighting Co., Elkhart (Ind.) Gas & Fuel Co., Jackson (Miss.) Light & Traction Co., Utah Gas & Coke Co.,	.Grand Rapids G. Rapids Savgs	Michigan . Bldg.

Company

Salt Lake; Wisconsin-Minnesota Light & Power Co., serving Eau Claire, La-Crosse, Chippewa Falls, and Menomi-nee, Wis, Red Wing and Winona, Minn.; Eastern Wis. Elec. Co., Fond du Lac, and upwards of 20 smaller com-munities in immediate vicinity. All operated by Kelsey-Brewer & Co., and all except Fond du Lac (which belongs to Kelsey-Brewer Co.) owned by Amer to Kelsey-Brewer Co.) owned by Amer. Pub. Util. Co.

HOWE, SNOW, CORRIGAN & BERTLES..Grand Rapids Control Emporia (Kans.) Gas Co., Mc-Alester (Okla.) Gas & Coke Co.; Choc-taw Natural Gas Co., Okla.

UNITED LIGHT AND RAILWAYS CO.... Grand Rapids (Financial offices: 72 W. Adams, Chi. Michigan Tr Operating hdqrs.: Grand Rapids and Michigan Michigan Trust Bldg. Operating hdqrs.: Grand Rapids and Davenport, Iowa.) Operates: Chattanooga (Tenn.) Gas Co.; Cedar Rapids (Ia.) Gas Co.; Mus-catine (Ia.) Ltg. Co.; Ft. Dodge (Ia.) Gas & Elec. Co.; Iowa City (Ia.) Lt. & Pr. Co.; Peoples Gas & Elec. Co., Mason City, Ia.; La Porte (Ind.) Gas & E. Co.; Cadillac (Mich.) Gas Lt. Co., Ottumwa (Ia.) Gas Co.; also The Peo-ples Power Co. of Moline and Rock Island, Ill., The Peoples Lt. Co. of Davenport, Ia. and the Davenport (Ia.)

Island, Ill., The Peoples Lt. Co. of Davenport, Ia., and the Davenport (Ia.) Gas & Elec. Co.

- MICHIGAN LIGHT CO..... Owns gas and electric plants at Jack-son, Flint, Bay City, Kalamazoo, Sagi-naw, Pontiac and Manistee.Jackson
- APPLEBY & WAGNER..... Own: Consumers' Gas Co., Waycross, Ga., Gratiot County Gas Co., Alma, Mich., Washington County Gas Co.,Saginaw Michigan Forester Temple Johnson City, Tenn.
- Minnesota 348 Security Bldg. Manitowoc, Wis.
- Missouri 1116 Commerce Bldg. & Lt. Co.
- THE LIGHT & DEVELOPMENT CO. OF

Missouri

Michigan

Michigan

Michigan

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—Continued	
Company City	State
Utilities Co.; Paris (Ky.) Gas and Elec. Co.; Ft. Scott (Kas.) Gas & Elec. Co.; Mitchell (S. D.) Power Co.' Oberlin (O.) Gas & Elec. Co.; Monmouth (Ill.) Pub. Serv. Co.	
THE WATTS ENGINEERING COSt. Louis Owns Columbia (Mo.) Gas Co.	Missouri
GAS CONSTRUCTION COOmaha Operate: Broken Bow (Neb.) Gas Co. 48th & Leavenw	Nebraska orth Sts.
UNION POWER & LIGHT COOmaha Operates North Platte (Nebr.) Lt. & 424 First Natl. Pr. Co. and Southern Ia. Elec. Co., Osceola, Ia.	Nebraska Bk. Bldg.
SIERRA-PACIFIC ELECTRIC COReno (Stone & Webster management) Controls: Carson City (Nev.) Coal Gas Co. and Reno Pr., Lt. & Water Co.	Nevada
CUMBERLAND COUNTY GAS COMillville Operates: Millville Gas Lt. Co.; Citi- zens Gas Co. of Landis Tp., N. J.; Pittsgrove (N. J.) Gas Co.; Fairfield (N. J.) Gas Co.; Citizens Gas Co., Vine- land, N. J.; Maurice River (N. J.) Gas Co.; The Commercial Gas Co., Port Norris, N. J.; Downe Township Gas Co., Newport, N. J., and Lawrence (Tp.) Gas Co., Cedarville, N. J., and Deer- field Gas Co., Rosenhayn, N. J.	New Jersey
PUBLIC SERVICE GAS CONewark Operates: (Essex Division) Essex & 80 Park Place Hudson Gas Co.; East Newark (N. J.) Gas Lt. Co.; Morristown (N. J.) Gas Lt. Co.; (Hudson Division Hudson Co. Gas Co.; (Passaic Division) Patterson & Passaic Gas & Elec. Co.; (Southern Division) South Jersey Gas; Elec. & Trac. Co.; Princeton Lt., Ht. & Pr. Co.; (Central Division) Somerset, Union & Middlesex Ltg. Co.; New Brunswick (N. J.) Gas Lt. Co.; Shore Ltg. Co.; (Bergen Division) Gas & Elec. Co. of Bergen Co.; Ridgewood (N. J.) Gas Co.	New Jersey
FLORIDA UTILITIES COTrenton Moon Clay & Kaolin Co. 715 Broad St. F Owns gas companies at Palm Beach, W. Palm Beach, and Ocala, Fla.	New Jersey Bk. Bldg.
BROOKLYN UNION GAS COBrooklyn Owns and operates: Flatbush Gas Co., 29th ward, Brooklyn; Newton Gas Co., 2d ward, Queens' Jamaica (Long Isl- and) Gas Co.; Woodhaven (L. I.) Gas Co.; Richmond Hill & Queens County Gas Lt. Co., 4th ward, Queens.	New York
EASTERN OIL CO.,	New York
other Maryland towns, and Terra Alta, W. Va.; West Union (W. Va.) Gas Co.; Salem (W. Va.) Natural Gas Co.; Glen- ville (W. Va.) Nat. Gas Co. SOUTH SHORE NATURAL CASE &	
SOUTH SHORE NATURAL GAS & FUEL CoBuffalo	New York
Owns: Dunkirk, N. Y., and other 842 Marine Bk. points.	Bldg.

AMERICAN GAS SYNDICATES AND THEIR HOLDINGS —Continued Company City Stat

Contin	ued	
Company	City	State
EMPIRE COKE CO Furnishes gas for Empire Gas & Elec Co., which does all the gas and elec- tric business in Auburn, Weedsport Cayuga, Seneca Falls; Waterloo Geneva, Phelps, Palmyra, Newark Lyons and Clyde.		New York
AMERICAN LIGHT & TRACTION CO Own practically all the capital stock of Binghamton (N. Y.) Gas Works; Con- solidated Gas Co., of Long Branch, N J., Detroit (Mich.) City Gas Co.; Grand Rapids (Mich.) Gas Light Co.; Madi- son (Wis.) Gas & Elec. Co.; Milwaukee (Wis.) Gas Lt. Co.; Muskegon (Mich.) Traction & Ltg. Co.; St. Joseph (Mo.) Gas Co.; St. Paul (Minn.) Gas Light Co.; Son Antonio (Tex.) Pub. Serv. Co.; So. St. Paul (Minn.) G. & E. Co.; West Allis (Wis.) Gas Co.; Wuawa- tosa (Wis.) Gas Co.		New York
AMERICAN POWER & LIGHT CO	New York	New York
CONCOLIDATION CALC TO	71 Broadway	
 CONSOLIDATED GAS CO	New York 124-130 E. 15th St.	New York
Controls: Homer and Cortland (N, Y.) Gas Light Co.; Norwich (N. Y.) Gas & Electric Co.; Ithaca (N. Y.) Gas & Elec. Co.; Oneonta (N. Y.) Lt. & Pr. Co.; Greenville (O.) Gas Lt. Co.; Van Wert (Ohio) Gas Light Co. and Ky. Service Co., with plants at Bowling Green, Owensboro, Frankfort, Hopkins- ville Ky. You Charlet II.	43 Exchange Place	New York
As Eastern States Pub. Service Co. operates N. J. Gas & Elec. Co., Dover, N. J.; Newton (N. J.) Gas & Elec. Co.; Lambertville (N. J.) Pub. Service Co. Also owns: Port Arthur (Tex.) Gas & Pr. Co. and Utah Valley Gas &	115 Broadway	New York
Coke Co., Provo, Utah. HENRY L. DOHERTY & CO. (Cities Service Co.)	New York 60 Wall St.	New York

Company

Company Natural Gas Co. of Hamilton, Ont., Can.; Fremont (Neb.) Gas, Elec. Lt. & Pr. Co.; Knoxville (Tenn.) Gas Co.; Lebanon (Pa.) Gas & Fuel Co.; Lin-coln (Neb.) Gas & Elec. Lt. Co.; Meri-dian (Miss.) Lt. & Ry. Co.; Montgom-ery (Ala.) Lt. & Water Pr. Co.; Pueblo (Colo.) Gas & Fuel Co.; Spokane (Wash.) Gas & Fuel Co.; Spokane (Wash.) Gas & Fuel Co.; Spokane (Wash.) Gas & Fuel Co.; Trumbull Public Service Co., Warren & Niles, O.; Toledo (O.) Ry. & Lt. Co.; Webb City & Carterville Gas Co., Webb City, Mo.; Woodstock (Can.) Gas Lt. Co.; Empire Gas & Fuel Co. of Kansas, Missouri and Oklahoma; Hattiesburg (Miss.) Traction Co.; Arkansas Valley Gas Co.; Glenwood Nat. Gas Co., Ltd. (Can.); Manufacturers Nat. Gas Co., Ltd.; S. W. Oklahoma Gas & Fuel Co.; Weshi-ta (Okla.) Gas & Fuel Co.; Western Oklahoma Gas & Fuel Co.; Western Oklahoma Gas & Fuel Co.; Western Oklahoma Gas & Fuel Co.; Medina (O.) Gas & Fuel Co.; Mansfield (O.) Gas Lt. Co.; Ingersoll (Can.) Gas Lt. Co.; Thoroid (Can.) Nat. Gas Co. (Intic S. U. Okla.); Reserve Gas Co. (Okla.); Toledo (O.) Ry. & Lt. Co.; Western Oklahom, Cas Co.; So. (Okla.); Reserve Gas Co. (Okla.); Toledo (O.) Ry. & Lt. Co.; Venture Gas Co., Morral, O.; Frost Gas Co. (Okla.); Reserve Gas Co. (Okla.); Toledo (O.) Ry. & Lt. Co.; Venture Gas Co., Morral, O.; Frost Gas Co., owning Brooton (N. Y.) Gas & Fuel Co.; Westfield, N. Y. Westfield, N. Y.

Westneid, N. Y.
ELECTRIC BOND & SHARE CO.........New York Fiscal Agents: Carolina Power & Lt. 71 Broadway Co.; Raleigh & Durham, N. C., operat-ing Asheville (N. C.) Pr. & Lt. Co.; Yadkin River Pr. Co.; Utah Securities Co., controlling Utah Pr. & Lt. Co., which controls Utah Lt. & Tr. Co. at Ogden, Salt Lake City, etc., Utah; American Pwr. & Lt. Co., operating Portland (Ore.) Gas & Coke Co.; Kan-sas Gas & Elec. Co. of Wichita, Kans., Nebr. Pr. Co., Omaha; Pacific Pr. & Lt. Co. of Vancouver, Yakima and Walla Walla, Wash.; Pendleton and Astoria, Ore., and Lewiston, Idaho; Texas Pr. & Lt. Co. of Brownwood, Denison, Cleburne, Paris and Waco, Texas; Galveston (Tex.) Gas Co.; El Paso (Tex.) Gas & Fuel Co.; Newton (Kans.) Gas & Fuel Co.; National Se-curities Corp., controlling Idaho Pr. Co.

FEDERAL LIGHT & TRACTION CO.....New York Operates. Albuquerque (N. M.) Gas & 60 Broadway Electric Co.; Consumers' Gas Co., Hot Springs, Ark.; Tucson (Ariz.) Gas. Elec. Lt. & Pr. Co.; Springfield (Mo.) Gas & Elec. Co.; Trinidad (Colo.) Elec.

New York

New York

City

State

Company	City	State
Trans. Ry. & Gas Co.; Gray's Harbor Ry. & Light Co., of Aberdeen, Wash.; and various electric and railway com- panies.		
GENERAL GAS & ELECTRIC CO Controls: Rutland (Vt.) Ry., Lt. & Pr. Co.; Sandusky (O.) Gas & Elec. Co.; Interurban Gas Co., Easton, Penn.	New York 50 Pine St.	New York
COMMONWEALTH PR. RY. & LT. CO Controls: Michigan Light Co. in Bay City, Flint, Jackson, Kalamazoo, Man- istee, Pontiac and Saginaw, Mich.: Springfield, Ill., Gas & Elec. Co.; Evansville (Ind.) Public Utilities Co.; Central Illinois Light Co., Peoria and Pekin, Ill.	New York 14 Wall St.	New York
GENERAL ENGINEERING & MANAGE- MENT CORP. Controls: Peoples Gas & Electric Co., Chilicothe, Mo.; Trenton (Mo.) Gas & Elec. Co.	New York 141 Broadway	New York
NASSAU & SUFFOLK LIGHTING CO Operates: Nassau & Suffolk Lighting Co.'s plants at Garden City, Hempstead, Freeport, Merrick, Mineola, Roosevelt and other Long Island points.	149 Broadway	New York
NATIONAL FUEL GAS CO Controls: United Natural Gas Co., Oil City, Pa.; Iroquois Natural Gas Co., Buffalo, N. Y.; Provincial Natural Gas Co. of Ontario, Niagara Falls, Ont., Can.; Pennsylvania Gas Co., Warren, Pa.; Clarion Gas Co., Oil City, Pa.	New York 26 Broadway	New York
Operates: Twin State Gas & Elec. Co. of Dover, N. H., Bennington and Brat- tleboro. Vt.	New York 111 Broadway	New York
NATIONAL UTHIJTIES CO Operates: Ft. Scott & Nevada (Mo.) Lt., Ht., Wtr. & Pr. Co.; N. J. Gas & Elec. Co., Dover, N. J.; Port Arthur (Tex.) Gas & Pr. Co.; Hillsboro (O.) Lt. & Fuel Co.	New York 61 Broadway	New York
THE NORTH AMERICAN CO Operates: St. Louis Co. Gas Co., Web- ster Groves, Mo.; Wisconsin Edison Co., operating Wisconsin Gas Electric Co. of Rache, Kenosha, Watertown and Bur- lington, Wis.; North Milwaukee Lt. & Pr. Co.; Wells Pr. Co.; Mil. Elec. Ry. & Lt. Co.; Mil. Lt., Ht. & Tr. Co	New York 30 Broad St.	New York
PEARSON ENGINEERING CORP'N Operates gas plant at Rio de Janeiro,	New York 115 Broadway	New York
Brazil, THE UNITED GAS & ELECTRIC EN- GINEERING CORPORATION Controls: Altoona (Pa.) Gas Light & Fuel Co.; Citizens Gas & Fuel Co., Terre Haute, Ind.; Colorado Springs (Colo.) Lt., Ht. & Pr. Co.; Consumers Electric Light & Pr. Co., New Orleans; Elmira (N. Y.) Water, Lt. & Rd. Co.; Harrisburg (Pa.) Lt. & Pr. Co.; Hous- ton (Tex.) Gas & Fuel Co.; Lockport (N. Y.) Lt., Ht. & Pr. Co.; Richmond (Ind.) Lt., Ht. & Pr. Co.; Richmond (Ind.) Lt., Ht. & Pr. Co.; Richmond	New York 61 Broadway	New York
(and) and at the the out of the day of		

	-Continu	ea	
Company		City	State
Electric Co., Bloomington, Barre (Pa.) Co.; Birming Ry., Lt. & Pr. Co.; Houston & Pr. Co.; New Orleans (La. Co.; Lancaster (Pa.) Gas Co.; Columbia Gas Co., Lan Leavenworth (Kans.) Lt., H	Ill.; Wilkes ham (Ala.) h (Tex.) Lt.) Gas Light Lt. & Fuel caster, Pa.; t. & Pr. Co.		
H. D. WALBRIDGE & CO Controls: Dallas (Tex.) County Gas Co., Dallas, T town (Pa.) Fuel Supply Pub. Serv. Co., Clearfield, F	Gas Co.; ex.; Johns- Co., Penn. Pa.	New York 14 Wall St	New York
 THE[*] J. G. WHITE MAN CORP'N. Operates the Associated G tric Co., controlling Greenvi Lt. Co.; Homer & Cortland Cortland, N. Y.; Ithaca (N. Co.; Norwich (N. Y.) Gas & Oneonta (N. Y.) Lt. & Pr Wert (O.) Gas Lt. Co.; al the Kentucky Public Service ating in Bowling Green, Hopkinsville and Owensborr Clarksville, Tenn.; Eastern & Pr. Co., Pottsville, P (Mont.) Lt. & Ry. Co., Gas & Elec. Co., Hastin Palatka (Fla.) Public Service 	as & Elec- lle (O.) Gas Gas Lt. Co., Y.) G. & E. & Elec. Co.; x. Co.; Van so operates e Co., oper- Frankfort, Pa. Lt., Ht. a.; Helena Thornapple gs, Mich.; Co.; San- Co.		ge Place New York
ALLEN & PECK Control Newport News and (Va.) Ry., Gas & Elec. Co.			New York lg.
(va.) GAS & ELEC. O Operates the Utica plant; Co Pr. Co., Canastota; Utica G. Little Falls; Utica G. & E. mer; Utica G. & E. Co., II Glens Falls (N. C.) Gas & Whitehall (N. Y.) Con. Lt. Sandy Hill & Ft. Edwar United Gas, Elec. Lt., Ht. & NOPTH CAPOUNA DUPLIC	entral N. Y. . & E. Co., Co., Herki- lion, N. Y.; Elec. Co.; & Pr. Co.; d (N. Y.) Fuel Co.	. Utica	· New York
CO. Operates: No. Car. Pub. Greensboro, Concord, High isbury and Spencer.	Ser. Co., Point, Sal-	. Greensboro	North Carolina
CAROLINA POWER & LIGH' Owns Carolina Pr. & Lt. Co N. C., and Raleigh, N. C.	o., Durham,	. Raleigh	North Carolina
CONTINENTAL GAS AND	ELECTRIC	Clausland	Ohis
CONSOLIDATED GAS, ELEC.	& Pr. Co., Co., Shen- Co., Platts-) Gas Co.; don (Man., aska Gas & Elec. Co., & WATER	Cuyahoga	Bldg. Ohio
Operates: Menominie (Wis. Hurley (Wis.) Gas Co., (Mich.) Gas Co., Iron Mount Gas Co.	J Gas. Co., Ironwood ain (Mich.)	Cleveland 1123 Illumi	Ohio inating Bldg.

Company	City	State
OHIO CITIES CO Owns Columbus (O.) Gas & Fuel Co.; Dayton (O.) Gas Co. and Springfield (O.) Gas Co.	.Columbus	Ohio
OHIO FUEL SUPPLY CO		Ohio
OHIO GAS LIGHT & COKE CO Operates, Plants at Napoleon, Wau- seon, Bryan, Stryker, Archbold, Mont- pelier and Delta, Ohio; Central States Gas Co., Vincennes, Ind. (operates at Vincennes and supplies Lawrenceville, Bridgeport, Sumner and Olney); Illinois Gas Co., Lawrenceville, Ill. (operates at Lawrenceville, Bridgeport, Sumner and Olney; Wabash Gas Co., Robinson, Ill. (operates at Robinson).	. Napoleon	Ohio
EMPIRE GAS & FUEL CO Owns, either directly or through owner- ship of securities, leases in Kansas and Oklahoma.		Oklahoma
PACIFIC POWER & LIGHT CO (See Elec. Bond & Share Co., N. Y.) Gas plants at Walla Walla, Yakima and Vancouver, Wash.; Astoria and Pen- dleton, Ore., and Lewiston, Idaho.	Portland Gasco Bldg.	Oregon
THE AMERICAN GAS CO Owns: Bangor (Me.) Gas Lt. Co.; Bur- lington (Ct.) Lt. & Pr. Co.; Consoli- dated Lt. & Pr. Co. of Kewanee, Ed- wardsville, Sheffield and Galva, Ill.; Kingston (N. Y.) Gas & Elec. Co.; Lu- zerne Co. Gas & Elec. Co. of Kingston, Nanticoke, Hazelton, Plymouth and Fort, Pa.; Phila. Suburban Gas & Elec. Co. of Chester, Coatesville, Potts- town, Wyncote, West Chester, Phoe- nixville, Royersford, Spring City and other Pa. points; Petersburg (Va.) Gas Co.; Portage (Wis.) American Gas Co.; Rockford (Ill.) Gas Lt. & Coke Co.; St. Clair Co. Gas & Elec. Co. of Bell- ville (also operating E. St. Louis (Ill.); Waukesha (Wis.) Gas & Elec. Co.; Waterloo (Ia.) Citizens Gas & Elec. Co.		
EASTERN LIGHT & FUEL CO Operates: New Jersey Gas Co., Glass- boro, N. J.; Schuylkill Haven (Pa.) Gas & Water Co.; Wildwood (N. J.) Gas Co.; Pottsville (Pa.) Gas Co.	.Philadelphia Real Estate Tru	Pennsylvania st Bldg.
DAY & ZIMMERMANN Operate gas plants of the Penn. Central Light & Power Co. at Huntingdon, Lewistown; Eastern Shore Gas & Elec. Co., controlling the Cambridge (Md.) Gas. Elec. Lt. & Pr. Co.	. Philadelphia	Pennsylvania
THE C. H. GEIST CO Operates: Freeport (III.) Gas Co.; Ro- anoke (Va.) Gas Lt. Co.; Atlantic City (N. J.) Gas Co.; Lansing (Mich.) Fuel & Gas Co., East Chicago (Ind.) & In- diana Harbor Water Co.; Northern Ala- bama Gas Co., of Florence, Ala.; Wil- mington (Del.) Gas Co.; Indianavolis 'Ind.) Water Co.	,Philadelphia Land Title Bldg.	Pennsylvania

Company	City	State
GIRARDVILLE GAS CO Operate: Girardville, Lansford and Frackville.	.Philadelphia 4014 Chestnut St.	Pennsylvania
GRIBBEL SYNDICATE CO Operates: Athens (Ga.) Gas Lt. & Fuel Co.; Helena (Ark.) Gas & Elec. Co.; Tampa (Fla.) Gas Co.	Philadelphia 1513 Race St.	Pennsylvania
Operates: Concord (N. C.) Gas Co, and the Georgetown (S. C.) Gas & Elec. Co.; Syracuse (N. Y.) Suburban Gas Co.; Gaston Co. Gas Co., Gastonia, N. C.; Choster City Cas Co. Choster S. C.	.Philadelphia Widener Bldg.	Pennsylvania
NATIONAL GAS, ELEC. LT. & PR. CO Operates: Cape May (N. J.) Lt. & Pr. Co.; Carbondale (Pa.) Gas Co.; Goshen (Ind.) Gas Co.; Joplin (Mo.) Gas Co.; Niles (Mich.) Gas Lt. Co.; Port Huron (Mich.) G. & E. Co.; Portsmouth (O.) Gas Co.; Quincy (III.) Gas, Elec. Lt. & Pr. Co.; Warsaw (Ind.) Gas Co. INTERURBAN GAS IMPROVEMENT CO.	.Philadelphia Witherspoon Bldg	Pennsylvania
PUBLIC SERVICE CO Operates: Bucks Co. Public Service Co., Newtown, Pa.; Doylestown (Pa.) Gas Co.; Southern Gas Improvement Co. of Elizabeth City, Henderson and Oxford, N. C.; Rock Hill (S. C.) Gas Co.	. Philadelphia Real Estate Trust . Philadelphia Real Estate Trust	Pennsylvania Bldg. Pennsylvania Bldg.
PHILADELPHIA SUBURBAN GAS & ELEC. CO. A consolidation of: Suburban Gas Co. of Philadelphia; Peoples Gas Co. of Pottstown; Coatesville Gas Co., Jen- kintown and Cheltenham Gas Co.; Hunt- ingdon Valley Light & Power Co., and Pottstown Light, Heat & Power Co., and others.	. Philadelphia S. W. Corner 7th Locust Sts.	Pennsylvania and
J. C. REED & CO Control: Key West (Fla.) Gas Co.; Colon (Republic of Panama) Gas Co.; Panama (Republic of Panama) Gas Co.		Pennsylvania
UNION RAILWAY SUPPLY CO Operates: Lewisburg (Pa.) Gas Co.; Ocean Co. Gas Co., Toms River, N. J.; Standard Gas Co. of Atlantic Highlands, Keansburg and Keyport, N. J.; Tucker- ton (N. J.) Gas Co.; Equitable Lt., Ht. & Pr. Co., Monmouth Shore Gas Co.	. Philadelphia Real Estate Trust	Pennsylvania Bldg.
& Pr. Co., Monmouth Shore Gas Co. UNITED GAS IMPROVEMENT CO Philadelphia Gas Works. Interested in Allentown-Bethlehem (Pa.) Gas Co.; Burlington (Ia.) Gas Lt. Co.; Charles- ton (S. C.) Con. Ry. & Ltg. Co.; Ches- ter Co. Gas Co., W. Chester, Pa.; Concord (N. H.) Lt. & Pr. Co.; Con- sumers Gas Co., Reading, Pa.; Counties Gas & Elec. Co., Philadelphia (operat- ing at Ardmore, Conshohocken, Nor- ristown); Des Moines (Ia.) Gas Co.; Fulton Co. Gas & Elec. Co., Glovers- ville, N. Y.; Harrisburg (Pa.) Gas Co.; Kansas City (Mo.) Gas Co.; Nashville (Tenn.) Gas & Heating Co.; New Gas	.Philadelphia Broad & Arch Sts	Pennsylvania

Company

Lt. Co., Janesville, Wis.; Northern In-diana Gas & Elec. Co.; Hammond, Ind.; (also operates Michigan City, South Bend and Ft. Wayne, Ind.; Northern Liberties Gas Co.; Pensacola (Fla.) Gas (Neb.) Gas Co.; Pensacola (Fla.) Gas (Neb.) Gas Co.; Pensacola (Fla.) Gas Co.; Peoples Gas Lt. Co.; Manchester, N. H.; St. Augustine (Fla.) Gas & Elec. Lt. Co.; Savannah (Ga.) Gas Co.; Sioux City (Ia.) Gas & Elec. Co.; Sioux Falls (S. D.) Gas Co.; Syracuse (N. Y.) Ltg. Co.; Vicksburg (Miss.) Gas Wks.; Wyandotte Co. Gas Co., Kansas City, Kan.; Northern Indiana Gas & Elec. Co., Hammond (see above) Hammond (see above).

- ARKANSAS NATURAL GAS CO.......Pittsburgh Pipes ratural gas to Little Rock and 223 Four many towns and cities in Arkansas.
- MANUFACTURERS LIGHT & HEAT CO. Pittsburgh Owns and controls: New Cumberland Water & Gas Co.; Venture Oil Co. and Sewickley Gas Co.
- OHIO FUEL SUPPLY CO.... Owns Northwestern Ohio Natural Gas Co., Toledo, Ohio; Point Pleasant (W. Va.) Natural Gas Co.; Miami Valley Gas & Fuel Co. Serves: Piqua, Troy, China Control Sidney, Covington, Tippecanoe City, Mt. Sterling, South Charleston, Tarlton, Starling, South Charleston, Tarlton, Williamsport, Urbana, Rockbridge, Bloomingburg, Sedalia, Fremont City, North Hampton, New Carlisle, Law-renceville, Beatty Town, Five Points, Lancaster, Middletown, Mt. Vernon, Xenia, Zanesville and 118 other Ohio towns towns.
- THE PHILADELPHIA CO......Pittsburgh Controls Chartiers Valley Gas Co.; 435 Sixth Mansfield & Chartiers Gas Co.; Penna. Mansheid & Charders Gas Co.; Peina. Nat. Gas Co., Philadelphia Co. of W. Va.; Union Gas Co. of McKresport; Allegheny Heating Co., Pittsburgh; Equitable Gas Co., Pittsburgh; Pitts-burg & W. Va. Gas Co.
- burg & W. Va. Gas Co. UNION NATURAL GAS CORP'N.......Pittsburgh Controls. Logan Natural Gas & Fuel Union Ba Co. of Lancaster, O.; Newark (O.) Natural Gas & Fuel Co.; Athens (O.) Gas Lt. & Elec. Co.; Buckeye Gas Co., of Circleville, O.; Bellevue (O.) Gas Co.; Marion (O.) Gas Co.; Fremont (O.) Gas, Elec. Lt. & Pr. Co.; Citizens Gas Lt. & Coke Co., Findlay, O.; Citizens Gas Lt. & Elec. Co. of Elyrla and Lorain; Manu-facturers Gas Co., Bradford, Pa.; War-ren & Chautauqua Gas Co., of Warren, Pa. Pa.
- WABASH GAS CO...... Serves: Marshall, Martinsville, Annap-.....Pittsburgh olis, Hutsonville and Porterville, Ill.

UNITED SERVICE CO..... Operates: Ohio Service Co., Coshocton, Cambridge, Dennison, New Phila-delphia; Warren (Pa.) Lt. & Pr. Co., Punxsutawney, Pa.; Wabash (Ind.) Water & Light Co.; E. Penna. Gas & Elec. Co., Bristol, Pa.; Hanover (Pa.)

City

State

Pennsylvania 223 Fourth Ave.

Pennsylvania 248 Fourth Ave.

.....Pittsburgh

435 Sixth Ave.

Pennsylvania

Pennsylvania

Pennsylvania Union Bank Bldg.

Pennsylvania Benedum-Trees Bldg.

....Scranton Pennsylvania 700 Scranton Life Bldg.

78

AMERICAN GAS SYNDICATES AND THEIR HOLDINGS

Company	City	C4++++
Lt., Ht. & Pr. Co.; Susquehanna (Pa.) Lt. & Pr. Co.	City	State
NORTHERN CENTRAL GAS CO Controls: Hagerstown (Md.) Lt. & Ht. Co.; Northern Central Gas Co. of Mil- ton, Watsontown, Dewart, Montgomery and Williamsport, Pa.	. Williamsport	Pennsylvania
BLACKSTONE VALLEY GAS & ELEC. CO. Controls: Pawtucket (R. I.) Elec. Co.; Pawtucket (R. I.) Gas Co.; Woonsocket (R. I.) Gas Co.	.Pawtucket	Rhode Island
TEXAS POWER & LIGHT CO Operates gas plants at Brownwood, Clebourne, Paris, Denison and Waco, the two latter natural gas.		Texas St. Broadway
TWIN STATE GAS & ELEC. CO Operates: Bennington Gas Lt. Co.; Brattleboro Gas Lt. Co.; Hoosick Falls Illuminating Co.; Dover Gas Light Co.; United Gas & Elec. Co.		Vermont .
SOUTHERN GAS & ELECTRIC CORP'N. Operates: Suffolk (Va.) Gas-Elec. Co.; Bluefield (W. Va.) Gas & Pr. Co.; Sum- ter (S. C.) Gas & Pr. Co., Henrico Co. Gas Co., Richmond, Va.; Gas Light Co. of Augusta, Ga.	.Richmond	Virginia
VIRGINIA RY. & PR. CO Owns City Gas Co., Norfolk, Va.	.Richmond	Virginia
NORTH PACIFIC PUBLIC SERVICE CO., Operates: Gray's Harbor Gas Co., Aberdeen, Wash.; Centralia and Che- halis Gas Co., Centralia, Wash.; Bre- merton-Charleston Lt. & Fuel Co.	.Tacoma 323 Tacoma Bldg	Washington 5.
BOYD E. HORNER SYNDICATE COLUMBIA GAS & ELECTRIC CO Controls: Union Gas & Elec. Co., Cin- cinnati; Union Lt., Ht. & Pr. Co., Cov- ington Ky	.Clarksburg .Huntington	West Virginia West Virginia
WISCONSIN SECURITIES CO Controls: Wis. Pub. Service Co. of Green Bay, De Pere and Two Rivers; Sheboygan Gas Light Co.; Wis. Ry., Lt. & Pr., La Crosse and Winona; Manito- woc & Northern Traction Co.; West Side Pr. Co. Manitowoc; Calumet Serv- ice Co., Chilton.	.Milwaukee First Nat. Bank	Wisconsin Bldg.
COLUMBUS GAS CONSTRUCTION CO Owns or controls: Little Falls-Darling Gas Co., Little Falls, Minn.; Taylor (Tex.) Gas Co.; Victoria (Tex.) Gas Co.; Oconomowoc (Wis.) Gas Co.	. Milwaukee Majestic Bldg.	Wisconsin
DOMINION GAS CO Owns: Dominion Natural Gas Co., Ltd., of Ontario, Canada; Brantford Gas Co.; Woodstock Gas Light Co., Ltd.; Beaver Oil & Gas Co.; Ingersoll Gas Co.; Thor- old Nat. Gas Co.; United Gas Co. Sup- plies: Natural gas from its own wells to Dunnville, Brantford, Galt, Tillson- burg, Simcoe, Paris, St. George, Dun- das Reartonville, Lowing Course and	.Hamilton O	ntario, Canada
QUEBEC RY., LT., HT. & PR. CO Operates: Quebec Ry., Lt. & Pr. Co.; Quebec Gas Co.; Frontenac Gas Co.;	.Quebec	Canada

By-Product Coke Plants in United States

		No. of
Owner or Operator	Location	Ovens
	Battle Creek, Mich. Detroit, Mich. Detroit, Mich. Flint, Mich. Kalamazoo, Mich. Wyandotte, Mich. Duluth, Minn. Duluth, Minn. St. Paul, Minn. St. Louis Mo.	
Camden Coke Co Seaboard By-Prod. Coke Co Semet-Solvay Co Empire Coke Co Dominion Iron & Steel Co Nova Scotia Steel & Coal Co. Dover By-Prod. Coke Co	Camden N. J. Jersey City N. J. Jersey City, N. J. Buffalo, N. Y. Geneva, N. Y. Syracuse, N. Y. Sydney, N. S. Sydney, N. S. Sydney, Mines Canal Dover, Ohio.	$\begin{array}{c} \dots 150 \\ 55 \\ \dots 110 \\ 60 \\ \dots 46 \\ \dots 40 \\ \dots 120 \\ \dots 520 \\ \dots 106 \\ \dots 26 \end{array}$
Cleveland Furnace Co River Furnace Co American Steel & Wire Co Hamilton Otto Coke Co	Canton, Ohio Cleveland, Ohio Cleveland, Ohio Cleveland, Ohio Hamilton, Ohio Ironton, Ohio Lorain, Ohio Portsmouth, Ohio. Toledo, Ohio. Youngstown, Ohio.	
Algoma Steel Co Algoma Steel Co Phil. Suburb. Gas & Elec Co Carnegie Steel Co	Youngstown, Ohio Youngstown, Ohio Hamilton, Ont Sault Ste. Marie, Ont Chester, Pa Clairton, Pa Dunbar, Pa.	$ \begin{array}{r} 60 \\ $
Cambria Steel Co. Cambria Steel Co. Bethlehem Steel Co. Bethlehem Steel Co. Lehigh Coke Co. Providence Gas Co.	Farrell, Pa Glassport, Pa. Johnstown, Pa. Johnstown, Pa. Lebanon, Pa. Steelton, Pa. Steelton, Pa. So. Bethlehem, Pa. Providence, R. I.	$\begin{array}{c} \dots & 300 \\ \dots & 147 \\ \dots & 462 \\ \dots & 318 \\ \dots & 60 \\ \dots & 120 \\ \dots & 424 \\ \dots & 40 \end{array}$
Memphis Gas & Electr. Co Seattle Lighting Co Fairmount By-Prod. Co LaBelle Iron Works National Tube Co Northwestern Iron Co Milwaukee Coke & Gas Co Northwestern Iron Co.	Memphis, Tenn. Seattle, Wash. Fairmount, W. Va. Follansbee, W. Va. Benwood, W. Va. Mayville, Wis. Milwaukee, Wis. Mayville, Wis. Chattanooga, Tenn.	$ \begin{array}{r} 110 \\ 94 \\ 120 \\ 72 \\ 160 \end{array} $

and Canada (Naphtha Producers)

	Ammonia	
Kind of oven Coal C	loke as NH3	
Gas Machinery, inclined chambers 36,000 2.	5,300 90	0
Semet-Solvay	2,000 2,160	
Semet-Solvay	9,000 3,690	
Semet-Solvay 1,343,300 1,000 Park Gas Machinery, inclined chambers 96,400 66 Parker-Russell, Horiz. 43,800 3	7,500 240 0,700 131	
Otto	5,800 235	
	0,000 1,500	
Otto 200,000 14	4,000 500	0
Koppers 380,000 27	3,600 1,140	
Koppers 320,000 24	0,000 880)
Otto	2,000 990	0
Koppers 340,500 25	5,350 937	
Koppers 681,000 51	0,700 1,974	
	9,500 965	
Semet-Solvay 146,000 10 Semet-Solvay 65,000 4	2,200 401 5,500 195	
Koppers	8,400 2,160	
Otto	8,080 4,576	
30 Bauer, 160 Bernard 159,000 11	0,000	
A Roberts 120,000 8	7,600 300	0
Koppers 280,000 20	4,400 770	0
	7,500 1,125	
ppers	9,000 3,578	
ppers	9,500 3,162	
	8,000 720	
	3,600 3,630 9,900 1,944	
	8,800 1,540	0
	7,600 1,435	8
Koppers	4,600 2,805	5
	9,000 3,575	
	4,500 1,788	
	0,400 779	
Wilputte	7,000 650	
	0,700 1,874	
	7,500 313 0,000 12,000	
	0,000 2,400	
	3,600 682	
	A. 1	
	1,000 2,283	
	5,000 650 0,000 5,000	
Otto 529.200 33	8,888 1,293	
Otto	3,700 3.440	
228 Otto, 90 Semet-Solvay 887,000 633	8,000 2,267	7
Koppers	0,000 1,031	
Semet-Solvay	1,500 1,354	
Koppers	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	
Gas Mach., including slots 59,000 4	1,300 148	
	9,200 90	-
Koppers	5,300 1,677	-
	9,000 743	
Otto 320,000 23	0,400 800	
	9,000 2,104	4
	7,000	
Semet-Solvay 173,000 12	4,000 432	5

The Flow of Oil in Pipes

The quantity of oil of the same viscosity as water discharged through a pipe is in accordance with the following formula: Q = av in which Q is the quantity discharged in cubic feet per

second

a is the pipe area in square feet

v is the velocity in feet per second

To find the velocity discharged from the pipe line, knowing the head, length and inside diameter use the following formula:

$$\frac{Q}{R} = v = m_{i} / \frac{hD}{L + 54 D}$$

in which $\mathbf{v} =$ approximate mean velocity in feet per second m = coefficient from table below

D = diameter of pipe in feet

h = total head in feet

L = total length of line in feet Value of Coefficient "m"

Diamet	er of Pipe	States - IS	Diamete	er of Pipe	
Feet	Inches	m	Feet	Inches	m
0.1	1.2	23	1.5	18	53
0.2	2.4	30	2.0	24	57
0.3	3.6	34	2.5	30	60
0.4	4.8	37	3.0	36	62
0.5	6.0	39	3.5	42	64
0.6	7.2	42	4.0	48	66
0.7	8.4	44	5.0	60	68
0.8	9.6	46	6.0	72	70
0.9	10.8	47	7.0	84	72
1.0	12.0	48	10.0	120	77

The above coefficients are averages deduced from a large number of experiments. In most cases of pipes carefully laid and in fair condition they should give results within 5 to 10% of the truth.

Example: Given the head, h = 50 feet, the length, L = 5280 feet, and the diameter D = 2 feet; to find the velocity and quantity of discharge.

The value of the coefficient m from the table when D = 2 feet is m = 57.

Substituting these values in the formula, we get:

$$v = 57 \left(\frac{50 \times 2}{5280 + 108}\right) = 57 \left(\frac{100}{5388}\right) = 57 \times 0.136 = 7.52 \text{ ft.}$$

per second.

To find the discharge in cubic feet per second, multiply this velocity by the area of cross section of the pipe in square feet. Thus, $3.1416 \ge 1.2 \ge 24.35$ cubic feet per second.

Since there are 7.48 gallons in a cubic foot, the discharge in gallons per second = $24.35 \times 7.48 = 182.1$.

The above formula is only an approximation, since the flow is modified by bends, joints, incrustations, etc. Wrought pipes are smoother than cast iron ones, thereby presenting less friction and less encouragement for deposits; and being in longer lengths, the number of joints is reduced, thus lessening the undesirable effects of eddy currents.

Principal Pipelines

Pipeline Mile Alluwe Pipeline Co. (Kas. Oil	age
Ref. Co.)	40 70
American Petroleum Co	20
Associated Oil Co	105
Associated Oil Co	60
Arkansas City Pipeline Co	
Associated Pipeline Co	281
Associated Pipeline Co	278
Bessemer Pipeline	
Buckeye Pipeline Co., Lima Division	700
Buckeye Pipeline Co., Macks- burg Division	350
Colive Oil Co Cosden & Co	
Cosden Pipeline Co	
Crescent Pipeline Co	315
Crown Pipeline Co	58
Cumberland Pipeline Co	475
Emery Pipeline Co	480
Empire Pipeline Co	85
Empire Pipeline Co	67
Empire Pipeline Co	70
Empire Pipeline Co	55
Empire Pipeline Co	17
Eureka Pipeline Co4	,300

Franklin Pipe Co	• • • •
General Pipeline Co	156
General Pipeline Co	52
Gulf Pipeline Co	458
Gulf Pipeline Co	76
Gulf Pipeline Co	117
Gulf Pipeline Co	124
Gulf Pipeline Co. of Okla	275

	pacity, barrels
From Alluwe Dist., Okla., to Coffeyville, Kans From Salt Lake Dist., Cal., to Los Angeles, Cal From Humble to E. Houston,	2,500
From Salt Lake Dist., Cal., to	9,000
From Humble to E. Houston,	0,000
From Coalings Dist Cal to	
Monterey, Cal.	15,000
Monterey, Cal From Santa Barbara Co., Cal. to Gaviota, Cal From Blackwell to Arkansas	23,000
From Blackwell to Arkansas	
From Blackwell to Arkansas City, Kans. From Kern River Dist., Cal., to Port Costa, Cal From Kern River Dist., Cal., to Port Costa, Cal From Titusville, Pa., to W. Pa. From Ohio-Ind. state bound- ary to Ohio-Penn. state boundary. From Eastern Ohio to Ohio- Penn. and Ohio-W. Va. boundary.	
to Port Costa, Cal From Kern River Dist Cal	13,000
to Port Costa, Cal	26,000
Pa	
From Ohio-Ind. state bound-	
boundary	75,000
From Eastern Ohio to Ohio-	
	10,000
From Healdton to Ardmore	
heart, Okla.	500
West Tulsa, Okla,	30.000
From Greggs, Pa., to Marcus	5 000
From Okmulgee, Okla., to	2,000
 From Healdton to Ardmore From adjacent wells to Bigheart, Okla. From various Okla. Grom Greggs, Pa., to Marcus Hook, Pa. From Okmulgee, Okla., to Muskogee, Okla. From Southeastern Kentucky to Kentucky-W. Va. bound. From adjacent oil dist. to Bradford, Pa. From Eldorado and Augusta, Kans. to Ponca City. Okla 	
to Kentucky-W. Va. bound.	10,000
From adjacent oil dist. to Bradford Pa	1.000
From Eldorado and Augusta,	-,
From Ponca City, Okla., to	
Norfolk, Okla.	
From Eldorado and Augusta, Kans., to Ponca City, Okla From Ponca City, Okla., to Norfolk, Okla., From northern' Oklahoma to Independence, Kans. From Healdton, Okla., to Gainesville, Tex. (Total) From Gainesville, Tex. to Red	
From Healdton, Okla., to	35,000
From Gainesville, Tex., to Red	55,000
River, Tex From Kentucky - W. Va.	8 inch
boundary and Ohio-W. Va.	
From Gainesville, Tex., to Ked River, Tex. From Kentucky W. Va. boundary and Ohio-W. Va. boundary to W. VaPa. boundary From adjacent fields to Franklin, Pa. Los Angeles and San Pedro From Liebere, Cal., to Mojave, Cal.	65,000
From adjacent fields to	150
From Midway Dist., Cal., to	100
Los Angeles and San Pedro	25,000
Cal.	5,000
From TexOkla. State Line to Port Arthur, Tex	28,000
From Batson, Tex., to Sour	14 000
From LaTex. State Line to	14,000
Lufkin Station, Tex	9,600
Fort Worth, Tex	7,000
From Liebere, Cal., to Mojave, Cal. From TexOkla. State Line to Port Arthur, Tex From Batson, Tex., to Sour Lake and Houston From LaTex. State Line to Lufkin Station, Tex From Saltillo Station, Tex to Fort Worth, Tex From Bartlesville, Okla., to OklaTex. boundary	25,000

PRINCIPAL PIPELINES—Continued

Pipeline	Mileage
Gulf Refining Co. of La	21
Gulf Pipeline Co	305
Gulf Pipeline Co	124
Gulf Pipeline Co	98
Gulf Pipeline Co	86
Gulf Pipeline Co	63
Hale Petroleum Co	20
Illinois Pipeline Co	1,300
Illinois Pipeline Co	25
Illinois Pipeline Co	20
Illinois Pipeline Co	20
Imperial Pipeline Co., Ltd	155
Indiana Pipeline Co	800
Magnolia Petroleum Co	569
Magnolia Petroleum Co	137
Magnolia Petroleum Co	150
Magnolia Petroleum Co. (Don Line) Magnolia Petroleum Co	uble 800 76
Maryland Pipeline Co	
Midwest Refining Co	90
National Pipeline Co	60
National Pipeline Co	110
National Transit Co	205
National Transit Co	175
National Transit Co	35
National Transit Co	70
National Transit Co	70
Natrona Pipeline Co	90
New York Transit Co	130
New York Transit Co	1,100
Northern Pipe Co	525
Oklahoma Pipeline Co	229
Paragon Refining Co	237
Pierce Pipeline Co	135
•	

	barrels
From Mansfield, La., to La Texas boundary	
From Olean, Tex., to Red	8 inch
From Fort Worth, Tex., to Saltillo. Tex.	6 inch
From Caddo Tex to Lufkin	6 inch
From Ranger, Tex., to Fort Worth, Tex. From Houston to Sour Lake,	8 inch
Tex	6 inch
From Eldorado, Kan., to Wichita, Kan From Alton, Ill., to Center- bridge, Pa From Grass Creek, Wyo., to Chatham, Wyo From Elk Basin, Wyo., to Frannie, Wyo From Big Muddy, Wyo., to Casper, Wyo From Sarnia, Ont., to Cygnet, Ohio	7,500
bridge, Pa	60,000
Chatham, Wyo From Elk Basin, Wyo., to	
Frannie, Wyo From Big Muddy, Wyo., to	• • • •
Casper, Wyo. From Sarnia, Ont., to Cygnet,	20,000
From Griffith, Ind., to In- diana-Ohio boundary From Electra, Tex., to Sabine,	8 inch
From Electra, Tex., to Sabine,	60.000
From Healdton, Okla., to Fort Worth. Tex.	60.000
From Cushing Dist., Okla., to Addington, Okla.	50,000
From Red River, Tex., to Beaumont, Tex.	8 inch
From Electra, Tex., to Sabine, Tex. From Healdton, Okla., to Fort Worth, Tex. From Cushing Dist., Okla., to Addington, Okla. From Red River, Tex., to Beaumont, Tex. From Electra, Tex., to Bowie, Tex. From Kay County, Okla., to Vonca City, Okla. From Salt Creek Dist., Wyo., to Caspar, Wyo. From oil fields in Wood Co., Ohio, to Findlay, Ohio From oil fields in southeastern Ohio to Marietta, Ohio	8 inch
From Kay County, Okla., to Ponca City, Okla	• • • •
to Caspar, Wyo From oil fields in Wood Co	13,000
Ohio, to Findlay, Ohio From oil fields in southeastern	1,000
Ohio to Marietta, Ohio From Nedska, Pa., to New	500
York-Pa. boundary From Colegrave, Pa., to Mil-	
From Milway, Pa. to Fawn Grove Pa	75 000
From Milway, Pa., to Point Breeze, Pa.	10,000
From Milway, Pa., to Center- bridge, Pa.	
From Salt Creek, Wyo., to Casper, Wyo	6 inch
ary to Buffalo, N. Y.	55,000
Onio, to Findlay, Onio From oli fields in southeastern Ohio to Marietta, Ohio From Nedska, Pa., to New York-Pa. boundary From Colegrave, Pa., to Mil- way, Pa From Milway, Pa. to Fawn Grove Pa From Milway, Pa., to Point Breeze, Pa From Milway, Pa., to Center- bridge, Pa From Salt Creek, Wyo., to Casper, Wyo From PaNew York bound- ary to Buffalo, N. Y From Olean, N. Y. to Bay- onne, N. J., and Long Is- land, N. Y From PaOhio boundary to PaN. Y. boundary From Creek County, Okla., to McCurtain, Okla.	5
From PaOhio boundary to PaN. Y. boundary	60,000
From Creek County, Okla., to McCurtain, Okla.	35,000
McCurtain, Okla From Sandusky County Ohio, to Toledo, Ohio From Healdton, Okla., to Fort Worth, Tex	4,000
Fort Worth, Tex.	

PRINCIPAL PIPELINES—Continued

Pipeline Mile Prairie Pipeline Co	age
Prairie Pipeline Co. (Double	
(Line) Prairie Pipeline Co	260 701
	,820
Prairie Pipeline Co	90
Prairie Pipeline Co	85
Producers' & Refiners' Pipe Line Co.	210
Producers' Transportation Co	41
Producers' Transportation Co	50
Producers' Transportation Co	39
Producers' Transportation Co	13
Producers' Transportation Co	3
Producers' Transportation Co	74
Pure Oil Pipeline Co	250
Rio Brava Oil Co	13
Pierce Pipeline Co	76
Sinclair-Cudahy Pipeline Co	750
Sinclair-Cudahy Pipeline Co	70
Sinclair-Cudahy Pipeline Co	340
Sinclair-Cudahy Pipeline Co	
Sinclair-Cudahy Pipeline Co	
Southern Pipeline Co1,	,130
Southwestern Penn. Pipelines1	,650
Standard Oil Co., Cal	281
Standard Oil Co., Cal	32
Standard Oil Co., Cal	29
Standard Oil Co., Cal	21
Standard Oil Co., Cal	24
Standard Oil Co., Cal	45
Standard Oil Co., Cal	32
Standard Oil Co. of La	522
Sun Co	250
Sun Pipeline Co	100
S'un Pipeline Co	53

	pacity, barrels
From Drumright Okla to	
Ranger, Tex., to Red	8 inch
From Kanger, Tex., to Red River, Tex. From Cushing Dist., Okla., to Humboldt, Kan., to Su- gar Creek, Mo., and Wood River, Ill.	8 inch
Humboldt, Kan.	100,000
gar Creek, Mo., and Wood	
River, Ill	94,000
Ida, La.	31,000
to Neodesha, Kan.	
Titusville, Pa	9,000
From Coalinga Dist., Cal., to Junction, Cal.	15,000
From Sunset Dist., Cal., to	20,000
From McCurtain, Okla., to Ida, La. From Eldorado-Augusta Kan., to Neodesha, Kan. From Watertown, Ohio, to Titusville, Pa. From Coalinga Dist., Cal., to Junction, Cal. From Sunset Dist., Cal., to Junction Cal. From Kern River Dist., Cal., to McKittrick, Cal. From Kern River Dist., Cal., to Trunk Line, Cal. Trunk Line, Cal.	20,000
From Lost Hills Dist., Cal.	
to Trunk Line, Cal From Belridge Dist. Cal. to	
Trunk Line, Cal From Junction, Cal., to Port	
San Luis, Cal., to Port From Morgantown, W. Va., to	30,000
From Morgantown, W. Va., to Marcus Hook, Pa.	10,000
Marcus Hook, Pa From Saratoga, Tex., to Sour Lake Tex	1,500
From Saratoga, Tex., to Sour Lake, Tex From Fort Worth, Tex., to Red River, Tex From Cushing Dist., Okla., to Kansas City and Chicago From Cushing Dist., Okla., to Coffeyville, Kan From branches and lateral in Okla. and Kansas	1,500
From Cushing Dist., Okla., to	8 inch
From Cushing Dist., Okla., to	
Coffeyville, Kan	
Okla. and Kansas	50,000
From Cushing field, Okla., to Whiting, Ind.	8 inch
From Cushing field to Heald- ton, Okla.	8 inch
from Cusning neid to Heald- ton, Okla. From PaW. Va. boundary to Philadelphia, Pa Operates exclusively in south- western Pennsylvania. From Kern River Dist., Cal., to Richmond, Cal.	51,000
Operates exclusively in south-	
From Kern River Dist., Cal., to	45,000
From Midway Dist., Cal. to	65,000
Bakersfield, Cal	65,000
Richmond, Cal. From Midway Dist., Cal., to Bakersfield, Cal From Coalinga Dist., Cal., to Mendota, Cal. From Lost Hills Dist., Cal., to Pand Cal	28,000
I Und, Call	20,000
El Segundo. Cal	27,000
From Newhall Dist., Cal., to Ventura, Cal.	1,400
From Santa Mina Dist., Cal.,	
From Northan Dist., Cal., to El Segundo. Cal	20,000
From Seneca and Wood Co.,	35,000
C., to Toledo, O	1,000
Yale, Okla.) to Sabine Pass,	
From Humble Tex to Sour	21,000
Lake, Tex.	6 inch

PRINCIPAL PIPELINES-Concluded

Pipeline	Mileage
Sun Pipeline Co	25
Sun Pipeline Co	23
Sun Pipeline Co	16
Sun Pipeline Co	4
Texas Co. (main lines)	742
Texas Co. (main lines)	160
Texas Co. (main lines)	253
Texas Co. (main lines)	96
Texas Co. (main lines)	60
Texas Co. (laterals)	222
Texas' Co	400
Texas Co	155
Texas Co	85
Texas Co. (two lines)	60
Texas Co	25
Texas Co	130
Texas Co	15
Texas Co	
Tidewater Pipe Co. (main lin	ne). 830
Tidewater Pipe Co. (laterals	5)1,929
Union Oil Co	65
Union Oil Co	43
Union Oil Co	51
Valley Pipeline Co	170
War Pipeline Co	
Wilburine Pipeline Co	125
Yarhola Pipeline Co	135
Yarhola Pipeline Co	400

	oarrels
From Spindle Top, Tex., to Sabine Pass, Tex From Sour Lake, Tex., to	
Sabine Pass, Tex	8 inch
From Sour Lake, Tex., to	
Spindle Top, Tex	8 inch
From Batson, Tex., to Sour	
Lake, Tex	8 inch
From Spindletop, Tex., to Sun	
Station, Tex.	6 inch
From Bartlesville, Okla., to	
Port Arthur, Tex	20,000
From Electra, Tex., to West	
Dallas, Tex.	17,000
From Vivian, La., to Port	
Arthur, Tex.	20,000
From Sour Lake, Tex., to Spindle Top, Tex., to Sour Lake, Tex. From Batson, Tex., to Sour Station, Tex., to Sour Station, Tex. From Bartlesville, Okla., to Port Arthur, Tex. From Electra, Tex., to West Dallas, Tex. From Vivian, La., to Port Arthur, Tex. From Vivian, La., to Port Garrison, Tex. From Healdton, Okla., to Sherman, Tex. From in Oklahoma and Texas to.	0 000
Garrison, Tex	9,600
From Healdton, Okla., to	10 000
Enom in Oklahoma and Horag	12,000
From in Oklanoma and Texas	
From Donnigon Tor to Dont	• • • • •
Arthur	6 inch
Arthur. From Logansport, Tex., to Port Port Arthur, Tex From Ranger, Tex., to Fort Worth, Tex. From Dallas, Tex., to Fort Worth, Tex.	o men
Port Arthur Toy	8 inch
From Banger Tex to Fort	6 men
Worth Tex., to Fort	8 inch
From Dallas' Tex to Fort	o men
Worth Tex	8 inch
Worth, Tex From Dayton, Tex., to Goose	o men
Creek.	8 inch
From Electra, Tex., to Fort	0 111011
Worth, Tex.	6 inch
Creek. From Electra, Tex., to Fort Worth, Tex. From Humble, Tex., to Hous-	
ton. Tex	6 inch
From Healdton, Okla., to	
Gates Station, Tex	8 inch
From Stoy, Ill., to Bayonne,	
N. J	11,000
From Humble, Tex., to Hous- ton, Tex	
and Ind	
and Ind From Orcutt, Cal., to Port San Luis, Cal Local lines in Ventura Coun- ty, Cal Local lines in Los Angeles,	
San Luis, Cal	
Local lines in Ventura Coun-	
ty, Cal.	
Local lines in Los Angeles,	
Orange County, Fields, Cal.	
From Coalinga Dist., Cal., to	
San Francisco Bay	25,000
From Cushing Field, Okla., to	0 to ab
From Shannopin, Pa., to War-	8 inch
ron Pa	5,000
From Healdton Okla to	5,000
From Healdton, Okla., to Cushing, Okla. From Cushing, Okla., to St.	9.000
From Cushing, Okla to St	0,000
Divon Ill	00.000

Capacity,

Losses in the Storage of Crude Petroleum

The principal losses in the storage of crude petroleum are due to evaporation, to fire and to seepage.

Oils having the greatest loss are the crude oils containing the most gasoline, since they are the most volatile, most readily form explosive and inflammable mixtures and due to their low viscosity most readily flow through walls of loose texture.

The loss from evaporation is greater the larger the amount of gasoline. The loss also depends upon the temperatures of storage, upon the amount of surface exposed to the atmospheric circulation. If the tank or container is perfectly tight, then there will be no loss by evaporation.

There are three general types of storage now in use in the Mid-Continent fields, the earthen reservoir, the steel tank with wooden roof and the steel tank with a steel gas tight roof. The 55,000 and 35,000 barrel steel tanks are the usual sizes. Al-

The 55,000 and 35,000 barrel steel tanks are the usual sizes. Altogether there are more than 3,000 of these large steel tanks in use in the Mid-Continent field.

The earthen storage is extremely wasteful from both seepage and evaporation. Petroleum standing in this type of reservoir has been known to shrink 40% in volume in two or three weeks. The shrinkage in value is of course much greater as the portion lost by evaporation is the best of the gasoline.

evaporation is the best of the gasoline. The following losses by evaporation took place in steel tanks with no seepage, with wooden roof covered with paper and tarred and apparently tight. The oil was of 40° Be' gravity and the tanks were of a diameter of $114\frac{1}{2}$ feet.

Capacity	Loss in Gauge	Actual Loss	Period Per	Cent Loss
55,000 bbls.	1 ft. 134 in.	2101 bbls.	5 mos.	4.2
55,000 bbls.	1 ft. 2% in.	2235 bbls.	4½ mos.	4.6
55,000 bbls.	11½ in.	1700 bbls.	3½ mos.	3.4
55,000 bbls.	1 ft. ½ in.	1910 bbls.	3¼ mos.	3.8

The above figures indicate that there might be a loss of 1% per month of storage in wood roof steel tanks and this might amount to as much as 6,000 barrels per year per tank. It has been claimed that oil stored in white tanks is subjected to

It has been claimed that oil stored in white tanks is subjected to 1 to $1\frac{1}{2}\%$ less evaporation than in red tanks and $2\frac{1}{2}\%$ less evaporation than in black tanks.

Various types of insulation have been used with success.

A typical storage temperature for the Mid-Continent field for oil stored above ground would be 80°F. A typical temperature of the ground for a submerged tank would be 60°F which would more nearly approach the storage temperature of the air for the whole year. If tanks could be successfully and cheaply built in the ground,

If tanks could be successfully and cheaply built in the ground, they would have the advantage of almost perfect insulation from outheat, and the oil would be stored at practically the temperature at which it comes from the ground. For this submerged type of tank, concrete construction would be proper if capable of perfect construction. It should be monolithic, well reinforced and lined with a coating impervious to water and gasoline. Next in quantity after the evaporation losses in the storage of crude oil is the loss due to fire. Petroleum fires destroyed 12,850,000 barrels of oil in the United States in 1918. From Jan. 1, 1908, to Jan. 1, 1918, approximately 12,850,000 barrels of oil and 5,024,506,000 cubic feet of gas were destroyed by fire in the United States entailing a total estimated property loss of \$25,254,000. During this period 503 fires were reported. Of these fires 310 were caused by lightning and 193 by other causes. The losses from the fires caused by lightning were estimated to be \$11,148,000 and from those due to other causes, \$14,106,200. Directly and indirectly the fires resulted in the deaths of nearly 150 persons and were responsible for almost as many more being permanently disabled.

Loss from fire in the oil field storage in the year 1916 amounted to about \$4,000,000.

The causes of fires are electrical discharges or open flames in the presence of an inflammable or explosive mixture of gasoline and air. The amount of gasoline vapor in air necessary for an explosive mixture is within the limits of $1\frac{1}{2}\%$ and 5% by weight. Less than the lower limit or more than the upper limit will not inflame. In an open tank if the amount at the surface of the oil exceeds $1\frac{1}{2}\%$ there is at some point an explosive mixture and an igniting temperature of 900°F. or over will cause it to take fire. In a perfectly tight tank with gasoline vapor in excess of the upper limit for an explosive mixture, there will be no fire unless the roof of the tank is open at some point.

The ingress of a flame through an opening may be prevented in the same way that the flame in the Davy miner's lamp is prevented from passing outward. This operates by having some metal screen or other material cool the flame and prevent it being propagated into the tank. This will not prevent ignition from an electrostatic discharge in the vapor space of the tank.

Methods for prevention of fires of oil in storage are as follows: 1st. Means of preventing the passage of the spark in a portion of the unfilled face of the tank.

2d. The maintenance of a mixture in the unfilled portion of the tank which is not an explosive mixture.

3d. A tank so placed and constructed that the cooling effect of the walls will tend to smother the flames and the ingress of air will be so arranged that the fire is not readily fed.

4th. A means for quickly eradicating the fire after it is ignited. Several more or less successful methods for extinction of oil tank fires have been in use. The best involves the use of mixtures of sodium bicarbonate and sulphuric acid which produce sufficient carbon dioxide to smother the flame. If some sort of saponifying agent is used the carbon dioxide will make a froth which will float on the surface of the oil and is very effective in extinguishing the flame.

The application of steam is very effective but in the storage of a very large amount of oil the steam is not always available when needed and at the point where needed.

For small oil fires dust or other finely divided mineral matter is effective in extinguishing the fire.

Fuel Oil Storage Tanks Regulations Drafted by Fire Protection Association

The Committee on Inflammable Liquids of the National Fire Protection Association has submitted the following tentative regulations covering the construction of concrete tanks for fuel oil storage.

tions covering the construction of concrete tanks for fuel oil storage. Setting of Tanks.—(a) Tanks, if underground, shall be buried so that the top of the tank will be not less than three feet below the level of the surface of the ground and below the level of any piping to which the tanks may be connected.

(b) Tanks shall be set on a firm foundation.

(c) All tanks shall be provided with a concrete or other noncombustible roof.

Material and Construction of Tanks.—(a) Reinforcement.—Sufficient steel reinforcement shall be used to resist the oil pressure, and the horizontal and vertical reinforcement shall be proportioned properly and located to reduce the shrinkage cracks, so that they will be too minute to permit leakage. The fiber stress in the steel shall not exceed 10,000 pounds per square inch. (Note. A fiber stress of 10,000 pounds per sq. in. should prevent shrinkage cracks although a number of tanks have been designed with a fiber stress of 6,000 to 8,000 pounds.)

(b) Concrete.-The concrete for floor and walls shall be at least 8 inches thick, mixed in the proportion of 1:2:3 or better 1:11/2:3 and having the coarse aggregate of clean, dense, crushed rock or gravel ranging in size from one inch down. The concrete shall be thoroughly mixed, carefully placed and worked around the reinforce-ment. The forms should not be held together by wire as is frequently done in building construction because leakage is likely to take place along the wire. The concrete shall preferably be poured in a continuous operation so as to form a monolithic construction. (Note. Where this cannot be done, the bottom shall be poured without joints and the walls as a second continuous operation. One method of making a tight joint between the bottom of the tank and the walls is by means of a strip of galvanized iron six inches wide with joints riveted and soldered, so as to form a continuous band. This strip should be vertically embedded three inches in the floor slab and on the center line of the wall. The floor slab under the walls should be thoroughly cleaned, and before pouring the walls a mixture of 1:1 mortar should be placed in the bottom of the forms and around the galvanized strip to make a tight joint.)

(c) Finish.—As soon as the wall and sides have been poured the floor shall be floated and troweled smooth. The wall forms shall be removed as soon as the concrete has hardened sufficiently to be selfsustaining and all projections and irregularities shall be removed from the surface and all cavities filled with a 1:1 mortar thoroughly rubbed in and troweled smooth. No plastering shall be applied.

(d) Aging.—The concrete shall be allowed to harden at least 30 days and longer if possible. (Note.—To assist in the setting of the concrete before it becomes oil soaked it is advantageous to use several priming coats of a 1:4 solution of 40° Baume' sodium silicate, followed by a finish coat of a 1:2 solution. This forms a glazed surface on the concrete, which although it is not permanent, gives the concrete an opportunity to harden until the protection from the silicate of soda is not longer necessary.)

Location of Pipe Connections.—All pipe connections to the tank shall be made through the top.

Venting of Tanks.—(a) Tanks shall be provided with a permanently open vent, or with a combined fill and vent fitting so arranged that the fill pipe cannot be opened without opening the vent pipe.

(b) Vent openings shall be screened (30x30 brass mesh or equivalent) and shall provide sufficient area for allowing proper flow of liquid during the filling operation. Permanently open vent pipes shall be provided with weatherproof hoods and terminate at a point at least twelve feet above the top of the fill pipe and never within less than three feet, measured horizontally and vertically, from any window or other building opening. Where a battery of tanks is installed vent pipes may be run into a main header. Individual vent pipes should, however, be screened between tank and header and connection to the header should be not less than one foot above the level of the top of the highest reservoir from which the tanks may be filled.

(c) Fill pipes shall be screened and when installed in the vicinity of a building, shall not be located within five feet of any door or other opening and shall terminate in a metal box or casting provided with means for locking.

Rules Governing the Shipment of Oil Samples by Express

Oils having a flash point of 20°F or below must not be shipped in quantities greater than one gallon.

This includes benzine, benzol, casinghead gasoline, casinghead naphtha, coal tar light oils, coal tar naphthas, distillates, petroleum ether, gas drips, gasoline, liquefied petroleum gas, naphthas, naphtha distillates, gas oil, pentane and toluol.

Not more than one gallon shall be in one outside container and the package containing the fluid must not be entirely filled.

The vacant space must be not less than 2% of the contents. If in tightly closed metal cans, the package must be packed in wooden boxes. The package shall be labeled with a red label.

Crude oils, crude petroleums or petroleum naphthas or liquids having a flash point above 20°F and below 80°F may be shipped in quantities less than six gallons in one package. The package, if a metal can, must be covered with wood or packed in wooden boxes.

Gasoline or naphtha with a flash point of 20°F or lower when shipped in glass must be in capacity of one pint or less and cushioned in fiber board or corrugated straw board containers and with not more than eight quarts in one package.

Lubricating oils, motor oils, coal oil, fuel oil, illuminating oil, kerosene and other petroleum oils with a flash point higher than 80°F are not subject to special rules governing the transportation of dangerous articles by express and do not require special labels. The red label is required on all inflammable liquids, including light crude oils and light distillates having a flash point below 80°F.

Ownership of Tank Cars

TANK CARS OWNED BY RAILROADS.

	Tank
Name and Location.	Cars.
Colorado & Southern	. 14
Delaware River & Union R. R	
Denver & Rio Grande	
East Jersey R. R.	. 120
El Paso & Western	
Kansas City Southern Ry. Co	193
Los Angeles & Salt Lake R. R. Co	214
Midland Valley R. R. Co	
Missouri, Kansas & Texas Ry	
Morenci Southern Ry. Co	. 2
New Orleans, Texas & Mexico R. R	. 75
Northwestern Pacific R. R. Co	. 34
Oregon-Washington R. R. & Nav. Co	44
Pacific Electric Ry. Co	. 29
Pennsylvania R. R. Co	
Philadelphia & Reading Ry. Co	. 20
St. Louis & San Francisco R. R. Co	629
St. Louis, Brownsville & Mexico Ry	. 59
St. Louis, Southwestern Ry. Co	. 29
San Antonio & Aransas Pass Ry. Co	. 81
Santa Fe Ry. Co	3.178
Santa Fe & Arizona Ry	4
Southern Pacific Ry	2,963
Texas & New Orleans R. R. Co	459
Trinity & Brazos Valley R. R.	25

TANK CARS OWNED BY OIL INDUSTRY.

9,813

Tonla

Name and Location.	Cars.
Akin Gasoline Co., Tulsa, Okla	3
Ajax Gasoline Co., Kansas City	4
American Oil Products Corp., Erie	3
American Oil Works, Titusville, Penn	57
American Refining Co., Tulsa	256
Anderson & Gustafson, Cushing, Okla	59
Asphaltum Oil & Refining Co., Los Angeles	3
Associated Oil Co., California	337
Atlantic Refining Co., Philadelphia	4
Atwood Refining Co., Oklahoma City,	23
Allied Renning Co., Okmulgee	63
Barkhausen Oll Co., Green Bay, Wis	1
Beaver Refining Co., Washington, Pa	13
J. B. Berry Sons Co., Oil City, Pa	105
Bigheart Petroleum Refining Co Bigheart Okla	25
F. W. Bird & Sons, E. Walnole Mass	3
blake Oli Co., Liberal, Kansas,	ĩ
Bliss Renning Co., Augusta, Kansas,	37
Boynton Gasoline Co., Tulsa	4
Brooks Oll Co. Cleveland Obio	$\hat{2}$
BUVNION REINING LO BOVNTON ()RIG	61
	200
F. A. BUSH CO., Palmer, Mass.	3
Butter County Renning Co., Bruin, Pa	85
Caddo Oil Refining Co., Shreveport, La.	120
Canneld Ull Renning Co., Coraonolis Pa	45
Canneld Tank Line, Cleveland O.	78
	55
	50
	47
Carbo Renning Co., Guinrie, Okia	30
Central Renning Co., Lawrenceville III	293
Champlin Oil & Refining Co., Enid, Okla	38

OWNERSHIP OF TANK CARS—Continued

	Tank
Name and Location.	Cars.
Chestnut & Smith, Tulsa, Okla	12
Cincinnati Oil Works, Cincinnati Clarendon Refining Co., Clarendon, Pa Cleveland Petroleum Refining Co., Cleveland, O	_1
Clarendon Refining Co., Clarendon, Pa	75
Cleveland Petroleum Refining Co., Cleveland, O	21
	13
Columbia Oli Co., New York.	39 24
Commonwealth Renning Co., Moran, Kansas	47
Conewango Reinfing Co., warren, Fa	200
Canadiani on Companies, Etd., Fetrona, Canada	500
Columbia Oil Co., New York. Commonwealth Refining Co., Moran, Kansas. Conewango Refining Co., Warren, Pa. Canadian Oil Companies, Ltd., Petrolia, Canada. Constantin Refining Co., Tulsa. Consumers Mutual Tank Line, Chicago.	88
Consumers Refining Co., Cushing, Okla Continental Oil Co., Denver	379
Continental Oil Co., Denver Continental Refining Co., Oil City, Pa Continental Refining Co., Bristow, Okla Cosden & Co. Tulsa.	. 8
Continental Refining Co., Oil City, Pa	50
Continental Refining Co., Bristow, Okla	50
Cosden & Co., Tulsa	2,163
Cosden & Co., Tulsa. Craig Oil Co., Toledo, Ohio. Crescent Refining Co., Newkirk, Okla. Crew Levick Co., Philadelphia.	161
Crescent Refining Co., Newkirk, Okla	80
Crew Levick Co., Philadelphia.	215
Crywn Gasoline & Oli Co., Pittsburgh, Pa	35
Crystal Wille Renning Co., Allen, Okla	35
Dallas Oil & Refining Co Dallas Teras	20
W H Daugherty & Son Petrolia Pa	10
Crew Levick Co., Philadelphia. Crown Gasoline & Oil Co., Pittsburgh, Pa. Crystal White Refining Co., Allen, Okla. Crystal Oil Works, Rouseville, Pa. Dallas Oil & Refining Co., Dallas, Texas. W. H. Daugherty & Son, Petrolia, Pa. El Dorado Refining Co., El Dorado, Kansas. Economy Oil & Refining Co. Blackwell Okla	136
Economy Oil & Refining Co., Blackwell, Okla. Elk Refining Co., Charleston, W. Va. Emery Mfg. Co., Bradford, Pa.	68
Elk Refining Co., Charleston, W. Va	48
Emery Mfg. Co., Bradford, Pa	90
Emlenton Refining Co., Emlenton, Pa. Empire Refineries, Tulsa. Empire Oil Works, Oil City, Pa. Ensign Oil Co., Norristown, Pa.	74
Empire Refineries, Tulsa.	2,100
Empire Oil Works, Oil City, Pa	980
Ensign Oil Co., Norristown, Pa	4
Evans-Thwing.	250 25
Evans-Thwing. Foco Oil Co., Franklin, Pa. D. W. Frauchot Co., Tulsa, Okla.	23 12
D. W. Frauchot Co., 1018a, Okla	10
D. W. Frauchot Co., Tulsa, Okla Franklin Quality Refining Co., Franklin, Pa Freeport-Mex. Fuel Oil Corp., New Orleans, La	350
	97
General Refining Co., Tulsa	70
General Refining Co., Tulsa	265
Golden Rule Refinery, Wichita, Kansas	30
Great American Refining Co., Jennings, Okla	116
Great Western Oil Co., Cleveland	21
Great Western Oil Refining Co., Erie, Kansas	. 86
Guil Refining Co., Pittsburgh, Pa.	1,411
Great Western Oil Refining Co., Erie, Kansas. Gulf Refining Co., Pittsburgh, Pa. Gasoline Corporation, New York. General Petroleum Co., Los Angeles, Calif.	59 10
General Petroleum Co., Los Angeles, Calif. Home Oil Refining Co., Yale, Okla. High Grade Petroleum Products Co., St. Mary's, W. Va. High Grade Petroleum Products Co., St. Mary's, W. Va. Humble Oil & Refining Co. (Dixie O. & F. Co.), San Antonio, Texas Humboldt Refining Co., Humboldt, Kansas. Hutchinson Refining Co., Hutchinson, Kansas. Illinois Oil Co., Cushing, Okla. Illinois Refining Co., Rock Island, Ill. Imperial Refining Co., Ardmore. Okla	195
Great Lakes Oil & Refining Co. Wallaceburg Can	12
Hillman Refining Co. Cushing Okla	49
High Grade Petroleum Products Co. St. Mary's, W. Va.	50
Humble Oll & Refining Co. (Dixie O. & F. Co.). San Antonio. Texas	33
Humboldt Refining Co., Humboldt, Kansas	3
Hutchinson Refining Co., Hutchinson, Kansas	35
Illinois Oil Co., Cushing, Okla	75
Illinois Refining Co., Rock Island, Ill	61
Imperial Refining Co., Ardmore, Okla	26
Imperial Refining Co., Ardinore, Okla Imperial Oil Co., of Canada Independent Refining Co., Oil City, Pa Indian Refining Co., St. Louis Indian Refining Co., Lawrenceville, Ill. Inland Refining Co., Tulsa International Oil Works, Ltd., St. Louis. International Refining Co., Tulsa International Refining Co., Tulsa	668 82
Indiahoma Befining Co., On City, Fa	600
Indian Refining Co. Lawrenceville Ill	1.032
Inland Refining Co., Tulsa.	1,052
International Oil Works, Ltd., St. Louis	3
International Refining Co., Tulsa	418
	15
Interstate Oil Co., Minneapolis, Minn. Island Petroleum Co., Pittsburgh Kansas City Oil Co., Kansas City, Kansas.	1
Island Petroleum Co., Pittsburgh	70
Kansas City Oil Co., Kansas City, Kansas	5

OWNERSHIP OF TANK CARS—Continued

	Tank
Name and Location.	Cars.
Kansas Oil Refining Co., Coffeyville, Kansas	94 181
Kansas City Refining Co., Kansas City, Kansas. Kansas Co-Operative Refining Co., Chanute, Kansas. Kendall Refining Co., Bradford, Pa. A. Knabb & Co., Marcus Hook, Pa. Lake Park Refining Co., Okmulgee, Okla	193
Kendall Refining Co., Bradford, Pa	28
A. Knabb & Co., Marcus Hook, Pa.	$20^{1}{5}$
Lawton Refining Co., Lawton, Okla	32
Lawton Refining Co., Lawton, Ökla. Leader Oil Co., Casey, Ill Lesh-National Refining Co., Arkansas City, Kansas. Liquified Petroleum Gas Co. Tulsa Louisiana Oil Refining Co., Shreveport, La. Magnolia Petroleum Co., Dallas, Texas. Manufacturers Paraffine Co., Chester, Pa. Marland Refining Co., Ponca City, Okla. Marshall Oil Co., Marshalltown, Ia Mexican Petroleum Co., Ltd., New York. Mid-Continent Oil Refining Co., East St. Louis, Ill. Mid-Continent Gasoline Co. Tulsa Midland Refining Co., Eldorado, Kansas.	13
Lesh-National Refining Co., Arkansas City, Kansas	45 8
Louisiana Oil Refining Co. Shrevenort La.	60
Magnolia Petroleum Co., Dallas, Texas	590
Manufacturers Paraffine Co., Chester, Pa	320 ¹
Marshall Oil Co. Marshalltown Ia	520
Mexican Petroleum Co., Ltd., New York	170
Mid-Co Gasoline Co., Tulsa	$166 \\ 14$
Mid-Continent Oll Renning Co., East St. Louis, 111	166
Midland Refining Co., Eldorado, Kansas	148
Midland Refining Co., Eldorado, Kansas. Midwest Refining Co., Denver	22
Miller S Oil Refining Works, Allegheny, Pa	44 59
Milliken Refining Co., St. Louis	70
Motor Fuel Co., Sapulpa, Okla	/ 24
Muskogee Refining Co., Muskogee, Okla	$150 \\ 82$
Mutual Refining Co. Ltd. Warren, Pa.	39
National Oil Co., New York	24
New Haven Gas Light Co., New Haven, Conn	5
Oconee Oil Befining Co. Athens Co.	415 10
Ohio Valley Refining Co., St. Mary's, W. Va	50
Oconee Oil Refining Co., Athens, Ga. Ohio Valley Refining Co., St. Mary's, W. Va. Oil Products Corp., New York. O. K. Refining Co., Niotaze, Kansas.	20
Oklahoma Petroleum & Gasoline Co. Tulsa	161 41
O. K. Renning Co., Niotaze, Kansas. Oklahoma Petroleum & Gasoline Co., Tulsa. Oklahoma Refining Co., Oklahoma City. Okmulgee Products & Refining Co., Okmulgee, Okla. Oil State Refining Co. Enid, Okla. Oklahoma Products & Refining Co., Tulsa. Oneta Refining Co., Oneta, Okla. Oriental Oil Co., Dallas, Texas. Ozark Refining Co., Fort Smith, Ark. Ohio Citles Gas Co.	93
Okmulgee Products & Refining Co., Okmulgee, Okla	20
Oklahoma Products & Refining Co. Tulco	30 22
Oneta Refining Co., Oneta, Okla.	31
Oriental Oil Co., Dallas, Texas	39
Ozark Refining Co., Fort Smith, Ark	13
Ohio Cities Gas Co National Refining Co., Cleveland	900 1,004
Pelican Oil Refining Co., New Orleans, La.	1,004
Pelican Oil Refining Co., New Orleans, La Pennsylvania Refining Co., Oil City, Pa Pennsylvania Refining Co., Karns City, Pa Pan American Bodring Co.	6
Pennsylvania Refining Co., Karns City, Pa. Pan-American Refining Co., Tulsa Panhandle Refining Co., Wichita Falls, Texas Paragon Refining Co., Toledo, Ohio Pawnee Refining Co., Oklahoma Pennsylvania & Delaware Oil Co., New York Pennsylvania & Delaware Oil Refining Co. Eldred Pa	260
Panhandle Refining Co., Wichita Falls, Texas	35
Paragon Refining Co., Toledo, Ohio	173
Penn-American Refining Co. Oil City Po	48
Pennsylvania & Delaware Oil Co., New York	19
Petroleum Products Co., Pittsburgh	12
Pierce-Fordyce Assn. Dallas. Texas.	164 403
Pierce Oil Corp., St. Louis	643
Pinal Dome Refining Co., Santa Maria, Calif	1 100
Ponca Lub, Oil Co., Ponca City, Okla.	$ 100 \\ 30 $
Ponca Refining Co., Ponca City, Okla.	140
Producers Refining Co., Oklahoma City	270
Frank Prox Co., Terre Haute, Ind	136
Petroleum Products Co., Pittsburgh Phoenix Refining Co., Tulsa Pierce-Fordyce Assn., Dallas, Texas Pierce Oil Corp., St. Louis. Pinal Dome Refining Co., Santa Maria, Calif. Pittsburgh Oil Refining Co., Pittsburgh. Ponca Lub. Oil Co., Ponca City, Okla. Ponca Refining Co., Ponca City, Okla. Producers Refining Co., Oklahoma City. Prod. & Ref. Co., Blackwell, Okla. Frank Prox Co., Terre Haute, Ind. Prudential Oil Corp., Baltimore, Md. Puente Oil Co., Los Angeles.	250
Puente Oil Co., Los Angeles	. 2

OWNERSHIP OF TANK CARS—Concluded

	Tank
Name and Location.	Cars.
Pure Oil Co., Minneapolis, Minn	74
Pure Oil Co., Minneapolis, Minn Railroad Men's Refinery, Eldorado, Kansas	3
Record Oil Refining Co., New Orleans	3 35
Red "C" Oil Mfg. Co., Highland Town, Md	92
Red "C" Oil Mfg. Co., Highland Town, Md	3
Richfield ()il (o Los Angeles	10
Riverside Western Oil Co., Tulsa	225
Royana Petroleum Corp. Tulsa	400
Rohinson Oil Refining Co. Rohinson Ill	. 9
Rosedale Refining Co., Rosedale, Kansas. Rucker Bros., Everett, Washington. Sapulpa Refining Co., Sapulpa, Okla. Sarco Petroleum Products Co., Independence, Kansas	90
Rucker Bros. Everett. Washington	2
Sapulpa Refining Co., Sapulpa, Okla	441
Sarco Petroleum Products Co., Independence, Kansas	183
Seneca Oil Works, Warren, Pa	67
Sinclair Refining Co., Chicágo	3,700
Levi Smith, Ltd., Clarendon, Pa	18
Levi Smith, Ltd., Clarendon, Pa Shell Co. of California, San Francisco	84
Southern Oil Corp., Tulsa	250
Standard Oil Ca (IInian Manla)	21.600
Stannard, C. A., Emporia, Kansas	14
Starlight Refining Co., Karns City, Pa	5
Stannard Oli Co (Union Tank). Stannard, C. A., Emporia, Kansas. Starlight Refining Co., Karns City, Pa. Sterling Oli & Refining Co., Wichita, Kansas. St Louis Oli & Refining Co. El Dorado Kansas.	36
St. Louis Oil & Refining Co., El Dorádo, Kansas	25
Southern Alberta Refineries, Ltd., Okotoks, Alta	1
Southern Refining Co Los Angeles	26
A. Speare's Sons Co., Boston	6
Superior Oll Works, Ltd., Warren, Pa.	25
Superior Refining Co., Covington, Okla	18
Terminal Oil Refining Co., Healdton, Okla	18
Superior Refining Co., Covington, Okla. Terminal Oil Refining Co., Healdton, Okla. The Texas Co., Houston, Texas	3,435
Tiona Refining Co. Clarendon, Pa	5
Titusville Oil Works, Titusville, Pa	50
Turner Oil Co., Los Ángeles Uncle Sam Oil Co., Cherryvale, Kansas	9
Uncle Sam Oil Co., Cherryvale, Kansas,	51
Union Oil Co. of California, Los Angeles	115
Union Petroleum Co., Philadelphia	105
Union Refining Co., East St. Louis, Ill	3
United O. & R. Co., Beaumont, Texas	5
United Oil Co., Denver. United Refining Co., Warren, Pa. U. S. Asphalt Co., E. Brooklyn, Md. Upson's Oil & Soap Co., Parkersburg, W. Va.	19
United Refining Co., Warren, Pa	40
U. S. Asphalt Co., E. Brooklyn, Md	300
Upson's Oil & Soap Co., Parkersburg, W. Va	7
Valley Refining Co., Tulsa	23
Valvoline Oil Works. Ltd., East Butler, Pa	89
Valley Refining Co., Tulsa Valvoline Oil Works, Ltd., East Butler, Pa Victor Refining Co., Yale, Okla	10
Vulcan Oil Refining Co., Cleveland, O	48
Wabash Refining Co., Robinson, Ill.	88
Wadhams Oil Co., Milwaukee	5
Warren Oil Co., Warren, Pa	50
Wannan Defining Co Wannan De	106
Waverly Oil Co., Pittsburgh	50
Webster Oil & Gas Co., Yale, Okla	5
Webster Refining Co., Humboldt. Kansas	4
Western Refining Co., Wichita, Kansas	22
West Virginia Oil Co., Parkersburg, W. Va	1
Waverly Oil Co., Witheling Co., Warren, Pa. Webster Oil & Gas Co., Yale, Okla Webster Refining Co., Humboldt, Kansas. Wester Refining Co., Wichita, Kansas West Virginia Oil Co., Parkersburg, W. Va. Wichita Independent Oil & Refg. Co. (Sterling), Wichita, Kansas	115
white our works, indu, warren, i d	61
Wilholt Refining Co., Springfield, Mo	51
Wilshire Oil Co	50
White Eagle Petroleum Co., Augusta, Kansas	260
Wright Pro. & Refining Co., Cherryvale, Kansas	11
Yaryan Rosin & Turpentine Co., Brunswick, Ga	5
Car manufacturers	7,969
	Contraction Providence

Total.

... 65,500

95

RULES GOVERNING THE LOCATION OF LOADING RACKS AND UNLOADING POINTS FOR CASINGHEAD GASOLINE, RE-FINERY GASOLINE, NAPHTHA OR ANY INFLAM-MABLE LIQUID WITH FLASH POINT BELOW 30 DEG. F. (Temporarily Suspended.)

The location of loading racks and unloading points for volatile inflammable liquids is considered of great importance to the safety of railroad property, and there is at present lack of uniformity in the enforcement of proper safeguards for the protection of life and property. The following rules shall govern all carriers under Federal control with respect to the location of loading racks or unloading points hereafter installed. As to present locations these rules shall be observed when practicable, and for locations not in accordance therewith carrier through its proper officer shall submit report with recommendation covering each such location to the director of the Division of Operation for instruction. Whenever practicable, throug efforts will be made to secure their removal to a safe distance, or such other remedy as the facts may justify will be applied.

Loading

1. Loading racks for refinery gasoline, benzine, naphtha or iny liquid with flash point below 30°F, must not be located nearer the eighty feet to a track over which trains or engines are operat. (This does not apply to the track serving the loading rack.) Loz racks for casinghead gasoline or casinghead blends must be loc not less than 160 feet distance from such tracks, whenever practicat and in no case should they be located at a less distance than 100 fe These rules apply to casinghead condensates or blends whether made by the compression or absorption process.

Unloading

2. a. The unloading of tank cars of casinghead gasoline, benzine, naphtha and similar petroleum products on railroad sidings must not be permitted, except where facilities exist for piping the contents from the tank cars to permanent storage tanks.

b. The part of any siding on which tank cars of gasoline, benzine, naphtha, or any liquid with flash point below 30°F are to be unloaded, must be located not less than 80 feet from a track over which trains or engines are operated. (This does not apply to the track serving the unloading point.) Where casinghead gasoline is to be unloaded the unloading point must be not less than 160 feet from such track, whenever practicable, but in no case should the distance be less than 100 feet.

c. If the unloading is done on a private siding into tank wagons, barrels or drums (not permanently located storage tanks) the distance at which this operation is permitted must not be less than 160 feet.

d. If tank cars of refinery gasoline, benzine, naphtha, or any liquid with flash point below 30° F are loaded or unloaded at a place within 80 feet from a track over which trains or engines are operated, such tank cars must be provided with a dome cover equipped with

a vent line to liberate any escaping vapors. This vent line must be carried to a point at least 80 feet distant from such track. For casinghead gasoline 160 feet will be required whenever practicable, but in no case shall the distance be less than 100 feet from the center of track over which trains or engines are operated. The end of the vent line must be covered with a proper screen of not less than 20x20 mesh.

Storage

3. a. Gasoline, benzine, naphtha, or any liquid with flash point below 30°F, when stored in properly constructed tanks, is comparatively safe. The following regulations will apply for the construction and location of such storage tanks:

b. These regulations apply only to above ground tanks. Underground tanks should be considered separately as occasion may arise. All storage tanks will be considered above ground unless they are buried so that the top of the tank is covered with at least three feet of earth.

c. All tanks should be set upon a firm foundation and be electrically grounded.

d. Each tank over 1,000 gallons in capacity should have all hanholes, hand holes, vent openings, and other openings which may ontain inflammable vapor, provided with 20x20 mesh brass wire creen or its equivalent, so attached as to completely cover the openngs and be protected against clogging; these screens may be made removable but should be kept, normally, firmly attached. Such a tank should also be vented or provided with a suitable safety valve set to

erate at not more than five pounds per square inch for both inior pressure and vacuum; manhole covers kept closed by their ght only will be considered satisfactory.

e. Tanks used with a pressure system may have a safety valve t at not more than one-half the pressure to which the tank was originally tested.

f. Tanks containing over 500 gallons and not exceeding 48,000 gallons of gasoline, benzine, naphtha, or any liquid with flash point below 30°F, should be located not less than 80 feet from a track over which trains or engines are operated.

g. For capacities exceeding 48,000 gallons the following distances shall govern:

	Distance	Distance
	from Railroad	from
	Tracks or	Other
Capacity of Tanks.	Property,	Tanks,
(In gallons.)	Feet.	Feet.
48,001 to 75.000	. 85	3
75.001 to 100.000	. 100	15
100.001 to 150.000	. 150	25
150,001 to 250,000	. 250	35
250,001 to 500,000		50
500.001 to 1,000,000	. 350	75
Unlimited		200

h. The above distances should be doubled for tanks containing casinghead gasoline or casinghead blends.

i. Where practicable, tanks should be located on ground sloping away from railroad property, if this is impracticable, then the tanks should be surrounded by dikes of earth, or concrete, or other suitable material, of sufficient capacity to hold all the contents of the tanks, or of such nature and location that in case of breakage of the tanks the oil will be diverted to point such that railway property and passing trains will not be endangered.

General

4. a. In measuring distances from any railroad track the center line of the track should be considered as the starting point.

b. During the time that the tank car is connected by loading or unloading connections, there must be signs placed on the track or car so as to give necessary warning. Such signs must be at least 12x15 inches in size and bear the words "Stop—Tank Car Connected," the word "STOP" being in letters at least 4 inches high and the other words in letters at least two inches high. The printing must be in white letters on a blue background.

c. In laying pipelines on railroad property for the loading or unloading of tank cars, they should be laid at a depth of at least three feet, and at points where such pipelines pass under tracks they should be laid at least four feet below the bottom of the ties.

d. All connections between tank cars and storage tanks must be in good condition, and must not permit any leakage. They must be frequently examined and replaced when they have become worn in order to insure at all times absolutely tight connections. Rubber, leather, or fabric hose must not be used. Tank cars must not be left connected to pipelines except when loading or unloading is going on and while a competent man is present and in charge.

e. Goose-necks, when used for loading and unloading tank cars, must be so constructed that when not in use they will automatically assume a stable position that will provide a horizontal clearance of not less than eight (8) feet from center line of track and be locked in that position. Where this method of unloading is used the rack supporting standpipe and goose-neck shall be of non-combustible material.

f. The ends of pipelines for loading or unloading tank cars from their bottom opening should be placed in shallow pits with brick or concrete walls not closer than 8 feet from center line of track. These pits should be ventilated and be protected by substantial onepiece covers, level with the surface of the ground, which must be kept locked in place when the pits are not in use. These pits should not be drained into a sewer or running stream. g. The loading or unloading of tank cars should not be per-

g. The loading or unloading of tank cars should not be permitted except during daylight, when artificial light is not required. The presence of nearby switch lights. lanterns, or other exposed lights or fires during the process of loading or unloading is prohibited.

Multiply or divide, as required, the weight-measure values by the specific gravity of the petrol-Specific gravity of average crude oil=0.850; fuel oil=0.900; gasoline=0.750; kerosene=0.820; oil=0.850. Specific gravity of average crude oil=0.850; fuel oil=0.900; gasoline=0.750; kerosene=0.820; oil=0.850. Petroleum XRIo Specific gravity of average crude oil=0.850; fuel oil=0.850; fuel oil=0.850; fuel oil=0.850; fuel oil=0.850; fuel oil=0.900; gasoline=0.750; kerosene=0.820; oil=0.850, fuel oil=0.850; fuel oil=0.900; gasoline=0.750; kerosene=0.820; oil=0.850, fuel oil=0.850; gasoline=0.750; kerosene=0.820; oil=0.850; fuel oil=0.850; fuel oil=0.850; gasoline=0.750; kerosene=0.820; fuel oil=0.850; fuel oil=0.850; fuel oil=0.850; gasoline=0.750; kerosene=0.820; fuel oil=0.850; gasoline=0.750; gasoline=0.750; kerosene=0.820; fuel oil=0.850; gasoline=0.750; gasoline=0.750; gasoline=0.750; gasoline=0.820; fuel oil=0.850; gasoline=0.820; fuel oil=0.900; gasoline=0.750; gasoline=0.820; fuel oil=0.850; gasoline=0.820; gasoline=0.820; gasoline=0.820; gasoline=0.800; gasoline=0.800; gasoline=0.800; gasoline=0.800; gasoline=0.800; gasoline=0.800; gasoline=0.800; gasoline=0.800	ast	used 1 Ires of	ly converting different measures of petroleum and water into each other.	Ine ronowing table is used in the calculation of capacities of reservoirs and tanks and in quick- inverting different measures of petroleum and water into each other.	r into eacl	h other.			
Inter, the weight-measure values by the specific gravity of the perage crude oil=0.850; fuel oil=0.900; gasoline=0.750; kerosene=0. n $(allon (allon $	URE	MEN	T OF WA	ATER AN	D PETRO	DLEUM A	T 60° F		
	ave	erage	crude oil=	=0.850; fu	el oil_0.9	oy une si 900; gaso	line=0.75	0; kerosei	le peur
	-	Cubie	U.S.	Imperial		Petroleum		Kilo-	Metric
7.48 6.23 28.317 0.1781 (2.57 28.29 0 .004329 .003005 .016387 1.300.10-4 .03600 .01637 1.000 .8228 3.785 .02830 8.388 3.782 1.201 1.000 .8228 3.785 .02830 10.01 4.541 1.201 1.000 4.545 .02830 10.01 4.541 1.201 1.000 1.000 2.303 3.782 .96034 42.00 34.98 150.3 1.000 330.2 158.65 .96034 42.00 34.98 150.3 1.000 .00286 1.000 .45360 42.00 34.88 150.3 1.000 330.2 138.65 .96034 42.01 .0264 .2305 1.000 360.2 1000 .45560 504.4 .2022 1001 .00286 1.000 .45560 .1000 504.4 .2023 1001 .006206 2.2055 <t< td=""><td>I</td><td>Inch</td><td>Gallon</td><td>Gallon</td><td>Liter</td><td>Barrel</td><td>Pound</td><td>gram</td><td>Ton</td></t<>	I	Inch	Gallon	Gallon	Liter	Barrel	Pound	gram	Ton
00 004:320 003005 016387 1.300.10-4 03000 01637 1.000 .83235 3.785 02381 8.338 3.782 1.000 .83235 3.785 02381 8.338 3.782 1.001 1.000 4.545 02830 10.01 4.541 1.201 1.000 4.545 02830 10.01 4.541 2.200 1.000 4.545 02830 10.01 4.541 42.00 34.98 159.3 1.000 350.2 158.65 42.00 34.98 159.3 1.000 350.2 158.65 42.00 34.98 159.3 1.000 350.2 158.65 1119 0990 .4539 092806 1.000 .45580 1119 2041 2202 1.001 .005296 2.005 1.000 204.4 2041 2001 6.296 2.005 1.000 .45580 204.4 2022	1728.		7.48	6.23	28,317	0.1781	62.37	28.20	62820.
1.000 .8238 3.755 .02381 8.338 3.782 1.201 1.000 4.545 .02360 10.01 4.541 1.201 1.000 4.545 .02360 10.01 4.541 2.202 2.200 1.000 2.203 2.903 .06004 42.00 34.86 150.3 1.000 360.2 188.85 .06004 42.00 34.86 150.3 1.000 360.2 188.85 .06004 41.1190 .00900 .4530 .002366 1.000 .45356 .002366 1.000 .45356 504.4 220.2 1001 6.366 2205 1000. 1.000 .45356 504.4 220.2 1001 6.306 2965 1.000 .100 .4538 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100 .100	1.(1.000	004329	.003005	.016387	1.303.10-4	.03009	.01637	1.637.10
	231.		1.000	.8328	3.785	.02381	8.338	3.782	.003782
2642 .2300 1.000 .00230 2.303 .060034 42.00 34.98 159.3 1.000 350.2 158.85 .11190 .0960 .4530 .002366 1.000 .4556 .11190 .0960 .4530 .002366 1.000 .4556 .1001 .2023 1.001 .002366 1.000 .4556 .2044 .2202 1.001 .002366 1.000 .4556 .264.4 .220.2 1001. 6.266 2205. 1000. 1. .4.331 3.607 10.40 0.1031 35.12 16.38 . 1.	277.4		1.201	1.000	4.545	.02859	10.01	4.541	.004541
42.00 34.98 150.3 1.000 370.2 158.85 .1190 .0080 .4589 .002856 1.000 .45850 .119 .0080 .4589 .002856 1.000 .45850 .2044 .2202 1.001 .6295 2.205 1.000 294.4 220.2 1001. 6.296 2.205 1000. 4.381 3.607 16.40 0.1061 38.12 16.88	61.03	03	.2642	.2200	1.000	.00629	2.203	.999034	666000*
.1190 .0090 .4539 .002566 1.000 .45359 i .2644 .2202 1.001 .0002566 2.2055 1.000 264.4 .220.2 1001. 6.206 2.2055 1000. 1. 264.4 220.2 1001. 6.206 2205. 1000. 1. 4.381 3.607 16.40 0.1081 36.12 16.38 .	9703.		42.00	34.98	159.3	1.000	350.2	158.85	.15885
.2644 .2202 1.001 .000206 2.205 1.000 264.4 220.2 1001. 6.206 2205. 1000. 1 4.831 3.607 16.40 0.1081 36.12 16.38 1	277.1	1	.1199	0660.	.4530	.002856	1.000	.45359	.0004536
220.2 1001. 6.296 2205. 1000. 1 11 3.607 16.40 0.1031 36.12 16.38 1	61.08	8	.2644	.2202	1.001	.006296	2.205	1.000	100*
3.607 16.40 0.1031 36.12 16.38	61080.		264.4	220.2	1001.	6.296	2205.	1000.	1.000
	1000.		4.331	3.607	16.40	0.1031	36.12	16.38	.01638

KANSAS CITY TESTING LABORATORY

Horizontal Cylindrical Tanks

C = Liquid contents in gallons
L = Length of tank in inches
d = Diameter of tank in inches
x = Depth of liquid contents in inches
C =
$$\frac{L}{231} \left(0.004363 \, d^2 \, \cos^{-1} \frac{d-2x}{d} - \frac{d-2x}{2} \sqrt{x(d-x)} \right)$$

Cos⁻¹ $\frac{d-2x}{d}$ means the value of the angular degrees whose cosine is
 $\frac{d-2x}{d}$
The cosine of an angle is the ratio in its right angled triangle, of the
side adjacent the angle to the hypothenuse of the triangle.
When L = 300 inches
d = 100 inches
x = 30 inches
 $\frac{d-2x}{d} = .4$
Cos⁻¹ .4 = 66.42° (From Trigonometric tables)
C = $\frac{300}{231} \left(0.004363 (10000) (66.42) - 20 \sqrt{2100} \right)$
= $\frac{300}{231} (2897 - 882.)$
= 2617 gallons.

Total capacity of horizontal cylindrical tank in gallons.

 $C = .0034 d^{2}L$

- d = diameter in inches. L = length in inches.
- c = capacity in U. S. Gallons.

Total capacity of horizontal cylindrical tanks in barrels.

- $C = 0.14 d^{2}L.$
- d = diameter in feet.
- L = Length in feet.
- $\mathbf{c} = \mathbf{capacity}$ in barrels.

Horizontal Cylindrical Tank Capacity Table

Diameter		Capacity	Capacity		Diameter
1%	=	.17%	1%	=	3.3%
2%	=	.48%	2%		5.2%
3%	=	.87%	3%	-	7.0%
4%	=	1.34%	4%	=	8.2%
5%	=	1.87%	5%	=	9.7%
6%	=	2.45%	6%	= .	11.0%
7%	_	3.08%	7%	=	12.2%
8%	=	3.75%	8%	=	13.4%
9%	=	4.46%	9%	=	14.5%
10%	=	5.20%	10%	Ξ	15.6%
11%	=	5.98%	10%	=	16.7%
12%		6.79%	12%		17.8%
12%	=	7.64%	12%	=	
	=		13%	=	18.8%
14%	=	8.51%	14%	=	19.8%
15%	=	9.41%	15%	=	20.8%
16%	=	10.33%	16%	=	21.7%
17%	=	11.27%	17%	=	22.6%
18%	=	12.24%	18%	=	23.6%
19%	=	13.23%	19%	=	24.5%
20%	=	14.24%	20%	=	25.4%
21%	=	15.27%	21%	=	26.3%
22%	=	16.31%	22%	=	27.2%
23%	=	17.37%	23%	=	21.8%
24%	=	18.45%	24%	=	29.0%
25%	=	19.55%	25%	=	29.8%
26%	=	20.66%	26%	=	30.6%
27%	=	21.78%	27%	=	31.5%
28%	=	22.92%	28%	=	32.4%
29%	=	24.07%	29%	=	33.2%
30%	=	25.23%	30%	=	34.0%
31%	=	26.40%	31%	=	34.8%
32%	=	27.58%	32%		35.7%
33%	=	28.78%	33%	=	36.5%
34%	_	29.98%	34%		37.3%
35%	=	31.19%	35%	-	38.1%
36%	=	32.41%	36%	=	38.9%
37%	=	33.63%	37%	=	39.7%
38%	=	34.87%	38%	-	40.5%
39%	=	36.11%	39%	=	41.3%
40%	=	37.35%	40%	=	42.1%
41%	=	38.60%	41%	=	42.9%
42%	=	39.86%	42%	=	43.7%
43%	=	41.12%	43%	-	44.5%
44%	=	42.38%	43%		45.3%
45%		43.64%	44%	= -	46.1%
45%	=	43.04%	45%	=	46.9%
40%	=	44.91%	40%	=	40.9%
41%	=	40.18%	41%		41.1%
	=		48% 49%	=	48.5%
49% 50%	=	48.73% 50.00%	49% 50%	=	49.2%
00%	-	00.00%	00%		50.070

Tank Car Outage Table

Showing Capacity of an 8,000-Gallon Tank Car at Different Levels

Wet Reading	Contents U. S. Gal.	Wet Reading	Contents U. S. Gal.	Wet Reading	Contents U. S. Gal.
Ft. In.		Ft. In.		Ft. In.	
1	20.4	3 4	4160.	6 7	8084.
2	63.7	3 5	4293.	6 8	8093.8
3	102.	3 6	4424.	6 9	8103.74
4	157.	3 7	4554.	6 10	8113.66
5	218.	3 8	4684.	6 11	8123.57
0	285.	3 9	4814.	7 0	8133.49
7	356.5	3 10	4945.	7 1	8143.4
0	434.6	3 11	5073.	7 2	8153.32
0	516.7	4 0	5201.	7 3	8163.23
10	602.	4 1	5330.	7 4	8173.14
11	692.2	4 2	5456.	7 5	8183.06
1 0	785.5	4 3	5582.	7 6	8192.97
1 1	881.3	4 4	5705.	7 7	8202.89
1 2	981.1	4 5	5829.	7 8	8213.05
1 3	1082.	4 6	5950.	7 9	8223.97
1 4	1187.	4 7	6071.	7 10	8234.88
1 5	1296.	4 8	6191,	7 11	8245.04
1 6	1456.	4 . 9	6308.	8 0	8254.96
1 7	1518.	4 10	6424.	8 1	8264.87
1 8	1630.	4 11	6536.	8 2	8274.79
1 9	1746.	5 0	6649.	8 3	8284.7
1 10	1863.	5 1	6758.	8 4	8294.62
1 11	1983.	$ 5 1 \\ 5 2 \\ 5 3 $	6867.	8 5	8304.53
2 0	2104.	5 3	6972.	8 6	8314.45
2 1	2225.	5 4	7073.	8 7	8324.36
2 9	2349.	5 5	7173.	8 8	8334.28
	2472.	5 6	7269.	8 9	8342.29
2 4	2598.	5 7	7362.	8 10	8346.98
2 5	2724.	5 8	7452.08	8 11	8349.48
2 6	2853.	5 9	7538.32	9 0	8351,98
2 7	2981.	5 10	7620.07	9 1	8354.48
2 8	3109.	5 11	7699.34	9 2	8356.98
2 9	3240.	6 0	7771.36	9 3	8359.49
2 10	3370.	6 1	7839.86	9 4	8361.49
$\begin{array}{c} 2 & 1 \\ 2 & 2 \\ 2 & 3 \\ 2 & 2 \\ 2 & 4 \\ 5 & 6 \\ 2 & 7 \\ 2 & 8 \\ 2 & 9 \\ 2 & 10 \\ 2 & 11 \\ \end{array}$	3500.	6 2	7902.96	9 5	8362.70
3 0	3630,	6 3	7960.76	9 6	8363.31
3 1	3761.	6 4	8002.86	9 7	8363.93
3 2	3894.	6 5	8051.24	9 8	8364.54
$ 3 2 \\ 3 3 $	4027.	6 6	8074.	0	0304.04
0 0	4047.	0 0	0014.		

GAUGING TABLE FOR EACH ONE-QUARTER INCH IN DEPTH FOR TANK AS DETAILED ON PETROLEUM IRON WORKS COMPANY DRAWING No. 2050-A

8050-Gallon	78-Inch	Diameter	Tank	With	Steam	Coils	for	Type
		"A" and	"A-1"	' Cars				

Depth	Bottom	Feet	Gallons	Feet	Gallons	Feet	Gallons	Feet	Gallons	Feet	Gallons	Feet	Gallons	Feet	Gallons
0		<u></u>	731	2	2037	3	3565	4	5142	5	6583	6	7695	7	8085
	 20 28 37 46 55	Steam Coils, Con.	754 777 801 825 849 875 898 823		2067 2097 2128 2159 2189 2220 2251 2282		3598 3631 3364 3697 3730 3765 3796 3830	14-1313411 14-1304 21	$\begin{array}{c} 5174\\ 5206\\ 5238\\ 5269\\ 5301\\ 5332\\ 5364\\ 5395\\ \end{array}$	100 000	6611 6638 6665 6692 6719 6746 6772 6798		7713 7731 7749 7766 7783 7800 7816 7832	19 4000 11 14 1000 01	8088 8091 8094 8097 8100 8103 8106 8109
-000*00 --000**	65 76 88 100 113 126 139 153	חשריאטיים בישריאנושיים	948 973 998 1024 1050 1076 1102 1128		$\begin{array}{c} 2313\\ 2344\\ 2375\\ 2406\\ 2437\\ 2468\\ 2499\\ 2531\\ \end{array}$		3863 3896 3929 3963 3997 4030 4063 4096		$\begin{array}{c} 5427\\ 5458\\ 5489\\ 5520\\ 5551\\ 5582\\ 5613\\ 5644\\ \end{array}$		3824 6850 6876 6901 6927 6952 6977 7002	Ha AST Nor-Aller CO AND Nor-Alle	7847 7862 7877 7891 7905 7918 7931 7943		8112 8115 8118 8120 8123 8126 8129 8132
	167 181 196 211 226 241 256 272	O ACKINAL CLASSIN-AL	$\begin{array}{c} 1154 \\ 1180 \\ 1207 \\ 1234 \\ 1261 \\ 1288 \\ 1315 \\ 1343 \end{array}$		2562 2594 2625 2657 2688 2720 2752 2784	0.410 Northan C1410 Northan	4130 4163 4196 4229 4262 4295 4328 4361	0 4000-40-014000-00	5675 5706 5737 5767 5799 5829 5859 5889	O POINT AL CIAONHAL	7027 7052 7077 7101 7125 7149 7173 7197		7954 7965 7976 7986 7995 8003 8010 8015	CAMBINAN CIABINAN	8135 8138 8141 8144 8147 8150 8153 8155
במוודהו ווהמוויניו 10 המוודהו 6 המוודהו 20 המוודהו 14 מוודהו	$\begin{array}{c} 287\\ 305\\ 319\\ 335\\ 352\\ 369\\ 403\\ 421\\ 439\\ 457\\ 476\\ 496\\ 516\\ 556\\ 556\\ 557\\ 598\\ 619\\ 640\\ 662\\ 684\\ 707\\ \end{array}$	אין 11 איזיאר 10 אינטאר 9 איז איזער 10 אינטאר 8 איזער 11 איז איזער 10 אינטאר 11 איז איזער 10 אינטאר	$\begin{array}{c} 1370\\ 1398\\ 1426\\ 1454\\ 1482\\ 1510\\ 1595\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1695\\ 1711\\ 1740\\ 1769\\ 1799\\ 1826\\ 1857\\ 1887\\ 1917\\ 1947\\ 1947\\ 1977\\ 2007 \end{array}$	אמואישה 11 אמואישה 10 אינוגראשה C אנגוגראשה 70 אמואישה 7 אינוגראשה	$\begin{array}{c} 2816\\ 2848\\ 2880\\ 2912\\ 2944\\ 2976\\ 3008\\ 3041\\ 3073\\ 3106\\ 3138\\ 3171\\ 3203\\ 3236\\ 3236\\ 3239\\ 3302\\ 3334\\ 3367\\ 3400\\ 3433\\ 3466\\ 3499\\ 3532 \end{array}$	14100047 1410048 4410049 9 14100490 11 1410041 1 141004	$\begin{array}{r} 4394\\ 4427\\ 4430\\ 4493\\ 4596\\ 4592\\ 4624\\ 4657\\ 4690\\ 4723\\ 4723\\ 4723\\ 4723\\ 4723\\ 4788\\ 4853\\ 4918\\ 4950\\ 4885\\ 4918\\ 4950\\ 4982\\ 5014\\ 5046\\ 5078\\ 5110\\ \end{array}$	המהואה ון המואראה 10 אמאראה כי אמואראה 20 אומראה 17 אמואראה 19	$\begin{array}{c} 5919\\ 5949\\ 5979\\ 6009\\ 6039\\ 6069\\ 6098\\ 6127\\ 6157\\ 6127\\ 6127\\ 6215\\ 6214\\ 6272\\ 6352\\ 6352\\ 6356\\ 6416\\ 6444\\ 6472\\ 6500\\ 6528\\ 6556\\ \end{array}$	אמוניי איד די אמנואי איד ס אמוניי איד ס אמוניר איד על אמוני איד על אמוניי איד	$\begin{array}{c} 7221\\ 7244\\ 7267\\ 7290\\ 7313\\ 7357\\ 7379\\ 7401\\ 7492\\ 7443\\ 7484\\ 7485\\ 7506\\ 7526\\ 7546\\ 7526\\ 7585\\ 7546\\ 7585\\ 7604\\ 7585\\ 7641\\ 7659\\ 7677\\ \end{array}$	аспенан Пакиман Сакимаан Сакиман соасиман чан сакиман	8018 8021 8024 8027 8030 8033 8039 8042 8045 8048 8050 8053 8056 8059 8055 8059 8062 8065 8065 8068 8071 8077 8080 8083	אמראיקא דן אמומאיר אין 10 אמונאי אין 00 אמואיראין 10 אמראיראין 14 אמונאיראין 14 אמונאיראין 14 אמראיראין 14 איז	

DOME 244 gallons = 11.60 gallons to one inch. Furnished by Pennsylvania Tank Car Company, Sharon, Pa.

Steam Coils 46 Gallons Deducted

Tank Car Outage Tables*

Calculated from 0.25 Inch to 5 Inches Out of Shell, at 60°F Capacity of Car in Gallons at 60°F

Inches	4231 Gallons	6000 Gallons	6641 Gallons	7000 Gallons	8087 Gallons	8102 Gallons	8505 Gallons	10000 Gallons
0.25	3	4	4	4	5	5	5	6
0.5	6	8	8	8	10	10	10	12
0.75	9	13	13	13	16	16	17	19
1.	13	18	18	18	23	23	25	26
1.25	18	24	25	25	31	31	33	36
1.5	23	31	33	33 .	39	39	45	46
1.75	29	38	41	41	48	48	56	58
2.	35	46	49	- 50	58	58	67	71
2.25	41	54	58	59	69	69	79	84
2.5	48	63	68	69	80	80	92	98
2.75	55	72	78	79	-90	91	105	111
3.	63	82	88	90	103	103	119	125
3.25	71	92	99	101	115	115	133	140
3.5	79	103	110	113	128	128	148	156
3.75	87	114	123	125	141	141	163	171
4.	96	125	134	137	154	154	178	186
4.25	105	136	146	150	167	167	194	203
4.5	114	148	159	163	181	181	211	220
4.75	123	160	172	176	195	195	288	237
5.	133	173	186	190	210	210	244	254

*Furnished by Phoenix Refining Co.

CONTENTS OF HORIZONTAL TANKS (GALLONS). Multiply Capacity in Tables by Length of Tanks in Inches.

36 Inches in Diameter	37 Inches in Diameter	38 Inches in Diameter	Depth Inches	39 Inches in Diameter	40 Inches in Diameter	41 Inches in Diameter
			201/2			2.858
			20		2.720	2.769
			$19\frac{1}{2}$	2.586		
		2.445	19	2.501	2.547	2.591
	2.327		181/2			
2.203	2.247	2.290	18	2.332	2.374	2.415
2.047	2.087	2.126	17	2.165	2.202	2.239
1.893	1.928	1.963	16	1.998	2.032	2.065
1.739	1.770	1.801	15	1.832	1.863	1.894
1.585	1.613	1.643	14	1.669	1.697	1.724
1.434	1.459	1.484	13	1.509	1.533	1.557
1.286	1.308	1.330	12	1.351	1.372	1.393
1.140	1.159	1.179	11	1.198	1.216	1.233
.999	1.015	1.032	10	1.047	1.063	1.079
.861	.875	.889	9	.903	.916	.929
.729	.740	.752	8	.763	.774	.785
.603	.612	.621	7	.631	.639	.648
.483	.490	.497	6	.505	.512	.518
371	.376	.382	5	.387	.392	.398
.268	.271	.275	4	.280	.283	.287
.175	.178	.180	3	.183	.185	.188
	.098	.099	2	.100	.102	.103
.034	.035	.035	1	.036	.036	.037

CONTENTS OF HORIZONTAL TANKS-Continued.

42 Inches in Diameter	43 Inches in Diameter	44 Inches in Diameter	Depth Inches	45 Inches in Diameter	46 Inches in Diameter	47 Inches in Diameter
			231/2			3.755
			23 221/2	3.442	3.597	3.653
	· · · · · · · · · · · · · · · · · · ·	3.291	22 72	3.344	3.397	3.450
	3.143	0.401	211/2	U.UTI	0.001	0.100
2,998	3.050	3.100	21	3.149	3.199	3.248
2.817	2.864	2.908	20	2.955	3.002	3.047
2.635	2.679	2.721	19	2.763	2.805	2.846
2.455	2.495	2.533	18	2.572	2.609	2.647
2.276	2.313	2.347	17	2.381	2.416	2.450
2.098	2.132	2.163	16	2.193	2.225	2.256
1.922	1.952	1.981	15	2.009	2.037	2.064
1.750	1.776	1.802	14 .	1.827	1.852	1.876
1.580	1.603	1.626 .	13	1.648	1.672	1.693
1.414	1.434	1.454	12	1.473	1.494	1.513
1.252	1.269	1.287	11	1.304	1.321	1.338
1.094	1.110	1.125	10	1.139	1.154	1.168 1.005
.942 .797	.955	.968	9 8	.980 .827	.838	.848
.657	.807 .665	.817 .675	7	.682	.691	.699
.526	.532	.540	6	.546	.552	.558
.403	.408	.414	5	.418	.424	.428
.291	.294	.297	4	.301	.304	.308
.190	.193	.194	3	.197	.199	.200
.104	.106	.107	2	.108	.110	.111
.037	.038	.038	1	.038	.039	.039

48 Inches in	49 Inches in	50 Inches in	Depth	51 Inches in	52 Inches in	53 Inches in
Diameter	Diameter	Diameter	Inches	Diameter	Diameter	Diameter
3 917 3 917 3.498 2.289 3.084 2.881 2.679 2.478 2.281 2.067 1.900 1.716 1.533 1.353 1.353 1.180 1.017 8.509 .708 .565 .432 .310 .201	4.082 3.975 3.555 3.555 3.345 3.555 3.345 2.928 2.792 2.517 2.316 2.118 1.924 1.754 1.550 1.370 1.195 1.027 .866 .575 .440 .317 2.205	4.250 4.023 3.817 3.002 3.388 3.175 2.964 2.755 2.548 2.344 1.145 1.948 1.756 1.569 1.386 1.210 1.040 8.78 .723 .578 442 3.19 2.208	$\begin{array}{c} 26^{1}/_{2}\\ 26\\ 25^{1}/_{2}\\ 25\\ 24^{1}/_{2}\\ 24\\ 22\\ 21\\ 22\\ 21\\ 19\\ 19\\ 18\\ 17\\ 16\\ 15\\ 14\\ 13\\ 12\\ 11\\ 10\\ 9\\ 8\\ 7\\ 6\\ 5\\ 4\\ 3\end{array}$	4.422 4.309 4.085 3.647 3.647 3.647 3.431 3.216 3.002 2.580 2.570 2.580 2.374 2.170 1.971 1.777 1.585 1.402 1.223 1.052 3.688 3.688 3.729 5.583 4.447 3.319 2.211	$\begin{array}{c} & 4.597 \\ \hline & 4.371 \\ \hline & 4.146 \\ \hline & 3.922 \\ 3.700 \\ 3.479 \\ 3.259 \\ 3.044 \\ 2.825 \\ 2.613 \\ 2.405 \\ 2.199 \\ 1.996 \\ 1.797 \\ 1.906 \\ 1.797 \\ 1.605 \\ 1.417 \\ 1.235 \\ 1.063 \\ .897 \\ .737 \\ .587 \\ .451 \\ .326 \\ .214 \end{array}$	$\begin{array}{r} 4.776\\ 4.600\\ \hline \\ 4.630\\ \hline \\ 4.431\\ \hline \\ 4.203\\ 3.976\\ 3.749\\ 3.523\\ 3.300\\ 3.078\\ 2.859\\ 2.644\\ 2.432\\ 2.222\\ 2.016\\ 1.815\\ 1.622\\ 1.433\\ 1.251\\ 1.077\\ .907\\ .746\\ .595\\ .454\\ .329\\ .214\\ \end{array}$
.113	.114	.114	2	.114	.117	.119
.040	.041	.041	1	.041	.041	.042

54 Inches in Diameter	55 Inches in Diameter	56 Inches in Diameter	Depth Inches	57 Inches in Diameter	58 Inches in Diameter	59 Inches in Diameter
	1.000					
			291/2			5.918
			29		5.719	5.790
			281/2	5.523		0.100
		5.331	28	5.399	5.467	5,535
	5.143	0.001	271/2	0.000	0.201	0.000
4.957	5.023	5.089	27	5.153	5.217	5.280
4.723	4.785	4.847	23	4.907	4.967	5.026
4.490	4.547	4.605	25	4.662	4.717	4.773
4.258	4.311	4.365	24	4.417	4.469	4.521
4.026	4.076	4.125	23	4.175	4.223	4.271
3.794	3.842	3.886	22	3.934	3.978	4.023
3,566	3.611	3.651	21	3.694	3.736	3.777
3.340	3.381	3.418	20	3.456	3.495	3.534
3.116	3.152	3.188	19	3.222	3.256	3.293
2:893	2.926	2.959	18	2.992	3.020	3.057
2.674	2.704	2.734	17	2.766	2.788	2.823
2.459	2.486	2.513	16	2.543	2.563	2.594
2,248	2.271	2.296	15	2.321	2.344	2.369
2.041	2.061	2.084	14	2.104	2.128	2.149
1.838	1.857	1.878	13	1.895	1.916	1.934
1.640	1.657	1.675	12	1.692	1.710	1.726
1.449	1.464	1.478	11	1.496	1.509	1.524
1.265	1.279	1.290	10	1.304	1.316	1.329
1.086	1.099	1.108	9	1.120	1.130	1.141
.915	.926	.936	8	.943	.953	.961
.755	.759	.769	7	.776	.784	.791
.602	.607	.614	6	.620	.626	.631
.461	.466	.470	5	.473	.479	.483
.331	.335	.337	4	.340	.344	.347
.217	.219	.220	3	.223	.225	.227
.119	.120	.121	$\frac{2}{1}$.122	.123	.124
.042	.042	.043	1	.043	.044	.044

60 Inches in Diameter	61 Inches in Diameter	62 Inches in Diameter	Depth Inches	63 Inches in Diameter	64 Inches in Diameter	65 Inches in Diameter
			321/2			7.182
			32		6.963	7.039
			311/2	6.747		
		6.535	31	6.610	6.686	6.755
	6.326		301/2			
6.119 ·	6.193	6.267	30	6.337	6.410	6.472
5.858	5.929	5.999	29	6.065	6.134	6.193
5.598	5.668	5.732	28	5.794	5.858	5.915
5.339	5.407	5.465	27	5.523	5.584	5.639
5.082	5.146	5.199	26	5.254	5.310	5.363
4.826	4.885	4.935	25	4.986	5.038	5.089
4.572	4.625	4.672	24	4.722	4.769	4.817
4.318	4.366	4.412	23	4.458	4.503	4.547
4.066	4.111	4.153	22	4.196	4.239	4.281
3.818	3.859	3.898	21	3.937	3.976	4.016
3.572	3.609	3.645	20	3.683	3.718	3.756
3.328	3.363	3.397	19	3.430	3.464	3.496
3.088	3.120	3.151	18	3.181	3.213	3.242
2.582	2.881	2.910	17	2.937	2.964	2.992
2.621	2.646	2.672	16	2.698	2.723	2.748
2.392	2.417	2.440	15	2.463	2.486	2.508
2.171	2.192	2.213	14	2.232	2.254	2.274
1.954	1.972	1.991	13	2.008	2.027	2.045
1.743	1.759	1.776	12	1.791	1.808	1.823
1.538	1.552	1.567	11	1.581	1.595	1.608
1.341	1.352	1.366	10	1.378	1.390	1.401
1.152	1.161	1.173	9	1.183	1.192	1.203
.971	.980	.988	8	.996	1.005	1.013
.799	.806	.812	7	.819	.827	.833
.634	.642	.648	6	.653	.659	.664
.487	.491	.496	5	.500	.504	.508
.349	.354	.357	4	.359	.362	.365
.229	.230	.233	3	.235	.238	.238
.125	.126	.128	2	.128	.129	.131
.045	.045	.045	1	.046	.046	.047

66 Inches in Diameter	67 Inches in Diameter	68 inches in Diameter	Depth inches	60 Inches in Diameter	70 Inches in Diameter	71 Inches in Diameter
			351/2			8.570
• • • • • • • • • • • • •	•••••		35	0.004	8.330	8.413
	• • • • • • • • • • • • • • • •	7.861	$\frac{341}{2}$ 34	8.094 7.944	8.026	8.107
	7.631	7.801	331/2	1.944	8.020	8.107
7.406	7.485	7.567	33	7.640	7.723	7.801
7.120	7.194	7.273	32	7.348	7.421	7.495
6.834	6.904	6.979	31	7.051	7.120	7.190
6.549	6.617	6.687	30	6.755	6.819	6.886
6.264	0.227	6.395	29	6.459	6.519	6.583
5.981	6.041	6,104	28	6.164	6.222	6.283
5.699	5.756	5.814	27	5.870	5.927	5.983
5.419	5.473	5.528	26	5.580	5.634	5.686
5.141	5.191	5.244	25	5.292	5.343	5.391
4.865	4.913	4.961	24	5.006	5.052	5.098
4.592	4.037	4.681	23	4.724	4.764	4.809
4.322	4.363	4.403	22	4.444	4.481	4.524
4.054	4.032	4.123	21	4.167	4.204	4.241
3.789	3.824	3.859	20	3.893	3.929	3.962
3.529	3.561	3.593	19	3.625	3.657	3,688
3.273	3.302	3.331	18	3.360	3,388	3.418
3.020	3.046	3.074	17	3.101	3.125	3.152
2.772	2.797	2.821	16	2.846	2.868	2.894
2.530	2.553	2.575	15	2.595	2.617	2.640
2.294	2.314	2.333	14	2.352	2.372	2.391
2.064	2.080	2.099	13	2.116	2.135	2.150
1.839	1.855	1.871	12	1.886	1.901	1.916
1.622	1.635	1.650	11	1.663	1.674	1.693
1.413	1.426	1.439	10	1.449	1.459	1.476
1.213	1.223	1.235	9	1.242	1.254	1.264
1.022	1.030	1.041	8	1.047	1.060	1.063
.841	.847	.855	7	.859	.871	.874
.670	.675	.680	6	.687	.689	.697
.512	.516	.529	5	.524	.528	.531
.368	.371	.374	4	377	.378	.382
.240	.243	.244	3	.246	249	.250
.131	.132	.133	2	.134	.135	.136
.047	.047	.047	1	.048	.048	.048

72 Inches in Diameter	73 Inches in Diameter	74 Inches in Diameter	Depth Inches	75 Inches in Diameter	76 Inches in Diameter	77 Inches i Diameter
				a carl	A COM LA	17 - Adama
			381/2			10.079
			33 ,		. 9.819	9.912
			371/2	9.562		
		9.309	37	· 9.400	9.489	9.579
	9.059		361/2			
8.813	8.899	8.989	36	9.076	9.160	9.246
8.500	8.582	8.669	35	8.752	8.832	8.914
8.188	8.267	8.349	34	8.428	8.505	8.583
7.887	7.953	8.030	33	8.104	8.178	8.253
7.567	7.639	7.712	32	7.782	7.782	7.924
7.259	7.326	7.395	31	7.461	7.528	7.596
6.952	7.015	7.080	30	7.142	7.205	7.268
6.645	6.706	6.766	29	6.824	6.885	6.944
6.341	6.397	6.454	28	6.509	6.567	6.622
6.038	6.091	6.145	27	6.195	6.250	6.302
5.736	5.786	5.839	26	5.885	5.938	5.988
5.439	5.485	5.535	25	5.578	5.628	5.675
5.144	5.188	5.232	24	5.274	5.320	5.364
4.852	4.892	4.934	23	4.975	5.014	5.056
4.563	4.599	4.639	22	4.677	4.715	4.753
4.278	4.311	4.374	21	4.383	4.418	4.453
3.997	4.025	4.062	20	4.094	4.127	4.161
3.719	3.748	3.781	19 .	3.809	3.839	3.871
3.446	3.474	3.501	18	3.529	3.556	3.585
3.179	3.204	3.229	17	3.255	3.280	3.305
2.917	2.938	2.962	16	2.985	3.008	3.032
2.658	2.681	2.702	15	2.723	2.744	2.764
2.408	2.429	2.447	14	2.467	2.485	2.503
2.167	1.184	2.200	13	2.216	2.234	2.250
1.932	1.946	1.960	12	1.978	1.990	2.003
1.703	1.716	1.727	11	1.742	1.753	1.767
1.483	1.494	1.505	10	1.515	1.527	1.538
1.272	1.281	1.291	9	1.300	1.309	1.318
1.071	1.079	1.086	8	1.095	1.102	1.110
.880	.887	.893	7	.899	.906	.912
.701	.707	.712	6	.717	.722	.727
.536	.540	.544	5	.548	.551	.555
.386	.388	.391	4	.393	.396	.399
.252	.253	.254	3	.256	.259	.260
.138	.138	.139	2	.140	'.141	.142
.048	.049	.049	1	.050	.050	.050

78 Inches in Diameter	79 Inches in Diameter	80 Inches in Diameter	Depth Inches	81 inches in Diameter	82 Inches in Diameter	83 Inches in Diameter
	1.0		411/2			11.711
• • • • • • • • • • • • •	••••	•••••			11.431	11.711
• • • • • • • • • • • • • •			41 401/2	11.154	11.451	11.001
• • • • • • • • • • • • •	•••••	10.000	40 72	10.978	11.075	11.172
• • • • • • • • • • • • • • •	010 of	10.880	391/2	10.978	11.075	11,172
10.040	10.610	10 500	3972	10.627	10.720	10.814
10.343	10.439	10.533	39	10.027	10.720	10.814
10.000	10.097	10.187	37			
9.666	9.756	9.841		9.927	10.012	10.098
9.329	9.416	9.496	36	9.578	9.659	9.741
8.994	9.076	9.151	35	9.231	9.307	9.385
8.659	8.737	8.809	24	5.532	5.574	5.615
8.325	8.398	8.468	33	8.538	8.608	8.679
7.992	8.060	8.128	32	8.194	8.260	8.328
7.660	7.724	7.789	31	7.854	7.916	7.980
7.330	7.391	7.454	30	7.514	7.575	7.633
7.001	7.059	7.120	29	7.176	7.234	7.286
6.676	6.734	6.788	28	6.842	6.893	6.947
6.354	6.407	6.458	27	6.508	6.557	6.610
6.035	6.085	6.132	26	6.181	6.228	6.274
5.719	5.764	5.809	25	5.583	5.899	5.943
5.406	5.449	5.490	24	5.532	5.574	5.615
5.096	5.138	5.175	23	5.212	5.252	5,291
4.791	4.829	4.864	22	4.900	4.933	4.970
4.487	4.523	4.557	21	4.592	4.624	4.657
4.189	4.224	4.254	20	4.286	4.316	4.436
3.897	3.928	3.956	19	3.987	4.013	4.043
3.610	3.637	3.665	18	3.691	3.717	3.742
3.329	3.355	3.377	17 -	3.403	3.426	3.450
3.053	3.076	3.098	16	3.120	3,141	3.164
2.784	2.804	2.825	15	2.846	2.863	2.883
2.522	2.540	2.558	14	2.576	2.592	2.612
2.267	2.282	2.299	13	2.315	2.329	2.345
2.019	2.033	2.047	12	2.062	2.074	2.089
1.779	1.791	1.804	11	1.816	1.827	1.840
1.548	1.560	1.570	10	1.582	1.591	1.606
1.328	1.336	1.345	9	1.355	1.365	1.372
1.118	1.126	1.132	8	1.141	1.148	1.156
.919	.925	.931	7	.937	.943	.950
.731	.736	742	7 5	.569	.574	.576
.559	.563	.565	6	.746	.752	.757
.401	.404	.407	4	.409	.412	.415
.261	.264	.265	3	.267	.269	.269
.143	.143	.145	2	.146	.147	.148
.051	.051	.051	ĩ	.052	.052	.053

HOH	RIZONTAL T	ANKS.	
Multiply Capacity in	Tables by Ler	ngth of Tanks	in Inches.

84 inches in Diameter	85 inches in Diameter	86 inches in Diameter	Depth Inches	87 inches in Diameter	88 inches in Diameter	89 inches in Diameter
						1.1
10.113			441%			13.466
•••••			44		13,165	
• • • • • • • • • • • • • •			431/2	12.867	19.109	• • • • • • • • • • • • • •
		12.573	43	12.679	12.783	12.887
	12.283		421/2	12.079	. 12.100	12.00/
11.995	12.099	12.201	42	12.303	12.401	12.501
11.632	11.731	11.829	41	11.927	12.019	12.116
11.269	11.363	11.457	40-	11.552	11.638	11.734
10.903	10.997	11.086	39	11.177	11.261	11.352
10.544	10.632	10.716	38	10.802	10.884	10.970
10.183	10.267	10.347	37	10.430	10.508	10.589
9.822	9.903	9.979	30	. 10.058	10.132	10.209
9.462	9.540	9.611	35	9.687	9.759	9.832
9.104	9.177	9.245	34	9.318	9.387	9.458
8.747	8.816	8.883	33	8.951	9.018	9.085
8.392		8,525	32	8.587	8.651	8.713
8.040	8.459		31	8.226	8.287	8.345
	8.105	8.164	30			
7.690	7.751	7.807	29	7.865	7.925	7.978
7.344	7.401	7.454		7.509	7.566	7.617
7.000	7.054	7.104	28	7.156	7.210	7.258
6.658	6.710	6.756	27	6.805	6.856	6.901
6.320	6.369	6.413	26	6.458	6.504	6 549
5.983	6.030	6.074	25	6.118	6.158	6.201
5.656	5.699	5.738	24	5.773	5.816	5.858
5.330	5.368	5.404	23	5.445	5.482	5.516
5.007	5.043	5.078	22	5.114	5.150	5.182
4.690	4.724	4.756	21	4.790	4.821	4.855
4.378	4.410	4.440	20	4.469	4.499	4.528
4.071	4.098	4.126	19	4.155	4.181	4.211
3.770	3.796	3.821	18	3.847	3.872	3.896
3.475	3.497	3.522	17	3.544	3.576	3.590
3.186	3.206	3.227	16	3.249	3.269	3.291
2.904	2.924	2.941	15	2.961	2.980	2.999
2.629	2.646	2.663	14	2.679	2.699	2.714
2.362	2.378	2.393	13	2.406	2.421	2.439
2.104	2.116	2.129	12	2.142	2.154	2.169
1.853	1.865	1.876	11	1.888	1.900	1.200
1.613	1.621	1.633	10	1.641	1.656	1.663
1.383	1.391	1.400	9	1.407	1.416	1.425
1.162	1.169	1.176	8	1.185	1.190	1.200
.954	.962	.967	7	.973	.979	.983
.760	.765	.770	6	.776	.778	.784
.580	.585	.587	5	.592	.595	.598
.417	.420	.422	4	.429	.429	.430
.272	.274	.275	3	.278	.279	.280
.148	.149	.151	2	.151	.153	.154
.053	.053	.053	1	.054	.055	.055

90 Inches in Diameter	91 Inches in Diameter	92 Inches in Diameter	Depth Inches	93 Inches in Diameter	94 Inches in Diameter	95 Inches in Diameter
			471/2			15.342
			47		15.021	15.136
	•••••	14.388	46½ 46	14.703 14.501	14.612	14.726
	14.078	11.000	451/2	11.001	11.012	
13.770	13.880	13.988	45	14.098	14.207	14.316
13.378	13.487	13,590	44	13.696	13.802	13.905
12.987	13.094	13.194	-43	13.296	13.397	13.495
12.597	12.701	12.798	42	12.896	12.993	13.086
12.209	12,308	12,403	41	12.497	12.590	12.679
11.822	11.915	12.008	40	12.098	12.187	12.273
11.436	11.525	11.613	39	11.699	11.785	11.867
11.051	11.137	11.218	38	11.301	11.384	11.463
10.667	10.750	10.826	37	10,906	10.983	11.061
10.284	10.363	10.438	36	10.513	10.587	10.662
9.903	9.977	10.050	35	10.123	10.193	10.265
9.524	9.596	9,665	34	9.733	9.800	9.870
9.184	9.216	9.281	33	9.344	9.410	9.476
8.773	8.837	8,900	32	8.962	9.024	9.084
8.403	8.463	8.523	31	8.580	8.639	8.697
8.035	8.093	8.149	30	8.200	8.257	8.313
7.670	7.724	7.777	29	7.827	7.880	7.932
7.308	7.358	7.409	28	7.456	7.506	7.553
6,948	6,996	7.046	27	7.089	7.138	7.182
6.593	6.638	6.687	26	6.727 -	6.771	6.812
	6.283		25	6.367	6.407	6.450
6.242	5.934	$6.331 \\ 5.976$	23	6.013	6.052	6.090
5.894			23	5.662	5.700	5.734
5.552	5.588	5.626	23 22	5.320	5.352	5.386
5.215	5.248	5.284	22			
4.883	4.916	4.948	21	4.979 4.647	$5.010 \\ 4.673$	5.042
4.656	4.587	4.617	20 19			4.701
4.235	4.264	4.292	19	4.317	4.343	4.368
3.921	3.946	3.972	18	3.996	4.021	4.045
3.611	3.635	3.657		3.681	3.703	3.727
3.309	3.331	3.353	16	3.375	3.393	3.414
3.014	3.035	3.056	15	3.073	3.091	3.109
2.729	2.747	2.763	14	2.781	2.796	2.814
2.452	2.468	2.480	13	2.497	2.510	2.524
2.183	2.196	2.210	12	2.222	2.232	2.248
1.922	1.934	1.946	11	1.957	1.966	1.981
1.673	1.682	1.696	10	1.703	1.714	1.723
1.433	1.443	1.455	9	1.459	1.469	1.474
1.204	1.214	1.216	8	1.226	1.232	1.240
.989	.995	1.000	7	1.007	1.010	1.019
.787	.793	.799	6	.803	.807	.812
.601	.605	.608	5	.613	.616	.618
.432	.435	.440	4	.440	.445	.445
.281	.284	.290	3	.290	.291	.292
.154	.155	.156	2	.157	.158	.160
.055	.055	.056	1	.056	.056	.056

96 Inches in Diameter	Depth Inches	97 Inches in Diameter	96 Inches in Diameter	Depth Inches	97 Inches in Diameter
	481/2	15,995	6,128	24	6,163
15.638	48	15.785	5.770	23	5.803
15.248	47	15.365	5.416	22	5.450
14.828	46	14.945	5.066	21	5.101
14.410	45	14.525	4.726	20	4.757
13.992	44	14.108	4.394	19	4.421
13.574	43	13.692	4.068	18	4.092
13.158	42	13.276	3.752	17	3.770
12.744	41	12.860	3.444	16	3.455
12.333	40	12.446	3.139	15	3.145
11.930	39	12.033	2.838	14	2.844
11.524	38	11.622	2.546	13	2.554
11.119	37	11.214	2.230	12	2.273
10.716	36	10.807	1.990	11	2.001
10.315	35	10,400	1.728	10	1.742
9.915	34	9,997	1.480	9	1.492
9.518	33	9.599	1.240	8	1.254
9.124	32	9.204	1.016	7	1.032
8.736	31	8.810	.804	6	.821
8.352	30	8,420	.620	5	.625
7.974	29	8,035	.447	4	.448
7.600	28	7.654	.292	3	.293
7.230	27	7.274	.160	2	.160
6.862	23	6.897	.057	1	.057
6.494	25	6.526			

98 Inches in Diameter	Depth Inches	59 Inches in Diameter	98 Inches in Diameter	Depth Inches	99 Inches in Diameter
	491/2	166.662	6.569	25	6.607
16.327	49	16.446	6.203	24	6.239
15.898	48	16.016	5.841	23	5.874
15.473	47	15.587	5.484	22	5.514
15.049	46	15.159	5.131	21	5.160
14.626	45	14.732	4.786	20	4.814
14.205	44	14.305	4.449	19	4.472
13.784	43	13.880	4.116	18	4.138
13.363	42	13.458	3.792	17	3.811
12.944	41	13.036	3.472	16	3,491
12.527	* 40	12,615	3.160	15	3,181
12.111	39	12.197	2.856	14	2.878
11.698	38	11.780	2.565	13	2.583
11.287	37	11.365	2.282	12	2,298
10.877	36	10,952	2.016	11	2.025
10.468	35	10.539	1.754	10	1.759
10 063	34	10.128	1.501	9	1.508
9.661	33	9.723	1.260	8 7	1.266
9.263	32	9.322	1.035	7	1.040
8.867	31	8.921	.823	6 .5	.828
8.473	30	8.526	.628	.5	.633
8.085	29	8.136	.453	4	.453
7.700	28	7.747	.295	4 3	.297
7.318	27	, 7.362	.162	2	.162
6.940	26	6.982	.058	1	.058

100 Inches in Diameter	Depth Inches	101 Inches in Diameter	100 Inches in Diameter	Depth Inches	101 Inches in Diameter
					-
	501/2	17.342	6.647	25	6.685
17.000	50	17.122	6.274	24	6.311
16.565	49	. 16.683	5.908	23	5.942
16.132	48	16.247	5.546	22	5.579
15.699	47	15.812	5.190	21	5.221
15.267	46	15.377	4.841	20	4.868
14.837	45	14.942	4.498	19	4.523
14.407	44	14.507	4.162	18	4.185
13.978	43	14.073	3.833	17	3.855
13.551	42	13.642	3.511	16	3.531
13.125	41	13.213	3.198	15	3.215
12.700	40	12.784	2.893	14	2.908
12.277	39	12.356	2.597	13	2.612
11.855	38	11.931	2.311	12	2.324
11.436	37	11,508	2.035	11	2.041
11.020	36	11.090	1.769	10	1.779
10.605	- 35	10.672	1.516	9	1.524
10.194	34	10.257	- 1.274	8	1.282
9,785	33	9.846	1.046	7	1.053
9.379	32	9.437	.833	6	.838
8.977	31	9.032	.636	5	.640
8.578	30	8.630	.456	4	.458
8.184	29	8.233	.297	4 3	.298
7.793	28	7.840	.162	2	.162
7.407	27	7.450	.058	1	.058
7.024	26	7.065			

102 Inches in Diameter	Depth Inches	103 Inches in Diameter	102 Inches in Diameter	Depth Inches	103 Inches in Diameter
	511/2	18.033	7.108	26	7.148
17.687	51	17.811	6,722	25	6.764
17.246	50	17.364	6.340	24	6.387
16.805	49	16.918	5.972	23	6.010
16.364	48	16.473	5.608	22	5,644
15.924	47	16.030	5.251	21	5.281
15.485	46	15.587	4.895	20	4.924
15.047	45	15.144	4.549	19	4.576
14.609	44	14.701	4.208	18	4.230
14.172	43	14.259	3.877	17	3.896
13.738	42	13.819	3.554	16	3.568
13.304	41	13.384	3.235	15	3.250
12.871	40	12.950	2.916	14	2.938
12.440	39	12.516	2.622	13	2.639
12.011	38	12.083	2.333	12	2.348
11.587	37	11.655	2.056	11	2.069
11.163	36	11.229	1.787	10	1.798
10.743	35	10.805	1.531	9	1.542
10.325	34	10.386	1.278	8 7	1.295
9.911	33	9.968	1.057	7	1.064
9.498	32	9.556	.854	6	.844
9.087	31	9.147	.642	5	.646
8.680	30	8.738	.458	4	.462
8.282	29	8.331	.300	3	.301
7.884	28	7.930	.163	2	.164
7.497	27	7.537	.058	1	.059

104 Inches in Diameter	Depth Inches	105 Inches in Diameter	104 Inches in Diameter	Depth Inches	105 Inches in Diameter
	521/2	18.742	7.190	26	7.229
18.387	52	18.513	6.804	. 25	6.841
17 936	51	18.057	6.423	24	6.457
17.485	50	17.603	6.046	23	6.075
17.035	49	17.150	5.671	22	5.704
16.587	48	16.697	5.308	21	5.336
16.140	47	16.245	4.950	20	4.978
15.693	46	15.794	4.599	19	4.626
15.247	45	15.343	4.255	18	4.277
14.802	44	14.893	3.920	17	3.938
14.357	43	14.447	3.588	16	3.608
13.912	42	14.002	3.267	15	3.285
13.470	41	13.558	2.955	14	2.971
13.032	40	13.116	2.653	13	2.667
12.597	39	12.675	2.361	12	2.373
12.164	38	12.237	2.080	11	2.090
11.732	37	11.802	1.809	10	1.814
11.297	36	11.371	1.548	9	1.556
10.872	35	10.940	1.300	8	1.308
10.450	34	10.511	1.068	7	1.074
10.029	33	10.088	.850	6	.853
9.610	32	9.666	.649	5	.652
9.198	31	9.249	.467	4	.469
8.789	30	8.837	.302	3	.304
8.382	29	8.430	.164	21	.165
7.978	28	8.025	.059	1	.059
7.582	27	7.623			

106 Inches in Diameter	Depth Inches	107 Inches in Diameter	106 Inches in Diameter	Depth Inches	107 Inches in Diameter
	531/2	19.463	7.668	27	7.710
19.101	53	19.230	7.272	26	7.312
18,639	52	18.766	6.877	25	6.919
18,180	51	18.303	6.491	24	6.526
17.723	50	17.841	6.111	23	6.143
17.266	49	17.381	5.733	22	5,767
16.810	48	16.922	5.396	21	5,395
16.354	47	16.463	5.005	20	5.029
15.898	46	16.004	4.648	19	4.673
15.444	45	15.545	4.300	18	4.323
14.991	44	15.087	3.960	17	3.980
14.539	43	14.629	3.626	16	3.643
14.089	42	14.176	3.302	15	3.320
13.642	41	13.724	2.988	14	3.001
13.196	40	13.275	2.680	13	2.696
12.752	39	12.828	2.384	12	2.398
12.310	38	12.384	2.101	11	2.110
11.899	37	11.943	1.824	10	1.834
11.434	36	11.503	1.564	9	1.571
11.005	35	11.069	1.314	8	1.320
10.576	34	10.635	1.077	8 7	1.084
10.150	33	10.055	.858	6	.862
9,725			.655	5	.658
9.725	32 31	9.779	.470	4	.473
9.303		9.354	.306	4 3	.306
	30	8.937	.166	2	.167
8.474	29	8.523	.059	1	.060
8.069	28	8.116	.039	1	.000

Line and Par	J Cupatin		5		
108 Inches in Diameter	Depth Inches	109 Inches in Diameter	108 Inches in Diameter	Depth Inches	109 Inches in Diamete
	541/2	20.198	7.756	27	7.796
19.828	54	19.962	7.352	26	7.391
19.359	53	19.490	6.953	25	6.993
18.892	52	19.019	6.560	24	6.597
18.426	51	18.548	6.176	23	6.209
17.961	50	18.077	5.797	22	5.827
17.496	49	17.607	5.428	21	5.453
17.031	48	17.137	5.059	20	5.084
16.567	47	16.670	4.696	19	4.720
16.103	46	16.203	4.343	18	4.367
15.639	45	15.737	4.000	17	4.022
15.178	44	15.272	3.661	16	3.682
14.719	43	14.810	3.335	15	3.353
14.263	42	14.349	3.020	14	3.032
13.810	41	13.890	2.711	13	2.723
13.359	40	13.435	2.409	12	2.422
12,910	39	12.983	2.121	11	2.131
12.464	38	12.531	1.843	10	1.852
12.019	37	12,083	1.575	9	1.586
11.576	36	11.639	1.323	8	1.336
11.135	35	11.197	1.085	8 7	1.095
10.698	34	10.758		6 5	.871
10.265	33	10.322	.662	5	.665
9.836	32	9.892	.476	4	.477
9.412	31	9.463	.309	3	.309
9.992	30	9.037	.169	3 2 1	.170
8.576	29	8.619	.060	1	.060
8.165	28	8.207			

110 Inches in Diameter	Depth Inches	111 Inches in Diameter	110 Inches in Diameter	Depth Inches	111 Inches in Diameter
	-551/2	20.946	8.244		8.290
20.570	55	20,703	7.833	27	7.878
20.093	54	20.219	7.428	26	7.468
19.616	53	19.738	7.026	25	7.063
19,140	52	19.259	6.628	24	6.665
18.664	51	18.781	6.238	23	6.274
18,188	50	18.305	5.850	22	5.888
17.715	49	17.829	5.481	21	5.509
17.244	48	17.353	5.116	20	5.136
16.774	47	16.877	4.754	19	4.771
16.304	46	16,403	4.396	18	4.413
15.836	45	15.932	4.046	17	4.059
15.368	44	15,461	3.704	16	3.718
14.905	43	14.992	3.366	15	3.385
14.444	42	14.523	3,036	14	3.062
13.983	41	14.064	2.724	13	2.748
13.524	40	13,589	2.428	12	2.445
13.066	39	13.130	2.140	11	1.153
12.608	38	12.676	1.864	10	1.870
12.155	37	12.223	1.599	9	1.600
11.704	36	11.772	1.347		1.347
11.258	35	11.323	1.102	8 7	1.106
10.816	34	10.879	.876	6	.880
10.378	33	10.437	.671	5	.671
9.944	32	10.002	.479	4	.480
9.514	31	9.570	.310		.312
9.087	30	9.141	.170	$3 \\ 2 \\ 1$.170
8.664	29	8.714	.060	1	.061

KANSAS CITY TESTING LABORATORY

HORIZONTAL TANKS. Multiply Capacity in Tables by Length of Tank in Inches.

56 ¹ / ₂ 50 55 54 53	$\begin{array}{r} 21.707 \\ 21.461 \\ 20.971 \\ 20.481 \end{array}$	8.338 7.919 7.507	28 27	8.383
50 55 54 53	$21.461 \\ 20.971$	7.919		
55 54 53	20.971			7.962
54 53			26	7.548
53		7.101	25	7.139
	19.991	6.703	24	6.736
. 52	19.504	6.307	23	6.339
51	19.017	5.916	22	5.948
50	18.530	5.536	21	5.560
49	18.044	5.163	20	5.188
48	17.559	4.795	19	4.817
47	17.074	4.434	19	4.017
46	16.590	4.081	18	4.437
40	16.112	3.738	16	4.101
44	15.638	3.402	15	3.419
43	15.165	3.077	14	
40	14.692	2.764	14	$3.091 \\ 2.772$
42	14.092	2.457		
			12	2.468 2.171
				1.887
				1.615
				1.357
				1.113
			0	.886
				.675
				.486
			3	.317
			2	.171
		.031	1	.062
	40 39 38 37 36 35 34 33 32 31 30 29	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

114 Inches in Diameter	Depth Inches	115 Inches in Diameter	114 Inches in Diameter	Depth Inches	115 Inches in Diameter
	571/2	22.482	8.855	29	8,898
22.093	57	22.230	8.425	28	8.468
21.599	56	21.733	8.003	27	8.040
21,105	55	21.236	7.583	26	7.622
20.611	54	20,740	7.176	25	7.213
20.117	53	20.244	6.770	24	6.806
19.624	52	19.748	6.359	23	6,401
19.132	51	19.252	5.978	22	6.007
18.643	50	18.756	5.592	21	5.619
18,155	49	18.262	5.212	20	5.238
17.668	48	17.772	4.841	19	4.865
17.181	47	17.282	4.476	18	4.499
16.695	46	16.795	4.120	17	4.139
16.212	45	16.309	3.771	.16	3.786
15.731	44	15.823	3.436	15	3.451
15.253	43	15.341	3.109	14	3.121
14.775	42	14.862	2.786	13	2,799
14.299	41	14.383	2.481	12	2.491
13.828	40	13.906	2.183	11	2.192
13.360	39	13.431	1.898	10	1.907
12.893	38	12.964	1.624	9	1.632
12,428	37	12.497	1.365	. 8	1.371
11.967	36	12.033	1.120	7	1.126
11.511	35	11.572	.890	6	.895
11.057	34	11.116	.681	5	.684
10.609	33	10.664	.488	4	.490
10.165	32	10.217	.317	3	.319
9.722	31	9.771	.172	2	.173
9.288	30	9.331	.062	1	.062

116 Inches in Diameter	Depth Inches	117 Inches in Diameter	116 Inches in Diameter	Depth Inches	117 Inches in Diameter
	581/2	23.271	8.944	29	8.988
22.875	58	23.016	8.513	28	8.555
22.371	57	22,506	8.086	27	8.125
21.868	56	21.998	7.663	26	7.701
21.366	55	21.493	7.247	25	7.282
20.865	54	20.989	6.838	24	6.870
20.365	53	20.485	6.434	23	6.460
19.866	52	19.992	6.036	22	6.065
19.368	51	19.479	5.645	21	5.675
18.870	50	18.977	5.262	20	5.292
18.373	49	18.476	4.888	19	4.913
17.877	48	17.975	4.519	18	4.541
17.382	47	17.478	4.160	17	4.179
16.888	46	16,984	3.813	16	3.826
16.398	45	16,491	3.468	15	3.483
15.911	44	15.999	3.136	14	3.149
15.427	43	15.510	2.813	13	2.828
14.944	42	15.024	2.502	12	2.516
14.462	41 .	14.540	2.201	11	2.215
13.981	40	14.056	1.914	10	1.925
13.501	39	13.578	1.639	9	1.645
13.023	38	13.102	1.376	8	1.385
12.549	37	12.632	1.131	7	1.136
12.079	36	12,162	.899	6	.903
11.613	35	11.698	.686	5	.689
11.152	34	11.238	.492	4	.496
10.697	33	10.778	.320		.321
10.250	32	10.323	.175	3 2 1	.175
9.812	31	9.872	.062	1	.063
9.377	30	9.428			

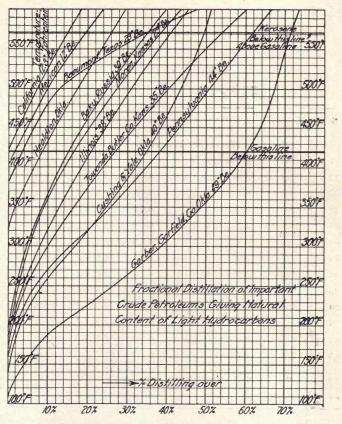
HORIZONTAL TANKS. Multiply Capacity in Tables by Length of Tank in Inches

118 Inches in Diameter	Depth Inches	119 Inches in Diameter	118 Inches in Diameter	Depth Inches	119 Inches in Diameter
	591/2	24.074	9.476	30	9.524
23.671	59	23.816	9.031	29	9.082
23.160	58	23.301	8.595	28	8.643
22.649	57	22.787	8.165	27	8.207
22.138	56	22.273	7.739	26	7.779
21.627	55	21.760	7.319	25	7.357
21.117	54	21.247	6.905	24	6.940
20.609	53	20,734	6.496	23	6,529
20.102	52	20,221	6.094	22	6.127
19.597	51	19.710	5.702	21	5.730
19.092	50	19.203	5.317	20	5.342
18.587	49	18,697	4.937	19	4.959
18.083	48	18.191	4.562	18	4.587
17.582	47	17.685	4.197	* 17	4.220
17.082	46	17.182	3.845	16	3.867
16.584	45	16.681	3.501	15	3.520
16.088	44	16.180	3.163	14	3.180
15.595	43	15.682	2.841	13	2.853
15.105	42	15.188	2.526	12	2.535
14.620	41	14.697	2.223	īī	2.232
14.137	40	14.209	1.932	10	1.938
13.654	39	13.725	1.655	9	1.659
13.174	38	13.245	1.390	8	1.396
12.698	37	12.767	1.141	7	1,146
12.225	36	12.291	.909	6	.910
11.758	35	11.818	.694	5	
11.292	34	11.350	.497	4	.498
10.832	33	10.888	.322	4 3	325
10.377	32	10.429	.175	2	.178
9.924	31	9.975	.063	2 1	.063

120 Inches in Diameter	Depth Inches	120 Inches in Diameter	Depth Inches	120 Inches in Diameter	Depth Inches
24.479	60	14.287	40	5.363	20
23.954	59	13.797	39	4.981	19
23.434	58	13.314	38	4.608	18
22.914	57	12.833	37	4.240	17
22.395	56 _	12.354	36	3.882	16
21.877	55	11.881	35	3.538	15
21.359	54	11.411	34	3.198	14
20.842	53	10.944	33	2.866	13
20.328	52	10.483	32	2.537	12
19.815	51	10.024	31	2.239	11
19.305	50	9.567	30	1.949	10
18.795	49	9.124	29	1.668	9
18.287	48	8.683	28	1.396	8
17.780	47	8.244	27	1.151	7
17.273	46	7.816	26	.915	6
16.767	45	7.393	25	.699	5
16.265	44	6.976	24	.501	4
15.768	43	6.561	23	.326	3
15.273	42	6.153	22	.178	2
14.779	41	5.751	21	.063	1

Content of Crude Oils (Typical Samples.)

Source	Specific Gravity	Natural Commercial Automobile Gasoline, % by Vol. to 410° F.	Kerosene, 410° F., 572° F., % by Vol.	Total Obtainable Gasoline, Natural and Artificial (KCTL Test), % by Vol.
Garber, Garfield Co., Oklahoma	49.5°Be' 0.780	60.0%	10.8%	91.0%
Pennsylvania (Light)	44.5° Be' 0.802	37.5%	12.7%	86.2%
Cushing, Oklahoma	40.1°Be' 0.823	35.0%	15.0%	83.7%
Towanda, Butler Co., Kansas	34.7°Be/ 0.850	26.5%	27.5%	77.9%
Neodesha, Wilson Co., Kansas	33.3°Be' 0.800	25.0%	17.0%	81.2%
Newkirk, Oklahoma	40.3° Be' 0.822	32.5%	24.0%	83.1%
Mexico (Panuco)	11.4°Be/ 0.990	2.0%	18.0%	44.5%
California	12.3°Be' 0.984	0.0%	12.3%	50.0%
Texas (Beaumont)	23.4°Be/ 0.912	4.0%	16.0%	61.0%
Russia	30.2° Be' 0.874	15.0%	20.0%	
Healdton, Oklahoma.	22.1°Be' 0.920	8.5%	17.5%	64.0%
Moran, Kansas (Allen County)	30.7°Be 0.871	15.0%	17.5%	. 74 5%
Kentucky (Wayne County)	37.7°Be' 0 835	28.0%	21.0%	
Wyoming (Lander County)	24.0°Be/ 0.909	13.0%	13.0%	
Ranger, Texas.	39.2°Be/ 0.829	30.0%	25.0%	84.0%
Burkburnett, Texas	40.1°Be' 0 8~4	41.0%	20.0%	83.5%
Pine Island, Louisiana	25.4°Be/ 0.902	0.0%	25.0%	57.0%
West Virginia, Cabin Creek	48 0° Be'	36.0%	24.0%	86.0%



Diagrammatic Proximate Composition of Crude Petroleum as to Gasoline and Kerosene.

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF BURKBURNETT, TEXAS, CRUDE OIL.

Gravity of % Temp. Gravity of Gravity of °F Fraction Total Over Stream 0 121 $0.686 = 74.8^{\circ} \text{Be'}$ 179 $0.693 = 72.7^{\circ} \text{Be'}$ $0.686 = 74.8^{\circ} \text{Be'}$ $0.686 = 74.8^{\circ} \text{Be'}$ 5 197 $0.701 = 70.4^{\circ} \text{Be'}$ 211 $0.710 = 67.8^{\circ} \text{Be'}$ $0.701 = 70.4^{\circ} \text{Be}'$ $0.693 = 72.7^{\circ} \text{Be'}$ 10 227 $0.720 = 65.0^{\circ} \text{Be'}$ 238 $0.729 = 62.6^{\circ} \text{Be'}$ $0.738 = 60.2^{\circ} \text{Be'}$ $0.702 = 70.1^{\circ}\text{Be'}$ $0.720 = 65.0^{\circ} \text{Be'}$ 15 253 268 $0.744 = 58.7^{\circ} \text{Be'}$ 20 283 $0.738 = 60.2^{\circ} \text{Be}'$ $0.711 = 67.5^{\circ} \text{Be'}$ $0.751 = 56.9^{\circ} \text{Be'}$ 295 $0.756 = 55.7^{\circ} \text{Be'}$ $0.719 = 65.3^{\circ} \text{Be'}$ 25 309 $0.751 = 56.9^{\circ} \text{Be'}$ $0.762 = 54.2^{\circ}\text{Be'}$ $0.769 = 52.5^{\circ}\text{Be'}$ 322 30 342 $0.762 = 54.2^{\circ} \text{Be}'$ $0.726 = 63.4^{\circ} \text{Be}'$ $0.776 = 50.8^{\circ} \text{Be'}$ 358 $0.782 = 49.4^{\circ} \text{Be'}$ $0.776 = 50.8^{\circ} \text{Be}'$ $0.733 = 61.6^{\circ} \text{Be'}$ 375 35 $0.789 = 47.8^{\circ} \text{Be}'$ 394 $0.789 = 47.8^{\circ} \text{Be}'$ $0.795 = 46.5^{\circ} \text{Be}'$ 40 410 $0.740 = 59.7^{\circ} \text{Be}'$ $0.801 = 45.2^{\circ} \text{Be'}$ 426 $0.807 = 43.8^{\circ} \text{Be'}$ $0.801 = 45.2^{\circ} \text{Be'}$ $0.747 = 57.9^{\circ} \text{Be'}$ 45 440 $0.813 = 42.5^{\circ} \text{Be'}$ 470 $0.754 = 56.2^{\circ} \text{Be'}$ $0.819 = 41.3^{\circ} \text{Be'}$ $0.813 = 42.5^{\circ} \text{Be}'$ 50 485 $0.825 = 40.0^{\circ} \text{Be'}$ $0.829 = 39.2^{\circ} \text{Be'}$ 508 $0.760 = 54.7^{\circ} \text{Be'}$ 55 529 $0.825 = 40.0^{\circ} \text{Be}'$ $0.834 = 38.2^{\circ} \text{Be'}$ 547 $0.838 = 37.4^{\circ} \text{Be'}$ $0.834 = 38.2^{\circ} \text{Be}'$ $0.766 = 53.2^{\circ} \text{Be}'$ 60 562 $0.842 = 36.6^{\circ} \text{Be'}$ 574 0.846 = 35.8°Be' 0.861 = 32.8°Be' 65 $0.842 = 36.6^{\circ} \text{Be'}$ $0.854 = 34.2^{\circ} \text{Be'}$ $0.772 = 51.8^{\circ} \text{Be'}$ 578 $0.785 = 48.7^{\circ} \text{Be'}$ 70 steam $0.868 = 31.5^{\circ} \text{Be'}$ $0.791 = 47.3^{\circ} \text{Be'}$ $0.877 = 29.8^{\circ} \text{Be'}$ 75 $0.797 = 46.0^{\circ} \text{Be'}$ 66 $0.887 = 28.0^{\circ} \text{Be'}$ $0.898 = 26.0^{\circ} \text{Be'}$ 80 $0.910 = 24.0^{\circ} \text{Be'}$ $0.803 = 44.7^{\circ} \text{Be'}$ 66 $0.913 = 23.4^{\circ} \text{Be}'$ 85 90 residue $0.916 = 22.9^{\circ} \text{Be}'$ $0.809 = 43.4^{\circ} \text{Be}'$

Specific Gravity, 0.821; 'Be' U. S., 40.5'; 'Be' Tag, 40.9'. Summary: 59.7'Be' Gasoline = 40.0%; 40.5'Be' Kerosene = 25.0%.

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF BIXBY, OKLA., CRUDE OIL.

Specific Gravity, 0.845; °Be' U. S., 35.7° ; °Be' Tag, 36.0° . Summary: 60.7° Be' Gasoline = 25.0%; 41.0° Be' Kerosene = 20.0%.

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
0	121			
5	173 213	0.695 = 72.1°Be'	0.695 = 72.1°Be'	$0.695 = 72.1^{\circ}\text{Be'}$ $0.712 = 67.2^{\circ}\text{Be'}$
10	253 274 293	0.729 = 62.6°Be'	0.712 = 67.2°Be'	$0.729 = 62.6^{\circ}Be'$ $0.734 = 61.2^{\circ}Be'$
15	293 309 337	$0.752 = 56.6^{\circ} \text{Be'}$	$0.725 = 63.6^{\circ} \text{Be'}$	$0.752 = 56.6^{\circ}Be'$ $0.761 = 54.4^{\circ}Be'$
20	370 391	0.770 = 52.2°Be'	0.736 = 60.7°Be'	$0.770 = 52.2^{\circ}Be'$ $0.778 = 50.3^{\circ}Be'$
25	403 437	0.787 = 48.3°Be'	0.746 = 58.1°Be'	$0.787 = 48.3^{\circ}Be'$ $0.794 = 46.7^{\circ}Be'$ $0.802 = 44.7^{\circ}Be'$
30	447	$0.802 = 44.9^{\circ} \text{Be'}$	$0.755 = 55.9^{\circ} \text{Be'}$	$0.802 \equiv 44.7$ Be $0.807 \equiv 43.8$ Be' $0.813 \equiv 42.5$ Be'
35	490 512	$0.813 = 42.5^{\circ} \text{Be'}$	$0.764 = 53.7^{\circ} \text{Be'}$	$0.819 = 41.2^{\circ} \text{Be'}$
40	534	0.826 = 39.8°Be'	$0.771 = 52.0^{\circ} \text{Be'}$	$0.826 = 39.8^{\circ}Be'$ $0.830 = 38.9^{\circ}Be'$
45	567	0.835 = 37.9°Be'	0.778 = 50.3°Be'	$0.835 = 37.9^{\circ}Be'$ $0.838 = 37.3^{\circ}Be'$ $0.842 = 36.5^{\circ}Be'$
50	600 steam	$0.842 = 36.5^{\circ}\text{Be'}$	$0.785 = 48.7^{\circ} \mathrm{Be'}$	$0.842 \equiv 30.5^{\circ}$ Be $0.848 \equiv 35.3^{\circ}$ Be' $0.855 \equiv 34.0^{\circ}$ Be'
55	44 44	0.855 = 34.0°Be'	$0.791 = 47.3^{\circ} \text{Be'}$	0.855 = 34.0 Be $0.860 = 33.0^{\circ}$ Be' $0.865 = 32.0^{\circ}$ Be'
60	44	$0.865 = 32.0^{\circ} \text{Be'}$	$0.797 = 46.0^{\circ} \mathrm{Be'}$	0.803 ± 32.0 Be $0.871 \pm 30.9^{\circ}$ Be' $0.878 \pm 29.6^{\circ}$ Be'
65	44 44	$0.878 = 29.6^{\circ} \mathrm{Be'}$	$0.803 = 44.7^{\circ} \text{Be'}$	0.878 = 23.0 Be' $0.884 = 28.5^{\circ}\text{Be'}$ $0.890 = 27.4^{\circ}\text{Be'}$
70	46 46	$0.890 \equiv 27.4^{\circ} \mathrm{Be'}$	$0.809 = 43.4^{\circ}\text{Be'}$	0.890 = 21.4 Be 0.894 = 26.7°Be' 0.899 = 25.9°Be'
75	-4	$0.899 = 25.9^{\circ} \text{Be'}$	$0.815 = 42.1^\circ \mathrm{Be'}$	$0.899 = 25.9^{\circ} \text{Be}'$ $0.903 = 25.2^{\circ} \text{Be}'$ $0.907 = 24.5^{\circ} \text{Be}'$
80	44 46	$0.907 = 24.5^{\circ} \text{Be'}$	$0.820 = 41.0^{\circ} \text{Be'}$	$0.901 = 23.8^{\circ}Be'$ $0.911 = 23.8^{\circ}Be'$ $0.915 = 23.1^{\circ}Be'$
85	**	$0.915 = 23.1^{\circ} \text{Be'}$	$0.827 = 39.6^{\circ} \text{Be'}$	0.913 = 23.1 Be $0.919 = 22.5^{\circ}\text{Be'}$ $0.923 = 21.8^{\circ}\text{Be'}$
90	**	$0.923 = 21.8^{\circ} \text{Be'}$	$0.833 = 38.4^{\circ}\text{Be'}$	
95-100	residue	$0.953 = 17.0^{\circ} \text{Be'}$		

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF CUSHING, OKLA., CRUDE OIL.

Specific Gravity, 0.824; °Be' U. S., 39.9°; °Be' Tag, 40.2°. Summary: 60.2°Be' Gasoline = 35.0%; 42.1°Be' Kerosene = 25.0%.

%	Temp. °F	Gravity of Fraction	Gravity of Total over	Gravity of Stream
0	130			
5	$\begin{array}{c} 163 \\ 179 \end{array}$	$0.685 = 75.0^{\circ} \text{Be'}$	$0.685 = 75.0^{\circ} \text{Be'}$	$0.685 \equiv 75.0^{\circ}\text{Be'}$ $0.695 \equiv 72.1^{\circ}\text{Be'}$
10	$\begin{array}{c} 205 \\ 229 \end{array}$	$0.706 = 68.9^{\circ} \text{Be'}$	$0.695 = 72.1^{\circ} \mathrm{Be'}$	$0.706 = 68.9^{\circ}Be'$ $0.716 = 66.1^{\circ}Be'$
15	$250 \\ 270$	$0.727 = 63.1^{\circ} \text{Be'}$	$0.706 = 68.9^{\circ} \text{Be'}$	$0.727 = 63.1^{\circ}\text{Be'}$ $0.736 = 60.7^{\circ}\text{Be'}$
20	$\begin{array}{c} 283 \\ 297 \end{array}$	$0.745 = 58.4^{\circ} \text{Be'}$	$0.715 = 66.4^{\circ} \text{Be'}$	$0.745 \equiv 58.4^{\circ}\text{Be'}$ $0.751 \equiv 56.9^{\circ}\text{Be'}$
25	$\begin{array}{c} 316\\327\end{array}$	$0.757 = 55.4^{\circ} \text{Be'}$	$0.724 = 63.9^{\circ} \mathrm{Be'}$	$0.757 \equiv 55.4^{\circ}\text{Be'}$ $0.762 \equiv 54.2^{\circ}\text{Be'}$
30 '	339 352	$0.768 = 52.7^{\circ} \mathrm{Be'}$	$0.731 = 62.0^{\circ} \mathrm{Be'}$	$0.768 = 52.7^{\circ} \text{Be'}$ $0.774 = 51.3^{\circ} \text{Be'}$
35	372 394	$0.780 = 49.9^{\circ} \mathrm{Be'}$	$0.738 = 60.2^\circ\mathrm{Be'}$	$0.780 = 49.9^{\circ} \text{Be'}$ $0.786 = 48.5^{\circ} \text{Be'}$
40	$414 \\ 427 \\ 447$	$0.793 = 46.9^\circ\mathrm{Be'}$	$0.745 = 58.4^{\circ}\mathrm{Be'}$	$0.793 = 46.9^{\circ}Be'$ $0.799 = 45.6^{\circ}Be'$
45	447 460 481	$0.805 = 44.2^{\circ} \mathrm{Be'}$	$0.751 = 56.9^\circ\mathrm{Be'}$	$0.805 = 44.2^{\circ}\text{Be'}$ $0.810 = 43.2^{\circ}\text{Be'}$
50	$507 \\ 523$	$0.816 = 41.9^{\circ} \mathrm{Be'}$	$0.758 = 55.1^\circ\mathrm{Be'}$	$0.816 = 41.9^{\circ}\text{Be'}$ $0.822 = 40.5^{\circ}\text{Be'}$ $0.826 = 39.4^{\circ}\text{Be'}$
55	542 559	$0.823 = 39.4^{\circ} \mathrm{Be'}$	$0.764 = 53.7^\circ\mathrm{Be'}$	$0.826 \equiv 39.4^{\circ} \text{Be}'$ $0.832 \equiv 38.5^{\circ} \text{Be}'$ $0.837 \equiv 37.4^{\circ} \text{Be}'$
60	588 steam	$0.837 = 37.4^{\circ} \mathrm{Be'}$	$0.770 \equiv 52.2^\circ\mathrm{Be'}$	0.837 = 37.4 Be $0.842 = 36.5^{\circ}$ Be' $0.847 = 35.5^{\circ}$ Be'
65	5000111 (i	$8.847 \equiv 35.5^\circ\mathrm{Be'}$	$0.779 \equiv 50.1^\circ\mathrm{Be'}$	0.847 = 35.5 Be $0.857 = 33.6^{\circ}$ Be' $0.867 = 31.7^{\circ}$ Be'
70		$0.867 = 31.7^{\circ} \mathrm{Be'}$	$0.785 \equiv 48.7^\circ\mathrm{Be'}$	0.807 = 31.7 Be $0.875 = 30.2^{\circ}$ Be' $0.884 = 28.5^{\circ}$ Be'
75	**	$0.884 = 28.5^{\circ} \mathrm{Be'}$	$0.792 \equiv 47.1^\circ\mathrm{Be'}$	$0.890 = 27.4^{\circ}Be'$ $0.896 = 26.4^{\circ}Be'$
80	66 66	$0.896 = 26.4^{\circ} \text{Be'}$	$0.798 \equiv 45.8^{\circ}\mathrm{Be'}$	0.890 = 20.4 Be $0.907 = 24.5^{\circ}$ Be' $0.909 = 24.1^{\circ}$ Be'
85	66 66	$0.909 = 24.1^{\circ} \text{Be'}$	$0.805 = 44.2^{\circ} \mathrm{Be'}$	$0.916 = 22.9^{\circ} \text{Be'}$ $0.924 = 21.6^{\circ} \text{Be'}$
90	· · · · · · · · · · · · · · · · · · ·	$0.924 = 21.6^{\circ} \mathrm{Be'}$	$0.811 = 42.9^{\circ} \text{Be'}$	$0.930 = 20.7^{\circ} \text{Be}'$
95-100	residue	$0.940 = 19.0^{\circ} \mathrm{Be'}$		

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF PINE ISLAND. NO. LOUISIANA, CRUDE OIL.

Specific Gravity, 0.902; °Be' U. S., 25.2°; °Be' Tag, 25.4°Be'. Summary: Gasoline = none; 31.0° Kerosene = 25.0% (Naphthene base oil)

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream	
0	365				
5	471	$0.839 = 37.2^{\circ} \text{Be}'$	$0.839 = 37.2^{\circ} \text{Be}'$	$0.849 = 35.1^{\circ} \text{Be'}$	
10	500	$0.860 = 33.0^{\circ} \text{Be}'$	$0.849 = 35.1^{\circ} \text{Be}'$	$0.864 = 32.2^{\circ} \text{Be}'$	
15	530	$0.869 = 31.3^{\circ} \text{Be}'$	0.856 = 33.8°Be'	$0.872 = 30.7^{\circ} \text{Be}'$	
20	549	$0.876 = 30.0^{\circ} \text{Be}'$	$0.861 = 32.8^{\circ} \text{Be'}$		
25	564	$0.880 = 29.3^{\circ} \text{Be}'$	$0.865 = 32.0^{\circ} \text{Be}'$	$0.881 = 29.1^{\circ} \text{Be}'$	
30	589	$0.883 = 28.8^{\circ} \text{Be'}$	$0.867 = 31.7^{\circ} \text{Be}'$	$0.886 = 28.2^{\circ}Be'$	Viscosity
35	steam	$0.889 = 27.7^{\circ} \text{Be}'$	$0.870 = 31.1^{\circ} \text{Be}'$	$0.890 = 27.4^{\circ} \text{Be}'$	
40	66	$0.892 = 27.1^{\circ} \text{Be}'$	$0.873 = 30.5^{\circ} \text{Be'}$	$0.893 = 26.9^{\circ} \text{Be'}$	
45	66	$0.894 = 26.8^{\circ} \text{Be}'$	$0.875 = 30.2^{\circ} \text{Be}'$	$0.894 = 26.7^{\circ} \text{Be}'$	
50	66	$0.895 = 26.6^{\circ} \text{Be'}$	$0.877 = 29.8^{\circ} \text{Be}'$	$0.895 = 26.6^{\circ} \text{Be'}$	
55	66	$0.896 = 26.4^{\circ} \text{Be'}$	$0.879 = 29.4^{\circ} \text{Be}'$	$0.896 = 26.4^{\circ} \text{Be}'$	
60	66	$0.897 = 26.2^{\circ} \text{Be'}$	$0.880 = 29.3^{\circ} \text{Be}'$		
65	66	$0.897 = 26.2^{\circ} \text{Be}'$	$0.880 = 29.3^{\circ} \text{Be}'$	$0.897 = 26.2^{\circ} \text{Be}'$	625
70	66	$0.897 = 26.2^{\circ} \text{Be}'$	$0.880 = 29.3^{\circ} \text{Be}'$		
75	66 -	$0.897 = 26.2^{\circ} \text{Be}'$	$0.880 = 29.3^{\circ} \text{Be'}$	$0.898 = 26.0^{\circ} \text{Be}'$	
80	66	$0.899 = 25.9^{\circ} \text{Be}'$	$0.885 = 28.3^{\circ} \text{Be'}$		
85	66	$0.900 = 25.7^{\circ} \text{Be}'$	$0.885 = 28.3^{\circ} \text{Be'}$	$0.901 = 25.5^{\circ} \text{Be'}$	
90	66	$0.902 = 25.4^{\circ}Be'$	$0.886 = 28.2^{\circ} \text{Be}'$	0.001 - 20.0 DC	waxy

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF BILLINGS,

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
0	116			
	169			$0.679 = 76.9^{\circ}Be$
5	191	$0.679 = 76.9^{\circ} \text{Be'}$	$0.679 = 76.9^{\circ} \text{Be'}$	$0.689 = 73.8^{\circ} \text{Be}$
10	207 223	0.700 = 70.6°Be'	0.689 = 73.8°Be'	$0.700 = 70.6^{\circ}Be$ $0.710 = 67.8^{\circ}Be$
10	235	0.100 - 10.0 De	0.009 - 10.0 De	0.710 = 67.8 Be
15	252	$0.720 = 65.0^{\circ} \text{Be}'$	0.699 = 70.9°Be'	$0.728 = 62.8^{\circ}Be$
	264	00000 - 0000 20	0.000 - 1000 20	$0.736 = 60.7^{\circ}Be$
20	277	$0.736 = 60.7^{\circ} \text{Be'}$	$0.708 = 68.3^{\circ} \text{Be'}$	$0.742 = 59.2^{\circ} \text{Be}$
	286	and a supervision of the supervi		$0.748 = 57.6^{\circ} \text{Be}$
25	303	0.748 = 57.6°Be'	$0.716 = 66.1^{\circ}\text{Be'}$	$0.753 = 56.4^{\circ}Be$
30	317	$0.761 = 54.4^{\circ} \text{Be'}$	$0.724 = 63.9^{\circ}Be'$	$0.761 = 54.4^{\circ}Be$
90	337 353	0.701 = 04.4 De	0.724 = 03.9 De	$0.767 = 52.9^{\circ}Be$ $0.774 = 51.3^{\circ}Be$
35	367	$0.774 = 51.3^{\circ}Be'$	$0.731 = 62.0^{\circ} \text{Be'}$	$0.779 = 50.1^{\circ}Be$
	381	0.111 = 01.0 DC	0.101 - 02.0 DC	$0.785 = 48.7^{\circ}Be$
40	396	$0.785 = 48.7^{\circ} \text{Be'}$	0.737 = 60.5°Be'	$0.790 = 47.6^{\circ}Be$
	413			$0.795 = 46.3^{\circ}Be$
45	431	$0.795 = 46.3^{\circ} \text{Be'}$	0.744 = 58.7°Be'	$0.801 = 45.1^{\circ}Be$
50	456			$0.808 = 43.6^{\circ}Be$
90	482 500	$0.808 = 43.6^{\circ} \text{Be'}$	0.750 = 57.1°Be'	$0.814 = 42.3^{\circ}Be$ $0.820 = 41.0^{\circ}Be$
55	513	$0.820 = 41.0^{\circ} \text{Be'}$	$0.756 = 55.6^{\circ} \text{Be'}$	$0.820 = 41.0^{\circ}Be$ $0.825 = 40.0^{\circ}Be$
	530	0.020 = 41.0 De	0.100 - 00.0 De	$0.820 = 38.9^{\circ}Be$
60	550	$0.830 = 38.9^{\circ} \text{Be'}$	0.763 = 53.9°Be'	$0.835 = 37.9^{\circ}Be$
	577		The Constant of Constant of Constant	$0.840 = 36.9^{\circ}Be$
65	593	$0.840 = 36.9^{\circ} \text{Be'}$	$0.768 = 52.7^{\circ} \text{Be'}$	$0.843 = 36.3^{\circ}Be$
-	steam			$0.847 = 35.5^{\circ}Be$
70	66	$0.847 = 35.5^{\circ} \text{Be'}$	0.774 = 51.3°Be'	$0.855 = 34.0^{\circ}Be$
- 75 -		0.864 = 32.3°Be'	$0.780 = 49.9^{\circ} \text{Be}'$	$0.864 = 32.3^{\circ}Be$ $0.870 = 31.1^{\circ}Be$
	66	0.001 - 02.0 De	0.100 - 19.9 De	$0.878 = 29.6^{\circ}Be$
80	66	$0.878 = 29.6^{\circ} \text{Be'}$	0.786 = 48.5°Be'	$0.886 = 28.2^{\circ}Be$
85	**	$0.893 = 27.0^{\circ} \text{Be'}$	0.792 = 47.2°Be'	$0.900 = 25.7^{\circ}Be$
90	66	$0.906 = 24.7^{\circ} \text{Be'}$	$0.798 = 45.8^{\circ} \text{Be'}$	$0.918 = 22.6^{\circ}Be$
5-100	residue	$0.930 = 20.7^{\circ} \text{Be'}$		

OKLA., CRUDE OIL. Specific Gravity, 0.812; 'Be' U. S. 42.4° ; 'Be' Tag 42.8° . Summary: 60.5° Be' Gasoline = 40.0%; 41.0° Be' Kerosene = 25.0%.

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF GARBER, OKLA., CRUDE OIL.

Specific Gravity, 0.780; °Be' U. S. 49.5°; °Be' Tag 49.9°. Sulphur = 0.05%. Summary: 59.2°Be' Gasoline = 55.0%; 40.5°Be' Kerosene = 20.0%.

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream	Olefins %
05	110			$0.670 = 79.7^{\circ}\text{Be'}$	1.0
10 15	188	$0.675 = 78.1^{\circ}\text{Be'}$	$0.675 = 78.1^{\circ}\text{Be'}$ $0.684 = 75.3^{\circ}\text{Be'}$	$0.694 = 72.4^{\circ} \text{Be'}$	1.0
20 25	226	$0.712 = 67.2^{\circ} \text{Be'}$		$0.726 = 63.4^{\circ} \text{Be'}$	1.0
30 35	264	0.739 = 59.9°Be'		$0.748 = 57.6^{\circ} \text{Be'}$	1.2
40 45	322 350	$0.757 = 55.4^{\circ} \text{Be'}$		$0.769 = 52.5^{\circ} \text{Be'}$	1.4
50 55	380 400		$0.733 \equiv 61.5^{\circ}\text{Be'}$ $0.742 \equiv 59.2^{\circ}\text{Be'}$	$0.793 = 46.9^{\circ} \text{Be'}$	1.5
60 65	420	$0.806 = 44.0^{\circ} \text{Be}'$	$0.745 \equiv 58.4^{\circ}\text{Be'}$ $0.751 \equiv 56.9^{\circ}\text{Be'}$	$0.821 = 40.8^{\circ} \text{Be}'$	1.7
70 75	550	$0.830 = 38.9^{\circ} \text{Be'}$	$0.757 \equiv 55.4^{\circ}\text{Be'}$ $0.763 \equiv 53.9^{\circ}\text{Be'}$	$0.850 = 34.9^{\circ}Be'$	
80 85		$0.855 = 34.0^{\circ} \text{Be'}$	$0.769 \equiv 52.5^{\circ} \text{Be'}$ $0.774 \equiv 51.3^{\circ} \text{Be'}$	$0.865 = 32.0^{\circ} \text{Be}'$	
90 95	4%		$0.779 = 50.1^{\circ} \text{Be'}$	$0.870 = 31.1^{\circ} \text{Be'}$	3.0

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF RANGER, TEX., CRUDE OIL. Specific Gravity, 0.829; °Be', U. S. 38.9°; °Be' Tag 39.2°. Summary: 57.4°Be' Gasoline = 30.0%; 41.1°Be'; Kerosene = 30.0%.

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
0 5	154	$0.705 = 69.2^{\circ}\text{Be}'$	$0.705 = 69.2^{\circ} \text{Be'}$	$0.717 = 65.8^{\circ}\text{Be}'$
10	$\begin{array}{c} 239 \\ 268 \end{array}$	0.703 ± 69.2 Be 0.729 ± 62.5 Be'	0.705 ± 69.2 Be 0.717 ± 65.8 °Be'	$0.717 = 60.5^{\circ}$ Be:
15	208	$0.745 = 58.4^{\circ}Be'$	$0.726 = 63.4^{\circ} \text{Be'}$	$0.752 = 56.6^{\circ} \text{Be'}$
20	325	$0.759 = 55.2^{\circ}Be'$	$0.734 = 61.3^{\circ} \text{Be'}$	$0.765 = 53.4^{\circ} \text{Be'}$
25	362	$0.771 = 52.0^{\circ} \text{Be'}$	$0.742 = 59.2^{\circ} \text{Be'}$	$0.777 = 50.6^{\circ} \text{Be'}$
30	390	$0.783 = 49.2^{\circ} \text{Be'}$	$0.749 = 57.4^{\circ} \text{Be}'$	$0.789 = 47.8^{\circ} \text{Be}'$
35	423	$0.796 = 46.4^{\circ}Be'$	$0.755 = 55.9^{\circ} \text{Be}'$	$0.800 = 45.3^{\circ} \text{Be}'$
40	460	$0.805 = 44.2^{\circ} \text{Be'}$	$0.762 \equiv 54.2^{\circ} \text{Be'}$	$0.811 = 42.9^{\circ} \text{Be'}$
45	494	$0.817 = 41.7^{\circ} \text{Be'}$	$0.768 = 52.7^{\circ} \text{Be}'$	$0.822 = 40.6^{\circ} \text{Be}'$
50	528	$0.827 = 39.6^{\circ} \text{Be'}$	$0.774 = 51.3^{\circ} \text{Be'}$	$0.831 = 38.7^{\circ} \text{Be}'$
55	558	$0.835 \equiv 38.0^{\circ} \text{Be'}$	$0.779 = 50.1^{\circ} \text{Be}'$	$0.837 \equiv 37.5^{\circ} \text{Be}'$
60	582	$0.840 = 37.0^{\circ} \text{Be'}$	$0.784 = 49.0^{\circ} \text{Be'}$	$0.843 = 36.3^{\circ} \text{Be}'$
65	vacuum	$0.851 = 34.8^{\circ} \text{Be'}$	$0.789 = 47.8^{\circ} \text{Be'}$	$0.858 = 33.4^{\circ} \text{Be}'$
70	66	$0.865 = 32.1^{\circ} \text{Be'}$	$0.795 = 46.4^{\circ} \text{Be}'$	$0.870 = 31.1^{\circ} \text{Be}'$
75	66	$0.875 = 30.2^{\circ} \text{Be'}$	$0.800 = 45.3^{\circ} \text{Be'}$	$0.880 = 29.3^{\circ} \text{Be'}$
- 80	- 64	$0.885 = 28.4^{\circ} \text{Be'}$	$0.805 = 44.2^{\circ} \text{Be'}$	$0.890 = 27.4^{\circ} \text{Be'}$
85	66	$0.895 = 26.6^{\circ} \text{Be'}$	$0.810 = 43.2^{\circ} \text{Be'}$	$0.897 = 26.2^{\circ} \text{Be}'$
90	56	$0.900 = 25.7^{\circ} \text{Be'}$	$0.815 = 42.1^{\circ} \text{Be'}$	$0.923 = 21.8^{\circ} \text{Be'}$
95-100	residue	$0.947 = 17.9^{\circ}Be'$	$0.823 = 40.4^{\circ} \text{Be'}$	0.020 - 21.0 BC

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF KENTUCKY CRUDE OIL. Specific Gravity, 0.8415; °Be' U. S. 36.7°; °Be' Tag. Summary: Gasoline = 27.5%; Kerosene = 22.5%.

6

%	Temp. °F	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
0	186			
5	259	0.720	$0.720 = 65.0^{\circ} \text{Be'}$	$0.728 = 63.8^{\circ} \text{Be'}$
10	285	0.737	$0.728 = 63.8^{\circ} \text{Be'}$	$0.742 = 59.2^{\circ} \text{Be'}$
15	310	0.748	$0.735 = 61.0^{\circ} \text{Be}'$	$0.755 = 55.9^{\circ} \text{Be}'$
20	342	0.762	$0.741 = 59.4^{\circ} \text{Be'}$	$0.769 = 52.5^{\circ} \text{Be'}$
25	387	0.776	$0.748 = 57.6^{\circ} \text{Be'}$	$0.784 = 48.9^{\circ} \text{Be}'$
30	422	0.793	$0.756 \equiv 55.6^{\circ} \text{Be'}$	$0.796 = 46.2^{\circ} \text{Be}'$
35	458	0.800	$0.762 = 54.2^{\circ} \text{Be'}$	$0.805 = 44.2^{\circ} \text{Be}'$
40	502	0.811	$0.768 \equiv 52.7^{\circ} \text{Be}'$	$0.816 = 41.9^{\circ} \text{Be}'$
45	542	0.822	$0.774 = 51.3^{\circ} \text{Be'}$	$0.827 = 39.6^{\circ} \text{Be}'$
. 50	582	0.833	$0.780 = 49.9^{\circ} \text{Be'}$	$0.839 = 37.1^{\circ} \text{Be}'$
55	600	0.845	$0.786 = 48.5^{\circ} \text{Be'}$	$0.849 = 35.1^{\circ} \text{Be}'$
60	600	0.853	$0.791 = 47.3^{\circ} \text{Be}'$	$0.857 = 33.6^{\circ} \text{Be'}$
65	600	0.862	$0.797 = 46.0^{\circ} \text{Be'}$	$0.871 = 30.9^{\circ} \text{Be}'$
70	600	0.880	$0.803 \equiv 44.7^{\circ} \text{Be'}$	$0.887 = 28.0^{\circ} \text{Be'}$
75	600	0.895	$0.809 \equiv 43.4^{\circ}\text{Be'}$	$0.919 = 22.4^{\circ} \text{Be}'$
80	residue	0.943	$0.817 = 41.7^{\circ} \text{Be'}$	

FRACTIONAL GRAVITY DISTILLATION OF CRUDE OIL FROM EASTERN ALLEN COUNTY, MORAN, KAN.

		Sp. $Gr. =$.8775					
Be. $Gr. = 29.7$								
Per	Temp.	Gravity	Gravity of	Gravity				
Cent	°F	of Fraction	Total Dist.	of Stream				
0 5								
5	342	.753	.753 = 56.4	.762 = 54.2				
10	384	.771	.762 = 54.2	.778 = 50.3				
15	422	.788	.770 = 52.2	.796 = 46.2				
20	459	.804	.779 = 50.1	.810 = 43.2				
25	490	.816	.786 = 48.5	.822 = 40.6				
30	529	.829	.792 = 47.1	.835 = 37.9				
35	562	.840	.800 = 45.3	.844 = 36.2				
40	592	.849	.806 = 44.0	.853 = 34.3				
45	600	.858	.812 = 42.7	.863 = 32.4				
50	600	.868	.817 = 41.7	.874 = 30.4				
55	600	.881	.823 = 40.4	.886 = 28.2				
60	600	.891	.829 = 39.1	.896 = 26.4				
65	600	.901	.834 = 38.1	.904 = 25.0				
70	600	.907	.839 = 37.1	.912 = 23.6				
75	620	.921	.845 = 35.9	.946 = 18.0				
Residue		.972	.853 = 34.3					
			and the second se					

Chemical Constitution of Petroleum

Petroleum is composed of carbon and hydrogen in chemical combination known as hydrocarbons. In conjunction with the carbon and hydrogen there frequently is oxygen, nitrogen and sulphur in much smaller amounts.

In crude oils the amount of carbon varies from 80 to 89%, the hydrogen from 10 to 15%, oxygen from 0.0 to 5.0%, nitrogen from 0.0 to 1.8% and sulphur from .01 to 5.0%.

Typical ultimate analyses of	f petroleum	products	are as fo	llows:
	Hydrogen			Oxygen
Pennsylvania Crude86.06%	13.88%	0.06%	0.00%	0.00%
Texas Crude	12.30	1.75	0.70	0.00
California Crude84.00	12.70	0.75	1.70	1.20
Mexican Crude	10.20	4.15		
Oklahoma Crude85.70	13.11	0.40	0.30	
Kas. Crude (Towanda).84.15	13.00	1.90	0.45	
Kansas Residuum85.51	11.88	0.71	0.32	0.63
Healdton (Okla.) Crude.85.00	12.90	0.76		
Kansas Air Blown				
Residuum	10.39	0.42	0.21	4.61
Byerlite Pitch	9.97	0.55	0.29	1.58
Grahamite	7.50	2.00	0.20	
Trinidad Asphalt82.60	10.50	6.50	0.50	
Commercial Gasoline84.27	15.73	0.00	0.00	0.00
Kerosene	15.26	0.01	0.00	0.00
Lubricating Oil	14.87	0.01		
(Paraffin)				
Lubricating Oil	12.51	0.01		
(Naphthene)				t:
Benzol	7.76	0.00	0.00	0.00 17

Paraffin $(C_{n}H_{2n+2})$ hydrocarbons largely compose the light or more volatile constituents of all petroleum. They are "saturated" hydrocarbons and have a very low ration of specific gravity to distilling temperature, are not acted upon by concentrated sulphuric acid or by fuming sulphuric acid (oleum), are not nitrated by nitric acid and are extremely resistant to all chemical reactions. The chief differences in petroleum are in the heavy constituents, the heavy hydrocarbons of the paraffin series being found chiefly in Pennsylvania and some Mid-Continent oils.

Naphthenes (C_nH_{2n}) ring or cyclic compounds are less common hydrocarbons in lighter portions of petroleum, but commonly found as heavy hydrocarbons of petroleum. They have a higher ratio of specific gravity to distilling temperature than the paraffin comopunds, are resistant to the action of sulphuric acid and some types may be distinguished by the "formolit" reaction. Oils containing light naphthenes are found in Russia and Louisiana. All heavy oils contain naphthenes.

C_n H_{2n} (NAPHTHENES) POLYMETHYLENE SERIES.

		Boiling	
	Formula	Temperature	Gravity
Cyclopropane	CaHa	$-35^{\circ}C = -31^{\circ}F$	
Cyclobutane	C4H8	$+12^{\circ}C = 54^{\circ}F$	$.709 = 67.5^{\circ} \text{Be}'$
Cyclopentane	C5H10	$49^{\circ}C = 120^{\circ}F$	$.769 = 52.1^{\circ} \text{Be}'$
Cyclohexane	C6H12	$81^{\circ}C = 178^{\circ}F$	$.799 = 45.2^{\circ} \text{Be}'$
Cycloheptane	C7H14	$117^{\circ}C = 243^{\circ}F$	$.089 = 43.1^{\circ} \text{Be}'$
Methyl Cyclopentane	C6H12	$72^{\circ}C = 162^{\circ}F$	$.766 = 52.8^{\circ} \text{Be'}$
Dimethyl Cyclopentane	C7H14	$91^{\circ}C = 196^{\circ}F$	$.778 = 50.0^{\circ} \text{Be}'$
Methyl Cyclohexane	C7H14	$98^{\circ}C = 208^{\circ}F$	$.778 = 50.0^{\circ} \text{Be}'$
Dimethyl Cyclohexane	C.H.16	$118^{\circ}C = 244^{\circ}F$	$.781 = 49.3^{\circ} \text{Be}'$
Trimethyl Cylohexane	C9H18	$198^{\circ}C = 388^{\circ}F$	$.787 = 47.9^{\circ} \text{Be}'$

Aromatic or Benzene Hydrocarbons (C_nH_{2n-6}) exist to some extent in certain California petroleums and have a very high ratio of specific gravity to distilling temperature. Gasoline made from the California petroleum is heavier than light gasoline with the same end point made from Mid-Continent petroleum. The aromatic compounds are acted upon by nitric acid forming nitro products. They are formed from paraffin and naphthene hydrocarbons by pyrogenic decomposition at temperatures above 1000°F. The production of aromatic compounds from petroleum has not been commercially satisfactory on account of incomplete conversion and difficulty of freeing from paraffin hydrocarbons.

Olefines or Ethylenes (C_nH_{2n}) are "unsaturated" hydrocarbons, ely if ever existing naturally in crude oil but commonly resulting om its exposure to high temperatures. These compounds contain less hydrogen and more carbon than paraffin hydrocarbons and are capable of taking in more hydrogen. They are removed from aromatic compounds, paraffin compounds and naphthene compounds by the tion of concentrated sulphuric acid in the usual process of refining

soline. These hydrocarbons give gasoline, to a large extent, its diseeable odor before refining. Their combination with sulphur gives more intense odor. Each of these groups of hydrocarbons is supposed to exist in a complete series, represented by the general formula given. The paraffin or methane series of "saturated" hydrocarbons

has been fairly well worked out and is given in the following table: According to Hofer, the following olefines have been isolated from "North American" netroleum:

Ethylene	C ₂ H 4	Heptylene	C 7H14	Dodecylene	C12H14
Propylene	C.H.	Octylene	C 8H16	Decatrilene	C18H38
Butylene	C.H.	Nonylene	C 9H18	Cetene	CieHas
Amylene	C5H10	Decylene	C10H20	Cerotene	C27H34
Hexylene	C.H.12	Endecylene	CuHa	Melene	C30H00

If the residue contains much wax, the crude is known as paraffin base oil, but if napthenes or similar hydrocarbons predominate, it is an "asphalt" base oil. Practically the "asphalt" is determined by the solubility of the solid hydrocarbons in pentane and by the gravity and physical character of the residue.

Among the light hydrocarbons of petroleum, either existing naturally or pyrogenically produced, the relation of the specific gravity to the distilling temperature affords a simple and practical method of estimating the amount of olefin, paraffin and aromatic compounds. This relation is set forth in the curves on page 227.

The value of crude oil is not measured by its ultimate analysis or by its "base" so much as by the amount of volatile constituents which it contains. The amount of volatile constituents obtained from various crude oils is shown by the curves on page 121.

Paraffin Hydrocarbons in Petroleum

GASEOUS HYDROCARBONS (Natural Gas)

		Sp. Gr.				
	Baume'			Melting	Boiling M	olecular
Name	Gravity		Formula		Point	Weight
Methane			CH	-184.0°C	-165.0°C	16.03
Ethane	194	0.432	C ₂ H ₆	-171.4	- 93.0	30.05
Propane	142	0.525	C ₃ H ₈	-195.0	- 45.0	44.07
Butane	109	0.585	C H10	-135.0	+ 1.0	58.08
Dutaile	109	0.000	· 41110		+ 1.0	00.00
"GASOLINE"	HYDRO	CARBO	ONS			
Pentane	92.2	0.630	C 5H12		36.3	72.10
Hexane	78.9	0.670	C 6H14		69.0	86.12
Heptane	70.9	0.697	C 7H16		98.4	100.13
Octane	65.0	0.718	C 8H18		125.5	114.15
Nonane	59.2	0.740	$C_{9}H_{20}$	- 51.0	150.0	128.16
Decane	56.7	0.750				
Undecane			C10H22	- 31.0	173.0	242.18
Undecane	54.2	0.760	C11H24	- 26.0	195.0	156.20
HEAVY LIQU	ID HVI	ROCA	RBONS	(Kerosene)		
Duodecane	51.8	0.770	C ₁₂ H ₂₄	-12.0	214.0	170.22
Tridecane	46.8	0.792	$C_{12}H_{28}$ $C_{18}H_{28}$	- 6.0	234.0	184.24
Tetradecane		0.192				
	45.0		C14H30	+ 5.0	252.0	198.25
Pentadecane	43.5	0.807	C15H32	10.0	270.0	212.26
Hexadecane	41.8	0.815	C16H34	28.0	287.0	226.27
Heptadecane	40.3	0.822	C17H36	22.0	295.0	240.28
Octadecane	38.6	0.830	$C_{18}H_{38}$	28.0	317.0	254.30
HEAVY SOLU	D HYDE	ROCAR	BONS		(vacuo)	
Eicosane	37.2	0.837	C20H42	37.0	117.5	282.34
Tricosane	36.5	0.841	C23H48	48.0	138.0	325.38
Tetracosane			C24H30	51.0	145.5	338.39
Pentacosane			C25H52	54.0	152.5	352.41
Hexacosane			C26H34	56.0	160.0	366.43
Mericyl			C27H56	59.4	167.0	370.45
Octocosane			C28H58	60.0	173.5	384.47
Nonocosane			C20 Han	63.0	179.0	398.48
Cervl			$C_{30}H_{62}$	65.6	186.0	422.49
Pentriacontane			C301162	68.0	193.5	436.52
Duotriacontane			C32He6	70.0	201.0	450.52
Tetratriacontan		• • • •	C321100	72.0	215.0	450.55
Pentatriacontar		0.846	Ca5H72			
a ontact lacontal	10 00.4	0.040	0851172	75.0	222.0	492.58

There is no natural petroleum composed exclusively of the paraffin series of hydrocarbons, even Pennsylvania and Garber, Oklahoma, crude oils having members of other series. The main body of the light petroleum is made up of paraffin hydrocarbons and the heavy residues are largely made up of napthenes.

Typical Refinery Practice

There is much variation in the practice of petroleum distillation in different refineries. This depends to a large extent upon the character of the crude oil used, the market to which the refiner sells and the ability of the refiner both as to knowledge and equipment.

The following outlines the progressive distillation and treatment of crude oil in a typical refinery:

1. Crude Benzine (Gasoline and Naphtha) includes all of the light distillate which vaporizes up to 410° F. In the ordinary Mid-Continent or Texas petroleum, 420° F indicates a gravity of the streem of distillate from the condenser in the receiving house of 46.5° Be' to 47.0° Be'. The gravity of the total distillate at this point varies with different types of crude. In some crudes this will be as high as 64.0° gravity, in others as low as 50.0° . For example, referring to pages 122 to 127, Burkburnett crude boiling at 410° F has a gravity of 59.7° of the total benzine and a stream gravity of 58.0° Be' and a stream gravity of 59.7° Be' and a stream gravity of 58.0° Be' and a stream gravity of 58.0° Be' and a stream gravity of 59.7° Be' and a stream gravity of 46.5° Be'; Bixby, Okla., crude benzine at 410° F has a gravity of 59.7° Be' and a stream gravity of 46.7° Be'; Cushing, Okla., crude benzine at 410° F has a gravity of 59.7° Be' and a stream gravity of 47.0° Be', Billings, Okla., crude has a gravity of 60° Be' at 410° F and a stream gravity of 40.7° Be'; Go Be' at 410° F and a stream gravity of 410° F of 56.6° Be' and a stream gravity of 46.7° Be'. The gravity of crude benzine depends upon the initial boiling point of the crude, the relative proportion of the different paraffin constituents and the chemical series of hydrocarbons to which the crude belongs (see page 227.

The crude benzine is run off with direct fire under the still, though after a temperature of 212°F is reached some steam may be put in. The steam decidedly sweetens the product and brings over the benzine at a lower temperature. In the use of steam the distillation must be entirely governed by the gravity of the stream in the receiving house and not by temperatures. In cases where the crude is of good quality it is not necessary to treat the benzine as it may merely be redistilled with steam. In many cases the refiner puts a good dephlegmator over on his crude still and makes a marketable gasoline without either treating it with acid or redistilling it with steam.

When a high sulphur or low grade petroleum is treated the distillate is put into an agitator with sulphuric acid, the mixing being perfected by blowing air through the acid in the bottom of the agitator, thus contacting it with all portions of the benzine. The acid is drained out and the benzine washed with water. Caustic soda or "doctor" solution is added to neutralize the acid and the benzine is thoroughly washed to remove the last traces of caustic or sulfonates. The benzine is redistilled in a steam still to give a gasoline of 58 to 60 gravity and about 430 end point, this depending largely upon the perfection of the dephlegmator. The last portion of the distillate is naphtha if a gasoline of high Baume' is desired. High gravity crudes are blended with low gravity crudes to eliminate the naphtha fraction.

are blended with low gravity crudes to eliminate the naphtha fraction. 2. Kerosene or Water White Distillate comes over just after the crude benzine, with the gravity of the stream in the receiving house at about 37.0 and a vapor temperature of 572°F. This will give a kerosene ordinarily of a 41 gravity, but this again varies greatly with the type of the oil. For example, a certain Wyoming crude oil under these conditions gives a 31.0° kerosene, whereas Cushing, Okla., and Bixby, Okla., crude oils give a 41.0° to 42.0° gravity kerosene. Pine Island cracked oil gives a 33-34°Be' kerosene. In distilling kerosene from the crude it is desirable to stop before there is discoloration from decomposition or cracking. Cracking may be very largely prevented and kerosene very greatly sweetened by using open steam throughout the entire distillation. The water white distillate or first run kerosene is now treated with acid and caustic in the agitator and exposed to heat, air and light in a shallow tank or bleacher in which all water is settled out. If the kerosene after treatment is not water white or has too high an end point it is redistilled with superheated open steam. The residue in the still may be mixed with the solar oil.

3. Solar Oil or Distillate Oil is taken out immediately following the kerosene, being a crude distillate not subjected to refining, and sold for use in explosion engines and as a high grade special fuel oil. The making of this product depends upon the market. It may be about a 36 gravity product or it may be combined with gas oil or straw oil.

4. Gas Oil is taken immediately following the distillate oil or kerosene and its distillation is continued until the residuum in the still has a gravity of 23 to 26°Be'. It is distinctly a destructive distillation and the yield depends largely upon the method and rate of firing. Gas oil is used in making gas and contains a considerable amount of olefins and cracked products, and is not refined except for special purposes. If a gas oil fraction low in olefins (straw oil) is desired it is necessary to distill using open steam and direct fire. Straight firing gives a more fluid residue on account of cracking.

Straight firing gives a more fluid residue on account of cracking. 5. Residuum or tar is sold as fuel oil or it may be used to produce lubricating oil. In the latter case it may be put into tar stills and run down to coke. The distillate is treated, refrigerated and the paraffin is removed by the filter press. If the crude oil contains no wax then the lubricants may be made by vacuum, steam or gas distillation, and the distillate is only filtered through fullers earth for use.

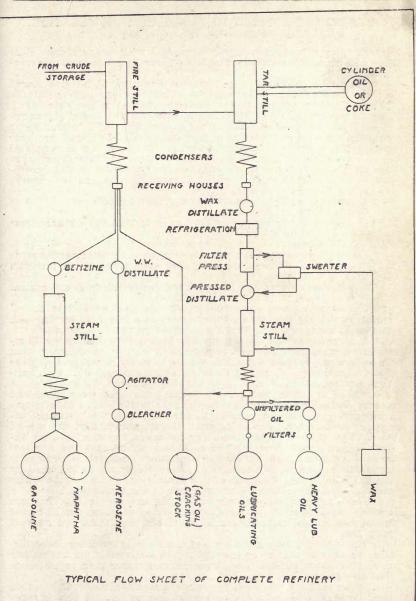
6. The filtrate from the paraffin wax or pressed distillate may be redistilled with steam to produce lubricating oil of the desired gravity, viscosity and cold test. The heaviest residual oil is the steam cylinder stock. Steam cylinder stock is the residue from the steam distillation of light colored crude oils, such as Cabin Creek, W. Va., and Ranger, Tex. The most careful refining is required for the automobile cylinder oil in order to obtain low fixed carbon to prevent separation of free carbon in the cylinder.

7. When asphalt is desired the residue from the gasoline and kerosene may be distilled by blowing superheated steam through it until the desired consistency is reached. Asphalt base oils or cracked paraffin base oils are necessary to make first class asphalt. An outline of the methods used for producing asphalts and road oils is given on page 191. Frequently, particularly for road oils, the stock remaining after cracking heavy gas oil is run down to a semi-solid or solid consistency. This gives a specially valuable road oil on account of its high asphalt content, good hardening or drying properties, low viscosity and excellent penetration.

For refining by cracking see pages 209-232.

For illustration of a refinery operation see flow sheet on page 133.

KANSAS CITY TESTING LABORATORY



PYROMETRY APPLIED TO PETROLEUM DISTILLATION.

C. Benton Kennedye, Pyrometric Engineer.

Refinery operation is largely dependent upon temperature. Considerable thought and study should be given to its correct measurement.

The most widely used instruments for measuring high temperatures are the Thermo-Electric Pyrometers. The improved high resistance Thermo-Electric Pyrometer for refinery application consists of a thermo-couple inserted eighteen inches into the still and a galvanometer. The thermo-couple is formed of two wires of different alloys welded at one end and when this junction is heated by the oil or vapor it generates a small current of electricity. The current thus generated operates a millivolt meter. As the temperature in the still or vapor line increases or decreases, the millivoltage generated by the thermocouple is increased or decreased in direct proportion and is indicated on the instrument in degrees.

The advantages of Pyrometers over Thermometers are:

- A. Ease of observation.
- B. Adaptability—recorders can be located any distance from stills or cracking plant.
- C. Robustness of apparatus, ease of repair.
- D. Availability for automatically making permanent records of temperature extending over considerable intervals of time.
- E. Indications can be noted and controlled from one central point by means of switch.

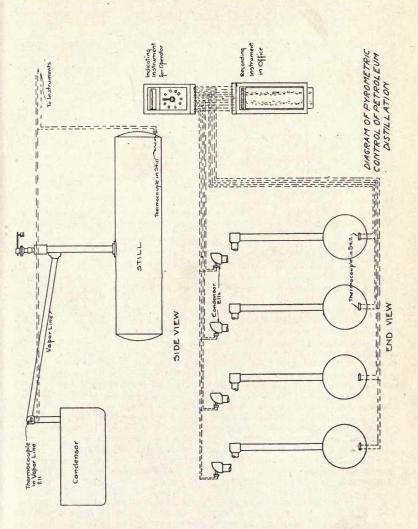
The pyrometers of one manufacturer are most popular in the oil industry owing to the fact that they maintain a free field service with competent engineers who periodically check up their equipment. A recent practical development is known as a Resistance Ther-

A recent practical development is known as a Resistance Thermometer, which will measure temperatures with great accuracy from three hundred degrees Fahr. below zero to eighteen hundred degrees Fahr. above zero. A coil of platinum or pure nickel wire with suitable protecting tube similar to diagram on the following page, is installed in the vapor line just before entering the condenser and another bulb installed in a similar manner in the still. With a constant source of current passing through the coil, the resistance increases or decreases, depending upon the temperature of the coil of wire. This change in resistance can be easily measured and an adjustable resistance balancing the resistance of the bulb and a galvanometer or deflector shows when the balance is reached. With suitable switching apparatus, any number of temperatures can be indicated by the one instrument. Gasoline of any desired end point may be easily secured by maintaining that temperature in the vapor line.

These instruments are accurate within one degree F. They are entirely unaffected by the length of leads connecting the bulb and instrument or the temperature of the leads, and there are no cold junctions.

The rapidly growing tendency of the trade to purchase oil on distillation tests as well as on gravity, makes it desirable for the refiner to make runs on oil and vapor temperatures rather than on tailhouse gravities alone.

In the cracking processes, the temperatures of the oil being treated must be maintained within very narrow ranges and for this purpose accurate pyrometers are absolutely necessary.



135

Cause of Color and Odor in Refined Petroleum

Most distillates from petroleum contain sufficient foreign matter to give an undesirable odor or a yellowish to red color.

The odor in natural distillates is due ordinarily to sulphur compounds, characteristic of which is hydrogen sulphide. Gasoline or light hydrocarbons produced by cracking have a more or less offensive odor even though sulphur is not present in appreciable quantity. In a general way, color is present in proportion as the odor is more disagreeable. The color of petroleum products is thought to be largely due to nitrogen compounds. Light hydrocarbons produced by cracking have a higher color the larger the amount of nitrogen in the heavy oils cracked, as a general rule. Cracked products from paraffin hydrocarbons such as those from Oklahoma give a yellowish color in the distillate above 300°F, though they may be colorless below 300°F. California and Mexico cracked gasoline give a red color, which is not noticeable immediately upon distilling, but becomes more intense as the gasoline is exposed to the action of the acid. This coloring matter on standing largely settles out or is oxidized so that the redistilled gasoline may be free from color.

Kerosene, the first refined product of petroleum marketed on a large scale, was a yellow or dark red liquid. It was first produced from coal, and it was found in 1857 that "coal oil" could be deodorized and decolorized by treatment with sulphuric acid, and this is the process that is in general use at the Present time. 66°Be' sulphuric acid is ordinarily used, as it reacts upon the unsaturated compounds, the sulphur compounds and the nitrogenous compounds in the oil by forming substances which dissolve largely in the sulphuric acid. The shrinkage of the oil treated may vary from almost nothing up to 10%, depending upon the character of the oil being refined. In ordinary natural distillates, one pound of acid per barrel is commonly sufficient, but with cracked oil as much as 10 pounds of acid are often required. Even then the treatment is often not sufficiently severe, and oleum or Nordhausen sulphuric acid, which contains an excess of sulphur trioxide, is necessary. This is the case with California and Towanda oil. After treatment with sulphuric acid, thorough washing and neutralization with caustic soda is always necessary. Other substances used for neutralizing the acid and acid sulfonates are soda ash, lime, silicate of soda and sodium plumbite.

Other chemicals may be quite successfully used in removing the odor of cracked gasoline, among these being sodium plumbite, copper oxide, manganese dioxide, potassium permanganate, sodium chromate, aluminum chloride and chlorine.

Dry hydrochloric acid gas (hydrogen chloride, HCl) is often highly effective in treating gasoline to remove the color.

The "bloom" or fluorescence of mineral oils is supposed to be due to the presence of asphalt-like or pitchy material in colloidal condition. This is overcome by the use of mono-nitro-napthalene $(C_{10}H_7NO_2)$ in small amounts. The physical means of removing color and to some degree odor is by filtration through fuller's earth. This is common practice with lubricating oils.

THE EFFECT OF SULPHUR IN THE REFINING OF PETROLEUM.

Sulphur is present in all petroleums. (See page 128.) It exists in the elementary form dissolved in the oil or in a chemically combined form as the sulphides of hydrocarbon groups. When it is found in very large amount there is usually a considerable amount of free or elementary sulphur. The alkyl or organic sulphides give to petroleum its characteristic odor. High sulphur petroleum residues such as Trinidad asphalt have characteristic odors of complex sulphur compounds. Lighter gasoline-bearing oils such as the Ohio and the Butler County, Kansas, oils have characteristic odors varying from that of pure hydrogen sulphide to that of the complex organic sulphides such as exist in natural asphalt. A typical distillation of a heavy crude oil by means of steam shows the following results as to distribution of sulphur:

Fraction	Specific Gravity	Sulphur
0-10%	$0.868 = 31.3^{\circ} \text{Be'}$	0.39%
10-20%	$0.877 = 29.6^{\circ} \text{Be'}$	0.35%
20-30%	$0.895 = 26.4^{\circ} \text{Be'}$	0.43%
30-40%	$0.909 = 24.0^{\circ} \text{Be'}$	0.53%
40-50%	$0.920 = 22.1^{\circ} \text{Be}'$	0.70%
50-60%	$0.920 = 22.1^{\circ} \text{Be}'$	0.70%
60-70%	0.917 = 22.7° Be'	0.70%
70-80%	$0.917 = 22.7^{\circ} \text{Be'}$	0.56%

This condition does not hold in the case of all oils, particularly the oils from Butler County, Kansas, which are characterized by the giving off of the rather large amount of hydrogen sulphide in the early part of the distillation.

Sulphur causes trouble in the refinery in the purification of the distilled products and in the corrosive effect of the oxidized sulphur, particularly on the condenser Pipes. At the time that the first sulphur oils were discovered in Ohio

At the time that the first sulphur oils were discovered in Ohio (.8% sulphur) they brought a price of only 14c per barrel, while at the same time the Pennsylvania oils (0.04% sulphur) sold at \$2.25per barrel. According to Frash it is a comparatively simple matter to free petroleum of elementary sulphur or hydrogen sulphide, but the sulphur compounds, which are the cause of the offensive odor, are very stable and cannot easily be broken up into hydrogen sulphide or other sulphur compounds which can be eliminated. It was because of the presence of these stable compounds that high sulphur oils for many years resisted all efforts to refine it. These complex sulphur compounds have the peculiarity of dissolving a number of metallic oxides. When the oil is saturated with all of the oxide which can be carried, the disagreeable odor disappears. It tends to reappear, however, when an attempt is made to separate the metal from the oil unless more oxide is used than is necessary to precipitate all of the sulphur, in which case complete desulphurization of the petroleum is effected. The Frash method, which has been successfully used for nearly thirty years by the Standard Oil Co., consists in the use of 1,000 pounds of the copper oxide to 2,000 barrels of distillate. The copper is recovered by filtering and roasting.

In distillation the chemical action of the sulphur may result from the direct combination of the sulphur with the iron or by the oxidation of the sulphur with formation of sulphonic acids, which pit the iron, particularly of the condensers.

The acid withdrawn from the agitator after treatment of oils to

remove color and odor is a black viscous material. Much of this sulphuric acid may be recovered by digestion to decompose the complex organic compounds and oxidation usually with air to burn out the carbonaceous material and preserve as much of the sulphur as SO_s instead of driving it off as SO_s .

Gasoline

Gasoline as now found on the market is a mixture of petroleum hydrocarbons, having an initial boiling point of from 80°F to 160°F, an end boiling point of from 368°F to 450°F, gravity of <u>56° to 61°Be'</u>., a sweet to oily aroma and a water white color.

The particular hydrocarbons composing it belong to a general group known as the paraffins. Other types of hydrocarbons are occasionally present in a very small amount. These are known as olefins and as benzenes. The olefins are removed by a thorough treatment with sulphuric acid, but the benzenes remain if originally present.

Ordinary gasoline made by the natural distillation of Mid Continent crude oil will contain several or all of the following substances:

	Name		Boiling point	Specific gravity	Baume'Gravity	
1.	Pentane		$97^{\circ}F$	0.630	92.2°	
2.	Hexane		$156^{\circ}F$	0.670	78.9°	
3.	Heptane		209°F	0.697	70.9°	
4.	Octane		$258^{\circ}F$	0.718	65.0°	
5.	Nonane		$302^{\circ}F$	0.740	59.2°	
6.	Decane		343°F	0.750	56.7°	
7.	Undecane	3	383°F	0.760	54.2°	

The following aromatic compounds are produced by pyrogenic decomposition of heavy hydrocarbons and rarely exist naturally in crude petroleum.

They are produced by the cracking of oil in the vapor phase and at high temperatures and occur in artificial or what has been called "synthetic" gasoline.

Name B	oiling point	Specific gravity	Baume' gravity
Benzol (C ₆ H ₆)	176°F	0.880	29.1°
Toluol (C ₈ H ₅ ĆH ₃)	232°F	0.872	30.6°
Xylene (C ₆ H ₄ (CH ₃)	2 291°F	0.882	28.7°

A small amount of these hydrocarbons in commercial gasoline very materially affects the gravity.

The character of gasoline is governed almost entirely by its use for automobiles. It is also used to some extent for stove gasoline and for cleaning purposes, in which case it has a lower end point and a higher Baume' gravity.

Gasoline is commonly blended and originates from one or more of the following sources:

1. The natural product distilled from crude oil. This constitutes about 73% of the total on the market (1917-18).

2. As a condensate from natural gas and known as casinghead gasoline. This constitutes about 7% of all gasoline and is always incorporated with heavy hydrocarbons such as naphtha or with gasoline distilled from a heavy crude or with gasoline made by cracking.

distilled from a heavy crude or with gasoline made by cracking. 3. The light hydrocarbons produced by the pyrogenic decomposition of heavy petroleum residua. This constitutes about 20% of the market gasoline and tends to have a considerable amount of aromatic compounds. The most desirable properties of gasoline are low end point and a low initial boiling point, the usual refiner's practice being to call everything gasoline which distills up to a temperature of 410°F. This practice in a light crude gives a 58°Be' product, although in the unusually light crudes a 61° product is obtained and in heavy crudes a gravity as low as 54° may be obtained. This heavy gasoline must be blended to make it satisfactory for ordinary market purposes. Page 227 shows the relation of the boiling point to the specific

Page 227 shows the relation of the boiling point to the specific gravity of ordinary market gasoline. Gasolines containing considerable olefins, aromatics or naphthenes have a higher relation of specific gravity to boiling point than do gasolines composed entirely of paraffin hydrocarbons.

Page 148 shows the relation of the boiling temperature to the percentage distilled over in ordinary commercial gasoline. These curves show that the gravity alone is not a good measure of the quality of a gasoline. For example, a 53° gravity gasoline in one case has an initial boiling point of less than 100°F and in another case has an initial boiling point of 190°F. A naphtha blended with casinghead will have a very high gravity test, but will show a very low initial boiling point and a very high end point.

The method of determining the quality of gasoline is described on page 307.

U. S. GASOLINE SPECIFICATIONS.

Specifications for standard tests of aviation gasoline, motor gasoline and fuel oil as announced in October, 1918, by the Inter-departmental Committee on Stardardization of Specifications for Petroleum Products.

AVIATION GASOLINE.

The specifications for aviation gasoline (export, fighting and domestic) as adopted are as follows:

1. Color.

The color shall be water white.

Test—Inspection of a column in a standard 4 ounce oil sample bottle.

2. Foreign Matter.

The gasoline shall be free from acid, undissolved water and suspended matter.

Acid Test—The residue remaining in the flask after distillation is complete is shaken thoroughly with 1 cc of distilled water. The aqueous extract must not be colored red on addition of a few drops of methyl orange solution. Water and suspended matter would be in evidence in the test for color.

3. Doctor Test.

The gasoline shall yield a negative doctor test.

Directions for making doctor test on gasoline:

(a) Preparation of Reagents: Sodium plumbite or "doctor solution." Dissolve approximately 125 grams of sodium hydroxide (NaOH) in a liter of distilled water. Add 60 to 70 grams of litharge (PbO) and shake vigorously for 15 to 30 minutes or let stand with occasional shaking for at least a day. Allow to settle and decant or siphon off the clear liquid. Filtration through a mat of asbestos may be employed if the solution does not settle clear. The solution should be kept in a bottle tightly stoppered with a cork.

Sulphur-Obtain pure flower of sulphur.

(b) Making a Test:-Shake vigorously together two volumes of

gasoline and one volume of the "doctor solution" (10 cc of gasoline and 5 cc of "doctor solution" in an ordinary test tube; or proportional quantities in a 4 ounce oil sample may be conveniently used). After shaking for about fifteen seconds a small pinch of flowers of sulphur should be added and the tube again shaken for 15 seconds and allowed to settle. The quantity of sulphur used should be such that practically all of the sulphur floats on the surface, separating the gasoline from the "doctor solution."

(c) Interpretation of Results—If the gasoline is discolored, or if the sulphur film is so dark that its yellow color is noticeably masked the test shall be reported as positive and the gasoline condemned as "sour". If the liquid remains unchanged in color and if the sulphur film is bright yellow, or only slightly discolored with gray or flecked with black the test shall be reported negative and the gasoline considered "sweet".

4. Corrosion and Gumming.

The gasoline, when subjected to the corrosion test, shall show no gray or black corrosion and no weighable amount of gum.

The apparatus used in this test consists of a freshly polished hemispherical dish of spun copper, approximately 3½ inches in diameter.

Fill this dish to within $\frac{3}{8}$ inch of the top with the gasoline to be examined and place the dish upon a steam bath. Leave the dish on the steam bath until all volatile portions have disappeared.

If the gasoline contains any dissolved elementary sulphur the bottom of the dish will be colored gray or black.

If the gasoline contains undesirable gum-forming constituents there will be a weighable amount of gum deposited on the dish. Acid residues will show as gum in this test.

Interpretation of Results.

Corrosion—It is specified that no gray or black deposit shall be formed. This wording is intended to admit gasolines that have so small a quantity of sulphur that the deposit is peacock colored.

Gum—It is specified that there shall be no weighable amount of gum. The intention is to refuse admittance to gasoline that shows an amount that can be readily weighed in this style of dish.

The distillation method and apparatus shall conform to those outlined and described in Bureau of Mines Technical Paper No. 166, entitled "Motor Gasoline, Properties, Laboratory Methods of Testing and Practical Specifications."

Volatility and Distillation Range-Export Grade.

When 5% of the sample has been recovered in the graduated receiver the thermometer shall not read more than $65^{\circ}C$ (149°F) or less than $95^{\circ}C$ (203°F).

When 50% has been recovered in the receiver the thermometer shall not read more than $95^{\circ}C$ (203°F).

When 90% has been recovered in the receiver the thermometer shall not read more than 150 °C (302 °F).

When 95% has been recovered in the receiver the thermometer shall not read more than 150° C (302° F) and the end point shall not exceed this temperature by more than 15° C (27° F).

At least 96% must be recovered in the receiver from the distillation.

The distillation loss shall not exceed 2% when the residue in the flask is cooled and added to the distillate in the receiver.

Volatility and Distillation Range-Fighting Grade.

When 5% of the samples has been recovered in the graduated receiver the thermometer shall not read more than $70^{\circ}C$ (158°F) or less than $60^{\circ}C$ (140°F).

When 50% has been recovered in the receiver the thermometer shall not read more than 95°C (203°F).

When 90% has been recovered in the receiver the thermometer shall not read more than 113°C (235°F).

When 96% has been recovered in the receiver the thermometer shall not read more than 113° C (235° F) and the end point shall not exceed this temperature by more than 15° C (27° F).

At least 96% must be recovered in the receiver from the distillation.

The distillation loss shall not exceed 2% when the residue in the flask is cooled and added to the distillate in the receiver.

The United States War Department requires fighting grade to be colored red after inspection and acceptance.

Volatility and Distillation Range-Domestic Range.

When 5% of the sample has been recovered in the graduated receiver the thermometer shall not read more than $75^{\circ}C$ ($167^{\circ}F$) or less than $50^{\circ}C$ ($122^{\circ}F$).

When 50% has been recovered in the receiver the thermometer shall not read more than $105^{\circ}C$ (221°F).

When 90% has been recovered in the receiver the thermometer shall not read more than 155°C (311°F).

When 96% has been recovered in the receiver the thermometer shall not read more than 175°C (347°F).

At least 96% must be recovered in the receiver from the distillation.

The distillation loss shall not exceed 2% when the residue in the flask is cooled and added to the distillate in the receiver.

MOTOR GASOLINE.

The specifications for motor gasoline are:

Quality.

Gasoline to be high grade, refined and free from water and all impurities, and shall have a vapor tension not greater than 10 pounds per square inch at 100°F temperature, same to be determined in accordance with the current "Rules and Regulations for the Transportation of Explosives and Other Dangerous Articles by Freight"—paragraph 1824 (k) as issued by the Interstate Commerce Commission.

Inspection and Tests.

Inspection—Before acceptance the gasoline will be inspected. Samples of each lot will be taken at random. These samples immediately after drawing will be retained in a clean, absolutely tight closed vessel and a sample for test taken from the mixture in this vessel directly into the test vessel.

Test—100 cc will be taken as a test sample. The apparatus and method of conducting the distillation test shall be that described in Bureau of Mines Technical Paper No. 166, Motor Gasoline:

(a) Boiling point must not be higher than 60°C (140°F).

(b) 20% of the sample must distill below 105°C (275°F).

(c) 45% must distill below 135°C (275°F).

(d) 90% must distill below 180°C (356°F).

(e) The end of dry point of distillation must not be higher than 220° C (428° F).

(f) Not less than 95% of the liquid will be recovered from the distillation.

MINERAL SPIRITS-1918.

1. General Specifications—General specifications for paint and painting materials, issued by the Railroad Administration, in effect at date of opening of bids, shall form part of these specifications. 2. The mineral spirits shall be a hydrocarbon distillate, water

2. The mineral spirits shall be a hydrocarbon distillate, water white, neutral, clear and free from suspended matter and water. It shall have no darkening effect when mixed with basic carbonate white lead.

3. Properties and Tests—When 100 cc are submitted to continuous distillation in an Engler flask with a condenser 22 inches long and at an angle of 30 degrees with the horizontal and cooled with water, the first drop shall issue from the condenser at a temperature of not less than .265°F and 97 per cent shall distill below 470°F.

4. When 10 cc of the distillate are placed in a glass crystallizing dish $2\frac{1}{2}$ inches in diameter, in a steam bath maintained at a temperature of 212° F and evaporated not more than 0.2 per cent of residue shall remain after $2\frac{1}{2}$ hours.

5. The flash point shall be not less than 85°F when determined by the closed Elliott tester method, the test being made in the usual official manner.

Summary of Gasoline Inspection Laws (By Dr. G. W. Gray.)

Arkansas.—Gravity shall be taken at 60°F, and marked on tank, can, cask, barrel or other vessel containing said gasoline.

California .- No law. Los Angeles has adopted motor transport specification.

Colorado .- Gravity shall be taken, but no products shall be offered for sale which contain more than 5 per cent of solid matter.

Georgia .- Gravity shall be taken and no product known as gasoline, benzine or naphtha shall be offered for sale unless casks, barrels or packages containing such products are labeled with figures de-noting gravity and the words "gasoline" "Benzine" or "naphtha", in large red letters.

Idaho .- The standard adopted by the Bureau of Mines shall be the standard for Idaho.

Indiana—Gravity shall not be less than 56°Be., and the correction for temperature shall be 1°Be. or 10°F.

Illinois.—There is no law except that gasoline must be branded "Condemned for illuminating purposes".

lowa.-Gravity shall be between 80°Be. and 70° Be. Boiling point shall not be below 150°F. and not above 210°F. All other products shall be branded "Substitute for gasoline" and these substitutes shall be sold under label, which label shall be printed in large, legible type, etc., defined as follows:

(a) Per cent of boiling below 135°F.

(b) Per cent of boiling between 135°F and 210°F.
(c) Per cent of boiling between 201°F. and 302°F.

(d) Per cent boiling above 302°F. Bills of lading and the labels of such substitutes shall call attention to the danger of such low boiling point.

Kansas-Gravity must not be heavier than 58°Be., initial boiling point shall not exceed 90°F., end boiling point not above 410°F. All products sold not meeting this test shall be known and sold as "Gasoline under test."

Michigan -No law. Grand Rapids, 20 per cent shall distill over at or below 320°F. Fifty per cent shall distill over at or below 300°F. End point not above 450°F. If product does not meet this test, it shall be known as a mixed gasoline-kerosene; Detroit, same law as Grand Rapids, but method of distillation is entirely different. Gaso-line passing Grand Rapids specification by their method of distillation, might be rejected by Detroit.

Minnesota-Gravity shall be taken and containers shall be marked "Unsafe for illuminating purposes." Missouri.—Gravity must not be less than 58°Be.

Montana .- Any gasoline used for heating, burning or power purposes in any automobile, engine or in any machinery which falls below 63°Be., shall be deemed below standard, but nothing in this act shall prevent the sale of a heavier product, when product is sold under its proper name and its specific gravity given.

Nebraska .- Gravity shall be taken and marked upon container.

New Mexico .- No gasoline for illuminating purposes can be sold which is less than 63° gravity, and it shall be conclusively presumed that all sales are for illuminating purposes, unless containers are marked "Not for illuminating purposes."

North Carolina.—The initial boiling point not higher than 158° F; 16 per cent off at 230° F; residue not more than 35 per cent at 302° F.; end point not higher than 437° F.

North Dakota.—Gravity shall be taken and all gasoline sold for household purposes shall show not less than 3 per cent off at 158°F. and not more than 6 per cent residue at 248°F.

Ohio.—Shall be branded according to its commercial name and with the word "Dangerous".

Oklahoma.—Gravity shall be taken. If gravity is greater than 74°F. it shall be deemed unsafe and sale is prohibited for use in vapor stoves or other domestic uses.

Oregon.-Gravity shall be not less than 56°Be.

South Dakota.—Gas machine gasoline, light gasoline, power gasoline, when made from Mid-continent crude shall be as follows: Gas machine gasoline, not less than 64° Be; residue not more than 4 per cent at 300° F; all off below 350° F. Light gasoline, gravity not less than 60° Be; residue not over 10 per cent above 300° F. and not over 25 per cent above 350° F. Power gasoline gravity not less than 57° Be; residue not more than 25 per cent at 300° F and not more than 3 per cent at 400° F. Below is a table giving gravities depending upor what crude the products are made from:

	Gravity in deg Mid-Conti- nent Field.	Penn.
Gasoline for use in automobile engines and		
other gasoline engines should have a gravit		
of not less than	57	62
Gasoline for household use in stoves, flatiron		1.1.1.1
gasoline lamps, dry cleaning, etc., shou have a gravity of not less than	62	65
Gasoline for use in gas machines for the pr		
duction of gasoline gas, should have a gravit		1
of not less than	70	80
Naphtha for use in engines and for other purpos		
should have a gravity of not less than	55	

In describing kerosene or gasoline by its gravity it is necessary to indicate the State or Territory producing the crude petroleum from which the finished product was distilled, because crude petroleum differs in different regions and its products differ likewise. In stating the crude petroleum fields above, the Western is taken to include Texas, Oklahoma, Kansas, Wyoming, Illinois and other oil-producing States in the west-central portion of the United States. The Pennsylvania field includes Pennsylvania, West Virginia and neighboring States.

South Carclina.—Flash point not more than 32°F; distillation test not less than 25 per cent off at 230°F; not more than 16 per cent of residue at 302°F; dry point not more than 392°F. Any product not meeting this specification must be sold under the name of "naphtha." Tennessee.—The container shall be branded "Gravity not less than "Be.; unsafe for illuminating purposes; for power purposes only."

Utah.—Standards adopted by the Bureau of Mines shall be standard for this State. No product sold shall contain more than 1 per cent of solid matter.

Washington.—Shall be inspected for its specific gravity and all containers shall be branded with the specific gravity.

Wisconsin .- Containers shall have gravity stamped on same.

Wyoming.—Gasoline for household use. Distillation test: Not less than 10 per cent off at 150°F; not less than 50 per cent off at 212°F; not less than 98 per cent off at 325°F. Gasoline for power purposes: Not less than 10 per cent off at 170°F; not less than 50 per cent off at 240°F; not less than 94 per cent off at 350°F.

Benzinum Purificatum (U. S. Pharmacopoeia)

Purified Petroleum Benzin. Benzin. Purif.—Petroleum Ether.

A purified distillate from American petroleum consisting of hyirocarbons, chiefly of the marsh-gas series. Preserve it carefully in vell-closed containers, in a cool place, remote from fire.

Purified Petroleum Benzin is a clear, colorless, non-fluorescent, volatile liquid, of an ethereal, or faint, petroleum-like odor, and having a neutral reaction. It is highly inflammable and its vapor, when mixed with air and ignited, explodes violently.

It is practically insoluble in water, freely soluble in alcohol, and miscible with ether, chloroform, benzene, volatile oils and fixed oils. with the exception of castor oil.

Specific gravity: 0.638 to 0.660 at 25°C.

It distil's completely between 40°C and 80°C (104°F to 176°F).

Evaporate 10 mils of Purified Petroleum Benzin from a piece of clean filter paper: no greasy stain remains, and the odor is not disagreeable or notably sulphuretted. Not more than 0.0015 Gm. of residue remains on evaporating 50 mils of Purified Petroleum Benzin at a temperature not exceeding 40°C. Boil 10 mils of Purified Petroleum Benzin for a few minutes with

Boil 10 mils of Purified Petroleum Benzin for a few minutes with one-fourth its volume of an alcoholic solution of ammonia (1 in 10) and a few drops of silver nitrate T. S.; the liquid does not turn brown (pyrogenous products and sulphur compounds).

Add 5 drops of Purified Petroleum Benzin to a mixture of 40 drops of sulphuric acid and 10 drops of nitric acid in a test tube, warm the liouid for about ten minutes, set it aside for half an hour, and dilute it in a shallow dish with water; no odor of nitrobenzene is evolved.

NAVY SPECIFICATIONS FOR GASOLINE.

Regular Gasoline.

The	havy specifications for gasoline are as follows:	
	nitial below	0
	20% off at20	0
	5% off at	5
	0% off at	6
	End point below	

Aero Gasoline.

The	e aero gasoline (for fighting planes) shall be:
	Not more than 5% shall distill below 60°C (140°F).
(b)	Not less than 5% shall distill below 70°C (167°F).
(c)	At least 50% shall distill below 95°C (202°F).
	At least 90% shall distill below 113°C (235°F).
(e)	At least 96% shall distill below 125°C (257°F).

Export Gasoline.

Export gasoline (for use in bombing planes): (a) Not more than 5% shall distill below $60^{\circ}C$ (140°F).

- (b) Not less than 5% shall distill below 75°C (167°F).
- (c) At least 50% shall distill below 10°C (212°F).
 (d) At least 90% shall distill below 125°C (257°F).
- (e) At least 96% shall disitll below 150°C (302°F).

Domestic Gasoline.

- Domestic gasoline (for use in training planes): (a) Not more than 5% shall distill below $60^{\circ}C$ (140°F).
- (b) Not less than 5% shall distill below 75°C (167°F).
- (c) At least 50% shall distill below 105°C (223°F).
 (d) At least 90% shall distill below 155°C (311°F).

(e) At least 96% shall distill below 175°C (347°F).

Comparison of Gasoline and	Benzol as Mo	tor Fuel.
Heat of combustion:	Benzol.	Gasoline.
B. T. U. per gallon	132330	129060
B. T. U. per pound	18054	20750
Freezing temperature.	41°F	50°F below Zero
Boiling temperature.	170-180	130-400°F
Rate of evaporation	Slower	Faster
Mileage per gallon (comparative)	110.	100.
Ignition temperature	Higher	
Preignition from carbon	Less trouble	More trouble
Carbon formed.	More	Less
Relative volume of air required per		
gallon	1.04	1.00
Relative volume of explosive gases		
produced per gallon	.92	1.00
Temperature of explosion	Higher	Lower
Rapidity of explosive force	Less sudden	More sudden
Benzol is most satisfactory if use	d mixed with	gasoline or alco-

hol, preferably the latter.

Possible Savings in Gasoline

The Bureau of Mines estimates that the following savings can be effected daily:

Tank wagon losses
Leaky carburetors, average 1/17th of a pint per car
Poorly adjusted carburetors, 1/2 pint per car
Motors running idle, ¼ pint per car
Wasted in garages, 10 pints per day 67,000
Saved by using kerosene in garages
Needless use of passenger cars, 1% pints per car

This makes a total of 1,500,000 gallons a day, or 561,000,000 gallons a year, whereas our war needs are 350,000,000 gallons a year, or less than two-thirds of what may be considered as wasted at the present time.

SUGGESTIONS TO GASOLINE USERS.

The following important suggestions for avoiding waste will not only save gasoline, but users of motor vehicles will be benefitted personally and individually through more efficient and more economical operation of cars:

1. Store gasoline in underground steel tanks. Use wheeled steel tanks with measuring pump and hose. They prevent loss by fire, evaporation and spilling.

2. Don't spill or expose gasoline to air—it evaporates rapidly and is dangerous.

3. Don't use gasoline for cleaning and washing—use kerosene or other materials to cut grease.

4. Stop all gasoline leakages. Form habit of shutting off gas at tank or feed pipe.

5. Adjust brake bands so they do not drag. See that all bearings run freely.

6. Don't let engine run when car is standing. It is good for starter battery to be used frequently.

7. Have carburetors adjusted at service stations of carburetor or automobile companies—they will make adjustments without charge.

8. Keep needle valve clean and adjust carburetor (while engine is hot) to use as lean mixture as possible. A rich mixture fouls the engine and is wasteful.

9. Pre-heat air entering carburetor and keep radiator covered in cold weather—this will insure better vaporization.

10. See that spark is timed correctly with engine and drive with spark full advanced—a late spark increases gas consumption.

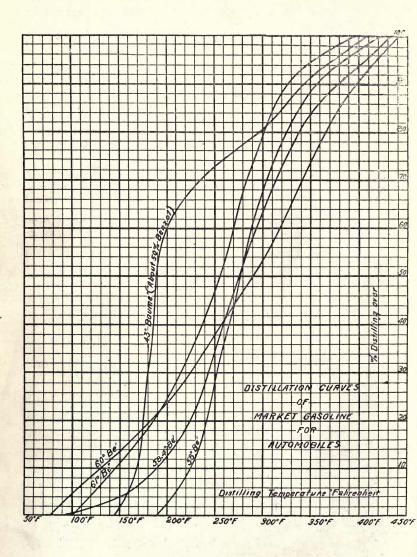
11. Have a hot spark, keep plugs clean and spark points properly adjusted.

12. Avoid high speed. The average car is most economical at 15 to 25 miles an hour.

13. Don't accelerate and stop quickly—it wastes gas and wears out tires. Stop engine and coast long hills.

14. Cut down aimless and needless use of cars. Do a number of errands in one trip.

15. Know your mileage per gallon. Fill tank full and divide odometer mileage by gallons consumed.



Kerosene, Coal Oil, Illuminating Oils

Kerosene in a general way may be defined as that fraction of crude petroleum or oil made by the pyrogenic decomposition of shales or coal and it distills at a temperature of from 302° F to 572° F (150- 300° C) and contains no gasoline or residuum. Its flash point is always greater than 100° F and usually greater than 120° F. Its color may be standard white, prime white, superfine white or water white. Its gravity ranges from 31 to 48° Baume'. Typical kerosene has a gravity of 41 to 42° Be'. Sulphur is usually almost completely absent from kerosene, being less than 0.03%. It consists chiefly of the paraffin series, particularly when the gravity is greater than 38. The principal constituents are nonane, decane, undecane, duodecane, tridecane, tetradecane, pentadecane, hexadecane and heptadecane. With lower gravities it contains naphthenes and aromatic compounds. This is particularly true of Louisiana oils and California oils.

The quality of good kerosene has been found to be within the following limits:

1. Specific gravity shall be between 0.760-0.860 (54.2-32.8°Be').

2. Flash point shall be over 100°F by closed tester.

3. Color shall be water white, with no turbidity.

- 4. Cold test shall be below 10°F.
- 5. End point shall be below 600°F.
- 6. Sulphur shall be below 0.03%.
- 7. Acid shall be absent.

8. It should not lose more than 1% on treatment with 66° sulphuric acid.

The U. S. Government specifications for various illuminating oils are as follows:

WATER WHITE KEROSENE.

Flash.—To be taken on the Tag closed cup A. S. T. M. standard, oil to be heated at the rate of 2°F per minute, test flame to be applied every 2 degrees commencing at 105°F.

Color.—To be determined on the Saybolt colorimeter or its equivalent.

Sulphur.—Test to be made by burning at least 2 grams of the oil in a small flask and absorbing the gases of combustion in a standard solution of sodium carbonate and titrating the excess of sodium carbonate with the standard solution of sulphuric acid.

Floc.—For making the test, take a hemispherical iron dish and place a small layer of sand in the bottom. Take a 500 cc Florence or Erlenmeyer flask and into it put 300 cc of the oil (after filtering if it contains suspended matter). Suspend a thermometer in the oil by means of a cork slotted on the side. Place flask containing the oil in the sand bath and heat bath so that the oil has reached a temperature of 240° F at the end of one hour. Hold oil at temperature of not less than 240F° nor more than 350° F for six hours. The oil may become discolored, but there should be no suspended matter formed in the oil. The flask should be given a slight rotary motion, and if there is a trace of "floc" it can be seen to rise from the center of the bottom.

Distillation Test—The oil shall all distill below temperature of 600°F. The test is made as described by the Bureau of Mines Technical Paper 166, using A. S. T. M. apparatus with wet bulb and total immersion thermometer.

Cloud Test.—For making test take a 4-ounce oil sample bottle and introduce therein 1½ ounces of the oil to be tested. Insert cork with cold test thermometer so that thermometer is suspended in the oil. Place bottle in a freezing mixture and cool to 0°F. Keep oil cooled to this temperature for 10 minutes. Bottle should be given a rotary motion occasionally so as not to supercool the sides. The oil should not be clouded from crystals of parafin wax at the end of ten minutes.

Reactions.—Two ounces of the oil should be shaken with $\frac{1}{2}$ ounce of warm neutral distilled water and allowed to cool and separate. The water when separated shall react neutral to methyl-orange and phenolphthalein.

Burning Test.—The oil must burn freely and steadily in a lamp fitted with a No. 1 sun hinge burner. It must give a good flame for a period of 18 hours without smoking or forming "ears" or "toadstools" on the wick. The chimney must be only slightly clouded or stained at the end of the test.

Specification Summary .-- Oil must be free from water, glue and suspended matter.

Flash.—Not less than 115°F, Tag closed cup, A. S. T. M. standard. Color.—To be 21 color on Saybolt colorimeter or its equivalent on

a Lovibond tintometer, these being equal to color of a solution of potassium bichromate containing 0.0048 gram per liter.

Sulphur.—Not more than 0.06%.

Floc.—Oil to be free from floc.

Distillation.—Oil to distill below temperature of 600°F. Cloud Test.—Oil should not show cloud at 0°F.

Reaction .- Must be neither acid nor alkaline.

Burning Test.—As stated above.

KEROSENE FOR U. S. NAVY. Water white kerosene for U. S. Navy use when specifically required for special fuel shall have a heating value of not less than 20,000 B. T. U. per pound.

When specifically provided for a representative sample of the oil delivered will be tested photometrically after burning for 1 hour in a lamp fitted with a No. 1 sun hinge burner. Five hours later another photometric test will be made to determine any change in intensity of the light. The maximum allowable loss shall be 5%. The flame shall show at least 6 candle power when compared photometrically with an incandescent lamp which has been standardized by the Bureau of Standards.

Otherwise specifications enumerated above apply for U.S. Navy kerosene.

LONG-TIME BURNING OIL.

Flash.—To be taken on the Tag closed cup, A. S. T. M. standard. Oil to be heated at the rate of 2°F per minute. Test flame to be applied every 2°, commencing at 105°F.

Color .- To be determined on the Saybolt colorimeter or its equivalent.

Floc.-For making test take a hemispherical iron dish and place a small layer of sand in the bottom. Take a 500 cc Florence or Erlenmeyer flask and into it put 300 cc of the oil (after filtering if it con-tains suspended matter). Suspend a thermometer in the oil by means of a cork slotted on the side. Place flask containing the oil in the sand bath and heat bath so that the oil has reached a temperature of 240°F at the end of one hour. Hold oil at temperature of not less than 240°F nor more than 250°F for 6 hours. The oil may become

discolored, but there should be no suspended matter formed in the oil. The flask should be given a slight rotary motion, and if there is a trace of "floc" it can be seen to rise from the center of the bottom.

Cloud Test .-- For making cloud test take a 4-ounce oil sample bottle and introduce therein 11/2 ounces of the oil to be tested. Insert cork with cold test thermometer, so that thermometer is suspended in the oil. Place bottle in a freezing mixture and cool to zero degrees Fahr. Keep oil cooled to this temperature for 10 minutes. Bottle should be given a rotary motion occasionally so as not to supercool the sides. The oil should not be clouded from crystals of paraffin wax at the end of 10 minutes.

Reaction .- Two ounces of the oil should be shaken with 1/2 ounce of warm neutral distilled water and allowed to cool and separate. The water when separated shall react neutral to methyl-orange and phenolphthalein.

Burning Test.-This test will be made by introducing 25 fluid ounces of oil into the pot of a standard Railway Signal Association semaphore lamp, fitted with the purchaser's standard burner, chimney and wick. The wick shall be new and previously washed with redistilled ether and dried at room temperature, the lamp to be protected from the direct rays of the sun but may be burned either outdoors or in a well-ventilated room. During the first hour of the test the wick will be adjusted so as to produce a flame %-inch high, measured from the top of the wick. The lamp shall burn continuously without readjusting the wick for 120 hours or until all of the oil is consumed.

The flame shall remain symmetrical and free from smoke throughout the test period.

The height of the flame at any time during the test shall be not less than three-quarters of an inch. The oil shall not produce any appreciable hard incrustation on the wick.

Oil must be free from water, glue and suspended matter. Flash.—Not less than 115°F, Tag closed cup, A. S. T. M. standard. Color.—Twenty-one color on Saybolt colorimeter or its equivalent. on a Lovibond tintometer, these being equal to color of a solution of potassium bichromate containing .0048 gram per liter. Floc.—Oil to be free from "floc."

Cloud Test.—Oil should not show cloud at 0°F. See Note 1 below. Reaction .- Must be neither acid nor alkaline.

Burning Test .- As stated above.

Note No. 1 Relative to Cloud Test .- Temperature of 0°F can be varied either up or down to suit the climatic conditions in the territory in which the oil is to be used.

LIGHT HOUSE OIL.

Oil for use by the Bureau of Light Houses shall be as described by the Department of Commerce, which specifications, etc., at the present time are as follows:

The kerosene must have a flash point of not less than 140°F (1)and fire point of not less than 160°F (Tag closed tester). (2) The kerosene must contain no free acids or mineral salts.

Litmus paper immersed in it for five hours must remain unchanged.

One hundred grams of kerosene shaken with 40 grams of (3)sulphuric acid (sp. gr. 1.73) must show little or no coloration).

(4) When distilled from a still so jacketed as not to allow of local heating at a rate of not over 10% in ten minutes the kerosene shall not distill below 350°F and 98% shall distill under 515°F, the temperature taken being that of the condensing vapor.

(5) When burned for 120 hours in a lens lantern supplied with a fifth order oil lamp, the kerosene must burn steadily and clearly without smoking with minimum incrustation of wick, slight dis-coloration of chimney and less than 10% loss of candie power. A lamp of this description will be loaned to successful bidder.

300 DEGREE MINERAL SEAL OIL.

Flash.—To be taken on the Cleveland open cup, oil to be heated at the rate of 7°F per minute, test flame to be applied every 5°, commencing at 210°F.

Fire Test.—After the flash point is obtained the oil shall be heated at the same rate (7° per minute), test flame to be applied every 5° after the flash point has been obtained.

Color.— To be determined on the Saybolt colorimeter or its equivalent.

Floc.—For making test take 500 cc Florence or Erlenmeyer flask and into it put 300 cc of oil (after filtering if it contains suspended matter). Oil to be heated at the rate of 10°F per minute to a temperature of 450°F and held at that temperature for 15 minutes. The oil shall show no floc or precipitate at that temperature or one hour after cooling.

Cloud Test.—For making this test take a 4-ounce oil sample bottle. Introduce therein 11/2 ounces of oil to be tested. Insert cork with cold test thermometer so that bulb is slightly below the surface of the oil. Place bottle in a freezing mixture and cool oil to a temperature of 32°F Keep oil cooled to this temperature for 10 minutes. Bottle should be given a rotary motion occasionally so as not to supercool the sides. The oil should not become cloudy from crystals of paraffin wax at the end of 10 minutes.

Reaction .- Two ounces of the oil should be shaken with 1/2 ounce of warm neutral distilled water and allowed to cool and separate. Water when separated shall react neutral to methyl-orange and phenolphthalein.

Burning Test .- This test will be made by introducing 20 fluid ounces of oil into a lamp fitted with a dual burner No. 3, dual chimney and duplex wicks. The lamp used shall be such that the distance from the top of the wick tube to the bottom of the inside of font is not less than $6\frac{1}{2}$ inches nor more than 7 inches. During the first hour of the test the wicks will be adjusted so as to produce a symmetrical flame approximately 1 inch high, measured from the top of the wicks. The lamp shall burn continuously without readjusting until all of the oil is consumed.

The flame shall remain symmetrical and free from smoke throughout the test period. The oil shall not produce any appreciable hard incrustation on the wick.

The oil must be free from water, flue and suspended matter.

Flash — Not less than 250°F, Cleveland open cup. Fire.—Not less than 300°F, Cleveland open cup. Color.—To be not less than 16 color on Saybolt colorimeter or its equivalent on the Lovibond tintometer, these being equal to color of a solution of potassium bichromate containing 0.012 grams per liter.

Floc.-Oil to be free from "floc."

Cloud Test -Oil should not show cloud at 32°F.

Reaction .- Must be neither acid nor alkaline.

Burning Test .- As stated above.

SIGNAL OIL.

Flash.—To be taken on the Cleveland open cup. Oil to be heated at the rate of $7^{\circ}F$ per minute and test flame to be applied every 5°, commencing at 210°F.

Fire 1est.—After the flash point is obtained the oil shall be heated at the same rate (7° per minute) and test flame to be applied every 5° after flash point has been obtained.

Cloud Test.—For making test take a 4-ounce oil sample bottle and introduce therein $1\frac{1}{2}$ ounces of oil to be tested. Insert cork with cold test thermometer so that bulb is slightly below the surface of the oil. Place the oil in a freezing mixture and cool to 32° F. Keep oil cooled to this temperature for 10 minutes. Bottle should be given a rotary motion occasionally so as not to supercool the sides. The oil should not become cloudy at the end of 10 minutes from crystals of paraffin wax or solid fats from the lard oil or sperm oil.

Burning Test.—This test is to be made in standard railway signal hand lantern, the burner of which is fitted with a 1-inch wick. The oil to be burned 24 hours without trimming or adjusting the wick, the pot of the lantern to be refilled if too small for a test of the duration named.

Oil must produce a satisfactory flame throughout the test period.

The oil must not produce an appreciable amount of hard incrustation on the wick.

The flame must stand all forms of railroad signaling in any kind of weather without being extinguished or smoking the globe.

Appearance.—The oil must be free from water, glue and suspended matter.

Composition.—To be 300° mineral seal oil as adopted by the Committee on Standardization of Petroleum Specifications, compounded with pure prime winter strained lard oil or sperm oil or compounded with a mixture of pure prime winter strained lard oil and sperm oil.

Flash.-Not less than 250°F, Cleveland open cup.

Fire .- Not less than 300°F, Cleveland open cup.

Cloud Test .- Oil should not show cloud at 32°F.

Percentage of Fatty Oil.—"A" grade must contain not less than 30% of fatty oil by volume.

"B" grade must contain not less than 22% of fatty oil by volume.

The "A" grade shall always be furnished unless "B" grade is specifically ordered.

Free Fatty Acids.—"A" grade must contain not over 0.60% free fatty acid calculated as oleic acid.

"B" grade must contain not over 0.45% free fatty acid calculated as oleic acid.

Burning Test.-As stated above.

Gravity.—It will be noted that there are not gravity specifications for any of the products mentioned above. It has been known for a number of years that the gravity of an oil, by itself, has no relation to the quality. Two oils may have exactly the same gravity and one might be an excellent oil while the other would be absolutely worthless. This difference in quality is due to the crude from which it has been made. Therefore no gravity was specified and the quality was left to be determined by other specifications.

Flash.—The Tag closed cup A. S. T. M. standard was adopted because it has been accepted by several societies and its measurements have been standardized.

Color.—The Saybolt colorimeter was adopted because most of the kerosene manufactured in this country is tested by this machine.

GAS OIL.

Gas oil is that fraction of petroleum distillation coming off after the kerosene or other illuminating oil. It is usually a destructive distillation resulting in a distilled product carrying a considerable amount of olefins and a residue having a lower viscosity than would be the case without a partially destructive distillation. When it is desired to avoid a destructive distillation, steam may be used, giving an oil suitable for absorption purposes sometimes known as straw oil.

Gas oil is used for making gas and for carburetting coal gas or water gas. It is also used to make Blaugas, which is a product liquified under a pressure of about 1,500 pounds. It is also used for Pintsche gas. A typical gas oil has the following properties:

gust in official gust out into the source of	
Specific gravity	$= 36.1^{\circ} \text{Be'}$
Flash point	90°C
Burning test	
Distillation test	
0°C-150°C	0.0%
150°C-300°C	44.0%
300°C up	
Coke	0.7%
	0/0

GAS OIL FOR DIESEL ENGINES (U. S. NAVY).

1. Flash point not lower than 150°F (Abel or Pennsky-Marten's closed cup).

2. Water and sediment-trace only.

3. Asphaltum-none.

Bunker Oil "B".—Specifications to be the same as for navy fuel oil except:

(c) Omit and substitute "The flash point shall not be lower than 150°F as a minimum (Abel or Pennsky-Marten's closed cup) or 175°F (Tagliabue open cup)."

(d) Omit and substitute "To have a minimum gravity of 15° Baume'."

(f) Omit.

Navy standard fuel oil only will be supplied to battleships, destroyers and other vessels subject to heavy forced draft conditions or required to run smokeless. It will also be supplied for cargo oil for all shipments abroad or to navy storage.

Bunker oil "A" will be used by other types of vessels requiring a light oil and by shore stations fitted with separate storage for yard use. It will not be used where Bunker oil "B" can be satisfactorily used.

Bunker oil "B" will be used by all transports and cargo vessels which can satisfactorily burn an oil not heavier than 15° Baume'.

The commander, Cruiser and Transport Force, or his representative and the District supervisor, Naval Overseas Transportation Service, shall determine the grade of oil to be used by vessels operating under their direction.

STRAW OIL (U. S. BUREAU OF STANDARDS). The characteristics of a straw oil for absorption of light oils from gas as recommended by some operators and which are concurred in by the committee of coal-tar products are substantially as follows:

Specific gravity not less than 0.860 (34°Be') at 15.5°C (60°F). 1.

Flash point in open cup tester not less than 135°C (275°F). 2.

Viscosity in Saybolt viscosimeter at 37.7°C (100°F) not more 3. than 70 seconds.

The pour test shall not be over 1.1°C (30°F). 4.

5. When 500 cc of the oil are distilled with steam at atmospheric pressure collecting 500 cc of condensed water, not over 5 cc of oil shall have distilled over.

The oil remaining after the steam distillation shall be poured 6. into a 500 cc cylinder and shall show no permanent emulsion.

The oil shall not lose more than 10% by volume in washing with 21/2 times its volume of 100% sulphuric acid when vigorously agitated with acid for five minutes and allowed to stand for two hours.

An additional set of specifications for wash oil which is used by one Government department is as follows:

Specific gravity shall not be greater than thirty-five and ninetenths degrees (35.9°) Baume' at 60°F, equivalent to specific gravity 0.844.

Viscosity shall not be more than 56 seconds in a Saybolt viscosimeter at 100° Fahrenheit.

The oil shall not thicken or cloud at 25°F in the cold test.

At least 95% of the oil shall separate as a clear layer within 10 minutes after 100 cubic centimeters of oil and 100 cubic centimeters of water have been shaken together vigorously for 20 seconds at a temperature of 70°F.

There shall not be more than 14% of loss in volume of oil when '1 volume of oil and 2½ volumes of 100% sulphuric acid are vigorously agitated for 5 minutes and allowed to settle for 2 hours.

The oil shall not begin to distill below 240°C.

Quality of Absorption Oil for Extracting Gasoline from Natural Gas (Westcott "Casinghead Gasoline").

Gravity.	35.6°
Initial boiling point	536°F
End point.	698°F
Fire test.	312.8°F
Saybolt viscosity @ 100°F	40.5

Distillation

Distillation.		
Initial.	273	°C
5%	295	°C
10%	300	°C
20%	305	°C
30%		
40%		
50%		
60%		
70%	329	°Ċ
80%	336.	5°C
90%		

Kerosene Regulations (March 1919)

The Martine States	Tabulation of Essential Points in State Laws.				
State.	Cup	Flash	Fire	Gravity	Distillation
Alabama			120		No law
Arizona	Open Tag		160	· · · · · · · · · · · · · · · · · · ·	No law
California	Foster			•••••	
Colorado Connecticut.	Open Tag	110	140		
Delaware	Open Tag		115		No law
Georgia	Elliott	190-			
Idaho	Open Tag				
Indiana	Indiana Elliott	120 100		50-46	
Kansas	Foster	110			
Kentucky	Tag	125	130		
Maine	Tag		120		
Maryland	Tag	100	110		
Michigan	Foster	120	120		
Mississippi.	Tag			40 min.	No law 4% resid.
Missouri				40 mm.	at 570
Montana	Foster.	110 112		42 min.	No law 7% resid.
					at 570
Nevada		100	120		No law
New Jersey	Tag	110			
New York	Tag	110	110		
North Carolina	Elliott	100			6% resid. at 570*
North Dakota.	Elliott	100	125 .		4% resid. at 570
	State State				6% max.
Ohio	Foster	120			at 310
Oklahoma	Tag	115		40-48	No law
Pennsylvania	Tag		110		
Rhode Island	Tag	100	110		6% res.
South Dakota.	Elliott.	105		47° Pa.	at 570
Doum Dakota	12110000	100	••••••	erude	Not more than 10%
				41° M-C	at 300 Not more
				crude	tban 4% at 570
Tennessee	Tag	120			
Texas	Foster.				No law
Vermont	Foster		110		No law
Washington	Tag elec. cup		120		
West Virginia.		105	120		No law
Wisconsin	Foster				5% resid. at 572
*North Carolina-If	ile are lighter th	0.0.47 0000	tre then nor	idaa maaat a	

*North Carolina-If oils are lighter than 47 gravity, then residue must not be more than 10 per cent. †Oklahoma-Oils 40-48 gravity must be branded Good. Oils less than 40 or more than 48 must be branded Inferior.

Specifications for Petroleum Products of the Kansas City Southern Railway Co.

Material.—The materials desired under this specification are the products of distillation and refining of petroleum, unmixed with any other substance, and conforming to the detailed specifications below.

Illuminating Oils.

General Requirements.—These oils must be water white in color, and free from sulphur in any form. "Cracked" oils are not desired. Products having an offensive odor or containing any admixture of other oils will not be accepted. All samples must show a neutral or slightly alkaline reaction.

Tests.—One sample shall be taken from each carload or fraction thereof, and subjected to the following tests:

Headlight or 150 Degree Oil.

Sample must not flash below a temperature of 130 degrees, or burn below a temperature of 150 degrees Fahrenheit, when heated at the rate of 2 degrees per minute. The test flame to be applied once every 5 degrees, beginning at 110. The above flash and fire tests will be made in the Tagliabue open cup tester.

Samples must remain clear and transparent when called to a temperature of 0 degrees and held there for ten minutes.

It must have a specific gravity of between 41 and 48 degrees Baume'.

Mineral Seal or 300 Degree Oil.

Sample must not flash below a temperature of 245 degrees or burn below a temperature of 300 degrees Fahrenheit, when heated at rate of 5 degrees per minute. The test flame to be applied once every 5 degrees, beginning at 180. The above flash and fire tests will be made in the Tagliabue open cup tester.

Sample must remain clear and transparent when called to a temperature of 32 degrees Fahrenheit, and held there for ten minutes.

It must have a specific gravity of between 38 and 43 degrees Baume'.

Gasoline.

General Requirements.-Gasoline shall be water white in color.

Tests.—A sample sufficiently large to provide for the following tests, taken at random, will represent the shipment:

1. Gasoline must not be heavier than specific gravity of 72 degrees Baume, but when specifically ordered stove gasoline may be furnished at specific gravity of 66 degrees Baume.

2. A portion of the sample must be entire'y volatile at a temperature not exceeding 100 degrees Fahrenheit.

3. When blotting paper is moistened with a few drops of the sample it must evaporate entirely, leaving no greasy stain.

Conditions.—If any portion of an accepted shipment is subsequently found to be damaged, or otherwise inferior to the original sample, that portion will be returned to the shipper at his expense.

Any sample failing to meet all the requirements of this specification will be condemned, and the shipment represented by it will be returned to the manufacturers, they paying freight both ways.

Pennsylvania Railroad Company No. 20-A.

SPECIFICATIONS FOR PETROLEUM PRODUCTS. Five different grades of Petroleum Products will be used.

These will be purchased in amounts as the demands of the service indicate.

2. The materials desired under this specification are the products of the distillation and refining of petroleum unmixed with any other substances, and conforming to the detail specifications below. Products having a very offensive odor, or being mixed with other oils. will not be accepted.

3. Shipments must be made as soon as possible after the order is received. It will be observed that the detail specifications provide for a change of cold test and flashing point in some of the oils on May 1st and October 1st. Shipments reaching destination on or after these dates must conform to the specifications characteristic of these dates and will be rejected if they fail, unless it can be shown that they have been more than a week in transit. No preliminary examination of samples will be required, but a limited amount of special preliminary examinations will be made on the request of the Purchasing Agent for use of parties desiring the information. Definite printed methods for determining flashing and burning points, for making cold test and for taking gravity will be furnished if desired, and in case of dispute these methods must be used.

A shipment being received at any shops, one sample of not 4. less than a pint must be taken from any barrel at random, for each shipment of a carload or less, and sent by R. R. S. to the Chemist, Altoona, Pa. This sample must be accompanied by a "Sample for Test" tag properly filled out, and must be sent in a proper can, enclosed in a "Sample for Test" box. In taking the sample, care must be exercised to prevent contaminating the sample with any other oil or any other substance, and a clean, dry can must always be used. This sample will represent the shipment. If it stands the tests, the shipment will be accepted, except as provided in Section 5. If the sample fails to stand the tests, the shipment will be rejected and returned to the shippers, who must pay return freight.

5. The examination of a shipment for oil that is cloudy from glue or suspended matter must be made by those by whom the oil is received. The examination applies especially to 150 degree and 399 degree Fire Test Oils. As this defect rarely characterizes all of the bar-rels of a shipment, it is obvious that the sample for test may fail to show it. Accordingly when any barrel or barrels in a shipment are found to be cloudy from glue or suspended matter, such barrels must be set aside and returned to the shipper, notwithstanding the Test Report has shown the shipment to be ready for use.

The following detail specifications will be enforced: 6.

150° Fire Test Oil.

This grade of oil will not be accepted if sample from shipment: Is not "water white" in color. 1.

- 2.
- Flashes below 130° Fahrenheit. Burns below 151° Fahrenheit. 3.

4. Is cloudy or shipment has cloudy barrels when received, from the presence of glue or suspended matter.

5. Becomes opaque or shows cloud when the sample has been 10 minutes at a temperature of 0° Fahrenheit.

300° Fire Test Oil.

This grade of oil will not be accepted if sample from shipment: 1. Is not "water white" in color.

Flashes below 249° Fahrenheit. 2.

Burns below 298° Fahrenheit. 3.

Is cloudy or shipment has cloudy barrels when received, from 4. the presence of glue or suspended matter.

5. Becomes opaque or shows cloud when the sample has been 10 minutes at a temperature of 32° Fahrenheit.

6. Shows precipitation when some of the sample is heated to 450° Fahrenheit.

The precipitation test is made by having about two fluid ounces of the oil in a six-ounce beaker, with a thermometer suspended in the oil, and then heating slowly until the thermometer shows the required temperature. The oil changes color but must show no precipitation.

Paraffine and Neutral Oils.

These grades of oil will not be accepted if the sample from shipment:

1. Is so dark in color that printing with long primer type cannot be read with ordinary daylight through a layer of the oil ½ inch thick.

2. Flashes below 298° Fahrenheit.

Has a gravity at 60° Fahrenheit below 24° or above 35° 3. Baume'.

From October 1st to May 1st has a cold test above 10° Fahren-4. heit, and from May 1st to October 1st has a cold test above 32° Fahrenheit.

The color test is made by having a layer of the oil of the prescribed thickness in a proper glass vessel, and then putting the print-ing on one side of the vessel and reading it through the layer of oil with the back of the observer toward the source of light.

Well Oil.

This grade of oil will not be accepted if the sample from shipment: 1. Flashes, from May 1st to October 1st, below 298°F, or from October 1st to May 1st below 249°F. 2. Has a gravity at 60°F below 28° or above 31° Baume'.

From October 1st to May 1st has a cold test above 10°F, and 3. from May 1st to October 1st has a cold test above 32°F.

Shows any precipitation when 5 cubic centimeters are mixed 4. with 95 cubic centimeters of gasoline.

The precipitation test is to exclude tarry and suspended matter. It is made by putting 95 cc of 88 degree B. gasoline, which must not be above 80 degrees Fahrenheit in temperature, into a 100 cc graduate, then adding the prescribed amount of oil and shaking thoroughly. Allow to stand 10 minutes. With satisfactory oil no separated or precipitated material can be seen.

500° Fire Test Oil.

This grade of oil will not be accepted if sample from shipment: 1. Flashes below 494° Fahrenheit.

Shows precipitation with gasoline when tested as described for well oil.

Lubricating Oil

The principal source of lubricating oil is petroleum, from which the lighter components (naphtha and kerosene) have been removed by distillation, the residue thus obtained being used directly as a lubricant or separated by distillation into various fractions. By removing some of the fractions, as well as by mixing others, a variety of products may be obtained with special properties (viscosity, flash point, cold test and specific gravity).

This is the principle on which the industry is based. The separate fractions are further refined to remove odor, resinous materials, etc., as well as to attain the desired lightness of color. This is accomplished by means of sulphuric acid, agitating with a stream of air, the acid being later removed by washing with alkali or water; the purification may also be brought about by filtration through fuller's earth (customary in the United States).

In Europe the oil is distilled with superheated steam, recently also with partial vacuum, direct firing being avoided to prevent decomposition. The temperature of the superheated steam is kept somewhat higher than that of the still. Commercially, the distillates are cool and separated according to specific gravity, flash point and viscosit

In the United States direct firing is much used in separating the crude oil fractions, thus increasing the yield of illuminating oils. Therefining, however, is carried on with superheated steam.

ECONOMY OF LUBRICATION.

The economical transmission of power is largely dependent upon the maximum reduction of friction.

The purpose of lubrication is to overcome friction in so far as, possible and to prevent wear and deterioration of adjacent moving parts.

It is claimed that from 40% to 80% of all power produced by machinery is lost in friction, and a very considerable part of this is lost in avoidable friction due to improper lubrication.

THEORY OF LUBRICATION.

A lubricant should prevent direct contact between the bearings and the moving parts of machinery, thus substituting for metallic friction and wear the much smaller internal friction of the lubricant. The more completely this result is attained under the conditions of temperature, speed and pressure, the more valuable the lubricant from a mechanical point of view. Whether the mechanically most efficient lubricant is the most economical depends somewhat on the ratio of efficiency, the amount used and the price of the material. Greases have a low mechanical efficiency compared with liquid oils, but from the point of economy and cleanliness they are far superior.

Only liquids with great tendency to adhere are suited for lubrication, since only these have the property to penetrate by capillarity where journal and bearings are the closest and where the danger of contact and wear is the greatest. The lubricating oils prevent direct contact of the metal surfaces because of their adhesion to these surfaces and because their viscosity keeps them from being squeezed out by the pressure on the bearing.

Experience has shown that the power to adhere to metals increases with the viscosity of the oil. Since the danger that an oil will be pressed out increases with the pressure on the bearings, it is advisable for high pressures to use oils of considerable viscosity.

With low pressure and high speed there should be used a very mobile oil, with higher pressure and great velocity more viscous oils. If, for example, a spindle rotating with practically no pressure but very rapidly were lubricated with a very viscous oil, it would mean a vish waste of power. But to lubricate a transmission gear with a mobile oil would be a waste of lubricant, while the use of a heavy

rease would be entirely suitable. In fact, the use of a solid lubricant, raphite, with heavy oils as a vehicle, has proven most desirable in the case of very heavy bearings and transmission gears with enormous pressures.

The oil should not lose its power of reducing friction by evaporation, gumming or by acting chemically on the metal of the bearings or journal.

The oil or grease should not solidify or greatly change its viscosity under conditions of use.

PHYSICAL TESTS FOR LUBRICANTS.

1. Flash and burning points of lubricants are the respective tematures at which the vapors arise in sufficient amount to ignite and burn continuously. They should be high enough to prevent any ager of fire in using the oil and to be assured that a light oil has been added to a heavy oil to regulate viscosity. With the same osity asphaltic base oils (Texas, California and Mexico) have a ver flash point and a higher specific gravity than paraffin base oils ennsylvania and West Virginia).

2. Specific gravity is the relation of the weight of a given volume f oil to the weight of the same volume of water. The oil trade usually uses the Baume' scale of gravity, which is entirely arbitrary (see ables). The raraffin oils with the same viscosity are lighter (have a nigher gravity--Baume') than the asphaltic or semi-asphaltic oil. Gravity is not a measure of the quality of a lubricating oil.

3. Viscosity is the most important property for lubrication. The viscosity is expressed in the terms of the Saybolt Universal Viscosimeter in this country, the Engler in Germany and the Redwood in England (see conversion factors). Paraffin oils lose their viscosity most readily in use in an explosion cylinder by reason of the greater ease in decomposing to lighter products than do asphaltic oils (see also cracked lubricating oils). They tend to be more viscous at higher temperatures than naphthene base oils (note).

4. Carbon. The fixed carbon is a most harmful property in lubricants for explosion motors, such as automobiles. High fixed carbon is found in poorly refined and blended oils. It is higher in asphaltic than in Pennsylvania or Mid-Continent oils with the same refining. Less carbon is present in light oils.

5. Cold test determines the lowest temperature at which the oil will pour. A low cold test is desirable for ease in circulating and handling in cold weather. A low cold test for motor oils indicates the absence of heavy ends that produce excessive carbon in the cylinder.

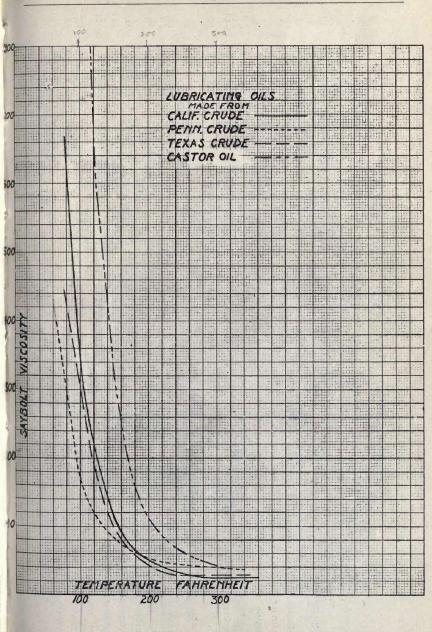
6. Color is not an index of the value of a lubricating oil. The lighter the color, other things being equal, the purer the oil.

7. Free acid should be, and usually is, absent. It is an indication of mineral acid that has not been neutralized and washed out in refining or of the presence of naphthenic acids.

The qualities of y	arious	lubricatio	ng oils a	re as fo	llows:	
Viscosity at Spindle	Light M'ch'n'	Heavy v M'ch'n'	Auto- v mobile	Engine	Steam	Large Cylinder
70°F	375-750	750-1875	470-1100	300-400		2800-4000
	180-220		160-400	130-150		
122°F75-90					1100	300-560
210°F	40-50	45-60	40-55	44-47	120-150	
Flash point, °F						
Min 140	160	390	350	430	525	450
Cold test, °F 10	5	10-40	10	25	45	40
Gravity, Be'			19-32	23-30	24-30	

Note.-See Lubricating Engineer's Handbook, by J. R. Battle.

KANSAS CITY TESTING LABORATORY



163

	DO	JLLI	SIII	1 1	0 11	DE	IL I'I	I.I.L	111	v c					
	Cold Test, F°	44	: :	40	35 34-36	32	30	28 26	20	:	55	20-25	0-4	20	:
	Fire Test, F°	069	600	500	450	450	490	450	470	445	550	470	420	470	445
	Flash Point, F°	620	550	450	400 400	400	440	$\frac{400}{395}$	420	395	500	420	350-360	420	395
ts	Saybolt Viscosity	250 @ 212°F	130-150 @ 212°F	300-325 @ 100°F	250-280 @ 100°F 190-240 @ 100°F	190-210 @ 100°F	150 @ 100°F	120 @ 100°F 75 @ 100°F	200 @ 100°F	150 @ 100°F	115 @ 212°F	200-380 @ 100°F	100 @ 70°F	200 @ 100°F	150 @ 100°F
ubricant	Baume Gravity Degrees	24.5-27	25-26.0	24.0	24.5 - 25 24 - 25	28-30	26.5	28.0 32.0	30.5	32.0	26.5	30.0	26-27	30.5	32.0
Properties of Various Lubricants	Trade Name or Machines and Conditions Per Cent Classification Steam engine; medium	Cylinder ollSteam enter	Steam e	treme)	Engine oil (heavy)Heavy bearings, lower duty than above	Engine oil (medium)	Crankcase oll (heavy)Steam engine; crankcase splashStraight mineral) splash	:	12	(neavy)	stage	:	:	Dynamo on (meanum)

164

BULLETIN NUMBER FIFTEEN OF

		KA	NS	AS	CT.	IY I	ES.	IING	LA	BO	RATO	JRY		- 6-
	18	5-15	40	30	:	:		20°F	35 50 10°F		1vania 0.2°Be' 3.0°Be'		% 0%	
	400	380	460	430	390	400		$\begin{array}{cccccccccccccccccccccccccccccccccccc$	550 450		No. 2 Pennsylvania .875 = 30.2°Be' .860 = 33.0°Be'	251	0.050%	600 40%
	345	325	410	380	340	360		210-420 415 435	440 485 380					
	90 @ 100°F	100 @ 70°F	300 @ 100°F	150 @ 100°F	H.01 (203 @ 100°F		212°F 100°F	100°F	DIL.	10.1 = 20.0°B = 22.3°B	652 229 0.00	0.08%	1000 8%
	906	100 @	300 @	150 @	60-150 @ 70°F	203 @		300-400 @ 212°F 2 185 @ 100°F 285 @ 100°F	470 0 135 0 652 0	ATING C	Sout 934 =	960		
	27.5	29.0	24.0	26.0	80-35	28.0		24.0 30.5 30.0 8	22.6	-UBRIC				
										I NO (I				••••
	nineral	nineral	nineral	mineral	nineral	rineral		Carbon Carbon	Carbon Carbon	EFFECT OF AIR-COOLED MOTOR (FRANKLIN) ON LUBRICATING OIL.	Crude from which manufactured South Texas Gravity before using $920 = 22.3^{\circ}$ Be Gravity tatfer using $920 = 22.3^{\circ}$ Be Viscosity at 100° to	Viscosity at 100° after use Free carbon before use	Free carbon after use. Contradion carbon before use.	Lubricating oil consumed
	HP; small ring oil bearingsStraight mineral ugh, slow speed bear-	ings, lineshafts, crush- ers, etc.; cheap workStraight mineral bists, elevators and gen-	etat macumery, cool runningStraight mineral	es, large lathes, efc Straight mineral ght machines for spin-	ning, automatics, etc. Straight mineral eaving machines; high	dium weight 		Pa. Petroleum	Tanger, Texas	FOR (FF				
2 to 5	ing oil bear-	workS	press-	etcS	s; high	weight rs and	ile and	N000	1 0.	ED MO	· · · · · · · · · · · · · · · · · · ·			
Small motors, 1-32 to 5	w speed	ings, lineshafts, crush- ers, etc.; cheap work	ng cools. print press-	es, large lathes, etc	Weaving machines; high	speed medium weight machines	feed lubrication of gasoline automobile and	eum	exas	R-COOL				
nall mot	HP; s bearings Rough, slo	ers, etc. bists, etc.	running tchine to	es, large	ning, au eaving	machines ts engine	feed 1 gasoline	. Petrol	unger, T xaco	T OF AI	be			• • •
S'r	Rc	y) H_0	ium).Ms	Li	W	Ga		Pa Pa	Re	EFFEC	ufactur	use	e use.	led
		ne (hear	ne (med								using	0° after fore use	ter use.	consum
Dynamo oil	Black machine	General machine (heavy)Holist, etc.; cheap work General machine (heavy)Holist, elevators and gen-	General machine (medium). Machine fools. print press-	Spindle oil	Loom oil	Gas engine oil		Motor (light) Motor (medium)	Cylinder oil.		from wl before	ty at 10 trbon be	son cart	ting oil
Dynam	Black	Genera	Genera	Spindle	Loom	Gas ei		Motor Motor	Cylind		Crude Gravity Gravity Viscosi	Viscosi Free ca	Conrad	Lubrice

KANSAS CITY TESTING LABORATORY

165

EFFECT OF CRACKING ON THE LUBRICATING QUALITIES OF OIL.

In the cracking of petroleum by heat the paraffin hydrocarbons are most readily accomposed into lighter hydrocarbons. The lubricating hydrocarbons remaining in cracked oil are therefore not paraffin but consist chiefly of naphthenes and aromatics. In other words, cracking reduces the viscosity of heavy hydrocarbon oils based (U. S. No. 1,167,834, Jan. 11, 1916) as follows: Lubricating fractions made from Mid-Continent Crude Petroleum

Baume' Gravity Viscosity at 100°

241	and citatio	15			`
		(2	Saybolt Visco	simeter	r)
	25.0		235		
	26.0		190		
	26.0		165		
	26.5		145		
	27.5		100		
Lubricating f	ractions made	from		opur	Petroleum
Bublicating	ume' Gravity	nom			retroteum
Dat			Viscosity a	1 100	
	18.8		449		
	20.4		235		
	20.6		339		
	21.6		146		
	21.8		167		
	22.5		139		
Lubricating	fractions made	from	Cracked Petr	oleum	Residua
Baume' Gravity			Gravity		Viscosity
28.9	36		15.2		88
26.5	38		15.0		89
23.8	42		14.7		97
21.5	45		14.1		105
21.1	51		13.2		110
20.2	52				
			13.0		116
18.7	58		12.0		158
17.8	62		10.8		198
17.2	65				
16.7	.66				
15.8	76				

166

NATURAL HYDROCARBONS—VACUUM DISTILLED. Table showing the properties of vacuum distilled hydrocarbons and atmospheric pressure forced fire distilled hydrocarbons of a heavy residuum from Mid-Continent oil.

residuam 110	m mu-Continent on.		
Fraction	Gravity	Viscosity	Sulphur
0-10%	0.868	46	0.39%
	31.3°Be'		and the second second
10-20%	0.877	60	0.35%
/ •	29.6°Be'		
20-30%	0.895	143	0.43%
	26.4°Be'		011070
30-40%	0,909	293	0.53%
00 10 /0	24.0°Be'		0.00 /0
40-50%	0.920	740	0.76%
10 00 /0	22.1°Be'	110	0.1070
50-60%	0.920	745	0.68%
00-0070	22.1°Be'	120	0.00/0
00 00d	0.920	1058	0 500
60-70%		1098	0.70%
	22.1°Be'		State of the state
70-80%	0.920	2600	0.56%
	22.1°Be'		

HYDROCARBONS FROM FORCED FIRE DISTILLATION OF SAME OIL

	SAME UIL.	
Fraction	Gravity	Viscosity
0-10%	0.864	51
	32.1°Be'	
10-20%	0.877	69
	29.6°Be'	
20-30%	0.888	109
	27.6°Be'	
30-40%	0.893	141
	26.7°Be'	
40-50%	0.894	141
	26.6°Be'	
50-60%	0.887	106
	27.0°Be'	
60-70%	0.878	75
	29.4°Be'	
70-80%	0.877	- 69
	29.6°Be'	

EFFECT OF TEMPERATURE ON VISCOSITY OF NATURAL MID-CONTINENT HEAVY OILS.

		Av'ge Mid-Conti- nent Fuel Oil 26.8°Be'	Heavy Kansas Crude 19.6°Be'
60°F	=	294.	
$70^{\circ}\mathrm{F}$	=	190.	3360.
100°F	=	94.	1250.
120°F	=	70.	680.
120°F		55.	328.
212°F	=	41.	105.
(Viscosity	is expressed in	terms of the Savbolt	Universal)

U. S. GOVERNMENT SPECIFICATIONS FOR LUBRICATING OILS

LIBERTY AERO OIL. SPECIFICATION NO. 3,501.

1. This specification covers the requirements of the army in all purchases of oil to be used for the lubrication of stationary cylinder aircraft engines.

2. It is intended to use the name "Liberty Aero Oil" for all oils approved for the lubrication of these engines. On account of the differences in characteristics of the high and low specific gravity oils, this specification is drawn to cover both types of oil and to in-clude products manufactured from crude petroleum oils from all fields. For the purposes of this specification oils are classified as follows:

CLASSIFICATION.

3. High Specific Gravity Oils-This class includes all oils having a specific gravity above 0.9100 or below 24 degrees Baume' conver-sion by the Tagliabue Manual, 9th edition, or below 25.85 degrees Baume' conversion by the Bureau of Standards' conversion table, Circular No. 57) and having a pour test below 15 degrees Fahrenheit. (Tested by the method of the American Society for Testing Materials.)

4. Low Specific Gravity Oils—This class includes all oils having a specific gravity below 0.9100 (or above 24 degrees Baume' conver-sion by the Tagliabue Manual, 9th edition, or above 23.85 degrees Baume' conversion by the Bureau of Standards' conversion table, Circular No. 57) and having a pour test above 15 degrees Fahrenheit. (Tested by the method of the American Society for Testing Materials.)

PHYSICAL PROPERTIES AND TESTS.

5. The oil must be made from pure, highly refined petroleum products, and must be suitable in every way for the entire lubrication of stationary cylinder aircraft engines operating under all conditions.

The oil must be neutral in action and must not show the 6. presence of moisture, sulphonates, soap, resin or tarry constituents which would indicate adulteration or lack of proper refining.

Viscosity-The viscosity of the oil, when tested in a Saybolt 7. Universal Viscosimeter at 212 degrees Fahrenheit, shall be as follows:

High specific gravity oil—70 seconds to 75 seconds. Low specific gravity oil—85 seconds to 90 seconds.

8. Pour Test—The oil must pass the following pour test: High specific gravity oil—not over 15 degrees Fahrenheit. Low specific gravity oil—not over 40 degrees Fahrenheit.

9. Flash Point-The oil must have a flash point over 350 degrees Fahrenheit in a Cleveland open cup.

10. Carbon-The oil must not show a carbon residue of over 1.5 per cent by the Conradson method. The carbon shown must be loose and flaky and must break up easily in the crucible.

11. Emulsion Test-One ounce of oil shall be placed in a standard four-ounce sample bottle with one ounce of distilled water. The mixture shall be heated to a temperature of 180 degrees Fahrenheit, and then shaken vigorously for five minutes. After standing for one hour, the oil must be clear and of the same color as before the test. All of the water must have settled and appear only slightly cloudy.

12. All tests must be made in accordance with methods adopted by the American Society for Testing Materials. Detailed descrip-tions of the Conradson carbon test and the pour test have been re-printed in Army Specification No. 3,525 which will be furnished on application.

MOTOR OIL FOR GASOLINE ENGINES. SPECIFICATION NO. 3,502.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer /Corps, the Medical Corps and the Signal Corps, United States Army.

GENERAL.

1. This specification covers the requirements of the Signal Corps for motor oil to be used for the lubrication of internal combustion 2. The oil shall be supplied in three grades—light, medium and

heavy. The light oil shall be used where specially specified. The medium oil shall be for general use in winter and for use in new engines at all times. The heavy oil shall be for general use in summer and for use in old engines.

PHYSICAL PROPERTIES AND TESTS.

3. The oil must be a refined and filtered mineral oil, or a mixture of such oils. It must be suitable in every way for the satisfactory lubrication of the internal combustion engines specified above.

4. Viscosity-The viscosity of the three grades of oil, when tested in a Saybolt viscosimeter at 100 degrees F., must be within the following limits:

Light oil-270 seconds to 230 seconds.

Medium oil-270 seconds to 330 seconds.

Heavy oil-470 seconds to 530 seconds.

5. Carbon-The carbon residue, determined by the Conradson method, must be as follows:

Light oil-not more than 0.2 per cent.

Medium oil-not more than 0.4 per cent.

Heavy oil-not more than 0.6 per cent.

6. Pour Test—One ounce of the oil must not congeal in a standard 4-ounce sample bottle when exposed to the following temperatures:

Light oil—25 degrees F. Medium oil—30 degrees F.

Heavy oil—40 degrees F. 7. All tests shall be made in accordance with methods adopted by the American Society for Testing Materials. Detailed descriptions of the Conradson carbon test and the pour test have been reprinted in army specification No. 3,525, which will be furnished on application.

AIRPLANE MACHINE GUN OIL.

SPECIFICATION NO. 3,503.

GENERAL.

1. This specification covers the requirements of the Army for gun oil for the lubrication of machine guns on airplanes and for gun oil for cleaning and oiling machine guns and small arms.

PHYSICAL PROPERTIES AND TESTS.

2. The oil must be a highly refined, highly filtered straight-run mineral oil, suitable in every way for the uses specified in paragraph 1. It must be a pure petroleum product, without the addition of

vegetable or animal oils or fats of any kind, and must contain no moisture.

3. The oil must be free from acids and from any material which might gum or corrode metals under any conditions.

4. Viscosity—The viscosity, when the oil is tested in a Saybolt universal viscosimeter at 100 degrees F., shall be as follows:

Seventy seconds to 95 seconds.

5. Acidity—The acidity of the oil must not be more than 0.03 per cent calculated as SO₃.

6. Carbon—The carbon residue must not be more than 0.003 per cent when determined by the Conradson method.

7. Pour Test—One ounce of the oil must not congeal in a standard 4-ounce bottle at 45 degrees below zero F.

8. All tests must be made in accordance with methods adopted by the American Society for Testing Materials. Detailed descriptions of the Conradson carbon test and the pour test have been reprinted in army specification No. 3,525, which will be furnished on application.

TRANSMISSION LUBRICANT.

SPECIFICATION NO. 3,504.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medical Corps and the Signal Corps, United States Army.

GENERAL.

1. This specification covers the requirements of the Army for a very adhesive mineral oil, which must be suitable in every way for the lubrication of transmission gears and bearings, differential gears, warm drives, winch drives and roller and ball bearings used in connection with such parts of the equipment of motor vehicles.

CHARACTERISTICS.

2. The lubricant must be a petroleum product only, without the addition of vegetable or animal oils or products, or residues or fats of any kind. It must be entirely free from fillers such as talc, resin, tar and all materials of every nature not related to the original product.

Physical Properties.

3. Viscosity.—The viscosity must be within the following limits when the lubricant is tested in a Saybolt universal viscosimeter at 212 degrees F.: 195 seconds to 220 seconds.

4. Adhesiveness.—The adhesiveness of the lubricant is one of the most essential qualities. As there is no satisfactory laboratory method for its determination, the adhesiveness will be determined by applying the lubricant to a set of gears operating under practical conditions and comparing the effect produced by the lubricant with the effect produced by a standard sample of army specifications No. 10 under the same conditions.

NON-FLUID TRANSMISSION LUBRICANT.

Specification No. 3,505.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medical Corps and the Signal Corps, United States Army.

General.

This specification covers the requirements of the Army for purchases of non-fluid transmission lubricant to be used for the axles and transmissions of motor trucks.

Physical Properties and Tests.

The lubricant shall be composed of calcium soap and mineral 2. oil manufactured in accordance with the best commercial process. It must have a consistency similar to that known to the trade as "No. 00 grease."

3. The lubricant must be a boiled grease containing not less than nor more than 1½ per cent of moisture in the finished product.
 4. Mineral Oil Base.—The mineral oil used in reducing the soaps,

when tested in a Saybolt universal viscosimeter at 100 degrees F.,

must show a viscosity of not less than 180 seconds. 5. Saponifiable Fat Base.—Not more than 10 per cent of either pure tallow oil, neatsfoot oil, lard oil or horse oil, singly or in combination, shall be used as a fat base.

Acidity .- The lubricant must not attack a sheet of polished 6. copper within a period of 48 hours.

7. Heat Test .- Two ounces of the grease shall be heated to 212 degrees F., or until the entire mass becomes liquid, and then allowed to cool. The soaps must not separate from the oils during this test, and the grease must return to its original consistency.

8. Fillers.—The grease shall contain no fillers, such as resin, resinous oils, soapstone, wax, talc, powdered mica, lamp black, sulphur, clay, asbestos or any other artificial thickening.

MEDIUM CUP GREASE.

Specification No. 3.506.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medical Corps and the Signal Corps, United States Army.

General.

1. This specification covers the requirements of the Army for a medium cup grease to be used for the lubrication of such parts of motor equipment and other machinery as are lubricated by means of compression cups.

2. The grease must be a well manufactured product, composed of calcium soap and mineral oil.

Physical Properties and Tests. 3. Mineral Oil Base.—The mineral oil used in reducing the soaps must show a viscosity of at least 180 seconds when tested in Saybolt universal viscosimeter at 100 degrees F.

4. Saponifiable Fat Base .- The grease must have a fat base of 15 to 20 per cent of either pure tallow oil, neatsfoot oil, lard oil or horse oil, used singly or in combination.

Consistency .-- The grease must be a medium cup grease simi-5. lar in consistency to that known to the trade as "No. 3 cup grease."

Moisture.-The grease must be a boiled grease containing not 6. less than 1 nor more than 3 per cent of moisture when finished.

7. Acidity.—The grease must not attack a sheet of polished copper within a period of 48 hours.

Ash .-- The ash shall not be greater than 2 per cent.

Heat Test .- Two ounces of the grease shall be heated to 212 9. degrees F., or until the entire mass becomes liquid and then allowed to cool. The soaps must not separate from the oils during this test, and the grease must return to its original consistency.

10. Fillers.-The grease must contain no fillers, such as resin, resinous oils, soapstone, wax, talc, powdered mica, lamp black, sulphur, clay, asbestos or any other filler or artificial thickening.

GUN OIL.

Specification No. 3,507.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medical Corps and the Signal Corps, United States Army.

General.

This specification covers the requirements of the Army for 1. gun oil to be used for the following purposes and where airplane machine gun oil (Specification No. 3,503) is not required:

For cleaning and oiling guns and small arms. For filling recoil cylinders of artillery and naval guns.

For oil switches and oil current breakers.

For transformers up to 6,600 volts.

For lubrication of the compressor and expander cylinders of ice machines.

For lubrication of pneumatic tools.

For hydraulic systems.

Physical Properties and Tests.

2. The oil must be a straight-run highly refined and highly fil-tered mineral oil, suitable in every way for the uses listed in paragraph 1.

3. The oil must be a petroleum product only, free from vegetable or animal oils or fats of any kind and entirely free from moisture.

Specific Gravity .-- The oil must have a Baume' gravity of not more than 23 degrees at a temperature of 60 degrees F.

5. Viscosity.-The viscosity must be within the following limits when the oil is tested in a Saybolt universal viscosimeter at 100 degrees F.: 95 seconds to 105 seconds.

6. Flash Point.-The flash point of the oil must not be less than 300 degrees F. in a Cleveland open cup.

7. Pour Test.—One ounce of the oil must not congeal in a standard sample bottle at 5 degrees below zero F.

8. Carbon .- The carbon residue must not be more than 0.003 per cent by the Conradson method.

Acidity.-The oil must not show an acid reaction of more 9. than 0.03 per cent, calculated at SO₃, and must not gum or corrode metals under any conditions.

10. All tests must be made in accordance with methods adopted by the American Society for Testing Materials. Detailed descriptions of the Conradson carbon test and the pour test have been reprinted in Army Specification No. 3,525, which will be furnished on application.

GEAR, CHAIN, WIRE ROPE LUBRICANT.

Specification No. 3,508.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medi-cal Corps and the Signal Corps, United States Army.

General.

1. This specification covers the requirements of the Army for a very adhesive, heavy-bodied, straight mineral oil, which must be suitable in every way for the following uses:

For the lubrication and protection of chains, wire ropes and gears of cranes, dredges, steam shovels and all other heavy equipment.

For the lubrication and protection of the gears and ropes of balloon hoists.

For swabbing the wires and cables of airplanes and seaplanes.

For slushing and protecting the bright and exposed metal parts of guns, machines and automobiles during storage or overseas shipment. When used for this purpose the lubricant shall be mixed with an equal amount of kerosene, so that it may be applied with a brush.

2. Kerosene may be used to remove this lubricant from the equipment.

Physical Properties and Tests.

3. The quality of the lubricant must be equal to or better than that of a standard sample of No. 1 wire rope lubricant, sample of which will be furnished by the Quartermaster-General, Fuel and Forage Division, Washington, D. C.

4. The lubricant must be a petroleum product only, free from vegetable or animal oils or products or residues or fats of any kind. It must be entirely free from fillers, such as talc, resin, tar and all materials of every nature not related to the original product.

5. Viscosity.—The viscosity must be within the following limits when the lubricant is tested in a Saybolt universal viscosimeter at 212 degrees F.: 900 seconds to 1,100 seconds.

6. Adhesiveness.—The adhesiveness of the lubricant is one of its most essential qualities. As there is no satisfactory laboratory method for the determination of this quality, the adhesiveness will be determined by applying the lubricant to a set of gears operating under practical conditions and comparing the effect produced with that produced by a standard sample of No. 1 wire rope lubricant mentioned above under the same conditions.

7. Corrosion Test.—When applied to a plate of polished steel the lubricant must protect the steel for a period of thirty days from chemical vapors, from the action of salt or fresh water and from the action of water containing from 10 to 25 per cent of sulphuric acid.

For the purposes of these tests the water and solutions shall be held at a temperature of 60 degrees F.

Drying Test.

8. When the lubricant is applied to a wire rope that has not been oiled with any other material, it must not crack, peel or chip after exposure to low atmospheric temperatures for sixty days.

Penetration Test.

9. When applied hot to the outside of a 1-inch wire rope that has not been oiled with any other material the lubricant must penetrate to and be absorbed by the fiber core, and at the end of sixty days, when the rope is put under strain, the oil must be forced out of the core between the wires of the strand.

MINERAL CYLINDER OIL. Specification No. 3,509.

This specification has been approved and adopted by the Ordnance Department, the Quartermaster Corps, the Engineer Corps, the Medical Corps and the Signal Corps, United States Army.

General.

1. This specification covers the requirements of the army for mineral cylinder oil known to the trade as "600 Steam Refined Cylinder Oil," to be used for steam engine lubrication, where a mineral oil is required, also as a stock oil for compounding, and as a light transmission lubricant for motor vehicles.

Physical Properties and Tests.

2. The oil must be a well refined, unfiltered oil, without compounding of any nature. It must be free from moisture, dirt and all foreign matter.

3. Viscosity.—The viscosity must be within the following limits when the lubricant is tested in a Saybolt universal viscosimeter at 212 degrees F.: 135 seconds to 165 seconds.

4. Flash Point.—The flash point of oil must be more than 475 degrees Fahrenheit in a Cleveland open cup.

5. Pour Test.—One ounce of the oil, in a standard 4-ounce sample bottle, must not congeal at 45 degrees F.

6. All tests must be made in accordance with methods adopted by the American Society for Testing Materials. Detailed description of the pour test has been reprinted in Army Specification No. 3,525, which will be furnished on application.

COMPOUND CYLINDER OIL. Specification No. 3,510.

This specification has been approved and adopted by the Signal Corps, Quartermaster Corps and the Engineer Corps, United States Army.

General.

1. This specification covers the requirements of the Army for compound cylinder oil to be used for the lubrication of steam cylinders of engines and pumps, where a compounded oil is required.

Physical Properties and Tests.

2. The cil must be a well refined, clean, mineral cylinder oil, known to the trade as "600 Steam Refined Cylinder Oil." It must be compounded with from 5 to 10 per cent of tallow oil. The finished oil must be free from moisture, dirt and all foreign matter.

3. Viscosity.—The viscosity, when the oil is tested in a Saybolt universal viscosimeter at 212 degrees F. must be as follows: 135 seconds to 150 seconds.

4. Flash Point.—The flash point of the oil must be over 475 degrees F. in a Cleveland open cup.

5. Pour Test.—One ounce of the oil must not congeal in a standard 4-ounce sample bottle at 50 degrees F.

6. All tests shall be made in accordance with methods adopted by the American Society for Testing Materials. Detailed description of the pour test has been reprinted in Army Specification No. 3,525, which will be furnished on application.

Lubricating Refinery Terminology

Aeroplane Oil.—This is oil used for aeroplanes, typical of which is the Liberty aeroplane oil under Specification No. 3501, page 168. It is frequently a light colored, straight production, viscous neutral oil.

Automobile Oils are usually viscous neutral oils with a flash point above 400°F and a Saybolt viscosity over 145 at 70°F. Axle Oil is a natural black lubricating oil, commonly summer

Axle Oil is a natural black lubricating oil, commonly summer black oil of a 500°F to 550°F fire test. It is used also as a tempering oil.

Bolt Oil is a viscous neutral oil with a gravity of about 30° Baume', viscosity of about 220, used for thread cutting.

Brick Oil is a light non-viscous neutral oil with a gravity about 34° Baume', a flash point of about 340° F, viscosity of about 80 at 70° F. It is also known in terms of Paint Oil.

Car Oil is the same as axle oil and summer black oil.

Castor Oil is oil from castor beans; has a specific gravity of .965 Baume', gravity of 15° , cold test of 5° F, viscosity on the Saybolt universal viscosimeter at 100° F of 1200

$125^{\circ}F$	600
$150^{\circ}F$	300
175°F	175
$200^{\circ}F$	110
250° F	60
300°F	40

Claroline Oil has a viscosity of 4.4° Engler at 70° F. It is essentially the same as straw oil or absorption oil.

Condenser, Compounded and Blown Oils are mixtures of mineral lubricating oils with seed oil, the seed oil usually being blown to increase the viscosity.

Cylinder Oil or Cylinder Stock is the residue obtained from distilling special grades of light crude oils with a very large amount of steam, avoiding cracking as much as possible, and from which the wax distillate has been removed. Cylinder oil varies in gravity from 20° to 27° Be', flash point 475° F to 650° F, viscosity at 210 Saybolt, 100 to 350, cold test 30 to 60° F. They usually are not filtered, but may be refined by filtering through Fuller's earth or bone black.

Cove Oil is 36° gravity mineral oil compounded with seed oil.

Cream Separator Oils are non-viscous oils of about 30 to 34° gravity, 70 to 200 viscosity at 70°F.

Cup Greases are mixtures of petroleum oil and lime soap with or without rosin oil.

Dynamo Oil is a viscous neutral oil, gravity 30-32°Be', flash point 400-425°F, fire test 450-500°F, cold test 15-30°F, Saybolt viscosity 140-225.

Engine Oil is a variable quality of lubricating oil, a common type of which has a Saybolt viscosity of 180-300, cold test 20-30°F, flash point of 400°F, gravity of 29°Be'.

Floor Oil is a light non-viscous neutral oil.

Gear Case Oil is a steam refined cylinder oil with a gravity of about 25°Be', flash point 600°F, cold test of 30°F, Saybolt viscosity at 210°F of 240.

Hammer Oil is a steam cylinder oil with a viscosity of about 220.

Harness Oil is a compounded oil containing petrolatum, leather oil and wax and some fatty oils.

Knitting Machine Oil is a spindle oil of 70-200 viscosity at 70°F. Leather Oil is a non-viscous neutral oil of low viscosity.

Machine Oil is made in various grades with viscosities of 290 to 800 Saybolt at 70°F. It is usually a red oil with a cold test of about 30°F.

Motor-Cycle Oil is a high viscosity lubricating oil similar to aeroplane oil.

Neutral Oils are oils obtained from pressed distillates.

Non-Viscous Neutral Oils are oils having a viscosity below 135 Saybolt at 100° F.

Viscous Neutral Oils are oils having a viscosity above 135 at 100°F.

Mineral Seal Oils are heavy burning oils obtained in the distillation for cylinder stock.

Oil Dag is a compound of deflocculated graphite suspended in petroleum lubricating oil covered by U. S. Patent No. 911,358 by Acheson.

Paraffin. Oil is the wax-free oil obtained by pressing wax distillate.

Paraffin Scale is crude wax.

Sweated Wax is crude wax freed from oil.

Refined Wax is sweated wax which has been filtered and decolorized with Fuller's earth.

Pressed Distillate or Pressed Oil is the oil after the wax distilled has been refrigerated and the wax removed from it.

Petroleum Coke is the residue in coking or tar stills and usue constitutes about 5% of the crude oil. Mid-Continent crude leaves residue ordinarily about 6 inches thick in the still, and Mexican cruc petroleum leaves a residue about 30 inches thick in the bottom of the still.

Roll Oil for tin, copper and brass rolls has the same qualities a engine oil.

Sewing Machine Oil is light neutral oil with a viscosity of 75 at 70°F, cold test 20°F or below, fire test 400°F, flash point 340° F, a gravity of 34.5° Be'.

Spindle Oils are the lighter lubricating oils usually of a gravity of $25-35^{\circ}$ Be', flash point $300-450^{\circ}$ F, viscosity 40-400 at 70° F, cold test $0^{\circ}-40^{\circ}$ F, a colorless to dark red.

Stitching Oil is a light non-viscous neutral oil used for stitching shoes.

Summer Black Oil is a black lubricating oil of about 500-600 fire test and is used for tempering and for concrete waterproofing.

Tar Stills or Coking Stills are large oil stills in which heavy portions of crude petroleum are distilled by intense firing until nothing but coke remains in the bottom of the still. This sort of distillation is desirable to produce a wax distillate from which the crystalline wax may be easily filtered.

Tower Stills are types of crude oil stills used to prevent the accumulation of coke in the still and does not subject the oil to as much cracking in the still as in the ordinary method.

Tempering Oil is a viscous neutral oil, frequently the same as hammer oil and summer black oil.

Thread Cutting Oil is viscous neutral oil of the quality of engine oil. It is sometimes compounded with 20-40% of lard oil.

Thickened Oils are mineral oils in which the viscosity is increased by the addition of unvulcanized rubber, aluminum soap or blown vegetable oil.

Transformer Oils are oils free from acid with flash point over 300°F and viscosity of about 100.

Transmission Oil is steam cylinder oil of a viscosity of 245 at 210°F.

Turbine Oil is a non-emulsifying oil of about 150 viscosity at 70° F and a flash point of about 420° F.

Vaseline is a semi-solid paraffin oil or wax composed of sufficient varieties of petroleum hydrocarbons to give an indistinct melting point.

Watch Oil is usually a non-petroleum oil and is ordinarily Dolphin oil.

Winter Oil is an engine oil having a cold test of -20°F.

Wax Distillate is the distillate coming from the coking stills and containing wax in crystalline form which immediately follows the gas oil.

Wool Oil is a sun bleached neutral oil sometimes compounded with lard oil and with a viscosity of 140-160 Saybolt, gravity of about $32^{\circ}Be'$ and flash point of $375^{\circ}F$.

PETROLATUM LIQUIDUM, U. S. P. Liquid Petrolatum.

ate

Petrolat. Lig.-Liquid Paraffin, Mineral Oil.

A mixture of liquid hydrocarbons obtained from petroleum. Precrve it in well closed containers, protected from light.

Heavy Liquid Petrolatum.—Heavy Liquid Petrolatum has a vissity of not less than 3.1 when determined by the test given below.

Light Liquid Petrolatum.—Light Liquid Petrolatum has a viscosty of not more than 3 when determined by the test given below and vaporizes freely.

Each variety conforms to the following description and tests:

Specific gravity for Liquid Petrolatum, 0.828 to 0.905 at 25°C.

A colorless, transparent, oil liquid, free or nearly free from fluorescence, odorless and tasteless when cold and possessing not more than a faint petroleum odor when heated. When cooled to 10°C Liquid Petrolatum does not become more

When cooled to 10°C Liquid Petrolatum does not become more than opalescent (solid paraffins).

Insoluble in water or alcohol; soluble in ether, chloroform, petroleum benzin or in fixed or volatile oils. Camphor, menthol, thymol and many similar substances are dissolved by Liquid Petrolatum.

Boil 10 mils. of Liquid Petrolatum with an equal volume of alcohol, the alcoholic liquid is not acid to litmus (acids).

Introduce into a glass-stoppered cylinder which has been previously rinsed with sulphuric acid 5 mils. of Liquid Petrolatum and 5 mils. of colorless sulphuric acid, heat in a water bath during 10 minutes, shaking well at intervals of 30 seconds; the oil remains unchanged in color and the acid does not become darker than pale amber (carbonized impurities).

Prepare a clear, colorless saturated solution of lead oxide in an aqueous solution of sodium hydroxide (1 in 5) and mix 2 drops of

this solution with 4 mils. of Liquid Petrolatum and 2 mils. of dehydrated alcohol; the mixture does not darken after heating for 10 minutes at 70°C and cooling (sulphur compounds).

Viscosity.-Make a permanent mark about 2 cm. below the bulb of a 50 mil. pipet of the usual type and note the time in seconds required at 25°C for the level of distilled water to fall from the upper to the lower mark as the liquid flows from the pipet. The time should not be less than 25 seconds nor more than 30 seconds for the pipet selected.

Draw the Liquid Petrolatum to be tested into this pipet, which should be clean and dry, and note the time in seconds required at 25°C for its level to fall from the same upper to the lower mark as used for the water. Divide the number of seconds thus noted by the number of seconds required for water to fall from the upper to the lower mark as above determined. The quotient indicates the viscosity. Distilled water at 25°C is taken as 1.

Average Dose.-Metric, 15 mils.; apothecaries, 4 fluidrachms.

PETROLATUM, U. S. P.

Petrolat.-Petrolatum Ointment. Petroleum Jelly.

A purified mixture of semi-solid hydrocarbons obtained from petroleum.

Petrolatum is an unctuous mass, varying in color from yellowish to light amber, having not more than a slight fluorescence even after being melted. It is transparent in thin layers, completely amorphous, free or nearly free from odor or taste.

Petrolatum is insoluble in water, almost insoluble in cold or hot alcohol or in cold dehydrated alcohol, freely soluble in ether, chloroform, carbon bisulphide, oil of turpentine, petroleum benzin, benzene or in most fixed or volatile oils.

Specific gravity, 0.820 to 0.865 at 60°C.

It melts between 38° and 54°C.

Heat about 2 gms. of Petrolatum in an open porcelain or platinum dish over a Bunsen burner flame. It volatilizes without emitting an acrid odor and on incineration not more than 0.05% of ash remains.

Shake melted Petrolatum with an equal volume of hot distilled water; the latter remains neutral to litmus (acid or alkalies).

Digest 10 grams of Petrolatum at 100°C for half an hour with 10 gms. of sodium hydroxide and 50 mils. of distilled water, then separate the aqueous layer and supersaturate it with sulphuric acid; no oils or solid substance separates (fixed oils, fats or rosin).

PETROLATUM ALBUM. U. S. P. White Petrolatum.

Petrolat. Alb .- White Petroleum Jelly.

Petrolatum wholly or nearly decolorized. White Petrolatum is a white or faintly yellowish unctuous mass, transparent in thin layers even after cooling to 0° C, completely amorphous.

In other respects White Petrolatum has the characteristics of and responds to the tests for identity and purity under Petrolatum.

PARAFFINUM, U. S. P. Paraffin.

A purified mixture of solid hydrocarbons usually obtained from petroleum.

Paraffin is a colorless or white more or less translucent mass, crystalline when separating from solution, without odor or taste and slightly greasy to the touch.

It is insoluble in water or alcohol, slightly soluble in dehydrated alcohol, freely soluble in ether, petroleum benzine, benzene, carbon disulphide, volatile oils or in most warm fixed oils.

Specific gravity, about 0.900 at 25°C.

It melts between 50° and 57°C.

When strongly heated it ignites, burns with a luminous flame and deposits carbon.

Heat about 0.5 gm. of paraffin in a dry test tube with an equal weight of sulphur; the mixture becomes black from separated carbon and hydrogen sulphide gas is evolved.

Paraffin is not acted upon or colored by concentrated sulphuric or nitric acid in the cold.

Shake melted paraffin with an equal volume of hot alcohol; the separated alcohol does not redden moistened blue litmus paper (acids).

Ichthyol is an artificial preparation obtained by the distillation of certain bituminous shales and subsequent sulphonation and neutralization with ammonia or soda. It comes on the market under the official name of Ammonii Icythyo-sulphonas or Ammonium Sulpho-ichthyolate. The specific gravity of the preparation is approximately 1.0, and it has a viscosity of 17.7 (Engler). A typical preparation contains 15% to 16% of sulphur, and it is to the sulphur that the value of the preparation is largely due. On account of the difficulty in duplicating exactly the original product and the scarcity of the original product, it has now attained a very high price.

Paraffin Wax

Paraffin wax is valued by the color, melting point and the specific pravity. The price of the crude wax having a melting point and the specific 103°F to 108°F is about 6c per pound, while the highly refined wax having a melting point of up to 140°F is worth about 17c per pound. Paraffin wax is ordinarily obtained from petroleum; also from shale oil and ozocerite. Paraffin exists in crude petroleum in the form of protoparaffin, in which condition it does not crystallize out and cannot be expressed from oil at low temperatures. In order to obtain it 6.0 Grams per 100 cc 1. Splability of Paraffin wax in gasoline *50. G. 0678 = 76.5° Be 5.0 2. Solubility of parolin wax in benzol. 4.0 3.0 20 1.0 Temperature 20° F 30°F 40°F 50°F 60°F 70°F

in condition for refrigeration and filtration, the heavy oil is subjected to a destructive distillation, thereby producing the crystalline pyroparaffin.

Pennsylvania petroleum furnishes from $1\frac{1}{2}\%$ to 2% paraffin wax, some petroleum such as one in Roumania giving as much as 10%.

The wax distillate from which paraffin is obtained contains ordinarily about 10% of wax. This distillate has a gravity of from $33^{\circ}Be'$ to $35^{\circ}Be'$ and distills over at a temperature of $500^{\circ}F$ to $700^{\circ}F$. The paraffin is freed from oil by the sweating process after filtration.

Fuel Oil

Petroleum as a fuel for use in steam plants has considerable variation, the only feature common to all oils coming under this class being that it is free from gasoline.

The gravity varies according to the character of the oil and the
amount of light constituents that have been distilled out of it. The
following table shows typical gravities of fuel oil from different
sources. Gravity
Mexican fuel oil12.6°Be'
Paraffin base fuel oil27.5°Be'
California fuel oil
Towanda fuel oil
Mid-Continent heavy fuel oil
Typical Mid-Continent oil
Garber, Oklahoma, fuel oil31.3°Be'

The chief property making fuel oil available for use is the ease with which it flows, or its viscosity. The viscosity is not proportional to the gravity, as is indicated by the following table: Viscosity and Gravity of Fuel Oils (See Pages 187-8).

SourceGravityat 70° FCalifornia crude16.9°Be'5400Residuum from same after cracking15.5414Heavy Kansas crude19.73360Residuum from same after cracking21.2178Heavy Mid-Continent fuel oil23.5810Residuum from same after cracking21.2135Garber, Oklahoma, fuel oil31.3183Residuum from same after cracking28.070Mexican heavy flux oil10.814500Residuum from same after cracking12.6530Average Mid-Continent fuel oil27.5272Residuum from same after cracking23.788			Viscosity
California crude 16.9°Be' 5400 Residuum from same after cracking 15.5 414 Heavy Kansas crude 19.7 3360 Residuum from same after cracking 21.2 178 Heavy Mid-Continent fuel oil 23.5 810 Residuum from same after cracking 21.2 135 Garber, Oklahoma, fuel oil 31.3 183 Residuum from same after cracking 28.0 70 Mexican heavy flux oil 10.8 14500 Residuum from same after cracking 12.6 530 Average Mid-Continent fuel oil 27.5 272	Source	Gravity	at 70°F
Heavy Kansas crude.19.73360Residuum from same after cracking.21.2178Heavy Mid-Continent fuel oil.23.5810Residuum from same after cracking.21.2135Garber, Oklahoma, fuel oil.31.3183Residuum from same after cracking.28.070Mexican heavy flux oil.10.814500Residuum from same after cracking.12.6530Average Mid-Continent fuel oil.27.5272	California crude	16.9°Be'	5400
Residuum from same after cracking.21.2178Heavy Mid-Continent fuel oil.23.5810Residuum from same after cracking.21.2135Garber, Oklahoma, fuel oil.31.3183Residuum from same after cracking.28.070Mexican heavy flux oil.10.814500Residuum from same after cracking.12.6530Average Mid-Continent fuel oil.27.5272	Residuum from same after cracking	15.5	414
Heavy Mid-Continent fuel oil.23.5810Residuum from same after cracking.21.2135Garber, Oklahoma, fuel oil.31.3183Residuum from same after cracking.28.070Mexican heavy flux oil.10.814500Residuum from same after cracking.12.6530Average Mid-Continent fuel oil.27.5272	Heavy Kansas crude	19.7	3360
Residuum from same after cracking.21.2135Garber, Oklahoma, fuel oil.31.3183Residuum from same after cracking.28.070Mexican heavy flux oil.10.814500Residuum from same after cracking.12.6530Average Mid-Continent fuel oil.27.5272			178
Residuum from same after cracking.21.2135Garber, Oklahoma, fuel oil.31.3183Residuum from same after cracking.28.070Mexican heavy flux oil.10.814500Residuum from same after cracking.12.6530Average Mid-Continent fuel oil.27.5272	Heavy Mid-Continent fuel oil	23.5	810
Residuum from same after cracking	Residuum from same after cracking	21.2	135
Mexican heavy flux oil10.814500Residuum from same after cracking12.6530Average Mid-Continent fuel oil	Garber, Oklahoma, fuel oil	31.3	183
Residuum from same after cracking12.6530Average Mid-Continent fuel oil27.5272	Residuum from same after cracking		70
Average Mid-Continent fuel oil	Mexican heavy flux oil	10.8	14500
			530
Residuum from same after cracking23.7 88	Average Mid-Continent fuel oil	27.5	272
			88

Fuel oil has a remarkably constant heating value based on British thermal units or bound of oil. Oil free from water has a higher B.T.U. per pound and a lower B.T.U. per gallon, the lighter the oil; and a lower B.T.U. per pound and a higher B. T. U. per gallon, the heavier the oil. This is set forth in the curves on page 189.

As compared with other sources of heat the theoretical amount of heat obtainable from petroleum or fuel oil as determined when the combustion is complete and the absorption of heat is complete is as follows:

1,000,000 B. T. U. of Petroleum @ \$1.00 per bbl. costs......\$0.165 1,000,000 B. T. U. of Cherokee slack coal @ \$3.00 per ton.... 0.136 1,000,000 B. T. U. of natural gas @ \$0.30 per 1,000 cu. ft. =... 0.33 1,000,000 B. T. U. of coal gas @ \$0.50 per 1,000 cu. ft. =... 0.79 1,000,000 B. T. U. of electricity @ 1c per k.w.hour =..... 2.93 The above data is from the following: Fuel oil based on specific

gravity $0.900 = 25.7^{\circ}$ Be', weight per gallon, 7.5 lbs., weight per barrel 315 lbs.

B. T. U. per lb. = 19,225, per ton = 38,450,000, per gallon = 144,200, cubic foot = 1,078,500, per barrel = 6,056,000.

Slack coal = 11,000 B. T. U. per pound. Natural gas = 900 B. T. U. per cubic foot.

Equivalents.

1 ton of coal = 3.6 bbls. oil = 24,500 cu. ft. of natural gas.

1 gallon of oil = 13.1 lbs. coal = 160 cu. ft. of natural gas.

1 barrel oil = 0.278 ton coal = 680.6 cu. ft. of natural gas.

1 pound oil = 1.75 lbs. coal = 21.3 cu. ft. of natural gas.

1 pound coal = 0.763 gallon oil = 12.2 cu. ft. of natural gas.

As to the actual heating value of fuel oils from various sources the following is representative:

Heating Value of Fuel Oils.

	Mid Con	1-				
	tinent	Light	Heavy			
	Av'ge	Mid	Mid	To'da		
	1255	Con-	Con-	Fuel	Gas	Mexi-
	Samples	tinent	tinent	Oil	Oil	can
Specific gravity	0.892	.863	0.922	0.921	0.856	0.975
Baume' gravity	26.9	32.2	21.8	22.0	33.5	12.6
Weight per gal. (lbs)	7.43	7.18	7.68	7.67	7.13	8.25
Heat value B. T. U.						
per pound	19376	19580	19170	19175	19635	18710
Heat value B. T. U.						
per gallon	.143950	140580	157220	147600	139990	154360
Flash point	$125^{\circ}F$	$110^{\circ}F$	$132^{\circ}F$	180°F	$170^{\circ} F$	100°F
Sediment		0.2 %	1.5 %	1.0 %	0.0 %	2.0 %
Sulphur		0.24%	0.65%	0.75%	0.05%	2.5 %
***** · · · · · · · · · · · · · · · · ·	1 1	1	1	1		1

It is to be noted that purchasers obtain more heat from a heavy fuel oil as it is purchased on the basis of the gallon.

The chief impurities found in fuel oil are water or brine and asphaltic sediment. The asphaltic sediment has almost as great heating value as the oil itself but the brine or water very greatly diminish the heatin value as well as interfere with the mechanical use of the oil.

The price of coal is the most important factor governing the price of fuel oil. In a general way it is claimed that one unit of heat from oil will produce the same amount of steam as 1.4 units of heat from coal. This takes into consideration the higher efficiency in using the oil, the greater ease in handling, the absence of certain mechanical features attendant upon the use of coal but does not consider the flexibility of the oil where this is a necessary feature of the power plant. According to this one pound of oil would be equivalent to 21/2 pounds of coal or one barrel of oil would be equivalent to .45 ton of coal. Oil at \$2.00 per barrel would be equivalent to slack coal at \$4.45 per ton. This assumes that the slack has a heating value of about 10,000 B. T. U. per pound.

Specifications for Fuel Oil of U.S. Navy.

The specifications for navy fuel oil, gas oil and bunker oil, Atlantic and Gulf ports, are:

1. Methods of Test.

Flash point will be taken as indicated in the specifications. (a)

(b) Viscosity will be taken by the Engler viscosimeter (see note under "2. Specifications")

(c) Water and sediment will be taken by the distillation method. When oil in small lots is consigned to naval vessels or to navy yards the centrifugal test will be used in order to obviate delay. In this test 30 cc. of oil and an equal quantity of best commercial benzol, 50% white will be used and the mixture heated to 100°F.

2. Specifications.

Fuel oil shall be a hydrocarbon oil free from grit, acid and (a) fibrous or other foreign matter likely to clog or injure the burners or valves. If required by the Navy Department it shall be strained by being drawn through filters of wire gauge having 16 meshes to the inch. The clearance through the strainer shall be at least twice the area of the suction pipe and strainers shall be in duplicate.

(b) The unit of quantity to be the barrel of 42 gallons of 231 cu. in. at a standard temperature of 60° F. For every decrease or increase of temperature of 10° F (or proportion thereof) from the standard 0.4 of 1% (or prorated percentage) shall be added or deducted from the measured or gauged quantity for correction.

The flash point shall not be lower than 150°F as a minimum (c) (Abel or Pennsky-Marten's closed cup) or 175°F Tagliabue open cup. In case of oils having a viscosity greater than 8 Engler at 150°F the flash point (closed cup) shall not be below the temperature at which the oil has a viscosity of E Engler.

(d) Viscosity shall not be greater than 40 Engler at 70°F.

Water and sediment not over 1%. If in excess of 1% the (e)excess to be subtracted from the volume or the oil may be rejected. Sulphur not over 1/5%. (f)

Note:-If the Engler viscosimeter is not available, the Saybolt standard universal viscosimeter may be used. Equivalent viscosities:

FUEL OIL FOR DIESEL ENGINES.

Explosion Engine oils should have the following properties: Specific gravity shall be below 0.920. Water shall be below 1%. 1.

2.

Flash point shall be between 60°C-100°C. 3.

- 4. Volatility shall be 80% or more at 350°C in Engler flask.
- Cold test shall be below 32°F. 5.

6. Coke shall be less than 3%.

7. Sulphur shall be below 0.75%.

8. Solubility in xylene shall be more than 99.5%.

Acids and alkalies shall be absent. 9.

Some of the advantages claimed for liquid fuel under boilers are; (Poole-Calorific Power of Fuels.)

1. Diminished loss of heat up the stack owing to the clean condition in which the tubes can be kept, and to the smaller amount of air which has to pass through the combustion-chamber for a given fuel consumption.

A more equal distribution of heat in the combustion-chamber 2. as the doors do not have to be opened and consequently a higher efficiency is obtained.

With oil there is no chance of getting dirty fires on a hard 3. run as with coal.

4. A reduction in cost of handling fuel, since oil is handled

mechanically or by gravitation while with solid fuel, manual labor is required.

No firing tools or grate bars are used, consequently the furn-5. ace lining and brickwork floors, etc., suffer less damage.

No dust nor ashes to cover or fill the tubes and diminish the 6. heating surface, nor to be handled or carted away.

7. Petroleum does not suffer while being stored, while the deterioration of coal under atmospheric influence is well known.

Ease with which fire can be regulated from a low to a most 8. intense heat in a short time.

9. Absence of sulphur or other impurities and longer life of plates, etc.

10. Lessening of manual labor of fireman.

Great increase of steaming capacity. 11.

For burning liquid fuel the best burner is that which atomizes or sprays the fuel. By thus forming a fine mist an approximation to the theoretical fuel, gas, is obtained. Several methods are in use for this purpose. By some the oils are vaporized by heat but this is applicable only to light oils which are not much used. The favorite method is by having the burner so constructed that the oil is forced out in a spray and at the same time mixed with the air necessary for its combustion.

To have the best results, the burner must be so regulated as to have a flame bordering on, but not quite, smoky. Thus sufficient and not too much air is obtained. The quantity of steam needed to atomize the oil is about 4% of the water evaporated.

MISCELLANEOUS FACTS CONCERNING HEATING BY OIL.

Good practice in the atomization of fuel oil requires an average of 0.3 pound of steam per pound of oil burned. One pound of fuel oil requires 14 to 15 pounds or 200 cubic feet

of air for complete combustion. 225 cubic feet is good practice.

The stack gases from an oil furnace for the highest efficiency should not contain less than 15% of carbon dioxide (over 13% is good).

The temperature of an oil flame with complete combustion and without an excess of air is about 3750°F. (Natural gas flame, 3250°F.)

One pound of oil will yield on combustion 16 to 17 pounds of gases of combustion or 400-500 cubic feet at a temperature of 400°F.

Oil is successfully used in melting iron and steel scrap. For this purpose it is much superior to coal on account of the absence of

mineral matter and the very much smaller amount of sulphur. One barrel of oil will melt one ton of steel in the reverberatory furnace, with the furnace walls already hot.

A typical malleable iron foundry by the changing of the furnaces from coal to oil fuel increased the strength of their castings 100% and increased the output 20%.

Diesel engines consume from .45 to .7 pound of heavy oil per brake H. P. per hour.

Oil requires 60% of stack area needed for coal firing.

Oil gives a fuel efficiency at least 10% greater than coal.

The advantages of oil fuel installations for locomotives and boats have been found to be as follows:

Economy of space reserved for carrying fuel; 50% more (a) fuel value per unit space.

(b) Ease in filling tanks.

(c) Rapidity of time in meeting a varying load on boiler. Fires may be instantly lighted.

(d) Ability to force boiler to extreme duty in case of emergency.(e) Short height of stack.

(g) Superior personnel available for the operation of the burners.

(h) Ability to secure and maintain higher speed with oil fuel than with coal. No deterioration in storage.

In the distillation of crude oil in which 50% of the crude is distilled off as benzine and kerosene, in good practice, 2.8 barrels of fuel oil are used per 100 barrels of crude oil treated.

For all refining purposes in the production of gasoline, naphtha and kerosene only, from 6 to 7 barrels of fuel oil are required for each 100 barrels of crude treated, assuming that 50% of the lighter hydrocarbons are distilled from the crude.

One-fourth of a gallon of fuel oil is required to produce one gallon of 58° Baume' gasoline by cracking according to a pressure distillation process now extensively used.

The specific heat of petroleum is about 0.5 (.49-.53), the heat of vaporization averages about 130 B. T. U. per pound and the heat of fusion 63 B. T. U. per pound (Paraffin). For Natural Dry Petroleum of Paraffin or Semi-Paraffin Base

For Natural Dry Petroleum of Paraffin or Semi-Paraffin Base the following relation of gravity (Baume-U. S.) and heating value holds:

B. T. U. per pound=18700+40 (Be'-10).

SAMPLING OF FUEL OIL.

The accuracy of tests depends upon the care with which an average representative sample of the fuel oil delivery has been taken and

the importance of obtaining such a sample cannot be overestimated. Top, middle and bottom samples should be taken with a standard "car thief" and these samples should be combined and thoroughly mixed to form one sample for car deliveries. Where oil is received in tanks or reservoirs the swing pipe should first be locked at a position well above the level of the water and sediment usually found in the bottom of such tanks. Tanks schould be sampled every foot for the first five feet above the bottom of the swing pipe, and at five-foot intervals from there to the surface of the oil. This sampling should be done with a standard tank thief, the samples "cut" individually, and deductions for impurities made on the separate volumes which these samples represent. If the tank is a large one, it should be sampled through at least two hatches. In receiving large deliveries of the more viscous oils it is necessary to take many samples in order to insure fair and average impurity (M. & B. S.) deductions. This is because water and sediment do not readily settle out of such oils.

Natural Gas Fuel and Producer Gas Costs

The following table of Producer Gas Costs includes fuel, power, repairs and maintenance, labor and supervision, interest and depreciation, in fact, every item of cost except the interest and taxes on the land occupied.

1000 C	er Gas Co u. Ft. fo osts Give	r Coal	Costs at Which Other Fuels Must be Bought to Obtain th Same Number of B. T. U. as When Buying Producer Gas With Coal at the Price Given							
	Hot Raw	Clean		al Gas) Cu. Ft.	Fue per G	l Oil allon	Carbu Water	Gas or iretted Gas per Ju. Ft.	Blue G 1000 C	as per u. Ft.
Cost of	Pro- ducer	Cold Pro-	Hot	Clean	Hot	Clean	Hot	Clean	Hot	Clean
One Ton		ducer	Raw	Cold	Raw	Cold	Raw	Cold	Raw	Cold
of Coal	Offtake	Gas	Gas	Gas	Gas	Gas	Gas	Gas	Gas	Gas
\$2.00	3.13c	4.15c	23.7c	31.5c	2.91c	3.86c	12.6c	16.72c	6.45c	8.590
2.50	3.55	4.57	26.9	34.67	3.3	4.25	14.3	18.40	7.34	9.45
3.00	3.96	4.98	30.1	37.84	3.69	4.64	16.6	20.09	8.20	10.32
3.50	4.38	5.40	33.3	41.01	4.08	5.03	17.65	21.77	9.07	11.18
4.00	4.79	5.82	36.3	44.18	4.46	5.42	19.3	23.45	9.92	12.05
4.50	5.21	6.24	39.5	47.35	4.85	5.81	21.	25.13	10.78	12.91
5.00	5.63	6.66	42.7	50.52	5.24	6.20	22.7	26.82	11.65	13.78
5.50	6.05	7.08	45.9	53.69	5.63	6.59	24.35	28.50	12.5	14.64
6.00	6.46	7.49	49.1	56.85	6.01	6.97	26.0	30.18	13.36	15.50

HEATING VALUES USED

Producer Gas Natural Gas

Fuel Oil

Coal Gas or Carburetted Water Gas Blue Gas

145 B. T. U. per cu. ft. 1,100 B. T. U. per cu. ft. 135,000 B. T. U. per gallon 585 B. T. U. per cu. ft. 300 B. T. U. per cu. ft.

These costs are based on the plant operating with a 100%Note: load factor, that is, operating at rated capacity 24 hours per day, 365 days per year. Comparatively few plants have a 100% load factor, therefore, it is necessary to take this very important point into consideration when estimating the cost of gas. The cost of Producer Gas, with a reasonable degree of accuracy

may be estimated for any load factor by applying the formula:

(Rx 400) C = T +-2.38A x B

Where C = Cost of Producer Gas per 1000 cu. ft. under conditions specified.

A = Number of feet of gas used per day.

B = Days per week plant is in operation.

T = Cost figures shown in table at 100% load factor.

R = Rated hourly capacity of plant in cubic feet.

It also must be kept in mind that furnace efficiencies have a very great bearing on the cost of the finished product. Without regeneration or recuperation Producer Gas cannot be used as efficiently as the more concentrated fuels.

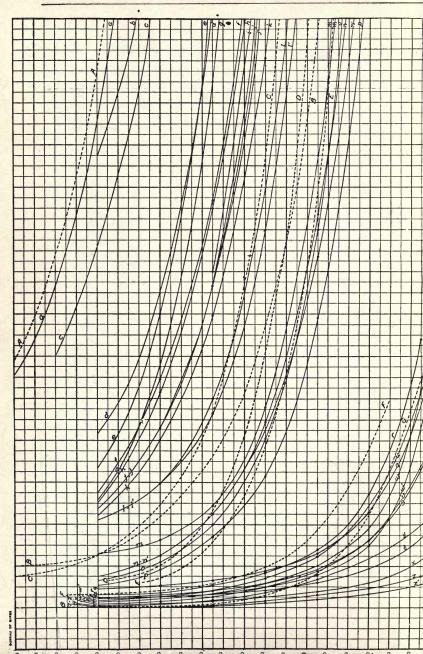
The expense of the distribution system and the furnaces also have an important bearing on the total cost of doing the work.

KEY TO PAGE 188.

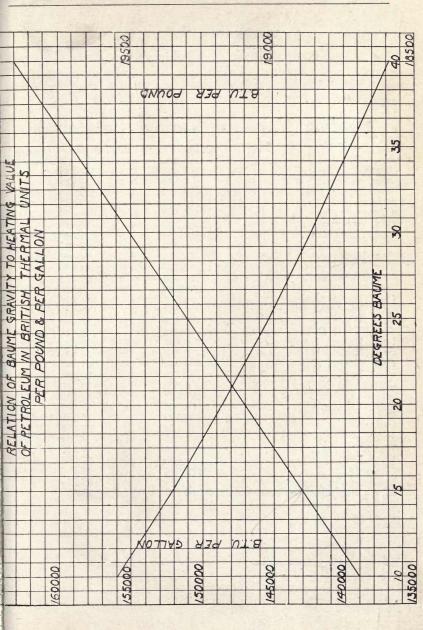
Curve No.	TYPE OF OIL	Gravi	ity	Flash point,
NO.		Specific	°Be⁄	°F
-	Solid curves			
8	Mexican residue	1.000	10.0	374
b	"Toltec fuel oil," Inter-Ocean Oil Co., N. Y	.988	11.7	220
c	"Toltec or Panuco oil," Inter-Ocean Oil Co	.986	12.0	124
d	"No. 102," Union Oil Co., Bakerstield, Cal	.980	12.9	280
0	"No. 18," Union Oil Co., Bakersfield, Cal	.980	12.9	285
f	"Standard" Mexican crude (lot 2)	.964	13.4	202
g	"No. 25," Union Oil Co., Bakersfield, Cal	.978	13.2	262
h	Mexican crude, Texas Co	.952	17.3	126
i	Sample No. 3, Anglo-Mex. Pet. Products Co	.952	17.3	164
i	"Gaviota Refinery," Associated Oil Co., Cal	.953	17.1	230
j j'	Mexican oil, Atlantic torpedo flotilla, March,			
	1914	.947	18.1	182
k	Standard Mexican crude (lot 1)	.954	17 0	145
1	Mexican oil, U. S. S. Arethusa "Nos. 1, 2, 3," Anglo-Mexican Pet. Products Co.	.950	17.6	182
ľ	"Nos. 1, 2, 3," Anglo-Mexican Pet. Products Co.	.955	16.8	188
m	Producers Crude No. 1 fuel oil, Union Oil Co.,			
n	California. "Coalinga Field," Associated Oil Co., Mont-	.959	16.1	174
	erev. Cal.	.957	16.5	186
n'	"Avon Refinery," Asso'd Oil Co., Avon, Cal	.953	17.1	168
0	Richmond, California	.953	17.1	228
р	Sun Co., Louisiana	.936	19.8	275
q	"Standard," Illinois	.893	27.3	146
r	Gulf Refining Co., Navy standard oil, U. S. S.	.892	27.5	180
8	"Standard " Indiana	.880	29.6	144
ť	Perkins. "Standard," Indiana. "Standard Star," California	.912	23.9	180
u	"Standard," Illinois (lot 4)	.893	27.3	146
v	"Standard," Indiana (lot 4)	.880	29.6	144
w	Gulf Refining Co., Navy contract	.882	29.3	170
W'	"Standard," Lima, Ohio, crude	.876	30.4	149
x	Camden Chemical Co., by-product of coal tar.	.010	00.2	145
	"Star" California	.912	23.9	180
y Z	"Star," California	.912	20.9	100
2	Roe.	.885	28.7	182
Z'	Standard Mexican gas oil	.856	34.2	151
	Indicates test results.			
	Dotted curves			
A	Panuca crude, Inter-Ocean Oil Co	.975	13.7	- 146
B	Mexican petroleum, Texas Co	.938	19.5	234
õ	Associated Oil Co., California	.971	14.2	257
D	Bakersfield, Cal., pipe line to Port Costa	.970	14.4	260
E	California Standard Oil Co., steamer Santa	.010	-1.1	200
10	Barbara.	.962	15.7	282
F	Beaumont, Tex., Gulf Refining Co	.907	24.8	200
G	Navy standard oil, Texas Co	.911 to .900	24 to 26	195 to 220

From "Oil Fuel Handbook."

BULLETIN NUMBER FIFTEEN OF



KANSAS CITY TESTING LABORATORY



Heating Value of Various Substances

	Calories per	B. T. U. per lb. of Com-
	gram.	bustible Matter.
Alcohol, grain	7,054	12,697
Alcohol, wood	5,330	9,594
Asphalt, 60° pen		17,159
Benzol.		18,054
Carbon or Coke		14,647
Gas, Acetylene		20,749
Gas, Coal, Min		7,990
Max		12,266
Gas, Methane		24,019
Gas, Water	2.350	4,230
Gas, Hydrogen		62,032
Iron		2,848
Coal, Pa. Anthracite		14,880
Coal, West Va. Bituminous	8.778	15,800
Coal, Wyoming Lignite	7.444	13,400
Coal, North Dakota Lignite	6.411	11.540
Coal, Kansas Bituminous	8.461	15,230
Coal, Illinois Bituminous	-8.056	14,500
Coal, Cannel (Missouri)	8,980	16,165
Coal, Peat	5.940	10,692
Cottonseed Oil.	9,500	17,100
Gasoline, avg		20,750
Fuel Oil, avg		19,500
Shale Oil.	10,970	19,750
Paraffin wax		20,050
Sulphur.		4,034
Wood		8,550
Naphthalene.		17,442
Gilsonite.		17,900
Hard Asphalt from petroleum.	9,989	17,980
Blown Asphalt from petroleum.	10,210	18,380
stown rispitate from postoreum.		10,000

Refining of Oil for Road Building and Paving Purposes

The various methods of refining which yield residues adaptable or used for road building and paving purposes are as follows:

Sedimentation.

Dehydration.

Fractional distillation by direct fire.

Forced fire distillation with direct fire.

Steam distillation.

Inert gas distillation.

Air blowing.

In the types of oil which are ordinarily used for making asphalt or road binders, water is one of the most common impurities. The water is ordinarily salt water and may contain more or less other mineral matter than the salt. These impurities are insoluble in the bitumen proper, and, as they differ from the bitumen in specific gravity, they may be removed wholly or in part by the process of sedimentation or separation by gravity. In the more fluid petroleums sedimentation occurs during storage in the large tanks and the water is ordinarily automatically drawn off from the bottom of the tank by reason of the different pressure produced by the salt water and by the oil. However, a small amount of emulsified water nearly always remains in all petroleums, so that there will always be a small amount of sediment. If the petroleum is very heavy and viscous, approximately equal in gravity to water, then the water will remain emulsified and will not separate by gravity. This type of oil happens to be the most suitable in quality for producing asphalt, and special means of removing this water is necessary before the oil can be reduced to the desired consistency. The dehydration processes are designed primarily for removal of the water in the bituminous material which will not completely separate by sedimentation. It is desirable to do this before distillation because of the fact that the presence of the water will cause foaming when the mixture is heated to the temperature of boiling water. Dehydrating plants vary considerably in design, but those more commonly used for petroleum in California are spoken of as topping plants. In this sort of plant the oil is pumped with or without pressure through a length of pipe containing many bends and turns, so that the oil is considerably stirred. The pipe coils are set in furnaces, so that they may be suitably heated to a temperature above that of boiling water. This pipe discharges the foam into a large expansion chamber, where the water and more volatile constituents separate in the form of vapor, which is condensed in an ordinary condenser for the recovery of the light products. This sort of plant is commonly spoken of as a pipe still. From the pipe still the oil passes through another line, direct to a large batch still, where it is subjected to the ordinary fractional distillation.

The essential principle in the distillation of an oil for road purposes is that it shall distill at a temperature sufficiently low to prevent the decomposition of the hydrocarbons: Since asphalt hydrocarbons begin to decompose at a temperature of 600°F or slightly below, it is desirable that the fire distillation be carried only to that temperature. After this temperature has been reached, the usual method is to blow superheated steam, which mechanically carries over the more volatile hydrocarbons at a temperature much below t⁺ actual boiling point.

This distillation has a special action in removing the paraffi. compounds which are particularly undesirable in that they have very little ductility and cementition value. The distillate will contain any light oils such as are used as spindle oils and for general lubrication, as well as any paraffin wax. It is particularly desirable in this distillation to prevent the formation of free carbon or coke. The distillat ion with steam may be carried down until the residue shows a penetration of about 10 millimeters.

A method of distillation which gives very great yields of solid or semisolid asphalt even from semiparaffin base oils is that of blowing the oil at moderately high temperature with air. This in many Mid-Continent oils gives much more asphalt than naturally exists in the oil. The action of the air is to produce a more viscous product which is very much less susceptible to temperature changes than the natural asphalt. It is strictly a chemical transformation process formed from the hydrocarbons in the oil which are ordinarily not useful for asphalt making purposes. It has been found from practical experience that this type of asphalt is not sufficiently cementitious and ductile to be used for ordinary paving purposes in producing firstclass asphalt pavement. It can, however, be successfully used and is in great demand for waterproofing purposes, for filler in brick and wood block pavement and for roofing purposes and for fluxing ductile asphalt.

The best types of petroleum for asphalt paving purposes are those from California, Mexico, Trinidad and Texas.

Asphalt production in 1917 from domestic petroleum was 701,8 short tons valued at \$7,734,690. This includes 327,142 tons of sersolid and 374,677 tons of semi-fluid asphalt. The total manufactur asphalt from Mexican petroleum was 645,613 tons. The import: native asphalt and asphalt rock in 1917 was 187,886 tons.

ASPHALT PAVEMENT

Asphalt is a black non-oxidized bituminous hydrocarbon, sen fluid to hard in consistency, the heavy residuum from petroleum or occurring naturally. The residua from petroleum are known as c' asohalts and come most largely from California, Mexican, Texas anu Mid-Continent petroleums. The most commonly used natural asphalts are Trinidad, Bermudez, Cuban and Gilsonite.

The term asphalt is commonly applied to bituminous pavements, being mixtures usually of oil asphalt with dust, sand, gravel or rock in varying proportions from 6% to 20%. The terms "bitumen" or "asphaltic cement" are commonly applied to the pure asphalt material.

The types of asphalt construction now commonly used are:

1. Asphaltic concrete. This mixture is very common in localities where Joplin chats are available. It is known also as "Topeka Specification Pavement" and "Bituminous Concrete," but it might be called bituminous gravel. The stone it carries is of $\frac{1}{2}$ " and $\frac{1}{4}$ " size.

2. Sheet asphalt is the original type of asphalt pavement laid in two courses, the bottom one with coarse stone, the top with sand mixed with the bitumen.

3. Bituminous concrete (Warren) is laid with coarse stone in the wearing surface.

4. Bituminous earth is laid without an appreciable amount of sand or rock.

There are two different basic principles involved in proportioning mineral matter of an asphalt pavement. One is to so grade the rse mineral particles that they support each other and interlock. The other is to produce a mastic of bitumen and finely divided earthy material that is rigid and self-supporting because of surface tension action. This mastic fills the voids in the coarse material and has a much higher melting point than the pure bitumen and does not so readily allow softening or movement of the pavement.

COMPOSITION OF NATURAL ASPHALT

	Natural	Ber-		Gra-	
	Trinidad	mudez	Gilsonite	hamite	Cuban
Bitumen	56.0%	94.0%	99.4%	94.1%	75.1%
Mineral Matter.	36.8%	2.0%	0.5%	5.7%	21.4%
Specific Gravity.	1.400	1.085	1.045	1.171	1.305
Fixed Carbon.	11.0%	13.5%	13.0%	53.3%	25.0%
Melting Point, °F	.190	180	300	Cokes	240
Penetration.	0.5	2.5	0	0	0
Free Carbon	6.0%	4.0%	0.1%	0.2%	3.5%
Sulphur (ash free basis)	6.5%	5.6%	1.3%	2.0%	8.3%
Petroleum ether soluble	65.0%	70.0%	30.0%	0.4%	41.1%
Total Carbon (ash free)	82.6%	82.5%		87.2%	
Hydrogen (ash free)	10.5%	10.3%		7.5%	
Nitrogen (ash free)	0.5%	0.7%		0.2%	

COMPOSITION OF OIL ASPHALTS

9,				Stanolind	
			((cracked-pres	-
202	M	Iid-Contine	nt	sure tar	
-In.	Mexican	Air Blown	California	residue)	
men	99.5%	99.2%	99.5%	99.8%	
eral Matter	0.3%	0.7%	0.3%	0.3%	
cific Gravity	1.040	0.990	1.045	1.060	
ed carbon		12.0%	15.0%	17.5	
lting Point °F	.140	180	140	135	
enetration		40	60	50	
Free Carbon	0.0	0.0	0.0	0.0	
alphur (ash free basis).	4.50%	0.60%	1.65%	0.35	
Petroleum Ether Soluble.		72.0%	67.0%	70.0%	
Cementing Properties	good	poor	good	good	
Ductility	45 cm	2 cm	70 cm	100+	
Loss at 32°F. 5 hrs	0.2%	0.1%	0.2%	0.1%	
Heat test		t smooth	adherent	scaly	

Stanolind

Composition of Rock Asphalt

ASPHALTIC LIMESTONES

· · · · · · · · · · · · · · · · · · ·	Ragusa Sicily	Seyssel France	Mons France		Buckhorn Oklahoma
Bitumen	9.9%	5.9%	8.9%	6.9%	5.9%
Passing 200 mesh	37.1	44.1	53.1	20.0	9.0
80 "	23.0	15.0	13.0	21.0	8.4
50 "	14.0	9.0	7.0	17.0	9.0
40 "	4.0	7.0	5.0	6.0	9.9
30 "	2.0	7.0	3.0	6.5	15.0
20 "	5.0	6.0	5.0	5.1	8.8
10 "	5.0	6.0	5.0	7.5	8.0
4 "	0.0	0.0	0.0	10.0	26.0
Calcium carbonate	89.0	91.3	90.0	92.9	96.0

ASPHALTIC SANDSTONES

E	Breckenridge	Buckhorn District	Higginsville,
	County, Ky.	Oklahoma	Missouri
Bitumen	9.2%	9.2%	7.9%
Passing 200 mesh	5.2	1.5	25.7
80 "	45.5	56.5	71.3
40 "		30.4	3.0
10 "	3.8	2.4	0.0
Calcium carbonate	0.0	0.0	0.0

Composition of Asphalt Pavements

The following table gives a comparison of a typical composition and properties of good mixtures representing the various types of asphalt wearing surface pavements:

Bitumi-	Bitumi-	Sheet	Bitumi-
nous	nous	As-	nous
Concrete	Concrete	phalt	Earth
(Topeka	(War-		"Na-
Spec.)	ren)		tional"
Asphaltic cement 8.0%	6.0%	10.0%	20.0%
Dust passing 200 mesh screen. 12.0	5.5	12.0	62.0
Dust passing 80 mesh screen. 12.0	2.8	16.0	15.0
Dust passing 40 mesh screen. 20.0	6.7	38.0	3.0
Dust passing 10 mesh screen. 20.0	24.5	24.0	0.0
Dust passing 4 mesh screen. 18.0	15.3	0.0	0.0
Dust passing 2 mesh screen. 10.0	13.3	0.0	0.0
Dust passing 1 mesh screen 0.0	25.0	0.0	0.0
100.0	100.0	100.0	100.0
Weight per sq. yd. 2 in. surface.215 lbs.	225 lbs.	205 lbs.	185 lbs.

SHEET ASPHALT PAVEMENT

Sheet asphalt is the standard asphalt pavement. Specifications call for two courses of the following composition and properties:

BINDER	OR	BOTTOM	COURSE
			Timita

	Limits	Standard
Bitumen.		6.0%
Mineral passing 200 mesh	7 -12	8.0
Mineral passing 80 mesh	10 -20	12.0
Mineral passing 40 mesh		15.0
Mineral passing 10 mesh		13.0
Mineral passing 4 mesh		17.0
Mineral passing 2 mesh		16.0
Mineral passing 1 mesh		13.0
		100.0
Thickness		.1½ in.
Density		ver 2.30
TOD CONDON		-
TOP COURSE		
	Limits	Standard
Bitumen.		10.0%
Mineral passing 200 mesh	12	13.0
Mineral passing 80 mesh	20 -34	23.0
Mineral passing 40 mesh	20 -40	27.5
Mineral passing 10 mesh	12 -35	-26.5
Mineral passing 4 mesh	0	0.0
Mineral passing 2 mesh	. 0 .	0.0
Mineral passing 1 mesh	0	0.0

MATERIALS REQUIRED FOR 1000 YARDS OF ASPHALTIC CON-CRETE PAVEMENT ARE AS FOLLOWS (Typical):

For wearing surface	For concrete base
"Chats" or Gravel = 32 tons	(6 inches of 1:3:6 mix)
Sand (Coarse) $= 32$ tons	Cement $= 732$ sacks $= 183$ barrels
Sand (Fine) $= 32$ tons	Sand $=$ 77 cubic yards
Dust $= 7 \text{ tons}$	Rock $= 155$ cubic yards
Asphaltic cement $= 8\frac{1}{2}$ ton	s Water $=$ 7,000 gallons

RELATION OF THE DEFECTS OF AN ASPHALT PAVEMENT TO ITS PHYSICAL PROPERTIES

- **Cracking** is caused by asphaltic cement without sufficient ductility, with too low penetration, insufficient in quantity or that has been over-heated; Imperfections in the base, such as a cracking in the base or the lack of a rigid base or lateral support; Insufficient compression when laid; Lack of traffic.
- Disintegration and Hole Formation are caused by asphaltic cement with poor ductility and cementing value, or insufficient to coat mineral aggregate and fill voids; Dirty sand; Non-uniform thickness of surface mixture; Weak foundations in spots; Water from beneath.
- Scaling of the Surface Mixture is caused by asphaltic cement lacking in cementing power, insufficient in quantity or subject to decomposition by the weather; Improper grading of mineral, particularly insufficient dust; Dirt conglomerates in sand; Insufficient density.
- Waviness and Displacement are caused by asphaltic cement without cementing power, too soft or in too large quantity; Irregularity of surface thickness, or of composition of asphaltic surface mixture; Insufficient dust or filler; Non-rigid base or expansion of the base; Street with heavy grade.
- Marking is caused by asphaltic cement that is too soft or in too large quantity; and that is too uniform; Insufficient dust or filler, Insufficient density.

FUNCTIONS OF VARIOUS CONSTITUENTS OF ASPHALTIC SUR-FACE MIXTURE.

- Gravel and Coarse Sand in proper relation diminish voids, insure greater stability and increase density, allow the use of less asphaltic cement, decrease tendency to displacement, waviness and marking, increase susceptibility to damage by erosion and abrasion.
- Sand in proper relation increases stability by filling voids in stone, increases capacity to resist abrasion, diminishes tendency to raveling.
- Filler or Very Fine Dust in proper relation increases density and stability by filling voids in sand, increases capacity to resist abrasion, allows wider range in penetration of A.C., diminishes or overcomes tendency to marking, displacement and waviness, increases cementition of mixture, increases capacity for A.C., increases the need for much compression and softer A.C. in laying mixture, eliminates lakes of A.C., decreases brittleness of pavement.
- A.C. in proper quantity and relation cements mineral particles together, keeps out water, imparts pliability, resiliency and noiselessness, prevents erosion and disintegration of coarse mineral of pavement.

Specifications for Asphaltic Cement for Asphalt Surface Mixture

Impurities.

The asphaltic cement shall contain no water, decomposition products, granular particles or other impurities, and it shall be homogeneous.

Ash passing the 200-mesh screen shall not be considered an impurity, but if greater than 1% corrections in gross weights shall be made to allow for the proper percentage of bitumen. Specific Gravity.

The specific gravity of the asphaltic cement shall not be less than 1.000 at $77^{\circ}F$.

Fixed Carbon.

The fixed carbon shall not be greater than 18%.

Solubility in Carbon Bisulphide.

The asphaltic cement shall be soluble to the extent of at least 99% in chemically pure carbon bisulphide at air temperature and based upon ash free material.

Solubility in Carbon Tetrachloride.

The asphaltic cement shall be soluble to the extent of at least 98.5% in chemically pure carbon tetrachloride at air temperature and based upon the ash free material.

Melting Point.

The melting point shall be greater than 128°F and less than 160°F (General Electric method).

Flash Point.

The flash point shall be not less than 400°F by a closed test. Penetration.

The asphaltic cement shall be of such consistency that at a temperature of $77^{\circ}F$ a No. 2 needle weighted with 100 grams in five seconds shall not penetrate more than 9.0 nor less than 5.0 millimeters. For asphaltic cement containing ash 0.2 millimeter may be added for each 1.0% of ash to give the true penetration.

Loss by Volatilization.

The loss by volatilization shall not exceed 2%, and the penetration after such loss shall be more than 50% of the original penetration. The ductility after heating as above shall have been reduced not more than 20%, the value of the ductility in each case being the number of centimeters of elongation at the temperature at which the asphaltic cement has a penetration of 5.0 millimeters. The volatilization test shall be carried out essentially as follows: Fifty grams of the asphaltic cement in a cylindrical vessel 55 mil-

Fifty grams of the asphaltic cement in a cylindrical vessel 55 millimeters in diameter and 35 millimeters high shall be placed in an electrically heated oven at a temperature of 325°F and so maintained for a period of 5 hours. The oven shall have one vent in the top 1 centimeter in diameter, and the bulb of the thermometer shall be placed adjacent the vessel containing the asphaltic cement. Ductility.

When pulled vertically or horizontally by a motor at a uniform rate of 5 centimeters per minute in a bath of water, a cylinder of asphaltic cement 1 centimeter in diameter at a temperature at which its penetration is 5 millimeters shall be elongated to the extent of not less than 10 centimeters before breaking.

EPITOME OF THE PURPOSES OF CERTAIN SPECIFICATIONS FOR ASPHALTIC CEMENT.

Impurities are a measure of the care with which the asphaltic cement has been refined and handled. Usually the presence of impurities in large quantities indicates a poor grade of asphalt. Water as an impurity would act as a diluent and would cause foaming in the kettle. Ash or mineral matter is not considered an impurity if it is a natural constituent of the asphaltic cement, but the mix and cementing value must be figured on the bitumen alone.

Specific Gravity of the asphaltic cement should be over 1.000. The advantage of a specific gravity more than 1.000 is that there will be less tendency for water to float out the asphaltic cement. The specific gravity is raised by the presence of mineral matter. Asphaltic oils of a penetration satisfactory for paving purposes always have a specific gravity greater than 1.000. Paraffin base oil and air-blown products usually have a specific gravity less than 1.000.

Fixed Carbon is a measure of the chemical constitution of an asphalt to some extent. Certain types of asphalt such as Mexican have naturally a constitution that yields a large amount of fixed carbon. Fixed carbon is largely used for determining the source and uniformity of an asphalt. Fixed carbon is not free carbon, but includes free carbon, which is practically absent in asphaltic cements.

Solubility in Carbon Bisulphide is a measure of the purity of an asphaltic cement. The cementing value, other things being equal, is proportional to the carbon bisulphide solubility. Any carbonaceous material such as coal tar or pitch is detected by the carbon bisulphide solubility test.

Solubility in Carbon Tetrachloride is very nearly the same as the solubility in carbon bisulphide. It is claimed that an asphalt having more than $1\frac{1}{2}$ % difference in the solubility in carbon bisulphide and carbon tetrachloride has been subjected to excessive heat in refining.

Melting Point is the temperature at which the asphaltic cement will flow readily. The melting point desired is dependent upon the mixture. If the amount of fine dust in the mineral aggregate is low, the asphalt should have a melting point higher than the highest temperature to which the pavement is subjected.

Flash Point is a measure of the amount of volatile hydrocarbons that are present in the asphalt and its readiness to decompose by heat.

Penetration is a measure of the consistency of the asphaltic cement. It is merely a quick, convenient test for checking up numerous individual samples. The penetration is expressed in degrees and in accordance with the method of the American Society for Testing Materials, each degree representing $\frac{1}{100}$ of a millimeter or $\frac{1}{1200}$ of an inch. The penetration, then, is the number of degrees that a No. 2 sewing needle when weighted with 100 grams will pass vertically into the A. C. at a temperature of 77°F (25°C) in 5 seconds. The penetration to be desired will depend upon the climate, the nature of the traffic, the grading of the mineral particles, the amount of voids, the amount of compression attainable, the ductility and cementing strength of the A. C. and the amount of dust filler.

Loss by Volatilization is a measure of the amount of light hydro-

carbons that are present in asphalt and is also a measure of the tendency of an asphalt to oxidize and to lose its ductility and penetration. Asphaltic cement which has no ductility after this volatilization test will not be satisfactory for paving purposes.

Ductility is the measure of the ability of an asphaltic cement to expand and contract without breaking or cracking. The same asphalt at a higher penetration should have a higher ductility, so all ductility tests should be based on a certain definite penetration regardless of the temperature, or should be based upon a temperature of 32°F. Ductility is also a measure of the cementing strength.

Viscosity is a measure of ability of the asphaltic cement to impart plasticity and malleability.

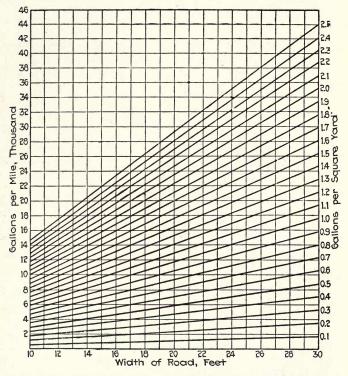
EFFECT OF MINERAL MATTER ON THE PENETRATION OF ASPHALTIC CEMENT (Typical Case).

% Dust	Penetration	Melting Point
0	200	100
35	128	110
55	92	120
70	34	150

In a general way, 1% of dust in asphaltic cement decreases the penetration 2 points with A. C. of ordinary penetration. This will vary somewhat according to the character of the asphaltic cement. A pavement having a relation of 2 parts dust and 1 part bitumen cannot soften or flow in hot weather.

FLUXING OF HARD ASPHALT.

As a general rule, 30% of $10-12^{\circ}$ Be' asphaltic flux is required to bring Trinidad asphalt to a penetration of 50. Less of paraffin flux is required. For each 1% of asphaltic flux added to about 50° asphalt the penetration is raised 3 points. For exact results a test should be made with the actual materials in question.



GALLONS OF ROAD OIL REQUIRED PER MILE OF ROAD AT GIVEN WIDTH AND RATE

Table for Calculating Voids in Sand and Hard Limestone

Weight in Pounds per Cubic Foot	% Voids	Weight in Pounds per Cubic Foot	% Voids	
		Cubic Poot	70 VOIUS	
60	63.9		10.0	
61	63.3	96	42.2	
62	62.6	97	41.6	
63	62.1	98	41.0	
64 65	$61.5 \\ 60.9$	99 100	40.4 39.8	
00	00.9	100	99.8	
66	60.3	101	39.2	
67	59.6	102	38.6	
68	59.1	103	38.0	
69	58.5	104	37.4	
70	57.9	105	36.7	
71	57.3	106	36.2	
72	56.7	107	35.6	
73	56.0	108	35.0	
74	55.4	109	34.4	
75	54.8	110	33.8	
76	54.2	111	33.2	
77	53.6	112	32.5	
78	53.0	. 113	32.0	
79	52.4	114	31.4	
80	51.8	115	30.7	
81	51.2	116	30.2	
82	50.6	117	29.6	
83	50.0	118	28.9	
84	49.4	119	28.3	
85	48.8	120	27.8	
86	48.2	121	27.2	
87	47.6	122	26.6	
88	47.0	123	26.0	
89	46.4	124	25.4	
90	45.8	125	24.7	
91	45.2	126	24.1	
92	44.6	127	23.5	
93	44.0	128	22.9	
94	43.4	129	22.3	
95	42.8	130	21.7	

Grams per 100 cc \times .6243 = pounds per cubic foot. % voids = 100 — (0.376 \times grams per 100 cc).

Typical Specifications for Wearing Surface of Asphaltic Concrete

The wearing surface shall be composed of a properly prepared mixture of bitumen, dust, sand and chats, gravel or trap rock.

The amount of asphaltic cement, dust, sand and chats shall be so regulated that the average mixture shall be within the following limits by weight:

		Size of			
		Opening	, Lower	Upper	Average
		In. Squar	e Limit	Limit	Typical
Bitumen			7.0%	10.0%	8.0%
Dust passing	200 mesh	0.0029	8.0	18.0	12.0
Sand passing	80 mesh	0.0068	10.0	20.0	12.0
Sand passing	40 mesh	0.0150	15.0	25.0	20.0
Sand passing	10 mesh	0.065	15.0	40.0	20.0
Sand passing	4 mesh	0.185	10.0	22.0	20.0
Sand passing	2 mesh	0.380	0.0	10.0	8.0

Ordinarily this mixture is to be obtained by the use of rock, coarse sand, fine bank sand and limestone dust or cement.

All of the mineral ingredients except the dust shall be heated and mixed in a suitable drier to a temperature of from 300 to 350°F. The bin containing the mineral shall be permanently equipped with a recording or an observation thermometer.

The asphaltic cement shall be added after it has been heated to a temperature not exceeding 360°F. The heating of the asphaltic cement must be by steam or if by direct fire vigorous mechanical stirring must be used. A recording thermometer should be used in the A.C. kettle and the aggregate.

The dust shall be added dry to each batch separately prior to the addition of the A.C. All materials shall be weighed.

The mixing shall be for a sufficient time to thoroughly and uniformly mix all materials and for a period of not less than one minute. The temperature of the mixture shall be between 270°F and 350°F

when it leaves the plant.

It shall be between 250° F and 350° F on the street (preferably 300° F).

The surface of the concrete shall be dry and clean at the time the surface mixture is applied.

The mixture shall be applied and raked to a uniform thickness, none being allowed to remain at the point of dumping and all lumps being thoroughly raked out.

The amount of hot mix applied shall be at least 210 pounds per square yard and shall be of a uniform thickness of 2 inches after rolling.

The compression shall be applied with a 5-ton roller until complete and sufficient in the judgment of the inspector and as indicated by the tests of the preceding day's laid surface. Hydraulic cement may be dusted over and rolled into the finished pavement.

dusted over and rolled into the finished pavement. The specific gravity of the compressed surface mixture shall average 2.20 or more and shall not at any time be less than 2.16. A piece of the compressed surface mixture after being placed in water for 24 hours shall not have absorbed water and shall not have become crumbly or weakened.

Kansas State Highway Commission Specifications for Road Oil

SPECIFICATION "A." Road Oil for the Surface Treatment of Earth Roads. (Cold Application.)

The road oil shall be homogeneous and free from water. It shall conform to the following requirements: Specific gravity, 25°C/25°C (77°F/77°F).. Not less than 0.910 Specific viscosity at 40°C (104°F)...... 100 to 25.0 Loss at 163°C (325°F), 5 hours...... Not over 25.0% Total bitumen...... Not less than 99.5% Per cent of total bitumen insoluble in 86° B. Naphtha..... Not less than 5.0% Fixed carbon......Not less than 4.0% SPECIFICATIONS "B." Road Oil for the Surface Treatment of Earth Roads. (Cold Application.) The road oil shall be homogeneous and free from water. It shall conform to the following requirements: Specific gravity, 25°C/25°C (77°F/77°F).. Not less than 0.890 Specific viscosity at 40°C (104°F)...... 10.0 to 25.0 Total bitumen......Not less than 99.5% Percentage of residue of 100 penetration...40 to 60 SPECIFICATION "M1" ASPHALT BINDER. The asphalt shall be homogeneous and free from water. It shall conform to the following requirements: Specific gravity, 25°C/25°C(77°F/77°F). Not less than 1.040 Flash point.....Not less than 163°C (325°F) Penetration at 25°C (77°F) 100 gm. 5 sec. When total bitumen is more than 90%.110 to 140 When total bitumen is 80% to 90%.... 90 to 120 When total bitumen is less than 80%... 80 to 100 Loss at 163°C (325°F), 5 hours..... Not over 4.0% Penetration of residue at 25°C (77°F) 100 gms. 5 sec. When total bitumen is more than 80%. Not less than 50 When total bitumen is less than 80%... Not less than 40 Total bitumen: Bermudez products..... Not less than 95.0% Cuban products...... Not less than 80.0% Trinidad products...... Not less than 65.0% Per cent of total bitumen insoluble in 86° B. Naphtha.....15.0 to 28.0 in diameter after being maintained at a temperature of 5°C (41°F) for 20 minutes shall bend 180° at any point without checking or breaking. The bending shall take place in one continuous operation requiring

not more than ten seconds.

SPECIFICATIONS "M2" ASPHALT BINDER.

The asphalt shall be homogeneous and free from water. It shall conform to the following requirements:

Brittleness Test.—A cylindrical prism of the asphalt 1 centimeter in diameter after being maintained at a temperature of 5° C (41°F) for twenty (20) minutes shall bend 180° at any point without checking or breaking. The bending shall take place in one continuous operation requiring not more than ten seconds.

SPECIFICATIONS "M3" ASPHALT BINDER.

The asphalt shall be homogeneous and free from water. It shall conform to the following requirements:

Specific gravity 25°C/25°C (77°F/77°F).0.970 to 1.020
Flash pointNot less than 200°C (392°F)
Ductility at 25°C (77°F)Not less than 15 cm
Penetration at 25°C (77°F) 100 gm. 5 sec.80 to 100
Loss at 163°C (325°F) 5 hrsNot over 2.0%
Penetration of residue at 25°C (77°F)
100 gms. 5 sec
Total bitumenNot less than 99.5%
Per cent of total bitumen insoluble in
86° B. Naphtha
Fixed Carbon

Brittleness Test.—A cylindrical prism of the asphalt 1 centimeter in diameter after being maintained at a temperature of 5° C (41°F) for twenty (20) minutes shall bend 180° at any point without checking or breaking. The bending shall take place in one continuous operation requiring not more than ten seconds.

SPECIFICATIONS "MT" REFINED TAR BINDER.

The refined tar shall be homogeneous and free from water. It shall conform to the following requirements: Specific gravity $25^{\circ}C$ ($77^{\circ}F$).....1.180 to 1.260 Float test $50^{\circ}C$ ($122^{\circ}F$).....110 sec. to 150 sec. Total distillate by weight: Not over To $170^{\circ}C$ ($338^{\circ}F$).....10% To $300^{\circ}C$ ($572^{\circ}F$).....15.0% Specific gravity to total distillate $25^{\circ}C$ ($77^{\circ}F$).....Not less than 1.030 Melting point of residue.....Not over $75^{\circ}C$ ($167^{\circ}F$) Solubility in carbon disulphide......77.0% to 88.0%Inorganic matter (ash).....Not over 0.5%

SPECIFICATIONS "ST1"

Refined Tar for Surface Treatment of Bituminous or Water-Bound Macadam Roads. (Hot Application.)

SPECIFICATIONS "ST2"

Refined Tar for Surface Treatment of Bituminous or Water-Bound Macadam Roads. (Cold Application.)

The refined tar shall be homogeneous and shall conform to the following requirements: Specific gravity 25°C/25°C (77°F/77°F).1.120 to 1.200 Specific viscosity at 40°C (104°F).....4.0 to 12.0 Total distillate by weight: Not over To 170°C (338°F).....5.0% To 300°C (572°F)......35.0% Specific gravity of total distillate 25°C (77°F)......Not less than 1.010 Melting point of residue......Not less than 1.010 Melting point of residue......Not over 65°C (149°F) Solubility in carbon disulphide.........88.0% to 96.0% Inorganic matter (ash).......Not over 0.5%

SPECIFICATIONS "S1."

Heavy Oil for Surface Treatment of Bituminous or Water-Bound Macadam Roads.

(Hot Application.)

The road oil shall be homogeneous, free from water and shall not foam when heated to 150°C (302°F). It shall conform to the following requirements:

Specific gravity 25°C/25°C (77°F/77°F).Not less than 0.980 Flash point.....Not less than 150°C (302°F) Specific viscosity to 100°C (212°F).....30.0 to 70.0

Per cent of total bitumen insoluble in

SPECIFICATIONS "S2."

Medium Oil for Surface Treatment of Bituminous or Water-Bound Macadam Roads.

(Hot Application.)

The road oil shall be homogeneous, free from water and shall not foam when heated to 100°C (212°F). It shall conform to the following requirements: Specific gravity 25°C/25°C (77°F/77°F).0.960 to 1.010 Total bitumen.Not less than 99.5% Per cent of total bitumen insoluble in **SPECIFICATIONS "S3"** Light Oil for Surface Treatment of Bituminous or Water-Bound Macadam or of Gravel Roads. (Cold Application.) The road oil shall be homogeneous and free from water. It shall conform to the following requirements: Specific gravity $25^{\circ}C/25^{\circ}C$ (77°F/77°F). 0.920 to 0.970 Specific viscosity at $25^{\circ}C$ (77°F)......30.0 to 70.0 Loss at 163°C (325°F) 5 hrs.......20.0% to 30.0% Per cent of total bitumen insoluble in

ASPHALT FILLER FOR BRICK. (Mexican Type for Vertical Fiber Brick.)

The joints between the paving blocks next the curb, railroad tracks and around manholes, or other street structures, shall be filled with asphalt filler complying with the following requirements:

The asphalt filler shall be composed of asphalt, or asphalts properly fluxed, if flux is necessary to bring it to the proper consistency. It shall contain at least 991/2 per cent bitumen soluble in carbon bisulphide. At least 99½ per cent of the contained bitumen soluble in carbon bisulphide shall be soluble in cold carbon tetrachloride.

The penetration shall conform to the following:

No. 2 Needle 5 Sec. 100 Grammes at 77°F 25 to 35.

No. 2 Needle 1 Min. 200 Grammes at 32°F not below 10. No. 2 Needle 5 Sec. 50 Grammes at 115°F not above 90.

The above filler shall be waterproof, shall adhere strongly to the brick, both in the joints and on the surface, and shall remain ductile and pliable at all climatic temperatures to which it may be subjected. It shall not run in the joints during the hottest summer weather, nor become hard or brittle when cold.

The melting point shall not be less than 165°F nor more than 200°F. The brick shall be dry and the asphalt filler shall be poured at a temperature of not less than 350° , nor more than 450° F and by means of squeegees, especially adapted to the purpose, the joints shall be thoroughly filled and shall provide sufficient hot material on the top surface as will fully cover and penetrate the etched or reticular surface with a thin coat of the hot asphalt.

The minimum amount of coarse sand, clean and free from dust, necessary to keep the fresh asphalt from sticking to traffic shall be immediately applied.

ASPHALT FILLER FOR BRICK.

(Texaco Type for Vertical Fiber.)

The joints between the paving blocks next the curb, railroad tracks and around manholes, or other street structures, shall be filled with

asphalt filler complying with the following requirements: The asphalt filler shall be composed of asphalt, or asphalts prop-erly fuxed, if flux is necessary to bring it to the proper consistency. It shall contain at least 98 per cent bitumen soluble in carbon bisulphide. At least 98½ per cent of the contained bitumen soluble in car-bon bisulphide shall be soluble in cold carbon tetrachloride.

The penetration shall conform to the following:

No. 2 Needle 5 Sec. 100 Grammes at 77°F 25 to 60. No. 2 Needle 1 Min. 200 Grammes at 32°F not below 10.

No. 2 Needle 5 Sec. 50 Grammes at 115°F not above 200. The above filler shall be waterproof, shall adhere strongly to the brick, both in the joints and on the surface, and shall remain ductile and pliable at all climatic temperatures to which it may be subjected. It shall not run in the joints during the hottest summer weather, nor become hard or brittle when cold.

The melting point shall not be less than 150°F nor more than 225°F. The brick shall be dry and the asphalt filler shall be poured at a temperature of not less than 375°, nor more than 450°F and by means of squeegees, especially adapted to the purpose, the joints shall be thoroughly filled and shall provide sufficient hot material on the top surface as will fully cover and penetrate the etched or reticular surface with a thin coat of the hot asphalt.

The minimum amount of coarse sand, clean and free from dust, necessary to keep the fresh asphalt from sticking to traffic shall be immediately applied.

ASPHALT FILLER FOR BRICK. (Sarco Type.)

The joints between the paving blocks next the curb, railroad tracks and around manholes, or other street structures, shall be filled with asphalt filler complying with the following requirements:

The asphalt filler shall be composed of asphalt containing at least 98 per cent of bitumen soluble in carbon bisulphide. At least 98½, per cent of the contained bitumen soluble in carbon bisulphide shal be soluble in cold carbon tetrachloride. The penetration shall be uniform in consistency and shall not vary more than seven and one-half $(7\frac{1}{2})$ points in penetration from the following standard:

The penetration shall conform to the following:

No. 2 Needle 1 Min. 200 Grammes at $32^{\circ}F$ 30. No. 2 Needle 5 Sec. 100 Grammes at $77^{\circ}F$ 40. No. 2 Needle 5 Sec. 50 Grammes at $115^{\circ}F$ 60.

The above filler shall be waterproof, shall adhere strongly to the brick, both in the joints and on the surface, and shall remain ductile and pliable at all climatic temperatures to which it may be subjected. It shall not run in the joints during the hottest summer weather, no become hard or brittle when cold.

The melting point shall be not less than 225°F, nor more t 275°F. The brick shall be dry and the asphalt filler shall be pour at a temperature of not less than 375°F, nor more than 450°F, and by means of squeegees, especially adapted to the purpose, the joints shall be thoroughly filled and shall provide sufficient hot material on the top surface as will fully cover the etched or reticular surface with a thin coat of the hot asphalt.

A top dressing of one-half (1/2) inch of coarse sand shall be spread immediately after the filler is applied and before the same has had its initial set, and shall immediately be rolled with a roller weighing not less than three nor more than five tons until the sand is thoroughly imbedded in the asphalt filler. As soon as the sand has been thoroughly ground into the top dressing of asphalt by traffic the surplus may then be swept off clean.

Chemical Nature of Cracking of Oil

When crude oil is subjected to ordinary distillation by fire the ht products naturally present in the oil are distilled off as such to a temperature of about 300°C (572°F) comprising both the gasoine and the kerosene. Above this temperature the hydrocarbons undergo partial decomposition while distilling, with the result that some light products are produced and distilled along with the heavy products. Olefins as well as paraffin compounds of lower molecular weight than the oil being heated are formed. By vigorous firing the entire oil residue may be distilled, leaving only a variable amount of residual carbon as a product of decomposition. The mount of carbon and gas formed by this pyrogenic decomposition is greater with the asphaltic or naphthene petroleums than with he paraffin base petroleums. A typical heavy Mid-Continent petroeum gives 4.5% of carbon and 4.0% of gas on distillation to coke or carbon. With pure paraffin base oils the amounts of carbon and gas formed are comparatively slight.

This property of all heavy petroleums in decomposing into hydrocarbons of lower molecular weight by heating is generally known as cracking. The chemical reactions involved in cracking are not definite. It was originally supposed that cracking involved the formation of a large amount of olefins according to the following reaction:

C_nH_{2n+2}	$= C_{n-m}H_{2(n-m)+2}$	$+C_mH_{2m}$
specific illus	tration of which would be	
C15H32	$=C_{8}H_{18}$	+C7H14
per Pentadeca		+Heptylene
This react	ion does not, however, accord	

and carbon are always formed in varying amount. A reaction which corresponds to the yields as experimentally found under certain conditions is the following:

$2C_nH_{2n+2}$	$=2C_{n-1}$	$H_{2(n-m)+2}$	$+mCH_4+mC$
or as a specific	illustration		
C15H32	$=C_8H_{18}$	+7CH.	+7C
Pentadecane	=Octane	+Methane	+Carbon
		onditions the amou	nt of gas formed is

very small, indicating that the following reaction was partly carried out. (2m+2) $C_nH_{2n+2} = (2n+2)$ $C_mH_{2n+2} + 2(n-m)C$ or as an illustration

9C15H22 Pentadecane	$= 16 C_8 H_{18}$ $= Octane$	+7C +Carbon
· childrectane		Our box

This last reaction is also indicated by the yields of gasoline obtained from some crude oils given in the table on page 120.

Pure paraffin wax of melting point of 130° F and specific gravity of 0.892 on repeated cracking confined under pressure up to 57 atmospheres at temperature of 400°C and with a vapor space twice the volume of the liquid, yielded 32.5% by volume of gasoline of $0.724=63.4^{\circ}$ Be' gravity or 29.1% by weight by each treatment or a total of 94.7% by weight, or 104% by volume.

s was as follows:
ginal paraffin
rinal paraffin
ginal paraffin
ginal paraffin
ginal paraffin
ginal paraffin

84.9%

The gasoline produced consisted of paraffin hydrocarbons as shown in curve on page 227.

That the cracking of oil is not simply a decomposition of the hydrocarbon molecules is shown by the curves on pages 211-2-3. These curves show the relation between the distilling temperature and the specific gravity of water white Cabin Creek distillate. Before cracking it had an end point of about 540°F and its heaviest ends had a specific gravity of 0.815. After cracking the end point was above 640°F and the end gravity above 0.900. Both heavier and higher boiling hydrocarbons as well as lighter and lower boiling hydrocarbons were produced simultaneously. There must have been polymerization to yield hydrocarbons of both higher boiling point and higher specific gravity. By continued cracking there may be made from water white distillate, solid and ductile asphaltic cement of typical conchoidal fracture. It may be that these polymerized products will make lubricating oils if they prove to be more resistant to heat decomposition and ordinary paraffin hydrocarbons.

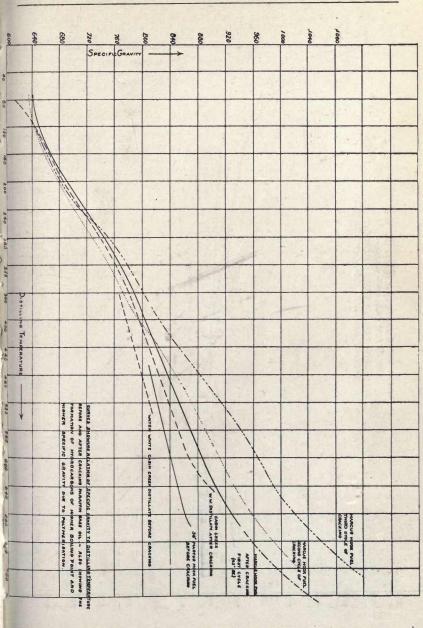
The gases produced by cracking likewise are not simple split-off hydrocarbons but vary according to the method of cracking. In liquid phase cracking the chief variation is in the olefin and hydrogen content. In a general way, there seems to be a tendency for low percentages of hydrogen to be associated with low percentages of olefins. A typical gas made in a Burton still gives the following analysis:

One of the problems in cracking is to limit the amount of hydrogen. This has been partially done by allowing the hydrogen to remain in contact with the cracked distillate under high pressure and at a temperature somewhat below the ordinary temperature of cracking. (See U. S. patent 1255138.)

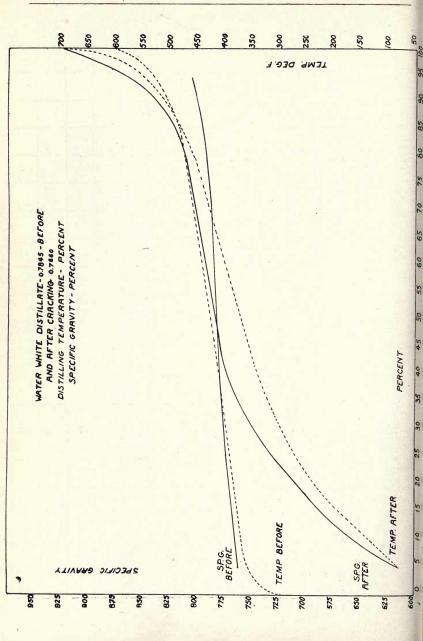
The chart on page 214 shows some of the relative properties of light hydrocarbons made by various processes used more or less in a commercial way for the production of gasoline from heavy oil. The smooth lines represent the distilling temperatures and the irregular lines represent the per cent of olefins in the fractions represented by the specific gravities indicated at the bottom of the chart. The per cent olefins or unsaturated hydrocarbons are shown at the right and the distilling temperatures are shown at the left.

The product marked B C is Burkburnett crude oil, B C O being the olefin curve. C is a gasoline made by a certain type of very high pressure equilibrium cracking in the liquid phase. W B is gasoline made by the Benton process. B and D are gasolines made by 85-170 lbs. pressure distillation. M and G are gasolines made by cracking in the liquid phase. C B is coal tar benzol as sold for blending for motor use.

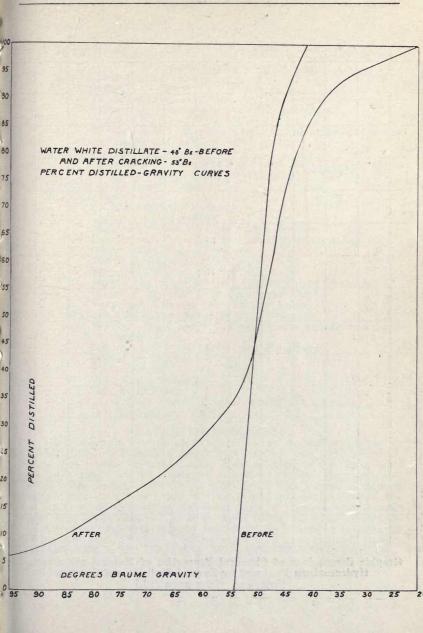
KANSAS CITY TESTING LABORATORY

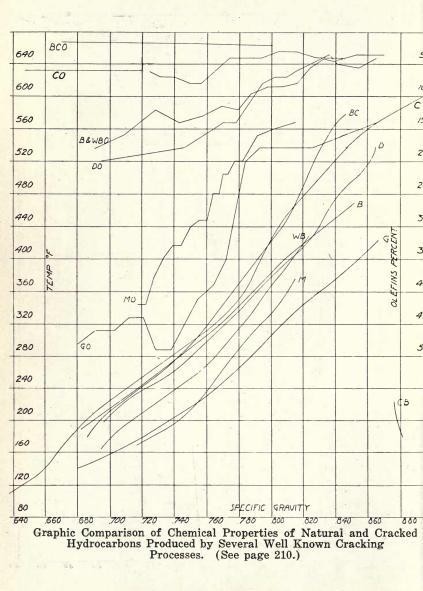


BULLETIN NUMBER FIFTEEN OF .



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Classification of Oil Cracking Processes

(Representative Patents)

I. Cracking in the vapor phase.

A Atmospheric Pressure.

Oil gas plants-very high temperature.

Pintsch Gas Plants-very high temperature.

Blaugas Plants-1000-12000°F.

Parker (W.M.) process-at 1000°F with or without steam. Greenstreet-Cherry red with steam.

With Increased Pressure. B

Rittman process-above 950°F and 200-300 lbs. pressure.

W. A. Hall process-1100°F and about 75 lbs. pressure.

II. Cracking in the Liquid Phase.

A With Distillation.

At Atmospheric Pressure. 1. Luther Atwood (1860). McAfee Process with aluminum chloride. Russian and American Practice for illuminating oils.

Above Atmospheric Pressure. Dewar & Redwood (1890). 2. Bacon & Clark at 100-300 lbs. Burton (Standard Oil Co.) 650-850°F and 60-85 lbs. Dubbs, J. A., over 10 lbs. and over 300°F.

3. Very high pressure (over 27 atmospheres). Without Distillation and with High Pressure.

B 1.

Without vapor space for equilibrium (continuous processes).

Benton (1886) 700-1000°F and 500 pounds.

Goebel-Wellman.

Mark (English).

2. With Vapor Space.

(a) Intermittent.

Palmer (below 27 atmospheres for aromatics).

(b) Continuous.

CATALYTIC PROCESSES

Many claims are made as to the virtue of certain substances in promoting the conversion of heavy hydrocarbons into light hydrocar-The writer has made many high pressure-liquid phase tests bons. with such substances as aluminum chloride, hydrogen chloride, manganese oxide, nickel, copper, lime, mercury, sodium nitrate, aluminum powder, zinc dust, iron dust, iron oxide and platinized pumice and has found in no case either increased rates of reaction or increased yields over those obtained by heat alone under the same conditions.

Electrical processes are not considered by informed refiners on the basis of cost alone and none have yet been demonstrated as having any virtue, in fact, other than as a means of applying heat.

In some instances a sweeter and whiter product resulted by use of added chemicals than with heat alone.

No Model.)

G. L. BENTON.

PROCESS OF BEFINING CRUDE PETROLEUM OIL.

No. 342,564. Patented May 25, 1886. Fig. 1 ſ d. 4 Fig. 2 Th D Fig. 3 h A . 4, INVENTOR Gurgo L. Bauton By Jillatto Amglow asty B & Woodruff

Development of Commercial Practice in Cracking of Oil

It has been stated that the commercial cracking of oil was accidentally discovered in the winter of 1861 by a stillman at Newark, New Jersey. However, this is probably not the case, since a patent was granted to Luther Atwood, of New York, May 15, 1860, No. 28,246, in the U. S. Patent Office, which provides for the production of light hydrocarbon illuminating oils from heavy oils, paraffin, etc. The apparatus provides for the cooling of the heavy oil vapors and their return to the still for further cracking. This is all carried out at atmospheric pressure.

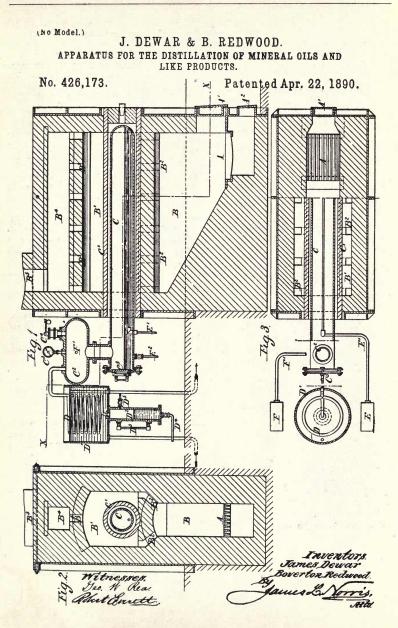
The first record of pressure distillation is apparently set forth by James Young in his patent, No. 3345 (English) of 1865, in which a distillation is described as being conducted in a vessel having a loaded valve or a partially closed stop cock through which the confined vapors escape under any desired pressure. Under these conditions, distillation takes place at higher temperature than the normal boiling points of the heavy hydrocarbons and partial cracking results. The patent was taken out for treatment of shale oil and in practice a pressure of 20 pounds to the square inch was recommended.

The first extremely high pressure process was that of Benton, U. S. patent No. 342,564, May 25, 1886. In this the oil is heated at a temperature of from 700 to 1000°F through a pipe not connected with a high pressure vapor chamber, but leading to a low pressure expansion chamber. The pressure used is as high as 500 pounds per square inch.

The most important patent in the present development of cracking processes is that issued to Dewar & Redwood which is described on the following two pages.

SPECIFICATIONS AND CLAIMS OF DEWAR & REDWOOD

"In distilling mineral oils—such as natural petroleum or similar oil made from shale, coal or other bituminous substances—in order to separate the lighter oils, suitable for lamps and other purposes, from the heavier oils, there is frequently a very large residue of heavy oil. Attempts have been made to obtain lighter oils from such residues or from heavy natural petroleums by causing the vapor generated in the still-boiler to pass a heavily-loaded valve, so that the vaporization takes place under considerable pressure. It has also been proposed to arrange the still-boiler with its upper part cooled,



so that the less volatile portions of the vapor may become more or less condensed and fall back into the hot liquid below, this mode of operating being commonly termed "cracking". Both these methods are objectionable, the former on account of the irregularity of the distillation and the latter on account of the waste of heat in conducting the cracking process and the slowness and insufficiency of the results."

"Our invention relates to a method of conducting the distillation by suitable apparatus in such a manner that we get the benefit of regular vaporization and condensation under high pressure, and that we may at the same time get such advantage as can be obtained from cracking. For this purpose we arrange a suitable boiler or retort, and a condenser in free communication with one another, without interposing any valve between them; but we provide a regulated outlet for condensed liquid from the condenser. We charge and keep charged the space in the boiler or retort and condenser that is not occupied by liquid with gas under considerable pressure, it may be with air or it may be with carbonic-acid gas or other gas that cannot act chemically on the matter treated. The distillation and condensation being thus conducted under considerable pressure, which can be regulated at will, we obtain from the heavy residue a quantity of more or less light oil suitable for illuminating and other purposes, which cannot be obtained by distillation under atmospheric pressure. We may also arrange the still-head or upper part of the boiler or retort so as to operate according to the cracking method above referred to, the cracking in this case taking place under high pressure instead of being carried on under atmospheric pressure.

"The apparatus for effecting distillation in the manner described may be arranged in various ways. The accompanying drawings show one form of apparatus for this purpose.

"By a pipe and cock or a suitably loaded safety-valve D^s gas may be withdrawn from the space above the liquid in the column D^s.

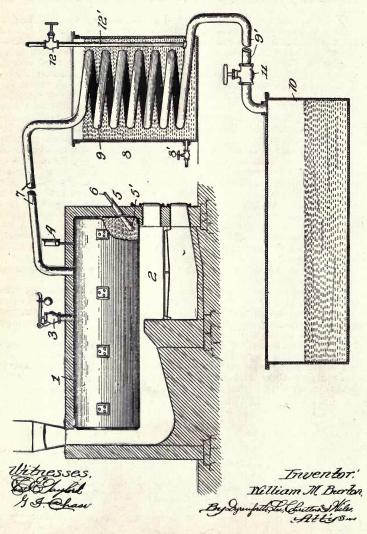
"By regulating the heat and pressure to which the retort is subjected the character of the distillate may be varied, and thus oils more or less light can be obtained to suit various uses. Also the proportions of the parts may be varied, and, if necessary, means of cooling may be applied to the still-head C^2 .

"Having thus described the nature of our invention and the manner of carrying the same into effect, we claim—the herein-described method of distilling mineral oils and like products, which consists in both vaporizing them and condensing the generated vapor under a regulated pressure of air or gas substantially as specified."

W. M. BURTON. MANUFACTURE OF GASOLENE. APPLICATION FILED JULY 3, 1912.

1,049,667.

Patented Jan. 7, 1913.



THE BURTON PROCESS

This is the process by which much of the artificial gasoline now on the market is made.

The sketch in the patent is shown on the opposite page.

In the practical operation of this process a very hot furnace is required on account of the very great radiation of heat from the return conduit 7.

Novelty in this process is claimed to lie in the maintenance of pressure on the condenser, though this is done in the Dewar & Redwood process already described (q.v.). The fact remains, however, that the Burton process is being successfully operated on a large scale and presumably with profit. In one of the Burton patents (1,105,961) it is claimed that $63\frac{1}{2}\%$ of the original charge of oil is converted into gasoline.

The actual operation of the Burton process has been described as follows.

The stills have a capacity of 200 barrels each and are heavy, horizontal steel cylinders, with walls one-half inch thick, thoroughly insulated with asbestos. From the top of the still is a long run-back, exposed to the air, which returns for cracking any undecomposed oil. The stills, the run-back and the condenser are all maintained under a pressure of about 85 pounds per square inch, the oil being heated to a temperature of about 750°F. Each still is charged every 48 hours, the yield being 57% of 51° naphtha. The carbon tends to be of a granular or mealy nature, rather than hard and adherent, and is cleaned out after each run.

Important modifications of the Burton process are shown in the Clark patents, 1,119,496, 1,129,034 and 1,132,163; A. S. Hopkins, 1,199,-464; R. E. Humphreys, 1,122,002, 1,122,003 and 1,119,700.

One of the Clark modifications allows the application of heat to tubes and seeks to overcome the danger of heating a large bulk of oil directly.

The Hopkins patent provides for introducing fresh oil supply into the run-back 7.

One of the Humphreys patents provides for plates in the bottom of the still to prevent the bad effect of carbon and to give a large metallic heating area.

The original Burton claims are as follows (Patent 1,049,667, filed July 3, 1912):

"1. The method of treating the liquid portions of the paraffin series of petroleum distillation having a boiling point upward of 500°F to obtain therefrom low-boiling point products of the same series, which consists in distilling at a temperature of from about 650 to about 850°F the volatile constituents of said liquid, conducting off and condensing said constituents and maintaining a pressure of from about 4 to about 5 atmospheres on said liquid of said vapors throughout their course to and while undergoing condensation.
2. The method of treating the liquid portions of the paraffin series

2. The method of treating the liquid portions of the paraffin series of petroleum distillation having a boiling point of upward of 500° F to obtain therefrom low-boiling point products of the same series, which consists in distilling off at a temperature of from about 650 to 850° F the volatile constituents of said liquid, conducting off and condensing said constituents, maintaining a pressure of from about 4 to about 5 atmospheres on said liquid of said vapors throughout their course to and while undergoing condensation, and releasing from time to time accumulations of gas from the product of condensation."

ADVANTAGES OF LIQUID PHASE CRACKING

All processes of making gasoline which have not involved the treatment of the oil strictly in the liquid phase are said to have met with only a questionable degree of success.

While the cracking of oil in the vapor phase would be highly desirable if the product and other conditions were satisfactory, it has been claimed by many that the advantages of applying the heat to the liquid phase are as follows:

- 1. A lower temperature is sufficient to induce cracking.
- 2. The rate of reaction is greatly increased, being greater the higher the pressure within certain limits.
- 3. A product containing smaller amount of olefins and aromatics is produced.
- 4. A higher yield of refined gasoline is obtained.
- 5. There is a better economy of heat.
- 6. There is a selective action on the oil or heavy portions of the petroleum by reason of the automatic conversion of the desired product into the vapor phase, thus freeing it from further liability to decomposition.
- 7. There is a high oil capacity with small plant dimensions.
- 8. There is a perfect control of temperature.
- 9. There is a rapid and more complete absorption of heat from the furnace and less tendency to local overheating on account of the much higher specific heat of oil than of the oil vapor.
- 10. There is the possibility of operating either by intermittent charging or by continuous treatment and distillation.
- 11. The carbon is deposited in a suspended condition in the oil and not on the retaining walls.
- 12. There is the possibility of the use of the automatically developed pressure for mechanical and condensing purposes. The chief disadvantage in cracking oil in the vapor phase and under high pressure seems to be the danger attendant upon a possible failure of steel parts. (See page 225.)

Refinery Engineering Data on Distilling and Cracking of Petroleum

The total capacity of a horizontal still is approximately 0.14 d²l, d being the diameter and l the length of the still in feet.

The heating area of a horizontal still is 1.0472 d l on the assumption that one-third of the shell is fired. In continuous stills a larger area may be fired on account of a higher minimum oil level.

Continuous stills give a greater crude oil capacity than batch stills on account of the time required for charging and discharging batch stills. The amount of benzine or crude gasoline distilled is 1.5 d l barrel per day with continuous operation and with no other products distilled.

The approximate amount of gasoline from crude oil stills per day per square foot of still bottom area not including charging time or time for bringing to distillation temperature is 1.0 barrel. This may vary according to the intensity of firing and the character of the crude.

The approximate total fuel consumption in producing one gallon of 58°Be' gasoline in a still by cracking at 85 pounds pressure is 50,000 B. T. U. or 0.4 gallon of fuel oil.

The approximate total fuel consumption by properly cracking in tubes at 750 pounds pressure in producing one gallon of 58° Be' gasoline is 20,000 B. T. U. or 0.15 gallon of fuel oil. The report of the Western Petroleum Refiner's Association of

The report of the Western Petroleum Refiner's Association of September, 1919, on a pressure distillation process operating at 135 pounds per square inch pressure may be analyzed as follows: 0.164 gallon of 58°Be' gasoline was produced per square foot

0.164 gallon of 58°Be' gasoline was produced per square foot of heating area per hour after the oil was brought to the cracking temperature.

0.8 gallon of fuel oil equivalent to 112,000 B. T. U. was required to produce 1 gallon of 58°Be' gasoline.

200 cubic feet of gas was produced for each barrel of 58°Be' gasoline.

7.0 pounds of still carbon was produced per barrel of 58°Be' gasoline.

A typical composition of the so-called carbon deposited in cracking stills is as follows. This sample was extracted with 70°Be' petroleum naphtha before testing:

Moisture (volatile at 105°C)	0.00%
Volatile (500°C)	13.08%
Fixed carbon.	80.42%
Ash	6.50%

												10	00.00%	
Sulphur.													1.83%	
Iron													2.76%	

The following data represents the operation covering a long period of time of a very extensively used process for cracking oil, based on one still.

100.000

Gallons of oil charged	8,000 gallons
Gallons of oil run in	1,800 gallons
Gallons of oil treated	9,800 gallons

Amount time feedings in ail	5 hours
Average time feeding in oil	
Total hours distilled	
Pounds of coal used to distill1	1,000 lbs. per run
Total distillate produced	5,295 gallons
Total 58.5° gasoline produced	3,018 gallons
% distillate	
% 58.5° gasoline in distillate	
% 58.5° gasoline of oil treated	30.8%
Amount of distillate per hour of distilling1	43.1 gallons
% distillate of total charge per hour of dis-	
tillation	1.46%
Amount of 58.5°Be' gasoline per hour of dis-	
tilling	81.6 gallons
% of 58.5° gasoline per hour of distilling	0.83%
Area of still bottom	70 sq. ft.
Gallons of 58.5° gasoline per hour per sq. ft.	
of heating area	0.302
Lbs. of coal per gallon of gasoline (58.5°)	3.625 lbs.
Equivalent gallons of fuel oil per gallon of	
58.5° gasoline.	0.25
0	

CALCULATION OF HEAT EXCHANGES IN REFINERY CON DENSERS

In calculating amount of water required for condenser, use the following formula:

200 g

$t_2 - t_1$

w = gallons of water required per hour.

w =

 $t_1 =$ incoming temperature of condenser water.

 $t_2 =$ outgoing temperature of condenser water.

g = gallons of gasoline to be condensed per hour.

Heat absorbed in condensing 1 gallon of gasoline to 60° F = 1550 B. T. U.

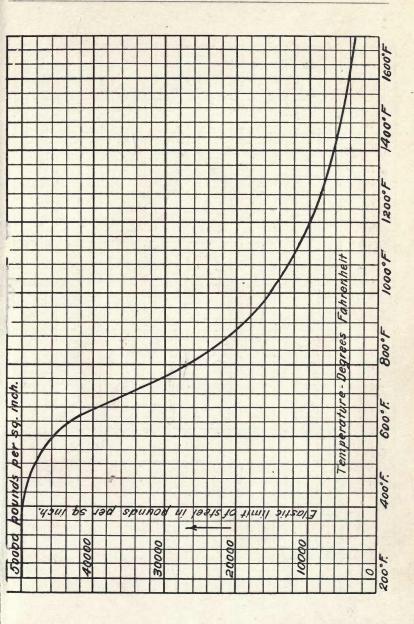
Heat absorbed in condensing 1 gallon of kerosene to $60^{\circ}F = 2400$ B. T. U.

Heat absorbed by oil in distilling off 50% from it as gasoline and kerosene is 2100 B. T. U. per gallon of crude oil.

Heat absorbed by oil in distilling to coke is approximately 3000 B. T. U. per gallon.

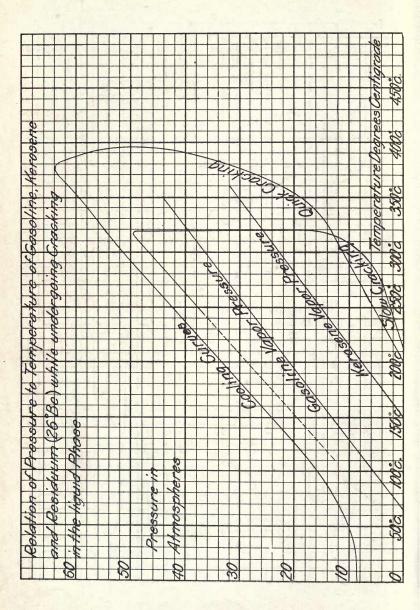
- Amount of condenser surface required to properly condense one gallon of gasoline per hour = 2 sq. ft.; 1 gallon of kerosene per hour = 1 sq. ft. This is lessened with cold water and with larger quantities of water and varies with the length and cross section of the condenser tubes.
- The cross section of the vapor line should be .05 sq. in. per gallon of gasoline per hour. The cross section of the condenser tubes may be reduced ½ after first ½ of length and ¼ more after second ½ of length.

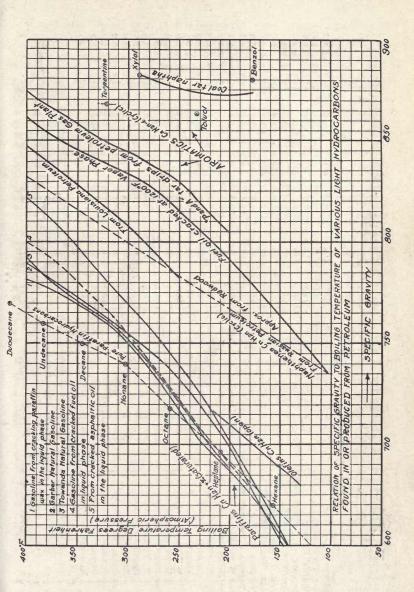
The same water used for condensing the benzine or gasoline fraction in crude distillation may be used to condense the kerosene fraction.



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BULLETIN NUMBER FIFTEEN OF





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Petroleum	
Heavy	
Fests on Different Heavy	ydrocarbons
no	ocal
Tests	Hydr
Cracking Te	
Equilibrium	

No. 10	0.994	10.8	500	500	50	410	45.0	12.5	8.5	350	0.898	25.9	110	70.0	30.0	109	21.8	0.746	57.6	48.2	0.982	12.6	530
No. 9	0.886	31.6	500	66 14	61	414	55.0	9.0	6.3	470	0.842	36.2	37	94.0	6.0	173	34.6	0.748	57.1	59.4	0.925	21.3	86
	0.820																						
No. 7	0.889	27.5	500	272	59.5	415	54.5	9.5	6.8	470	0.861	32.6	42	94.0	6.0	157	31.4	0.754	55.6	68.6	0.911	23.6	88
No. 6	0.946	18.0	500	038	58.5	412	54.5	11.5	8.0	442	0.887	27.8	47	88.4	11.6	118	23.6	0.753	55.9	64.8	0.944	18.3	218
	0.953																						
	0.820																						
	0.868																						
No. 2	0.935	19.7	500	360	60	420	55	10	7	460	0.862	32.4	47	92.0	8.0	139.5	27.9	0.746	57.6	64.1	0.926	21.2	178
No. 1	Specific Gravity 0.912					417		g (Atms.). 10	Gas % by Weight 7	Recovered—cc465 4	0.862	Baume' Gravity 32.4	47	93.0	7.0	.c127	25.4	0.743	Baume' Gravity 58.4	Residuum % Volume 67.6	Specific Gravity 0.926	Baume' Gravity 21.2	Viscosity at 70°F135 1

No. 1 = Mid-Continent fuel oil average of 48 cars on Kansas' City market.

California heat treated and skimmed.

11 11 11

Mexican flux oil (natural). Mid-Continent kerosene. Mid-Continent gas oil. Healdton crude.

1

il 9 8 10 00 NO. 0. No.

> Heavy Kansas crude oil from Allen County. Sarber residuum f1om Enid, Oklahoma. Ш NO. 3 NO. 4 NO. 5

11

California crude oil. araffin wax. j I Ш

228

BULLETIN NUMBER FIFTEEN OF

Effect of Varying Pressure on the Products of Cracking

KEROSENE.

Using kerosene of specific gravity	0.8155	in vesse	el with a	relation
of vapor space to oil of 2 to 1.				
Pressure, atmospheres	40	55	75	90
% distillate to 410°F28.0	32.5	38.0	43.7	45.9
Shrinkage, volume % 0.0	0.4	2.4	5.0	7.0
Specific gravity of cracked oil810	.808	.807	.806	.805
Specific gravity of residue828	.833	.845	.871	.888
Cold pressure, atmospheres 2.5	4.0	6.5	10.0	11.8

FUEL OIL.

Fuel oil with specific gravity of	0.908 in	vessel	with rela	tion of
vapor space to oil of 2 to 1.				
Pressure, atmospheres	40	55	75	90
% distillate to 410°F14.3	22.3	25.4	32.5	38.7
Shrinkage, volume % 3.0	3.3	9.0	12.0	14.0
Specific gravity of cracked oil879	.869	.862	.837	.818
Specific gravity of residue914	.918	.926	.930	.932
Cold pressure, atmospheres 5	6	10	13	15.5

Properties of Water White Kerosene Distillate Before and After Cracking

(See page 212.) Distilling Temperature Gravity of Stream Before After Before After % Cracking Cracking Cracking Cracking 294°F 0 Room 2.5 355 Room $.766 = 53.2^{\circ} \text{Be}$ 80°F 5.0 363 $.614 = 98.9^{\circ}Be$.767 = 52.9°Be .634 = 91.7°Be' 7.5 366 105 10.0 367 130 $.768 = 52.7^{\circ} \text{Be}'$ $.654 = 84.8^{\circ} \text{Be}'$ $.769 = 52.5^{\circ} \text{Be}'$ $.667 = 80.6^{\circ} \text{Be}'$ 12.5 370 158 $.770 = 52.2^{\circ} \text{Be}'$ 15.0 379 188 $.680 = 76.6^{\circ} \text{Be}$ $.771 = 52.0^{\circ} \text{Be}'$ 17.5 381 218 $.695 = 72.1^{\circ} \text{Be}^{\circ}$ 20.0 382 237 $.772 = 51.8^{\circ} \text{Be}^{\prime}$ $.710 = 67.8^{\circ} \text{Be'}$ 22.5 $.773 = 51.5^{\circ} \text{Be}'$ 384 256 $.720 = 65.0^{\circ} \text{Be}'$ 25.0 391 269.774 = 51.3°Be $.730 = 63.3^{\circ} \text{Be}'$ 27.5 $.774 = 51.3^{\circ} \text{Be}'$ 395 282 $.739 = 59.9^{\circ} \text{Be}'$ $.749 = 57.4^{\circ} \text{Be}'$ 30.0399 296 $.775 = 51.0^{\circ} \text{Be}'$ 32.5 402 310 $.776 = 50.8^{\circ} \text{Be}'$ $.756 = 55.6^{\circ} \text{Be}'$ 35.0 .777 = 50.6° Be' $.764 = 53.7^{\circ} \text{Be}'$ 406 319 37.5 408 328 $.777 = 50.6^{\circ} \text{Be}'$ $.769 = 52.5^{\circ} \text{Be}'$ 40.0 410 340 $.778 = 50.3^{\circ} \text{Be}'$ $.775 = 51.0^{\circ} \text{Be}'$ 42.5 352 $.779 = 50.1^{\circ} \text{Be}'$ $.777 = 50.6^{\circ} \text{Be}'$ 414 45.0 417 359 $.780 = 49.9^{\circ} \text{Be}'$ $.780 = 49.9^{\circ} \text{Be}$ $.780 = 49.9^{\circ} \text{Be}'$ 47.5 420 366 $.782 = 49.4^{\circ} \text{Be'}$ 50.0 423 371 $.781 = 49.6^{\circ} \text{Be'}$ $.785 = 48.7^{\circ} \text{Be}'$ 52.5425 376 $.782 = 49.4^{\circ} \text{Be'}$ $.787 = 48.3^{\circ} \text{Be}'$ 55.0 431 386 $.783 = 49.2^{\circ} \text{Be}'$ $.790 = 47.6^{\circ} \text{Be}'$ 57.5 433 396 $.784 = 48.9^{\circ} \text{Be'}$ $.792 = 47.1^{\circ} \text{Be}'$ 60.0 437 405 $.785 = 48.7^{\circ} \text{Be}'$ $.793 = 46.9^{\circ} \text{Be}'$ 62.5 440 $.786 = 48.5^{\circ} \text{Be'}$ $.795 = 46.4^{\circ} \text{Be}'$ 414 65.0 $.787 = 48.3^{\circ} \text{Be}'$ 444 418 $.798 = 45.8^{\circ} \text{Be'}$ 67.5 448 422 $.788 = 48.0^{\circ} \text{Be}'$ $.798 = 45.8^{\circ} \text{Be}'$ 70.0453 429 $.789 = 47.8^{\circ} \text{Be}'$ $.800 = 45.4^{\circ} \text{Be}'$ 72.5 457 $.802 = 44.9^{\circ} \text{Be}'$ 436 $.790 = 47.6^{\circ} \text{Be}'$ 75.0 462 443 $.792 = 47.1^{\circ} \text{Be}'$ $.805 = 44.2^{\circ} \text{Be}'$ 77:5 468 450 $.793 = 46.9^{\circ} \text{Be}'$ $.808 = 43.6^{\circ} \text{Be}'$ 473 80.0 459 $.794 = 46.7^{\circ} \text{Be}'$ $.812 = 42.7^{\circ} \text{Be}'$ 82.5 479 468 $.795 = 46.4^{\circ} \text{Be}'$ $.817 = 41.7^{\circ} \text{Be}'$ 85.0 485 484 $.797 = 46.0^{\circ} \text{Be}'$ $.823 = 40.4^{\circ} \text{Be}'$ 87.5 493 500 $.800 = 45.3^{\circ} \text{Be}'$ $.830 = 38.9^{\circ} \text{Be'}$ 90.0 506 523 $.803 = 44.7^{\circ} \text{Be'}$ $.837 = 37.5^{\circ} \text{Be}'$ 92.5 516 547 $.807 = 43.8^{\circ} \text{Be}'$ $.851 = 34.7^{\circ} \text{Be}'$ 95.0 533 600 $.812 = 42.7^{\circ} \text{Be}'$ $.866 = 31.9^{\circ} \text{Be}'$ 97.5 560648 $.936 = 19.6^{\circ} \text{Be}$ 100.0 608 700

Gravity of sample

.7845=48.9°Be'

 $.766 = 53.2^{\circ}\mathrm{Be'}$

FRACTIONAL GRAVITY DISTILLATION ANALYSIS

of Benton Process Gasoline; Specific Gravity, 0.758; °Be' U. S., 54.7 °Be' Tag, 55.1°; Olefins, 16.0%.

%	Time	Temp. °F.	Gravity of Fraction.	Gravity of Total Over	Gravity of Stream
	10:09	1821		and the second	
0	10:14	85 155			1.2
ō	10:22	164 171	0.694=72.4°Be'	0.694=72.4°Be/	0.694=72.4°Be
10	10:28	176 184	0.695=72.1°Be	0.694=72.4°Be	0.689=71.2° Be
15	10:35	188 193	0.701=70.3°Be'	0.696=71.8°Be	0.705=69.2°Be
20	10:42	199 206	0.710=67.8°Be'	0.700=70.6° Be'	0.714=66.6° Be
25	10:48	211 216	0.718=65.5°Be⁄	0.704=69.5°Be'	0.722=64.4°Be
30	10:54	222 228	0.727=63.1°Be'	0.707=68.6°Be	0.731=62.0°Be
35	10:58	234 238	0.735=61.0°Be'	0.711=67.5°Be⁄	0.738=60°2Be
40	11:03	244 248	0.742=59.2°Be'	0.715=66.4°Be	0.745=58.4°Be
45	11:00	254 258	0.748=57.6°Be'	0.719=65.3°Be'	0.751=56.9°Be
50	11:14	264 270	0.755=55.9°Be	0.722=64.4°Be'	0.758=55.1°Be
55	11:19	278 283	0.761=54.4°Be	0.729=62.6°Be	0.770=52.2°Be
60	11:25	290 297	0.767=52.9°Be	0.729=62.6° Be	0.770=52.2°Be
(5	11:29	306 312	0.773=51.5°Be'	0.732=61.8°Be/	0.776=50.8°Be
70	11:34	320 328	0.779=50.1°Be	0.736=60.7°Be	0.781=49.6°Be
75	11:41	336 348	0.784=48.9°Be	0.739=59.9°Be	0.788=48.0°Be
80	11:46	362 371	0.793=46.9°Be	0.742=59.2°Be/	0.797=46.0°Be
85	11:53	388 406	0.801=45.1°Be	0.746=58.1°Be	0.808=43.6°Be
90	11:59	428 460	0.815=42.1°Be	0.749=57.4°Be	0.823=40.4°Be
95	12:05	492	0.832=38.5°Be'	0.754=56.1°Be'	

Remarks: 36 cc. residuum; loss, 1/2 %.

FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF COAL TAR BENZOL.

Laboratory Number, 44118; Specific Gravity, 0.880; °Be' U. S., 29.0°; Cold test, 40° F.

%	Time	Temp. °F.	Gravity of Fraction.	Gravity of Total Over	Gravity of Stream
0	3:25 3:31	173 178			
5	3:37	179 180	0.882=28.9°Be'	0.882=28.9°Be'	0.881=29.1°Be
10	3:42	180 180	0.881=29.1°Be'	0.881=29.1°Be	0.882=28.9°Be
15	3:47	180 180	0.883=28.7°Be'	0.882=28.9°Be'	0.882=28.9°Be
20	3:51	180 180	0.882=28.9°Be'	0.882=28.9°Be'	0.882=28.9°Be
25	3:56	180 180	0.882=28.9°Be'	0.882=28.9°Be'	0.882=28.9°Be
30	4:00	181 181	0.882=28.9°Be'	0.882=28.9°Be	0.882=28.9°Be
35	4:05	182 182	0.882=28.9°Be'	0.882=28.9°Be/	0.881=29.1°Be
40	4:10	182 182	0.881=29.1°Be′	0.881=29.1°Be′	0.881=29.1°Be
45	4:15	182 182	0.881=29.1°Be'	0.881=29.1°Be'	0.881=29.1°Be
50	4:19	182 183	0.881=29.1°Be′	0.881=29.1°Be/	0.880=29.3°Be
55	4:23	183 183	0.880=29.3°Be'	0.881=29.1°Be'	0.880=29.3°Be
60	4:28	184 184	0.880=29.3°Be'	0.881=29.1°Be/	0.880=29.3°Be
65	4:33	184 185	0.880=29.3°Be'	0.881=29.1°Be'	0.880=29.3°Be
70	4:38	186 186	0.880=29.3°Be'	0.881=29.1°Be'	0.880=29.3°Be
75	4:43	187 188	0.880=29.3°Be'	0.881=29.1°Be'	0.880=29.3°Be
80	4:48	189 190	0.880=29.3°Be'	0.881=29.1°Be'	0.879=29.4°Be
85	4:53	192 196	0.879=29.4°Be'	0.880=29.3°Be'	0.879=29.4°Be
90	4:57	199 205	0.879=29.4°Be	0.880=29.3°Be'	0.877=29.8°Be'
95 100	5:01 5:10	216 225	0.876=30.0°Be' 0.876=30.0°Be'	0.880=29.3°Be' 0.880=29.3°Be'	0.876=30.0'Be° 0.876=30.0'Be°

Information Concerning Oil Shales

The chief occurrences of oil shale in the United States are in Western Colorado—Northeastern Utah—Kentucky—Elko, Nevada— Great Falls, Montana—Parkfield, California—New Brunswick, Canada—Alabama—Tennessee and Virginia. It is estimated that in Colorado there are enough oil shales to produce 20,000 million barrels of oil and 300 million tons of ammonium sulphate.

The shale oil industry started in England in 1694. The oil was used for medicinal purposes, later for varnishes and in 1815 for ammonia.

The chief commercial operations on oil shale are in Scotland and were begun in 1847. These industries were demoralized when Pennsylvania petroleum first appeared on the market, but later recovered partially and are now operated with profit. The amount of oil obtainable from one ton of shale varies from one gallon to 90 gallons. In Scotland it is 23 gallons. In Colorado alone there is said to be enough shale to produce 20,000,000 barrels of oil and 300,000,000 tons of ammonium sulphate.

Gasoline made from shale is of inferior quality, containing large amounts of olefins and aromatic compounds and giving a large shrinkage on refining.

Shale oil is especially adapted to the uses to which the heavy products of petroleum are now put, such as fuel oil, paraffin wax, lubricants, gas oil and illuminating oil. It is not likely to be so satisfactory for the production of gasoline as is the cracking of heavy petroleums. The character of the oil recovered and the amount of ammonium sulphate produced from shale depend largely upon the method of distillation.

Oil shale rock is a tough brownish to black shale-like rock. As it naturally exists it contains no oil and oil cannot be extracted from it by solvents or by any of the means used for asphaltic sandstone or limestone. The oil is produced from complex organic matter by decomposing it at high temperatures.

The mineral base of oil shales is of the nature of kaolin and contains potash in water-insoluble form.

Cannel coal is of the same chemical nature as oil shale both as to the bitumen and the mineral matter. The hydrocarbons of oil shale and cannel coal more nearly approach petroleum than coal in their calorific value.

Unlike coal, cannel "coal" has no structure or evidence of the former presence of or origin from vegetable matter. It breaks with a conchoidal fracture and is usually free from mineral sulphides such as pyrites of iron. It commonly occurs on the top of the Mississippian (subcarboniferous) and may lie immediately above deposits of galena or sphalerite (zinc).

Presumptive Operation of 1000-Ton Shale Oil Plant in Western Colorado (Based upon 1 ton of shale.)

(Dased upon 1 ton of shale.)		
	918	1913
54 gallons of oil (405 lbs.)\$	2.70	\$ 1.00
34 pounds of ammonium sulphate		1.09
	5.16	\$ 2.09
Costs.		
*Cost of mining\$	1.35	\$ 0.90
Cost of distilling oil and ammonia	.65	.50
Cost of acid for ammonia	.55	.16
*Freight on acid to plant	.12	.12
Cost of preparation of ammonium sulphate for		
market.	.10	.06
*Freight on ammonium sulphate to market	.17	.17
Treight on annonium suphate to market		
*Freight on oil	1.00	1.00
Overhead expense.	.40	.25
\$	4.34	\$ 3.16
*Donond upon local conditions to a large extent		

*Depend upon local conditions to a large extent.

PROFITS IN SHALE INDUSTRY BY COMPANIES IN SCOTLAND IN 1910

Companies. Broxburn.	Dividends.
Broxburn.	
Oakland	15.0
Pumpherston	
Tarbrax	15.0
Youngs.	6.0
Delmenv.	5.0

SHALE OIL PRODUCTS

Yields from "Oil Shale" from Colorado

(100,000 million tons of shale of this quality are said to	o be available.)
Oil $= 405$ lbs. $= 54$ gallons	=20.25%
Water $= 83$ lbs. $=10$ gallons	= 4.08%
Gas ==1605 cu. ft.	= 8.86%
Ammonium Sulphate=34 lbs. from nitrogen	= 0.90%
Carbon (not separable)=101 lbs.	= 5.05%
Mineral matter=1219.2 lbs.	=60.96%

COMPOSITION OF MINERAL ASH IN	SHALE
Loss on ignition	= 11.05%
Silica	= 37.10%
Alumina. (Al_2O_3)	= 20.30%
Iron Oxide. (Fe_2O_3)	= 9.20%
Lime	= 12.05%
Magnesia	= 5.10%
Sulphur. (SO_3)	= 4.80%
Alkalies and difference	= 0.40%

100.00%

PROPERTIES OF SHALE OIL

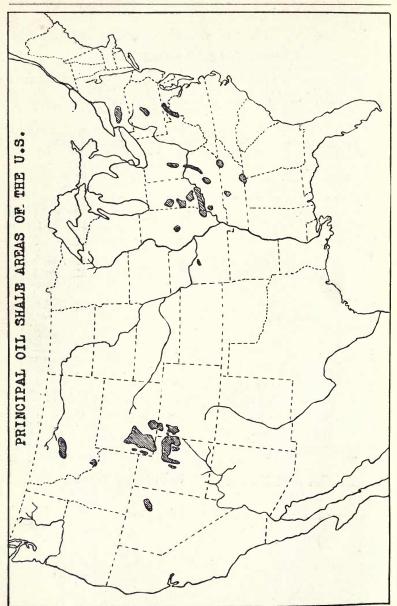
Commercial Fractions.		
Naphtha (410°F) "gasol	ine"	10.0% (46° Baume')
Burning oil		
Gas and lubricating oil		
Scale.		10.0%
Fractional Distillation of oil.		
Fraction	Boiling Point	Specific Gravity (25°C)
0-10	100°C	0.794=46.3°Be'
10-20	194	0.822=40.3
20- 30	230	0.846 = 35.5
30-40	255	0.867=31.5
40- 50	285	0.885=28.2
50- 60	309	0.899=25.7
60- 70	328	0.912=23.5
70- 80	337	0.900 = 25.5
80-90	345	0.910=23.8
90-100	350	0.910=23.8

CANNEL COAL FROM CENTRAL MISSOURI (Large quantities of this hydrocarbon are found in Missouri.) Sample Sample

	Sample	Sample
	a	b
Moisture.	8.14%	2.56%
Volatile hydrocarbons.	41.16	44.78
	36.63	42.72
Fixed carbon.	14.07	9.94
Ash	14.07	9.94
	100.00	100.00
Fusing of bitumen	none	none
Total combustible	77.79	87.50
Heating value in B. T. U., per lb	12575	14095
B. T. U., per lb. of combustible	16165	16110
	2.10%	1.70%
Sulphur.	1.50	1.65
Nitrogen.		
	gallons	72 gallons
Ammonium sulphate, per ton 50	pounds	55 pounds
Coke, per ton		200 pounds
The second		-

Silica Iron and Alumina	COMPOSITION OF ASH IN	(SiO ₂)=43.28%	46.16
Magnesia. Sulphur. Phosphoru	s	$\frac{\dots (MgO)}{\dots (SO_3)}$)= 1.49)= 1.01)= 0.84)= 0.73)= 3.00	

BULLETIN NUMBER FIFTEEN OF



FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF SHALE OIL BEFORE CRACKING.

Laboratory Number 46258, Original Shale Oil. Specific Gravity, 0.920; °Be' U. S. 22.1°; °Be' Tag 22.3°. Color, Brownish Black; Sulphur=0.49% B. T. U.%18,425.

76	Temp. °F.	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
-		In sole in the		0.790=47.6°Be
5	330 368	0.790=47.6°Be	0.790=47.6°Be'	0.802=44.9° Be' 0.814=42.3° Be'
10	378 398	0.814=42.3°Be'	0.802=44.9°Be/	0.823=40.4° Be/ 0.833=38.3° Be/
15	413 426	0.833=38.3°Be'	0.812=42.7°Be/	0.839=37.1°Be 0.845=35.9°Be
20	446 464	0.845=35.9°Be'	0.820=41.0°Be′	0.853=34.4°Be 0.861=32.8°Be
25	479 494	0.861=32.8°Be'	0.828=39.4°Be/	0.869=31.3°Be 0.876=30.0°Be
30	516 530	0.876=30.0°Be'	0.836=37.7°Be	0.883=28.7°Be/ 0.890=27.5°Be/
35	543 552	0.890=27.5°Be'	0.844=36.1°Be'	0.895=26.6°Be/ 0.900=25.7°Be/
40	576 586	0.900=25.7°Be'	0.851=34.8°Be	0.905=24.8°Be/ 0.909=24.1°Be/
45	599 604	0.909=24.2°Be'	0.857=33.6°Be′	0.910=24.0° Be' 0.911=23.8° Be'
50	613	0.911=23.8°Be'	0.807=31.7°Be/	0.916=23.0°Be' 0.922=21.9°Be'
53	Gas	0.922=21.9°Be/	0.872=30.7°Be′	0.928=21.0°Be' 0.934=20.0°Be'
60	Gas	0.934=20.0°Be/	0.877=29 8° Be'	0.937=19.5°Be' 0.940=19.1°Be'
65	Gas	0.940=19.0°Be'	0.882=28.9°Be'	0.943=18.5°Be 0.947=17.9°Be
70	Gas	0.947=17.9°Be'	0.887=28.0° Be	0.950=17.4°Be

Summary:

Water. 42.7° Benzine or Naphtha 31° Illuminating oil, unrefined	12.9% 25.0%	Olefins Aromatics Naphthenes an	 27.0%
24° Gas. Oil or Distillate	10.0%		
18.5° Wax Distillate	30.0%		
Residue			
Ammonia in water portion		42% as NH.	

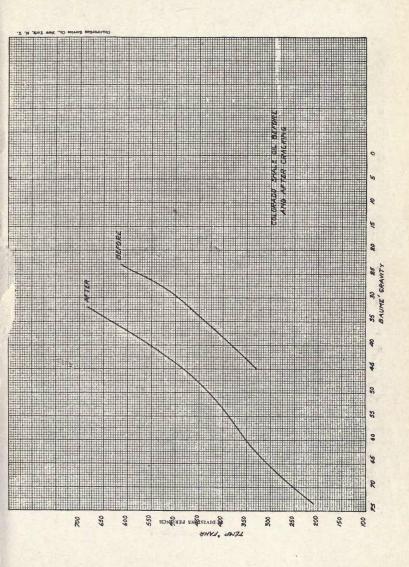
Ammonia in water portion = 0.442% as NH₃.

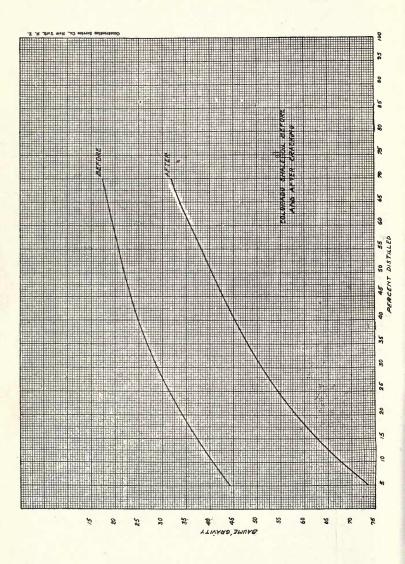
FRACTIONAL GRAVITY DISTILLATION ANALYSIS OF SHALE OIL RESIDUE.

Laboratory Number 46258, Shale Oil Residue Cracked at 800 lbs. Pressure. Specific Gravity, 0.896; °Be' U. S. 26.2; °Be' Tag 26.4. Color, Dark Red; Olefins 27.5%.

%	Temp. °F.	Gravity of Fraction	Gravity of Total Over	Gravity of Stream
0	119		1 - 1 - 1 - 1 - 1	
~	110		E -	0.681=76.3°Be
5	210	0.681==76.3°Be	0.681=76.2°Be	0.690=73.6° Be
			the second second second	0.699=70.9°Be
10	281	0.717=65.8°Be	0.699=70.9°Be'	0.710=67.8°Be
				0.721=64.7°Be
15	334 *	0.765=53.5°Be	0.721=64.7°Be	0.730=62.3°Be
			1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1. 1	0.740=59.7° Be
20	369	0.798=45.8°Be'	0.740=59.7°Be'	0 748=57.7°Be
1.5	All the second second	the second s		0.757=55.4°Be
25	. 395	0.823=40.4°Be'	0.757=55.4°Be'	0.764=53.7° Be
		The second second second		0.771=52.0°Be
30	435	0.846=35.7°Be'	0.771=52.0°Be	0.777=50.6° Be
				0.784=490°Be
35	454	0.861=32.8°Be'	0.784=49.0°Be	0.790=47.6°Be
				0.796=43.2°Be
40	486	0 881=29.1°Be'	0.796=46.2°Be	0.801=45.1°Be
		a start for a set of		0.807=43.8° Be
45	518	0.898=26.1°Be/	0.807=43.8°Be'	0.81 =42.7°Be
-	a second second		A Contract of the second second	0 818=41.5°Be
50	543	0.911=23.8°Be'	0.818=41.5°Be'	0 823=40.4°Be
				0 828=30.4° Be
55	582	0.930=20.7°Be'	0.828=39.4°Be	0.833=38 3°Be
~			K	0 838=37.3°Be
60	623	0.945=18.2°Be'	0.838=37.3°Be'	0 844=36.1°Be
CE	071	0.050 10.000.		0 855=34 0°Be
€5	651	0.959=16.0°Be	0.855=34 0° Be'	0 8 9=33 2°Be
70	070	0.025 15 1970-/	0.000 00.007	0.862=32.6°Be
10	679	0.935=15.1°Be'	0.832=32.6°Be'	0 865=32.0°Be

Naphtha in oil Synthetic Oil-	charged	None





Products of Refining of Light Oil of Gas Works

	Carbon Disul- phide	Benzene	Toulene	M-zylene	Naph- thalene
Molecular weight	76.12	78.05	92.06	106.08	128.06
Pounds per United States Gal. (60° F).	10.57	7.36	7.27	7.26	9.60
opecific gravity (0°C/4°C)	1.2921	.8999	.8845	.8823	
Specific gravity (10°C/4°C)	1.2773	.8893	.8757	.8738	
Specific gravity (15°C/4°C)	1.2698	.8839	.8714	.8697	1.1517
Specific gravity (20°C/4°C)	1.2623	.8786	.8659	.8655	
Specific gravity (30°C/4°C)	1.2473	.8679	.8573	.8574	
Change of Spec. Grav. per 1°C	.00125	.0012	.0010		
Boiling point at 760 mmHg (°C)	43.2	80.36	110.3	139.1	217.7
Increase in boiling point (°mmHg)	.041	.043	.047	.052	.059
Vapor pressure mmHg (0°C)	127.9	26.63	7.20	1.75	.022
Vapor pressure mmHg (10°C)	198.5	45.68	13.02	3.45	.047
Vapor pressure mmHg (15°C)	244.1	58.90	17.22	4.74	.062
Vapor pressure mmHg (20°C)	298.0	75.21	22.53	6.43	.080
Vapor pressure mmHg (30°C)	434.6	119.34	37.46	11.43	.135
Pounds per cu. ft. vapor (60°F=30 in.)	.202	.209	.244	.281	.339
Kil. per cu. m. vapor (0°C-760mm)	3.42	3.54	4.14	4.76	5.72
Heat combustion (net) 15°C-760mmHg.	0.40	0.01			
Calories per kil. liquid	.3480	.9960	10.150	10.230	.9700
Calories per liter, liquid		.8805	.8850	.8910	11.170
B. T. U. per pound, liquid		17.930	18.270	18.410	17.460
B. T. U. per U. S. gal., liquid		132.100	132.600	133.500	167.300
Calories per cu. meter, vapor		33.600	40,150	46.500	52,400
B. T. U. per cu, ft. vapor		.3780	.4500	.5210	.5910
Specific heat (calories per kil.)		0.419	0.440	0.383	0.814
Heat of vaporiz. (calories per kil.)		92.9	83.55	78.25	0.011
Sol. in water (22°C) grm. subs. in 100	00.0	04.0	00.00	10.40	
	.219	.072	Insol.	Insol.	Insol.
		.241	Insol.	Insol.	Insol.
Grams H2O in 100g subs Melting point (°C)		+5.4	-92.4	-54.8	+80.0

Gas-Manufacturing Processes in Use in the United States

The manufactured gas distributed in the United States is of three principal kinds: Coal gas, carbureted water gas and oil gas.

The manufacture of water gas consists essentially of an intermittent process in which a bed of anthracite coal or coke is brought to a high temperature by an air blast and then steam under pressure is blown through the fuel, forming carbon monoxide, hydrogen and a small amount of carbon dioxide by reaction with the carbon in the fuel. The resultant gas, called blue water gas, has a heating value of approximately 300 B.T.U. per cubic foot and almost no luminosity when burned in an open flame. It is conducted into a fire-brick-lined chamber called the carburetor, which contains staggered rows of fire bricks, called checker brick, heated to incandescence during the blow period. Gas oil or fuel is sprayed into the carburetor while the gas is passing through, forming an oil gas which enriches the blue water gas to any desired heating value or candlepower. Another checkerbrick-filled chamber, called the superheater, converts most of the oil-gas vapors into permanent gases, which will not condense again upon cooling. During the formation of the oil gas certain portions of the hydrocarbons which compose the oil are changed in their composition to form benzol, toluol and related hydrocarbons called aromatic compounds. Considerable tar is formed at the same time. This is condensed, scrubbed and washed out of the gas by various means, but usually at a temperature which permits most of the aromatics to go forward with the gas. The sulphur in the gas is removed by ironoxide purifiers and the gas is metered and leaves the plant at or slightly above atmospheric temperature.

The manufacture of coal gas' is essentially different from that of water gas. In this process certain classes of bituminous coals are distilled in fire clay or silica retorts or ovens and the resulting gases are condensed, scrubbed, washed and purified to remove water vapor, tar, ammonia and sulphur. As in the water gas process, certain of the hydrocarbons given off by the coal are transformed by the heat of the retort to aromatic compounds. A small part of these aromatics is washed out of the gas by the wash water and tar, but the larger part remains in the gas. In fact, the cooling of the gas is usually so regulated that most of these substances will remain in the gas to in-crease its heating value and candlepower. Coal gas retorts take a variety of forms. Among these are coke ovens, chamber ovens, horizontal D-shaped retorts, vertical retorts, inclined retorts, etc. Even those of a given class differ among themselves in details of construction. In most of them the distillation is an intermittent process, but some continuous methods are used. In all these processes the gas produced consists of the same constituents in somewhat different proportions. The form of apparatus used in a given case depends largely upon economic considerations or is governed by certain special qualities which are desired in one or more of the products produced. In all of these coal gas processes coke remains in the retort after distillation. In some of them, as for example in coke ovens, coke is the principal product, but in city gas plants gas is the chief product. The operation is carried out in any case to give most satisfactory qualities to the principal product and at the same time obtain as high yields and good quality as possible of the secondary or by-products.

Mixed gas is usually understood to be a mixture of carbureted water gas and coal or coke-oven gas. It is supplied in many cities in the United States where the requirements permit of a mixed gas being supplied. The manufacturing installation for mixed gas is practically two complete installations, one for coal gas and one for carbureted water gas, with their auxiliary scrubbing, condensing, purifying, and metering apparatus entirely independent and separate. The manufactured mixed gas, however, is stored in common holders and delivered through a single distribution system. The coal and water gas thus supplement each other. The uniform but more cumbersome coal-gas production furnishes coke as fuel for the water-gas plant. This in turn takes care of the irregularities of the output, and, where necessary, increases the quality of the gas production, especially where a high candlepower standard is in force.

The oil gas process is at present confined chiefly to the Pacific Coast States, where comparatively cheap oil and expensive coal make the coal and water gas processes less feasible. In oil gas manufacture oil alone is used as fuel for heating the checker bricks of the fixing chambers and oil is sprayed by steam into the chambers where, in contact with the bricks, lampblack and permanent gases are formed. In this process also aromatic compounds are included among the constituents of the gas.

Note.-See Bulletin of Bureau of Standards.

Average Content of Light Oils in Various Gases

The amount of benzol and toluol formed in any one of these processes is by no means definite. It depends upon the operating conditions and the quality of the raw materials (coal or oil). It would therefore be impossible to predict exactly what the yield of products in a given case would be, but an extensive inquiry into the operation of a number of typical plants has given the following tabulation as the usual range of figures for the various processes. Individual results may vary widely from them in a particular case.

 TABLE 1.—Approximate Yields of Crude Light Oil and Pure Products and Approximate Composition of Crude Light Oil.

APPROXIMATE YIELD OF CRUDE LIGHT OIL.

Coal	gas.
------	------

Horizontal retort	3.0-4.0 gallons per short ton coal car-
Continuous mention) actent	bonized
	1.5-2.5 gallons per short ton coal car- bonized
Inclined retort	1.8-2.3 gallons per short ton coal car- bonized
the second s	2.6-3.6 gallons per short ton coal car- bonized
Carbureted water gas	8-10 per cent of vol. of gas oil used 0.2-0.3 gal. per 1000 cu. ft. of gas.
	0.2-0.0 gai. per 1000 cu. 1t. 01 gas.

APPROXIMATE COMPOSITION OF CRUDE LIGHT OIL.

Solvent

			Naphtha.
	Benzol	Toluol	Wash Oil,
			Naphthalene,
Coal gas:	Per Cent	Per Cent	
Horizontal retort.	50	13-18	35
Continuous vertical retort	30	10-15	55
Inclined retort.	45	13-18	40
Coke-oven gas, run of oven	50	14-18	35
Carbureted water gas	40	20-25	37
Oil gas	80	8-10	10
APPROXIMATE YIELD			
Gallons per short ton coal carbonized:		Benzol	Toluol
Coal gas-	1.2.2.2.1	Denzoi	101001
Horizontal retort.		1.5	0.4-0.5
Continuous vertical retort	• • • • • • • • • •		.23
Inclined retert	• • • • • • • • • •	.6	
Inclined retort.	• • • • • • • • • •	.9	.24
Coke-oven gas, run of oven		1.5	.35
Gallons per 1000 cubic feet of gas:		1.	01100
Carbureted water gas	• • • • • • • • • •	.15	.0610
Oil gas		.25	.0203
			Degrees
	States of the second		Boiling
Paraffins	Specifi	c Gravity	Point in
NT 1		16. 20 A.	Centigrade
N-heptane.	. 0.712, :	at 16°C	97
Triethylmethane.	689. :	at 27°C	96
Noctane.	708. 2	at 12.5°C	125
Diispbutyl.	714, :	at 0°C	108.5

Natural Gas

produces and for the second states of

Natural gas is found trapped in the various strata of the earth, principally in sandstone formations of loose texture, in shale seams and in cavities. It is usually associated with petroleum or coal and occurs in the carboniferous strata or in more recent formations. In coal mines it constitutes what is known as fire damp, being given off from the exposed seams of coal. It is most commonly associated with petroleum in petroleum bearing sand and occupies the space in the sand above the oil. Occasionally it occurs in strata without any oil being present, in which case it is of a slightly different composition than the gas which is found in contact with the oil. In many cases it appears that the gas has been obtained from the atmosphere, the oxygen having been removed by its combination with reducible substances such as sulphides, leaving a residue of nitrogen. This gives to such natural gases the peculiarity of having a very large amount of nitrogen. Associated with the nitrogen there occasionally is found a small amount of Helium which is also an ordinary constituent of air in small quantities. It may be that the difference of solubility of the different gases of the air in water may account for the tendency of accumulation of Helium in such instances. As a rule, however, natural gas consists of hydrocarbons of the same type as petroleum and identical with the hydrocarbons which are given off by the cracking of petroleum.

The proportions in which the different hydrocarbons exist in ordinary gas such as is delivered to Kansas City, Missouri, is something like the following:

Iethane	1.7%
Ethane	.4%
Propane	.0%
Butane	.3%
Nitrogen	

This gas has the greater portion of the heavy hydrocarbons condensed out on account of the high pressure in the pipe lines. Such a gas is a mixture of methane with a varying amount of the other gases. As shown by the above table, the gases ethane, propane and butane furnish much of the heating value of the gas. A gas with a considerable amount of gasoline vapor in it will have a considerably higher heating value than one from which it has been removed, or known as a dry gas.

The compositions of the natural gas used in eight cities in the United States are as follows:

Stranger Warden Barris	Methane	Ethane	Nitrogen
City	Per Cent	Per Cent	Per Cent
Pittsburgh, Pa	79.2	19.6	1.2
Louisville, Ky	77.8	20.4	1.8
Buffalo, N. Y	79.9	15.2	4.9
Cincinnati, O		19.5	.7
Cleveland, 0		18.2	1.3
Springfield, O		14.7	5.0
Columbus, 0		18.1	1.5
Chelsea, Okla		17.7	6.6

These analyses were made by the ordinary combustion method and hence show only the two predominating paraffin hydrocarbons.

The composition of gases found in Kansas and Oklahoma as given by Allen and Lyder are shown by the following table: D M IT man

				B.I.U. per
Location	Methane	Ethane	Nitrogen	Cubic Foot
Augusta, Kas	. 10.54	1.64	87.69	129
Cowley County, Kas	. 16.27	3.01	80.23	209
Chautauqua County, Kas		1.85	55.29	441
Chautauqua County, Kas		3.89	46.67	541
Elsworth, Kas		1.09	37.20	609
Ponca City, Okla		14.86	40.10	688
Kay County, Okla		9.89	31.65	735
Chautauqua County, Kas		0.15	12.95	839
Chautauqua County, Kas		7.79	11.39	894
Butler County, Kas		18.38	18.64	930
Montgomery County, Kas		8.54	7.95	970
Blackwell, Okla		18,65	9.32	1025
Cushing, Okla		21.64	7.49	1059
Bartlesville, Okla	. 70.50	24.60	3.21	1125

The presence of such a large amount of nitrogen in some cases makes the gas almost valueless unless some process is used whereby the nitrogen may be adapted to chemical processes.

While natural gas has a very high heating value in comparison with water gas, water gas has the advantage in that it gives a more intense flame. The comparison of various commercial gases is shown in the following table:

PROPERTIES OF NATURAL AND MANUFACTURED GASES. Ave

				Tel Line	Producer
Avg.	Avg.		Avg.		Gas from
Constituents Pa. and O W. Va.		Avg.	Coal		ituminous
	Ind.	Kansas	Gas	Gas	Coal
Marsh gas, CH4 80.85	83.60	93.65	40.00	2.00	2.05
Other hydrocarbons 14.00	.30	.25	4.00	.00	.04
Nitrogen 4.60	3.60	4.80	2.05	2.00	56.26
Carbonic acid CO ₂ 00	.20	.30	.45	4.00	2.60
Carbonic oxide CO40	.50	1.00	6.00	45.50	27.00
Hydrogen	1.50	.00	46.00	45.00	12.00
Hydrogen sulphide00	.15	.00	.00	.00	.00
Oxygen trace	.15	.00	1.50	1.50	.05
Total	100.00	100.00	100.00	100.00	100.00
Pounds in 1,000 cu. ft. 47.50	48.50	49.00	33.00	45.60	75.00
Sp. grav. air being 1.00 0.624			0.435	0.600	0.935
B.T.U. per cu. ft 1,145	1.095	1.100	755	350	155

(a) 1,000 cu. ft. of air at an atmospheric pressure of 14.7 pounds (a) 1,000 cu. it. of all at an atmospheric pressure of 1.1, points and at a temperature of 62°F weighs 76.1 pounds and is a mechanical mixture of 23 parts of oxygen and 77 parts of nitrogen by weight.
(b) B.T.U. equals British thermal units, which indicate the heat necessary to raise one pound of pure water at 39°F one degree.

Natural gas may have its origin from a sand which is entirely separated from sand containing oil or it may come from above the oil in the same sand as oil.

In the latter case the lighter portions of the oil will have been volatilized and carried into the gas. Such a gas is known as a "wet" gas. In other words, the wet gas is composed of the usual constituents of dry gas; that is, methane, ethane, propane and butane, and in addition pentane, hexane and heptane. These last three are liquid at ordinary temperatures and are the most desirable components of gasoline.

Gas coming from a sand containing no oil is "dry" gas and does not contain the pentane, hexane and heptane.

A "wet" gas coming from an unknown sand indicates the presence of oil in that sand.

In the ordinary oil well the gas is allowed to escape between the casing of the well and the tube which has been inserted for withdrawal of the oil. The gas so collecting in the casing is known as casinghead gas and may be used or allowed to escape.

This gas collecting in the casinghead of an oil well is "wet" gas and contains some of the gasoline from the oil. The gasoline which may be compressed from it or refrigerated from it is then known as "casinghead" gasoline.

The lighter the oil with which the casinghead gas has been associated, the greater ordinarily will be the amount of gasoline contained in the gas.

Ever since natural gas has been conducted in pipe lines it has been known that gasoline could be separated by pressure and much has been incidentally so produced. More recently the great demand for gasoline has encouraged the design of hundreds of special plants for the extraction of gasoline from natural gas.

In 1904, at Titusville, Pennsylvania, Fasenmeyer made casinghead gasoline by pumping the gas under pressure through a coil under water.

In the early methods pressures of about 50 pounds per square inch were used. Later condensing with a pressure of 400 pounds per square inch was found to produce too "wild" a gasoline or one that escaped too easily on handling. A pressure of 250 pounds per square inch is now used, and the pressure of the condensed liquid is controlled by absorbing it directly into heavier naphtha.

At first the compression was done in one stage, but it is the custom now to do it in two stages. The gravity of the product is from 80 to 100° Baumé.

The amount of casinghead gasoline present in a gas will depend upon the character of the oil associated with it, the temperature, the pressure, the compactness of the sand and the condition in the sand at the point tapped.

The amount of gasoline obtained from casinghead gas in the Mid-Continent field varies from ½ to 8 gallons per 1,000 cubic feet. A typical gas yields 2½ gallons per 1,000 cubic feet. Many yield 3 to 4 gallons per 1,000 cubic feet.

The total production of casinghead gasoline in the United States is shown on page 24.

The cost of plants for producing casinghead gasoline has varied from \$12 to \$25 per thousand cubic feet of gas handled, and the operation of the plants has been uniformly successful and highly profitable.

While the type of plant ordinarily constructed is for compression methods, it is probable that the absorption method will be more generally adopted. The operation of the absorption method is similar to that of extracting toluol from coal gas and may be applied to a natural gas capable of yielding 1 pint of gasoline per 1,000 cu. ft. By the use of the absorption process 50 million cu. ft. of natural gas would be available per day and 100 million gallons of light gasoline would be made.

Yield of Gasoline from Casinghead Natural Gas by Compression Method, Corresponding to Absorption and Specific Gravity Tests.

Absorp- tion by Oil, per cent	Specific Gravity (Air=1)	Yield of Gasoline, Gallons per 1.000 Cubic Feet of Gas	Absorp- tion by Oil, per cent	Specific Gravity (Air=1)	Yield of Gasoline, Gallons per 1,000 Cubic Feet of Gas
16	0.64	None	50	1.29	3.00
23	.83	1.00	48	1.37	3.50
30	.90	1.75	44	1.38	3.50
37	1.00	2.00	65	1.38	4.00
39	1.03	2.50	84	1.41	4.50
38	1.07	3.00	86	1.46	5,00
54	1.21	3.50			

One casinghead plant figures its probable yield of gasoline in relation to the gravity (G) of the air free gas as follows: 2(15G - 10)

Recovery in gallons per 1,000 cu. ft. = 3

Two-thirds of this amount is marketed.

Helium in Natural Gas.

Locality	Helium	Nitrogen	Methane	Ethane
Dexter, Kansas	1.84%	82.70%	14.85%	0.41%
Eureka, Kansas		46.40	51.40	0.00
Fredonia, Kansas		16.40	82.25	0.00
Kansas City, Missouri		3.65	87.20	7.03
By H. P. Cady and D. F. I	McFarlan	d.—J. A. C.	S., Vol. 24,	p. 1530,

1907.

The chief helium producing natural gas is in Kansas with smaller amounts in the gas at Petrolia, Texas.

Properties of Incombustible Gases in Natural Gas.

	Helium (He)	Nitrogen (N)
Combining weight	1.99	14.01
Molecular weight	. 3.99	28.02
Specific gravity $(air = 1)$. 0.1368	0.96737
Liquefying point	268.5°C	-195.5°C
Freezing point	269.0°C	-210.5°C
Solubility in cold water	1.487%	2.348%
Absorption by platinum	great	little
Weight per cubic foot, pounds	01105	.07831
(Air = 0.080728)		

Extracting Helium From Natural Gas

The process is essentially one of liquefaction by cold and pressure. All of the constituents in natural gas are liquefied except the helium and then separated from the latter.

When the armistice was signed about 45,000 cubic feet of helium had been extracted and was waiting shipment overseas. Several million dollars had been invested in plant equipment in Texas. The cost of extraction was estimated at 10c per cubic foot.

Natural gas to be valuable as a source of helium should contain at least 0.50 per cent of the gas. It is probable that even this quantity will offer great difficulty in the extraction work, although with experience and cheaper methods which will come with practice even smaller quantities may be valuable. The largest quantity ever discovered in natural gas is something over 2 per cent.

smaller quantities may be valuable. The largest quantity ever discovered in natural gas is something over 2 per cent. The presence of helium in natural gas was discovered by H. P. Cady and D. F. McFarland of the University of Kansas in 1907. (See Journal of American Chemical Society, Vol. XXIX, p. 1523, November, 1907.)

Lifting Power of Gases in Balloons.

	Pounds per	Compared with
	1,000 Cu. Ft.	Hydrogen
Hydrogen	75.138 lbs.	100.0 %
Helium	69.748 lbs.	92.84%
Ammonia		44.16%
Natural gas (methane)	36.088 lbs.	48.03%

Gas Carbon Black From Natural Gas

About 1,000 cubic feet of natural gas of specific gravity of .86 are required to make one pound of carbon black.

The operation of making carbon black consists of burning the gas without air under a series of sheet iron shields which collect the carbon from the yellow flame.

The type of burner used is the old style lava tip originally used for lighting purposes with artificial gas. Many thousand tips are used at one plant.

The carbon is scraped off the shields and packed for shipment in 12½-pound sacks.

Plants of this character require very little labor and can be run under the supervision of the plant foreman, thus carrying little or no overhead expense.

Carbon black is mainly used in printers' ink and is a necessary article.

Carbon black is far superior for filler in rubber tires.

The market price of carbon black is from 12c to 25c per pound.

One thousand feet of natural gas contains 35 to 40 pounds of carbon. Practically no plants get over 2 pounds of gas carbon from 1,000 feet of gas, an average of about 1 pound. The smallest practical size unit should handle two million feet of gas per day, producing 2,000 to 3,000 pounds of carbon black. Ordinarily such a plant would cost not less than \$60,000.

About Natural Gas and Its Usefulness

An average sample of natural gas has 950 B.T.U. per cubic foot. 1 lb. mill coal will evaporate 9 lbs. water.

1 gal. oil will evaporate 100 lbs. water.

1 cu. ft. gas will evaporate 0.85 water.

1 ton coal used under boilers = 18,500 cu. ft. of gas.

1 bbl. oil (42 gal.) under boilers = 5,000 cu. ft. of gas.

40 to 50 cu. ft. of gas = 1 boiler H.P.

Gas Engines:

Highest grade gas engines develop a brake H.P. on 8,500 B.T.U. Average engine develops a H.P. on 10,500 B.T.U.

Oil well engine develops a H.P. on 20,000 B.T.U.

In a steam turbine plant of over 500 K.W. capacity 30 cu. ft. gas per K.W. is a fair average.

It requires 40,000 cu. ft. of gas to pump one million gallons of water against 200-foot head.

Brick Plants-Gas Used per Thousand Brick Made:

1,800 cubic feet for power.

1,800 cubic feet for drying.

15,000 cubic feet for kilns.

Ice Plants:

2,000 feet gas per ton of refrigeration.

Zinc Plants:

15,000 cubic feet for roasting per ton of metal produced.

65,000 cubic feet for smelting per ton of metal produced.

20,000 cubic feet for power and miscellaneous uses per ton of metal produced.

Cement Plants:

60 to 100 cubic feet per barrel for power.

80 to 100 cubic feet per barrel for roasters.

1,800 to 2,600 cubic feet per barrel for kilns.

Salt Plants:

Direct-fire pans, 9,000 cubic feet per ton.

Stream pans, 10,000 cubic feet per ton.

Single-effect vacuum pan, 15,000 cubic feet per ton.

Double-effect vacuum pan, 10,000 cubic feet per ton.

Triple-effect vacuum pan, 6,000 cubic feet per ton.

Flour Mills:

200 to 400 cubic feet per barrel.

Gas Compressors:

Horsepower required to compress 1,000 cu. ft. of gas per minute:

То	15 lbs.	50 H.P.	
То	30 lbs.	85 H.P.	
То		111 H.P.	
То		134 H.P.	
	80 lbs.	117 H.P. (2 stages)	
	100 lbs.	151 H.P. (2 stages)	
To	200 lbs.	212 H.P. (2 stages)	
Horsepower	required	to compress 1,000 cu. ft. of gas per hr.	
То	15 lbs.	1 H.P.	
	30 lbs.	1.75 H.P.	
	45 lbs.	2.25 H.P.	
То	60 lbs.	2.75 H.P.	

The specific heat of average natural gas is 0.60 B.T.U. per pound, or 0.028 B.T.U. per cubic foot at 32° F.

Properties of Hydrocarbons Found in Natural Gas and Casinghead Gas

	Methane	Ethane	Propana	Butane	Pentane	Hexane	Heptane	Octane
Formula.	CH4	C2H6	C3H8	C4H10	C5H12	C6H14	C7H16	C8H18
Molecular weight	16.03	30.05	44.07	58.08	72.10	86.12	100.13	114.15
Specific gravity of liquid.		.432=	.515=	.585=	.630=	.670=	.697=	.718=
appendix a second second	-	194° Be'	142° Be	109ª Be	92.2° Be	78.9° Be'	70.9°	65.0°
Specific gravity of gas	0.555	1.049	1.526	2.008	2.496	2.982	8.467	8.952
Boiling point at atmospheric pressure	165°C	93°O	-45°C	+1°0	36.3°C	69°O=	98.4°C	125.5°C
and the second second		=135° F				166° F		=258° F
Pressure to liquify at 60°F lbs.		475	105	35	6.5	1.8	0.5	0.15
Vapor pressure 70°F in percent of atmosphere.	100+	100+	100+	100+	55	10	2.7	0.7
Gallons per 1000 cu. ft. @ B. P. reduced to 60°F		4.13	7.17	10.72	14.35	18.22	22.05	25.86
Weight per 1000 cu. ft. @ B. P. reduced to 60°F, lbs.	42	79.7	116	152.6	189.7	226.6	263.5	800
Shrinkage in volume by 1 gal. liquid removed per 1000 cu. ft					7.0%	5.5%	4.5%	8.9%
Max. possible removable gal. per 1000 cu. ft. @					0.070	5.570	E. 070	a.s./0
70°F, gal					7.8	1.8	0.6	0.18
Heating value in B. T. U. per cu. ft	1065	1861	2685	3447	4250	5012	5780	6542
B. T. U. per lb	25360	23350	23150	22590	22400	22120	21935	21807
Ou. ft. air to burn 1 cu. ft. gas.	9.57	16.72	23.92	81.10	35.28	46.46	53.6	60.8
Carbon per cent	75.0	80.0	81.8	82.8	83.3	83.7	84.0	84.2
Explosive mixture per			0110					
cent in air, maximum Minimum.	14.5	5.0 3.0	3.5 2.1	3.0	2.5 1.3	2.2	1.9	1.6

Gasoline and Natural Gas Explosions

An explosion or a detonation is a chemical reaction which goes on with increasing velocity and is accompanied by a rise of temperature. The lowest temperature at which combustion or explosion of a mixture may take place is called the ignition temperature. This varies greatly with different kinds of gases, being with ordinary hydrocarbon gases, such as natural gas, about 650°C. The vapors of some substances such as carbon bisulphide and hydrogen sulphide are capable of ignition at much lower temperatures, even as low as 100°C. Some gases even inflame spontaneously at room temperature. These are phosphorous dihydride, boron and silicon hydride and cacodyl. Ordinarily, explosive mixtures are ignited by the presence of a flame or spark at any point in the mixture ordinarily greater than .2 of a millimeter in length. In order that the gaseous mixture explodes it is necessary that the heat generated by the local combustion be greater than the heat absorbed by the surrounding gases. This means of course that if the mixture is heated to a high temperature it will be more readily explosive though the pressure will exert very little influence. An excess of either the combustible agent or the oxidizing agent in the mixture will have the same cooling effect that is exerted by any inert gas. The result is that the limits of explosibility of various mixtures of combustible gases and air are dependent upon the heat generated by the combination and by the heat absorbed in raising the temperature of the gases. For ordinary gases the fol-lowing limits hold as to the range of combustion with combustible mixtures when air is the oxidizing agent:

Limits of Explosibility of Mixtures of Combustible Gases and Air

Gasoline vapor	1.5- 6.0%	by	volume	of	mixture	
Methane	5.5-14.5	by	volume	of	mixture	*
Ethane	2.5 - 5.0	by	volume	of	mixture	
Natural gas		by	volume	of	mixture	
Acetylene					mixture	
Artificial Illuminating gas.					mixture	
Hydrogen					mixture	mersie
Carbon Monoxide					mixture	19
Blast furnace gas					mixture	
Water gas					mixture	
Coal gas.		by	volume	of	mixture	
Ethene	4.0-22.0	by	volume	of	mixture	

The striking back of a flame in a burner is caused by the presence of an explosive mixture in the burner. While the usual rate of striking back of the flame or the propagation of an explosion is over 6000 feet per second and about seven times the rate of sound in the same medium; this rate exists only when there is no retardation of the explosive wave caused by the cooling effect of the orifice or tube through which it passes.

Testing of Capacity of Casinghead Gas Wells

To use the orifice well tester the specific gravity of the gas must be taken. This is fully described on page 350.

To test a well, close all openings but one or if the well is shut in at the casinghead, blow off the well before inserting the orifice well tester. Allow the well to blow into the atmosphere for half an hour or until there is no appreciable decrease in the volume of the gas flowing from it. Screw in the orifice well tester, which carries a twoinch thread, and allow the gas to flow into the atmosphere through the proper size of orifice.

Connect a syphon gauge to the nipple on the side of the orifice well tester, using a short piece of common three-eighths-inch rubber hose. The syphon gauge should be filled with water up to the zero mark on the scale. If the well appears to be large use the large-sized orifice. To correctly determine the proper size of orifice it is necessary to read the gauge and note the height of the water in the glass. Read both sides of the scale and add them together. In other words, measure the difference between the two water levels which is the true pressure in inches of water. By referring to tables that accompany each instrument, or as found on pages 263-5 the flow of a well for a twenty-four hour period will be found under the proper gravity and opposite the pressure.

The specific gravity bottle can be used to take the water pressure of the gas flowing through the orifice in place of the syphon gauge. In this case measure the difference between the two levels of the water.

Use as large an orifice as possible so as not to permit the gas to create a back pressure in the well. A back pressure in the well will decrease the flow of the gas.

NATURAL GAS PRODUCED IN THE UNITED STATES IN 1916.

1 2104 1 44.28	Quantity	Price, cents	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1
State	M.cu.ft.	per M.cu.ft.	Value
West Virginia.	.299,318,907	15.90	47,603,396
Pennsylvania	.129,925,150	18.74	24,344,324
Oklahoma.		9.70	11,983,774
Ohio		22.32	15,601,144
Louisiana		8.29	, 2,660,445
Kansas		15.31	4,855,389
California		17.19	5,440,277
Texas		18.89	3,143,871
New York.		29.37	2,524,115
Illinois		11.22	396.357
Arkansas		10.13	241,896
Kentucky		35.73	752,635
Indiana		29.34	503.373
Wyoming and Colorado		14.97	86.077
Montana		18.21	38,855
Dakotas and Alabama		40.75	31,573
Missouri		25.41	17,594
Tennessee		57.50	1,150
Michigan		73.04	948
Iowa		100.00	275
Totals		15.96	120,227,468

14

SPECIFIC HEAT OF GASES ENCOUNTERED IN NATURAL GAS AND "CRACKED" GAS.

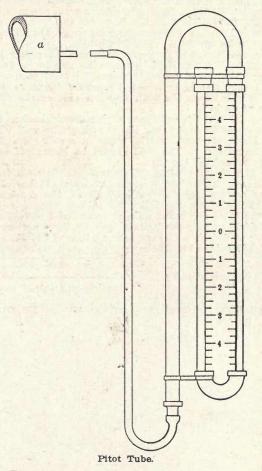
(H. L. Payne, J. A. &	Appl. Chem.)	
	B. T. U. per lb.	B. T. U per lb.
	per 1°F	per 1°F
Air	0.234	0.018
Carbon dioxide.	0.234	0.027
Carbonic oxide.		0.019
Hydrogen		0.019
"Illuminants".		0.040
Methane.		0.027
Nitrogen.		0.019
Oxygen		0.019
Aqueous vapor	0.480	

CALORIFIC VALUE OF NATURAL AND OIL GASES IN BRITISH THERMAL UNITS PER CUBIC FOOT.

		60°F	32°F Initial	Ignition	
Name	Symbol	Initial	32°F Final	Point °F	
Hydrogen	H2	326.2	345.4	1085	
Carbonic oxide		323.5	341.2	1200	
Methane.		1009.2	1065.0	1230	
Illuminants			2000.0		
Ethane.		1764.4	1861.0	1140	
Propane		2521	2657.0	1015	
Butane.		3274	3441.0		
Pentane.			4255.0		
Hexane.			5017.0	1400	
Ethylene		1588	1674.0	1010	
Propylene		2347.2	2509.0	940	
Benzene.		3807.4	4012.0		
Acetylene.		1476.7	1477.0	788	

Pitot Tube for Testing Open Flow of Gas Wells

The most accurate way of testing the flow of a gas well is by means of the Pitot tube, which is an instrument for determining the velocity of flowing gas by means of its momentum. The instrument,



as shown in figure usually consists of a small tube, with one end bent at right angles, which is inserted in the flowing gas, just inside the pipe or tubing a, at a point between one-third and one-fourth of the pipe's diameter from the outer edge of the pipe. The plane of the opening in the tube is held at right angles to the flowing gas. At a convenient distance, varying from 1 to 2 feet, an inverted siphon or U-shaped gage, usually half filled with mercury or water, is attached to the other end. If the pressure of the flow is more than 5 pounds per square inch, a pressure gage is required.

In small-sized wells with a flow of not more than 4,000,000 cubic feet per 24 hours, a 12-inch U-gage with water can be used for flows ranging from 4,000,000 to 15,000,000 feet, mercury in a 12-inch U-gage; for 15,000,000 to 35,000,000 feet, a 50-pound spring gage, and for more than 35,000,000 feet, a 100-pound spring gage should be used. The foregoing figures are based on a 6-inch hole.

For convenience, a scale graduated from the center in inches and tenths of an inch is attached between the two limbs of the U-gage. The distance above and below this center line at which the liquid in the gage stands should be added, the object being to determine the exact distance between the high and low side of the fluid in inches and tenths of an inch.

The top joint of the tubing or casing should be free from fittings for a distance of 10 feet below the mouth of the well where the test is made. The test should not be made in a collar or gate or at the mouth of any fitting. The well should be blown off at least three hours prior to making the test.

After the velocity pressure of the gas flowing from the well tubing has been determined in inches of water, inches of mercury, or pounds per square inch, as outlined above, the corresponding flow may be obtained from the following table*. The quantities of gas stated in the table are based on a pressure of 4 ounces above atmospheric, or 14.65 pounds per square inch absolute pressure, a flowin temperature of 60° F., a storage temperature of 60° F., and a specigravity of 0.60 (air = 1). If the specific gravity is other than 0.60 t

flow should be multiplied by

and the set of the set of the set of the

0.60

*Westcott, H. P.: Handbook of Natural Gas, 1915, pp. 176, 177.

	sizes	
	different	
ans of Pitot Tube.	(Figures show the rate of flow of gas or the pecific gravity from gas well tubing of different	pressures.)
Wells by Me	c gravity from	for different
Gas Gas	01 CL - pecific	per 24 hours
Determining	flow of gas	in cubic feet
Table for	the rate of	
	(Figures show	

	Fressure		Volume o Hours	Volume of Gas, in Cubic Feet per Hours, Discharged Through-	lume of Gas, in Cubic Feet per Hours, Discharged Through-	et per 24 ugh-	Pressure.	Volume	Volume of gas, in euble feet per hours, discharged through-	ume of gas, in cubic feet pe hours, discharged through-	u per 24 igh-
Inches of Water	Inches of Mercury	Pounds per Square Inch	1-Inch Tubing	2-Inch Tubing	3-Inch Tubing	4-Inch Tubing	Pounds per square inch.	1-inch tubing.	2-inch tubing.	3-inch tubing.	4-inch tubing.
0.10			11.850	47.520	106.920	190.080	2.75	321.000	1.284.000		5.136.00
20			17.130	68.544	154 224	274.176	03	340.200	1.300,800	3.061.800	5.443.200
30			20.568	89.979	185.112	320.088	3.25	354.120	1.416.480		5.665.920
07			062 26	04 080	089 116	0K45 312	2 60	SAT ASO	1 470 790		6 100 1
05.			000 544	100,170	000 000	104 101	0.0	000,100	1 501 610	0,000,000	0,000,000
ne.			20,044	0/1'001	000,002	424,104	0.10	000° #000	1,021,000		4'000'A
00.			211,62	116,448	262,008	405,792	4.00	302,880	1,571,520	ľ	6,286,080
1.			31,440	125.700	282,960	503.040	4.25	405.000	1.620,000	3.645,000	6,480,000
x			33.674	134.496	302.616	537.984	4.50	416.640	1.606.500		6.666.240
0			35 640	149 5420	2007 005	570 940.	4 75	086 864	1 713 190		6.852.480
01			003 40	140 980	225, 990	F07 190	R OD	420 090	1 750 690	2 050 980	7 086 7
1.04			010010	007 011	000,000	001,100	M.0	070,020			1,000,1
1.20			41,712	100,848	370,408	(967,392	9	476,040			0'010')
1.5			45,900	183,840	413,640	735,300	7	517,320			8,277,1
1.75	0.12		49.680	198.720	447.120	794.880	80	542.400		4.881.000	8.678.400
06	747		53,136	979, 544	478 994	850 176	0	500.640			0,114.9
i c	1.04		EQ ADD	002 000	594 BOO	020 400	OF	EDE EGO	0 000 010		0 202 0
	a or	001	002100	000,000	POC POC	001 LTO F		000 000	2,000,010	010 LOG 2	0.051.900
0.0	77.	DOL.	000,000	200,002	000,192	004'THO'T	11	008'TZ0	2,401,040		0'TOR'A
3.5	107.	1.20	70,272	R90'187.	652,448	1,124,352	12	642,600			10,281,000
4.0	£67.	141	75,120	300,480	676,080	1,201,920	13	064,680			10,634,8
4.6	.331	.162	79.704	318,810	717.336	1.275.261	14	683.880			10,942,080
5.0	.368	.18	84.000	336.000	756.000	1.344.000	15	703.080	2.812.320	6.327.720	11.249.280
6.	.441	216	92.016	348,000		1.472.956	16	080 1.64			11.537.280
7.	515	253	00 360	307 440	804 940	1 580 760	17	738 190			0.008 11
	699	886	646 901	495.098		1 700 259	10	752 000	2 015 240	R 785 640	19 063 34
50	600	100	110,011	A RO ONA			2.0	100,000			10 500 9
		200	0000011	EDU,UNE	*00'0TO'T	1,000,000 F	38	000,001	000,011,00	1,000,000	10,000,000
	001.	00.	110,000	NNZ*014			72.	803,280	3,213,120		12,502,40
.11	D.	002.	125,160	000,640	-	2,002,500	. 22	824,830	3,419.520		13.678,00
12.	88.	.432	130,128	520,512	-	2,082,048	30	910.680	3,642,720		14,570,8
	1.02	5	138.960	555.840	1.250.640	2.223.300	. 35	000.000	3.843.840	8.648.640	15.375.300
	1.52	.75	170.280	061.190	-	0.794 490	40	1.006,680	4.096.790		16.106.880
	2.03	1 00	108 680	064 984		000 JAC 200	TH I	1 DAR ROD	A 196 080	0.418 680	16 744 2
	964	1 95	010 000	040 040		0 610 900	2.5	000 100 h	1 000 000		17 010 71
		1 2 1	00000000	0101000		0001 1000	80	072 TON'T	100, 100 F		10100
	3.00	C.1	121,UH2	102,200	2,100.480	3,801,020	60	1,137,120	4.048 480		10,193,820
	3.55	1.70	259,920	1.039,680	2,339,280	4,158,720	75	1,223,400	5,217,000	000'010'11	19,574,400
	4.07	2.00	272,640	1.090,500	2.453.760	4.362.240	00	1.304.400	5.217.000		20,870,40
	4.67	2.25	294.600	1.178.400	2.051.400	4.773.600	. wit	1 824 000	N 947 080		PT 0008 19
									The second se		TAUNA TR

For pipe diameters other than those given in the preceding table, the following multipliers should be applied to the figures for 1-inch tubing given in the table.

Multipliers for Pipe Diameters Ranging from $1\frac{1}{2}$ to 12 inches.

Diameter of pipe, inches.	Multi- plier.	Diameter of pipe, inches.	Multi- plier.	Diameter of pipe, inches.	Multi- plier.
$ 1\frac{1\frac{1}{2}}{2\frac{1}{2}} \\ 4\frac{1}{4} \\ 4\frac{5}{8} $	2.25 6.25 18 21.39	5 5% 6 61/4 6%	25 31.64 36 39 43.9	8 8¼ 9 10 12	64 68 81 100 144

Flow of Gas in Pipes-Low Pressure The following formulae are intended for low pressure distribution of gas, with comparatively small differences between the initial and final pressures:

Pole's Formula

 $\mathbf{Q} = 1350 \sqrt{\frac{\mathbf{d}^5 \mathbf{h}}{\mathbf{sl}}}$ Molesworth's Formula $Q = 1000 \sqrt{\frac{d^3h}{sl}}$ $Q = 1291 \sqrt{\frac{d^{3}h}{d^{3}h}}$

Gill's Formula

$$s (1+d)$$

Where Q = quantity of gas discharged in cubic feet per hour.

d = inside diameter of pipe in inches.

h = pressure in inches of water.

s = specific gravity of gas, air being 1. 1 = length of main in yards.

Oliphant's Formula. A formula determined by F. H. Oliphant for the discharge of gas when the specific gravity is 0.60 is

$$Q = 42a \sqrt{\frac{P_1^2 - P_2^2}{L}}$$

Where Q = discharge in cubic feet per hour at atmospheric pressure.

 $P_1 =$ initial pressure in pounds per square inch (absolute).

 $P_2 =$ final pressure in pounds per square inch (absolute).

L = length of main in miles.

a = coefficient (see table below).

For gas of any other specific gravity, s, multiply the discharge by / 0.60

---, for temperature of flowing gas when observed above 60°F 1 . 197

deduct 1 per cent for each 5° and add a like amount for temperatures less than 60°F.

KANSAS CITY TESTING LABORATORY

According to Oliphant, the discharge is not strictly proportional to

----- Using a coefficient of unity for 1-inch pipe he gives

		u		
		1	1.61	
a	-	- 1-		+
u		V		1

ds

Inside
ameter
inches a
12 556
16 1160
18 1570
20 2055
24 3285
30 5830
36 9330
nside dia. $a = 863$
nside dia. $a = 1025$
nside dia. $a = 1410$
inside dia. $a = 1860$

Capacity of Pipe Lines

"(Metric Metal Works.)

Tables to Find the Cubic Feet, per Day of 24 Hours, of Gas of .6 Specific Gravity at Certain Pressure in Pipe Lines

of Various Diameter and Lengths.

Select in table A the number opposite the gauge pressures, in pounds, then from table B select the number opposite the length of line in miles. Multiply these two numbers together and result is the cubic feet that a 1-inch line will discharge for the pressures and length named in twenty-four hours. If the diameter of the pipe is other than one inch, select the number in table C which corresponds with the diameter and multiply this number by the discharge for one inch already secured. The result is the quantity in cubic feet in twenty-four hours discharged by a line whose diameter was selected.

If there are other pressures and lengths not given in the table they can be secured by interpolation. Example—Suppose it is required to find the discharge per day of twenty-four hours of a pipe line having an intake of 200-pound gauge pressure and 25 pounds at the discharge end, the length being 20 miles, and the diameter 8 inches. In table A we find opposite 200 and 25 the number 211.25, and in table B opposite 20 miles, 22.5, multiplying these two numbers the result being 47,637 cubic feet that under the above condition of pressure and length a 1-inch pipe would convey, but the required diameter is 8 inches. Under this number in table C it will be found that 198 corresponds; therefore 47,637 \times 198=9,433,126, which is the cubic feet discharged in 24 hours.

If the pressure were twenty pounds instead of twenty-five at the discharge end it would be found very closely by adding the figures opposite 15 and 25 and dividing by 2, the result would be 9,469,154.

TAB	\mathbf{LE}	A.
-----	---------------	----

ntake, Lbs.	Dis- charge, Lbs.	Re- sultant	Intake, Lbs.	Dis- charge, Lbs.	Re- sultant	Intake, Lbs.	Dis- charge, Lbs.	Re- sultan
í	1/	4.7	40	5	51.2	110	75	86.8
1	1/4 1/2	3.9	40	10	49.0	110	85	75.0
1 2 2	1/2	6.9	40	15	46.1	110	100	49.0
2	1 12	4.7	40	20	42.4	125	5	138.6
2	11/2	4.0	40	25	37.8	125	15	136.8
2334445555666	1	8.1	40	30	31.6	125	25	134.2
3	$\overline{2}$	5.8	40 -	35	22.9	125	35	130.8
4	1	10.1	50	5	61.8	125	50	124.0
4	2	8.4	50	10	60.0	125	75	107.2
. 4	$2 \\ 3 \\ 1$	6.0	50	15	57.7	125	100	79.8
5	1	11.8	50	20	54.8	125	110	63.1
5	2	10.4	50	25	51.2	135	5	148.7
5	3	8.6	50	30	46.9	135	15	147.0 144.6
5	4	6.2	50	35	41.5	135	25	144.6
6	1	13.4	50	40	34.6	135	35 50	141.4
6	3	10.6	50	45	25.0	135	50 75	135.2 120.0
6	5	6.3	60	5	72.3 70.7	135 135	100	120.0
7	1	14.9 12.5	60 60	10 15	68.8	150	5	96.3 163.8
7	35	9.0	60	20	66.3	150	15	162.3
	6	6.5	60	25	63.4	150	25	160,1
6	1	16.3	60	30	60.0	150	40	155.6
77788888	2	14.1	60	40	51.0	150	50	151.7
0	3 5 7	11.2	60	50	37.4	150	75	138.3
8	7	6.6	60	55	26.9	150	100	118.3
9	i	17.6	70	5	82.6	150	120	94.9
9	3	15.6	70	10	81.2	175	5	188.9
9	5	13.1	70	20	77.5	175	15	187.6
9	8	6.8	70	30	72.1	175	25	185.7
10	1	19.2	70	40	64.8	175	35	183.3
10	2	18.3	70	50	54.7	175	50	178.5
10	46	16.3	70	60	40.0	175	75	167.3
10	6	13.6	80 80	5 10	92.8	175	100	151.2
10	8	9.8	80	20	91.6 88.3	175	150	94.2
10 12	9 1	$7.0 \\ 21.8$	80	30	83.7	200 200	5 15	214.1 212.9
12 12	3	20.1	80	40	77.5	200	25	212.9
12	6	17.0	80	50	69.2	200	35	209.1
12	. 8	14.1	80	60	58.3	200	50	204 0
12	10	10.2	80	70	42.4	200	75	195.3
15	1	25.4	90	5	103.1	200	100	181.7
15	3	24.0	90	10	102.0	200	125	163.2
15	6	21.4	90	20	99.0	200	150	137.9
15	9	18.0	90	30	94.9	200	175	100.6
15	12	13.1	90	40	89.4	200	190	64.8
20	1	31.1 29.4	90 90	50 60	82.5 73.5	220	5 15	234.2
20	4	29.4 26.4	90 90	70	61.6	220	15 25	233.1 231.6
20 20	8 .10	24.5	90	80	44.7	220 220	25 35	231.6
20	15	18.0	100	5	113.3	220	50	229.6
20	15	11.7	100	10	112.3	220	75	225.8
25	18	36.7	100	15	111.0	220	100	204.9
25 25 25 25	3	35.7	100	20	109.5	220	125	188.8
25	6	34.0	100	25	107.8	220	150	167.3
25	10	31.2	100	35	103.6	220	175	138.3
25 25	15	26.5	100	50	94.9	220	200	94.9
25	18	22.6	100	75	71.6	230	5	244.1
30	1	42.1	100	85	56.8	230	15	243.2
30	3	41.2	100	95	33.5	230	25	241.7
30	6	39.8	110	5	123.4	230	35	239.8
30	10	37.4	110	15 25	121.4	230	50	236.2
30	15 20	33.5 28.3	110 110	25 35	118.4 114.6	230 230	75 100	227.9 216.3
30	20	40.0	110					

KANSAS CITY TESTING LABORATORY

TABLE A—Continued.

Intake, Lbs.	Dis- charge, Lbs.	Re- sultant	Intake, Lbs.	Dis- charge, Lbs.	Re- sultant	Intake, Lbs.	Dis- charge, Lbs.	Re- sultant
230	200	117.5	325 .	250	213.0	400	225	338.6
230	215	84.4	325	275	177.5	400	250	319.4
250	5	264.2	325	285	160.0	400	275	296.9
250	15	263.3	325	300	128.0	400	300	270.2
250	25	262.0	350	5	364.5	400	325	238.0
250	35	260.2	350	15	363.8	400	350	197.5
250	50	256.9	350	25	362.8	400	375 .	141.9
250	75	249.3	350	35	361.6	425	5	439.6
250	100	238.8	350	50	359.2	425	15	439.0
250	125	225.0	350	75	\$53.7	425	25	438.2
250	150	207.4	350	100	346.4	425	35	437.2
250	175	184.7	350	125	337.1	425	50	435.2
250	200	154.9	350	150	325.6	425	75	430.7
250	230	101.0	350	175	311.7	425	100	424.7
275	5	289.3	350	200	295.0	425	125	417.1
275	15	288.4	350	225	275.0	425	150	407.9
275	25	287.2	350	250	251.0	425	175	396.9
275	35	285.7	350	275	221.6	425	200	383.9
275	50	282.6	350	300	184.4	425	225	368.8
275	75	275.7	350	325	132.8	425	250	351.3
275	100	266.2	375	5	389.5	425	275	330.9
275	150	238.5	375	15	388.8	425	300	307.2
275	200	194.6	375	25	387.9	425	325	279.3
275	250	117.8	375	35	386.8	425	350	245.7
300	5	314.4	375	50	384.6	425	375	203.7
300	15	313.6	375	75	379.5	425	400	146.2
300	25	312.5	375	100	372.7	450	5	464.6
300	35	311.0	375	125	364.0	450	15	464.0
300	50	308.2	375	150	353.4	450	25	463.3
300	75	301.9	375	175	340.6	450	35	462.3
300	100	293.8	375	200	325.4	450	50	460.4
300	125	282.2	375	225	307.4	450	75	456.2
300	150	268.3	375	250	286.1	450	100	450.5
300	175	251.3	375	275	260.8	450	125	443.4 434.7
300	200	230.2	375	300	230.0	450	150 175	434.7
300	250	170.3	375	325	191.1	450		424.4 412.3
300	275	123.0	375	350	137.4	450	200 225	412.3
325	5	339.4	400	5	414.5	450	225	398.3
325	15	338.7	400	15	413.9	450 450	230	363.5
325	25	337.6	400	25	413.1		300	303.5
325	35	336.3	400	35 50	412.0	450 450	300	342.1
325 325	50 75	333.7	400		409.9	450	325 350	288.1
	100	327.9	400	75	405.1 398.8	450	300	253.2
325	100	320.0	400	100	398.8 390.2	450	400	203.2
325 325	125 150	309.8	400	125 150	390.2 380.8	450	400	150.4
		297.3	400	150	380.8	400	420	485.7
325 325	175 200	281.9 263.4	400	200	369.0	475 500	50	510.0
320	200	200.4	400	200	200.0	000	00	010.0

Miles	Multipliers	Miles	Multipliers	Miles	Multipliers
1/	2880.	19	231.2	61	129.1
78	2016.	20	225.5	62	128.1
74	1652.4	20 21	220.1	63	126.9
78	1419.7	21 22	214.9	64	120.9
72	1275.9	23	214.9	65	120.0 125.1
78	1158.6	23	205.7	66	123.1 124.1
1/8 1/4 3/8 1/2 5/8 3/4 7/8	1083.7	24	205.7	67	124.1 123.1
1 18	1008.0	20	197.6	68	123.1
	826.2	20 27	197.0	60	122.2
$1\frac{12}{1\frac{3}{4}}$	763.6	21 28	193.0	69 70 72	121.3 120.4
1%	703.0	28	190.5 187.0	70	120.4 118.7
2	638.0	30	187.0	74	117.2
21/2 23/4	607.2	30	185.9	14	117.2 115.6
3	582.7	32	178.0	76 78	114.2
31/2	539.0	33	175.6	80	112.7
3 1/2	504.0	34	172.9	82	111.2
41/2	475.5	35	170.3	84	109.9
5	450.0	36	168.0	88 .	108.7
51/2	428.9	37	165.8	88	107.5
6	411.4	38	163.6	90	106.2
61/2	395.3	39	161.3	.92	105.1
7	380.4	40	159.5	94	103.9
71/2	367.9	41	157.5	96	102.9
8	356.2	42	155.6	98	101.8
81/2	345.2	43	153.7	100	100.8
9	333.0	44	152.0	102	99.8
91/2	327.3	45	150.2	105	98.3
10	319.0	46	148.7	107	97.5
101/2	311.1	47	146.9	110	96.0
11	303.6	48	145.4	112	95.3
111/2	297.3	49	144.0	115	93.9
12	291.3	50	142.6	118	92.8
121/2	284.7	51	141.2	120	92.0
13	276.4	52	139.8	122	91.2
131/2	274.6	53	138.5 137.1	125	90.2
14	269.5	54	137.1	130	88.4
141/2	264.6	55	135.8	135	86.8
15	260.5	56	134.8	140	85.2
$15\frac{1}{2}$	255.8	57	133.5	145	83.7
16	252.0	58	132.3	150	82.3
17	244.7	59	131.2		
18	237.5	60	130.1		

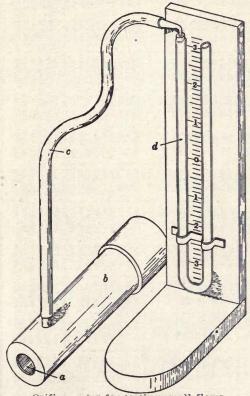
TABLE B.

TABLE C.

IADLE U.
Multipliers for diameters other than 1 inch.
$\frac{1}{4}$ inch = .0317 3 inch = 16.50 12 inch = 556
$\frac{1}{2}$ inch = .1810 4 inch = 34.10 16 inch = 1160
$\frac{3}{4}$ inch = .5012 5 inch = 60.00 18 inch = 1570
1 inch = 1.0000 5% inch = 81.00 20 inch = 2055
$1\frac{1}{2}$ inch = 2.9300 6 inch = 95.00 24 inch = 3285
2 inch = 5.9200 8 inch = 198.00 30 inch = 5830
$2\frac{1}{2}$ inch = 10.3700 10 inch = 350.00 36 inch = 9330
For wrought iron pipes greater than 12 inches in diameter the
measure is taken from outside, and for pipes of ordinary thick-
ness the corresponding inside diameters and multipliers are as follows:
Outside dia. of 15-inch pipe gives $14\frac{1}{4}$ in. inside dia. = 863
Outside dia. of 16-inch pipe gives $15\frac{1}{4}$ in. inside dia. = 1025
Outside dia. of 18-inch pipe gives $17\frac{1}{4}$ in. inside dia. = 1410
Outside dia. of 20-inch pipe gives 191/4 in. inside dia. = 1860

Measuring the Flow of Natural Gas ORIFICE METER.

An instrument known as the orifice meter, for testing small flows of natural gas, is shown in the figure. This instrument is simple in construction, consisting of a short 2-inch nipple, b, with pipe thread



Orifice meter for testing small flows of natural gas.

on one end and a thin plate disk on the other. The disk carries a 1inch orifice, a, and a hose connection, c, for taking the pressure. The meter is especially intended for testing small gas wells and "casinghead" gas from oil wells. As a rule the flow of gas from an oil well is rather small, and it is not advisable to test the flow with a Pitot tube such as is used in testing large gas wells. In using the orifice tester, it is necessary to know the specific gravity of the gas in order to obtain the flow.

Before the orifice well tester is attached to the casinghead the well should be permitted to blow into the atmosphere until the head of the gas is reduced and the flow has become normal. Then the tester is attached by simply screwing it into the end of a 3-foot length of 2-inch pipe and the pressure is read in inches of water on the siphon gage, d.

262-3, the flow of the well with values for the gas of different gravities is opposite the gage reading. The orifice in the instrument should be kept dry and uninjured; otherwise the gage reading will not be correct.

*Westcott H. P.: Handbook of Natural Gas, 1915, pp. 545-548.

Gas	
Capacities of Orifices for Testing Flows of Natural Gas From Small Gas Wells and Casinghead Gas From Oil Wells.	
and	1 7-
Wells	
Gas	
Small	
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Gas I Wells	
Natural rom Oil	a second se
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(Temperature, 60°F.; atmospheric pressure, 14.4 pounds per square inch.) ONE-INCH ORIFICE IN PLATE 3, INCH THICK.

16,750 223,700 223,700 335,620 35,620 35,620 35,620 45,100 52,100 52,100 57,400 42,500 (0,200 (1,200) (0,200) (0,200) (1,200) (1,200) (1,200) (1,200) (1,200) (1,200) (1,200) (2,000) (1,200) (2,000) 1.5 뀀 95,400 116,700 1124,700 1124,700 1152,700 1176,400 1176,400 1176,400 1176,400 1176,400 206,800 200,8000 200,80 117,320 30,400 331,400 331,400 337,920 37,920 56,780 56,780 56,780 56,400 44,000 62,300 76,300 88,200 1.4 112,000 121,200 129,400 1144,700 1144,700 1171,200 1194,200 2204,700 2214,700 2214,700 42,930 55,930 56,930 56,930 56,930 61,680 45,700 64,700 91,500 991,500 25,480 31,560 35,760 39,360 1.3 47,500 67,300 82,500 95,300 18,720 26,540 32,850 37,220 37,220 37,220 44,680 44,680 44,680 51,690 51,690 61,320 61,320 64,170 116,000 126,100 134,700 150,600 165,000 178,200 213,100 223,400 90,000 202,100 1.2 $\begin{array}{c} 19,120\\ 27,120\\ 38,550\\ 38,550\\ 38,040\\ 38,040\\ 38,040\\ 58,500\\ 56,160\\ 55,100\\ 55,100\\ 62,600\\ 62,600\\ 65,500\\$ 48,600 68,800 88,300 116,300 1116,300 1118,800 1135,900 1135,900 1135,0000 1135,000 1135,0000 1155,000000000000000000000 1.15 Capacity in Cubic Feet per 24 Hours, at Specific Gravity of-49,600 70,300 88,500 88,500 1107,600 1117,300 1177,3000 1177,300 1177,3000 1177,3000 1177,3000 1177,3000 1177,3000 1177, 1.1 50,800 72,000 88,200 101,800 110,100 124,700 114,000 1161,000 156,400 190,000 216,100 227,800 20,010 228,330 339,790 339,770 339,770 551,500 551,500 66,5500 66,5500 68,610 68,610 248,400 203,700 238,900 1.05 $\begin{array}{c} 20,520\\ 229,080\\ 386,000\\ 386,000\\ 386,000\\ 444,880\\ 444,880\\ 556,640\\ 63,240\\ 66,240\\ 66,240\\ 66,240\\ 66,240\\ 66,240\\ 66,240\\ 70,320\\ 67,200\\ 67,200\\ 67,200\\ 67,200\\ 67,200\\ 68,840\\$ $\begin{array}{c} 52,100\\ 73,800\\ 90,400\\ 1104,400\\ 1127,800\\ 127,800\\ 138,200\\ 147,600\\ 165,000 \end{array}$ 221,500 233,500 244,800 254,000 180,800 195,300 208,800 -53,400 75,600 92,600 107,000 1115,700 131,000 141,000 151,300 169,200 169,200 214,000 227,000 239,300 250,900 200,200 261,000 0.95 21,000 30,640 37,940 37,940 51,600 55,630 65,480 65,480 65,480 67,270 70,800 71,800 54,900 77,800 95,300 110,000 118,900 134,700 145,600 155,600 174,000 205,800 205,800 225,100 226,100 226,100 228,400 0.9 56,500 80,000 97,900 113,100 122,400 1138,600 1149,800 1178,900 1178,900 1178,900 1178,900 1178,900 221,700 221,700 221,700 225,400 276,000 276,000 0.85 0.8 23,000 38,600 38,600 51,790 56,490 66,350 66,350 66,350 81,140 81,140 00,100 85,100 104,300 0.75 24,500 43,000 43,000 43,000 53,490 65,090 65,090 67,680 67,680 80,300 84,000 84,000 0.7 25,410 36,040 36,040 56,550 60,720 60,720 60,720 87,150 87,150 87,150 87,150 87,150 64,000 91,500 112,000 1129,400 1171,300 1171,300 1171,300 204,600 204,600 242,100 258,900 274,600 289,500 303,600 315,700 0.65 26,400 37,510 46,440 57,880 68,140 68,140 68,140 68,140 88,580 88,580 90,720 90,720 315,800 328,400 0.0 1/2..... /2..... 1/2..... Pressure ches of Vater ches of 1/2 ercury

BULLETIN NUMBER FIFTEEN OF

1,410 2,190 2,739 3,950 3,950 3,950 3,950 3,950 3,950 3,950 3,950 5,130 5,240 5,240 1.5 1.4 Capacities of Orifices for Testing Flows of Natural Gas From Small Gas Wells and Casinghead Gas $\begin{array}{c} 3,050\\ 6,210\\ 6,210\\ 6,210\\ 6,210\\ 8,170\\ 8,170\\ 9,000\\ 9,000\\ 9,000\\ 9,000\\ 0,330\\ 10\\ 0,330\\ \end{array}$ (1,54)(2,350)(3,292)(3,291.3 1,610 2,450 2,450 2,450 2,450 2,450 2,410, 3,1805,5005,5006,4006,4006,4005,5009,5309,5309,5300,3800,3800,3800,3800,3800,3800,7000,3800,7000,3800,7000,3800,7000,3800,7000,3800,7000,3800,7000,3800,7000,3800,9000,7000,9000,9000,7000,9000,9000,7000,7000,9000,9000,701.2 3,250 6,710 6,710 8,000 8,000 9,730 9,730 9,730 9,730 10,610 1.15 1/8 INCH THICK. 1% INCH THICK. Capacity in Cubic Feet per 24 Hours, at Specific Gravity of- $\begin{array}{c} 3,320\\ 6,750\\ 6,750\\ 8,880\\ 9,450\\ 9,950\\ 0,430\\ 10,850\\ 11,230\\ 11,230\\ \end{array}$ 1.1 $\begin{array}{c} 3,400\\ 4,730\\ 6,910\\ 7,730\\ 9,020\\ 9,070\\ 9,070\\ 10,190\\ 10,100\\ 11,100\\ 11,500$ 1.05 From Oil Wells.-Continued PLATE $\begin{array}{c} 3,480\\ 6,120\\ 6,120\\ 7,080\\ 7,920\\ 9,310\\ 9,310\\ 9,310\\ 9,910\\ 11,380\\ 11,780\\$ 1,700 2,230 3,740 3,740 3,740 1,474 1,875 5,152, PLATE -THREE-EIGHTHS INCH ORIFICE IN 3,570 4,970 6,280 8,120 8,120 8,800 9,550 9,550 110,710 110,710 111,670 1,810 3,420 5,230 5,200 5,000 5,0000 5,000000000 0.95 INCH ORIFICE IN 0.9 3,770 5,290 6,640 6,640 9,570 9,370 110,750 110,750 111,320 111,320 112,340 112,340 112,340 112,340 112,340 112,340 112,340 112,340 112,750 110,750 110,750 110,750 110,750 110,750 10 6,910 6,910 6,910 6,450 6,450 6,450 6,450 6,9100 6,9100 6,9100 6,9100 6,9100 6,91000000000000000000000000000000000 0.85 ONE-HALF $\begin{array}{c} 3,800\\ 6,840\\ 6,840\\ 7,910\\ 9,600\\ 111,670\\ 111,670\\ 112,720\\ 13,170\\ 1$ 1.970 3,000 5,730 5,450 5,450 5,450 5,450 5,450 5,450 5,450 5,450 5,450 5,450 5,760 5,760 7,020 7,020 7,180 0.8 4,020 5,000 8,170 9,140 9,140 10,750 10,750 11,440 11,440 12,6400 12,6400 12,6400 12,6400 12,6400 12,6400 12,64000 0.75 4,160 5,790 8,460 9,470 9,470 9,470 9,470 9,470 111,130 110,130 110,130 110,130 110,130 110,130 110,130 110,130 110,10 0.7 4,320 6,010 8,780 9,820 9,820 9,820 11,5500 11,5500 11,5500 11,5500 11,5500 11,5500 10 0.65 4,490 6,260 9,140 9,140 9,140 111,150 112,800 112,800 113,480 114,130 114,130 114,690 114,690 115,210 2,270 3,460 5,470 5,470 5,770 5,770 5,770 5,770 8,250 8,270 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 8,700 0.6 Pressure ches of Water 12. 100 3 2 2 2 200 × 2

KANSAS CITY TESTING LABORATORY

Orifice Capacity

Diame	eter Inches	nches	Morse Drill Gage Size	Cubic Feet Per Hour		
Frac.	Decimal	- Area Square Inch		Coal Gas 0.43 sp. gr. 2" Press	Water Gas 0.62 sp. gr. 2" Press	Natural Ga 0.62 sp. gr. 4½ Oz. Pres
	0.0135	0.000143	80	1.04	0.85	1.67
	0.0145	0.000165	79	1.16	0.97	1.89
1/64	0.0156	0.00019		1.26	1.05	2.05
	0.016	0.00020	78	1.32	1.10	2.14
200	0.018	0.00025	77	1.35	1.13	2.20
	0.020 0.021	0.00031 0.00035	76 75	1.62 1.80	1.35	2.63
1.1	0.021	0.00035	74	2.16	1.52 1.80	2.96 3.51
	0.024	0.00045	73	2.29	1.80	3.70
	0.025	0.00049	72	2.46	2.05	4.00
1. 1	0.026	0.00053	71	2.70	2.25	4.38
	0.028	0.00062	70	2.79	2.33	4.54
	0.0292	0.00067	69	3.08	2.57	4.97
1.00	0.031	0.00075	68	3.23	2.70	5.26
1/32	0.031	0.00076		3.26	2.73	5.32
1.0	0.032	0.00080	67	3.42	2.85	5.56
	0.033	0.00086	66	3.53	2.94	5.73
1000	0.035	0.00096	65 64	3.69	3.08	6.00
2.12	0.036 0.037	0.00102 0.00108	63	3.86	3.23	6.30
00.2	0.037	0.00108	62	4.05 4.11	3.38 3.51	6.60 6.84
letro (p	0.039	0.00119	61	4.50	3.75	7.31
1	0.040	0.00126	60	4.95	4.12	8.04
1.1.1	0.041	0.00132	59	5.22	4.35	8.48
	0.042	0.00138	58	5.40	4.50	8.77
1.24	0.043	0.00145	57	5.67	4.71	9.2
	0.0465	0.00170	56	6.57	5.47	10.6
3/64	0.0469	0.00173		6.75	5.63	11.0
100	0.0520	0.0021	55	8.9	6.75	13.2
	0.0550	0.0023	54 53	9.0	7.50	14.6
1/16	0.0595 0.0625	0.0028 0.0031	03	10.8	9.0	17.5
1/10	0.0635	0.0032	52	11.7 11.9	9.7	19.0 19.3
	0.0670	0.0035	51	12.6	9.9 10.5	20.5
	0.070	0.0038	50	13.5	11.2	21.8
14.19	0.0730	0.0042	49	14.4	12.0	23.4
	0.076	0.0043	48	15.3	12.7	24.8
5/64	0.0781	0.0048		15.7	13.1	25.5
T. 19 4.	0.0785	0.0018	47	15.8	13.2	25.7
1.94	0.081	0.0051	46	16	13.5	26
	0.082	0.0053	45	17	14.3	28
	0.086 0.089	0.0058 0.0062	44 43	18	15	29
	0.0935	0.0069	42	19	16.5	32
3/32	0.0937	0.0069	34	20 21	17 18	33 35
-1-0	0.096	0.0072	41	22	18	30
	0.098	0.0075	40	23	20	39
	0.0995	0.0078	39	24	20.5	40
	0.1015	0.0081	38	25	21	41
	0.104	0.0085	37	26	22	43
TION	0.1065	0.0090	36	27	22.5	44
7/64	0.1093	0.0094		28	23	45
	0.110	0.0095	35	29	24	47
	0.111 0.113	0.0097 0.0100	34 33	30	25	49
313	0.115	0.0100	32	31 32	26 27	51 53

ORIFICE CAPACITY—Continued.

Diam	eter Inches		Morse	Cub	ubic Feet Per Hour		
Frac.	Decimal	- Area Square Inch	Drill Gage Size	Coal Gas 0.43 sp. gr. 2" Press	Water Gas 0.62 sp. gr. 2" Press	Natural Gas 0.62 sp. gr. 4½ Oz. Press	
	0.120	0.0113	31	33	28	55	
1/8	0.125	0.0123		36	30	58	
	0.1285	0.0130	30	39	32	62	
	0.136	0.0145 0.0155	29 28	43 44	35 37	68 72	
9/64	0.1405 0.1406	0.0155	20	45	38	74	
3/01	0.144	0.0163	27	47	39 .	75	
	0.147	0.0174	26	48	40	78	
	0.1495	0.0175	25	51	42	82	
	0.152	0.0181	24	52	43	84	
5/90	$0.154 \\ 0.156$	0.0186 0.0192	23	53 54	44 45	86 88	
5/32	0.157	0.0192	22	55	46	90	
	0.159	0.0198	21	57	47	91	
	0.161	0.0203	20	58	48	94	
	0.166	0.0216	19	60	50	97	
	0.1695	0.0226	18	62	52	101	
11/64	0.1719	0.0232	17	63	53	103	
	0.173 0.177	0.0235 0.0246	16	65 68	54 56	105 109	
1999	0.180	0.0254	15	69	58	113	
	0.182	0.0260	14	71	59	115	
4	0.185	0.0269	13	72	61	119	
3/16	0.1875	0.0276		75	62	121	
	0.189	0.0280	12	76	63	123	
	0.191	0.0286	11 10	77 79	64 66	125 129	
122	0.1935 0.196	0.0294 0.0302	9	80	67	129	
	0.199	0.0311	8	83	69	134	
	0.201	0.0317	7	84	70	136	
13/64	0.203	0.0324		86	71	138	
	0.204	0.0327	6	87	72	140	
	0.205	0.0332	5 4	89 93	74 77	144 150	
	0.209 0.213	0.0343 0.0356	* 3	95	79	154	
7/32	0.2187	0.0375		97	80	156	
.,	0.221	0.0384	2	99	82	160	
	0.228	0.0408	1	104	86	168	
15/64	0.2344	0.0442		108	90	175	
1/4 17/64	0.250 0.2656	0.0491 0.0554		119 131	99 109	193 212	
9/32	0.2812	0.0621		142	119	232	
19/64	0.2969	0.0692		153	128	250	
5/16	0.3125	0.0767		164	136	265	
21/64	0.3281	0.0845		176	146	285	
11/32	0.3437	0.0928		187	155 165	302 322	
23/64 3/8	0.3594 0.375	0.1014 0.1104		198 209	174	340	
25/64	0.3906	0.1198		221	184	360	
13/32	0.4062	0.1296		231	193	376	
27/64	0.4219	0.1398		241	201	392	
7/16	0.4375	0.1503		254	211	412	
29/64	0.4531	0.1612		264	220 230	430 448	
15/32 31/64	0.4687 0.4844	0.1725 0.1843		277 286	239	466	
1/2	0.500	0.1963		299	249	485	
33/64	0.5156	0.2088		309	257	500	
17/32	0.5312	0.2216		320	267	520	
35/64	0.5469	0.2349		331	276	539 556	
9/16	0.5625	0.2485 0.2625		340 353	285 295	576	
37/64 19/32	0.5781 0.5937	0.2625		365	303	590	
10/04	0.0301	0.2100		000	000		

Diameter Inches		Morse		Cubic Feet Per Hour			
Frac.	Decimal	- Area Square Inch	Drill Gage Size	Coal Gas 0.43 sp. gr. 2" Press	Water Gas 0.62 sp. gr. 2" Press	Natural Gas 0.62 sp. gr. 4½ Oz. Press	
39/69	0.6094	0.2917		376	313	610	
5/8	0.625	0.3068		387	323	630	
41/64	0.6406	0.3223		399	333	650	
21/32	0.6562	0.3382		410	341	665	
43/64	0.0719	0.3546		421	350	682	
11/16	0.6875	0.3712		431	369	720	
45/64	0.7031	0.3883		443	370	722	
23/32	0.7187	0.4057		454	378	737	
47/64	0.7344	0.4236		466	387	755	
3/4	0.750	0.4418		476	397	774	
49/64	0.7656	0.4604	1	488	406	792	
25/32	0.7812	0.4794		499	415	810	
51/64	0.7969	0.4988		510	424	827-	
13/16	0.8125	0.5185		520	433	845	
53/64	0.8281	0.5386		532	443	865	
27/32	0.8438	0.5591		543	453	884	
25/64	0.8594	0.5801		554	461	900	
7/8	0.875	0.6013		565	472	920	
57/64	0.8906	0.6229		576	480	938	
29/32	0.9062	0.6450		588	490	955	
59/64	0.9219	0.6675		599	500	976	
15/16	0.9375	0.6903		510	507	985	
61/64	0.9531	0.7134		620	517	1010	
31/32	0.9687	0.7371		632	526	1025	
63/64	0.9844	0.7611		644	536	1047	
1	1.0000	0.7854		655	545	1062	
			-				

ORIFICE CAPACITY—Continued.

NOTE:—The above table is based upon data obtained from gas orifices that are ordinarily used in gas appliances such as the ones used in Hale Gas Mixers.

ARTIFICIAL GAS:—The above figures are based upon 2-inch pressure; for higher pressures these figures should be increased by a percentage as shown below:

 $\begin{array}{l} 3\text{-inch} = 25 \ \% \\ 4\text{-inch} = 50 \\ 5\text{-inch} = 62.5 \\ 6\text{-inch} = 75 \\ 7\text{-inch} = 87.5 \\ 10\text{-inch} = 120 \\ 12\text{-inch} = 140 \\ 16\text{-inch} = 180 \\ 20\text{-inch} = 210 \end{array}$

NATURAL GAS:—The above figures for natural gas are based on a gas under $4\frac{1}{2}$ oz. pressure having a specific gravity of 0.62, which is the ordinary gravity of natural gas sold in cities supplied by gas from the Mid Continent, Pennsylvania and West Virginia fields. When the pressure is greater than $4\frac{1}{2}$ oz. the figures in the table should be increased as shown below:

 $5 \text{ oz.} = 10\% \\
6 \text{ oz.} = 20 \\
7 \text{ oz.} = 30 \\
8 \text{ oz.} = 39 \\
9 \text{ oz.} = 47.5 \\
10 \text{ oz.} = 60$

Outline of Methods of Analysis of Petroleum Products

- Specific Gravity and Baume' Gravity. 1.
 - With the hydrometer for fluid petroleum products. A.
 - **B**. With the picnometer.
 - C. With the Westphal balance.
 - For asphalt and semi-solid petroleum products by fluid sus-D. pension.
 - For rigid asphalt surface mixtures. E.
- 2. Color of Petroleum.
 - A. By the Saybolt Chromometer.
 - By the Lovibond Tintometer. B.
 - With Potassium Bichromate solutions. C.
 - With Iodine solutions. D.
- 3. Odor of oil. V
- Transparency. 4.
- Viscosity or Fluidity. 5.
 - A. With the Saybolt Universal Viscosimeter (A. S. T. M.), the Engler and the Redwood.
 - Ubbelohde Viscosimeter for thin petroleum products. B.
 - C. MacMichael disk friction viscosimeter.
 - D.
 - Float test for viscosity of road oils. Zero Viscosity for semi-solid petroleum products. E.
- 6. Melting Point.
 - Ring and Ball Method (A. S. T. M.). A.
 - Cube Method. **B**.
 - "General Electric" method. C.
 - Titer method for wax. D.
- 7. Cold Test.
 - Cloud test.) A.
 - B. Pour test.
 - C. Cold test.
- 8. Water and Bottom Settlings.
 - A. By centrifuge.
 - By distillation. **B**.
- Distillation tests of Petroleum. 9
 - A. Proximate distillation for water, gasoline, kerosene and residuum.
 - End point distillation (A. S. T. M. and Bureau of Mines). B.
 - C. Fractional-Gravity distillation analysis.
 - Fractional-Sample distillation. D.
- 10. Flash and Burning Points.
 - A. Illuminating oils with closed tester.
 - (Standard A. S. T. M .- "Tag" tester.)
 - B. All types of Petroleum products with the Elliott or New York closed tester.
 - C. Lubricants and asphalt with Cleveland open cup.
- 11. Pressure-heat tests.
 - Cracking test under high pressure and temperature. A.
 - Β. Vapor pressure test at high pressure.
 - C. Vapor pressure of casinghead and light gasoline.

- 12. Carbon residue.
 - A. Conradson Carbon test (A. S. T. M.).
 - **B**. Fixed carbon and ash in asphalt.
- 13. Emulsification test of lubricating oils.
- 14. Heat of combustion. 4
 - By bomb calorimeter. Α.
 - В. By calculation from gravity.
- Sulphur in petroleum products. > 15.
 - A. By bomb calorimeter.
 - By Eschka method. B.
 - C. By Parr chemical bomb.
- 16. Ultimate Analysis. V
 - Carbon and Hydrogen. A.
 - **B**. Nitrogen.
- 17. Doctor test for refined distillates.
- 18. Olefins, ethylenes or unsaturated hydrocarbons.
 - A. Babcock method (B. of M.).
 - Cylinder method (Egloff). Β.
 - C. Refining loss.
- 19. Aromatic and paraffin hydrocarbons in petroleum.
 - A. Nitrating method.
 - **B**. Distillation method.
- 20. Free acid in petroleum products.
- 21. Floc test.
- Corrosion and Gumming test of gasoline.
 Penetration or Consistency of asphalt.
 Ductility of asphalt.

- 25. Resistance of asphalt and oil to evaporation.
- 26. Determination of natural asphalt or semi-solid hydrocarbons in petroleum.
- 27. Solubility of asphalt.
 - A. In Petroleum ether-Petrolenes and Asphaltenes.
 - B. In Carbon bisulphide-total bitumen.
 - In Carbon tetrachloride-non-carbenes. C.
- 28. Resistance of asphalt to oxidation.
- 29. Paraffin wax or scale determination.
- 30. Bitumen and grading of asphalt-mineral mixtures. A. By burning.
 - Β. By extraction.
- Tensile and Cementing strength of asphaltic surface mixture. 31.
- Specific Gravity of Gas. 32.
 - A. Effusion or Viscosity method.
 - B. Edwards Gas balance.
- 33. Gasoline determination in gas (see also specific gravity). By absorption test. A.
 - Β. Freezing test.
- 34. Complete Chemical Analysis of Gas with preparation of reagents.
- 35. Heat of Combustion of Gas.
 - A. By the calorimeter.
 - Β. By oxygen consumption.
 - C. By calculation from chemical analysis.

Note.—The Kansas City Testing Laboratory will give information to anyone concerning supply houses from whom any of the following oil testing instruments may be obtained.

Index to Applications of Methods of Analysis

19.5	Product	Routine test	Occasional test	Rarely used	Can be used but not spe- cially adapted
А.	Crude Petroleum	1, 2, 3, 4, 8, 9A	5A, 9C, 14 15, 26, 29	2D, 7B, 9D, 10B, 16, 18	5D, 9B, 11, 12, 13, 19, 25
B.	Gasoline, Ben- zine and Naphtha	1, 2, 3, 4, 9B, 11C 17, 18, 22	9C, 14, 19, 20	5B, 7A, 15, 16	9D, 10
C.	Kerosene and Illuminat i ng Oils	1, 2, 3, 4, 5B, 7, 9B 10A, 15, 17, 21	10B, 14A, 20, 22	9C, 11B, 16, 18, 19	12A, 13
D.	Gas Oil, Straw Oil, Absorp- tion Oil	7, 9C, 10	5, 11A, 12A, 13, 17, 18	16, 19, 20, 21	
E.	Lubricants, Paraffin Oil	1, 2, 3, 4, 5A, 7, 10 12A, 13 15, 20	14, 17, 18	16, 19, 21	9, 11, 22
F.	Fuel Oil, Diesel Engine Oil		5, 11, 26 27A, 29	2D, 3, 9, 12, 16, 18, 19	13
G.	Road Oil, Flux Oil	1AB, 3, 5AD, 8, 10, 12, 25, 26, 27	7B, 14, 15, 29	2D, 11, 16	13, 28, 5A
H	Asphalt and Pitch	1D, 5E, 6, 8B, 12, 23, 24, 25, 27	10, 15, 28, 29	2D, 3, 14 16	5A
I.	Wax	1D, 2, 3, 4 6D	25	11A, 12A, 14, 15, 16, 17, 18, 19, 20	7A, 10
J.	Grease	1, 2, 3, 4, 5CDE, 8, 12B, 27	25	16	6, 7, 10
K	Asphalt Surface Mix				
L.	Gas	32, 33, 34, 35	16		

Note:-See special specifications for other tests of Petroleum Products.

1. Specific Gravity and Baume' Gravity.

A. With the hydrometer.

Specific gravity is the relation by weight of the same volume of oil and of water. Unless some other temperature is specifically mentioned the gravity refers to 60°F. Specific gravity is determined by means of the hydrometer, the Westphal balance, the picnometer and by displacement methods. The absolute specific gravity scale is not commonly used in the oil industry. Instead, the Baume' gravity scale, an entirely arbitrary standard, is used. Two Baume' gravity scales are in use in the oil industry; one is that adopted by the U. S. Bureau of Standards and its relation to specific gravity is indicated by the following formula:

140

Specific Gravity = $\frac{130 + \text{Baume'}^{\circ}}{130 + \text{Baume'}^{\circ}}$ for liquids lighter than

Another scale possibly more commonly used is that of instruments made by the Tagliabue Mfg. Co., which is based upon the following relation to specific gravity: 141.5

Specific Gravity = $\frac{1}{131.5 + \text{Baume'}^{\circ}}$ for liquids lighter than

The difference between the two readings varies from nothing with very heavy oils to as much as 0.5° Be' for ordinary gasoline. When the oil is heavier than water a different formula is used for calculating the Baume' gravity, the following being in general use: 145

Degrees Baume' = $145 - \frac{145}{\text{Specific Gravity}}$ for liquids heavier

Oils heavier than water are not commonly encountered. The method of using the hydrometer is the same in all cases whether its reading is in terms of the U. S. Bureau of Standards Baume' scale, the Tagliabue Baume' scale, Baume' scale for liquids heavier than water, or for direct specific gravity. The ideal instrument for all purposes is of course that reading directly in specific gravity. By the use of tables these readings can be converted into the Baume' reading desired and without any misunderstanding as to which scale is intended.

The correct method of reading the hydrometer is illustrated in Figs. 1 and 2, page 275. The sample of oil is placed in a clear jar or cylinder and the hydrometer carefully immersed in it to a point slightly below that to which it naturally sinks and is then allowed to float freely. The reading should not be taken until the oil and the hydrometer are free from air bubbles and are at rest.

In taking the reading the eye should be placed slightly below the plane of the surface of the oil (Fig. 1) and then raised slowly until this surface, seen as an ellipse, becomes a straight line (Fig. 2). The point at which this line cuts the hydrometer scale should be taken as the reading of the instrument (Fig. 2).

In case the oil is not sufficiently clear to allow the reading to be made as above described, it will be necessary to read from above the oil surface and to estimate as accurately as possible the point to which the oil rises on the hydrometer stem. It should be remembered, however, that the instrument is calibrated to give correct indications when read at the principal surface of the liquid. It will be necessary, therefore, to correct the reading at the upper meniscus by an amount equal to the height to which the oil creeps up on the stem of the hydrometer. The amount of this correction may be determined with sufficient accuracy for most purposes by taking a few readings on the upper and the lower meniscus in a clear oil and noting the differences.

A specific gravity hydrometer will read too low and a Baume' hydrometer too high when read at the upper edge of the meniscus. The correction for meniscus height should therefore be added to a specific gravity reading and subtracted from a Baume' reading.

The magnitude of the correction will obviously depend upon the length and value of the subdivisions of the hydrometer scale and must be determined in each case for the particular hydrometer in question.

Specific gravity and Baume' gravity readings of oil are conveniently taken at room temperature and these readings must be converted to the gravity at 60° F. As a general rule it may be said that petroleum oil expands with heat so that 0.0004 must be added as a correction to the specific gravity readings for each degree Fahr. that the oil is above 60° F or must be subtracted for each degree Fahr. below 60° F. On the Baume' scale .1°Be' may be subtracted for each degree Fahr. above 60° F or added for each degree Fahr. below 60° F. For exact temperature corrections for specific gravity, see pages 334 to 418. For exact temperature corrections for Baume' gravity, see pages 376-383. For conversions of Baume' to and from specific gravity, see pages 370-375.

1B. Specific Gravity with the picnometer.

Various types of picnometers may be used for this purpose, each of which has special advantages. Some are plain bottles with capillary openings in a well made ground glass stopper; others have graduated tubes in the stoppers, vacuum walls and inserted thermometers. The Sprengel picnometer is particularly adapted to the handling of very viscous oils as it prevents the including of air bubbles in the instrument. With any of the various types the perfectly dry and clean picnometer is weighed at 60° F to the nearest 0.0001 gram. It is filled with distilled water at 60° F and weighed. It is then dried completely and filled with the oil to be tested at 60° F. The net weight of the oil divided by the net weight of the distilled water gives the specific gravity of the oil. For conversion into degrees Baume' the formulae given on page 272 or the tables given on pages 370 to 375 are used.

1C. Specific Gravity with the Westphal balance.

This is a very convenient instrument where a great variety of petroleum products are to be tested as it covers any range of specific gravity and can be used for practically any type of material. Its character is shown by the figures on page 273. The oil is put into the jar and the weights or riders are adjusted on the beam until the pointer is in exact poise. The readings are in specific gravity based on a water temperature of 60° F at which temperature the instrument is standardized. The specific gravity may be converted to Baume' scale with the tables.

1D. Specific Gravity for semi-solid petroleum materials.

A convenient method of taking the specific gravity of asphaltic cement and similar semi-solid petroleum materials is the following (see upper figure on page 277). Roll up a ball of the asphalt about 1 cm. in diameter, being careful that no water is included. Place this in a cylinder of cold distilled water from which the air has been removed by previous boiling. If the ball of asphalt floats, denatured alcohol is added until it shows no tendency to go either up or down when placed in the middle of the cylinder. The specific gravity of the liquid is then taken with the Westphal balance or with the hydrometer. If the ball of asphalt sinks a saturated solution of sodium chloride or common salt is added until the asphalt when placed in the center of the cylinder shows no tendency to go either up or down. The specific gravity is taken with a hydrometer for liquids heavier than water or with the Westphal balance. It is necessary in performing this test that the bubbles of air which tend to adhere to the surface of the asphalt be occasionally removed and that the solution be thoroughly mixed. The usual temperature required for the gravity of this material is 77° F or 25° C.

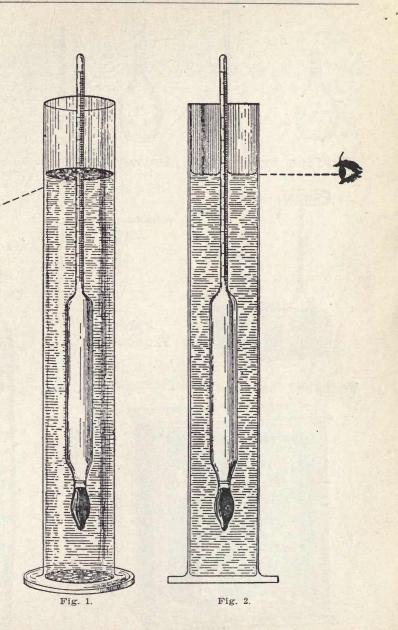
1E. Specific Gravity of solid oil materials.

A fragment of bituminous material is suspended by means of a silk thread from a hook of one pan support of the balance and about $\frac{1}{2}$ inch above the pan and weighed. This weight is "a." It is then immersed in water at 25°C and suspended, the water container not being allowed to touch the balance and is weighed again. This weight

a (see lower figure on page 277).

The sample of asphaltic surface mixture for this test should be cut out of the street after the pavement has been rolled and cooled. This test is a very good measure of the all around quality of the work. The sample is weighed in the air and in water, the weight in air divided by the loss of weight in water gives the specific gravity. This times 62.4 gives the weight per cubic foot and times 93.6 gives the weight per square yard of 2-inch surface.

KANSAS CITY TESTING LABORATORY



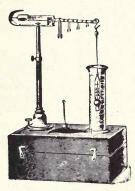
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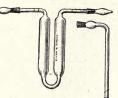




Various types of

picnometers





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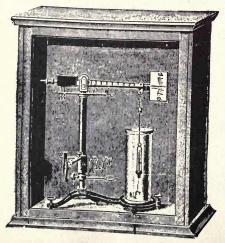
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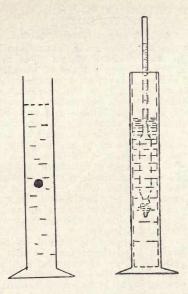
10 15

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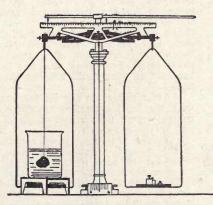
Sprengel Picnometer

Westphal balance





Specific Gravity of Asphaltic Cement.



Specific Gravity by Displacement.

2A. METHOD FOR DETERMINING THE COLOR OF REFINED PETROLEUM.

(Saybolt Universal Chromometer.)

The apparatus consists of two color comparison tubes, one being arranged for insertion of a standard yellow glass in the bottom, the other being graduated for different lengths of oil column (see figure).

Two like-colored yellow glass discs are supplied with each Chromometer. By the use of one singly or both together, color shades can be definitely determined between below Zero to + 25—Zero being Standard White and + 21 Water White—and as indicated by the accompanying table of inches corresponding to color shades.

The two glasses shall be used to determine color shades up to and including + 15, and only one glass from + 16 to + 25.

An excess of oil above that necessary to equal the working standard in color should be filled into the graduated tube so that in drawing off the excess, the eye can follow the color of oil under examination from dark to lighter, thereby making it easier to detect the point at which the oil and standard coincide.

The apparatus should be set at a window having a one-light sash so that a good light is reflected from the mirror, but not in the direct rays of the sun, and care should be taken that no colored light is reflected toward the instrument from surrounding buildings, tanks or other objects.

To clean the Chromometer before making a new test, simply allow some of the oil to be tested to run through the graduated tube. Even this need not be done between tests of similar oils if the previous oil is well drained through the pet cock, and the tube well filled with the next oil, because the influence of a drop or two of the previous oil remaining can not be seen against the half or nearly full tube of the next oil to be tested.

After using, do not let the instrument stand with the light reflecting up the tubes but move the reflecting mirror out of place, or better yet, put on the cover.

When not in use, always put the color glasses in the pockets prepared for them which will be found on the back of the upright.

For the purpose of most easily determining color shades, the column of oil when nearing the point of coincidence with the standard glass discs, shall be lowered shade by shade by use of the pet cock, until a point is reached where it

is questionable as to which is the lighter or darker shade.

Then lower the column of oil one shade more and if the oil column now shows without doubt whiter than the standard glass disc,

Saybolt Chromometer. the colorating of the oil shall be one shade above this last whiter point, or in other words, at the question point, where it was impossible to detect any difference between the oil and the glass disc.

TABLE	OF COLOR SHA	DES.	
	Inches of Oil	Color	
	in Tube	Shades	
	20	25	
Use One Disc.	18	24	
	16	23	
	14	22	
	12	21 = water	white
	10-6/8	20	
	9-4/8	19	
	8-2/8	18	
	7-2/8	17	
	6-2/8	16	
Use Two Discs.	10-4/8	+15	
	9-6/8	+14	
	9-0/8	+13	
	8-2/8	+12	
	7-6/8	+11	
	7-2/8	+10	
	6-6/8	+ 9	
	6-4/8	+ 8 + 76 + 55 + 44 + 33 + 22 + 1	
	6-2/8	+7	
	6-0/8	+6	
	5-6/8	+5	
	5-4/8	+ 4	
	5-2/8	+3	
	5-0/8	+2	
	4-6/8	+1	
	4-4/8	0 = Standa	ard
	4-2/8	-1 white	
	4-0/8	$\frac{-2}{-3}$	
	3-6/8	- 3	
	3-5/8	- 4	
	3-4/8	- 5	
	3-3/8	- 6	
	3-2/8	- 7	
	3-1/8	- 8	
	3-0/8	- 9	
It is evident that no oil		red with one d	isc unle

It is evident that no oils are to be compared with one disc unless they positively show whiter at 10-4/8 inches with two discs.

Moreover, a full tube (20 inches) of white oil that shows whiter than one (1) disc must rate +25 and up (better than +25).

2-B. COLOR BY LOVIBOND TINTOMETER. The Lovibond color units and divisions are shown below, together with the color, series and number of each glass. These slides are used for determining the color of the refined products—gasoline, naphtha and kerosene.

Lovibond color unites	with sp	ecifications for	the slides:
Slide Color		Series	Number
Water WhiteYellow		510	2.3
Red		200	1.6
1 to 12.0Amber		500	0.1 to 12.0

If the oil is darker than the water white glass, slides are added to the slot containing the standard water white until the color of the oil is matched. When the .2 slide is added in this manner, the color is reported as W.W. -0.2, the minus sign indicating that the oil is darker than the standard water white. If the color of the oil is lighter than that of the water white glass, additional slides are placed in the slots in front of the oil and should the color be matched in this manner with, say the .5 slide and the .2 slide, the color is reported W.W. +0.70.

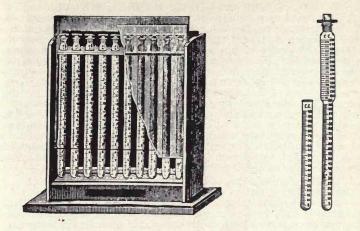
The color equivalent of water white, the standard color for gasoline and naphtha, has been defined as the equivalent of a column 404.6 mm. long of a 0.00027% acidulated solution of potassium chromate. A potassium bichromate solution, however, duplicates the tint of refined petroleum products more closely than the lower oxide. In standardizing the Stammer and Hellige colorimeters, L. Ubbelohde used a solution of 0.06 gram of potassium bichromate in one liter of water as the standard color.

2-C. COLOR WITH POTASSIUM BICHROMATE SOLUTIONS.

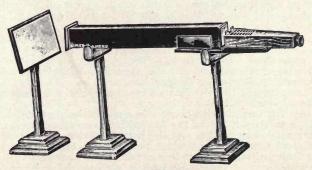
Comparison between Saybolt and Lovibond colorimeter values with equivalent potassium bichromate solutions.

In the absence of an instrument, standard acidulated solutions may be prepared to correspond with the solutions indicated in the following table. Each of these solutions when placed in four-ounce sample bottles and marked with the equivalent Saybolt and Lovibond values may be used to match samples.

es may be used to ma	tch samples.	
Potassium	Lovibond	Saybolt
Bichromate	Colorimeter	Colorimeter
Mgm. per liter of	18 cells with	Colorimeter
1% H ₂ SO ₄ Solution.	W.W. Slide	
2.0	W.W. 1 0.5	1 05 (1:.1.1)
2.9		1 25 (light)
	W.W.	1 24 (light)
3.8	W.W 0.3	1 23
4.7	W.W 0.5	1 22
5.6	W.W. — 0.8	1 21
6.5	W.W. — 1.3	1 20
7.5	W.W. — 1.5	1 19
8.5	W.W 2.0	1 18
9.5	W.W 2.2	1 17 (light)
10.5	W.W 2.8	1 16 (light)
11.5	W.W 3.0	1 16 (light)
12.5	W.W 3.8	1 15
13.5	W.W 4.5	1 14
14.5	W.W 5.2	1 13
15.5	W.W 5.7	1 13
16.5	W.W 6.0	1 13 13 11 12
17.5	W.W. = 6.5	1 12 11
18.6	W.W. = 6.9	
19.7	W.W. = 0.9 W.W. = 7.4	1 10
		$ \begin{array}{ccc} 1 & 9 \\ 1 & 8 \\ 1 & 7 \end{array} $
20.8	W.W 9.0	1 8
21.9	W.W 9.4	
23.0	W.W. = 10.0	1 6
24.1	W.W 10.2	1 5
25.3	W.W 11.0	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$
26.6	W.W 11.0	$ \begin{array}{ccc} 1 & 3 \\ 1 & 1 (light) \end{array} $
28.0	W.W 11.2	1 1 (light)
		(0)



Color comparison tubes for the determination of the color of petroleum products by the iodine method.



Lovibond Tintometer.

2-D. COLOR OF OIL BY IODINE METHOD.

This method may be applied to all dark colored petroleum products. In determining the color by the iodine method a solution is made containing in 1 liter of very pure distilled water, 10 grams of iodine and 20 grams of potassium iodide. This is kept in a glass stoppered bottle. The apparatus necessary is that indicated on page 281 which may be a set of carbon color tubes or two tubes such as are required in the determination of manganese in steel. For crude oil, road oil, fuel oil and other black oils a dilution of 1-1000 in colorless benzol is made by diluting 1 cc. to 10 cc. of benzol and then 1 cc. of this to 100 cc. with benzol. This is thoroughly mixed in one of the glass stoppered color tubes. 1 cc. of the standard iodine solution is put into the large color tubes. 1 cc. of the standard iodine solution is put into the large color tubes which holds 250 cc. It is diluted with distilled water until its color matches that of the oil under test. The color is calculated, as follows: I = milligrams of iodine in 100 cc. ofwater in the tube containing the diluted iodine.

d = The number of cc. of benzol to 1 cc. of oil.

Color = I (d+1).

For gas oil, lubricating oils and yellow oils, a dilution of 1-100 with benzol is sufficient. For gasoline, naphtha, kerosene and illuminating oils there is no dilution with benzol, the comparison being made directly.

The terms applied to the color of crude oil are black, brownish black, blackish brown, brown, reddish brown, green, greenish brown, brownish green and bluish green. The kerosene is spoken of as being water white, superfine white, prime white, standard white, prime light straw, light straw and straw. Other colors are designated by yellow, dark yellow, reddish yellow, brownish yellow, yellowish brown, brown, red, blood red and yellowish red.

3. ODOR OF OIL.

The odor of oil may be spoken of as sweet, ethereal, aromatic, tarry, fatty, creosotic, acid, sour, sulphurous, sulphuretted hydrogen, pyridine and pungent.

4. TRANSPARENCY OF OILS.

Transparency may be expressed by the thickness of oil in centimeters through which the filament of a 50 watt Mazda electric lamp is visible. It may be also noted whether the oil is fluorescent and the character of the fluorescence, whether bluish, greenish or yellowish by reflected light; also whether any turbidity is of a smoky, granular or flocculent character.

5-A. VISCOSITY OF LIQUID PETROLEUM PRODUCTS. (SAYBOLT UNIVERSAL.)

The apparatus is shown on page 285.

This is the tentative test for the viscosity of lubricants adopted by the American Society for Testing Materials:

1. Viscosity shall be determined by means of the Saybolt standard universal viscosimeter.

2. (a) The Saybolt viscosimeter is made entirely of metal. The standard oil tube is fitted at the top with an overflow cup and the tube is surrounded by a bath. At the bottom of the standard oil tube is a small outlet tube through which the oil to be tested flows into a receiving flask, whose capacity to a mark on its neck is 60 (+0.15) cc. The lower end of the outlet tube is enclosed by a larger tube, which when stoppered by a cork acts as a closed air chamber and

prevents the flow of oil through the outlet tube until the cork is removed and the test started. A looped string is attached to the lower end of the cork as an aid to its rapid removal. The bath is provided with two stirring paddles and operated by two turn-table handles. The temperatures in the standard oil tube and in the bath are shown by thermometers. The bath may be heated by a gas ring burner, steam U-tube, or electric heater. The standard oil tube is cleaned by means of a tube cleaning plunger, and all oil entering the standard oil tube shall be strained through a 30-mesh brass wire strainer. A stop watch is used for taking the time of flow of the oil and a pipette, fitted with a rubber suction bulb, is used for draining the overflow cup of the standard oil tube.

(b) The standard oil tube should be standardized by the United States Bureau of Standards, Washington, and shall conform to the following dimensions: Minimum Normal Maximum

bliowing almensions:	Minimum,	Normal,	Maximum,	
Dimensions.	CM.	CM.	CM.	
Inside diameter of outlet tube	0.1750	0.1765	0.1780	
Length of outlet tube	1.215	1.225	1.235	
Height of overflow rim above bot-				
tom of outlet tube	. 12.40	12.50	12.60	
Diameter of container of standard				
oil tube		2.975	2.995	
Outer diameter of outlet tube at				
lower end	. 0.28	0.30	0.32	

3. Viscosity shall be determined at $100^{\circ}F$ ($37^{\circ}.8$ C), $130^{\circ}F$ ($54^{\circ}.4C$), or $210^{\circ}F$ ($98^{\circ}.9C$). The bath shall be held constant within $0^{\circ}.25$ F ($0.14^{\circ}C$) at such a temperature as will maintain the desired temperature in the standard oil tube. For viscosity determinations at 100 and $130^{\circ}F$, oil or water may be used as the bath liquid. For viscosity determinations at $210^{\circ}F$, oil shall be used as the bath liquid. The oil for the bath liquid should be a pale engine oil of at least $350^{\circ}F$ flash point (open cup). Viscosity determinations shall be made in a room free from draughts, and from rapid changes in temperature. All oil introduced into the standard oil tube, either for cleaning or for test, shall first be passed through the strainer. To make the test, heat the oil to the necessary temperature and

To make the test, heat the oil to the necessary temperature and clean out the standard oil tube with the plunger, using some of the oil to be tested. Place the cork stopper into the lower end of the air chamber at the bottom of the standard oil tube. The stopper should be sufficiently inserted to prevent the escape of air, but should not touch the small outlet tube of the standard oil tube. Heat the oil to be tested, outside the viscosimeter, to slightly below the temperature at which the viscosity is to be determined and pour it into the standard oil tube until it ceases to overflow into the overflow cup.

By means of the oil tube thermometer keep the oil in the standard oil tube well stirred and also stir well the oil in the bath. It is extremely important that the temperature of the oil in the oil bath be maintained constant during the entire time consumed in making the test. When the temperature of the oil in the bath and in the standard oil tube are constant and the oil in the standard oil tube is at the desired temperature, withdraw the oil tube thermometer; quickly remove the surplus oil from the overflow cup by means of a pipette so that the level of the oil in the ode-the dot the level of the oil in the tube proper; place the 60-cc. flask in position so that the oil from the outlet tube will flow into the flask without making bubbles; snap the cork from its position, and at the same instant start the stop watch. Stir the liquid in the bath during the run and carefully maintain it at the previously determined proper temperature. Stop the watch when the bottom of the meniscus of the oil reaches the mark on the neck of the receiving flask.

The time in seconds for the delivery of 60-cc. of oil is the Saybolt viscosity of the oil at the temperature at which the test was made. The approximate factors for conversion of readings of the Saybolt Universal to other instruments are as follows:

· °F	°F.	°F	°F
70	100	212	338
50	.55	.60	.65
50	1.00		
		.46	.72
	.030	.028	.027
25	.28	.51	
	.47	.51	.94
13	.13		
83	.85	.88	.90
1.25	1.04	2.00	
1.90	1.85	1.68	1.30
	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

These values are not exact as they vary greatly with the actual viscosity readings. For exact conversion to Engler and Redwood values, see page 287.

 70° F may be used for light oils, gas oils, "straw" oils, engine oils, dynamo oils, auto oils, cottonseed oils and the like.

100°F may be used for Engine oils, machine oils and occasionally cylinder oils.

212°F may be used for cylinder oils, road oil, other heavy oils and asphaltic fluxes.

and asphalte fluxes. 338°F may be used for asphalt, fluxes, paraffin wax and residues. Other viscosimeters in use are the Engler, Tagliabue, Scott, Redwood, Penn. Ry. pipet, McMichael, Lamansky-Nobel, Ostwald, Martens, Stormer, Ubbelohde, Lepenau, Kuenkler, Albrecht, Arvine, Barbey, Cockrell, Doolittle, Gibbs, Mason, Napier, Nasmyth, Phillips, Reischauer, Magruder (see page 286). The Engler viscosimeter is used most extensively in Germany end its dimensions are as follows:

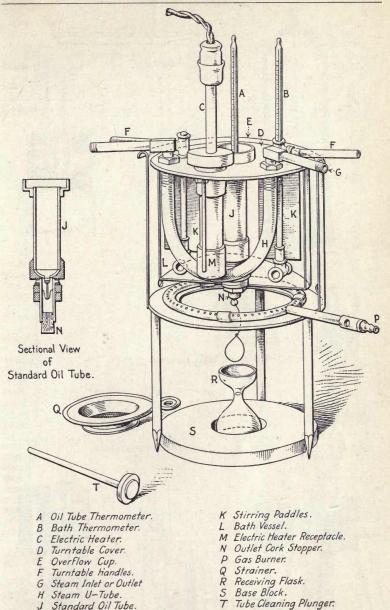
The quotient of the time of outflow of 200 cc. of oil divided by the time of outflow of 200 cc. of water is taken as a measure of the viscosity or is the so-called Engler degree.*

The Redwood viscosimeter: is used extensively in England and its value can be calculated from the Engler or the Saybolt in the tables on pages 288-9.

*Holde, Examination of Hydrocarbon Oils. ‡Redwood, Treatise on Petroleum.

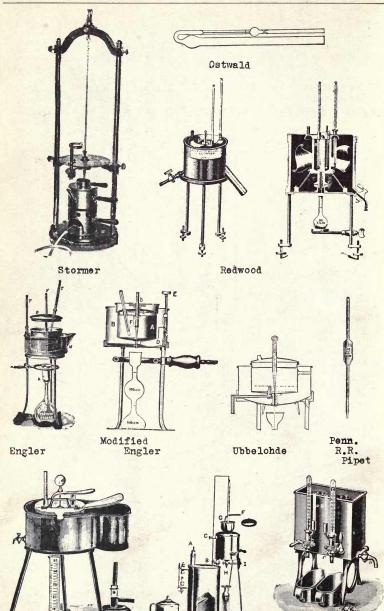
KANSAS CITY TESTING LABORATORY

285



Sectional View of Saybolt Standard Universal Viscosimeter

BULLETIN NUMBER FIFTEEN OF



Scott

Tagliabue

Lepenau

Factors to Reduce Saybolt Times to Engler Numbers or to Redwood Times

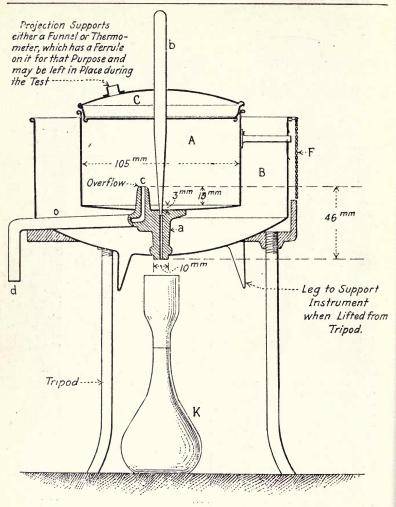
Saybolt Time, Seconds 28 30 32 34 36 38 40 42 44 46 48 50 55 60 65 70 75 80 85 90 95 100 110 120 130 140 160 180 200	Factor to Reduce Saybolt Time to Engler Number 0.0357 0.0352 0.0346 0.0342 0.0337 0.0334 0.0330 0.0320 0.0327 0.0323 0.0320 0.0317 0.0314 0.0308 0.0302 0.0297 0.0293 0.0293 0.0289 0.0284 0.0284 0.0284 0.0282 0.0280 0.0280 0.0297 0.0280 0.0271 0.0272 0.0271 0.0269 0.0269 0.0267	Factor to Reduce Saybolt Time to Redwood Time 0.95 0.95 0.94 0.94 0.94 0.93 0.92 0.92 0.92 0.91 0.91 0.90 0.89 0.89 0.88 0.87 0.86 0.86 0.85 0.85 0.85 0.85 0.85 0.85 0.85 0.85 0.84 0.84 0.84 0.84 0.84
200 1800	0.0267	0.84
2000		

Factors to Reduce Redwood Times to Saybolt Times or to Engler Numbers

Redwood Time 26 28 30 32 34 36 38 40 42 44 46 48 50 55 60 65	$\begin{array}{c} {\rm Factors \ to \ Reduce} \\ {\rm Redwood \ Time \ to} \\ {\rm Saybolt \ Time} \\ 1.05 \\ 1.05 \\ 1.06 \\ 1.06 \\ 1.07 \\ 1.07 \\ 1.08 \\ 1.09 \\ 1.10 \\ 1.10 \\ 1.10 \\ 1.11 \\ 1.12 \\ 1.13 \\ 1.14 \\ 1.15 \end{array}$	$\begin{array}{c} \mbox{Factors to Reduce} \\ \mbox{Redwood Time to} \\ \mbox{Redwood Time to} \\ \mbox{Underse} \\ 0.0377 \\ 0.0372 \\ 0.0368 \\ 0.0364 \\ 0.0364 \\ 0.0361 \\ 0.0355 \\ 0.0355 \\ 0.0355 \\ 0.0351 \\ 0.0349 \\ 0.0347 \\ 0.0344 \\ 0.0340 \\ 0.0337 \end{array}$
75 80 85 90	$1.17 \\ 1.18 \\ 1.18 \\ 1.18 \\ 1.18$	$\begin{array}{c} 0.0331 \\ 0.0330 \\ 0.0329 \\ 0.0328 \end{array}$
95 100 110 120 130 140 160 180	$1.19 \\ 1.19 \\ 1.19 \\ 1.20 \\ $	$\begin{array}{c} 0.0327\\ 0.0326\\ 0.0325\\ 0.0324\\ 0.0322\\ 0.0321\\ 0.0321\\ 0.0321\\ 0.0320\\ \end{array}$
1500	1.20	0.0320

Factors to Reduce Engler Numbers to Saybolt or to Redwood Times

Engler Number 1.00 1.05 1.10 1.15 1.20 1.25 1.30 1.35 1.40 1.45 1.50 1.60 1.70 1.80 1.90 2.00 2.10 2.20 2.30 2.40 2.50 2.60 2.70 2.80 2.90 3.00 3.50 4.00 4.50 5.00 6.00	Factors to Reduce Engler Number to Saybolt Time 28.1 28.4 29.5 29.8 30.1 30.4 30.8 31.1 31.5 32.0 32.5 33.0 33.5 33.9 34.2 34.5 34.8 35.5 35.5 35.7 35.9 36.1 36.2 36.7 37.0 37.3 37.4 37.5	Factors to Reduce Engler Number to Redwood Time 26.7 27.0 27.2 27.4 27.4 27.6 27.8 28.0 28.2 28.3 28.5 28.6 28.6 28.6 29.0 29.2 29.4 29.6 29.7 29.9 30.0 30.1 30.2 30.3 30.3 30.4 30.4 30.5 30.7 30.9 31.1 31.2 31.3
6.00 50.00	37.5 37.5	31.3



- A. Brass Oil Container.
- B. Bath.
- C. Cover of Oil Container.
- a. Capillary.
- b. Ivory or Wooden Skewer.
- c-o-d. Overflow Channel.
- F Plumb-bob for Leveling Instrument.
- IK. Flask holding 100 cu.cm at 20 deg. Centigrade.

The Ubbelohde Viscosimeter.

5-B. METHOD FOR DETERMINING THE VISCOSITY OF KEROSENE AND GASOLINE.

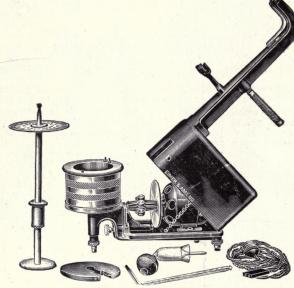
The apparatus used for this test is essentially that described on pages 55, 56 and 57 of Holde's "Examination of Hydrocarbon Oils." A diagram of the apparatus is shown on page 290. The instrument is known as the Ubbelohde viscosimeter.

Mammal

The dimensions are as follows:

NOLU	41
Instrum	ent
0.125	centimeters
0.125	66
1.0	"
3.0	66
10.5	66
4.6	66
3.992	66
200	seconds
132	cubic centimeters
	Instrum 0.125 0.125 1.0 3.0 10.5 4.6 3.992 200

The apparatus is placed in a horizontal position by means of the plummet F, the outflow tube is examined by looking through from the top with a sheet of white paper underneath to determine if there are any obstructions or dirt. If dirty, the outflow tube is cleaned by drawing a silk thread back and forth through it. Water or cracked ice, depending upon the temperature desired, is placed in the outer vessel B, the plug is put in place and an excess of kerosene or gasoline introduced into A. The excess runs out of the overflow pipe C. The plug b is loosened sufficiently to allow just a drop of liquid to pass out to the jet. When the proper temperature has been maintained for 15 minutes the plug is withdrawn and the time required to fill the 100 cc. flask is determined with the stop watch. This time divided by the time required for water gives the viscosity. For example, if the time of outflow of kerosene is 320 seconds and the water is 200 seconds, the viscosity is 1.6.



The MacMichael Viscosimeter

5-C. VISCOSITY WITH THE MacMICHAEL VISCOSIMETER.

In the MacMichael Viscosimeter a disk is suspended in a cup of fluid. The force exerted by the rotation of the fluid on the plunger is measured. This force is equal and opposite to that applied to the cup. Viscosities of oils are quickly and easily obtained at normal temperatures, also at very high and very low temperatures.

The disk is suspended in the cup of fluid by a torsion wire about ten inches long running down through the stem of the plunger and fastened near the bottom. The head of the torsion wire is triangular and is held between two grooved pins at the top of the standard. The cup and plunger may be removed and replaced without manipulating any catches or fastenings. All surfaces are smooth and rounded and may be easily cleaned.

The cup is oil jacketed, being formed of two pieces of heavy spun brass. Within the oil jacket is immersed an electric heating coil. This coil draws current from the same line as the motor, only one connection being necessary. The fluid to be tested is heated in place, no other heating device being required. The operation is very rapid. Stirring is effected by a slight vertical movement of the plunger. For low temperature work, the fluid and the adjacent parts are chilled in an ice bath or brine solution.

A bent thermometer inserted through an opening in the cover indicates the temperature, the bulb being immersed in the liquid. The temperature during test may be controlled to within a small fraction of one degree.

The graduated dial at the top of the plunger is secured to the stem by a friction disk, permitting the adjustment of the zero mark to its proper location. The fine adjustment is effected by means of the steel wire pointer at the head of the standard. The dash pot on the stem of the plunger is frictionless and automatic in action, requiring no attention from the operator. Its function is to check incipient vibrations and to permit quicker readings by damping the action.

The speed control is of the phonograph type and gives excellent results. The motor is furnished for 110 or 220 volts either A. C. or D. C. and is adapted for ordinary lighting circuits. Variations in voltage do not affect the accuracy of the determinations.

In operating, the cup is filled to the mark on the side with the oil or asphalt to be tested. This requires about 100 cc. The temperature is raised or lowered by means of the heating coils. The deflection noted on the dial is the viscosity of the fluid.

The operation is very rapid, so that the drop in temperature on ordinary work is entirely negligible. For extreme accuracy, the temperature may be raised slightly above the desired point, and an allowance made for the drop up to the moment of reading. This will seldom be found necessary in actual practice. The readings are in degrees of angular deflection, 300° to the circle, designated as °M. The practical working unit is 1/1000 of the absolute unit. As water at 20° C or 68° F has exactly 1/100 of the absolute unit of viscosity, water at this temperature reads 10° M. Thus by shifting the decimal point practical units, absolute units and specific viscosity may be obtained at one reading. Readings are taken directly from the dial, no intermediate calculations being required.

5-D. FLOAT TEST (VISCOSITY) OF PETROLEUM RESIDUES.

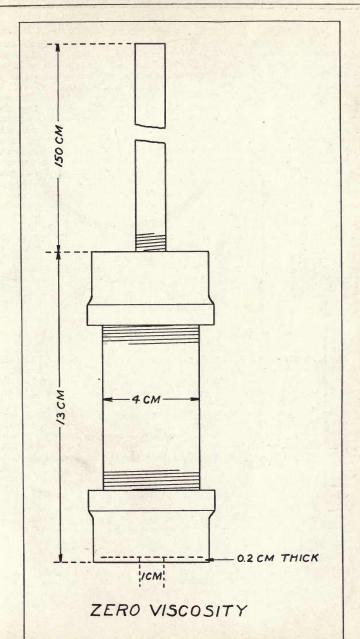
The special apparatus for the float test consists of an aluminum saucer having a diameter of 8.89 centimeters and a depth of 2.54 cm. and a radius of curvature of 5.16 cm. At the bottom there is an opening into which a collar may be screwed. This conical collar is 2.22 cm. long, is 0.95 cm. in diameter at the small end, 1.27 cm. in diameter at the large end and has a wall 0.13 cm. thick. This apparatus and method of operating is shown in the figures on page 296 In making the test the brass conical collar is placed with the

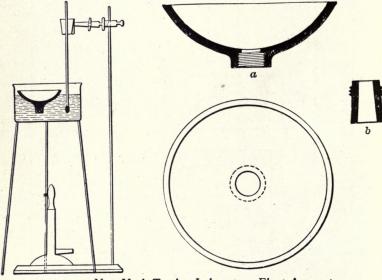
In making the test the brass conical collar is placed with the small end down on a brass plate which has been previously amalgamated with mercuric chloride. A small quantity of the material to be tested is carefully heated until quite fluid. It is then poured into the collar until slightly more than level with the top. The collar and plate are placed in ice water until rigid. The excess of material protruding from the collar is cut off with a warm knife. A pan of water is now heated to the desired temperature. The material should be kept in the ice water at least 15 minutes at a temperature of 5° C. The collar with the material is quickly screwed into the aluminum float which is immediately placed in the warm bath. As the plug of material becomes warm and fluid it is forced upward and out of the collar until the water breaks through the residue is determined with the stop watch and is recorded as the measure of the consistency of the material. Unless otherwise specified, the float test is made at 50°C, but it would necessarily be higher with the more viscous

5-E. ZERO VISCOSITY FOR SEMI-SOLID PETROLEUM PRODUCTS.

The apparatus used is a cylinder as shown in the sketch and may be constructed from ordinary iron pipe. The cylinder is 4 cm. in diameter and 13 cm. long with an opening centrally located in the bottom 1 cm. in diameter and with lips 2 mm. thick. A tube 150 cm. long is screwed into the cap on the top.

In making the test the melted asphalt is poured into the cylinder with the cap off of the top and the 1 cm. opening on the flat surface. It is cooled and topped with more asphalt, the cap is put on with 150 cm. tube and the cylinder is packed in pulverized ice and supported horizontally so that the bottom rests on a circular ring at least 1 cm. high which keeps the ice away from the orifice. The tube is filled with mercury and after some of the asphalt has protruded from the orifice it is trimmed off flush with the outer edge. The apparatus is now supported vertically at the temperature of 0°C for 5 hours. The weight of asphalt or bituminous material protruding from the orifice after this time expressed in decigrams is the zero viscosity. KANSAS CITY TESTING LABORATORY





New York Testing Laboratory Float Apparatus

6-A. MELTING POINT OF BITUMINOUS MATERIALS.

(SOFTENING POINT.) (Ring and Ball Method.) The apparatus consists of a brass ring 5%-inch in diameter, 1/4inch deep, 3/32-inch wall suspended 1 inch above the bottom of the beaker; a steel ball %-inch in diameter weighing between 3.45 and 3.50 grams, a standardized thermometer and a 600 cc. glass beaker.

Carefully melt the sample and fill the ring with the material to be tested, removing any excess. Place the ball in the center of the ring and suspend in the beaker containing 400 cc. of water at a temperature of 5° C. Set the thermometer bulb within $\frac{1}{2}$ inch of the sample and at the same level. Apply heat uniformly, preferably with a 200 watt electric hot plate over the bottom of the beaker sufficiently to raise the temperature of the water 5°C per minute. Record the temperature at starting the test and every minute thereafter until the test is completed. The softening point is the temperature at which the specimen touches the bottom of the beaker. For temperatures above 99°C glycerin should be used instead of water. Tests should check within 3°C.

6-B. MELTING POINT OF BITUMINOUS MATERIALS. (Cube Method.)

The bituminous material is carefully melted and poured into the ½-inch brass cubical mold which has been amalgamated with mercury and which is set on an amalgamated brass plate. The hot material should slightly more than fill the mold and when cold the excess may be cut off with a hot spatula. The cube is removed from the mold and fastened upon the lower arm of a No. 12 wire B. & S. gauge bent at right angles and suspended beside a thermometer in a tall covered beaker of 400 cc. capacity.

This tall form beaker is set in an 800 cc. low form beaker which is arranged for the application of heat. The wire is passed through the center of the two opposite faces of the cube which is suspended with its base one inch above the bottom of the inside beaker. The inner beaker cover has two openings, one for the wire and one for the thermometer. The wire is held in place by a cork in the cover. The bulb of the thermometer is level with the cube and at an equal dis-tance from the sides of the beaker. Heat is applied to the liquid in the outer vessel in such manner that the thermometer registers an increase of 5° C per minute and the temperature at which the bitumen touches a piece of paper placed in the bottom of the beaker is taken as the melting point. Determinations should check within 2°. The temperature at the beginning of the test should be approximately room temperature.

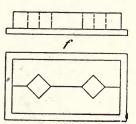
6-C. MELTING POINT OF BITUMINOUS MATERIALS. (General Electric Method.)

Mold one gram of the bituminous material so that it completely and uniformly covers the short bulb of a thermometer graduated to at least 500°F. Fit this thermometer with a cork into a 5 x 6-inch test tube with a side tubulation or air vent so that the bulb of the thermometer is ¾-inch from the bottom of the tube. Support the thermometer and tube with a clamp and immerse the tube to a depth of four inches in 400 cc. of commercial concentrated sulphuric acid in a 600 cc. beaker. The beaker of sulphuric acid is heated by direct contact with an electric hot plate of 220 watt capacity and 41/2 inches in diameter.

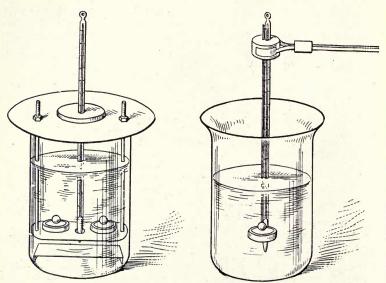
The melting point is taken from readings of the thermometer when the bituminous material flows sufficiently that a tear strikes the bottom of the tube.





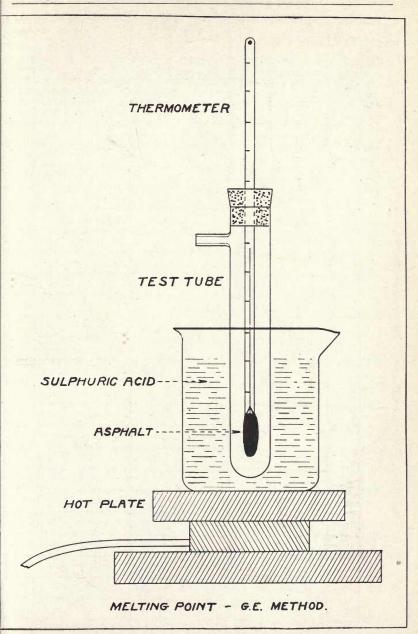


Melting Point-Cube Method



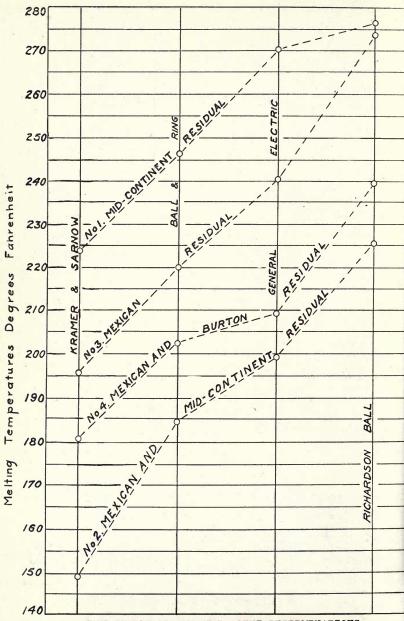
Melting Foint-Ring and Ball Method

KANSAS CITY TESTING LABORATORY



BULLETIN NUMBER FIFTEEN OF

300



COMPARISON OF MELTING POINT DETERMINATIONS.

6-D. METHOD OF DETERMINING THE MELTING POINT OF PARAFFIN WAX.

Use the apparatus consisting of a 1x8 inch test tube fitted into a bottle all as shown on page 302.

Pour the melted paraffin into the test tube to a depth of three inches and insert a special wax or titer test thermometer graduated to $.1^{\circ}F$. Place it exactly in the center of the tube so that the bottom of the thermometer is one-half inch from the bottom of the test tube. Let the apparatus stand in a warm place preferably in a blood temperature incubator or at 100°F and take readings of the thermometer every minute.

Continue the readings until the wax is nearly solid.

The melting point is the average of the three one-minute readings which are most nearly identical. In the case of high melting point wax these readings are practically identical. In the case of low melting point wax there would be some difference in the readings.

This method is graphically illustrated on page 303.

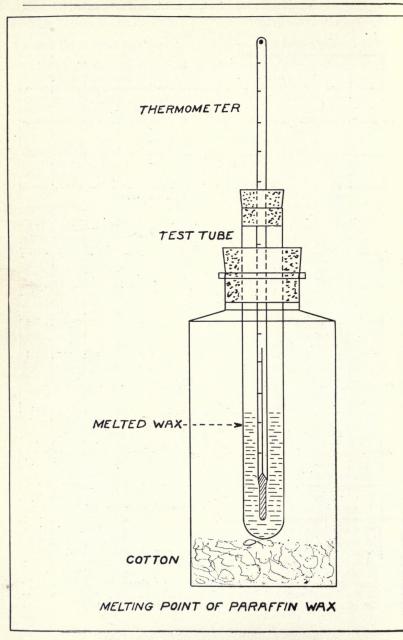
Note.—See Scientific Paper No. 340 of U. S. Bureau of Standards, Sept. 12, 1919. 7-A. CLOUD, POUR AND COLD TESTS OF PETROLEUM OILS.

7-A. CLOUD, POUR AND COLD TESTS OF PETROLEUM OILS. Put the oil to be tested in a glass jar or bottle approximately 1¼ inches inside diameter and 5 inches high, to a height of about 1¼ inches or sufficient to reach ¼-inch above the mercury bulb of the thermometer. The thermometer should have a bulb about %-inch long and is held centrally in the jar with a tightly fitting cork and with the lower end of the bulb ½-inch from the bottom of the jar.

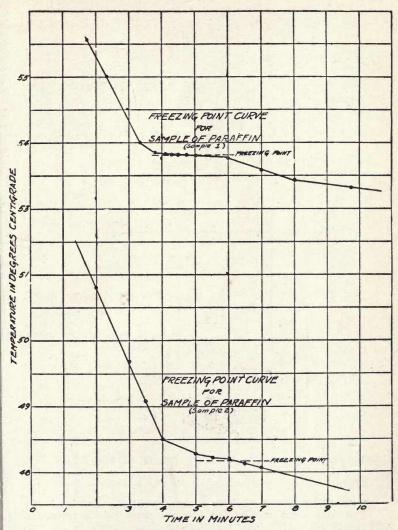
Now place the cold test bottle in a metal or glass jacket about 5 inches high having a diameter about $\frac{1}{2}$ -inch greater than the outside diameter of the bottle. A disc of cork $\frac{1}{4}$ -inch in thickness is placed in the bottom of the jar. Put the apparatus in the refrigerating mixture and place the bottle so that it does not touch the jacket. At every drop in temperature of 2°F when near the expected cloud test, remove the jar from the jacket, being careful not to disturb the oil by moving the thermometer. When the lower half of the sample becomes opaque through chilling, read the thermometer. This temperature is the cloud test of the oil. 7-B. For the POUR TEST (usually called Cold Test) continue the

7-B. For the POUR TEST (usually called Cold Test) continue the cloud test, and at each drop in temperature of 5°F remove the bottle from the jacket and tilt it until the oil begins to flow. When the oil has become solid around the thermometer and will not flow, the previous 5° point shall be taken as the pour test. For oil solidifying above 35°F pounded ice may be used for the refrigerating mixture. For temperatures below this and down to -5°F a mixture of 2 parts of pulverized ice and 1 part of salt may be used. From -5°F and to -25°F a mixture of equal parts of pulverized ice and calcium chloride may be used. A universal frozen mixture for all these temperatures can be made as follows:

Put a sufficient amount of dry acetone into a covered metal beaker. Put the beaker into an ice salt mixture and when the temperature of the acetone reaches 10° F or below, slowly add carbon dioxide snow until the desired temperature is reached. A temperature as low as -70° F can be thus attained. To get the solid carbon dioxide snow invert an ordinary carbon dioxide cylinder. Open the valve slowly and let the liquid run out into a close mesh bag. By rapid evaporation the carbon dioxide becomes solid. (Continued p. 304.)

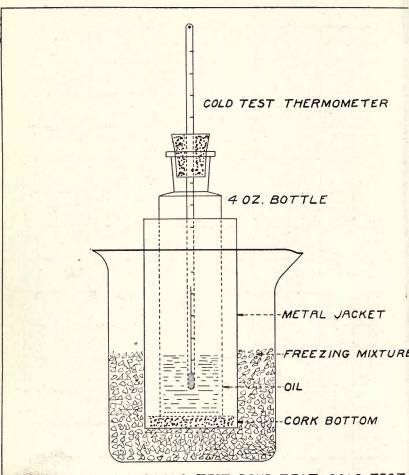


KANSAS CITY TESTING LABORATORY



Solidification curves for paraffin

BULLETIN NUMBER FIFTEEN OF



APPARATUS FOR CLOUD TEST, POUR TEST, COLD TEST.

7-C. COLD TEST specially for steam cylinder and black oils. The same bottle used in the pour and cloud tests is filled ¼ full and frozen with a freezing mixture. A thermometer is then introduced into the frozen mass and after it has become cold the bottle containing the solidified oil is removed from the cooling mixture. The solidified oil is thoroughly stirred with the thermometer until the

mass will run from one end of the bottle to the other and at this moment the temperature indicated is recorded. This reading is the cold test of the oil.

8-A. SEDIMENT, WATER, DIRT AND BOTTOM SETTLINGS IN PETROLEUM.

(Apparatus is shown on page 337.)

50 cc. of the oil are thoroughly mixed with 50 cc. of benzol and the mixture is poured into a 100 cc. graduated V-shaped centrifuge tube such as is shown on page 337. This is exactly counter-balanced and run in the electric centrifuge at a speed of approximately 2,000 R. P. M. for 5 minutes or until there is a sharp line of demarkation between the sediment or dirt, the water and the oil, if any water or dirt are present. The amount of sediment or dirt is read off by volume and expressed in percentage by volume. The water is also expressed in percentage by volume.

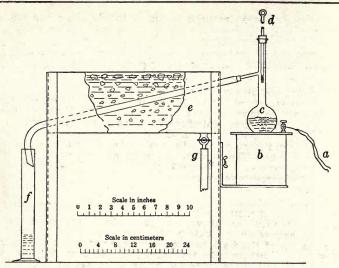
8-B. WATER BY DISTILLATION OF PETROLEUM.

100 cc. of the oil are placed in a flask connected to a distillation apparatus as shown on page 308. This is heated until foaming starts when an auxiliary flame is applied to all parts of the upper portion of the flask causing any water vapor to pass over into the condenser without allowing water to collect in the neck of the flask. This heating also tends to prevent the extension of the foam into the condenser. The flame beneath the flask must be applied very gently. This is continued until all foaming ceases and all water has been distilled over from the condenser. The number of cubic centimeters of water collected in the receiver is the percentage of water.

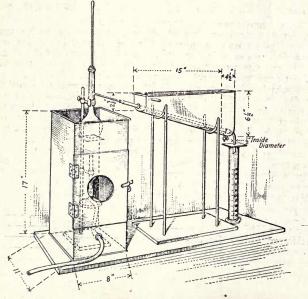
9-A. PROXIMATE DISTILLATION OF PETROLEUM.

400 cc. of the petroleum are poured into a 1,000 cc. flask which is connected to a condenser (as shown on page 310). The thermometer is inserted so that the top of the bulb is just below the outlet of the flask. The flame is gradually applied to the oil so that any foaming will tend to make itself evident. If there is foaming it will be necessary to heat the upper portion of the flask. Before the application of the flame to prevent foaming, it is necessary to get the temperature at which the first drop falls into the receiver. This is the initial boiling point. The distillate is collected until a temperature of 410° F is reached when distillation is proceeding at the rate of 5 cc. per minute. The fraction collected up to this temperature is the gasoline or naphtha, the gravity of which is determined. If the gravity is less than 57, it is classified as naphtha, if above this, it is classified as gasoline. The distillation is continued at the same rate until a temperature of 572° F is reached. This fraction is kerosene and its gravity is determined. The residue in the flask is fuel oil and is used for the determination of wax or asphalt, gas oil or lubricants. The information given by this distillation is:

water	/0
Gasoline (410°F)	$(Gr. =Be^{\circ})\%$
	$(Gr. =Be^{\circ})$
Fuel Oil—Residuum	$(Gr. =Be^{\circ})\%$



-Apparatus used by the Bureau of Mines for distillation test of gasoline. a Wires connecting with electric mains through a suitable rheostat. b Electric heater. c Engler distillation flask filled with charge of gasoline partly distilled. d Thermometer. e Condenser, with trough filled with lee and water. f Receiving graduate, g Cock for draining condenser trough.



American Society for Testing Materials Apparatus.

9B. END POINT DISTILLATION TEST OF GASOLINE,

NAPHTHA AND BENZINE.

This method is essentially that of the American Society for Testing Materials, page 606 of 1918 Book of Standards, and is the method given by the Bureau of Mines in Technical Paper 166 with slight modifications.

The apparatus used in the distillation is as follows: The flask used shall be the standard 100 cc. Engler flask. The dimensions are as follows:

Cm.	Inches
Diameter of bulb 6.5	2.56
Diameter of neck 1.6	0.63
Length of neck	5.91
Length of water tube	3.94
Diameter of vapor tube 0.6	0.24

Position of vapor tube 9 cm. (3.55 in.) above surface of oil when flask contains its charge of 100 cc. The tube is approximately in the middle of the neck.

The flask shall be supported on a ring of asbestos having a circular opening 1¹/₄ in. in diameter; this means that only this limited portion of the flask is to be heated. The use of a sand bath is not approved.

The condenser tube shall consist of a thin walled tube of metal (brass or copper) $\frac{1}{2}$ inch internal diameter and 22 inches long. It shall be set at an angle of 75° from the perpendicular and shall be surrounded with a water jacket of the trough type. The lower end of the condenser shall be cut off at an acute angle and shall be curved down for a length of 3 inches. The condenser jacket shall be 15 inches long.

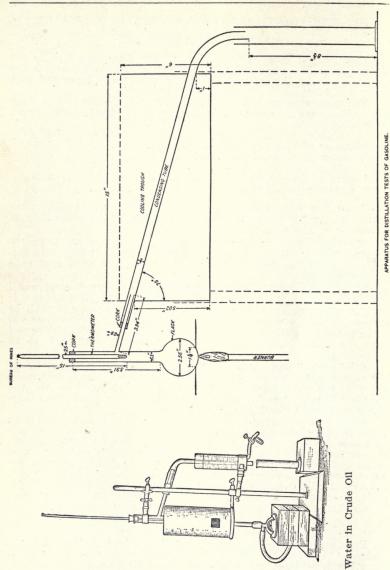
Briefly the thermometer should be an accurate "nitrogen-filled" instrument with a short bulb (length 10 to 15 mm. 0.39 to 0.59 inch) and with the mark for 35° C (95° F) at a distance between 100 and 120 mm. (3.94 to 4.73 inches) from the top of the bulb. The thermometer should be scaled for total immersion.

The above requirements insure that the lowest temperatures registered may be read above the cork of the distillation flask and variations because of the so-called "stem correction" will always be practically the same. The stem correction should not be applied but it should be understood that the results of distillations are expressed in terms of thermometer readings, not of actual temperatures. The use of partial-immersion thermometers is not recommended for distillations as these instruments are no more likely to agree with one another than are the more common total immersion thermometers.

Method of Distillation.

Pour some of the gasoline or naphtha to be tested through the condenser tube just before the distillation flask is connected with it and allow it to drain before placing the receiver. Likewise the distillation flask is rinsed out with the gasoline or naphtha to be distilled and drained before the charge of 100 cc. is added to it. The 100 cc. graduated cylinder may be used without drying as the receiving vessel for the distillation.

BULLETIN NUMBER FIFTEEN OF



The thermometer bulb should be covered with a thin film of absorbent cotton. This keeps the glass always wet with the condensate from the vapor and thus prevents possible fluctuations in the temperature. It also tends to prevent superheating of the bulb at the end of the distillation and thus makes possible an accurate determination of the dry point.

Heat should be applied to the flask in regulated degree, care being taken that the whole distillation from beginning to end shall proceed at a rate of not less than 4 nor more than 5 cc. a minute (about 2 to 3 drops per second). Readings of the thermometer shall be made as each 5% distills. The temperature at which the first drop falls from the exit of the condenser tube is the initial boiling point.

The dry point, end point or highest temperature reading at the end of the distillation shall also be recorded. It is the temperature when the last drop is vaporized and a puff of white vapor appears in the flask. The distillation loss shall be determined by adding the percentage of residue in the distilling flask, after cooling, to the percentage of total distillates held in the receiver. If the distillation loss is over 3%, a check distillation shall be made, as excessive loss may indicate that the rate of distillation at the beginning was too rapid. In case the magnitude of the loss is confirmed this fact is of importance in indicating that the gasoline contains very volatile constituents, particularly those derived from added casinghead gasoline.

The condenser trough shall be filled with a mixture of finely cracked ice and water (not dry cracked ice) and during the distillation sufficient ice shall be kept in the trough to prevent the temperature of the cooling water exceeding 4° C (39° F).

If distillations are made at high altitudes or when barometric pressures are low, allowances may be made for this factor. In general, recording the barometric pressure read at the time of the distillation will suffice and it is recommended that whenever there is possibility of dispute over the results of a distillation this should be done.

In finishing the distillation there is always a small amount of naphtha remaining in the flask in the vapor phase in excess of that required to wet the inside of the flask. If the residue in the flask has not been poured into the receiver the end point of 98% is to be read as 100% and any loss is to be calculated as the difference between the 98% and the amount actually recovered after the condenser tube has thoroughly drained.

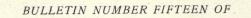
This method is identical with that of the Bureau of Mines with the following exceptions:

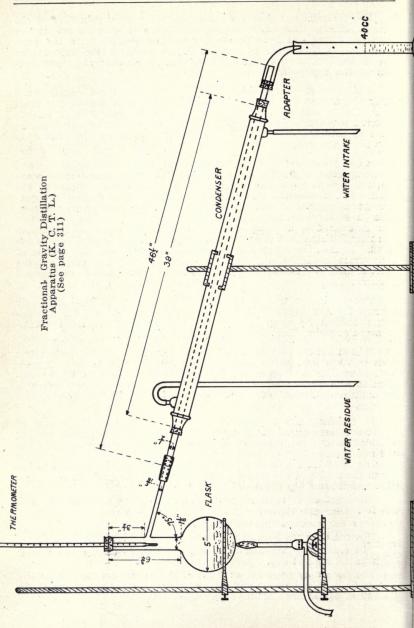
Five cc. readings instead of 10 cc. are made.

The condenser is kept at 4°C or below (B. of M. is below 8°C).

The condenser tube and flask are moistened with the gasoline or naphtha to be distilled.

The preceding page shows a cut of the apparatus for distillation. The A. S. T. M. apparatus and the electrically heated apparatus of the Bureau of Mines are shown on page 306.





9C. FRACTIONAL GRAVITY DISTILLATION OF PETROLEUM.

Use a 1,000 cc. distilling flask of heavy pyrex glass having a diameter of 13 centimeters, a neck 17 centimeters long with a 3 centimeter diameter and with a side tubulus 8 centimeters above the body of the flask. The tubulus is set at approximately 75° to the neck of the flask. The condenser tube is 36 inches long and the water jacket is 30 inches long. The details of the set-up are shown on page 310.

The oil to be used should be as nearly as possible free from water. Eight hundred cubic centimeters are poured into the distillation flask, the thermometer used is preferably for 5 inch immersion reading to 750° F. It is inserted so that the top of the mercury bulb is even with the bottom of the tubulus and is in the center of the neck of the flask.

Distillation is begun using a smoky flame of a strong Tirrell burner, the flask being supported on a ring as shown in the diagram. The burner is protected from air drafts and the flask is blanketed with asbestos paper if necessary. The flame is controlled by a screw pinch cock on the rubber tubing.

The temperature at which the first drop falls from the condenser is the initial boiling point. The rate of distillation after the first 5% is 8 cubic centimeters or 1% per minute. Five per cent fractions are collected in the 100 cubic centimeter cylinder. These 40 cubic centimeters are poured into a 50 cc. graduate, allowing the distillate to mix thoroughly. The specific gravity is taken and the corrections are made to 60°F. The end point of each $2\frac{1}{2}$ % fraction is recorded and the distillation is continued, taking the gravity of each 5% fraction. In operating on a crude oil in which the natural content is desired, the distillation with straight fire is stopped when the first fraction with a temperature above 572°F is completed. Beyond this temperature inert gas, such as natural gas, coal gas or carbon dioxide is introduced sufficiently to carry the distillation at the same rate of speed but such that the temperature at no time exceeds 650°F. After the gas is used the water is removed from the condenser and the condenser tube is kept warm to prevent wax occluding the tube.

Ninety per cent should be carried over and the gravity of the residue taken.

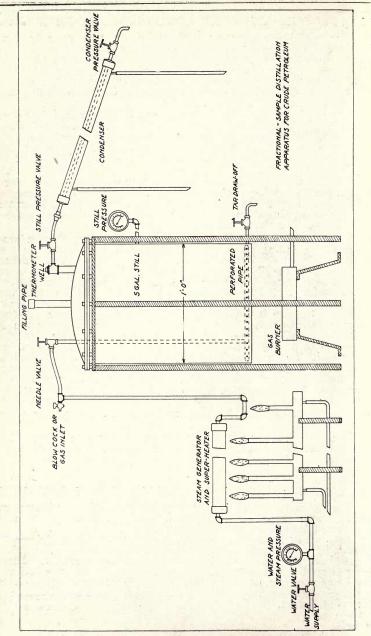
The data obtained by this distillation is shown on pages 122 to 127 for crude oil, on pages 231-2 for gasoline and page 230 for heavy distillate.

9D. SAMPLE PREPARATION DISTILLATION OF CRUDE OIL.

The apparatus consists of a 5-gallon steel still, condenser, gas burner, water supply under pressure, steam producers, superheater gauges and connections as shown on page 312.

Ten thousand cubic centimeters is a convenient charge, giving a 5% fraction of 500 cc., which is sufficient for special tests. The still is covered with chicken wire and asbestos cement for insulation. Direct firing is used until a temperature of slightly above 500° F is indicated in the vapor or a gravity of 40° Be' (0.825 specific gravity) is shown in the distillate fraction. At this temperature superheated steam or gas is introduced.

BULLETIN NUMBER FIFTEEN OF



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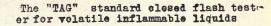
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Thermometer, indicating the temperature of the oil.

Thermometer, indicating the temperature of the water bath.

A miniature oil well to supply the test flame when gas is not available, mounted on the axis about which the test-flame burner is rotated, which axis is hollow and provided with connection on one and for gas hose, and provided also with needle valve for controlling gas supply, when gas is available, the gas passing through the empty oil well.

Gas or oil tip for test flame.

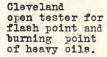
- Cover for oil cup, provided with three openings, which are in turn covered by a movable slide operated by a knurled hand knob, which also operates the test flame burner in unison with the movable slide, so that by turning this knob. the test flame is lowered into the middle opening in the cover, at the same time that this opening is uncovered by the movement of the slide.
- Oil cup (which cannot be seen in the illustration), of standardized size, weight and shape, fitting into the top of the water bath.

Overflow spout.

- Water bath, of copper, fitting into the top of the body, and provided with an overflow spout and openings in its top, to receive the oil cup and water bath thermometer.
- Body of metal, attached to substantial cast metal base provided with three feet.

Alcohol lamp for heating the water bath.

Gas hote.



and the second sec



New York closed tester for all types of oils. (Elliott)

10A. FLASH POINT OF KEROSENE AND OTHER VOLATILE INFLAMMABLE LIQUIDS.

(With Standard "TAG" Closed Tester.)

This is essentially in accordance with the method of the American Society for Testing Materials, Tentative Standards, 1917, pages 445-6.

The test must be performed in a dim light so as to see the flash plainly.

Surround the tester on three sides with an inclosure to keep away drafts. A shield about 18 inches square and 2 feet high, open in front, is satisfactory. See that tester sets firmly and level. For accuracy, the flash point thermometers which are especially

designed for the instrument should be used as the position of the bulb of the thermometer in the oil cup is essential.

Put the water-bath thermometer in place. Place a receptacle under the overflow spout to catch the overflow. Fill the water bath with water at such a temperature that when testing is started, the temperature of the water bath will be at least 10°C below the probable flash point of the oil to be tested. Put the oil cup in place in the water bath. Measure 50 cc. of

the oil to be tested in a pipet or a graduate and place in oil cup. The temperature of the oil must be at least 10°C below its probable flash point when testing is started. Destroy any bubbles on the surface of the oil. Put on cover with flash point thermometers in place and gas tube attached. Light pilot light on cover and adjust flame to size of the small white bead on cover.

Light and place the heating lamp, filled with alcohol in base of tester and see that it is centrally located. Adjust flame of alcohol lamp so that temperature of oil in cup rises at the rate of about 1°C (1.8°F) per minute or not faster than 1°C (1.8°F) nor slower than 0.9°C (1.6°F) per minute.

Record the "time of applying the heating lamp," record the "temperature of the water bath at start," record the "temperature of the oil sample at start."

When the temperature of the oil reaches about 5°C below the probable flash point of the oil, turn the knob on the cover so as to introduce the test flame into the cup and turn it promptly back again. Do not let it snap back. The time consumed in turning the knob down and back should be about one full second, or the time re-quired to pronounce distinctly the words "one thousand and one." Record the "time of making the first introduction of the test flame" and record the "temperature of the oil sample at time of first

flame" and record the "temperature of the oil sample at time of first test."

Repeat the application of the test flame at every 0.5°C rise in temperature of the oil until there is a flash of the oil within the cup. Do not be misled by an enlargement of the test flame or halo around it when entered into the cup or by slight flickering of the flame; the true flash consumes the gas in the top of the cup and causes a very slight puff.

Record the "time at which the flash point is reached," and the "flash point."

If the rise in temperature of the oil from the "time of making the first introduction of the test flame" to the "time at which the flash point is reached" was faster than 1.1°C or slower than 0.9°C per minute, the test should be questioned and the alcohol heating lamp adjusted so as to correct the rate of heating. It will be found that the wick of this lamp can be so accurately adjusted as to give a uniform rate of rise in temperature of $1^{\circ}C$ per minute and remain so.

Repeat Tests.—It is not necessary to turn off the test flame with the small regulating valve on the cover, but leave it adjusted to give the proper size of flame.

Having completed the preliminary test, remove the heating lamp, lift up the oil cup cover and wipe off the thermometer bulb. Lift out the oil cup and empty and carefully wipe it. Throw away all oil samples after once using in making test.

Pour cold water into the water bath, allowing it to overflow into the receptacle until the temperature of the water in the bath is lowered to 8° C below the flash point of the oil as shown by the previous test. With cold water of nearly constant temperature it will be found that a uniform amount will be required to reduce the temperature of the water bath to the required point.

Place the oil cup back in the bath and measure into it a 50 cc. charge of fresh oil. Destroy any bubbles on the surface of the oil, put on the cover with its thermometer, put in the heating lamp, record time and temperature of oil and water and proceed to repeat test as described above. Introduce test flame for first time at a temperature 5° C below the flash point obtained on the previous test.

Precautions.—Be sure to record barometric pressure either from laboratory barometer or from nearest Weather Bureau station. Record temperature of room.

Note and record any flickering of the test flame or slight preliminary flashes when the test flame is introduced into the cup before the proper flash occurs. Record time and temperature of such flickers or slight flashes if they occur.

10B. FLASH AND BURNING POINT OF ALL TYPES OF PETROLEUM OILS AND ASPHALTS. (With New York or Elliott Closed Tester.)

The bath surrounding the oil cup is filled with very high flash fluid oil or is left unfilled if the oil to be tested has a very high flash point. The oil cup is filled with the material to be tested to within 3 millimeters of the flange joining the cup and the vapor chamber above. The glass cover is then placed on the oil cup and the thermometer adjusted so that its bulb is just covered by the oil or bitumen. The flame is applied to the bath in such manner that the temperature is raised at the rate of about 5°C per minute. Every half minute the testing flame is inserted in the opening in the cover and about halfway between the surface of the material and the cover. The first appearance of a faint bluish flame on the entire surface of the bitumen or oil shows that the flash point has been reached, and this temperature is recorded.

The burning point of the material is now obtained by removing the glass cover and replacing the thermometer in the frame. The temperature is raised at the same rate and material tested as before. The temperature at which the oil or bitumen ignites and burns is recorded as the burning point. The flame should be extinguished with the metal cover very promptly after the burning point is reached.

10C. FLASH AND BURNING POINT OF LUBRICANTS. (With the Cleveland Open Cup.)

The lubricating oil is poured into the oil cup to within 5 mm. of the top. The flame is then applied to the air bath in such manner that the temperature of the oil in the cup is raised at the rate of 5°C per minute. The testing flame is made from a piece of drawn glass tubing, making a flame about 5 mm. in length. This flame is applied to the surface of the oil every half minute. A distinct flicker or flash over the entire surface of the oil shows that the flash point is reached and the temperature at this time is recorded.

The burning point of the oil is obtained by continuing the test and noting the temperature at which the vapor arising from the surface of the oil ignites and burns continuously. The thermometer is quickly withdrawn and the metal cover used to extinguish the flame.

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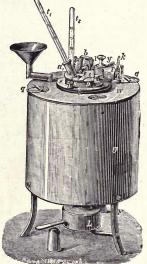
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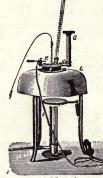


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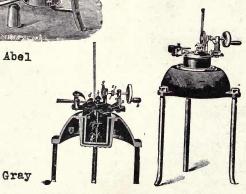


Scott





Pensky-Martens



11A. CRACKING TEST FOR HEAVY PETROLEUM

HYDROCARBONS. (See P. 320.) The apparatus is set up as shown in sketch. (a) is a cylindrical tube tested out to a pressure of 3,000 pounds such as is ordinarily used for dispensing oxygen gas. (b) is a thermometer well or plug with a tapered thread and of sufficient length that it protrudes well into the interior of the vessel (a). This plug has an opening from the outside into which the thermometer (c) is inserted. This mercury thermometer is graduated preferably in single degrees Centigrade and is of borosilicate glass, nitrogen filled and reading up to a temperature of 550 °C. (d) is an extra heavy ammonia pipe fitting connected to a valve (e) and a pressure gauge (f). Pressure gauge (f) should read to at least 200 atmospheres or 200 kilograms per square centimeter. Heat is applied by gas burners (g) such as are used in combustion furnaces and the whole apparatus is supported on a stand with the end carrying the pressure gauge slightly elevated. The capacity of the bomb is 1,500 to 1,600 cubic centimeters

and 500 cc. of oil to be tested are poured into it at a temperature of approximately 20°C. The plug (b) is inserted and screwed in very tightly, using Stilson wrenches. The threads on the plug may be dressed with a mixture of equal parts of glycerin, litharge and cop-per oxide. The flame is applied so that it does not excessively heat the portion of the container not in contact with the oil. The total time consumed for the test after the beginning of the application of the heat should be between 55 minutes and 70 minutes. The heating is carried on until a pressure of 55 atmospheres is attained, based on a temperature of 400°C. It is desirable to keep the container covered with a sheet of asbestos during the operation. The temperature should not ordinarily exceed 420°C. The apparatus is cooled to about 20°C before opening.

The constants in this test are the dimensions of the apparatus, the amount of oil used, the rate of application of heat and maximum pressure at 400°C.

The variables are the percentage by volume of oil recovered after cracking, the amount of carbon formed, the amount of gas formed, the specific gravity of the gasoline and the total yield of gasoline. (See proximate distillation of crude oil.)

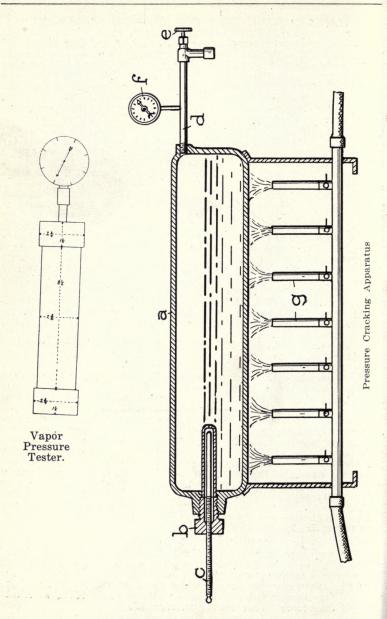
Variations are due to the character of the oil treated, the spe-cific gravity of the gasoline being higher, the recovery higher, the carbon and gas formation less and the total amount of oil recovered greater with paraffin base and with low gravity oils than with naphthene base and high gravity oils.

From one such equilibrium test it is possible to approximately estimate the amount of total gasoline which it would be possible to obtain from an oil. This may be calculated from one equilibrium test by taking into consideration the shrinkage on cracking and the increase in specific gravity of the residue above 210°C after cracking. (See pages 228-9.)

11B. VAPOR PRESSURE.

The vapor pressure of light petroleum hydrocarbons is determined with the same apparatus used for making the cracking test. The pressure readings with the corresponding temperature readings should be taken every 30 pounds and a curve plotted for intermediate points. The temperature should not be carried above 350°C as cracking will take place. (See curves on page 226.)

BULLETIN NUMBER FIFTEEN OF



11C. VAPOR PRESSURE TESTS FOR LIGHT GASOLINE MADE FROM GAS.

(See Westcott, Handbook of Casinghead Gasoline.)

Apparatus shown on page 320 consists of iron or steel pipe of 2 inch size, with caps screwed on ends. Upper cap has 0.25 inch nipple screwed in and is connected by a coupling to a 3 inch 30 lb. pressure gauge. Gauge is known as Inspector's Gas Gauge. All joints must be entirely tight. Joints between large pipe and caps are best sealed with solder. Approximate external dimensions are indicated on sketch. In addition to apparatus indicated in test, there is also required a tin cylinder for filling test tube, 12 by 3 inches, that can be slipped over outside of tube for convenience in carrying when not in use. The tin cylinder is provided with a lip for pouring. A small tin cover 0.75 inch deep, fitting over the bottom of the tin cylinder may be removed and used for measuring off one-tenth capacity of test tube. A small tin funnel 2.5 inches in diameter with stem 3 inches long and three-sixteenths inch in diameter should be used.

Remove the gauge from the tube and fill tube to 90 per cent of its capacity. Fill tube preferably by lowering it into the storage tank in upright position by means of a cord or wire. Leave the tube entirely immersed for several minutes, withdraw it and pour off sufficient liquid so that tube will contain 90% of its capacity. A small measure having capacity of 10% of the test tube should be used for that purpose.

In case it is impracticable to lower the tube into the storage tank, draw the liquid off into the vessel of capacity about equal to the test tube. Pour liquid into the test tube until about half filled. Shake tube and contents gently in order to bring both to the same temperature. After standing for several minutes, pour out all the liquid from the tube. Draw another sample from the storage tank into the cylinder and pour through funnel into the tube until the latter is entirely full. Withdraw one-tenth as before. Screw gauge tightly into position, using a little liquid shellac on joint to insure a tight fit.

Immerse the tube in water at temperature of 70°F and allow it to remain for five minutes. Then remove it from the water and unscrew the gauge sufficiently to relieve the pressure indicated by the gauge for a period of 20 seconds and screw the gauge tightly into the tube again. Then place the tube in water at a temperature of 100°F (90°F from Nov. 1st to March 1st). The level of the water must be just below the lower edge of the pressure gauge. Stir the water continually and maintain the temperature exactly constant for ten minutes, then tap the gauge lightly with the finger and read the pressure.

A correction of pressure figures should be made according to the initial temperature of the gasoline. This correction should be as follows:

For tests on samples taken at a temperature of 50 to 59°F, inc., deduct 1 lb.

For tests on samples taken at a temperature of 40 to 49°F, inc., deduct 2 lbs.

For tests on samples taken at a temperature below 40°F, deduct 3 lbs.

The gravity of the liquid, the temperature of liquid gas placed in test tube, the pressure at 70°F before venting tube, the corrected pressure at 100°F (90°F from Nov. 1st to March 1st) after venting at 70°F should all be recorded.

12A. CARBON RESIDUE IN LUBRICANTS AND DISTILLATES. (Conradson Method.)

The apparatus consists of:

(a) Porcelain crucible, wide form, glazed throughout, 25 to 26 cc. capacity, 46 mm. in diameter.

(b) Skidmore iron crucible, 45 cc. (1½ oz.) capacity, 65 mm. in diameter, 37 to 39 mm. high with cover, without delivery tubes and one opening closed.

(c) Wrought iron crucible with cover, about 180 cc. capacity, 80 mm. diameter, 58 to 60 mm. high. At the bottom of this crucible a layer of sand is placed about 10 mm. deep, or enough to bring the Skidmore crucible with cover on nearly to the top of the wrought iron crucible.

(d) Triangle, pipe stem covered, projection on side so as to allow flame to reach the crucible \in n all sides.

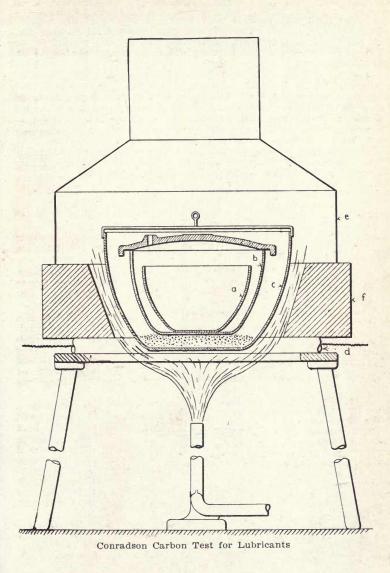
(e) Sheet iron or asbestos hood provided with a chimney about 2 to $2\frac{1}{2}$ inches high, $2\frac{1}{8}$ to $2\frac{1}{4}$ inches in diameter to distribute the heat uniformly during the process.

(f) Asbestos or hollow sheet iron block, 6 to 7 inches square, $1\frac{1}{4}$ to $1\frac{1}{2}$ inches high, provided with opening in center $3\frac{1}{4}$ inches in diameter at the bottom and $3\frac{1}{2}$ inches in diameter at the top. The test shall be conducted as follows:

Ten grams of the oil to be tested are weighed in the porcelain crucible, which is placed in the Skidmore crucible and these two crucibles set in the larger iron crucible, being careful to have the Skidmore crucible set in the center of the iron crucible, covers being applied to the Skidmore and iron crucibles. Place on triangle and suitable stand with asbestos block and cover with sheet iron or asbestos hood in order to distribute the heat uniformly during the process.

Heat from a Bunsen burner or other burner is applied with a high flame surrounding the large crucible, as shown in Fig. 1, until vapors from the oil start to ignite over the crucible, when the heat is slowed down so that the vapor (flame) will come off at a uniform rate. The flame from the ignited vapors should not extend over 2 inches above the sheet iron hood. After the vapor ceases to come off, the heat is increased as at the start and kept so for five minutes, making the lower part of large crucible red hot after which the apparatus is allowed to cool somewhat before uncovering the crucible. The porcelain crucible is removed, cooled in a dessicator and weighed.

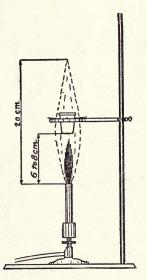
The entire process should require about one-half hour to complete when heat is properly regulated. The time will depend somewhat upon the kind of oil tested, as a very thin, rather low flashpoint oil will not take as long as a heavy, thick, high flash-point oil. (See A. S. T. M. 1918 Standards, page 620.)



12B. FIXED CARBON AND ASH IN OIL AND BITUMINOUS MATERIALS.

The apparatus used is that shown below, or if the apparatus used for the analysis of coal is available, the special furnace shown on page 325 may be used, or the electric furnace shown on page 348, such as is used for burning out mineral aggregates, is quite satisfactory.

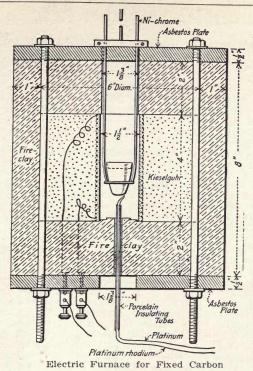
Between .4500 and .5500 gram of the material is placed in a 20-gram platinum crucible having a tightly fitting cover. It is heated for seven minutes with the full flame of a Bunsen burner, as shown, or at 950° C in the electric furnace. With the open flame the crucible should be supported with its bottom 6 or 8 cm. above the top of the burner and the flame should be at least 20 cm. high when burning freely. A shield is used to protect from drafts. The crucible while remaining covered is placed in a dessicator, cooled and weighed, then ignited with lid removed until nothing but the ash remains. The loss is the fixed carbon and the residue is the ash.

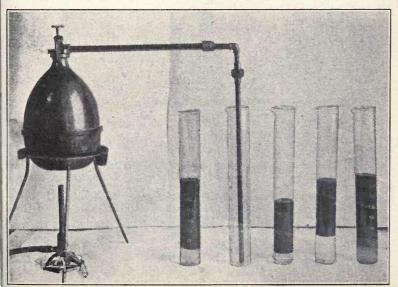


Apparatus for the Determination of Fixed Carbon

324

KANSAS CITY TESTING LABORATORY





13. EMULSIFICATION OF MINERAL LUBRICATING OILS. (A. S. T. M.—1916.)

P. H. Conradson. Apparatus.

The apparatus consists of a 4-pint copper retort, provided with a delivery tube, which is joined to a metal or glass pipe having an inside diameter of about 5/16 in. and about 15 in. long from the elbow. The lower end of this pipe is cut off diagonally to prevent thumping.

The glass cylinders are graduated to 250 cc. They have an inside diameter of about 17/16 in. and a length of about $9\frac{1}{2}$ in. from the bottom to the 250 cc. mark. They are $11\frac{1}{2}$ to 12 in. in over-all length, and are made of thin glass, with a flat bottom. In place of a copper retort for the generation of steam, a glass

In place of a copper retort for the generation of steam, a glass flask or any other suitable source of steam supply may be used; likewise, ordinary 250 cc. graduated glass cylinders, of dimensions given above, may be used where emulsion tests are required only occasionally.

Method of Testing.

The cylinder is filled with distilled water up to the 20 cc. mark, then 100 cc. of the oil to be tested are added. To churn the mixture, steam at ordinary pressure is conducted through this oil-water mixture for ten minutes. The amount of steam passed through is regulated in such a way so as to prevent the mixture from splashing over the top of the cylinder, but the rate may be as rapid as is practical. This is easily regulated by the height of the gas flame.

The churning is begun from the time the temperature of the mixture has reached 200°F, or when the steam as such passes off the mixture. It usually takes from 1 to $1\frac{1}{2}$ minutes to reach this temperature, depending somewhat on the body or viscosity of the oil. However, even churning with steam for 15 minutes does not seem to make any difference in the results.

When the churning is completed, the cylinder is immersed for one hour in a water bath, kept at a temperature of 130°F. During this time the cylinder and its contents are momentarily inspected at intervals to note the behavior of the oil mixture. At the expiration of one hour the cylinder is removed from the water bath and the contents are examined for the following:

1. The number of cubic centimeters of separated clear or turbid water.

2. The number of cubic centimeters of separated emulsified layer.

3. The number of cubic centimeters of separated clear or turbid oil above the emulsified layer; and

4. The percentage of water or moisture in the separated oil above the emulsified layer.

The number of cubic centimeters and condition of the emulsified layer is an indication of the emulsion-forming property, or quality of the oil.

The number of cubic centimeters of clear or turbid oil above the emulsified layer, less the percentage of water or moisture contained in the oil, is the percentage of demulsibility of the oil.

The condition of the separated water or watery liquid under the emulsified layer, if any, gives an indication also of the behavior of the oil in actual service. The amount of water held in the oil above the emulsified layer may be determined as follows:

The oil above the emulsified layer after the expiration of the test is carefully drawn off and shaken; then 20 cc. are mixed with 80 cc. of 88° Baume' gasoline (from Pennsylvania Crude) in a graduated, flat-glass precipitating tube having the lower end drawn out. The oil-gasoline mixture is kept at a temperature not over 80°F for one hour, or the water or watery liquid may be separated from the oil-gasoline mixture by means of a centrifuge. The amount of water or watery liquid is read off and calculated to percentage by volume and subtracted from the oil above the emulsified layer. Of course, this determination is only necessary when the oil above the emulsified layer appears to contain an appreciable amount of water.

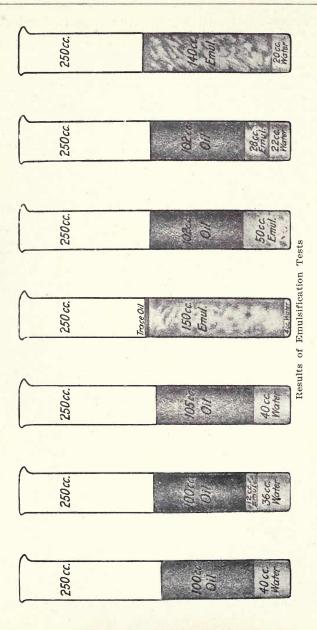
Interpretation of Results.

Page 328 illustrates the behavior of seven representative oils with this method, as they appear after expiration of the tests.

Table I gives detailed results of tests of these oils. The first turbine oil shows the oil entirely free from emulsifying property or elements, only retaining a very small percentage of moisture or water after the expiration of the test. The second turbine oil shows quite

TABLE IEMULSIFICATION TESTS OF LUBRICATING OILS.									
Oil No.	Kind of oil	Separated water, etc.	Condition of Water	Emulsified layer, etc.	Kind of emulsion	Separated oil, etc.	Condition of oil	Moisture in oil %	Demulsibility %
1	Turbine	40	Clear	None		100	Slightly turbid	0.2	99.8
2	Turbine	36	Slightly turbid	12	Light foamy	100	Turbid		99.0
3	Crank case	40	Clear	None		105	Very turbid	5.0	95.0
							Pract- ically all		
4	Crank case	4	Clear	150	Heavy thick	Trace	emul- sion		0.0
5	Engine	None		50	Thick milky	102	Very turbid	4.0	96.0
6	Engine	22	Clear	28	Light foamy	102	Turbid	3.0	97.0
7	Spindle	20	Milky	140	Thick milky	None	All emul- sion		0.0

327



a little of emulsified layer, but the condition of the emulsifier layer is light and foamy, not compact or creamy. The amount of water or moisture retained in the oil is much higher than in the first oil.

Consider next the two samples of crank-case oil: The first oil shows ready separation of water, which is clear and has no emulsified layer, but the oil after the test retains about 5 per cent of water. With the other crank-case oil only a very few cubic centimeters of water are separated at the expiration of the test, and a very large amount of emulsified layer of a heavy thick nature is shown; in fact the whole mixture is a heavy emulsion without separation.

The first sample of engine oil shows at the end of the test a thick milky emulsion with practically no separation of water, and the separated oil above the emulsified layer about 4 per cent of water. The second sample of engine oil shows considerable amount of separation of water and much smaller amount of emulsified layer; this layer is of a light foamy nature.

It should be particularly noticed in these two cases that while 100 cc. of oil were used in the tests, 102 cc. of separated turbid oil were found; deducting the amount of moisture or water found in the separated oil, 4 and 3 per cent, respectively, gives 96 and 97 per cent of demulsibility. This is a clear illustration of the importance of giving a complete statement in the report of the behavior of the oil or oils in the emulsifying test, as simply stating the percentage of demulsibility is clearly insufficient, and in cases of this kind would be seriously misleading.

14. HEATING VALUE OR CALORIFIC VALUE OF PETROLEUM PRODUCTS.

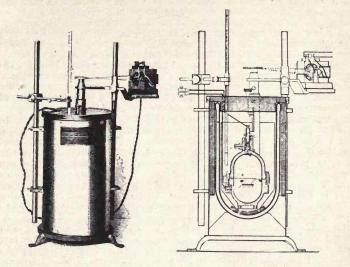
The apparatus used for the heating value, calorific value or British thermal units of petroleum products are shown on pages 331 and 334.

Any type of oxygen bomb calorimeter is satisfactory. Among these are the Atwater, Mahler, Parr and Kroeker bombs. The so-called Parr chemical calorimeter is not satisfactory either in operation or results when applied to oil. The description of the operation of one bomb calorimeter is typical of all.

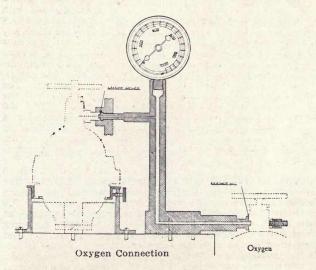
The lower half of the bomb is placed in the cast iron holder. About one gram of the oil is weighed to the nearest 0.0001 gram into the fuel pan and is placed in the bomb on the fuel pan holder. If the oil is volatile it is not advisable to pour the fuel directly into the fuel pan. For this purpose, small gelatine capsules weighing .1 gm. are used and may be filled with ignited asbestos and into this the light oil is discharged from a weighing pipet. The capsule is immediately closed leaving a minimum amount of air space. A similar capsule has been previously weighed and its calorific value determined. A stock of standardized capsules should be kept on hand in an air tight receptacle. The platinum fuse wire is cut equal in length to the taper pin wrench which is connected to the terminal being careful that it does not touch the pan. The wire is bent down so that it is covered by the oil or by the lips of the capsule. The upper half of the bomb is carefully fitted on the lead gasket to the lower half. The nut is screwed down over the upper half being careful not to cross the threads. The bomb nut is now tightened by the use of the long wrench, being careful to cause no sudden jerking or vibrating which will throw the oil from the pan. The bomb is now carefully lifted out and placed on the swivel table and connected with the oxygen piping. The valve in the top of the bomb is opened about one turn and the valve in the oxygen cylinder is carefully and slowly opened so that the pressure in the bomb as shown by the indicator rises to 300 pounds. The bomb valve is now closed and the oxygen cylinder valve is closed. Exactly 1900 grams of water at a tempera-ture of about 4° below room temperature is weighed into the calorimeter water bucket. This is placed in the calorimeter container. bomb is connected with the electric wire and is introduced into the water being careful to place it in the center of the bucket. Two 100 watt lamps placed in parallel are in series with the fuse wire when a 110 volt circuit is used for firing. The stirring motor is placed in series with a 60 watt lamp on a 110 volt circuit. The cover is put on, the connections to the bomb wire are made and the stirrer is introduced as far down as it will go. It should not touch the bomb. The thermometer is introduced and stirring is continued for about 5 minutes. The temperature is read and the stirring continued for exactly 5 minutes and the temperature is again read and the charge is fired by quickly throwing in the switch and withdrawing it. The stirring is continued for 5 minutes, the temperature being read at minute intervals or at the end of 5 minutes unless extreme accuracy is required. The stirrer is then run for an additional 5 minutes and the temperature is again read. The thermometer is corrected in accordance with the corrections furnished by the Bureau of Standards. The radiation corrections may be applied to each one minute interval

330

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Calorimeter Equipped With Vacuum Walled Jacket





Weighing Bottle for Liquid Fuels. etc

but for ordinary purposes 1/5 of the radiation for the 5 minute period before firing is applied on the 5 minute period immediately after firing and 4/5 of the radiation in the third 5 minute period is applied on the 5 minute period immediately after firing. The calorimeter constant (usually about 2400) is determined by a blank test using exactly 1 gram of benzoic acid. This constant always remains the same with the same calorimeter but must be determined each time a change is made in the calorimeter. In the case of oil in which it has been necessary to use the capsule the correction made must be applied for the calorific value of the cassule. This is most conveniently applied to the corrected net rise in temperature of the thermometer. To convert British thermal units per pound to calories per gram, multiply by 5/9. To obtain the water evaporative power, multiply the B. T. U. per pound by 1.035. To obtain the B. T. U. per gallon, multiply the B. T. U. per pound by the weight per gallon.

An approximation of the heating value of fuel oil can be obtained by the following formula:

B. T. U. in lbs. per gallon = 18700 + 40 (°Be'-10).

15-A. SULPHUR FROM THE BOMB CALORIMETER.

The calorimeter is opened by gradually allowing the pressure to diminish and the bomb is carefully and thoroughly washed out with distilled water. The pan is placed in the beaker with the washings and about 10 cc. of hydrochloric acid is added. The contents of the beaker are treated with bromine, heated to boiling temperature for about 10 minutes, filtered and washed and the sulphur in the filtrate precipitated with 10 cc. of barium chloride solution. The precipitated barium sulphate is filtered, washed and weighed in the usual manner. The weight of the barium sulphate \times 13.733 and divided by the weight of the sample gives the percentage of sulphur in the oil.

15-B. SULPHUR BY THE ESCHKA METHOD.

This method is not good for oils, in most instances giving a low result, but may be used where accuracy is not necessary. Weigh out approximately 1 gram of the oil and mix it with 2.5 grams of sodium carbonate and 5 grams of calcined magnesia in a platinum dish or crucible. Heat gradually increasing the temperature until the mass has a low red color and the mixture on cooling has a grayish tint. Cool and wash into a 500 cc. beaker with distilled water and add about 1 cc. of bromine. Mix until the bromine is thoroughly dissolved and allow some time for the bromine to react. Now add hydrochloric acid until the reaction is decidedly acid, the beaker being covered in the meantime to prevent any mechanical loss. Filter off and wash any undissolved residue. Precipitate in the usual manner with barium chloride and weigh as barium sulphate.

15-C. SULPHUR WITH THE PARR CHEMICAL CALORIMETER.

Weigh 0.25 - 0.40 gram of the oil from a weighing pipet into the Parr chemical bomb container along with a mixture of 1 gram of 100 mesh potassium chlorate and 1 full measure, using the cup furnished with the instrument, of pure sodium peroxide. Add 0.2500 gm. of sulphur free lampblack. Immediately close and lock the bomb and be sure that the spring in the plunger valve is strong. Shake thoroughly until the lampblack, the oil, the potassium chlorate and the sodium peroxide are thoroughly mixed. Place the bomb in the calorimeter with the stirring wings adjusted on it and add 2000 cc. of water.

Put the cover on the calorimeter and introduce the thermometer and stir. Heat a hot wire slug about ¼-inch long and when red, quickly introduce into the stem of the calorimeter. As quickly as possible with a quick thrust of the plunger allow it to fall into the bomb. Stir until the temperature ceases to rise. Remove the bomb, open it and place it in a beaker. Pour boiling hot water into it until effervescence ceases. Rinse off the bomb into the beaker and remove the wire. Add hydrochloric acid until acid in reaction.

Filter, wash and to the filtrate add barium chloride. Boil a few minutes, allow to stand hot at least 15 minutes or until the supernatent liquid is clear, filter off the barium sulphate and weigh it in the usual manner. The barium sulphate \times 13.73 divided by the weight of the oil taken is the percentage of sulphur.

16-A. CARBON AND HYDROGEN IN PETROLEUM PRODUCTS.

The most convenient method is to burn the oil in a special calorimeter bomb of the type of the Kroeker (see page 334).

The bomb must be perfectly dried on the inside by drawing dry air through the apparatus.

Approximately one gram of oil is now burned exactly as in the determination of heat of combustion (which see).

The bomb is taken from the calorimeter and is connected on the tube side with Drechsel bottles containing moist soda lime in the first bottle and calcium chloride in the second bottle. The outlet of the bomb is now connected in series with a U tube containing granulated zinc to decompose any acid formed in the combustion, with a glass stoppered U tube filled with calcium chloride of about 10 mesh size, with a glass stoppered U tube filled in the first arm with soda lime containing 10% water and the upper part of the second arm with calcium chloride connected then with an aspirator bottle.

The outlet of the bomb is gradually opened so that at least 10 minutes is required to release all of the pressure.

The bomb is now heated and the aspirator is run at such a rate that about five gallons of air are drawn through the bomb during a period of between one and two hours. The carbon is calculated from the increase in weight of the soda lime U tube and the hydrogren is calculated from the increase in weight of the calcium chloride U tube.

 $\frac{\text{CO}_2 \times 27.273}{\text{weight of sample}} = \% \text{ carbon}$ $\text{H}_2\text{O} \times 11.190$

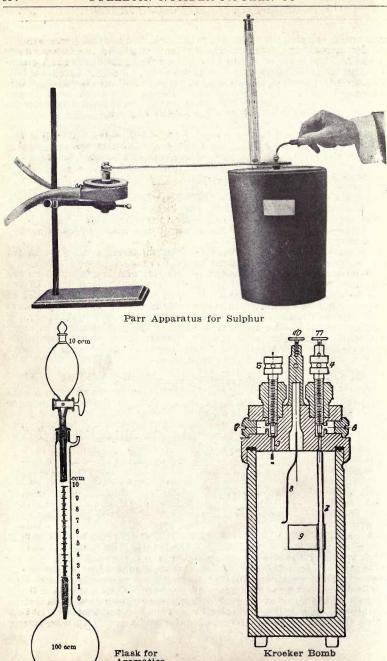
_____ = % hydrogen

weight of sample

16-B. DETERMINATION OF NITROGEN IN PETROLEUM OR ASPHALT. BY THE KJELDAHL METHOD.

5 grams of the sample are weighed into a pyrex Kjedahl digesting flask. 50 cc. of the digestion mixture composed of concentrated sulphuric acid containing 20% of phosphorous pentoxide is added to the flask. About one-third gram of mercuric oxide is added and the contents of the flask are heated with a strong flame until the solution has become pale yellow or colorless. The digested material is now cooled, diluted with about 150 cc. of water and neutralized with strong

BULLETIN NUMBER FIFTEEN OF



KANSAS CITY TESTING LABORATORY

caustic soda solution. Zinc shavings and some Potassium sulphide are added. The flask is quickly connected with the condenser tube and the ammonia is distilled off into a 25 cc. of N/10 sulphuric acid. The excess of acid is titrated with N/10 alkali. Each cubic centimeter of sulphuric acid consumed is equivalent to .001404 gram of nitrogen.

Reagent.

17. DOCTOR TEST FOR GASOLINE.

Solium plumbite or "doctor" solution—Dissolve 125 grams of sodium hydroxide (NaOH) in a liter of distilled water. Add 70 grams of litharge (PbO) and shake vigorously for 15 or 30 minutes or let stand with occasional shaking for at least a day. Allow to settle and decant off the clear liquid. Filtration through a mat of asbestos may be employed if the solution does not settle clear. The solution should be kept in a bottle tightly stoppered. Test.

Shake vigorously for about 15 seconds two volumes of gasoline and one volume of the "doctor" solution. Note color. A small pinch of flowers of sulphur should be added and the tube again shaken for 15 seconds and allowed to settle. The quantity of sulphur used should be such that practically all of the sulphur floats on the surface, separating the gasoline from the "doctor" solution. Interpretation.

If the gasoline is discolored or if the sulphur film is so dark that its yellow color is noticeably masked, the test shall be reported as positive, and the gasoline condemned as "sour." If the liquid remains unchanged in color and if the sulphur film is bright yellow, or only slightly discolored with gray or flecked with black, the test shall be reported negative and the gasoline considered "sweet."

18-A. OLEFINS OR UNSATURATED HYDROCARBONS AND REFINING LOSS IN PETROLEUM PRODUCTS.

Use apparatus and equipment as shown on page 337. Method using a Babcock cream bottle.

Weigh up a clean and dry 30% Babcock cream bottle, add to it exactly 5 cc. of the oil to be tested. Weigh again giving the amount of oil used. Cool in ice water and add 10 cc. of concentrated commercial sulphuric acid, letting the acid run down the sides of the bottle. Shake while cooling in the ice water. Keep stoppered with a rubber stopper. Let stand for $\frac{1}{2}$ hour with occasional shaking and constant cooling. Add sufficient concentrated sulphuric acid (commercial) to bring the reading about to the top of the scale on the neck of the bottle. Centrifuge for five minutes in the No. 1 centrifuge with the resistance at the first notch from the left. This gives a speed of 1000 R. P. M. Keep the rubber stopper in while centrifuging so that there will be no evaporation. The stopper shall be large enough so that it is not forced into the bottle.

The reading on the neck of the bottle divided by 5 is the net amount of saturated hydrocarbons contained. This multiplied by 20 and take from 100 gives the per cent of unsaturated hydrocarbons. For great accuracy the oil may be corrected for specific gravity and temperature and for the amount adhering to the sides of the pipet in which case the weighings are used. The waste acid from the Babcock bottle is poured into a bottle from which the sulphuric acid may be recovered by separating the oil and oxidising the organic material in the acid.

18-B. METHOD USING A 10 CC. GLASS STOPPERED CYLINDER. (Egloff.)

Use apparatus and equipment as shown on page 337.

Add exactly 5 cc. of the oil to be tested to the cylinder and 2 cc. of sulphuric acid of gravity 1.84. Shake thoroughly for about 5 minutes and place in centrifuge and centrifuge at the rate of 1000 R. P. M. for 5 minutes. The shrinkage of the oil in cubic centimeters \times 20 is the percentage of olefins.

18-C. REFINING LOSS OF PETROLEUM PRODUCTS.

Use the color tube as shown on page 281.

To a 50 cc. color tube that is graduated in .1 cc. and glass stoppered, add 45.0 cc. of the oil. Add exactly 1 cc. of 66° Baume' sulphuric acid. Shake thoroughly for about 5 minutes. Set vertically in a rack for at least one hour and preferably over night. The increase in volume of the acid in the bottom of the tube x 2-2/9 is the refining loss.

19-A. METHOD FOR DETERMINING AROMATIC AND PARAF-FIN HYDROCARBONS IN PETROLEUM PRODUCTS.

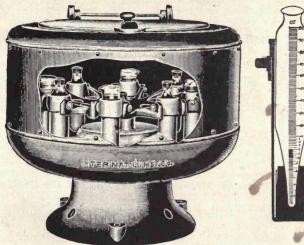
The apparatus is shown in the figure on page 334. The flask containing 30 cc. of fuming nitric acid (specific gravity 1.52) is cooled to -10° C by a salt ice freezing mixture. The separatory funnel is filled to the 10 cc. mark with the oil under test. The oil is rr drop by drop with continuous shaking into the cooled acid during period of not less than 45 minutes. With uncracked petroleun products 15 minutes is sufficient. The mixture is allowed to stand 15 minutes after completion of the reaction and then enough nitriacid (ordinary concentrated) at -10° temperature is added to the contents of the flask until the oil under the surface is brought into the graduated neck. The volume is read when the neck is at room temperature, the body of the flask being in the freezing mixture. This volume represents the paraffin hydrocarbons.

The mixture is transferred to a separatory funnel, the lower layer run off into a 500 cc. measuring flask containing 150 cc. of water. The neck should be graduated for a 10 cc. portion into 1/10th cc. The temperature will rise in proportion to the amount of olefins and aromatics present and more or less oil will separate according to the amount of paraffin hydrocarbons present.

to the amount of paraffin hydrocarbons present. The unattacked oily layer in the separatory funnel is washed with water and then examined for specific gravity and boiling point. The aqueous layer of nitric acid is warmed for 15 minutes to dissolve as completely as possible the resinous substances formed. The cooled liquid is shaken with 100 cc. of ether, the aqueous layer separated and the ether layer again washed free from acid with water, then with a solution of caustic potash containing 50 grams of KCH in 500 cc. of water with 50 cc. of alcohol.

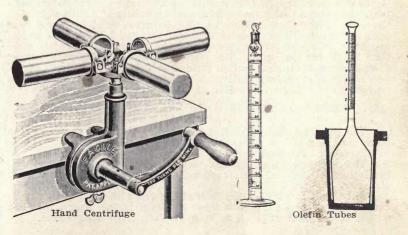
The caustic potash is drawn off and again the ether layer is washed with water. It is now dried with calcium chloride, filtered, the ether evaporated and the residue weighed. The residue consists of reddish brown oil, aromatic nitro-derivatives. The weight divided by 1.15 gives the percentage of aromatic hydrocarbons.

KANSAS CITY TESTING LABORATORY



Good type of electric centrifuge.

Centrifuge tube for water and "B.S. or sediment.



The difference between the aromatic and cyclic hydrocarbons and the paraffin hydrocarbons and 100% is the amount of olefins. This may be checked by direct determination as shown under olefins.

19-B. SHORT METHOD FOR AROMATIC AND CYCLIC HYDRO-CARBONS.

Distillation of 800 cc. of the hydrocarbons under examination may be made in a 1 liter distilling flask in accordance with the appar-atus set forth on page 310. Cuts may be made at 95°, 120° and 150°C and the percentage of aromatic compounds calculated from the specific gravity using the following specific gravities as the basis:

of Aromatic

0.880

0.871

Specific Gravity Specific Gravity of

Hydrocarbon Hydrocarbon

Non-Aromatic

0.720 0.730

0.760

Temperature of Cut 95°C 120°C 150°C

0.869 This is in accordance with the Bulletin No. 114 of the Bureau of Mines, page 95.

20. FREE ACID IN LUBRICATING AND OTHER PETROLEUM OILS.

Weigh 10 grams of the oil into an Erlenmeyer flask. Add 50 cc. of 95% denatured alcohol which has been previously neutralized with dilute caustic soda. Heat over a gauze to the boiling point. Shake thoroughly to dissolve the acid. Titrate while hot with N/10 caustic soda using phenolphthalein as indicator shaking thoroughly after each addition of NaOH, continuing to the first persistent pink color. Express results in terms of oleic acid. 1 cc. of N/10 NaOH = 0.0292gm. of oleic acid.

21. FLOC TEST.

Take a hemispherical iron dish and place a small layer of sand in the bottom. Take a 500 cc. Florence or Erlenmeyer flask and into it put 300 cc. of the oil (after filtering if it contains suspended mat-ter). Suspend a thermometer in the oil by means of a cork slotted on the side. Place flask containing the oil in the sand bath and heat bath so that the oil has reached a temperature of $240^{\circ}F$ at the end of one hour. Hold oil at temperature of not less than $240^{\circ}F$ nor more than 250°F for six hours. The oil may become discolored but there should be no suspended matter formed in the oil. The flask should be given a slight rotary motion and if there is a trace of floc, it can be seen to rise from the center of the bottom.

22. CORROSION AND GUMMING TEST OF GASOLINE AND NAPHTHA.

The gasoline when subjected to the corrosion test shall show no black corrosion and no weighable amount of gum.

Directions for making test:

The apparatus used in this test consists of a freshly polished hemispherical dish of spun copper, approximately 31/2 inches in diameter.

Fill this dish within % the inch of the top with the gasoline to be examined and place the dish upon a steam bath. Leave the dish on the steam bath until all volatile portions have disappeared. If the gasoline contains any dissolved elementary sulphur the bottom of the dish will be blackened.

If the gasoline contains undesirable gum-forming constituents there will be a weighable amount of gum deposited on the dish. Acid residues will show as gum in this test.

23. PENETRATION OF PETROLEUM ASPHALTS AND OTHER BITUMINOUS MATERIALS.

The apparatus used for this test is that shown on page 340.

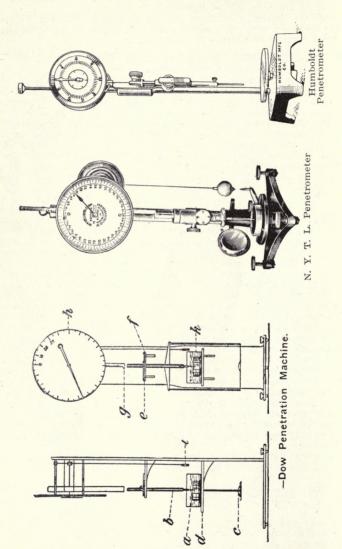
The penetration is the consistency of a bituminous material expressed as the distance that a standard needle vertically penetrates a sample of the material under known conditions of loading, time and temperature. When the conditions of test are not specifically mentioned the load, time and temperature are understood to be 100 grams, 5 seconds, 25° C (77°F) respectively and the units of penetration indicate hundredths of a centimeter. The container for holding the material to be tested should be a flat bottomed cylindrical dish 23/16 inches in diameter and 1% inches deep or the American Can Co. Gill style ointment box, deep pattern, 3 ounce capacity.

The needle is a cylindrical steel rod 2 inches long and with a diameter of 0.04 inch and turned on one end to a sharp point having a taper of $\frac{1}{4}$ -inch. The bath for the sample and the penetrometer should hold at least 10 liters of water. The sample should be melted at the lowest possible temperature and stirred until it is homogeneous and free from air bubbles. It is then poured into the sample container to a depth of about $\frac{3}{4}$ -inch and is allowed to cool for one hour in the air. It is now placed in the water bath maintained within 0.1°C of the temperature of penetration for one hour.

In making the test, the sample is immersed in water and the needle loaded with the specified weight is adjusted to make contact with the surface of the sample. This may be accomplished by making contact of the actual needle point with its image reflected by the surface of the sample or contact may be meted by slightly turning the container so that a faint scratch on the surface of the bitumen is observed. The needle is then released for the specified time and the distance measured by the means provided with the machine. At least three tests shall be made at different points on the surface of the sample and after each test the needle shall be wiped clean of all bituminous matter. The reported penetration is the average of at least three tests whose values do not differ more than four points between the maximum and minimum. Other conditions for penetrations particularly for oil asphalt filler and roofing material shall be the following:

At 0°C (32°F) 200 grams weight 60 seconds.

At 46.1°C (115°F) 50 gram weight 5 seconds.



340

24. DUCTILITY OF SEMI-SOLID BITUMINOUS MATERIALS. The apparatus used for this test is shown on page 342.

It consists of a machine for uniformly pulling a cylinder or briquet of the asphaltic cement at a slow rate of speed until it breaks. Two types of molds for the sample are used, one is a square mold having a cross section of 1 sq. cm. at its narrowest point, the other is a round mold separated by a cylindrical section of 1 cm. diameter. Either mold is used on the same machine.

The asphalt is carefully melted as in the penetration test and the mold is completely filled so that it bulges at the top enough to allow for shrinkage. The brass is amalgamated with mercury to prevent sticking. The temperature at which the test is made should correspond to that temperature at which the sample being tested has a penetration of 50°. The rate of pulling may be either the slow rate of 5 cm. per minute or the fast rate of 60 cm. per minute, the machine being adapted for either speed. The pulling is continued until the thread of asphalt breaks. The distance that it has been drawn out without breaking, expressed in centimeters is the ductility. By the square mold ductility a result is obtained which is about 3½ times that of the round ductility.



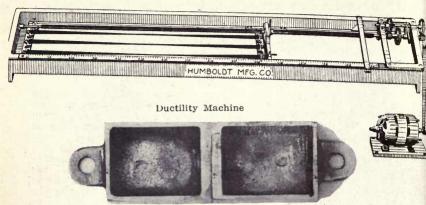
25. LOSS ON HEATING OF OIL AND ASPHALTIC COMPOUNDS.

The loss in weight by oil and asphaltic compounds when they are heated in an oven at a temperature of 163° C (325° F) is determined on 50 grams of the water free substance contained in a flat bottomed dish, the inside dimensions of which are approximately 23/16 inches in diameter and 1% inches deep (this is the 3 ounce Gill style ointment box, deep pattern).

The oven in which the substance is to be heated is brought to temperature before the sample is introduced and the temperature of the sample under test shall be regarded as that of a similar quantity of the same material immediately adjoining it in the oven in which the bulb of a standardized thermometer is immersed. The oven may be any well constructed type either circular or rectangular and the source of heat may be either gas or electricity. The samples under test rest in the same relative position in a single row upon a perforated shelf 9.75 inches in diameter as shown on page 343. A good type of oven is also shown on same page. The shelf is suspended by a vertical shaft midway in the oven which is revolved by mechanical means at the rate of from 5 to 6 R. P. M.

26. ASPHALT IN OIL AND ASPHALTIC COMPOUNDS.

50 grams of the crude oil, fuel oil, lubricating oil, road oil or other material are weighed into a 3 ounce Gill style ointment box, deep pattern, and placed in an oven heated either by electricity or gas and with good circulation to a temperature of approximately 500° F. Heat is maintained until the consistency of the residue is such that at a temperature of 77° F it has a penetration of 100. The amount of asphalt is reported in terms of the 100° penetration material.



Round Ductility Mold

27-A. ASPHALTENES IN OIL AND ASPHALTIC PRODUCTS (SOLUBILITY IN PETROLEUM NAPHTHA).

Weigh up one gram of the oil or asphaltic material into a 150 cc. Erlenmeyer flask. Add exactly 100 cc. of paraffin base petroleum naphtha of a gravity of $38^{\circ}Be'$. Shake until the material is well disintegrated and allow to stand for at least 5 hours and preferably over night, the flask being tightly corked. Prepare a Gooch crucible having a uniform layer of asbestos fiber in the bottom about $\frac{1}{2}$ -inch thick. Dry and ignite the prepared crucible. Place the crucible in a vacuum filter as shown in the figure on page 346 and pour the petroleum naphtha solution of the material through it. Drain the Erlenmeyer flask as thoroughly as possible and rinse out with 100 cc. of $88^{\circ}Be'$ petroleum naphtha so that the last of the naphtha is not stained with the bituminous material. Care must be taken to prevent undue disturbance of the asbestos mat. Draw air through the residue on the Gooch crucible for a minute with a suction pump and place it in the drying oven at 105°C for $\frac{1}{2}$ hour. Weigh, ignite and weigh again. The difference between the two weighings is the asphaltene. This taken from the original may also be recorded as the solubility in petroleum ether.

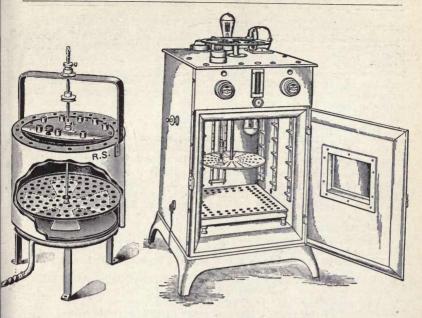
27-B. SOLUBILITY IN CARBON BISULPHIDE. (TOTAL BITUMEN.)

This test is performed in the same way as for asphaltenes or solubility in petroleum naphtha except that a 5-gram sample is preferably used. The same apparatus is used.

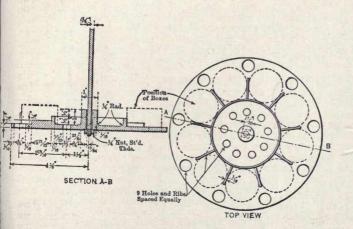
27-C.—SOLUBILITY IN CARBON TETRACHLORIDE.

This test is performed in exactly the same manner as with carbon bisulphide except that the flask containing the carbon tetrachloride must be kept in a dark place. The difference between the solubility in carbon bisulphide and carbon tetrachloride represents the CARBENES.

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Cas oven Electric oven Electric oven beating at 325°F.



Detail of shelf used in loss test.

343

28. RESISTANCE OF ASPHALTIC CEMENT TO OXIDATION.

After being subjected to the following tests the film of asphalt should be brilliant and lustrous, should not be scaly and fragile, should adhere firmly to the metal and should not be dull and cheesy in texture.

A strip of thin sheet iron 2 inches wide and 6 inches long is covered on its lower 4 inches with the melted asphaltic cement. This strip is placed in an oven at 275°F for 15 minutes and allowed to thoroughly drain.

It is removed from the oven and allowed to cool, then placed in an electrically heated oven at a temperature of 450°F for one hour. At the end of the hour, the door of the oven is opened and the heat is turned off, the specimen being allowed to remain in the oven.

The oven shall be one having an outside diameter of 12x12x12 inches with an opening in the top 1 cm. in diameter, the heating elements being in the bottom of the oven. The resistance shall be so distributed that the heat is uniform throughout the oven. The lower end of the strip shall be suspended so that it is at least 3 cm. from the bottom of the oven.

The resistance is preferably so arranged that three different heats can be maintained with a snap switch such that the lowest heat is 325° F, the medium heat is 400° F and the highest heat is 450° F.

29. PARAFFIN WAX OR SCALE IN PETROLEUM AND BITU-MINOUS PRODUCTS.

The apparatus used is shown on page 347.

Instead of the metal retort, a glass distilling flask with a glass air condenser may be used if desired. 100 grams of the oil, bitumen or material under examination are weighed into the retort and distilled as rapidly as possible to dry coke. The distillate is caught in a 150 cc. Erlenmeyer flask, the weight of which has been previously ascertained. During the early stages of distillation a cold, damp towel wrapped around the stem of the retort will serve to condense the distillate. After high temperatures have been reached, this towel may be removed. When the distillation is completed, the distillate is allowed to cool to room temperature and is then weighed in the flask. This weight minus that of the flask gives the weight of the total distillate.

Five grams of the well mixed distillate is then weighed into a 100 cc. Erlenmeyer flask and mixed with 25 cc. of Squibb's ether. 25 cc. of Squibb's absolute alcohol is then added, after which the flask is packed closely in a freezing mixture of finely crushed ice and salt maintained at -18° C in a quart tin cup. After remaining 30 minutes in this mixture, the solution is quickly filtered through a No. 575 C. S. & S. 9 cm. hardened filter paper placed in a glass funnel which is packed in a freezing mixture as shown in figure. Vacuum should be employed to hasten filtration. The freezing-mixture reservoir shown in the figure may be made by cutting in half a round glass bottle measuring approximately 120 millimeters in diameter and using the upper half in an inverted position. Any precipitate remaining on the paper should be washed until free from oil with about 50 cc. of a 1 to 1 mixture of Squibb's ether and absolute alcohol cooled to -18° C.

After the paper has been sucked dry, it should be removed from the funnel and the adhering paraffin scale should be scraped off into a weighed crystallizing dish and dried on a steam bath. The dish and contents should then be cooled in a dessicator and weighed. The weight of the paraffin scale so obtained, divided by the

weight of the distillate taken and multiplied by the percentage of the total distillate obtained from the original sample, equals the percentage of the paraffin scale.

30-A. BITUMEN AND GRADING OF ASPHALT SURFACE MIXTURE.

The asphaltic surface is softened by warming and is thoroughly mixed. 100.0 grams are weighed into a thin porcelain dish. This is placed in a gas or electric muffle, as shown on page 348. and heated with good aeration at a temperature not exceeding 700°C. preferably about 500°C, or at a barely perceptible red heat. It is well to use a pyrometer in the muffle. Usually about two

hours is required for the complete combustion of the carbonaceous material. The dish and contents are now removed from the muffle, allowed to cool and weighed. The loss in weight is the percentage of bitumen. The mineral matter is now screened through a nest of screens containing the 1, 2, 4, 10, 20, 40, 80, 200 meshes to the lineal inch. The amount passing each screen and retained on the next is recorded. The exact description of the sizes is as follows:

	Opening in	Opening in	Diameter of
Mesh	Inches	Millimeters	Wire, Inch
1	1.050	26.67	0.149
2	0.525	13.33	0.105
4	0.1850	4.699	0.065
10	0.0650	1.651	0.035
20	0.0340	0.864	0.016
40	0.0150	0.381	0.010
80	0.0068	0.173	0.00575
200	0.0029	0.074	0.0021

BITUMEN AND GRADING OF ASPHALTIC SURFACE 30-B. MIXTURE BY EXTRACTION.

The bituminous mixture should be warmed until it may be readily

broken apart by hand, without fracturing any of the stony particles. Five hundred grams of the disintegrated mixture should be packed as tightly as possible in the wire basket and then covered with a disc of cotton or felt of one-quarter inch to one-half inch thickness.

One hundred and seventy-five to two hundred cc. of carbon disulphide, carbon tetrachloride, chloroform, or benzole is placed in the inside vessel in which the wire basket is suspended.

Cool water should be circulated through the inverted cone condenser, which is also the cover of the apparatus, and is not intended to fit tight.

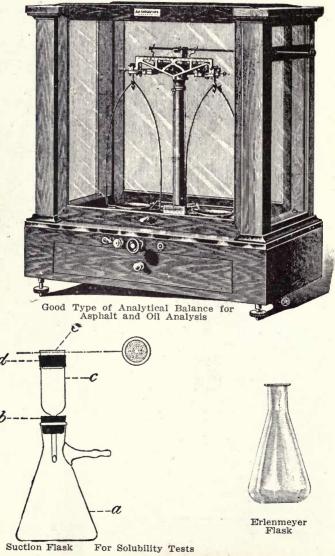
A 16 c. p. carbon filament incandescent lamp is the source of heat.

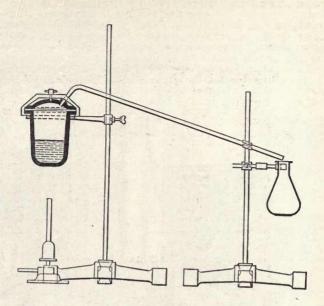
A 500-gram sample of mixture should extract clean with carbon disulphide in about three hours. From 200 to 300 grams of asphalt block or Topeka type mixture is a sufficiently large sample for that type of mixture.

After extraction, the solvent and matter removed from the sample during the analysis should be burnt to recover any fine mineral particles which may have passed into the extract.

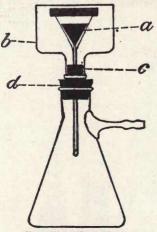
This method has the advantage of giving the true soluble bitumen

and of leaving the mineral matter in such condition that it is more easily screened. However, it has the disadvantage of requiring a longer time, a considerable amount of solvent and of giving slightly higher results in percentage of bitumen unless the extracted matter is burned out. Extraction may also be made by the Rotarex centrifuge.





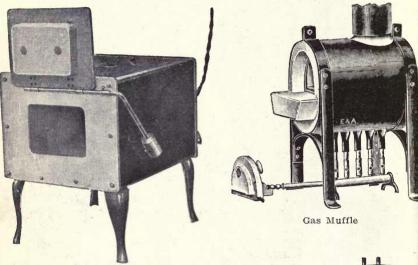
Paraffin Scale Distillation



Paraffin Filter Flask

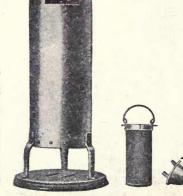
31. TENSILE STRENGTH OF BITUMINOUS SURFACE MIXTURE.

The surface mixture to be tested is heated to over 240°F to soften it and is thoroughly compressed into a standard cement testing briquet mold. The mold is then packed in ice for at least two hours. It is now quickly put in the tensile strength machine used for testing portland cement and pulled until it fails. Good bituminous surface mixture will give a tensile strength of as high as 600 lbs. per sq. in. Poorly cemented material will give a tensile strength usually lower than 200 lb. per sq. in.



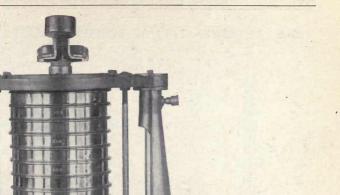
Electric Muffle

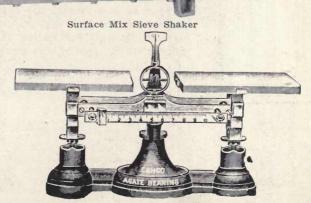




N. Y. T. L. Surface Mix Extractor.

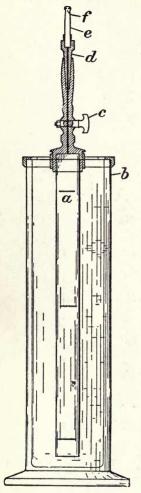
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Mineral Aggregate Grading Balance

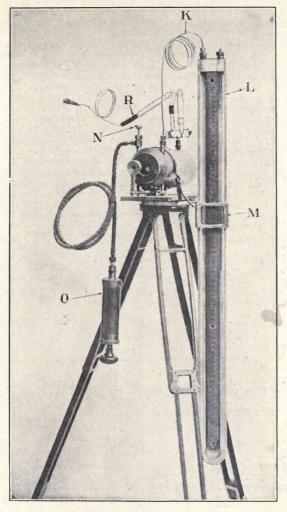
32-A. DETERMINATION OF SPECIFIC GRAVITY OF GAS.



The apparatus consists of a glass jar, b. with a metal top into which fits a brass column having suspended from its base a long, graduated tube, a, and at its top a cock. c. and a ground-joint socket d. into which sets a socket holding a small glass tip, e, closed at the top with a thin piece of platinum, f. In this platinum is a minute hole to permit the passage of gas or air at a very slow rate. All the metal parts are nickeled. The mode of operation is as follows: The glass jar is filled with water to the top graduation mark of the tube or to a point a little above it. The tube is then withdrawn so that it may be filled with air. The cock on the standard is then closed and the tube is replaced with air. The cock is then opened, and the number of seconds required for the water to pass from the lowest graduation mark to the graduation mark above it is recorded with a stop watch. The tube is then withdrawn and filled with gas and the procedure repeated. The specific gravity (air=1) is obtained by dividing the gas time squared by the air time squared. Thus, if A represents the time required for the gas to pass through the orifice, and B represents the time required for the air to pass through the orifice, the specific gravity of the gas will be represented by

 $\left\{\frac{A}{B}\right\}^2$

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Edwards Gas Balance

32-B. SPECIFIC GRAVITY OF GAS BY THE EDWARDS GAS BALANCE.

Above, the figure shows the Edwards Gas Balance completely assembled with mercury manometer "L" at the right in the foreground, hand pump "O" at the left for evacuating the balance chamber, and connection "R" to the gas sample by means of the stop cock on the back end of the balance chamber. On page 353 is shown the balance beam consisting of an air tight bulb of spun brass, counter-weighted with adjustable balancing weights. The bearing points are also adjustable, allowing the center of gravity of the beam to be raised or lowered, thus providing a control of the sensibility. The needle points rest on glass bearings.

The beam is adjusted so that it will come to equilibrium in atmosphere with the counter-weight end slightly below a horizontal plane through the bearing points. In this position a partial vacuum is required to bring it to a level position which position is effected by bringing into alignment the cross hair mounted permanently on glass and the line on the end of the balance beam. The air that is allowed into the chamber when making this balance must be drawn through some drying agent assuring dry air. The vacuum reading is then observed on the "U" gauge. This should be repeated and checked. The balancing chamber is then purged of air and the gas allowed to fill it to a pressure sufficient to bring the beam to the same position of equilibrium again. The pressure is then observed on the "U" gauge. These pressures are then reduced to absolute pressure, knowing the barometric pressure at the time of making the test. The specific gravity of the gas is the quotient of the absolute air pressure divided by the absolute gas pressure. (Air=1.000.)

a = Barometric pressure.

b = Balancing pressure air.

 $\mathbf{c} =$ Balancing pressure gas.

a-b

Specific gravity = ---

When air is present in gas it is determined with an Orsat apparatus, or other convenient apparatus. The correction of the observed specific gravity to the actual specific gravity is made with following formula:

- $g = \frac{100 \text{ d} a}{100 \text{ d} a}$
- $g = \frac{100}{100} a$

a = % air.

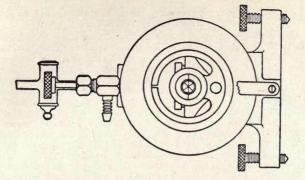
d = determined or observed specific gravity.

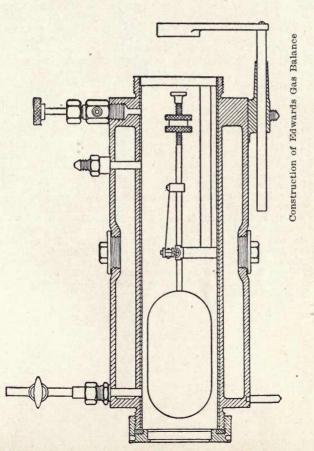
g = actual specific gravity.

Example:---

Barometric pressure (millimeters)		756.4
Barometric pressure (millimeters)	396.7	
0 0	390.3	
Pressure (milimeters)		-6.4
Total pressure (millimeters		750.0
Gage readings with gas (millimeters)	189.3	
0	597.4	- 3
Pressure (millimeters)		+408.1
Total Pressure (millimeters)		1164.5
Specific gravity	-750.0	=0.6441
	1164.5	

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33-A. ABSORPTION METHOD FOR TESTING NATURAL AND CASINGHEAD GAS.

Fill the two-armed pipet commonly known as the Hofman apparatus with distilled water. The glass stop cock at the top of the closed graduated arm is a two-way cock, so that the tube above the stop cock can be completely cleared of air. The end of the stop cock through which the outside discharge takes place is closed with a rubber tube and pinch cock. A funnel is set on top of the tube, water is introduced and the tube is washed out with distilled water. The pinch cock is closed, the funnel is removed and the gas is introduced in the usual manner by displacement with water until about 50 cc. are in the graduated arm. The level of the water is made the same in the two arms and the reading of the quantity of gas is made after it has adjusted itself to the room temperature.

25 cc. of Claroline oil or straw oil are introduced into the open arm. The open arm is now stoppered or held with the thumb so that no air can gain access and the oil is shaken over into the other arm so that it overlies the water. The water is now withdrawn through the stop cock at the lower end of the U. The arm is now filled and kept filled with Claroline or straw oil shaking until the gas ceases to be absorbed. The absorption is calculated in percentage.

The amount of gasoline that may be obtained by absorption from the gas may be approximately calculated from the following table:

Casinghead Gas Yield.

		Yield of Gasoline
Absorption		Gallons per 1000
Percentage		Gallons per 1000 Cu. Ft. of Gas
25	 	

One gallon of gasoline obtained from 1000 cu. ft. of gas reduces the volume about 25 to 30 cu. ft. and reduces the heating value about 75 to 100 B. T. U. per cu. ft. or $7\frac{1}{2}$ to 10%. One gallon of gasoline at 20c a gallon would then extract .6c from the value of gas at 20c per 1000 cu. ft. About one-half of the natural gas of the United States contains gasoline in commercially obtainable quantity. Some casinghead gas such as at Sisterville, West Va., gives 13 gallons of gasoline per 1000 cu. ft. and has a heating value of 2500 B. T. U. per cu. ft. Shellac is the best thread dressing material for gasoline and oil joints since it is not soluble in gasoline nor water.

354

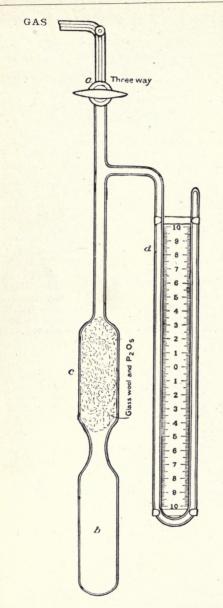
33-B. FREEZING METHOD FOR TESTING NATURAL GAS FOR GASOLINE CONTENT.

This method is from Technical Paper 104, Bureau of Mines, page 26. The sample of natural gas or casinghead gas is introduced in the usual manner into the apparatus shown on page 356.

In this apparatus (a) is a three-way stop cock, (c) is a tube filled with glass wool and phosphorus pentoxide for the purpose of drying, (b) is a portion of tube which is introduced into liquid air, (d) is a manometer tube containing mercury and is closed at the further end.

In filling the manometer, the apparatus must be completely exhausted of its air. Sufficient mercury is introduced so that its level rests at the zero point of the scale when under a vacuum. The three-way stop cock at (a) connects to the vacuum pump and to the gas sample container. The sample of gas is drawn in at ordinary atmospheric pressure and the stop cock (a) is closed and the bulb (b) is introduced into the cooling medium. The temperature below 100°C is taken. At this temperature all of the gasoline constituents are com-While maintained at this low temperature, the pletely liquefied. vapor above the liquefied gasoline is exhausted with the vacuum pump thus removing the non-condensible gas. The bulb is now taken out of the refrigerant and allowed to warm up to the temperature at the beginning of the test. The mercury level in the manometer is read. the pressure indicated being the partial pressure of the gasoline in the sample before the dry gas had been removed. The percentage by volume of gasoline vapor is 100 a, a being the partial pressure of the gasoline vapor after the b test, b being the original atmospheric pressure of the sample. The percentage of gasoline vapor gives the number of pints of gasoline that may be expected in the manufacture of gasoline from the gas under test by the absorption process.

BULLETIN NUMBER FIFTEEN OF



Freezing Method for Gasoline in Gas.

356

34. COMPLETE ANALYSIS OF GAS.

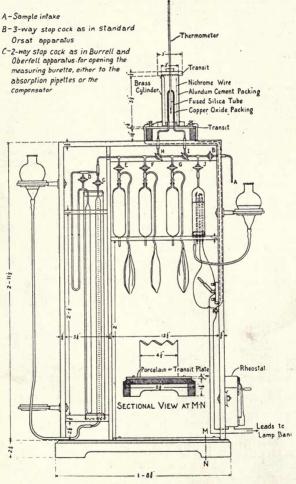
This apparatus is that described in the Journal of Industrial & Engineering Chemistry by G. A. Burrell and G. G. Oberfell, Vol. 8, page 229.

It is designed for the analysis of a gas mixture containing carbon dioxide, unsaturated hydrocarbons, principally ethylene, oxygen, carbon monoxide, methane, ethane, hydrogen and nitrogen.

In the analysis the capillary train and U tube are swept free of gases by drawing a sample of air into the buret and passing it into the alkaline pyrogallate pipet G to remove oxygen. The residual nitrogen is then passed into all the pipets and through the CuO tube to sweep out other gases that may have been contained therein. The electric current is now turned on the electric heating oven, the temperature having been established by previous experiments. About a 100 watt furnace is required. The temperature desired is between 275 and 300°C. Some of the gas mixture is now drawn into the buret, measured and passed into the pipets E, F and G for the removal respectively of carbon dioxide, illuminants, and oxygen. After these constitutents have been removed the stop cocks H, I and J are turned so that communication is made between the buret and the pipet cor-responding to J and through the CuO tube. The gas mixture is passed back and forth through the tube furnace until no further diminution in volume is noted by reading the gas volume in the buret. 15 minutes is usually required, the carbon monoxide being converted to carbon dioxide and the hydrogen to H_2O . The CO burns more rapidly if any hydrogen is present. When the gas is cooled and no further contraction takes place the remaining volume is read in the The carbon dioxide is now removed by placing the gas mixhuret. ture into the KOH pipet E. After the hydrogen and carbon monoxide have been determined the residual gas is placed in the KOH pipet for storage and the stop cock is closed. Enough oxygen to burn the paraffin hydrocarbons is then drawn into the buret, measured and passed into the slow combustion pipet J and the platinum spiral is heated to almost white heat. The residual gas is now withdrawn from the pipet E into the buret and from there slowly passed at the rate of not more than 10 cc. per minute into the pipet J. While operating it is well to cover the slow combustion pipet with gauze as occasionally if the gas is passed in too rapidly an explosion takes place. After combustion is complete, the contraction and the carbon dioxide are measured and the gas again passed into the slow combustion pipet and burned again. A small amount of further contraction may take place but may be ignored unless excessive.

For calculation of results the following example and formulae are useful:

(Continued on page 359.)



Burrell-Orsat Apparatus.

	Analysis of Gas From Pressure	Stills.
a.	Volume of sample taken	44.1 cc
	Volume after KOH absorption	44.0 cc
c.	C 1 D1 11 CO	0.1 cc = 0.22%
	Volume after Br ₂ or Oleum absorption	39.4 cc
e.		4.6 cc = 10.43%
	Volume after alkaline pyrogallate	
~.	absorption	39.3 cc
ø.		0.1 cc = 0.22%
h.	Volume after burning in CuO	35.2 cc
i	Hydrogen, H.	4.1 cc = 9.30%
i	Volume after absorption in KOH	35.0 cc
k	Hydrogen, H ₂ Volume after absorption in KOH Carbon Monoxide CO	0.2 cc = 0.45%
1.	Volume taken for slow combustion	17.5 cc
m.		75.6 cc
n.	Total volume	93.1 cc
0.		61.5 cc
p.	Contraction from burning	32.6 cc
		45.0 cc
q. r.	Contraction from CO ₂	16.5 cc
	Methane in sample	16.0 cc = 72.56%
	Ethane in sample	0.3 cc = 1.36%
u.	Nitrogen in sample	1.2 cc = 5.46%

To calculate amount of methane in the sample from the contraction from burning, "p," and the absorption with KOH, "r," use the following formulae:

Methane (s)	$=\frac{4p-5r}{3}$
Ethane (t)	$=$ $\frac{4r-2p}{3}$
or to obtain % in	original gas
	100 js
% Methane	=al
X Sector Design	100 jt
% Ethane	100 jt
70 Homane	al
	100 ju
% Nitrogen	=

REAGENTS USED IN GAS ANALYSIS.

(1) Potassium Hydroxide.

(a) For carbon dioxide determination.

500 grams of commercial potassium hydroxide are dissolved in 1 liter of distilled water. 1 cc. of this solution absorbs 40 cc. of CO. (b) For the preparation of potassium pyrogallate for oxygen

testing.

120 grams of potassium hydrate are dissolved in 100 cc. of water. 5 grams of crystalline pyrogallic acid are used with 100 cc. of this solution.

(2) Potassium Pyrogallate.

This solution is prepared when used except for charging absorption pipet. Five grams mixed with 100 cc. of potassium hydrate (b) gives a solution in which 1 cc. absorbs 2 cc. of oxygen.

(3) Sodium Hydroxide.

One hundred grams are dissolved in 300 grams of water and may be used instead of potassium hydrate where given above.

(4) Cuprous Chloride.

Method of preparation is to place a layer of copper oxide about % inch deep in the bottom of a two-liter acid bottle. Add an excess of long pieces of heavy copper wire reaching from the top to the bottom of the bottle and fill the bottle with hydrochloric acid of about 1.10 specific gravity. The absorption capacity of this reagent is 4 cc. of carbon monoxide CO for each 1 cc. of reagent. Metallic copper must always be maintained with the reagent to keep it in good condition.

(5) Ammoniacal Cuprous Chloride.

The acid cuprous chloride as prepared above is treated with ammonia until a faint odor of ammonia is perceptible. Likewise an excess of copper wire is maintained. The absorption capacity is 1 cc. of CO to 1 cc. of reagent.

(6) Sodium Hypobromite.

This is made of two solutions, one containing 100 grams of caustic soda with 250 cc. of distilled water, making 284 cc. of solution. The other, 25 grams of liquid bromine, 25 grams of potassium bromine and 200 cc. of water. The two solutions are not mixed until ready to use when equal parts are mixed. This reagent is very good for the determination of illuminants.

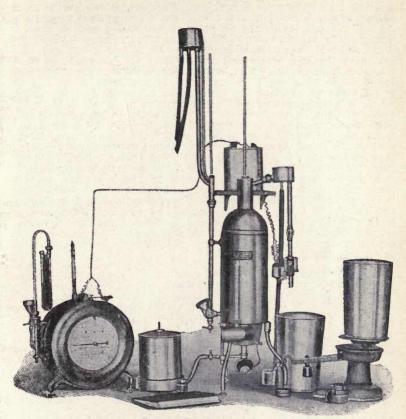
(7) Fuming Sulphuric Acid.

Ordinary concentrated sulphuric acid is mixed with an equal weight of sulphuric anhydride. One cc. of this reagent absorbs 8 cc. of olefins or illuminants.

(8) Palladium Chloride.

Five grams of palladium wire are dissolved in a solution of 30 cc. of hydrochloric acid and 2 cc. of nitric acid.

The solution is evaporated to dryness on a water bath, 5 cc. of hydrochloric acid are added and 25 cc. of water and complete solution is made. The solution is diluted to 750 cc. It contains one per cent palladous chloride and 1 cc. absorbs 2/3 of 1 cc. of hydrogen.



Sargent Gas Calorimeter.

35A. DETERMINATION OF THE HEATING VALUE OF GAS. The most common instruments used for determining heating value of gas are the Junker, the Parr and the Sargent calorimeters. The Sargent is a very convenient instrument and is described as follows:

The figure on page 363 shows a section of the calorimeter body in which the inlet body has a constant heat at the wier A, the temperature of which is taken at B, passes down the tube C and enters the

calorimeter at D. The quantity of water admitted is regulated by the graduated cock between C and D. When water reached D it is spread by the baffle plates E and F and flows upward around the tubes G through which the products of combustion pass downward. The partially heated water on leaving the tubes spreads out over the dome sheet H, where it is heated by the hottest gases and then passes to the wier K through the baffle plate I around the ther-mometer J, where the outlet temperature is taken. From the wier K it overflows to the waste until test begins, after which it goes through the cock below the wier to the automatic tipping bucket, which is a two-compartment funnel mounted on pivots held in extreme position by the latch so that the water to be weighed runs from one compartment to the receiving pail, while the meter needle is making one revolution or a tenth of a foot of gas is burned. As soon as the circuit is closed by the meter needle the current passing through the solenoid adjacent to the tipping bucket raises the armathrough the solehold adjacent of water flowing through one compart-ment of the tipping bucket to swing it to a new position, thereby discharging water for the next tenth of a foot of gas burned into the empty pail. While this pail is filling the filled pail is weighed and the B. T. U. may be determined and recorded while another tenth of a foot of gas is burned and continuous and correct results may be obtained and recorded as long as desired. The general set-up of the calorimeter is shown on page 361. The following method of calculating the B. T. U. is used:

 $t_1 = temperature of incoming water.$

 $t_2 = temperature of outgoing water.$

w = pounds of water passed through.

c = pounds of water condensed (average for each 0.1 cu. ft.).

From which B. T. U. per cubic foot =

10 (w + c + 0.02) ($t_2 - t_1$) - 9704 c.

Example:

 $t_1 = 63.0^{\circ} F$

 $t_2 = 111.0^{\circ} F.$

w = 1.7531 lbs. c = 0.0091 lb.

10 (1.7531 + 0.0091 + 0.02) (111.0 - 63.0) - (9704) (.0091) =855.3 - 88.3 = 767 B. T. U. per cubic foot.

35-B. APPROXIMATE HEATING VALUE OF NATURAL GAS BY CALCULATION.

The natural gas is burned with an excess of oxygen in a regular combustion pipet J as shown in the apparatus on page 358. V

B. T. U. per cu. ft. is equal to 504 - where $V_0 =$ volume of Vn

oxygen consumed in burning Vn volumes of natural gas.

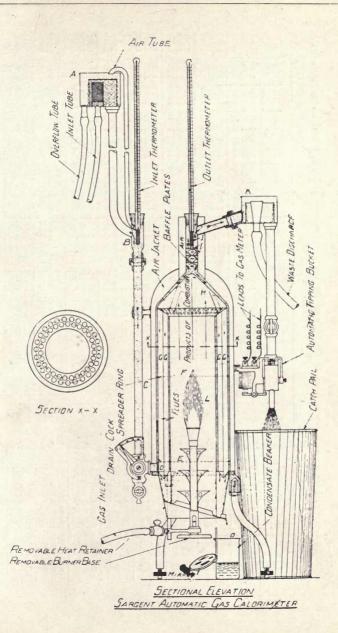
35-C. B. T. U. OF GAS BY CALCULATION FROM ANALYSIS.

The heating value of natural gas or any other gas may be calculated as follows:

Percentage	of	illumina	tes x	20.00	=	
Percentage	of	CO	x	3.41	-	
Percentage	of	CH4	x	10.65	=	
Percentage				3.45		
sum of thes	io i	tho R	TI	nor ou	hie .	f+

The sum of these is the B. T. U. per cubic ft. =

KANSAS CITY TESTING LABORATORY



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									NETERMONIVE											
1	28.6	28.7	28.8	28.9	29.0	29.1	29.2	29.3	29.4	29.5	29.6	29.7	29.8	29.9	30.0	30.1	30.2	30.3	30.4	30.5
885	0.909 0.971 0.974	0.973 0.975 0.975 0.977	0.976 0.978 0.981	0.980 0.982 0.984	0.983 0.985 0.988	0.986 0.986 0.991	0.990 0.992 0.995	0.993 0.998 0.998	0.997 0.999 1.002	1.000 1.003 1.005	1.004 1.006 1.009	1.007 1.010 1.012	1.011 1.013 1.016	1.014 1.017 1.019	1.018 1.020 1.023	1.021 1.024 1.026	1.025 1.027 1.030	1.028 1.031 1.033	1.031 1.034 1.034	1.035 1.035 1.040
1899	0.976 0.979 0.981	0.980 0.982 0.985	0.988	0.987 0.989 0.992	0.990 0.995 0.995	0.994	$\begin{array}{c} 0.997\\ 1.000\\ 1.002\end{array}$	1.003 1.003 1.006	$1.004 \\ 1.007 \\ 1.009 \\ 1.009 \\ 1.009 \\ 1.009 \\ 1.000 \\ 1.00$	1.008 1.010 1.013	1.011 1.014 1.016	1.015 1.017 1.020	1.018 1.021 1.023	1.022 1.024 1.021	1.025 1.028 1.028	1.029	1.032 1.035 1.035 1.037	1.036	1.030 1.042 1.042 1.044	1.043 1.045 1.045 1.045
1-010	0.984 0.986 0.989	0.992 0.992	0.993 0.996 0.996	0.994	0.998 1.000 1.003	1.001	1.005 1.007 1.010	1.011	1.012 1.014 1.017	1.015 1.018 1.020	1.019	1.022 1.025 1.027	1.026	1.029 1.032 1.034	1.035 1.035 1.035	1.036	1.040 1.042 1.045	1.043 1.046 1.046 1.048	1.047 1.049 1.052	1.050
499	0.995	0.994 0.997 0.999	0.998 1.001 1.003	1.001 1.004 1.006	1.005 1.008 1.010	1.011	1.012 1.015 1.017	1.015 1.018 1.020	1.019 1.022 1.024	1.022 1.025 1.025	1.026	1.029	1.038	1.036 1.039 1.041	1.040 1.043 1.045	1.043 1.046 1.046	1.050 1.050 1.052	1.053	1.054 1.057 1.059	1.060
1901	$\begin{array}{c} 0.998\\ 1.000\\ 1.002 \end{array}$	1.001 1.004 1.006	1.005 1.007 1.010	1.009 1.011 1.013	1.012 1.014 1.014	1.016 1.018 1.020	1.019	1.023 1.025 1.028	1.026 1.028 1.031	1.030 1.032 1.035	1.034 1.036 1.036	1.037 1.039 1.042	1.041	1.044 1.046 1.046 1.049	1.048	1.051	1.055 1.057 1.059	1.058	1.062 1.064 1.066	1.065
82.58	1.005 1.007 1.009	1.009 1.011 1.013	1.012 1.015 1.017	1.016 1.018 1.020	$1.020 \\ 1.022 \\ 1.024 \\ 1.024$	1.023 1.025 1.025	1.027 1.029 1.039	1.030 1.032 1.032 1.035	$1.034 \\ 1.036 \\ 1.038 \\ 1.038 $	1.037 1.039 1.042	1.041 1.043 1.045	1.044	1.048 1.050 1.052	1.051 1.053 1.056	1.057	1.058 1.060 1.063	1.062 1.064 1.067	1.065	1.009 1.072 1.074	1.073 1.076 1.078
838	$1.012 \\ 1.014 \\ 1.016 \\ 1.016$	1.015 1.018 1.020	$1.019 \\ 1.022 \\ 1.024 $	1.022 1.025 1.027	1.026 1.029 1.031	1.030 1.032 1.034	1.033 1.036 1.036	1.037 1.040 1.042	1.041 1.043 1.046	1.044	1.048 1.050 1.053	1.051	1.055	1.058	1.062 1.064 1.067	1.065 1.068 1.070	$1.069 \\ 1.072 \\ 1.072 \\ 1.074$	1.073 1.075 1.075	1.077 1.079 1.082	1.080
8888	1.019 1.021 1.023	1.023 1.025 1.025 1.027	1.027 1.029 1.031	1.030 1.032 1.032 1.034	$1.034 \\ 1.036 \\ 1.038 \\ 1.03$	1.037 1.039 1.042	1.041 1.043 1.045	1.044 1.047 1.049	1.048 1.056 1.056	1.051 1.054 1.056	1.055 1.057 1.060	1.058 1.061 1.063	1.062 1.064 1.067	1.066	1.069 1.072 1.072	1.073 1.075 1.078	1.077 1.079 1.083	1.081	1.085 1.087 1.090	1.091

TEMPERATURE 'F.

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51	28.6	28.7	28.8	28.9	20.0	29.1	29.2	20.3	29.4	29.5	29.6	29.7	29.8	29.9	30.0	30.1	30.2	30.3	30.4	30.5
	0.820	0.820	0.823	0.829	0.832	0.832	0.835	0.838	0.841	0.844	0.847	0.851	0.854	0.857	0.800	0.864	0.86	0.870	0.874	0.881
	0.827	0.830	0.833	0.830	0.830	0.842	0.845	0.848	0.851	0.854	0.858	0.861	100.0	108.0	128.0	0.874	10.01 9.878	188.0	0.884	0.887
	0.830 0.834	0.834	0.837 0.840	0.840 0.843	0.843	0.850	0.849 0.853	0.852 0.856	0.855	0.858	0.862	0.805	0.868	0.871 0.874	0.874 0.878	0.878 0.881	0.881	0.885	0.888	0.894
	0.837 0.840 0.840	0.840 0.843 0.847 0.847	0.843 0.846 0.850	0.846 0.853 0.853	0.850 0.853 0.857 0.857	0.853 0.856 0.866 0.800	0.856 0.856 0.868 0.868	0.869 0.862 0.862 0.866	0.862 0.865 0.865 0.865	0.805 0.808 0.808 0.872	0.808 0.872 0.872 0.876	0.875 0.875 0.879	0.875 0.878 0.878	0.878 0.881 0.885	0.885 0.885 0.885 0.885	0.885 0.888 0.892	0.888 0.891 0.895 0.895	0.891 0.895 0.898 0.898	0.895 0.895 0.898 0.902	0.808 0.901 0.905
	0.850 0.854	0.850 0.853 0.853 0.857	0.853 0.856 0.856 0.856	0.856 0.859 0.859 0.863	0.860 0.863 0.867	0.808 0.806 0.870	0.870 0.873 0.873	0.800 0.873 0.876	0.872 0.876 0.879	0.875 0.879 0.882	0.879 0.882 0.886	0.885 0.885 0.889	0.885 0.888 0.892	0.891 0.891 0.805	0.892 0.894 0.898 0.898	0.895 0.896 0.901	0.9080	0.902 0.905 0.908	0.905 0.905 0.905 0.911	0.908 0.911 0.914
	0.867 0.800 0.803	0.860 0.863 0.866	0.863 0.869 0.869	0.806	0.870 0.873 0.873	0.873 0.876 0.876	0.876 0.879 0.883	0.879 0.882 0.882	0.880	0.885 0.888 0.891	0.889 0.892 0.895	0.892 0.896 0.896 0.896	0.805 0.805 0.901 0.901	0.808 0.901 0.904	0.901	0.904	0.908 0.911 0.914 0.914	0.911 0.914 0.918	0.914 0.918 0.921	0.918 0.921 0.924
1	0.806 0.869 0.872	0.809 0.872 0.875	0.872 0.875 0.875	0.875 0.875 0.875 0.875	0.879 0.882 0.885	0.885 0.885 0.885	0.885 0.886 0.892	0.892 0.895 0.895	0.802 0.895 0.896 0.896	0.808 0.901 0.901	0.808 0.902 0.905	0.902 0.906 0.908	0.904 0.904 0.911	0.907 0.911 0.914	0.910 0.914 0.917	0.914 0.917 0.920	0.917 0.921 0.924	0.921 0.924 0.927	0.924 0.928 0.931	0.938 180.0
1	0.875 0.878 0.878 0.881	0.878 0.881 0.884	0.885 0.885 0.885	0.885 0.885 0.885 0.885 0.888	0.892 0.895 0.895	0.802 0.806 0.808	0.806 0.806 0.901 0.901	0.901	100.0 100.0	0.904 0.907 0.910	0.907 0.910 0.913	0.910 0.913 0.916 0.916	0.914 0.917 0.920	0.920 0.923 0.923	0.920 0.923 0.926 0.926	0.926 0.926 0.926 0.929	0.930	0.981 0.984 0.987	0.934 0.934 0.940	0.937 0.940 0.943
	0.884 0.887 0.889	0.887 0.890 0.893 0.893	0.890 0.893	0.804 0.809	0.898 0.900 0.903	0.903	0.904	0.907 0.909 0.909 0.912	0.910 0.913 0.913	0.913 0.916 0.916 0.919	0.916 0.919 0.925	0.919 0.929 0.925	0.923 0.926 0.926	0.926 0.929 0.932	0.932 0.932 0.935	0.936 0.936 0.939	0.936 0.939 0.939 0.942	0.940 0.946 0.946 0.946	0.943 0.946 0.946	0.946 0.949 0.952
	0.895 0.895 0.895	0.895 0.896 0.898 0.901	0.800	0.905	0.906 0.908 0.911	0.909	0.912 0.914 0.917	0.915 0.918 0.921	0.918 0.921 0.924	0.924	0.924	0.931	0.934	0.935 0.937 0.937 0.9340	0.938 0.941 0.944	0.945 0.945 0.948 0.948	0.945	0.949 0.951 0.954	0.952 0.954 0.957	0.960

TEMPERATURE 'F.

10	358	222	8888	686	1000	1.005	1.013	1.020	1.028
30.5	0.963	0.975	0.980	0.989	0.997				
30.4	0.960 0.966 0.966	0.968 0.971 0.974	776.0 088.0 288.0	0.985 0.988 0.988	0.993 0.996 0.998	1.001	1.009 1.011 1.014	1.016 1.019 1.021	1.024 1.027 1.029
30.3	$\begin{array}{c} 0.957\\ 0.960\\ 0.962\end{array}$	0.965 0.9771	$\begin{array}{c} 0.973\\ 0.976\\ 0.979\end{array}$	$\begin{array}{c} 0.981\\ 0.984\\ 0.987\end{array}$	$\begin{array}{c} 0.989\\ 0.993\\ 0.995\end{array}$	$\begin{array}{c} 0.997\\ 1.000\\ 1.003\end{array}$	1.005 1.008 1.010	1.013 1.016 1.018	1.021 1.023 1.023
30.2	0.954 0.956 0.959	0.962 0.964 0.967	0.970 0.972 0.975	0.978 0.980 0.983	0.986 0.989 0.992	$\begin{array}{c} 0.994 \\ 0.997 \\ 1.000 \end{array}$	$1.002 \\ 1.005 \\ 1.007 \\ 1.00$	$1.010 \\ 1.012 \\ 1.014 \\ 1.014$	1.017 1.020 1.022
30.1	0.950 0.958 0.956	0.958 0.961 0.963	0.939 0.939 0.972	0.975 0.977 0.980	0.983 0.985 0.988	$\begin{array}{c} 0.991 \\ 0.996 \\ 0.996 \end{array}$	$\begin{array}{c} 0.999 \\ 1.001 \\ 1.004 \end{array}$	1.005 1.009 1.011	1.014 1.016 1.019
30.0	0.946 0.949 0.952	$\begin{array}{c} 0.955\\ 0.958\\ 0.958\\ 0.960\end{array}$	0.963 0.963 0.968	0.971 0.974 0.977	$\begin{array}{c} 0.979\\ 0.985\\ 0.985\end{array}$	$\begin{array}{c} 0.987\\ 0.993\\ 0.993\end{array}$	0.995 0.998 1.000	1.003 1.005 1.007	1.010
29.9	$0.943 \\ 0.946 \\ 0.949 \\ 0.949$	$\begin{array}{c} 0.951 \\ 0.954 \\ 0.957 \end{array}$	0.960 0.963 0.965	0.968 0.973 0.973	0.976 0.976 0.981	0.984 0.986 0.989	166.0 166.0	$\begin{array}{c} 0.999\\ 1.002\\ 1.004 \end{array}$	1.009
29.8	0.940 0.943 0.946	0.948 0.951 0.954	$\begin{array}{c} 0.957\\ 0.960\\ 0.962\end{array}$	0.965 0.967 0.970	$\begin{array}{c} 0.972 \\ 0.975 \\ 0.978 \\ 0.978 \end{array}$	0.980 0.985 0.985	0.988 0.991 0.993	0.995 0.998 1.000	1.003
29.7	0.937 0.939 0.942	0.945 0.945 0.950	0.953 0.956 0.959	0.961 0.964 0.967	$\begin{array}{c} 0.969\\ 0.972\\ 0.974\end{array}$	0.977 0.980 0.982	0.985 0.987 0.990	0.992 0.997 0.997	1.000
29.6	0.933	0.942 0.944 0.947	0.950 0.953 0.955	0.958 0.963 0.963	0.966 0.968 0.971	0.973 0.976 0.979	0.981 0.984 0.986	0.989 0.992 0.994	0.996
29.5	0.930 0.933 0.936	0.938 0.941 0.943	0.947 0.949 0.952	$\begin{array}{c} 0.954 \\ 0.957 \\ 0.960 \end{array}$	0.962 0.965 0.968	0.970 0.973 0.975	0.978 0.981 0.983	0.986 0.988 0.991	0.993
29.4	0.927 0.930 0.932	$\begin{array}{c} 0.935\\ 0.938\\ 0.938\\ 0.940 \end{array}$	$\begin{array}{c} 0.943 \\ 0.946 \\ 0.949 \end{array}$	0.951 0.954 0.957	0.959 0.962 0.964	0.967	0.975 0.977 0.980	0.983 0.985 0.988	0.9990
29.3	0.923 0.926 0.929 0.929	0.931 0.935 0.937	0.940 0.945 0.945	$\begin{array}{c} 0.948\\ 0.950\\ 0.953\end{array}$	0.956	0.963 0.966 0.969	0.971 0.974 0.976 0.976	0.979 0.981 0.984	0.986
29.2	$\begin{array}{c} 0.920\\ 0.923\\ 0.926\\ 0.926\end{array}$	0.928 0.931 0.934	$\begin{array}{c} 0.937\\ 0.940\\ 0.942\\ 0.942 \end{array}$	0.945 0.947 0.950	$\begin{array}{c} 0.952 \\ 0.955 \\ 0.958 \\ 0.958 \end{array}$	0.960 0.963 0.963	0.968 0.973	0.976 0.978 0.980	0.985
29.1	0.917 0.920 0.923	0.925 0.928 0.931	0.933 0.936 0.939	0.941 0.944 0.947	0.952 0.952 0.954	0.957	0.969 196.0	0.972 0.975 0.977	0.982
29.0	0.914 0.917 0.917	0.925 0.925 0.928	0.930 0.935 0.935	0.938 0.941 0.944	0.946 0.949 0.951	$\begin{array}{c} 0.954 \\ 0.956 \\ 0.959 \end{array}$	0.961 0.964 0.966	0.939 0.974 0.974	0.976
28.9	0.910 0.914 0.916 0.916	0.919 0.921 0.924	0.930 0.932 0.932	0.935 0.937 0.937 0.940	$\begin{array}{c} 0.942 \\ 0.945 \\ 0.948 \end{array}$	0.950 0.953 0.955	0.958	0.966	0.973
28.8	0.907 0.910 0.913	0.915 0.918 0.921	$\begin{array}{c} 0.924 \\ 0.926 \\ 0.929 \end{array}$	0.931 0.934 0.937	$\begin{array}{c} 0.939 \\ 0.942 \\ 0.944 \end{array}$	$\begin{array}{c} 0.947\\ 0.949\\ 0.952\end{array}$	0.954 0.957 0.959	0.967 0.967	0.969
28.7	0.904	0.912 0.915 0.917	0.920 0.925 0.925	0.928 0.931 0.933	0.936 0.938 0.941	$\begin{array}{c} 0.944 \\ 0.946 \\ 0.949 \end{array}$	0.954	0.959 0.953 0.953	0.968
28.6	0.901 0.906 0.906	0.909 0.911 0.914	0.917 0.920 0.922	0.925	0.935 0.935 0.938	0.941 0.943 0.945	$\begin{array}{c} 0.947\\ 0.950\\ 0.952\end{array}$	0.955	0.965
	8228	1222	73	528	858	848	858	218 20	18182

BULLETIN NUMBER FIFTEEN OF

TEMPERATURE °F.

Comparison of Temperatures by the Fahrenheit and Centigrade Scales

Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr.
-273°	-459.4						
Absolut							
-200°	-328.0	- 5.6	+22.0	15.6	60.0	36.1	97.0
Tempera		- 5.0	+23.0	16.0	60.8	36.7	98.0
Liquid	Air	- 4.4	+24.0	16.1	61.0	37.0	98.6
-130°	-202.0	- 4.0	+24.8	16.7	62.0	37.2	99.0
Pure Grain		- 3.9	+25.0	17.0	62.6	37.8	100.0
Free		- 3.3	+26.0	17.2	63 0	38.0	100.4
-70°	-94.0	- 3.0	+26.6	17.8	64.0	38.3	101.0
Ammonia		- 2.8	+27.0	18.0	64.4	38.9	102.0
-(75		- 2.2	+28.0	18.3	65.0	39.0	102.2
-40°	-40.	- 2.0	+28.4	18.9	66.0	39.4	103.0
Mercury		- 1.7	+29.0	19.0	66.2	40.0	104.0
(39		- 1.1	+30.0	19.4	67.0	40.6	105.0
-30°	-22	- 1.0	+30.2	20.0	68.0	41.0	105.8
mmonia	Liquefies	- 0.6	+31.0	20.6	69.0	41.1	106.0
at3		0.	+32.0	21.0	69.8	41.7	107.0
28	-18.4	+ 0.6	+33.0	21.1	70.0	42.0	107.6
-26	-14.8	1.0	33.8	21.7	71.0	42.2	108.0
-24	-11.2	1.1	34.0	22.0	71.6	42.8	109.0
-22	- 7.6	1.7	35.0	22.2	72.0	43.0	109.4
-20	- 4.0	2.0	35.6	22.8	73.0	43.3	110.0
-19	- 2.2	2.2	36.0	23.0	73.4	43.9	111.0
-18	- 0.4	2.8	37.0	23.3	74.0	44.0	111.2
-17.8	- 0.0	3.0	37.4	23.9	75.0	44.4	112.0
-17.2	+ 1.0	3.3	38.0	24.0	75.2	45.0	113.0
-17.0	+ 1.4	3.9	39.0	24.4	76.0	45.6	114.0
-16.7	+ 2.0	4.0	39.2	25.0	77.0	46.0	114.8
-16.1	+ 3.0	4.4	40.0	25.6	78.0	46.1	115.0
-16.0	+ 3.2	5.0	41.0	. 26.0	78.8	46.7	116.0
-15.6	+ 4.0	5.6	42.0	26.1	79.0	47.0	116.6
-15.0	+ 5.0	6.0	42.8	26.7	80.0	47.2	117.0
-14.4	+ 6.0	6.1	43.0	27.0	80.6	47.8	118.0
-14.0	+ 6.8	6.7	44.0	27.2	81.0	48.0	118.4
-13.9	+ 7.0	7.0	44.6	27.8	82.0	48.3	119.0
-13.3	+ 8.0	7.2	45.0	28.0	82.4	48.9	120.0
-13.0	+ 8.6	7.8	46.0	28.3	83.0	49.0	120.2
-12.8	+ 9.0	8.0	46.4	28.9	84.0	49.4	121.0
-12.2	+10.0	8.3	47.0	29.0	84.2	50.0	122.0
-12.0	+10.4	8.9	48.0	29.4	85.0	50.6	123.0
-11.7	+11.0	9.0	48.2	30.0	86.0	51.0	123.8
-11.1	+12.0	9.4	49.0	30.6	87.0	51.1	124.0
-11.0	+12.2	10.0	50.0	31.0	87.8	51.7	125.0
-10.6	+13.0	10.6	51.0	31.1	88.0	52.0	125.6
-10.0	+14.0	11.0	51.8	31.7	89.0	52.2	126.0
- 9.4	+15.0	11.1	52.0	32.0	89.6	52.8	127.0
- 9.0	+15.8	11.7	53.0	32.2	90.0	53.0	127.4
- 8.9	+16.0	12.0	53.6	32.8	91.0	53.3	128.0
- 8.3	+17.0	12.2	54.0	33.0	91.4	53.9 54.0	129.0
- 8.0	+17.6	12.8	55.0	33.3	92.0	54.4	129.2 130.0
- 7.8	+18.0	13.0	55.4	33.9	93.0	55.0	
- 7.2	+19.0	13.3	56.0	34.0	93.2	55.6	131.0 132.0
- 7.0	+19.4	13.9	57.0	34.4	94.0		
- 6.7	+20.0	14.0	57.2	35.0	95.0	56.0	132.8
- 6.1	+21.0	14.4	58.0	35.6	96.0	56.1	133.0
- 6.0	+21.2	15.0	59.0	36.0	. 96.8	56.7	134.0

Temperature Conversion Tables

Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr
57.0	134.6	77.8	172.0	98.3	209.0	119.0	246.2
57.0	135.0	78.0	172.4	98.9	210.0	110.4	247.0
57.2 57.8	136.0	78.3	173.0	99.0	210.0	$119.4 \\ 120.0 \\ 120.6$	248.0
01.8	130.0		173.0	99.4	210.2	120.0	243.0
58.0	136.4	78.9	1/4.0	100.0	211.0	120.0	249.8
58.3	137.0	79.0	174.2	100.0	212.0 213.0	121.0	249.0
58.9	137.0 138.0 138.2	79.4	175.0	100.6	213.0	$121.1 \\ 121.7$	250.0
59.0	138.2	80.0	176.0	101.0	213.8 214.0	121.7	251.0
59.4	139.0	80.6	176.0 177.0 177.8	101.1	214.0	122.0	251.6
60.0	140.0	81.0	177.8	101.7	215.0	122.2	252.0
60.6	141.0	81.1	178.0	102.0	215.6	122.8	253.0
61.0	141.8	81.7	179.0	$102.2 \\ 102.8$	216.0 217.0	123.0 123.3 123.9	253.4
61.1	142.0	82.0	179.6	102.8	217.0	123.3	254.0
61.7	143.0	82.2	180.0	103.0	217.1 218.0	123.9	255.0
62.0	143.6 144.0	82.8	181.0	103.3	218.0	124.0	255.2
62.2	144.0	83.0	181.4	103.9	219.0	124.4	256.0
62.8 63.0	145.0	83.3	182.0	104.0	219.0 219.2	$124.4 \\ 125.0$	257.0
63.0	145 4	83.9	183.0	104.4	220.0 221.0	125.6	258.0
63.0	146.0	84.0	$183.0 \\ 183.2$	105.0	221.0	126.0	258.8
63.9	146.0 147.0 147.2	84.4	184.0	105.6 106.0	222.0	$126.0 \\ 126.1 \\ 126.7 \\ 127.0 $	259.0
64.0	147.9	85.0	185.0	166.0	222.8	126.7	260.0
64.4	148.0	85.6	100.0	106.1	223.0	127.0	260.0
65.0	149.0	85.0 86.0	186.0 186.8 187.0	106.7	>24.0	127 2	261.0
00.0	149.0	0.06	180.8	107.0	224.6	127.8 128.0	262.0
65.6	150.0	86.1	187.0	107.2	225.0	128.0	262.4
66.0	150.8	86.7	188.0	107.8	226.0	198 2	263.0
66.1	151.0	87.0	188.6	108.0	226.0 226.4	198.0	264.0
66.7	152.0	87.2	189.0	$108.0 \\ 108.3$	227.0	128.3 128.3 128.9 129.0 129.4	264.2
67.0	152.6 153.0	87.8	190.0	108.9	008.0	120.0	265.0
67.2	153.0	88.0	190.4	109.0	228.0 228.2 229.0 230.0	130.0	266.0
67.8	154.0	88.3	191.0	109.4	220.2	190.6	267.0
68.0	154.4	88.9	192.0	110.0	920.0	191.0	267.8
68.3	155.0	89.0	192.2	110.0	230.0	130.6 131.0 131.1 131.7 132.0	268.0
68.9	156.0	89.4	193.0	111.0	231.0 231.8	101.1	269.0
69.0	156.2	90.0	194.0	111.0 111.1 111.7 112.0	232.0	192.0	269.6
69.4	157.0	90.6	195.0	111.4	232.0	102.0	270.0
70.0	158.0	91.0	195.8	111.7	233.6	102.2	271.0
70.6	159.0	91.1	196.0	112.0	234.0	199.0	271.4
71.0	159.8	91.7	197.0	112.2	231.0	100.0	272.0
71.1	160.0	92.0	197.6	112.5	235.4	100.0	273.0
71.7	161.0	92.2	198.0	113.0	2-30.4	100.0	273.2
72.0	161.6	92.8	199.0	113.3	236.0 237.0	$132.2 \\ 132.8 \\ 133.0 \\ 133.3 \\ 133.0 \\ 134.0 \\ 134.4 \\ 135.6 \\ 136.0 \\ 136.1 \\ 136.7 \\ 137.0 \\ 137.$	274.0
72.2	162.0	93.0	199.4	113.9	227.2	104.4	275.0
72.8	163.0	93.3	200.0	114.0	238.0	130.0	276.0
73.0	163.4	93.9	201.0	114.4	235.0	130.0	276.8
73.3	164.0 165.0	94.0	201.2	115.0	239.0	130.0	270.8
73.9	165.0	94.4	202.0	115.6	240.0	136.1	277.0 278.0
74.0	165.2	95.0	203.0	116.0	240.8	130.7	218.0
74.4	166.0	95.6	203.0 204.0	116.1	241.0	137.0	278.6
75.0	167.0	96.0	204.8	116.7	242.0	137.2	279.0
75.6	168.0	96.1	205.0	117.0	242.6	137.8	280.0
76.0	167.0 168.0 168.8	96.7	206.0	117.2	243.0	138.0	280.4
76.1	169.0	97.0	206.6	117.8	244.0	138.3	281.0
76.7	170.0	97.2	207.0	118.0	241.4	138.9	282.0
77.0	170.6	97.8	208.0	118.3	245.0	139.0	282.2
77.2	171.6	98.0	208.4	118.9	246.0	139.4	283.0
11.4	1/1.0	20.0	200.4	110.8	240.9	159.4	205.0

TEMPERATURE CONVERSION TABLES—Continued.

Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr.	Cent.	Fahr.
140.0	284.0	215.0	419.0	590.0	1094.0	1360.0	2480.0
140.6	285.0	220.0	428.0	600.0	1112.0	1380.0	2516.0
140.0	285.8	225.0	437.0	610.0	1130.0	1400.0	2552.0
141.0	286.0	230.0	446.0	620.0	1148.0	1420.0	2588.0
141.7		235.0	455.0	630.0	1166.0	1440.0	2624.0
142.0	287.0 287.6	240.0	464.0	640.0	1184.0	1460.0	2660.0
142.0		245.0	473.0	650.0	1202.0	1480.0	2696.0
142.2	288.0	250.0	482.0	660.0	1202.0	1500.0	2090.0
142.8	289.0 289.4	254.0	489.2	670.0	1238.0	1520.0	2768.0
143.3	289.4	255.0	491.0	680.0	1256.0	1540.0	2804.0
143.9	290.0	260.0	500.0	690.0	1230.0	1560.0	2840.0
143.9	291.0	265.0	509.0	700.0	1292.0	1580.0	2840.0
144.4	291.2 292.0	270.0	518.0	710.0	1310.0	1600.0	2912.0
144.4	292.0	275.0	527.0		1328.0		2912.0
145.6	293.0	280.0	536.0	720.0 730.0	1346.0	1620.0 1640.0	2918.0
140.0		283.0	541.4			1040.0	3020.0
146.0 146.1	294.8 295.0	283.0 285.0	545.0	740.0 750.0	1364.0 1382.0	1660.0 1680.0	3020.0
140.1 146.7		288.0	550.4			1080.0	
	296.0	290.0	554.0	760.0	1400.0	1700.0	3092.0
147.0 147.2	296.6	295.0	563.0	770.0	1418.0	1720.0	3128.0
	297.0			780.0	1436.0	1740.0	3164.0
147.8	298.0	300.0 305.0	572.0	790.0	1454.0	1760.0	3200.0
148.0	298.4		581.0	800.0	1472.0	1780.0	3236.0
148.3	299.0	310.0 315.0	590.0	810.0	1490.0	1800.0	3272.0
148.9	300.0		599.0	820.0	1508.0	1825.0	3317.0
149.0	300.2	320.0 325.0	608.0	830.0	1526.0	1850.0	3362.0
149.4	301.0		617.0	840.0	1544.0	1875.0	3407.0
150.0	302.0	330.0	626.0	850.0	1562.0	1900.0	3452.0
152.0	305.6	335.0	635.0	860.0	1580.0	1925.0	3497.0
154.0	309.2	310.0	644.0	870.0	1598.0	1950.0	\$542.0
156.0	312.8	345.0	653.0	880.0	1616.0	1975.0	3587.0
158.0	316.4	350.0	662.0	890.0	1634.0	2000.0	3632.0
160.0	320.0	360.0	680.0 698.0	900.0	1652.0	2400.0	3812.0
162.0	323.6	370.0		920.0	1688.0	2500.0	4532.0
164.0	327.2	380.0 390.0	716.0 734.0	940.0	1724.0	3000.0	5432.0
166.0	330.8		752.0	960.0	1760.0	3500.0	6332.0
168.0	334.4	400.0 410.0	770.0	980.0	1796.0	4000.0	7232.0
170.0	338.0			1000.0	1832.0	5000.0	9032.0
172.0 174.0	341.6	420.0 430.0	788.0 806.0	1020.0	1868.0	6000.0	10832.0
174.0	345.2		824.0	1040.0	1904.0		
176.0	348.8	440.0 450.0	842.0	1060.0	1940.0		
178.0	352.4	450.0	860.0	1080.0	1976.0		
180.0	356.0	400.0 470.0	878.0	1100.0	2012.0		
182.0	359.6			1120.0	2048.0		
184.0	353.2	480.0	896.0	1140.0	2084.0		
186.0	366.8	490.0	914.0	1160.0	2120.0		
188.0	370.4	500.0	932.0	1180.0	2156.0		
190.0	374.0	510.0	950.0	1200.0	2192.0		
192.0	377.6	520.0	968.0	1220.0	2228.0		
194.0	381.2	530.0	986.0	1240.0	2264.0		
196.0	384.8	540.0	1004.0	1260.0	2300.0		
198.0	388.4	550.0	1022.0	1280.0	2336.0		100
200.0	392.0	560.0	1040.0	1300.0	2372.0		
205.0	401.0	570.0	1058.0	1320.0	2408.0		
210.0	410.0	580.0	1076.0	1340.0	2444.0		

TEMPERATURE READING CONVERSION FACTORS. Temp. Centigrade = 5/9 (F,-32) = 5/4 R. Temp. Fahrenheit = 9/5 C. + 32 = 9/4 R. + 32. Temp. Reaumur = 4/5 C. = 4/9 (F,-32).

BULLETIN NUMBER FIFTEEN OF

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON. (U. S. BUREAU OF STANDARDS.)

	0	1	2	3	4	5	6	7	8	9
		1			-			-		
10	1.0000	.9993	.9986	.9979	.9972	.9964	.9957	.9950	.9943	.9936
	8.325	8.322	8.317	8.311	8.305	8.299	8.293	8.287	8.281	8.275
11	.9929	.9922	.9915	.9908	.9901	.9894	.9887	.9880	.9873	.9866
	8.269	8.263	8.258	8.252	8.246	8.240	8.234	8.228	8.223	8.217
2	.9859	.9852	.9815	.9838	.9831	.9825	.9818	.9811	.9804	.979
	8.211	8.205	8,194	8.194	8.188	8,182	8.176	8.171	8.165	8.159
3	.9790	.9783	.9777	.9770	.9763	.9756	.9749	.9743	.9736	.972
	8.153	8,148	8.142	8.137	8.131	8.125	8.119	8.114	8.108	8.102
4	.9722	.9715	.9709	.9702	.9695	.9688	.9682	.9675	.9669	.966
	8.096	8.091	8.086	8.080	8.074	8.069	8.063	8.058	8.052	8.047
15	.9655	.9649	.9642	.9635	.9629	.9622	.9615	.9609	.9602	
	8.041	8.035	8.030	8.024	8.019	8.013	8.007			.959
6								8.002	7.997	7.991
0.	.9589	.9582	.9582	.9569	.9563	.9556	.9550	.9543	.9537	.953
	7.986	7.980	7.975	7.969	7.964	7.959	7.653	7.948	7.942	7.937
17	.9524	.9517	.9511	.9504	.9498	.9492	.9485	.9479	.9472	.946
	7.931	7.926	7.921	7.915	7.910	7.904	7.899	7.894	7.888	7.883
18	.9459	.9453	.9447	1440	9434	.9428	.9421	.9415	.9409	.940
	7.877	7.872	7.867	7.861	7.856	7.851	7.846	7.841	7.835	7.830
9	.9396	.9390	.9383	.9377	.9371	.9365	.9358	.9352	.9346	.934
	7.825	7.820	7.814	7.809	7.804	7.799	7.793	7.788	7.783	7.778
0	.9333	.9327	.9321	.9315	.9309	.9302	.9296	.9290	.9284	.9278
	7.772	7.767	7.762	7.757	7.752	7.747	7.742	7.736	7.731	7.726
1	.9272	.9265	.9259	.9253	.9247	.9241	.9235	.9229	.9223	.921
1	7.721	7.716	7.711	7.706	7.701	7.696	7.690	7.685	7.680	7.675
2	.9211	.9204	.9198	.9192	.9186	.9180	.9174	.9168	.9162	.9150
~	7,670	7.665	7.660	7.655	7.650	7.645	7.640	7.635	7.630	7.625
3	.9150	.9144	.9138	.9132	.9126	.9121	.9115	.9109	.9103	.9097
~	7.620	7.615	7.610	7.605	7.600	7.595	7.590	7.585	7.580	7.575
4	.9091	.9085	.9079			.9061	.9056	.9050	.9044	
	7.570	7.565	7.561	.9073 7.556	.9067 7.551	7.546	7.541	7.536	7.531	.9038 7.526
5	.9032	.9026				.9003	.8997			
G			.9021	.9015	.9009			.8992	.8986	.8980
0	7.522	7.517	7.512	7.507	7.502	7.497	7.493	7.488	7.483	7.478
6	.8974	.8969	.8963	.8957	.8951	.8946	.8940	.8934	.8929	.8923
	7.473	7.469	7.464	7.459	7.454	7.449	7.445	7.440	7.435	7.430
7	.8917	.8912	.8906	.8900	.8895	.8889	.8883	.8878	.8872	.8866
	7.425	7.421	7.416	7.411	7.407	7.402	7.397	7.393	7.388	7.383
8	.8861	.8855	.8850	.8844	.8838	.8833	.8827	.8822	.8816	.8811
	7.378	7.374	7.369	7.365	7.360	7.355	7.351	7.346	7.341	7.337
9	.8805	.8799	.8794	.8788	.8783	.8777	.8772	.8766	.8761	.8755
	7.332	7.328	7.323	7.318	7.314	7.309	7.305	7.300	7.295	7.291
0	.8750	.8745	.8739	.8734	.8728	.8723	.8717	.8712	.8706	.8701
	7.286	7.282	7.277	7.273	7.268	7.264	7.259	7.254	7.249	7.245
1	.8696	.8690	.8685	.8679	.8674	.8669	.8663	.8658	.8653	.8647
	7.241	7.236	7.232	7.227	7.223	7.218	7.214	7.210	7.205	7.201
2	.8642	.8637	.8631	.8626	.8621	.8615	.8610	.8605	.8600	.8594
-	7.196	7.192	7.187	7.183	7.178	7.173	7.169	7.165	7.161	7.156
3	.8589	.8584	.8578	.8573	.8568	.8563	.8557	.8552	.8547	.8542
	7.152	7.147	7.143	7.139	7.134	7.130	7.125	7.121	7.117	7.113
L	.152	.8531	.143	.139	.134	.8511	.8505	.8500	.8496	.8490
1	.8537					7.087	7.082		.8490	
-		7.104	7.100	7.095	7.091			7.078		7.069
5	.8485	.8480	.8475	.8469	.8464	.8459	.8454	.8449	.8444	.8439
	7.065	7.061	7.057	7.052	7.048	7.044	7.039	7.035	7.031	7.027
5	.8434	.8429	.8424	.8419	.8413	.8408	.8403	.8398	.8393	.8388
	7.022	7.018	7.014	7.010	7.006	7.001	6.997	6.993	6.989	6.985
7	.8383	.8378	.8373	.8368	.8363	.8358	.8353	.8348	.8343	.8338
	6.980	6.976	6.972	6.968	6.964	6.960	6.955	6.951	6.947	6.943
8	.8333	.8328	.8323	.8318	.8314	.8309	.8304	.8299	.8294	.8289
	6.939	6.935	6,930	6.926	6.922	6.918	6.914	6.910	6.906	6.902

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON—Con. U. S. BUREAU OF STANDARDS—Con.

	0	1	2	3	4	5	6	7	8	9
39	.8284	.8279	.8274	.8269	.8264	.8260	.8255	.8250	.8245	.8240
	6.898	6.894	6.889	6.885	6.881	6.877	6.873	6.869	6.865	6.861
40	.8235	.8230	.8226	.8221	.8216	.8211	.8206	.8202	.8197	.8192
	6.857	6.853	6.849	6.845	6.841	6.837	6.833	6.829	6.825	6.821
41	.8187	.8182	.8178	.8173	.8168	.8163	.8159	.8154	.8149	.8144
	6.817	6.813	6.809	6.805	6.801	6.797	6.793	6.789	6.785	6.781
42	.8140	.8135	.8130	.8125	.8121	.8116	.8111	.8107	.8102	.809
	6.777	6.773	6.769	6.765	6.761	6.758	6.754	6.750	6.746	6.742
43	.8092	.8088	.8083	.8078	.8074	.8069	.8065	.8060	.8055	.8051
	6.738	6.734	6.730	6.726	6.722	6.718	6.715	6.711	6.707	6.703
44	.8046	.8041	.8037	.8032	.8028	.8023	.8018	.8014	.8009	.8008
	6.699	6.695	6.691	6.688	6.684	6.680	6.676	6.672	6.668	6.665
45	.8000	.7995	.7991	.7986	.7982	.7977	.7973	.7968	.7964	.7956
	6.661	6.657	6.653	6.649	6.646	6.642	6.638	6.634	6.630	6.627
46	.7955	.7950	.7946	.7941	.7937	.7932	.7928	.7923	.7919	.7914
	6.623	6.619	6.616	6.612	6.608	6.604	6.600	6.597	6.593	6.589
47	.7910	.7905	.7901	.7896	.7892	.7887	.7883	.7878	.7874	.7870
	6.586	6.582	6.578	6.574	6.571	6.567	6.563	6.560	6.556	6.552
48	.7865	.7861	.7856	.7852	.7848	.7843	.7839	.7834	.7830	.7826
	6.548	6.545	6.541	6.537	6.534	6.530	6.526	6.523	6.519	6.515
49	.7821	.7817	.7812	.7808	.7804	.7799	.7795	.7791	.7786	.778
	6.511	6.508	6.504	6.501	6.497	6.494	6.490	6.486	6.483	6.479
50	.7778	.7773	.7769	.7765	.7761	.7756	.7752	.7748	.7743	.773
	6.476	6.472	6.468	6.465	6.461	6.458	6.454	6.450	6.447	6.443
51	.7735	.7731	.7726	.7722	.7717	.7713	.7709	.7705	.7701	.7697
	6.440	6.436	6.432	6.429	6.425	6.421	6.418	6.415	6.411	6.408
52	.7692	.7688	.7684	.7680	.7675	.7671	.7667	.7663	.7659	.7654
	6.404	6.401	6.397	6.394	6.390	6.387	6.383	6.380	6.376	6.373
53	.7650	.7646	.7642	.7638	.7634	.7629	.7625	.7621	.7617	.7613
	6.369	6.366	6.362	6.359	6.355	6.351	6.348	6.345	6.341	6.338
54	.7609	.7605	.7600	.7596	.7592	.7588	.7584	.7580	.7576	.7572
	6.334	6.331	6.327	6.324	6.321	6.317	6.314	6.311	6.307	6.304
55	.7568	.7563	.7559	.7555	.7551	.7547	.7543	.7539	.7535	.7531
	6.300	6.296	6.293	6.290	6.287	6.283	6.280	6.276	6.273	6.270
56	.7527	.7523	.7519	.7515	.7511	.7507	.7503	.7499	.7495	.7491
	6.266	6.263	6.250	6.256	6.253	6.249	6.246	6.243	6.240	6.236
57	.7487	.7483	.7479	.7475	.7471	.7467	.7463	.7459	.7455	.745
	6.233	6.229	6.226	6.223	6.219	6.216	6.213	6.209	6.206	6.203
58	.7447	.7443	.7439	.7435	.7431	.7427	.7423	.7419	.7415	.7411
	6.199	6.196	6.193	6.190	6.186	6.183	6.180	6.176	6.173	6.170
59	.7407	.7403	.7400	.7396	.7392	.7388	.7384	.7380	.7376	.7372
	6.166	6.163	6.160	6.157	6.154	6.150	6.147	6.144	6.141	6.137
60	.7368	.7365	.7361	.7357	.7353	.7349	.7345	.7341	.7338	.7334
	6.134 .	6.131	6.128	6.124	6.121	6.118	6.115	6.112	6.108	6.105
61	.7330	.7326	.7322	.7318	.7315	.7311	.7307	.7303	.7299	.7298
	6.102	6.099	6.096	6.093	6.090	6.086	6.983	6.080	6.077	6.073
62	.7292	.7288	.7284	.7280	.7277	.7273	.7269	.7265	.7261	.7258
	6.070	6.067	6.064	6.060	6.057	6.054	6.051	6.948	6.045	6.042
63	.7254	.7250	.7246	.7243	.7239	.7235	.7231	.7228	.7224	.7220
	6.038	6.035	6.032	6.029	6.026	6.023	6.020	6.017	6.014	6.010
34	.7216	.7213	.7209	.7205	.7202	.7198	.7194	.7191	.7187	.7189
	6.007	6:004	6.001	5.998	5.995	5.992	5.989	5.986	5.983	5.980
65	.7179	.7176	.7172	.7168	.7165	.7161	.7157	.7154	.7150	.7147
	5.976	5.973	5.970	5.967	5.964	5.961	5.958	5.955	5.952	5.949
66	.7143	.7139	.7136	.7132	.7128	.7125	.7121	.7117	.7114	.7110
	5.946	5.943	5.940	5.937	5.934	5.931	5.928	5.925	5.922	5.919
67	.7107	.7103	.7099	.7096	.7092	.7089	.7085	.7081	.7078	.7074
	5.916	5.913	5.910	5.907	5.904	5.901	5.898	5.895	5.892	5.889

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON—Con U. S. BUREAU OF STANDARDS—Con.

	0	1	2	3	4	5	6	7	8	9
68	.7071	.7067	.7064	.7060	.7056	.7053	.7049	.7046	.7042	.7039
	5.886	5.883	5.880	5.877	5.874	5.871	5.868	5.865	5.862	5.859
69	.7035	.7032	.7028	.7025	.7021	.7018	.7014	.7011	.7007	.7004
	5.856	5.853	5.850	5.848	5.845	5.842	5.839	5.836	5.833	5.830
70	.7000	.6997	.6993	.6990	.6983	.6983	.6979	.6976	.6972	.6669
	5.827	5.824	5.821	5.818	5.815	5.812	5.810	5.807	5.804	5.801
71	.6965	.6962	.6958	.6955	.6951	.6948 5.784	.6944	.6941	.6938	.6934 5.772
72	5.798 .6931	5.795	5.792	5.789 .6920	5.786 .6917	.6914	5.781 .6910	5.778 .6907	5.775	.6900
12	5.769	5.766	5.763	5.760	5.758	5.755	5.752	5.749	5.746	5.744
73	.6897	.6893	.6890	.6886	.6883	.6880	.6876	.6873	.6869	.6866
10	5.741	5.738	5.735	5.732	5.729	5.727	5.724	5.721	5.718	5.715
74	.6863	.6859	.6856	.6853	.6849	.6846	.6843	.6839	.6836	.6833
	5.712	5.710	5.707	5.704	5.701	5.698	5.696	5.693	5.690	5.687
75	.6829	.6826	.6823	.6819	.6816	.6813	.6809	.6806	.6803	.6799
	5.685	5.682	5.679	5.676	5.673	5.671	5.668	5.665	5.662	5.660
76	.6796	.6793	.6790	.6786	.6783	.6780	.6776	.6773	.6770	.6767
	5.657	5.654	5.652	5.649	5.646	5.643	5.640	5.638	5.635	5.632
77	.6763	.6760	.6757	.6753	.6750	.6747	.6744	.6740	.6737	.6734
-	5.629	5.627	5.624	5.621	5.618	5.616	5.613	5.610	5.608	5.605
78	.6731	.6728	.6724	.6721	.6718	.6715	.6711	.6708	.6705	.6702
-	5.602	5.600	5.597	5.594	5.592	5.589	5.586	5.584	5.581	5.578
79	.6699 5.576	.6695 5.573	.6692 5.570	.6689 5.568	.6686 5.565	.6683 5.562	.6679 5.560	.6676 5.557	.6673 5.554	.6670 5.552
80	.6967	.6663	.6660	.6657	.6654	6651	.6648	.6645	.6641	.6638
00	5.549	5.546	5.543	5.541	5.538	5.536	5.533	5.531	5.528	5.525
81	.6635	.6632	.6629	.6626	.6623	.6619	.6616	.6613	.6310	.6607
0.	5.522	5.520	5.517	5.515	5.512	5.510	5.507	5.504	5.502	5.490
82	.6604	.6601	.6598	.6594	.6591	.6588	.6585	.6582	.6579	.6576
	5.497	5.494	5.491	5.489	5.486	5.484	5.481	5.478	5.476	5.473
83	.6573	.6570	.6567	.6564	.6560	.6557	.6554	.6551	.6548	.6545
	5.471	5.468	5.466	5.463	5.460	5.458	5.455	5.453	5.450	5.448
84	.6542	.6539	.6536	.6533	.6530	.6527	.6524	.6521	.6518	.6515
	5.445	5.443	5.440	5.437	5.435	5.432	5.430	5.427	5.425	5.422
85	.6512	.6509	.6506	.6503	.6500	.6497	.6494	.6490	.6487	.6484
00	5.420	5.417	5.415	5.412	5.410	5.407	5.405	5.402	5.400	5.397
86	.6482	.6479	.6476	.6473	.6470	.6467	.6464 5.380	.6491 5.377	.6558	.6455 5.372
87	5.395	5.392 .6449	5.390 .6446	5.387	5.385	5.382	.6134	.6431	5.375 .6428	.6425
01	.6452 5.370	5.367	5.365	.6443 5.362	.6440 5.360	.6437 5.357	5.355	5.352	5.350	5.347
88	.6422	.6419	.6416	.6413	.6410	.6407	.6404	.6401	.6399	.6396
00	5.345	5.343	5.340	5.338	5.335	5.333	5.330	5.328	5.325	5.323
89	.6393	.6390	.6387	.6384	.6391	.6378	.6375	.6372	.6369	.6367
~	5.320	5.318	5.316	5.313	5.311	5.308	5.306	5.304	5.301	5.299
90	.6364	.6361	.6358	.6355	.6352	.6349	.6346	.6343	.6341	.6338
	5.296	5.294	5.291	5.289	5.286	5.284	5.281	5.279	5.277	5.275
91	.6335	.6332	.6329	.6326	.6323	.6321	.6318	.6315	.6312	.6309
	5.272	5.270	5.267	5.265	5.263	5.261	5.258	5.256	5.253	5.251
92	.6306	.6303	.6301	.6298	.6295	.6292	.6289	.6286	.6284	.6281
~~	5.248	5.246	5.214	5.241	5.239	5.236	5.234	5.232	5.230	5.227
93	.6278	.6275	.6272	.6270	.6267	.6264	.6251	.6258	.6256 5.206	.6253 5.204
04	5.225	5.222	5.220	5.218 .6242	5.216 .6239	5.213 .6236	5.210 .6233	5.208 .6231	.6228	.6225
94	.6250 5.201	.6247 5.199	.6214 5.196	5.194	5.192	5.190	5.187	5.185	5.183	5.180
95	.6222	.6219	.6217	.6214	.6211	.6208	.6206	.6203	.6200	.6197
00	5.178	5.176	5.174	5.171	5.169	5.169	5.164	5.162	5.160	5.157
96	.6195	.6192	.6189	.6186	.6184	.6181	.6178	.6176	.6173	.6170
	5.155	5.153	5.150	5.148	5.146	5.144	5.142	5.140	5.137	5.135
97	.6167	.6165	.6162	.6159	.6157	.6154	.6151	.6148	.6146	.6143
	5.132	5.130	5.128	5.126	5.124	5.121	5.119	5.116	5.114	5.112
98	.6140	.6138	.6135	.6132	.6130	.6127	.6124	.6122	.6119	.6116
	5.110	5.108	5.106	5.103	5.101	5.099	5.096	5.094	5.092	5.090
99	.6114	.6111	.6108	.6106	.6103	.6100	.6098	.6095	.6092	.6090
	5.088	5,085	5.083	5.081	5.079	5.076	5.074	5.072	5.070	5.068
00	.6087		•••••	•••••	•••••			•••••		•••••
	5.066									

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON. (MODULUS 141.5 TAGLIABUE.)

	0	1	2	3	4	5	6	7	8	9
10	1.0000	.9993	.9986	.9979	.9972	.9965	.9958	0051	.9944	.9937
	8.331	8.325	8.319	8.314	8.308	8.302	8.296	.9951 8.290	8.284	8.279
11	.9930 8.273	.9923 8.267	.9916 8.261	.9909 8.255	.9902 8.249	.9895 8.244	.9888 8,238	.9881 8.232	.9874 8.226	.9868 8.221
12	.9831	.9854	.9847	.9840	.9833	.9826	.9820	.9813	.9806	.9799
10	8.215 .9792	8.209	8.204 .9779	8.198 .9772	8.192 .9765	8.186	8.181	8.175	8.169	8.164
13	8.158	8.153	8.147	8.141	8.135	.9759 8.130	.9752 8.124	.9745 8.119	.9738 8,113	.9732 8.108
14	.9725	.9718	.9712	.9705	.9698	.9092	.9685	.9679	.9672	.9665
15	8.102 .9659	8.096 .9652	8.091 .9646	8.085	8.079 .9632	·8.074 .9626	8.069 .9619	8.064	8.058	8.052
	8.047	8.041	8.036	8.030	8.024	8.019	8.014	8.009	8.003	7.998
16	.9593 7.992	.9587 7.987	.9580 7.981	.9574 7.976	.9567 7.970	.9561 7.965	.9554 7.959	.9548 7.954	.9542	.9535
17	.992	.9522	.9516	.9509	.9503	.9497	.9490	7.954	7.949	.9471
	7.939	7,933	7.928	7.922	7.917	7.912	7.906	7.901	7.896	7.890
18	.9465	.9459 7.880	.9452	.9446 7.869	.9440 7.864	.9433	.9427	.9421	.9415 7.814	.9408 7.838
19	7.885	.9396	7.874	.809	.9377	7.859	7.854	7.819	.9352	.9346
	7.833	7.828	7.823	7.817	7.812	7.807	7.802	7.797	7.791	7.786
20	.9340 7.781	.9334 7.776	.9328 7.771	.9322 7.766	.9315 7.760	.9309 7.755	.9303 7.750	.9297 7.745	.9291 7.740	.9285 7.735
21	.9279	.9273	.9267	.9260	.9254	.9248	.9242	.9236	.9230	.9224
	7.730	7.725	7.720	7.715	7.710	7.705	7.700	7.695	7.690	7.685
22	.9218 7.680	.9212 7.675	.9206 7.670	.9200 7.665	.9194 7.660	.9188 7.655	.9182 7.650	.9176 7.645	.9170 7.640	.9165 7.635
23	.9159	.9153	.9147	.9141	.9135	.9129	.9123	.9117	.9111	.9106
	7.630	7.625	7.620	7.615	7.610	7.605	7.600	7.595	7.590	7.586
24	.9100 7.581	.9094 7.576	.9088 7.571	.9082 7.563	.9076 7.561	.9071	.9065 7.552	.9059 7.547	.9053 7.542	.9047 7.537
25	.9042	.9036	.9000	.9024	.9018	.9013	.9007	.9001	.8906	.8990
	7.533	7.528	7.523	7.518	7.513	7.509	7.504	7.499	7.495	7.490
26	.8984 7.485	.8978 7.480	.8973 7.475	.8967 7.471	.8961 7.465	.8956 7.461	.8950 7.456	.8944 7.451	.8939 7.447	.8933 7.442
27	.8827	.8922	.8916	.8911	.405	.8899	.8894	.8988	.8883	.8877
	7.437	7.433	7.428	7.424	7.419	7.414	7.410	7.405	7.400	7.395
28	.8871 7.390	.8936 7.386	.8860 7.381	.8855 7.377	.8849 7.372	.8814 7.368	.8838 7.363	.8833 7.359	.8827 7.354	.8822 7.350
29	.8813	.8811	.8805	.8800	.8794	.8789	.8783	.8778	.8772	.8767
	7.345	7.340	7.335	7.331	7.326	7.322	7.318	7.313	7.308	7.304
30	.8762 7.300	.8756 7.295	.8751 7.290	.8745 7.285	.8740 7.281	.8735 7.277	.8729 2.272	.8724 7.208	.8718 7.263	.8713 7.259
31	.8708	.8702	.8697	.8692	.8686	.8681	.8676	.8670	.8665	.8660
~	7.255	7.250	7.245	7.241	7.236	7.232	7.228	7.223	7.219	7.215
32	.8654 7.210	.8649 7.205	.8644 7.201	.8639	.8633 7.192	.8628 7.188	.8623 7.184	.8618 7.180	.8612 7.175	.8607 7.170
33	.8602	.8597	.8591	.8586	.8581	.8576	.8571	.8565	.8560	.8555
~	7.166	7.162	7.157	7.153	7.149	7.145	7.141	7.136	7.131	7.127
34	.8550 7.123	.8545 7.119	.8540 7.115	.8534 7.110	.8529 7.106	.8524	.8519 7.097	.8514 7.093	.8509 7.089	.8504 7.085
35	.8498	.8493	.8488	.8483	.8478	.8473	.8468	.8463	.8458	.8453
	7.080	7.076	7.071	7.067	7.063	7.059	7.055	7.051	7.046	7.042
36	.8418 7.038	.8443 7.034	.8438 7.030	.8433 7.025	.8428 7.021	.8423	.8418 7.013	.8413 7.009	.8408 7.005	.8403 7.001
37	.8398	.8393	.8388	.8383	.8378	.8373	.8368	.8363	.8358	.8353
-	6.996	6.992	6.988	6.984	6.980	6.976	6.971	6.967	6.963	6.959
38	.8348	.8343	.8338	.8333 6.942	.8328	.8324	.8319 6.931	.8314 6.926	.8309 6.922	.8304 6.918

	0	1	2	3	4	5	6	7	8	9
			· · · ·							
39	.8299 6.914	.8294 6.910	.8289 6.906	.8285 6.902	.8280 6.898	.8275 6.894	.8270 6.890	.8265 6.886	.8260 6.881	.8250 6.878
40	.8251	.8246	.8241	.8236	.8232	.8227	.8222	.8217	.8212	.8208
47	6.874	6.870	6.866	6.861	6.858 .8184	6.854 .8179	6.850 .8174	6.846 .8170	6.841 .8165	6.838 .8160
41	.8203 6.834	.8198 6.830	.8193 6.826	.8189 6.822	6.818	6.814	6,810	6.806	6.802	6.798
42	.8156	.8151	.8146	.8142	.8137	.8132	.8128	.8123	.8118	.8114
4.0	6.795	6.791	6.786	6.783	6.779	6.775	6.771	6.767	6.763	6.760
43	.8109	.8104	.8100	.8095	.8090	.8086	.8081	.8076	.8072	.8067
10	6.756	6.751	6.748	6.744	6.740	3.736	6.732	6.728	6.725	6.721
44	.8063	.8058	.8053	.8049	.8044	.8040	.8035	.8031	.8026	.8022
	6.717	6.713	6.709	6.706	6.701	6.698	6.694	6.691	6.868	6.683
45	.8017	.8012	.8008	.8003	.7999	.7994	.7990	.7985	.9781	.7976
	6.679	6.675	6.671	6.667	6.664	6.660	6.656	6.652	6.649	6.645
46	.7972	.7967	.7963	.7958	.7954	.7949	.7945	.7941	.7936	.7932
	6.641	6.637	6.634	6.630	6.626	6.623	6.619	6.616	6.611	6.608
47	.7927	.7923	.7918	.7914	.7909	.7905	.7901	.7896	.7892	.7887
10	6.604	6.601	6.596	6.593	6.589	6.586 .7861	6.582 .7857	6.578 .7852	6.575 .7848	6.571 .7844
48	.7883	.7879 6.564	.7874 6.560	.7870 6.556	.7865 6.552	6.549	6.546	6.542	6,538	6.535
49	6.567 .7839	.7835	.7831	.7826	.7822	.7818	.7813	.7809	.7805	.7800
49	6.531	6.527	6.524	6.520	6.517	3.513	6,500	6.506	6.502	6.498
50	.7796	.7792	.7788	.7783	.7779	.7775	.7770	.7766	.7762	.7758
~	6.495	6.492	6.488	6.484	6.481	3.477	6.473	6.470	6.467	6.463
51	.7753	.7749	.7745	.7741	.7736	.7732	.7728	.7724	.7720	.7715
	6.459	6.456	6.452	6.449	6.445	6.442	6.438	6.435	6.432	6.427
52	.7711	.7707	.7703	.7699	.7694	.7690	.7686	.7682	.7678	.7674
	6.424	6.421	6.417	6.414	6.410	6.407	6.403	6.400	6.397	6.393
53	.7669	.7665	.7661	.7657	.7653	.7649	.7645	.7640	.7636	.7632
	6.389	6.386	6.382	6.379	6.376	5.372	6.369	6.365	6.362	6.358
54	.7628	.7624	.7620	.7616	.7612	.7608	.7603 6.334	.7599	.7595	.7591
	6.355	6.352	6.348	6.345	6.342	3.338 .7567	0.334 .7563	6.331 .7559	6.327 .7555	6.324 .7551
55	.7587 6.321	.7583 6.317	.7579 6.314	.7575 6.311	.7571 6.307	3.304	6.301	6.297	6.294	6.291
56	.7547	.7543	.7539	.7535	.7531	.7527	.7523	.7519	.7515	.7511
00	6.287	6.284	6.281	6.277	6.274	3.271	6.267	6.264	6.261	6.257
57	.7507	.7503	.7499	.7495	.7491	.7487	.7483	.7479	.7475	.7471
0.	6.254	6.251	6.247	6.244	6.241	3.237	6.234	6.231	6.227	6.224
58	.7467	.7463	.7459	.7455	.7451	.7447	.7443	.7440	.7436	.7432
	6.221	6.217	6.214	62.11	6.207	5.204	6.201	6.198	6.195	6.191
59	.7428	.7424	.7420	.7416	.7412	.7408	.7405	.7401	.7397	.7393
	6.188	6.185	6.182	6.178	6.175	3.172	6.169	6.166	6.162	6.159
60	.7389	.7385	.7381	.7377	.7374	.7370	.7366	.7362	.7358	.7354
~	6.156	6.152	6.149	6.146	6.143	6.140	6.137	6.133	6.130	6.127
61	.7351	.7347	.7343	.7339	.7335	.7332	.7328	.7324	.7320	.7316
62	6.124 .7313	6.121 .7309	6.117 .7305	6.114 .7301	6.111 .7298	.7294	6.105 .7290	6.102 .7286	6.098 .7283	6.095 .7279
02	6.092	6.089	6.086	6.082	6.080	6.077	6.073	6.070	6.067	6.064
63	.7275	.7271	.7268	7264	.7260	.7256	.7253	.7249	.7245	.7242
00	6.061	6.057	6.055	6.052	6.048	6.045	6.042	6.039	6.036	6.033
64	.7238	.7234	.7230	.7227	.7223	.7219	.7216	.7212	.7208	.7205
~*	6.030	6.027	6.023	6.021	6.017	6.014	6.012	6.008	6.005	6.002
65	.7201	.7197	.7194	.7190	.7186	.7183	.7179	.7175	.7172	.7168
	5.999	5.996	5.993	5.990	5.987	5.984	5.981	5.977	5.975	5.972
66	.7165	.7161	.7157	.7154	.7150	.7146	.7143	.7139	.7136	.7132
	5.969	5.966	5.962	5,960	5.957	5.953	5.951	5.948	5.945	5.942
67	.7128	.7125	.7121	.7118	.7114	.7111	.7107	.7103	.7100	.7096
-	5.938	5.936	5.933	5.930	5.927	5.924	5.921	5.918	5.915	5.912

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON---Con. (MODULUS 141.5.)

BAUME', SPECIFIC GRAVITY AND POUNDS PER GALLON—Con. (MODULUS 141.5.)

	0	1	2	3	4	5	6	7	8	9
S	.7093	.7089	.7086	.7082	.7079	.7075	.7071	.7068	.7064	.706
	5,909	5.906	5.903	5.900	5.898	5.894	5.891	5.888	5.885	5.883
0	.7057	.7054	.7050	.7047	.7043	.7040	.7036	.7033	.7029	.702
	5.879	5.877	5.873	5.871	5.868	5.865	5.862	5.859	5.856	5.853
)	.7022	.7019	.7015	.7012	.7008	.7005	.7001	.6998	.6995	.699
	5.850	5.848	5.844	5.842	5.838	5.836	5.833	5.830	5.828	5.824
1	.6988	.6984	.6981	.6977	.6974	.6970	.6967	.6964	.6960	.695
-	5.822	5.818	5.816	5.813	5.810	5.807	5.804	5.802	5.798	5.796
	.6953	.6950	.6946	.6943	.6940	.6936	.6933	.6929	.6926	.692
2						.0900				
	5.793	5.790	5.787	5.784	5.782	5.778	5.776	5.773	5.770	5.768
3	.6919	.6916	.6912	.6909	.6906	.6902	.6899	.6896	.6892	.688
	5.764	5.762	5.758	5.756	5.753	5.750	5.748	5.745	5.742	5.739
Ł	.6886	.6882	.6879	.6876	.6872	.6869	.6866	.6862	.6859	.685
	5.737	5.733	5.731	5.728	5.725	5.723	5.720	5.717	5.714	5.712
5	.6852	.6849	.6846	.6842	.6839	.6836	.6832	.6829	.6826	.682
	5.708	5.706	5.703	5.700	5.698	5.695	5.692	5.689	5.687	5.684
	.6819	.6816	6813	.6809	.6806	.6803	.6800	.6796	.6793	.679
	5.681	5.678	5.676	5.673	5.670	5.668	5.665	5.662	5.659	5.657
					.6774	.6770	.6767	.6764	.6761	.675
	.6787	.6783	.6780	.6777						
	5.654	5.651	5.648	5.646	5.643	5.640	5.638	5.635	5.633	5.629
1	.6754	.6751	.6748	.6745	.6741	.6738	.6735	.6732	.6728	.672
	5.627	5.624	5.622	5.619	5.616	5.613	5.611	5.608	5.605	5.603
1	.6722	.6719	.6716	.6713	.6709	.6706	.6703	.6700	.6697	.669
	5.600	5.597	5.595	5.593	5.589	5.587	5.584	5.582	5.579	5.576
	.6690	.6687	.6684	.6681	.6678	.6675	.6671	.6668	.6665	.666
	5.573	5.571	5.568	5.566	5.563	5,561	5.558	5,555	5.553	5.550
	.6659	.6656	.6653	.6649	.6646	.6643	.6640	.6637	.6634	.663
		5.545		5.540	5.537	5.534	5.532	5.529	5.527	5.524
	5.548		5.543							
	.6628	.6625	.6621	.6618	.6615	.6612	.6609	.6606	.6603	.660
	5.522	5.519	5.516	5.513	5.511	5.508	5.506	5.503	5.501	5.498
	.6597	.6594	.6591	.6588	.6584	.6581	.6578	.6575	.6572	.656
	5.496	5.493	5.491	5.488	5.485	5.483	5.480	5.478	5.475	5.473
	.6566	.6563	.6560	.6557	.6554	.6551	.6548	.6545	.6542	.653
	5.470	5.468	5.465	5.463	5.460	5.458	5.455	5,453	5.450	5.448
	.6536	.6533	.6530	.6527	.6524	.6521	.6518	.6515	.6512	.650
	5.445	5.443	5.440	5.438	5.435	5.433	5.430	5.428	5.425	5.423
	.6506	.6503		.6497	.6494	.6491	.6488	.6485	.6482	.647
			.6500			5.408	5,405	5,403		
	5.420	5.418	5.415	5.419	5.410				5.400	5.398
	.6476	.6478	.6470	.6467	.6464	.6461	.6458	.6455	.6452	.644
	5.395	5.393	5.390	5.388	5.385	5.383	5.380	5.378	5.375	5.373
	.6446	.6144	.6441	.6438	.6435	.6432	.6429	.6426	.6423	.642
	5.370	5.368	5.366	5.363	5.361	5.358	5.356	5.353	5.351	5.349
1	.6417	.6414	.6411	.6409	.6406	.6403	.6400	.6397	.6394	,639
	5.346	5.344	5.341	5.339	5.237	5.334	5.332	5.329	5.327	5.324
	.6388	.6385	.6382	.6380	.6377	.6374	.6371	.6368	.6365	.636
	5.322	5.319	5.317	5.315	5.313	5.310	5.308	5,305	5.303	5.300
						.6345				
	.6360	.6357	.6354	.6351	.6348		.6342	.6340	.6337	.633
	5.299	5.296	5.294	5.291	5.289	5.286	5.284	5.282	5.279	5.277
	.6331	.6328	.6325	.6323	.6320	.6317	.6314	.6311	.6309	.6300
	5.274	5.272	5.269	5.268	5.265	5.263	5.260	5.258	5.256	5.254
	.6303	.6300	.6297	.6294	.6292	.6289	.6286	.6283	.6281	.6278
	5.251	5.249	5.246	5.244	5.242	5.239	5.237	5.234	5.233	5.230
	.6275	.6272	.6269	.6267	.6264	.6261	.6258	.6256	.6253	.6250
	5.228	5.225	5.223	5.221	5.219	5.216	5.214	5.212	5.209	5.207
	.6247	.6244	.6242	.6239	.6236	.6233	.6231	.6228	.6225	.622
						5.193	5.191	5.189	5.186	5.184
	5.204	5.202	5.200	5.198	5.195					
	.6220	.6217	.6214	.6212	.6209	.6206	.6203	.6201	.6198	.6198
	5.182	5.179	5.177	5.175	5.173	5.170	5.168	5.166	5.164	5.161
	.6193	.6190	.6187	.6184	.6182	.6179	.6176	.6174	.6171	.616
1	5.159	5.157	5.154	5.152	5.150	5.148	5.145	5.144	5.141	5.139
1	.6166	.6163	.6160	.6158	.6155	.6152	.6150	.6147	.6144	.614
1	5.137	5.134	5.132	5.130	5.128	5.125	5.124	5.121	5.119	5.116
1	.6139	.6136	.6134	.6131	.6128	.6126	.6123	.6120	.6118	.611
	·0100	5.112	5.110	5.108	5.105	5.104	5.101	5.099	5.097	5.094

[This table shows the degrees Baumé at 60° F of eils having, at the designated temperatures, the observed degrees Baumé indicated. For example, if the observed degrees Baumé is 20.0 at 78° F, the true degrees Baumé at 60° F will be 19.0. Intermediate values not given in the table may be conveniently interpolated. For example, if the observed degrees Baumé is 20.4 at 78° F, the true degrees Baumé at 60° F will be 19.4. The headings "Observed degrees Baumé" and "Observed temperature" signify the true indication of the hydrometer and the true temperature of the oil—that is, the observed readings corrected, it necessary, for instrumental errors.]

				Obse	erved de;	grees Ba	umé .			
Observed temperature in	17.0	18.0	19.0	20.0	21.0	22.0	23.0	24.0	25.0	26.0
		·	Co	rrespond	ing degr	ees Bau	mé at 60°	F		
30 32 34 36	18.6 18.6 18.5 18.3 18.2	19.7 19.6 19.5 19.4 19.3	20.7 20.6 20.5 20.4 20.3	21.7 21.6 21.5 21.4 21.3	22.7 22.6 22.5 22.4 22.3	23.7 23.6 23.5 23.4 23.3	24,8 24.7 24.6 24.5 24.4	25.8 25.7 25.6 25.5 25.4	26.9 26.8 26.7 26.5 26.4	27.9 27.8 27.7 27.5 27.4
38	18.1 18.0 17.9 17.8 17.6	19.3 19.1 19.0 18.9 18.8 18.7	20.3 20.1 20.0 19.9 19.8 19.7	21.2 21.1 20.9 20.8 20.7	22. 2 22. 1 21. 9 21. 8 21. 7	23.2 23.1 22.9 22.8 22.7	24.2 24.1 23.9 23.8 23.7	25.2 25.1 24.9 24.8 24.7	26.2 26.1 26.0 25.9 25.8	27.2 27.1 27.0 26.9 26.8
50 52 54 56 58	17.5 17.4 17.3 17.2	18.6 18.5 18.3 18.2 18.1	19.6 19.5 19.3 19.2 19.1	20.6 20.5 20.3 20.2 20.1	21.6 21.5 21.3 21.2 21.1	22.6 22.5 22.3 22.2 22.1	23.6 23.5 23.3 23.2 23.1	24.6 24.5 24.3 24.2 24.1	25.6 25.5 25.4 25.3 25.1	26.6 26.5 26.4 26.3 26.1
60 62 64 66 68	17.0	18.0 17.9 17.8	19.0 18.9 18.8 18.7 18.6	20.0 19.9 19.8 19.7 19.5	21.0 20.9 20.8 20.7 20.5	22.0 21.9 21.8 21.7 21.5	23.0 22.9 22.8 22.7 22.5	24.0 23.9 23.8 23.7 23.5	25.0 24.9 24.7 24.6 24.5	26.0 25.9 25.7 25.6 25.5
70 72 74 76 78		17.5 17.4 17.2 17.2 17.1	18.5 18.4 18.2 18.1 18.0	19.4 19.3 19.2 19.1 19.0	20.4 20.3 20.2 20.1 19.9	21.4 21.3 21.2 21.1 20.9	22.4 22.3 22.2 22.1 21.9	23.4 23.3 23.2 23.1 22.9	24.4 24.3 24.1 24.0 23.9	25.4 25.3 25.1 25.0 24.9
80 82 84 86 88			17.9 17.8 17.7 17.6 17.5	18.9 18.8 18.7 18.6 18.4	19.8 19.7 19.6 19.5 19.4	20.8 20.7 20.6 20.5 20.4	21.8 21.7 21.6 21.5 21.3	22.8 22.7 22.6 22.5 22.3	23.8 23.7 23.5 23.4 23.3	24.8 24.7 24.5 24.4 24.3
90 92 94 96 98			17.0	18.3 18.2 18.1 18.0 17.9	19.3 19.2 19.1 19.0 18.8	20.3 20.2 20.1 20.0 19.8	21.2 21.1 21.0 20.9 20.8	22.2 22.1 22.0 21.9 21.8	23.2 23.1 23.0 22.8 22.7	24.2 24.1 24.0 23.8 23.7
100 102 104 106 108				17.8 17.7 17.6 17.5 17.3	18.7 18.6 18.5 18.4 18.2	19.7 19.6 19.5 19.4 19.2	20.7 20.5 20.4 20.3 20.2	21.7 21.5 21.4 21.3 21.2	22.6 22.5 22.4 22.3 22.2	23.6 23.5 23.4 23.3 23.1
110. 112. 114. 116. 118.				17.2 17.1 17.0	18.1 18.0 17.9 17.8 17.7	19.1 19.0 18.9 18.8 18.7	20.1 20.0 19.9 19.8 19.6	21.1 21.0 20.9 20.8 20.6	22.0 21.9 21.8 21.7 21.5	23.0 22.9 22.8 22.7 22.5
120						18.6	19.5	20.5	21.4	22.4

A CARA				Obs	erved de	grees Ba	umé			
Observed temperature in	27.0	28.0	29.0	30.0	31.0	32.0	33.0	34.0	35.0	36.0
	1		Ca	rrespond	ling degr	ees Bau	mé at 60'	F		
30	29.0	30.0	31.0	32.0	33.1	34.1	35.2	36.2	37.3	38.3
32	28.8	29.8	30.9	31.9	33.0	34.0	35.0	36.0	37.1	38.1
34	28.7	29.7	30.8	31.8	32.8	33.8	34.8	35.8	36.9	38.0
36	28.5	29.5 29.4	30.6	31.6	32.7	33.7 33.5	34.7	35.7	36.8	37.8
30	60.4	69.4	30.3	51.5	34.3	33.3	34.5	33.5	30.0	31.1
40	28.3	29.3	30.4	31.4	32.4	33.4	34.4	35.4	36.5	37.5
42	28.2	29.2	30.2	31.2	32.2	33.2	34.3	35.3	36.3	37.3
44	28.1 27.9	29.1 28.9	30.1 29.9	31.1	32.1	33.1 32.9	34.2 34.0	35.2	36.2	37.2
48	27.8	28.8	29.8	30.8	31.8	32.8	33.9	34.9	35.9	36.9
		-				1.3.2		1000		
50	27.6	28.6	29.7	30.7	31.7	32.7	33.7	34.7	35.7	36.7
52	27.5	28.5	29.6	30.6	31.6	32.6	33.6	34.6	35.6	36.6
54	27.3	28.3	29.3	30.3	31.3	32.3	33.3	34.3	35.3	36.3
58	27.1	28.1	29.1	30.1	31.1	32.1	33.1	34.1	35.1	36.1
								1.		
60	27.0	28.0	29.0 28.9	30.0	31.0	32.0	33.0 32.9	34.0	35.0	36.0
64	26.7	27.7	28.7	29.7	30.7	31.7	32.7	33.7	34.7	35.9
66	26.6	27.6	28.6	29.6	30.6	31.6	32.6	33.6	34.6	35.6
68	26.5	27.5	28.4	29.4	30.4	31.4	32.4	33.4	34.4	35.4
70	26.4	27.4	28.3	29.3	30.3	31.3	32.2	33.2	34.2	35.2
72	26.3	27.3	28.2	29.2	30.2	31.2	32.1	33.1	34.1	35.1
74	26.1	27.1	28.1	29.1	30.1	31.1	32.0	33.0	33.9	34.9
76	26.0	27.0	27.9	28.9	29.9	30.9	31.8	32.8	33.8	34.8
78	25.8	26.8	27.8	28.8	29.8	30.8	31.7	32.7	33.6	34.6
80	25.7	26.7	27.7	28.7	29.7	30.7	31.6	32.6	33.5	34.5
82	25.6	26.6	27.6	28.6	29.5	30.5	31.5	32.5	33.4	34.4
84	25.5	26.5	27.5	28.5	29.4	30.4	31.3	32.3	33.2	34.2
86	25.4	26.4	27.3	28.3	29.2	30.2	31.2 31.0	32.2	33.1 33.0	34.1 34.0
00		20.2	61.6	20.2	. 63. 1	50.1	51.0	36.0	33.0	34.0
90	25.1	26.1	27.0	28.0	29.0	30.0	30.9	31.9	32.9	33.9
92	25.0	26.0	26.9	27.9	28.9	29.9	30.8	31.8	32.7	33.7
94	24.9	25.9	26.8	27.8	28.8 28.6	29.8 29.6	30.7	31.6	32.6 32.5	33.6 33.5
98	24.6	25.6	26.6	27.6	28.5	29.5	30.4	31.4	32.3	33.3
00	24.5	25.5	26.4	27.4	28.3	29.3 29.2	30.3 30.2	31.3 31.2	32.2	33.2
04	24.3	25.3	26.2	27.1	28.1	29.2	30.0	31.0	31.9	32.9
.06	24.2	25.2	26.1	27.0	28.0	29.0	29.9	30.9	31.8	32.7
.08	24.0	25.0	25.9	26.9	27.8	28.8	29.7	30.7	31.6	32.6
10	23.9	24.9	25.8	26.8	27.7	28.7	29.6	30.6	31.5	32.5
12	23.8	24.8	25.7	26.7	27.6	28.6	29.5	30.4	31.3	32.3
14	23.7	24.7	25.6	26.6.	27.5.	28.4	29.3	30.3	31.2	32.2
16	23.6	24.6	25.5	26.4	27.3	28.3	29.2	30.2	31.1	32.1
	23.4	24.4	25.3	26.3	27.2	28.2	29.1	30.1	31.0	32.0
20	23.3	24.3	25.2	26.2	27.1	28.1	29.0	30.0	30.9	.31.9

				Obs	erved de	grees Ba	umé			
Observed temperature in	37.0	38. 0	39. 0	40. 0	41.0	42.0	43. 0	44.0	45. 0	46. 0
			Co	rrespond	ling degr	ees Bau	me at 60°	F		
30.	39. 3	40. 3	41. 4	42. 4	43.5	44. 5	45.6	46. 6	47.7	48. 7
32.	39. 2	40. 2	41. 3	42. 3	43.4	44. 3	45.4	46. 4	47.5	48. 5
34.	39. 0	40. 0	41. 1	42. 1	43.2	44. 2	45.3	46. 3	47.3	48. 3
36.	38. 9	39. 9	41. 0	42. 0	43.1	44. 0	45.1	46. 1	47.2	48. 2
38.	38. 7	39. 7	40. 8	41. 8	42.9	43. 9	45.0	46. 0	47.0	48. 0
40	38. 5	39. 5	40. 6	41.6	42.7	43. 7	44. 8	45. 8	46. 8	47.8
	38. 4	39. 4	40. 5	41.5	42.5	43. 5	44. 6	45. 6	46. 6	47.6
	38. 2	39. 2	40. 3	41.3	42.4	43. 4	44. 4	45. 4	46. 4	47.4
	38. 1	39. 1	40. 1	41.1	42.2	43. 2	44. 2	45. 2	46. 2	47.2
	37. 9	38. 9	39. 9	40.9	42.0	43. 0	44. 1	45. 1	46. 1	47.1
50 52 54 56 58	37. 8 37. 6 37. 4 37. 3 37. 1	38. 8 38. 6 38. 4 38. 3 38. 1	39. 8 39. 6 39. 5 39. 3 39. 3 39. 1	40. 8 40. 7 40. 5 40. 3 40. 1	41.8 41.7 41.5 41.3 41.1	42.8 42.6 42.5 42.2 42.1	43.9 43.7 43.5 43.3 43.1	44. 9 44. 7 44. 5 44. 3 44. 1	45. 9 45. 7 45. 5 45. 3 45. 2	46. 9 46. 7 46. 5 46. 3 46. 2
50	37. 0	38. 0	39. 0	40. 0	41.0	42.0	43. 0	44.0	45.0	46.0
52	36. 9	37. 9	38. 9	39. 9	40.9	41.9	42. 9	43.9	44.9	45.9
54	36. 7	37. 7	38. 7	39. 7	40.7	41.7	42. 7	43.7	44.7	45.7
56	36. 6	37. 6	38. 6	39. 5	40.5	41.5	42. 5	43.5	44.5	45.5
58	36. 4	37. 4	38. 4	39. 4	40.4	41.4	42. 4	43.3	44.3	45.3
70	36. 2	37. 2	38. 2	39. 2	40. 2	41. 2	42.2	43.1	44. 1	45.1
72	36. 1	37. 1	38. 1	39. 1	40. 0	41. 0	42.0	43.0	44. 0	45.0
74	35. 9	36. 9	37. 9	38. 9	39. 8	40. 8	41.8	42.8	43. 8	44.8
76	35. 8	36. 8	37. 8	38. 7	39. 7	40. 7	41.7	42.7	43. 6	44.6
78	35. 6	36. 6	37. 6	38. 6	39. 5	40. 5	41.5	42.5	43. 4	44.4
30	35. 5	36. 5	37.5	38.5	39. 4	40. 4	41.3	42.3	43. 2	44. 2
32	35. 3	36. 3	37.3	38.3	39. 2	40. 2	41.2	42.2	43. 1	44. 1
34	35. 2	36. 2	37.2	38.2	39. 1	40. 1	41.0	42.0	42. 9	43. 9
86	35. 1	36. 1	37.0	38.0	38. 9	39. 9	40.9	41.9	42. 8	43. 8
88	34. 9	35. 9	36.9	37.9	38. 8	39. 8	40.7	41.7	42. 6	43. 6
90	34. 8	35. 8	36. 7	37.7	38. 6	39. 6	40. 5	41.5	42.5	43.5
	34. 6	35. 6	36. 6	37.6	38. 5	39. 5	40. 4	41.4	42.3	43.3
	34. 5	35. 5	36. 4	37.4	38. 3	39. 3	40. 2	41.2	42.2	43.2
	34. 4	35. 4	36. 3	37.3	38. 2	39. 2	40. 1	41.1	42.0	43.0
	34. 2	35. 2	36. 1	37.1	38. 0	39. 0	39. 9	40.9	41.8	42.8
00	34. 1 33. 9 33. 8 33. 6 33. 5	35. 1 34. 9 34. 8 34. 6 34. 5	36. 0 35. 8 35. 7 35. 5 35. 4	37. 0 36. 8 36. 7 36. 5 36. 4	37.9 37.7 37.6 37.4 37.3	38. 9 38. 7 38. 6 38. 4 38. 3	39. 8 39. 6 39. 5 39. 3 39. 3 39. 2	40. 7 40. 6 40. 4 40. 3 40. 1	41.6 41.5 41.3 41.2 41.0	42. 6 42. 5 42. 3 42. 2 42. 0
10	33. 4	34. 4	35. 3	36. 3	37. 2	38. 1	39. 0	40. 0	40. 9	41.8
12	33. 2	34. 2	35. 1	36. 1	37. 0	38. 0	38. 9	39. 8	40. 7	41.6
14	33. 1	34. 1	35. 0	36. 0	36. 9	37. 8	38. 7	39. 7	40. 6	41.5
16	33. 0	34. 0	34. 9	35. 9	36. 8	37. 7	38. 6	39. 5	40. 4	41.4
18	32. 9	33. 9	34. 8	35. 7	36. 6	37. 5	38. 4	39. 4	40. 3	41.2
20	32.8	33.7	34.6	35.6	36.5	37.4	38. 3	39.2	40.1	41.0

				Obs	erved de	grees Ba	ume			
Observed temperature in	47.0	48.0	49.0	50. 0	51.0	52.0	53.0	54.0	55.0	56.0
			Ca	rrespond	ling degr	ees Bau	mé at 60°	F		-
30	49.8	50.8	51.9	53.0	54.1	55.1	56.2	57.3	58.4	59.
4	49.6	50.6 50.4	51.7 51.5	52.8 52.6	53.9 53.7	54.9	56.0 55.8	57.1 56.8	58.2 57.9	59. 2 58. 9
6	49.3	50.3	51.4	52.4	53.5	54.5	55.6	56.6	57.7	58.7
8	49.1	50.1	51.2	52.2	53.3	54.3	55.4	56.4	57.5	58. 5
0	48.9	49.9	51.0	52.0	53:0	54.1	55.2	56.2	57.2	58. 2
2	48.7	49.7	50.8	51.8	52.8	53.8	54.9	56.0	57.0	58. 0
4 5	48.5 48.3	49.5 49.3	50.6	51.6 51.4	52.6 52.4	53.6 53.4	54.7 54.5	55.7 55.5	56.8	57.8
8	48.1	49.1	50.2	51.2	52.2	53.2	54.2	55.2	56.3	57.3
Contraction of the	47.9	48.9	50.0	51.0	52.0	53.0	54.0	55.0	56, 1	. 57. 1
2	47.9	48.9	49.8	50.8	51.8	52.8	53.8	54.8	55.9	56.9
	47.6	48.6	49.6	50.6	51.6	52.6	53.6	54.6	55.6	56.
	47.4	48.4	49.4	50.4	51.4	52.4	53.4	54.4 54.2	55.4	56.
	47.2	48.2	49.6	50. 2	31.6	366	33.6	34. 6	33. 4	30.
)	47.0	48.0	49.0	50.0	51.0	52.0	53.0	54.0	55.0	56.
	46.9	47.9	48.8	49.8	50.8	51.8	52.8	53.8 53.6	54.8	55.
	46.7	47.7	48.6	49.6 49.4	50.6 50.4	51.6	52.6	53. 6	54.6	55.
8	46. 3	47.3	48.3	49.3	50.3	51.3	52. 2	53.2	54.2	55.
0	46.1	47.1	48.1	49.1	50.1	51.1	52.0	53.0	54.0	55.
2	46.0	47.0	47.9	48.9	49.9	50.9	51.8	52.8	53.8	54.
·····	45.8	46.8	47.7	48.7	49.7	· 50.7	51.6	52.6	53.5	54.
6 8	45.6	46.6	47.5	48.5 48.3	49.5	50.5	51.4	52.4 52.2	53.3 53.1	54.
0	45.2	46.2	47.2	48.2	49.1	50.1	51.0	52.0	52.9	53.9
2	45.1 44.9	46.1 45.9	47.0	48.0	48.9	49.9	50.8 50.6	51.8	52.7 52.5	53.
6	44.7	45.7	46.6	47.6	48.5	49.5	50.4	51.4	52.3	53.
8	44.5	45.5	46.4	47.4	48.3	49.3	50.2	51.2	52.1	53.
	44.4	45.4	46.3	47.3	48.2	49.2	50.1	51.0	51.9	52
2	44.2	45.2	46.1	47.1	48.0	49.0	49.9	50.9	51.8	52.
4 5	44.1	45.1	46.0	46.9	47.8	48.8	49.7	50.7	51.6	52.
8	43.9 43.7	44.9	45.8	46.7	47.6	48.6	49.5	50.5 50.3	51.4	52.
)									51.0	51.
2	43.5	44.5	45.4	46.4	47.3	48.3	49.2	50.1 49.9	50.8	51.
ŧ	43.2	44.1	45.0	46.0	46.9	47.9	48.8	49.7	50.6	51.
6	43.1	44.0	44.9	45.8	46.7	47.7	48.6	49.5	50.4	51.
B	42.9	43.9	44.8	45.7	46.6	47.5	48.4	49.4	50.3	51.
0	42.7	43.7	44.6	45.6	46.5	47.4	48.3	49.2	50.1	51.
2	42.5	43.5	44.4	45.4	46.3	47.2	48.1	49.0	49.9	50.
4	42.4	43.4	44.3	45.3	46.2	47.1 46.9	48.0	48.8	49.7	50.
8	42.1	43.1	44.0	45.1	40.0	46.7	47.6	48.4	49.3	50.
						1.0				
0	41.9	42.9	43.8	44.7	45.6	46.5	47.4	48.2	49.1	50.

379

				Obs	erved de	grees Ba	umé			
Observed temperature in	57.0	58.0	59.0	60.0	61.0	62.0	63.0	64.0	65.0	66.0
•	2		C	orrespond	ling degr	ees Bau	mé at 60°	F		
j	60.5	61.6	62.7	63.7	64.8	65.8	66.9	67.9	69.0	70.0
2	60.3 60.0	61.3	62.4 62.1	63.4 63.1	64.5 64.2	65.5	66.6 66.3	67.7 67.4	68.8 68.5	69.8 69.5
	59.8	60.8	61.9	62.9	64.0	65.0	66.1	67.1	68.2	69.2
	59.5	60.5	61.6	62.6	63.7	64.7	65.8	66.8	67.9	68.9
	59.3	60.3	61.4	62.4	63.5	64.5	65.5	66.5	67.6	68.0
	59.1	60.1	61.2	62.2	63.3	64.3	65.3	66.3	67.4	68.4
	58.9	59.9	61.0	62.0	63.0	64.0 63.7	65.0	66.0	67.1	68.1
	58.6 58.4	59.6 59.4	60.7 60.4	61.7	62.7	63.5	64.5	65.8 65.5	66.8 66.5	67.8
	58.1 57.9	59.1 58.9	60.2 60.0	61.2 61.0	62.2 62.0	63.2 63.0	64.2 64.0	65.2 65.0	66.2 66.0	67.
	57.7	58.7	59.8	60.8	61.8	62.8	63.8	64.8	65.8	66.8
	57.5	58.5	59.5	60.5	61.5	62.5	63.6	64:6	65.6	66.
	57.3	58.3	59.3	60.3	61.3	62.3	63.3	64.3	65.3	66. 3
	57.0	58.0	59.0	60.0	61.0	62.0	63.0	64.0	65.0	66.
	56.8	57.8	58.8	59.8	60.8	61.8	62.7	63.7	64.7	65.
	56.6	57.6	58.6	59.6	60.5	61.5	62.5	63.5	64.5	65.
	56.4 56.1	57.4 57.1	58.3	59.3 59.1	60.3	61.3 61.1	62.3 62.1	63.3 63.1	64.2 64.0	65.0
2	55.9 55.7	56.9 56.7	57.9 57.7	58.9 58.7	59.8 59.6	60.8 60.6	61.8 61.6	62.8 62.6	63.8 63.5	64.8
	55.5	56.5	57.4	58.4	59.3	60.3	61.3	62.3	63.2	64.
5	55.3	56.3	57.2	58.2	59.1	60.1	61.0	62.0	63.0	64.1
3	55.0	56.0	57.0	58.0	58.9	59.9	69.8	61.8	62.8	63.8
	54.8	55.8	56.8	57.8	58.7	59.7	60.6	61.6	62.6	63.
	54.6	55.6	56.5	57.5	58.4	59.4	60.4	61.4	62.3	63.
	54.4 54.2	55.4 55.2	56.3 56.1	57.3	58.2 58.0	59.2 59.0	60.1 59.9	61.1 60.9	62.0 61.8	63.0
	54.0	55.0	55.9	56.9	57.8	58.8	59.9	60.6	61.5	62.
	53.8	54.8	55.7	56.7	57.6	58.6	59.5	60.4	61.3	62.3
	53.6	54.6	55.5	56.5	57.4	58.4	59.3	60.2	61.1	62.
	53.4	54.3	55.2	56.2	57.1	58.1	59.0	59.9	60.8	61.4
	53.2 53.0	54.1 53.9	55.0 54.8	56.0 55.8	56.9 56.7	57.9	58.8 58.5	59.7 59.5	60.6	61.
	52.8 52.6	53.7 53.5	54.6 54.4	55.6 55.4	56.5	57.4	58.3 58.1	59.3 59.0	60.2 59.9	61.
	52.0	53.3	54.2	55.2	56.1	57.0	57.9	58.8	59.9	60.
	52.2	53.1	54.0	55.0	55.9	56.8	57.7	58.6	59.5	60.
••••••	52.1	53.0	53.9	54.8	55.7	56.6	57.5	58.4	59.3	60.3
	51.9	52.8	53.7	54.6	55.5	56.4	57.3	58.2	59.1	60.0
2	51.7	52.6	53.5	54.4	55.2	56.2	57.1	58.0	58.9	59.8
	51.5 51.3	52.4 52.2	53.3 53.1	54.2 54.0	55.1 54.9	56.0 55.8	56.9	57.8	58.7 58.4	59.0 59.1
	51.5	52.0	52.9	53.8	54.9	55.6	56.5	57.4	58.2	59.
	50.9	51.8	52.7	53.6	54.5	55.4	56.3	57.2	58.0	58.9

				Obs	erved de	grees Ba	umé			
Observed temperature in	67.0	68.0	69.0	70.0	71.0	72.0	73.0	74.0	75.0	76.0
		1	Ca	rrespond	ling degi	ees Bau	mé at 60°	F		
30	71.1	72.1	73.2	74.3	75.4	76.4	77.5	78.5	79.6	80.7
32	70.9	71.9	73.0 .	74.0	75.1	76.1	77.2	78.2	79.3	80.4
34	70.6	71.6	72.7	73.7	74.8	75.8	76.9	77.9	79.0	80.1
36	70.3	71.3	72.4	73.4	74.5	75.5	76.6	77.6	78.7 78.4	79.7
40	69.7	70.7	71.8	72.8	73.9	74.9	76.0	77.0	78.1	79.1
42	69.4	70.4	71.5	72.5	73.6	74.6	75.7	76.7	77.8	78.8
44	69.1	70.1	71.2	72.2	73.3	74.3	75.4	76.4	77.5	78.5
46	68.8 68.6	69.8 69.6	70.9 70.6	71.9 71.6	73.0	74.0	75.1 74.8	76.1 75.8	77.1 76.8	78.1
50	68.3	69.3	70.4	71.4	72.5	73.5	74.5	75.5	76.5	77.5
52	68.0	69.0	70.1	71.1	72.2	73.2	74.2	75.2	76.2	77.2
54	67.8	68.8	69.9	70.9	71.9	72.9	73.9	74.9	75.9	76.9
56	67.6 67.3	68.6 68.3	69.6 69.3	70.6	71.6	72.6	73.6	74.6	75.6	76.6
60	67.0	68.0	69.0	70.0	71.0	72.0	73.0	74.0	75.0	76.0
62	66.7	67.7	68.7	69.7	70.7	71.7	72.7	73.7	74.7	75.7
64	66.4	67.4	68.4	69.4	70.4	71.4	72.4	73.4	74.4	75.4
66	66.2 66.0	67.2 67.0	68.2 67.9	69.2 68.9	70.1 69.8	71.1 70.8	72.1 71.8	73.1 72.8	74.1 73.8	75.1
70	65.7	66.7	67.6	68.6	69.5	70.5	71.5	72.5	73.5	74.5
72	65.4	66.4	67.4	68.4	69.3	70.3	71.2	72.2	73.2	74.2
74	65.2	66.2	67.2	68.2	69.1	70.1	71.0	72.0	72.9	73.9
76	64.9 64.7	65.9 65.6	66.9 66.6	67.9 67.6	68.8 68.5	69.8 69.5	70.8	71.8	72.7	73.7
80	64.5	65.4	66.4	67.4	68.3	69.3	70.2	71.2	72.1	73.1
82	64.2	65.2	66.1	67.1	68.0	69.0	69.9	70.9	71.8	72.8
84	63.9	64.9	65.8	66.8	67.7	68.7	69.6	70.6	71.5	72.5
86	63.7 63.4	64.7 64.4	65.6 65.3	66.6 66.3	67.5 67.2	68.4 68.2	69.3 69.1	70.3	71.3 71.0	72.3
90	63.2	64.2	65.1	66.1	67.0	68.0	68.9	69.9	70.8	71.7
92	63.0	64.0	64.9	65.8	65.7	67.7	68.6	69.6	70.5	71.4
94	62.7 62.5	63.7 63.5	64.6	65.6	66.5	67.4	68.3	69.3 69.0	70.2	71.1
96 98	62.2	63.2	64.4 64.1	65.1	66.3 66.0	66.9	67.8	68.8	69.7	70.8
	62.0	63.0	63.9	64.9	65.8	66.7	67.6	68.5	69.4	70.4
02	61.8	62.8	63.7	64.6	65.5	66.4	67.3	68.2	69.1	70.1
04	61.6 61.3	62.5	63.4	64.3	65.2	66.1	67.0	67.9	68.,8 68.6	69.8 69.5
108	61. 1	62.0	63.2 62.9	64.1 63.8	65.0 64.8	65.9 65.7	66.8 66.6	67.5	68.4	69.3
10	60.9	61.8	62.7	63.6	64.5	65.4	66.3	67.2	68.1	69.0
12	60.7	61.6	62.5	63.3	64.2	65.2	66.1	67.0	67.8	68.7
114	60.5 60.2	61.4	62.3	63.1 62.9	64.0 63.8	64.9 64.7	65.8	66.7 66.5	67.6	68.5 68.3
118	60.0	60.9	61.8	62.7	63.6	64.5	65.4	66.8	67.1	68.0
20	59.8	60.7	61.6	62.5	63.3	64.2	65.1	66.0	66.8	67.7

381

Observed temperature in	Observed degrees Baumé											
	77. 0	78.0	79. 0	80.0	81.0	82.0	83. 0	84. 0	85.0	86. 0		
	Corresponding degrees Baumé at 60° F											
30 32 34 36 38	81.8 81.5 81.2 80.8 80.5	82.9 82.6 82.2 81.9	84. 0 83. 7 83. 3 83. 0 82. 6	85. 0 84. 7 84. 3 84. 0 83. 6	86. 1 85. 8 85. 4 85. 1 84. 7	87.1 86.8 86.4 86.1 85.7	88. 2 87. 9 87. 5 87. 2 86. 8	89. 3 89. 0 88. 6 88. 2 87. 8	90. 4 90. 1 89. 7 89. 3 88. 9	91.5 91.1 90.7 90.3 89.9		
40 42 44 46 48	80. 3 80. 1 79. 8 79. 5 79. 2 78. 9	81.5 81.1 80.8 80.5 80.2 79.9	82. 2 81. 9 81. 6 81. 3 81. 0	83. 2 82. 9 82. 6 82. 3 82. 0	84. 3 84. 0 83. 7 83. 4 83. 0	85. 3 85. 0 84. 7 84. 4 84. 0	86. 4 86. 1 85. 8 85. 4 85. 1	87.4 87.1 86.8 86.5 86.1	88.5 88.2 87.8 87.5 87.1	89.5 89.2 88.8 88.5 88.1		
50	78.6	79.6	80. 6	81.6	82.6	83. 6	84. 7	85. 7	86. 7	87. 7		
52,	78.2	79.2	80. 3	81.3	82.3	83. 3	84. 3	85. 3	86. 3	87. 3		
54	77.9	78.9	79. 9	81.0	82.0	83. 0	84. 0	85. 0	86. 0	87. 0		
56	77.6	78.6	79. 6	80.6	81.6	82. 6	83. 7	84. 7	85. 7	86. 7		
58	77.3	78.3	79. 3	80.3	81.3	82. 3	83. 3	84. 3	85. 3	86. 3		
60	77.0	78.0	79.0	80. 0	81.0	82. 0	83.0	84. 0	85. 0	86. 0		
62	76.7	77.7	78.7	79. 7	80.7	81. 7	82.7	83. 7	84. 7	85. 7		
64	76.4	77.4	78.4	79. 4	80.4	81. 4	82.3	83. 4	84. 3	85. 3		
66	76.1	77.1	78.1	79. 1	80.0	81. 0	82.0	83. 0	84. 0	85. 0		
68	75.8	76.8	77.7	78. 7	79.7	80. 7	81.7	82. 7	83. 7	84. 7		
70	75.5	76.5	77.4	78.4	79. 4	80. 4	81.4	82. 4	83. 3	84. 3		
72	75.2	76.2	77.1	78.1	79. 1	80. 1	81.1	82. 1	83. 0	84. 0		
74	74.9	75.9	76.8	77.8	78. 8	79. 8	80.7	81. 7	82. 7	83. 7		
76	74.6	75.6	76.5	77.5	78. 4	79. 4	80.4	81. 4	82. 4	83. 4		
78	74.3	75.3	76.2	77.2	78. 1	79. 1	80.1	81. 1	82. 0	83. 0		
80	74.0	75.0	75.9	76. 9	77.8	78.8	79.8	80. 8	81. 7	82. 7		
82	73.7	74.7	75.6	76. 6	77.5	78.5	79.4	80. 4	81. 3	82. 3		
84	73.4	74.5	75.3	76. 3	77.2	78.2	79.1	80. 1	81. 0	82. 0		
86	73.2	74.1	75.0	76. 0	76.9	77.9	78.8	79. 8	80. 7	81. 7		
88	72.9	73.9	74.8	75. 8	76.7	77.6	78.5	79. 5	80. 4	81. 4		
90	72.6	73.6	74. 5	75. 5	76. 4	77. 3	78. 2	79. 2	80. 1	81. 1		
92	72.3	73.3	74. 2	75. 2	76. 1	77. 0	77. 9	78. 9	79. 8	80. 8		
94	72.0	73.0	73. 9	74. 9	75. 8	76. 7	77. 6	78. 6	79. 5	80. 5		
96	71.7	72.7	73. 6	74. 6	75. 5	76. 4	77. 3	78. 3	79. 2	80. 2		
98	71.5	72.4	73. 3	74. 3	75. 2	76. 1	77. 0	78. 0	78. 9	79. 8		
100	71. 2	72.1	73.0	74.0	74. 9	75.8	76. 7	77.6	78.5	79.5		
102	71. 0	71.9	72.8	73.7	74. 6	75.5	76. 4	77.3	78.2	79.2		
104	70. 7	71.6	72.5	73.4	74. 3	75.2	76. 1	77.0	77.9	78.8		
106	70. 4	71.3	72.2	73.1	74. 0	74.9	75. 8	76.7	77.6	78.5		
108	70. 1	71.0	71.9	72.8	73. 7	74.6	75. 5	76.4	77.3	78.2		
110	69.8	70. 7	71.6	72.5	73. 4	74. 3	75. 2	76. 1	77.0	77.9		
112	69.6	70. 5	71.4	72.3	73. 2	74. 1	74. 9	75. 8	76.7	77.6		
114	69.4	70. 3	71.2	72.1	72. 9	73. 8	74. 6	75. 5	76.4	77.3		
116	69.1	70. 0	70.9	71.8	72. 6	73. 5	74. 3	75. 2	76.1	77.0		
118	68.8	69. 7	70.6	71.5	72. 3	73. 2	74. 0	74. 9	75.8	76.7		
120	68. 5	69.4	70.3	71. 2	72. 0	72. 9	73. 7	74.6	75.5	76.4		

Observed temperature in °F	Observed degrees Baumé											
	87.0	88.0	89.0	90.0	91.0	92. 0	93.0	94.0	95.0	96.0		
	1	-	Ca	rrespond	ling degr	ees Bau	mé at 60	F				
30	92.6	93.6	94.7	95.7								
32	92.2	93. 2	94.3	95.3		12.00		1	100			
34	91.8	92.9	93.9 93.6	94.9	95.9				1.1			
36	91.4 91.0	92.5	93. 0	94.6 94.2	95.6 95.2		120					
40	90.6	91.7	92.8	93.8	94.9	95.9				1.		
42	90.3	91.3	92.4	93.4	94.5	95.5	1					
44	89.9	90.9	92.0	93.0	94.1	95.1	96.1	1				
46	89.6 89.2	90.6	91.7 91.3	92.7 92.3	93.7 93.3	94.7 94.3	95.7 95.3					
								1.1	1.000	1.0		
50	88.8	89.8	90.9	91.9	92.9	93.9	94.9	95.9				
52	88.4 88.0	89.4 89.0	90.5	91.5 91.1	92.5 92.1	93.5 93.1	94.5 94.1	95.5 95.1				
56	87.7	88.7	89.7	90.7	91.7	92.7	93.7	94.7	95.7	2.00		
58	87.3	88.3	89.4	90.4	91.4	92.4	93.4	94.4	95.4	221		
60	87.0	88.0	89.0	90.0	91.0	92.0	93.0	94.0	95.0	96.0		
62	86.7	87.7	88.6	89.6	90.6	91.6	92.6	93.6	94.6	95.6		
64	86.3	87.3	88.3	89.3	90.3	91.3	92.2	93. 2	94.2	95.2		
66	86.0	87.0	-88.0	89.0	89.9	90.9	91.8	92.8	93.8	94.8		
68	85.6	86.6	87.6	88.6	89.5	90.5	91.4	92.4	93.4	94.4		
70	85.3	86.3	87.3	88.3	89.2	90.1	91.0	92.0	93.0	94.0		
72	85.0	86.0	86.9	87.9	88.8	89.8	90.7	91.7	92.7	93.7		
74	84.6 84.3	85.6 85.3	86.5 86.2	87.5 87.2	88.4 88.1	89.4 89.1	90.3	91.3 91.0	92.3	93.3 93.0		
78	84.0	85.0	85.9	86.9	87.8	89.1	89.6	90.6	92.0 91.6	92.6		
80	83.6	84.6	85.5	86.5	87.4	88.4	89.3	90.2	91.2	92.2		
82	83. 2	84.2	85.1	86.1	87.0	88.0	88.9	89.8	90.8	91.8		
84	82.9	83.8	84.7	85.7	86.6	87.6	88.5	89.4	90.4	91.4		
86	82.6	83.5	84.4	85.4	86.3	87.3	88.2	89.1	90.0	91.0		
88	82.3	83.2	84.1	85.1	-86.0	87.0	87.9	88.8	89.7	90.7		
90	82.0	82.9	83.8	84.8	85.7	86. 6	87.5	88.4	89.3	90.3		
92	81.7	82.6	83.5	84.4	85.3	86.2	87.1	88.1	89.0	90.0		
94	81.3	82.2 81.9	83.1 82.8	84.1 83.7	85.0 84.6	85.9 85.6	86.8 86.5	87.7 87.4	88.6 88.3	89.6 89.3		
98	80.7	81.6	82.5	83.4	-84.3	85.2	86.1	87.0	88.0	89.0		
100	80.4	81.3	82.2	83.1	84.0	84.9	85.8	86.7	87.6	88.6		
102	80.1	81.0	81.9	82.8	83.7	84.6	85.5	86.4	87.3	88.3		
104	79.7	80.6	81.5	82.5	83.4	84.3	85.2	86.1	87.0	87.9		
106	79.4	80.3	81.2	82.1	83.0	83.9	84.8	85.7	86. 6 86. 3	87.6		
108	79.1	80.0	80.9	81.8	82.7	83.6	84.5	85.4	86.3	87.2		
110	78.8	79.7	80.6	81.5	82.4	83. 3	84.2	85.1	86.0	86.9		
112	78.5	79.4	80.3	81.2	82.1	83.0	83.8	84.7	85.6	86.6		
114	78.2	79.1	80.0 79.7	80.9	81.7	82.6	83.5 83.2	84.4	85.3 85.0	86. 2 85. 9		
118	77.5	78.8	79.7	80.6 80.2	81.4 81.1	82.3 82.0	83.2	84.1 83.7	84.6	85.6		
			1.1				1. 1. 1. 1					
120	77.2	78.1	79.0	79.9	80.8	81.7	82.5	83.4	84.3	85. 2		

Reduction of Specific Gravity Readings to 60°F

This table shows the specific gravities at $60^{\circ}/60^{\circ}$ F of oils having, at the designated temperatures, the observed specific gravities indicated. For example, if the observed specific gravity is 0.610 at 80° F, the true specific gravity at $60^{\circ}/60^{\circ}$ F will be 0.621. The headings "Observed specific gravity" and "Observed temperature" signify the true indication of the hydrometer and the true temperature of the oil; that is, the observed readings corrected, if necessary, for instrumental errors.)

Observed cemperature in °F	0.610	0.611	0.612	0.613	0.614	0.615	0.616	0.617	0.618	0.619
			Corres	sponding	g specifi	c gravi	ties at (80°/60° I	r	
62							[0.62
64									0,6200	.62
66								0.6200	.6210	.62
68						0.6200	.6205	.6215	.6225	.62
70					0.6200	.6210	.6215	.6225	.6235	.62
72				0.6200	.6210	.6220	.6225	.6235	.6245	.62
74		•••••	0.6200	.6210	.6220	.6230	.6235	.6245	.6255	.62
6		0.6200	.6210	.6220	.6230	.6240	.6245	.6255	.6265	.62
78	0.6200	.6210	.6220	.6230	.6240	.6250	.6255	.6265	.6275	.62
00	.621	600	.623	.624	.625	.626	.626	.627	.628	.62
80 82	.621	.622 .623	.623	.625	.626	.627	.628	.629	.630	.63
84	.623	.624	.625	.626	.627	.628	.629	.630	.631	.63
86	.623	.624	.626	.627	.628	.629	.630	.631	.632	.63
88	.625	.626	.627	.628	.629	.630	.631	.632	.633	.63
90	.626	.627	.628	.629	.630	.631	.632	.633	.634	.63
92	.627	.628	.629	.630	.631	.632	.633	.634	.635	.63
94	.628	.629	.630	.631	.632	.633	.634	.635	.636	.63
96	.629	.630	.631	.632	.633	.634	.635	.636	.637	.63
98	.630	.631	.632	.633	.634	.635	.636	.637	.638	.63
00	.631	.632	.633	.634	.635	.636	.637	.638	.639	.64
02	.632	.633	.634	.635	.636	.637	.638	.639	.640	.64
04	.633	.634	.635	.636	.637	.638	.639	.640	.641	.64
06	.634	.635	.636	.637	.638	.639	.640	.641	.642	.64
08	.635	.636	.637	.638	.639	.640	.641	.642	.643	.64
10	600	0.00	000	,639	010	041	619	.643	.644	.64
10	.636	.637	.638	.640	.640	.641	.642	.644	.645	.64
12	.637	.638 .639	.640	.641	.641	.643	.644	.645	.646	.64
14 16	.638	.639	.641	.642	.643	.643	.645	.646	.647	.64
18	.640	.641	.642	.643	.644	.645	.640	.647	.648	.64
10	.020	.041	.0122	.010	1094	.040	.010	.011	.010	.01
20	.641	.642	.643	.644	.645	.646	.647	.648	.649	.65

REDUCTION OF SPECIFIC GRAVITY READINGS TO 60°F-Con.

		Observed specific gravities												
Observed temperature in °F	0.620	0.621	0.622	0.623	0.624	0.625	0.626	0.627	0.628	0.629				
		Corresponding specific gravities at 60°/60° F												
44										0.6200				
43 48					•••••			0.6200	0.6200 .6210	.6210				
50 52					0.6200	0.6200	0.6205	.6215 .6230	.6225	.6235				
54				0.6200	.6210	.6220	.6230	.6240	.6250	.6260				
56			0.6200	.6210	.6220	.6230	.6240	.6250	.6260	.6270				
58		0.6200	.6210	.6220	.6230	.6240	.6250	.6260	.6270	.6280				
60	0.6200	.6210	.6220	.6230	.6240	.6250	.6260	.6270	.6280	.6290				
62	.6210	.6220	.6230	.6240	.6250	.6260	.6270	.6280	.6290	.6300				
64	.6220	.6230	.6240	.6250	.6260	.6270	.6280	.6290	.6300	.6310				
66	.6230	.6240	.6250	.6260	.6270	.6280	.6290	.6300	.6310	.6320				
68	.6245	.6255	.6265	.6275	.6285	.6295	.6305	.6315	.6325	.6335				
70	.6255	.6265	.6275	.6285	.6295	.6305	.6315	.6325	.6335	.6345				
72	.6265	.6275	.6285	.6295	.6305	.6315	.6325	.6335	.6345	.6355				
74	.6275	.6285	.6295	.6305	.6315	.6325	.6335	.6345	.6355	.6365				
76	.6285	.6295	.6305	.6315	.6325	.6335	.6345	.6355	.6365	.6375				
78	.6295	.6305	.6315	.6325	.6335	.6345	.6355	.6365	.6375	.6385				
80	.630	.631	.632	.633	.634	.635	.636	.637	.638	.639				
82	.632	.633	.634	.635	.636	.637	.637	.638	.639	.640				
84	.633	.634	.635	.636	.637	.638	.638	.639	.640	.641				
86	.634	.635	.636	.637	.638	.639	.639	.640	.641	.642				
83	.635	.636	.637	.638	.639	.640	.640	.641	.642	.643				
90	.636	.637	.638	.639	.640	.641	.641	.642	.643	.644				
02	.637	.638	.639	.640	.641	.642	.642	.643	.644	.645				
94	.638	.639	.640	.641	.642	.643	.643	.644	.645	.646				
96	.639	.640	.641	.642	.643	.644	.644	.645	.646	.647				
98	.640	.641	.642	.643	.644	.645	.645	.646	.647	.648				
00	.641	.642	.643	.644	.645	.646	.646	.647	.648	.649				
02	.642	.643	.644	.645	.648	.647	.647	.648	.649	.650				
04	.643	.644	.645	.646	.647	.648	.648	.649	.650	.651				
06	.644	.645	.646	.647	.648	.649	.649	.650	.651	.652				
08	.645	.646	.647	.648	.649	.650	.650	.651	.652	.653				
10	.646	.647	.648	.649	.650	.651	.651	.652	.653	.654				
12	.647	.648	.649	.650	.651	.652	.652	.653	.654	.655				
14	.648	.649	.650	.651	.652	.653	.653	.654	.655	.656				
16	.649	.650	.651	.652	.653	.654	.654	.655	.656	.657				
18	.650	.651	.652	.653	.654	.655	.655	.656	.657	.658				
20	.651	.652	.653	.654	.655	.656	.656	.657	.658	.659				

385

REDUCTION OF SPECIFIC GRAVITY READINGS TO 60°F-Con.

	Observed specific gravities												
Observed temperature in °F	0.630	0.631	0.632	0.633	0.634	0.635	0.636	0.637	0.638	0.639			
			Corres	ponding	g specifi	c gravit	ties at 6	10°/60° I					
30							0.620	0,621	0.622	0.623			
32						0.620	.621	.622	.623	.624			
34					0.620	.621	.622	.623	.624	.625			
36				0.620	.621	.622	.623	.624	.625	.626			
38			0.620	.621	.622	.623	.624	.625	.626	.627			
40		0.6200	.6210	.6220	.6230	.6240	.6255	.6265	.6275	.6285			
42	0.6200	.6210	.6220	.6230	.6240	.6250	.6265	.6275	.6285	.6295			
44	.6210	.6220	.6230	.6240	.6250	.6260	.6275	.6285	.6295	.6305			
46	.6220	.6230	.6240	.6250	.6?60	.6270	.6285	.6295	.6305	.6315			
48	.6230	.6240	.6250	.6260	.6270	.6280	.6295	.6305	.6315	.632			
50	.6245	.6255	.6265	.6275	.6285	.6295	.6305	.6315	.6325	.63%			
52	.6260	.6270	.6280	.6290	.6300	.6310	.6320	.6330	.6340	.6350			
54	.6270	.6280	.6290	.6300	.6310	.6320	.6330	.6340	.6350	.6360			
56	.6280	.6290	.6300	.6310	.6320	.6330	.6340	.6350	.6360	.6370			
58	.6290	.6300	.6310	.6320	.6330	.6340	.6350	.6360	.6370	.6380			
60	.6300	.6310	.6320	.6330	.6340	.6350	.6360	.6370	.6380	.6390			
62	.6310	.6320	.6330	.6340	.6350	.6360	.6370	.6380	.6390	.6400			
64	.6320	.6330	.6340	.6350	.6360	.6370	.6380	.6390	.6400	.6410			
66	.6330	.6340	.6350	.6360	.6370	.6380	.6390	.6400	.6410	.6420			
68	.6345	.6355	.6365	.6375	.6385	.6395	.6400	.6410	.6420	.6430			
70	.6355	.6365	.6375	.6385	.6395	.6405	.6410	.6420	.6430	.6440			
72	.6365	.6375	.6385	.6395	.6405	.6415	.6420	.6430	.6140	.6450			
74	.6375	.6385	.6395	.6405	.6415	.6425	.6430	.6440	.6450	.6460			
76	.6385	.6395	.6405	.6415	.6425	.6435	.6440	.6450	.6460	.6470			
78	.6395	.6405	.6415	.6425	.6435	.6445	.6450	.6460	.6470	.6480			
80	.640	.641	.642	.643	.644	.645	.646	.647	.648	.649			
82	.641	.642	.643	.644	.645	.646	.647	.648	.649	.650			
84	.642	.643	.644	.645	.646	.647	.648	.649	.650	.651			
86	.643	.644	.645	.646	.647	.648	.649	.650	.651	.652			
89	.614	.645	.646	.647	.648	.649	.650	.651	.652	.653			
90	.645	.646	.647	.648	.649	.650	.651	.652	.653	.654			
92	.646	.647	.648	.649	.650	.651	.652	.653	.654	.655			
94	.647	.648	.649	.650	.651	.652	.653	.654	.655	.656			
96 98	.648	.649 .650	.650 .651	.651	.652	.653 .654	.654 .655	.655 .656	.656 .657	.657 .658			
	850	.651	.652	652	654	CEF	050	.657	650	6FO			
100 102	.650	.652	.652	.653 .654	.654 .655	.655 .656	.656 .657	.057	.658 .659	.659			
		.652	.003	.054	.000	.657	.658		.660	.660			
104 106	.652	.653	.655	.656	.657	.658	.659	.659 .660	.660	.662			
108	.654	.655	.656	.657	.658	.659	.660	.661	.662	.663			
110	.655	.656	.657	.658	.659	.660	.661	.662	.663	.664			
112	.656	.657	.658	.659	.660	.661	.662	.663	.664	.665			
114	.657	.658	.659	.660	.661	.662	.663	.664	.665	.666			
116	.658	.659	.660	.661	.662	.663	.634	.665	.666	.667			
118	.659	.660	.661	.662	.663	.664	.665	.666	.667	.668			
120	.660	.661	.662	.663	.664	.665	.666	.667	.668	.669			

Sec. and				Observ	ed speci	ific gray	vities			
Observed		0.011	0.010	0.019	0.044	10.015	0.040	0.047	0.040	0.010
emperature in F	0.640	0.641	0.642	0.643	0.644	0.645	0.646	0.647	0.648	0.649
			Corresp	onding s	specific	gravitie	s at 60°	/60° F		
30	0.624	0.625	0.626	0.627	0.628	0.629	0.630	0.631	0.632	0.633
32	.625	.626	.627	.628	.629	.630	.631	.632	.633	.634
34	.626	.627	.628	.629	.630	.631	.632	.633	.634	.635
36	.627	.628	.629	.630	.631	.632	.633	.634	.635	.636
38	.628	.629	.630	.631	.632	.633	.634	.635	.633	.637
40	.6295	.6305	.6315	.6325	.6335	.6345	.6355	.6365	.6375	.638
42	.6305	.6315	.6325	.6335	.6345	.6355	.6365	.6375	.6385	.639
44	.6315	.6325	.6335	.6345	.6355	.6365	.6375	.6385	.6395	.640
46 48	.6325 .6335	.6335 .6345	.6345 .6355	.6355 .6365	.6365 .6375	.6375 .6385	.6385 .6395	.6395 .6405	.6405 .6415	.642
50	.6345	.6355	.6365	.6375	.6385	.6395	.6410	.6420	.6430	.644
52	.6360	.6370	.6380	.6390	.6400	.6410	.6420	.6430	.6440	.645
54	.6370	.6380	.6390	.6400	.6410	.6420	.6430	.6440	.6450	.646
56	.6380	.6390	.6400	.6410	.6420	.6430	.6440	.6450	.6460	.647
58	.6390	.6400	.6410	.6420	.6430	.6440	.6450	.6460	.6470	.649
60	.6400	.6410	.6420	.6430	.6440	.6450	.6430	.6470	.6480	.649
62	.6410	.6420	.6430	.6440	.6450	.6460	.6470	.6480	.6490	.650
€4	.6420	.6430	.6440	.6450	.6450	.6470	.6480	.6490	.6500	.651
66 68	.6430 .6440	.6440 .6450	.6450 .6460	.6460 .6470	.6470 .6490	.6480 .6490	.6490 .6500	.6500 .6510	.6510 .6520	.652 .653
70	.6450	.6460	.6170	.6480	.6490	.6500	.6510	.6520	.6530	.654
72	.6460	.6470	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.655
74	.6470	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.656
76	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.6560	.657
78	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.6560	.6570	.658
80	.650	.651	.652	.653	.654	.655	.656	.657	.658	.659
82	.651	.652	.653	.654	.655	.656	.657	.658	.659	.660
84	.652	.653	.654	.655	.656	.657	.658 .659	.659 .660	.660 .661	.661
83 88	.653 .654	.654 .655	.655 .656	.656 .657	.657 .658	.658 .659	.660	.661	.662	.663
90	.655	.656	.657	.658	.659	.660	.661	.662	.663	.664
92	.055	.657	.658	.659	.660	.661	.662	.663	.654	.665
91	.657	.658	.659	.660	.661	.662	.663	.664	.665	.666
96	.658	.659	.600	.661	.662	.663	.664	.665	.636	.667
98	.659	.660	.661	.662	.663	.664	.665	.666	.667	.668
100	.630	.661	.662	.663	.664	.665	.663	.667	.668	.669
102	.661	.662	.663	.664	.665	.666	667	.668	.639	.670
104	.662	.663	.664	.665	.666	.667	.668	.669	.670 .671	.671
103 108	.663 .664	.664 .665	.665 .666	.666 .667	.667	.608	.669 .670	.670 .671	.672	.673
			1	0.25	1 8					
10	.635	.666	.667	.668	.669	.670	.671	.672	.673	.674
112	.666	.667	.668	.669	.670	.671	.672	.673	.674 .675	.678
114	.667	.668	.639	.670	.671	.672	.673 .674	.674 .675	.676	.677
116 118	.668 .669	.669 .670	.670 .671	.671 .672	.672	.673 .674	.675	.673	.677	.678
							.676	.677	.678	.679
120	.670	.671	.672	.673	.674	.675	.010	.011	.010	.0/3

				Observ	ed spec	ific gra	VILLES			
Observed temperature in °F	0.650	0.651	0.652	0.653	0.654	0.655	0.656	0.657	0.658	0.659
			Corresp	onding	specific	gravitie	es at 60°	°/60° F		
00	0.004	0.005	0.000	0.00	0.000	0.000	0.010	0.011	0.010	0.010
30	0.634	0.635	0.636	0.637 .638	0.638	0.639	0.640	0.641	0.642	0.643
32	.635 .636	.636	.638	.639	.640	.641	.641 .642	.642	.643	.644
34								.643	.644	.645
33	.637	.638	.639	.640	.641	.642	.643	.644	.645	.646
38	.638	.639	.640	.641	.642	.643	.644	.645	.646	.647
40	.6395	.6405	.6415	.6425	.6435	.6445	.6455	.6435	.6475	.6485
42	.6405	.6415	.6425	.6435	.6445	.6455	.6465	.6475	.6485	.6495
44	.6415	.6425	.6435	.6445	.6455	.6465	.6475	.6485	.6495	.6505
46	.6425	.6435	.6445	.6455	.6435	.6475	.6485	.6495	.6505	.6515
48	.6435	.6445	.6455	.6465	.6475	.6485	.6495	.6505	.6515	.6525
50	.6450	.6430	.6470	.6480	.6490	.6500	.6510	.6520	.6530	.6540
52	.6460	.6470	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.6550
54	.6470	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.6560
56	.6480	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.6560	.6570
58	.6490	.6500	.6510	.6520	.6530	.6540	.6550	.6560	.6570	.6580
60	.6500	.6510	.6520	.6530	.6540	.6550	.6560	.6570	.6580	.6590
62	.6510	.6520	.6530	.6540	.6550	.6560	.6570	.6580	.6590	.6600
64	.6520	.6500	.6540	.6550	.6560	.6570	.6580	.6590	.6600	.6610
66	.6530	.6540	.6550	.6500	.6570	.6580	.6590	.6600	.6610	.6620
68	.6540	.6550	.6530	.6570	.6530	.6590	.6300	.6610	.6620	.6630
70	.6550	.6560	.6570	.6580	.6590	.6600	.6310	.6620	.6630	.6640
72	.6560	.6570	.6580	.6590	.6600	.6610	.6620	.6630	.6640	.6650
74	.6570	.6580	.6590	.6600	.6610	.6620	.6330	.6640	.6650	.6660
73	.6580	.6590	.6600	.6610	.6320	.6630	.6640	.6650	.6660	.6670
78	.6590	.6600	.6610	.6620	.6330	.6.40	.6650	.6660	.6670	.6680
90	000	001	000	000	001	0.05	000	0.017	000	000
80 82	.660 .661	.661	.662	.663	.664	.665	.663	.667	.668	.669
84:	.632	.662 .663	.663 .664	.664	.665 .663	.666 .667	.667 .668	.668 .669	.669	.670 .671
86	.663	.664	.665	.665 .666	.667	.668	.609	.670	.670 .671	.672
88	.664	.665	.666	.667	.668	.669	.670	.671	.672	.673
	cor	000	0.017	000	020	070	-	070	070	074
90	.665	.666	.667	.668	.639	.670	.671	.672	.673	.674
92	.663 .667	.667	.668	.669	.670	.671	.672	.673	.674	.675
94 93	.668	.668 .669	.669 .670	.670 .671	.671 .672	.672 .673	.673 .674	.674 .675	.675 .676	.676 .677
98	.669	.670	.671	.672	.673	.674	.675	.676	.677	.678
100	.670	.671	.672	.673	.674	.675	.676	.677	.678	.679
102	.671	.672	.673	.674	.675	.676	.677	.678	.679	.680
104	.672	.673	.674	.675	.676	.677	.678	.679	.680	.681
106	.673	.674	.675	.676	.677	.678	.679	.680	.681	.682
108	.674	.675	.676	.677	.678	.679	.679	.680	.681	.682
110	.675	.676	.677	.678	.679	.680	.680	.681	.682	.683
112	.676	.677	.678	.679	.680	.681	.681	.682	.683	.684
114	.677	.678	.679	.680	.681	.682	.682	.683	.684	.685
116	.678	.679	.680	.681	.68?	.683	.683	.684	.685	.686
118	.679	.680	.681	.682	.683	.684	.684	.685	.686	.687
120	.680	.681	.682	.683	.684	.685	.685	.686	.687	.688

				Observ	ed spec	ific grav	vities			
Observed temperature in "F	C.660	0.661	0.662	0.663	0.664	0.665	0.666	0.667	0.668	0.669
		- Jin	Corresp	onding	specific	gravitie	s at 60°	/60° F		
30	0.664	0.645	0.646	0.647	0.648	0.649	0.650	0.651	0.652	0.653
32	.645	.646	.647	.648	.649	.650	.651	.652	.653	.654
34	.645	.647	.648	.649	.650	.651	.652	.653	.654	.655
36	.647	.648	.649	.650	.651	.652	.653	.654	.655	.655
38	.648	.649	.650	.651	.652	.653	.655	.656	.657	.658
40	.6495	.6505	.6515	.6525	.6535	.6545	.6560	.6570	.6580	.659
42	.6505	.6515	.6525	.6535	.6545	.€555	.6570	.6580	.6590	.660
44	.6515	.6525	.6535	.6545	.6755	.6565	.6580	.6590	.6600	.661
46	.6525	.6535	.6545	.6555	.6565	.6575	.6590	.6600	.6610	.662
48	.6535	.6545	.6555	.6565	.6575	.6585	.6600	.6610	.6620	.6630
50	.6550	.6560	.6570	.6580	.6590	.6600	.6610	.6620	.6630	.664
52	.6530	.6570	.6580	.6590	.6600	.6610	.6620	.6630	.6640	.665
64	.6570	.6580	.6590	.6600	.6610	.66.0	.6630	.6640	.6650	.666
56	.6580	.6590	.6600	.6610	.6620	.6630	.6640	.6650	.6660	.667
58	.6590	.6000	.6610	.6620	.6630	.6640	,6650	.6660	.6670	.668
60	.6600	.6610	.6620	.6630	.6640	.6650	.6660	.6670	.6680	.639
62	.6610	.6620	.6630	.6640	.6650	.6660	.6670	.6380	.6690	.670
64	.6620	.6630	.6640	.6650	.6660	.6670	.6680	.6690	.6700	.671
66	6630	.6640	.6650	.6360	.6370	.6680	.6690	.6700	.6710	.672
68	.6640	.6650	.6660	.6670	.6080	.6690	.6700	.6710	.6720	.673
70	.6650	.6660	.6670	.6680	.6690	.6700	.6710	.6720	.6730	.674
72	.6660	.6670	.6680	.6690	.6700	.6710	.6720	.6730	.6740	.675
74	.6670	.6680	.6690	.6700	.6710	.6720	.6730	.6740	.6750	.676
76	.6680 .6690	.6690 .6700	.6700 .6710	.6710 .6720	.6720 .6730	.6730 .6740	.6740 .6750	.6750 .6760	.6760 .6770	.677
					2010					
80	.670	.671	.672	.673	.674	.675	.676	.677	.678	.679
82	.671	.672	.673	.674	.675	.676	.677	.678	.679	.680
84	.672	.673	.674	.675	.676 .677	.677 .678	.678 .679	.679 .680	.680 .681	.681
86	.673 .674	.674 .675	.675 .676	.676 .677	.678	.679	.679	.680	.681	.682
90	.675	.676	.677	.678	.679	.680	.680	.681	.682	.683
92	.676	.677	.678	.679	.680	.681	.681	.682	.683	.684
94	.677	.678	.679	.680	.681	.682	.682	.683	.684	.685
96	.678	.679	.680	.681	.682	.683	.683	.684	.685	.686
98	.679	.680	.681	.682	.683	.684	.684	.685	.686	.687
60	.680	.681	.682	.683	.684	.685	.685	.686	.687	.688
02	.681	.682	.683	.684	.685	.686	.686	.687	.688	.689
04	.682	.683	.684	.685	.686	.687	.687	.688	.689	.690
03	.683	.684	.685	.686	.687	.688	.688	.689	.690	.691
08	.683	.684	.685	.686	.687	.688	.689	.690	.691	.692
10	.684	.685	.686	.687	.688	.689	.690	.691	.692	.693
12	.685	.686	.687	.688	.689	.690	.691	.692	.693	.694
14	.683	.687	.688	.689	.690	.691	.692	.693	.694	.695
16	.687	.688	.689	.690	.691	.692	.693	.694	.695	.696
18	.688	.689	.690	.691	.692	.693	.694	.695	.696	.697
20	.689	.690	.691	.692	.693	.694	.695	.696	.697	.698

301-				Observ	ed speci	fic grav	rities			
Observed temperature in °F	0.670	0.671	0.672	0.673	0.674	0.675	0.676	0.677	0.678	0.679
			Corresp	onding	specific	gravitie	s at 60°	/60° F		
90	0.051	0.655	0.656	0.657	0.658	0.659	0.661	0.662	0.663	0.664
30	0.654				.659	.600	.632	.663	.664	.665
32	.655	.656	.657	.658						
34	.656	.657	.658	.659	.660	.661	.663	.664	.665	.666
33	.657	.658	.659	.660	.661	.662	.664	.665	.666	.667
38	.659	.660	.661	.662	.663	.664	.605	.666	.667	.668
40	.6600	.6610	.6620	.6630	.6610	.6650	.6660	.6670	.6630	.6690
42	.6610	.6620	.6630	.6640	.6650	.6060	.6670	.6680	.6690	.6700
44	.6620	.6630	.6640	.6650	.6660	.6670	.6680	.6690	.6700	.6710
46	.6630	.6640	.6650	.6660	.6670	.6680	.6690	.6700	.6710	.6720
48	.6640	.6650	.6660	.6670	.6630	.6390	.6700	.6710	.6720	.6730
50	.6650	.6060	.6670	.6680	.6690	.6700	.6710	.6720	.6730	.6740
52	.6660	.6670	.6680	.6690	.6700	.6710	.6720	.6730	.6740	.6750
54	.6670	.6680	.6390	.6700	.6710	.67.0	.6730	.6740	.6750	.6760
56	.6680	.6690	.6700	.6710	.6720	.6730	.6740	.6750	.6760	.6770
58	.6690	.6700	.6710	.6720	.6730	.6740	.6750	.6760	.6770	.6780
60	.6700	.6710	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790
62	.6710	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790	.6800
64	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790	.6800	.6810
63	.6730	.6740	.6750	.6760	.6770	.6780	.6790	.6800	.6810	.6820
68	.6740	.6750	.6760	.6770	.6780	.6790	.6800	.6810	.6820	.6830
70	.6750	.6760	.6770	.6780	.6790	.6800	.6810	.6820	.6830	.6840
72	.6760	.6770	.6780	.6790	.6800	.6810	.6820	.6830	.6840	.6850
74	.6770	.6780	.6790	.6800	.6810	.6820	.6830	.6840	.6850	.680
76	.6780	.6790	.6800	.6810	.6820	.6830	.6835	.6845	.6855	.686
78	.6790	.6800	.6810	.6820	.6830	.6840	.6845	.6855	.6865	.687
80	.680	.681	.682	.683	.684	.685	.685	.686	.687	.688
82	.681	.682	.683	.684	.685	.686 .	.686	.687	.688	.689
84	.682	.683	.684	.685	.686	.687	.687	.688	.689	.690
86	.683	.684	.685	.686	.687	.688	.688	.689	.690	.691
88	.683	.684	.685	.686	.687	.688	.689	.690	.691	.692
90	.684	.685	.683	.687	.688	.689	.690	.691	.692	.693
92	.685	.686	.687	.688	.689	.690	.691	.692	.693	.694
94	.686	.687	.688	.689	.690	.691	.692	.693	.694	.695
93	.687	.688	.689	.690	.691	.692	.693	.694	.695	.696
98	.688	.689	.690	.691	.692	.693	.694	.695	.696	.697
100	.689	.690	.691	.692	.693	.694	.695	.696	.697	.698
102	.690	.691	.692	.693	.694	.695	.693	.697	.698	.699
104	.691	.692	.693	.694	.695	.696	.697	.698	.699	.700
103	.692	.693	.694	.695	.696	.697	.698	.699	.700	.701
108	.693	.694	.695	.696	.697	.698	.699	.700	.701	.702
110	.694	.695	.696	.697	.698	.699	.700	.701	.702	.703
112	.695	.696	.697	.698	.699	.700	.701	.702	.703	.704
114	.696	.697	.689	.699	.700	.701	.702	.703	.704	.705
116	.697	.698	.699	.700	.701	.702	.702	.703	.704	.705
118	.698	.699	.700	.701	.702	.703	.703	.704	.705	.706
		.700	.701		1.150	.704	.704	.705	.706	.707
120	.699	.100	.701	.702	.703	.104	.104	.100	.100	.101

5-1-0				Observ	ed spec	ific grav	vities			
Observed temperature in °F	0.680	0.681	0.682	0.683	0.684	0.685	0.686	0.687	0.688	0.689
			Correspo	onding	specific	gravitie	s at 60°	/60° F		
30	0.635	0.666	0.667	0.668	0.669	0.670	0.671	0.672	0.673	0.674
32	.666	.667	.668	.669	.670	.671	.672	.673	.674	.675
34	.667	.668	.669	.670	.671	.672	.673	.674	.675	.676
36 38	.668 .669	.609 .670	.670 .671	.671 .672	.672 .673	.673 .674	.674 .675	.675 .676	.676 .677	.677
40	.6700	.6710	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790
42	.6710	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790	.6800
44	.6720	.6730	.6740	.6750	.6760	.6770	.6780	.6790	.6800	.6810
4 3 48	.6730 .6740	.6740 .6750	.6750 .6760	.6760 .6770	.6770 .6780	.6780 .6790	.6790 .6800	.6800 .6810	.6810 .6820	.6820 .6830
50	.6750	.6760	.6770	.6780	.6790	.6800	.6810	.6820	.6830	.6840
52	.6760	.6770	.6780	.6790	.6800	.6810	.6820	.6830	.6840	.6850
54	.6770	.6780	.6790	.6800 .6810	.6810	.6820	.6830 .6840	.6840	.6850 .6860	.6860
56 58	.6780 .6790	.6790 .6800	.6800 .6810	.6820	.6830	.6840	.6850	.6860	.6870	.6880
60	.6800	.6810	.6820	.6830	.6840	.6850	.6860	.6870	.6880	.6890
62	.6810	.6820	.6830	.6840	.6850	.6860	.6870 .6880	.6880 .6890	.6890	.6900
64 66	.6820 .6830	.6830 .6840	.6840 .6850	.6850 .6860	.6860 .6870	.6880	.6890	.6900	.6910	.6920
68	.6840	.6850	.6860	.6870	.6880	.6890	.6900	.6910	.6920	.6930
70	.6850	.6860	.6870	.6880	.6890	.6900	.6910	.6920	.6930	.6940
72	.6860	.6870	.6880	.6890	.6900	.6910 .6920	.6920 .6925	.6930	.6940 .6945	.6950 .6955
74	.6870 .6875	.6880 .6885	.6890 .6895	.6900	.6915	.6925	.6935	.6945	.6955	.6965
78	.6885	.6895	.6905	.6915	.6925	.6935	.6945	.6955	.6965	.6975
80	.689	.690	.691	.692	.693	.694	.695	.696	.697	.698
82 84	.690 .691	.691 .692	.692 ' .693	.693 .694	.694 .695	.695 .696	.696 .697	.697 .698	.698	.699 .700
86	.692	.693	.694	.695	.696	.697	.698	.699	.700	.701
88	.693	.694	.695	.696	.697	.698	.699	.700	.701	.702
90	.694	.695	.696	.697	.698	.699	.700	.701	.702	.703
92 94	.695 .696	.696 .697	.697 .698	.698 .699	.699	.700 .701	.701 .702	.702	.703	.704
96	.697	.698	.699	.700	.701	.702	.703	.704	.705	.706
98	.698	.699	.700	.701	.702	.703	.704	.705	.706	.707
100	.699	.700	.701	.702	.703	.704	.705	.706	.707	.708
102	.700	.701	.702	.703	.704	.705	.706	.707	.708	.709
104	.701 .702	.702 .703	.703	.704 .705	.705	.706	.707 .708	.708	.710	.711
108	.703	.704	.705	.706	.707	.708	.708	.709	.710	.711
110	.704	.705	.706	.707	.708	.709	.709	.710	.711	.712
112	.705	.706	.707	.708	.709	.710	.710	.711	.712	.713
116	.706	.707	.708	.709	.710	.711	.711	.712	.713	.714
116 118	.706 .707	.707 .708	.708 .709	.709	.710 .711	.711 .712	.712	.713	.714	.715
	.708	.709	.710	.711	.712	.713	.714	.715	.716	.717

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.690	0.691	0.692	0.693	0.694	0.695	0.696	0.697	0.698	0.699
		(Correspo	onding s	pecific	gravitie	s at 60°	/60° F		
30	0.675	0.676	0.677	0.678	0.679	0.680	0.681	0.682	0.683	0.684
32	.676	.677	.678	.679	.680	.681	.682	.683	.684	.685
34	.677	.678	.679	.680	.681	.682	.683	.684	.685	.686
36	.678 .679	.679 .680	.680 .681	.681 .682	.682 .683	.683 .684	.684 .685	.685 .686	.686 .687	.687 .688
40	.6800	.6810	,6820	.6830	.6840	,6850	.6865	.6875	.6885	.6895
42	.6810	.6820	.6830	.6840	.6850	.6860	.6875	.6885	.6895	.6905
44	.6820	.6830	.6840	.6850	.6860	.6870	.6885	. 6895	.6905	.6915
46	.6830 .6840	.6840 .6850	.6850 .6860	.6860 .6870	.6870 .6880	.6880 .6890	.6895 .6900	.6905 .6910	.6915 .6920	.6925 .6930
50	.6850	.6860	.6870	.6880	.6890	.6900	.6910	.6920	.6930	.6040
52	.6860	.6870	.6880	.6890	.6900	.6910	.6920	.6930	.6940	.6950
54	.6870	.6880	.6890	.6900	.6910	.6920	.6930	.6940	.6950	.6960
56 58	.6880 .6890	.6890 .6900	.6900 .6910	.6910 .6920	.6920 .6930	.6930 .6940	.6940 .6950	.6950 .6960	.6960 .6970	.6970
60	.6900	.6910	.6920	.6930	.6940	.6950	.6960	.6970	.6980	.6990
62	.6910	.6920	.6930	.6940	.6950	.6960	.6970	.6980	.6990	.7000
64	.6920 .6930	.6930 .6940	.6940 .6950	.6950 .6960	.6960 .6970	.6970 .6980	.6980 .6990	.6990 .7000	.7000	.7010
66 68	.6930	.6940	.6960	.6970	.6980	.6990	.7000	.7010	.7020	.7030
70	.6950	.6960	.6970	.6980	.6990	.7000	.7010	.7020	.7030	.7040
72	.6960	.6970	.6980	.6990	.7000	.7010	.7015	.7025	.7035	.704
74	.6965 .6975	.6975	.6985 .6995	.6995 .7005	.7005	.7015	.7025	.7035	.7045	.7055
78	.6985	.6995	.7005	.7015	.7025	.7035	.7045	.7055	.7065	.7075
80	.699	.700	.701	.702	.703	.704	.705	.706	.707	.708
82	.700	.701	.702	.703	.704	.705	.706	.707	.708	.709
84 86	.701	.702 .703	.703	.704 .705	.705	.706	.707	.708 .709	.709	.710
88	.703	.704	.705	.706	.707	.708	.709	.710	.711	.712
90	.704	.705	.706	.707	.708	.709	.710	.711	.712	.713
92	.705	.706	.707	.708 .	.709	.710	.711	.712 .713	.713	.714
94 96	.707	.708	.708	.709	.711	.711	.712	.713	.714	.715
98	.708	.709	.710	.711	.712	.713	.713	.714	.715	.716
100	.709	.710	.711	.712	.713	.714	.714	.715	.716	.717
102	.710	.711	.712	.713	.714	.715	.715	.716	.717	.718
104 106	.711 . .712	.712 .713	.713 .714	.714 .715	.715	.716 .717	.716 .717	.717 .718	.718 .719	.719 .720
108	.712	.713	.714	.715	.716	.717	.718	.719	.720	.721
110	.713	.714	.715	.716	.717	.718	.719	.720	.721	.722
112	.714	.715	.716	.717	.718	.719	.720	.721	.722	.723
114 116	.715 .716	.716 .717	.717 .718	.718 .719	.719	.720 .721	.721	.722 .723	.723 .724	.724
118	.710	.718	.719	.720	.721	.722	.722	.723	.724	.725
120	.718	.719	.720	.721	.722	.723	.723	.724	.725	.726

				Observ	ed speci	fic grav	vities			
Observed temperature in °F	0.700	0.701	0.702	0.703	0.704	0.705	0.706	0.707	0.708	0.709
			Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30 32	0.685	0.686	0.687 .688	0.688	0.689	0.690	0.691	0.692	0.693	0.694
34	.687	.688	.000	.690	.691	.692	.693	.694	.694	.695
33 38	.683 .689	.689 .690	.690 .691	.691 .692	.692 .693	.693 .694	.694 .695	.695 .696	.696 .697	.697
										.698
40 42	.6905 .6915	.6915 .6925	.6925 .6935	.6935 .6945	.6945 .6955	.6955 .6965	.6965 .6975	.6975 .6985	.6985 .6995	.6995
44	.6925	.6935	.6945	.6955	.6965	.6975	.6985	.6995	.7005	.7015
46	.6935	.6945	.6955	.6965	.6975	.6985	.6995	.7005	.7015	.7025
48	.6940	.6950	.6960	.6970	.6980	.6990	.7005	.7015	.7025	.7035
50	.6950	.6960	.6970	.6980	.6990	.7000	.7015	.7025	.7035	.7045
5?	.6930	.6970	.6980	.6990	.7000	.7010	.7025	.7035	.7045	.7056
54	.6970	.6980	.6990	.7000	.7010	.7020	.7030	.7040	.7050	.7060
56 58	.6980 .6990	.6990 .7000	.7000	.7010	.7020	.7030 .7040	.7040	.7050	.7060	.7070
					1			.1000	.1010	
60	.7000	.7010	.7020	.7030	.7040	.7050	.7060	.7070	.7080	.7090
62	.7010	.7020	.7030	.7040	.7050 .7030	.7060 .7070	.7070	.7080	.7090	.7100
64 63	.7020	.7030	.7040	.7050	.7070	.7080	.7080	.7090	.7110	.7110
68	.7040	.7050	.7060	.7070	.7080	.7090	.7095	.7105	.7115	.7125
70	.7050	.7060	.7070	.7080	.7090	.7100	.7105	.7115	.7125	.7135
72	.7055	.7065	.7075	.7085	.7095	.7105	.7115	.7125	.7135	.714
74	.7065	.7075	.7085	.7095	.7105	.7115	.7125	.7135	.7145	.7155
76	.7075 .7085	.7085 .7095	.7095	.7105	.7115 .7125	.71°5 .7135	.7135 .7145	.7145 .7155	.7155 .7165	.7165
80	.709	.710	.711	.712	.713	.714	.715	.716	.717	.718
82	.710	.711	.712	.713	.714	.715	.716	.717	.718	.719
84	.711	.712	.713	.714	.715	.716	.717	.718	.719	.720
86 88	.712 .713	.713	.714	.715	.716	.717	.718	.719 .720	.720 .721	.721
	.110	.114	.715	.716	.111	.718	.719	.120	.141	.122
90	.714	.715	.716	.717	.718	.719	.720	.721	.722	.723
92 94	.715 .716	.716	.717	.718	.719 .720	.720 .721	.720 .721	.721 .722	.722 .723	.723 .724
93	.716	.717	.718 .718	.719 .719	.720	.721	.721	.723	.723	.725
98	.717	.718	.719	.720	.721	.722	.723	.724	.725	.726
00	.718	.719	.720	.721	.722	.723	.724	.725	.726	.727
02	.719	.720	.721	.722	.723	.724	.725	.726	.727	.728
04	.720	.721	.722	.723	.724	.725	.723	.727	.728	.729
06	.721 .722	.722 .723	.723 .724	.724 .725	.725 .726	.726 .727	.727 .728	.728 .729	.729 .730	.730 .731
10	.723	.724	.725	.726	.727	.728	.729	.730	.731	.732
12	.724	.724	.725	.725	.728	.728	.729	.730	.731	.733
14	.725	.726	.727	.728	.729	.730	.731	.732	.733	.734
16	.726	.727	.728	.729	.730	.731	.731	.732	.733	.734
18	.726	.727	.728	.729	.730	.731	.732	.733	.734	.735
20	.727	.728	.729	.730	.731	.732	.733	.734	.735	.736

				Observ	ea speci	fic grav	Tues			
Observed emperature in °F	0.710	0.711	0.712	0.713	0.714	0.715	0.716	0.717	0.718	0.719
			Correspo	onding s	specific	gravitie	s at 60°	/60° F		
			(-	
30	0.695	0.696	0.967	0.698	0.699	0.700	0.701	0.702	0.703	0.704
2	.696	.697	.698	.699	.700	.701	.702	.703	.704	.705
4	.697	.698	.699	.700	.701	.702	.703	.704	.705	.706
6 8	.698 .699	.699 .700	.700 .701	.701 .702	.702	.703 .704	.704 .705	.705	.706 .707	.707
0	.7005	.7015	.7025	.7035	.7045	.7055	.7065	.7075	.7085	
2	.7015	.7025	.7035	.7045	.7055	.7065	.7075	.7085	.7085	.709
4	.7025	.7035	.7045	.7055	.7065	.7075	.7085	.7095	.7105	.711
6	.7035	.7045	.7055	.7065	.7075	.7085	.7095	.7105	.7115	.712
8	.7045	.7055	.7065	.7075	.7085	.7095	.7105	.7115	.7125	.713
0	.7055	.7065	.7075	.7085	.7095	.7105	.7115	.7125	.7135	.714
2	.7065	.7075	.7085	.7095	.7105	.7115	.7125	.7135	.7145	.715
4	.7070	.7080	.7090	.7100	.7100	.7120	.7130	.7140	.7150	.716
6 8	.7080 .7090	.7090	.7100	.7110 .7120	.7120	.7130 .7140	.7140 .7150	.7150 .7160	.7160 .7170	.717
	.7100	.7110	.7120	.7130						
0	.7110	.7120	.7120	.7140	.7140	.7150	.7160 .7170	.7170 .7180	.7180 .7190	.719
4	.7120	.7130	.7140	.7150	.7160	.7170	.7180	.7190	.7200	.721
6	.7130	.7140	.7150	.7160	.7170	.7180	.7185	.7195	.7205	.721
8	.7135	.7145	.7155	.7165	.7175	.7185	.7195	.7205	.7215	.722
0	.7145	.7155	.7165	.7175	.7185	.7195	.7205	.7215	.7225	.723
2	.7155	.7165	.7175	.7185	.7195	.7205	.7215	.7225	.7235	.724
4	.7165	.7175	.7185	.7195	.7205	.7215	.7225	.7235	.7245	.725
6 8	.7175 .7185	.7185 .7195	.7195 .7205	.7205 .7215	.7215	.7225 .7235	.7235 .7245	.7245 .7255	.7255 .7265	.726
0	.719	.720	.721	.722	.723	.724	.725	.726	.727	.728
3?	.720	.721	.722	.723	.724	.725	.726	.727	.728	.729
4	.721	.722	.723	.724	.725	.726	.727	.728	.729	.730
6 8	.722 .723	.723 .724	.724 .725	.725 .726	.726 .727	.727 .728	.728	.729	.730	.731
							.729	.730	.731	.732
0	.724 .724	.725 .725	.726 .726	.727 .727	.728	.729	.729	.730	.731	.732
4	.725	.726	.727	.728	.728 .729	.729	.730	.731 .732	.732 .733	.733
6	.726	.727	.728	.729	.730	.731	.732	.733	.734	.735
8	.727	.728	.729	.730	.731	.732	.733	.734	.735	.736
0	.728	.729	.730	.731	.732	.733	.734	.735	.736	.737
2	.729	.730	.731	.732	.733	.734	.735	.736	.737	.738
4	.730	.731	.732	.733	.734	.735	.736	.737	.738	.739
6 8	.731 .732	.732 .733	.733 .734	.734 .735	.735 .736	.736 .737	.737 .737	.738 .738	.739 .739	.740
0	.733	.734	.735	.736	.737	.738	.738	.739	.740	.741
2	.734	.735	.736	.737	.738	.739	.739	.740	.740	.741
4	.734	.735	.736	.737	.738	.739	.740	.741	.742	.743
6	.735	.736	.737	.738	.739	.740	.741	.742	.743	.744
18	.736	.737	.738	.739	.740	.741	.742	.743	.744	.745
0	.737	.738	.739	.740	.741	.742	.742	.743	.744	.745

36044°--16--3

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.720	0.721	0.722	0.723	0.724	0.725	0.726	0.727	0.728	0.729
		1.	Corresp	onding	specific	gravitie	es at 60°	/60° F		
30 32 54 36 38	0.705 .703 .707 .708 .709	0.706 .707 .708 .709 .710	0.707 .708 .709 .710 .711	0.708 .709 .710 .711 .712	0.709 .710 .711 .712 .713	0.710 .711 .712 .713 .714	0.712 .713 .714 .715 .716	0.713 .714 .715 .716 .717	0.714 .715 .716 .717 .718	0.715 .716 .717 .718 .719
40 42 44 43 48	.7105 .7115 .7125 .7125 .7135 .7145	.7115 .7125 .7135 .7145 .7155	.7125 .7135 .7145 .7155 .7155 .7165	.7135 .7145 .7155 .7165 .7165 .7175	.7145 .7155 .7165 .7165 .7175 .7185	.7155 .7165 .7175 .7185 .7185 .7195	.7165 .7175 .7185 .7195 .7205	.7175 .7185 .7195 .7205 .7215	.7185 .7195 .7205 .7215 .7225	.7195 .7205 .7215 .7225 .7225
50 52 54 56 58	.7155 .7165 .7170 .7180 .7190	.7165 .7175 .7180 .7190 .7200	.7175 .7185 .7190 .7200 .7210	.7185 .7195 .7200 .7210 .7220	.7195 .7205 .7210 .7220 .7230	.7305 .7215 .7220 .7230 .7230 .7240	.7215 .7225 .7230 .7240 .7250	.7225 .7235 .7240 .7250 .7260	.7235 .7245 .7250 .7260 .7270	.7245 .7255 .7260 .7270 .7280
60 62 64 66 68	.7200 .7210 .7220 .7225 .7235	.7210 .7220 .7230 .7235 .7245	.7220 .7230 .7240 .7245 .7255	.7230 .7240 .7250 .7255 .7265	.7240 .7250 .7260 .7265 .7275	.7250 .7260 .7270 .7275 .7285	.7200 .7270 .7280 .7285 .7295	.7270 .7280 .7290 .7295 .7305	.7280 .7290 .7300 .7305 .7315	.7290 .7300 .7310 .7315 .7325
70 72 74 76 78	.7245 .7255 .7265 .7265 .7275 .7285	.7255 .7265 .7275 .7285 .7295	.7265 .7275 .7285 .7295 .7305	.7275 .7285 .7295 .7305 .7315	.7285 .7295 .7305 .7315 .7325	.7295 .7305 .7315 .7325 .7335	.7305 .7315 .7325 .7330 .7340	.7315 .7325 .7335 .7340 .7350	.7325 .7335 .7345 .7350 .7360	.7338 .7348 .7358 .7360 .7370
80 82 84 83 88	.729 .730 .731 .732 .733	.730 .731 .732 .733 .733	.731 .732 .733 .734 .735	.732 .733 .734 .735 .735	.733 .734 .735 .736 .736	.734 .735 .736 .737 .738	.735 .736 .737 .737 .737 .738	.736 .737 .738 .738 .738	.737 .738 .739 .739 .739 .740	.738 .739 .740 .740 .741
90 92 94 96 98	.733 .734 .735 .736 .737	.734 .735 .736 .737 .738	.735 .736 .737 .738 .739	.736 .737 .738 .739 .740	.737 .738 .739 .740 .741	.738 .739 .740 .741 .742	.739 .740 .741 .742 .743	.740 .741 .742 .743 .744	.741 .742 .743 .744 .744 .745	.742 .743 .744 .745 .746
100 102 104 106 108	.738 .739 .740 .741 .741	.739 .740 .741 .742 .742	.740 .741 .742 .743 .743	.741 .742 .743 .744 .744	.742 .743 .744 .745 .745	.743 .744 .745 .746 .746	.743 .744 .745 .746 .747	.744 .745 .746 .747 .748	.745 .746 .747 .748 .749	.746 .747 .748 .749 .750
110 112 114 116 118	.742 .743 .744 .745 .745	.743 .744 .745 .746 .747	.744 .745 .746 .747 .748	.745 .746 .747 .748 .748 .749	.746 .747 .748 .749 .750	.747 .748 .749 .750 .751	.748 .749 .749 .750 .751	.749 .750 .750 .751 .752	.750 .751 .751 .752 .753	.751 .752 .752 .753 .753 .754
120	.746	.747	.748	.749	.750	.751	.752	.753	.754	.755

S				Observ	ed spec	ific grav	vities			
Observed temperature in °F	0.730	0.731	0.732	0.733	0.734	0.735	0.736	0.737	0.738	0.739
	•		Corresp	onding	specific	gravitie	s at 60°) /60° F		
30	0.716	0,717	0.718	0.719	0,720	0.721	0.722	0,723	0.724	0.725
32	.717	.718	.719	.720	.721	.722	.723	.724	.725	.726
34	.718	.719	.720	.721	.7.2	.723	.724	.725	.726	.727
36 38	.719 .720	.720 .721	.721 .722	.722 .723	.723 .724	.724 .725	.725 .726	.726 .727	.727 .728	.728 .729
40	.7205	.7215	.7225	.7235	.7245	.7255	.7270	.7280	.7290	.730
42	.7215	.7225	.7235	.7245	.7255	.7265	.7275	.7285	.7295	.730
44 4 3	.7225 .7235	.7235 .7245	.7245 .7255	.7255	.7265 .7275	.7275	.7285	.7295	7305	.731
48	.7245	.7245	.7205	.7275	.7285	.7285 .7295	.7255 .7305	.7305 .7315	.7315 .7325	.732 .733
50	.7255	.7265	.7275	.7285	.7295	.7305	.7315	.7325	.7335	.734
52	.7265	.7275	.7285	.7295	.7305	.7315	.7325	.7335	.7345	.735
54	.7270	.7280 .7290	.7290	.7300 .7310	.7310 .7320	.7320 .7330	.7330 .7340	.7340 .7350	.7350 .7360	.736
58	.7290	.7300	.7310	.7320	.7330	.7340	.7350	.7360	.7370	.737 .738
60	.7300	.7310	.7320	.7330	.7340	.7350	.7360	.7370	.7380	.739
62	.7310	.7320	.7330	.7340	.7350	.7360	.7370	.7380	.7390	.740
66	.7320 .7325	.7330 .7335	.7340 .7345	.7350 .7355	.7360 .7365	.7370 .7375	.7375 .7385	.7385 .7395	.7395 .7405	.740
68	.7335	.7345	.7355	.7365	.7375	.7385	.7395	.7405	.7405	.742
70	.7345	.7355	.7365	.7375	.7385	.7395	.7405	.7415	.7425	.743
72	.7355	.7365	.7375	.7385	.7395	.7405	.7410	.7420	.7430	.744
74	.7365	.7375	.7385	.7395	.7405	.7415	.7420	.7430	.7440	.745
76	.7370 .7380	.7380 .7390	.7390 .7400	.7400 .7410	.7410 .7420	.7420 .7430	.7430 .7440	.7440 .7450	.7450 .7460	.746
80	.739	.740	.741	.742	.743	.744	.744	.745	.746	.747
82	.740	.741	.742	.743	.744	.745	.745	.746	.747	.748
84 86	.741	.742 .742	.743 .743	.744	.745 .745	.746 .746	.746 .747	.747	.748	.749
88	.742	.743	.744	.745	.746	.747	.748	.748 .749	.749 .750	.750 .751
90	7.43	.744	.745	.746	.747	.748	.749	.750	.751	.752
92	.744 .745	.745 .746	.746 .747	.747 .748	.748 .749	.749 .750	.750	.751	.752	.753
93	.746	.747	.748	.749	.750	.751	.751 .751	.752 .752	.753 .753	.754
98	.747	.748	.749	.750	.751	.752	.752	.753	.754	.755
00	.747	.748	.749	.750	.751	.752	.753	.754	.755	.756
04	.748 .749	.749 .750	.750 .751	.751 .752	.752 .753	.753 .754	.754 .755	.755 .756	.756 .757	.757 .758
06	.750	.751	.752	.753	.754	.755	.756	.757	.758	.759
08	.751	.752	.753	.754	.755	.756	.756	.757	.758	.759
10	.752	.753	.754	.755	.756	.757	.757	.758	.759	.760
12	.753 .753	.754 .754	.755	.756	.757	.758	.758	.759	.760	.761
16	.754	.755	.755 .756	.756	.757 .758	.758 .759	.759 .760	.760 .761	.761 .762	.762 .763
18	.755	.756	.757	.758	.759	.760	.761	.762	.763	.764
20	.756	.757	.758	.759	.760	.761	.761	.762	.763	.764

KANSAS CITY TESTING LABORATORY

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.740	0.741	0.742	0.743	0.744	0.745	0.746	0.747	0.748	0.749
			Corresp	onding	specific	gravitie	ns at 60°	'/60° F		1
30 32 34 36 38	0.726 .727 .728 .729 .730	0.727 .728 .729 .730 .731	0.728 .729 .730 .731 .732	0.729 .730 .731 .732 .733	0.730 .731 .732 .733 .733 .734	0.731 .732 .733 .734 .735	0.732 .733 .734 .735 .736	0.733 .734 .735 .736 .737	0.734 .735 .736 .737 .738	0.735 .730 .737 .738 .739
40 42 44 46 48	.7310 .7315 .7325 .7335 .7345	.7320 .7325 .7335 .7345 .7355	.7330 .7335 .7345 .7355 .7365	.7340 .7345 .7355 .7365 .7365 .7375	.7350 .7355 .7305 .7375 .7375 .7385	.7360 .7365 .7375 .7385 .7395	.7370 .7380 .7390 .7400 .7405	.7380 .7390 .7400 .7410 .7415	.7390 .7400 .7410 .7420 .7425	.7400 .7410 .7420 .7430 .7435
50 52 54 56 58	.7355 .7365 .7370 .7380 .7390	.7365 .7375 .7380 .7390 .7400	.7375 .7385 .7390 .7400 .7410	.7385 .7395 .7400 .7410 .7420	.7395 .7405 .7410 .7420 .7430	.7405 .7415 .7420 .7430 .7430 .7440	.7415 .7425 .7435 .7435 .7440 .7450	.7425 .7435 .7445 .7450 .7460	.7435 .7445 .7455 .7460 .7470	.7445 .7455 .7465 .7470 .7470
60 62 64 66 68	.7400 .7410 .7415 .7425 .7435	.7410 .7420 .7425 .7435 .7445	.7420 .7430 .7435 .7445 .7455	.7430 .7440 .7445 .7455 .7465	.7440 .7450 .7455 .7465 .7475	.7450 .7460 .7465 .7475 .7485	.7460 .7470 .7475 .7485 .7485	.7470 .7480 .7485 .7495 .7505	.7480 .7490 .7495 .7505 .7515	.7490 .7500 .7500 .7511 .7512
70 72 74 76 78	.7445 .7450 .7460 .7470 .7480	.7455 .7450 .7470 .7480 .7480 .7490	.7465 .7470 .7480 .7490 .7500	.7475 .7480 .7490 .7500 .7510	.7485 .7490 .7500 .7510 .7520	.7495 .7500 .7510 .7510 .7510 .7530	.7505 .7510 .7520 .7530 .7530 .7540	.7515 .7520 .7530 .7540 .7550	.7525 .7530 .7540 .7550 .7560	.753 .7540 .7550 .7560 .7570
80 82 84 86 88	.748 .749 .750 .751 .752	.749 .750 .751 .752 .753	.750 .751 .752 .753 .754	.751 .752 .753 .754 .755	.752 .753 .754 .755 .750	.753 .754 .755 .756 .756	.754 .755 .756 .757 .758	.755 .756 .757 .758 .759	.756 .757 .758 .759 .760	.757 .758 .759 .760 .761
90 92 94 96 98	.753 .754 .755 .755 .756	.754 .755 .756 .756 .757	.755 .756 .757 .757 .757 .758	.756 .757 .758 .758 .758	.757 .758 .759 .759 .759 .760	.758 .759 .760 .760 .761	.759 .759 .760 .761 .762	.760 .760 .761 .762 .763	.761 .761 .763 .763 .764	.762 .762 .763 .764 .765
00 02 04 06 08	.757 .758 .759 .760 .760	.758 .759 .760 .761 .761	.759 .760 .761 .762 .762	.760 .761 .762 .763 .763	.761 .762 .763 .764 .764	.762 .763 .764 .765 .765	.763 .764 .764 .765 .766	.764 .765 .765 .766 .767	.765 .766 .676 .767 .768	.766 .767 .767 .768 .769
10 12 14 16 18	.761 .763 .763 .764 .765	.762 .763 .764 .765 .766	.763 .764 .765 .766 .767	.764 .765 .766 .767 .768	.765 .766 .767 .768 .769	.766 .767 .768 .769 .770	.767 .768 .768 .769 .770	.768 .769 .769 .770 .771	.769 .770 .770 .771 .772	.770 .771 .771 .771 .772 .773
20	.765	.766	.767	.768	.769	.770	.771	.772	.773	.774

			84 B.							
Observed temperature in °F	0.750	0.751	0.752	0.753	0.754	0.755	0.756	0.757	0.758	0.759
			Corresp	onding	specific	gravitie	s at 60°	/60° F		
30 32	0.736	0.737	0.738	0.739 .740	0.740	0.741	0.742 .743	0.743 .744	0.744 .745	0.745 .746
34 36 38	.738 .739 .740	.739 .740 .741	.740 .741 .742	.741 .742 .743	.742 .743 .744	.743 .744 .745	.744 .745 .746	.745 .746 .747	.746 .747 .748	.747 .748 .749
40 42 44 46 48	.7410 .7420 .7430 .7440 .7445	$\begin{array}{r} .7420 \\ .7430 \\ .7440 \\ .7450 \\ .7455 \end{array}$	$\begin{array}{r} .7430 \\ .7440 \\ .7450 \\ .7460 \\ .7465 \end{array}$.7440 .7450 .7460 .7470 .7475	.7450 .7430 .7470 .7480 .7485	.7460 .7470 .7480 .7490 .7495	.7475 .7480 .7490 .7500 .7510	.7485 .7490 .7500 .7510 .7520	.7495 .7500 .7510 .7520 .7530	.7505 .7510 .7520 .7530 .7540
50 52 54 56 58	.7455 .7465 .7475 .7480 .7490	.7465 .7475 .7485 .7490 .7500	.7475 .7485 .7405 .7500 .7510	.7485 .7495 .7505 .7510 .7520	.7495 .7505 .7515 .7520 .7530	.7505 .7515 .7525 .7530 .7540	$\begin{array}{r} .7515 \\ .7525 \\ .7535 \\ .7540 \\ .7550 \end{array}$.7525 .7535 .7545 .7550 .7560	.7535 .7545 .7555 .7560 .7570	.7545 .7555 .7565 .7565 .7570 .7580
60 62 64 66 68	.7500 .7510 .7515 .7525 .7535	.7510 .7520 .7525 .7535 .7545	.7520 .7530 .7535 .7545 .7555	.7530 .7540 .7545 .7555 .7565	.7540 .7550 .7555 .7565 .7575	.7550 .7560 .7565 .7575 .7585	.7560 .7570 .7575 .7585 .7580	.7570 .7580 .7585 .7595 .7600	.7580 .7590 .7595 .7605 .7610	.7590 .7600 .7605 .7615 .7620
70 72 74 76 78	.7545 .7550 .7560 .7570 .7580	.7555 .7560 .7570 .7580 .7590	.7565 .7570 .7580 .7590 .7590	.7575 .7580 .7590 .7600 .7610	.7585 .7590 .7600 .7610 .7620	.7595 .7600 .7610 .7620 .7630	.7600 .7610 .7615 .7625 .7635	.7610 .7620 .7625 .7635 .7645	.7620 .7630 .7635 .7645 .7655	.7630 .7640 .7645 .7655 .7655
80 82 84 86 88	.758 .759 .760 .761 .762	.759 .760 .761 .762 .763	.760 .761 .762 .763 .764	.761 .762 .763 .764 .765	.732 .763 .764 .765 .766	.763 .764 .765 .766 .767	.764 .765 .766 .767 .767	.765 .766 .767 .768 .768	.766 .767 .768 .769 .769	.767 .768 .769 .770 .770
90 92 94 96 98	.763 .763 .764 .765 .766	.764 .764 .765 .766 .767	.765 .765 .766 .766 .767 .768	.766 .766 .767 .768 .769	.767 .767 .768 .769 .770	.768 .768 .769 .770 .771	.768 .769 .770 .771 .771	.769 .770 .771 .772 .772	.770 .771 .772 .773 .773	.771 .772 .773 .774 .774
00 02 04 06 08	.767 .768 .768 .769 .770	.768 .769 .769 .770 .771	.769 .770 .770 .771 .772	.770 .771 .771 .772 .773	.771 .772 .772 .773 .773 .774	.772 .773 .773 .774 .775	.772 .773 .774 .775 .775	.773 .774 .775 .776 .776	.774 .775 .776 .777 .777	.775 .776 .777 .778 .778
10 112 114 116 118	.771 .772 .772 .773 .774	.772 .773 .773 .773 .774 .775	.773 .774 .774 .775 .776	.774 .775 .775 .776 .777	.775 .776 .776 .777 .777 .778	.776 .777 .777 .778 .778 .779	.776 .777 .778 .779 .780	.777 .778 .779 .780 .781	.778 .779 .780 .781 .782	.779 .780 .781 .782 .783
20	.775	.776	.777	.778	.779	.780	.780	.781	.782	.783

				Observ	ed spec.	ific grav	VILIES			
Observed temperature in °F	0.760	0.761	0.762	0.763	0.764	0.765	0.766	0.767	0.768	0.769
		(Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30 32 34 36 38	0.746 .747 .748 .749 .750	0.747 .748 .749 .750 .751	0.748 .749 .750 .751 .752	0.749 .750 .751 .752 .753	0.750 .751 .752 .753 .754	0.751 .752 .753 .754 .755	0.753 .754 .755 .756 .757	0.754 .755 .756 .757 .758	0.755 .756 .757 .758 .759	0.756 .757 .758 .759 .760
40 42 44 46 48	.7515 .7520 .7530 .7540 .7550	.7525 .7530 .7540 .7550 .7560	.7535 .7540 .7550 .7560 .7560 .7570	.7545 .7550 .7560 .7570 .7580	.7555 .7560 .7570 .7570 .7580 .7590	.7565 .7570 .7580 .7590 .7600	.7575 .7585 .7590 .7600 .7610	.7585 .7595 .7600 .7610 .7620	.7595 .7605 .7610 .7620 .7630	.7605 .7615 .7620 .7630 .7640
50 52 54 56 58	.7555 .75.5 .7575 .7580 .7590	.7565 .7575 .7585 .7590 .7600	.7575 .7585 .7595 .7600 .7610	.7585 .7595 .7005 .7610 .7620	.7595 .7605 .7615 .7620 .7630	.7605 .7615 .7625 .7630 .7640	.7620 .7625 .7635 .7645 .7650	.7630 .7635 .7645 .7655 .7650	.7640 .7645 .7655 .7665 .7670	.7650 .7653 .7666 .7675 .7675
60 62 64 66 68	.7600 .7610 .7615 .7625 .7630	.7610 .7620 .7025 .7635 .7640	.7620 .7630 .7635 .7645 .7650	.7630 .7640 .7645 .7655 .7650	.7640 .7650 .7655 .7665 .7670	.7650 .7660 .7665 .7675 .7680	.7630 .7670 .7675 .7685 .7690	.7670 .7680 .7685 .7695 .7700	.7680 .7090 .7695 .7705 .7710	.7690 .7700 .7706 .7715 .7715
70 72 74 78	.7640 .7650 .7655 .7665 .7675	.7650 .7660 .7665 .7675 .7685	.7660 .7670 .7675 .7685 .7695	.7670 .7680 .7685 .7695 .7705	.7680 .7090 .7695 .7705 .7715	.7690 .7700 .7705 .7715 .7725	.7700 .7710 .7715 .7725 .7735	.7710 .7720 .7725 .7735 .7745	.7720 .7730 .7735 .7745 .7755	.7730 .7740 .7745 .7758 .7762
80 82 84 86 88	.768 .769 .770 .771 .771	.769 .770 .771 .772 .772	.770 .771 .772 .773 .773	.771 .772 .773 .774 .774	.772 .773 .774 .775 .775	.773 .774 .775 .776 .776	.774 .775 .776 .776 .776 .777	.775 .776 .777 .777 .777 .778	.776 .777 .778 .778 .778 .779	.777 .778 .779 .778 .778
90 92 94 96 98	.772 .773 .774 .775 .775	.773 .774 .775 .776 .776	.774 .775 .776 .777 .777	.775 .776 .777 .778 .778	.776 .777 .778 .779 .779	.777 .778 .779 .780 .780	.778 .779 .780 .780 .781	.779 .780 .781 .781 .781	.780 .781 .782 .782 .782 .783	.781 .782 .783 .783 .783 .784
100 102 104 103	.776 .777 .778 .779 .779	.777 .778 .779 .780 .780	.778 .779 .780 .781 .781	.779 .780 .781 .782 .782	.780 .781 .782 .783 .783	.781 .782 .783 .784 .784	.782 .783 .784 .784 .785	.783 .784 .785 .785 .785 .786	.784 .785 .786 .786 .786 .787	.785 .786 .787 .787 .787 .788
110 112 114 116 118	.780 .781 .782 .783 .783 .784	.781 .782 .783 .784 .785	.782 .783 .784 .785 .785	.783 .784 .785 .786 .786	.784 .785 .786 .787 .788	.785 .786 .787 .788 .789	.786 .787 .787 .788 .788 .789	.787 .788 .788 .789 .789	.788 .789 .789 .790 .791	.789 .790 .790 .791 .791 .792
20	.784	.785	.786	.787	.788	.789	.790	.791	.792	.793

		4		Observ	ed spec	ific gra	VILLES			
Observed temperature in °F	0.770	0.771	0.772	0.773	0.774	0.775	0.776	0.777	0.778	0.779
			Corresp	onding	specific	gravitie	es at 60°	°/60° F	1	
30	0.757	0.758	0.759	0.760	0.761	0.762	0.763	0.764	0.765	0.766
32	.758	.759	.760	.761	.762	.763	.764	.765	.706	.767
34	.759	.760	.761	.762	.763	.764	.765	.766	.767	.768
36 38	.760 .761	.761 .762	.762 .763	.763 .764	.764 .765	.765 .766	.766 .767	.767 .768	.768 .769	.769 .770
40	.7615	.7625	.7635	.7645	.7655	.7665	.7675	.7685	.7695	.770
42	.7625	.7635	.7645	.7655	.7665	.7675	.7685	.7695	.7705	.771
44	.7630 .7640	.7640 .7650	.7650	.7665 .7670	.7670	.7680 .7690	.7695 .7700	.7705 .7710	.7715 .7720	.772 .7730
46 48	.7650	.7660	.7670	.7680	.7690	.7700	.7710	.7720	.7730	.7740
50	.7660	.7670	.7680	.7690	.7700	.7710	.7720	.7730	.7740	.7750
53	.7665	.7675	.7685	.7695	.7705	.7715	7725	.7735	.7745	.7758
54	.7675	.7685 .7695	.7695	.7705	.7715 7725	.7725 .7735	.7735 .7745	.7745	.7755	.7765
56 58	.7685 .7690	.7700	.7705 .7710	.7715 .7720	.7730	.7740	.7750	.7760	.7770	.7778
60	.7700	.7710	.7720	.7730	.7740	.7750	.7760	.7770	.7780	.7790
62	.7710	.7720	.7730	.7740	.7750	.7760	.7770	.7780	.7790	.7800
64 66	.7715 .7725	.7725	.7735 .7745	.7745 .7755	.7755 .7765	.7765 .7775	.7775 .7785	.7785 .7795	.7795 .7805	.7805
68	.7730	.7740	.7750	.7760	.7770	.7780	.7790	.7800	.7810	.7820
70	.7740	.7750	.7760	.7770	.7780	.7790	.7800	.7810	.7820	.7830
72	.7750 .7755	.7760	.7770	.7780	.7790 .7795	.7800 .7805	.7810 .7815	.7820 .78°5	.7830 .7835	.7840
74	.7765	.7765 .7775	.7775 .7785	.7785 .7795	.7805	.7815	.7825	.7835	.7845	.7855
78	.7775	.7785	.7795	.7805	.7815	.7825	.7835	.7845	.7855	.7865
80	.778	.779	.780	.781	.782	.783	.784	.785	.786	.787
82 84	.779 .780	.780	.781 .782	.782 .783	.783 .784	.784 .785	.785 .785	.786 .786	.787 .787	.788
86	.780	.781	.782	.783	.784	.785	.786	.787	.788	.789
88	.781	.782	.783	.784	.785	.786	.787	.788	.789	.790
90	.782	.783	.784	.785	.786 .787	.787	.788	.789	.790	.791
92 94	.783 .784	.784 .785	.785 .783	.786 .787	.788	.788 .789	.789 .789	.790 .790	.791	$.792 \\ .792$
96	.784	.785	.786	.787	.788	.789	.790	.791	.792	.793
98	.785	.786	.787	.788	.789	.790	.791	.792	.793	.794
00	.786 .787	.787 .788	.788 .789	.789	.790 .791	.791 .792	.792 .792	.793	.794	.795
04	.788	.788	.789	.790 .791	.791	.792	.792	.793	.794	.795
06	.788	.789	.790	.791	.792	.793	.794	.795	.796	.797
08	.789	.790	.791	.792	.793	.794	.795	.796	.797	.798
10	.790	.791	.792	.793	.794	.795	.795	.796	.797	.798
12	.791	.792	.793	.794	.795	.796	.796	.797	.798	.799
14	.791	.792	.793	.794	.795	.796	.797	.798	.799	.800 .801
16 18	.792 .793	.793	.794 .795	.795 .796	.790	.797 .798	.798 .799	.799 .800	.800 .801	.801
.20	.794	.795	.796	.797	.798	.799	.799	.800	.801	.802

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.780	0.781	0.782	0.783	0.784	0.785	0.786	0.787	0.788	0.789
			Corresp	onding	specific	gravitie	es at 60°	°/60° F		
30 32 34 36 38	0.767 .768 .769 .770 .771	0.768 .769 .770 .771 .771	0.769 .770 .771 .772 .773	0.770 .771 .772 .773 .774	0.771 .772 .773 .774 .775	0.772 .773 .774 .775 .776	0.773 .774 .775 .776 .777	0.774 .775 .776 .777 .778	0.775 .776 .777 .778 .779	0.776 .777 .778 .779 .780
40 42 44 46 48	.7715 .7725 .7735 .7740 .7750	.7725 .7735 .7745 .7750 .7760	.7735 .7745 .7755 .7760 .7770	.7745 .7755 .7765 .7770 .7770	.7755 .7765 .7775 .7780 .7790	.7765 .7775 .7785 .7790 .7800	.7780 .7785 .7795 .7805 .7810	.7790 .7795 .7805 .7815 .7820	.7800 .7805 .7815 .7825 .7830	.7810 .7815 .7825 .7835 .7836
50 52 54 53 58	.7760 .7765 .7775 .7785 .7790	.7770 .7775 .7785 .7795 .7800	.7780 .7785 .7795 .7805 .7810	.7790 .7795 .7805 .7815 .7820	.7800 .7805 .7815 .7825 .7830	.7810 .7815 .7825 .7835 .7840	.7820 .7830 .7835 .7845 .7850	.7830 .7840 .7845 .7855 .7860	.7840 .7850 .7855 .7865 .7870	.7850 .7860 .7865 .7875 .7875 .7890
60 62 64 66 68	.7800 .7810 .7815 .7825 .7830	.7810 .7820 .7825 .7835 .7840	.7820 .7830 .7835 .7845 .7845 .7850	.7830 .7840 .7845 .7855 .7800	.7840 .7850 .7855 .7865 .7870	.7850 .7860 .7865 .7875 .7880	.7860 .7865 .7875 .7885 .7890	.7870 .7875 .7885 .7895 .7900	.7880 .7885 .7895 .7905 .7910	.7890 .7895 .7905 .7915 .7920
70 72 74 76 78	.7840 .7850 .7855 .7865 .7865 .7875	.7850 .7860 .7865 .7875 .7875 .7885	.7860 .7870 .7875 .7885 .7895	.7870 .7880 .7885 .7895 .7905	.7880 .7890 .7895 .7905 .7915	.7890 .7900 .7905 .7915 .7925	.7900 .7905 .7915 .7925 .7930	.7910 .7915 .7925 .7935 .7940	.7920 .7925 .7935 .7945 .7450	.7930 .7935 .7945 .7955 .7960
80 82 84 86 88	.788 .789 .789 .790 .791	.789 .790 .790 .791 .791 .792	.790 .791 .791 .792 .793	.791 .792 .792 .793 .793	.792 .793 .793 .794 .795	.793 .794 .794 .795 .796	.794 .794 .795 .796 .797	.795 .795 .796 .797 .798	.796 .796 .797 .798 .798	.797 .797 .798 .799 .800
90 92 94 96 98	.792 .793 .793 .794 .795	.793 .794 .794 .795 .795	.794 .795 .795 .796 .797	.795 .796 .796 .797 .797 .798	.796 .797 .797 .798 .798	.797 .798 .798 .799 .800	.798 .798 .799 .800 .801	.799 .799 .800 .801 .802	.800 .800 901 .802 .803	.801 .801 .802 .803 .804
100 102 104 106 108	.796 .796 .797 .798 .799	.797 .797 .798 .799 .800	.798 .798 .799 .800 .801	.799 .799 .800 .801 .802	.800 .800 .801 .802 .803	.801 .801 .802 .803 .804	.801 .802 .803 .804 .804	.802 .803 .804 .805 .805	.803 .804 .805 .806 .806	.804 .805 .806 .807 .807
110 112 114 116 118	.799 .800 .801 .802 .803	.800 .801 .802 .803 .804	.801 .802 .803 .804 .805	.802 .803 .804 .805 .806	.803 .804 .805 .806 .807	.804 .805 .806 .807 .808	.805 .806 .807 .807 .808	.806 .807 .808 .808 .808 .809	.807 .808 .809 .809 .809 .810	.808 .809 .810 .810 .811
120	.803	.804	.805	.806	.807	.808	.809	.810	.811	.812

BULLETIN NUMBER FIFTEEN OF

0.790 0.777 .778 .779 .780 .781 .7820 .7825	0.791 .0778 .779 .780 .781 .782	0.779 .780 .781	0.793 onding 0.780 .781	0.794 specific 0.781	0.795 gravitie	0.796 s at 60°	0.797 /60° F	0.798	0.799
.778 .779 .780 .781 .7820	.0778 .779 .780 .781	0.779 .780 .781	0.780		gravitie	s at 60°	/60° F		
.778 .779 .780 .781 .7820	.779 .780 .781	.780 .781		0.781					
.778 .779 .780 .781 .7820	.779 .780 .781	.780 .781			0.782	0.784	0.785	0.786	0.787
.779 .780 .781 .7820	.780 .781	.781		.782	.783	.784	.785	.786	.787
.781 .7820			.782	.783	.784	.785	.786	.787	.788
.7820	.782	.782	.783	.784	.785	.786	.787	.788	.789
		.783	.784	.785	.786	.787	.788	.789	.790
	.7830	.7840	.7850	.7800	.7870	.7880	.7890	.7900	.7910
	.7835	.7845	.7855	.7865	.7875	.7890	.7900	.7910	.7920
.7835	.7845	.7855	.7865	.7875	.7885	.7895	.7905	.7915	.7925
.7845	.7855	.7865	.7875	.7885	.7895	.7905	.7915	.7925	.7935
.7850	.7860	.7870	.7880	.7890	.7900	.7910	.7920	.7930	.7940
7960	7070	7000	7800	7000	7010	7020	7020	7040	7050
									.7950
									.7965
									.7975
.7890	.7900	.7910	.7920	.7930	.7940	.7955	.7965	.7975	.7985
7900	7010	7990	7930	7040	7950	7060	7970	7080	.7990
									.7995
									.8005
									.8015
.7930	.7940	.7950	.7960	.7970	.7980	.7990	.8000	.8010	.8020
7940	7950	7960	7970	7980	7990	8000	8010	8020	.8030
									.8035
.7955	.7965	.7975	.7985	.7995	.8005	.8015	.8025	.8035	.8045
.7965	.7975	.7985	.7995	.8005	.8015	.8020	.8030	.8040	.8050
.7970	.7980	.7990	.8000	.8010	.8020	.8030	.8040	.8050	.8000
.798	.799	.800	.801	.802	.803	.804	.805	.806	.807
.798	.799	.800	.801	.802		.804	.805	.806	.807
									.808
									.809
.801	.802	.803	.804	805	.806	.807	.808	.809	.810
.802	.803	.804	.805	.806	.807	.808	.809	.810	.811
									.811
									.812
									.813
.805	.806	.807	.808	.809	.810	.811	.812	.813	.814
.805	.806	.807	.808	.809	.810	.811	.812	.813	.814
		.808	.809	.810	.811	.812	.813	.814	.815
									.816
									.816
.808	.809	.810	.811	.812	.813	.814	.815	.816	.817
.809	.810	.811	.812	.813	.814	.815	.816	.817	.818
.810	.811	.812	.813	.814	.815	.816	.817	.818	.819
.811	.812	.813	.814	.815	.816	.816	.817	.818	.819
.811		.813	.814	.815	.816	.817	.818	.819	.820
.812	.813	.814	.815	.816	.817	.818	.819	.820	.821
.813	.814	.815	.816	.817	.818	.819	.820	.821	.822
	.7860 .7870 .7870 .7885 .7885 .7985 .7915 .7925 .7925 .7925 .7925 .7925 .7925 .7940 .7945 .7940 .7945 .7945 .7945 .7945 .7945 .795 .7970 .798 .798 .800 .801 .802 .802 .804 .805 .805 .805 .805 .805 .805 .805 .808 .808	7890 .7870 .7870 .7880 .7875 .7885 .7885 .7895 .7880 .7900 .7900 .7910 .7905 .7915 .7915 .7925 .7925 .7933 .7940 .7950 .7945 .7955 .7955 .7955 .7965 .7975 .7965 .7975 .7970 .7980 .798 .799 .798 .799 .798 .799 .801 .802 .802 .803 .804 .805 .805 .806 .805 .806 .806 .807 .808 .809 .808 .809 .808 .809 .808 .809 .808 .809 .809 .811 .811 .812 .811 .812	7.7800 .77870 .77880 7.7870 .7880 .7890 7.7875 .7895 .7805 7.7850 .7505 .7505 7.7850 .7505 .7505 7.7850 .7505 .7505 .7900 .7910 .7920 .7905 .7915 .7925 .7925 .7935 .7945 .7925 .7935 .7945 .7920 .7940 .7950 .7945 .7955 .7965 .7055 .7965 .7965 .7970 .7980 .7090 .798 .799 .800 .798 .799 .800 .798 .799 .800 .798 .799 .800 .799 .800 .801 .801 .802 .803 .801 .802 .803 .802 .803 .804 .803 .804 .805 .805	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	7860 77870 77880 77890 77900 77870 77885 77890 77900 77910 77875 77885 77805 77905 77915 77920 77885 77800 77910 77920 77920 77930 77880 77900 77910 77920 77930 77940 77905 77915 77925 77935 77945 77945 77905 77945 77945 77945 77955 77965 77940 77950 77955 77965 77965 77965 77965 77940 77955 77955 77955 77965 79955 79955 77955 77955 77955 77955 79955 79955 79955 77975 77955 77955 77955 79955 79955 79955 77975 77955 77955 79955 79955 79955 79955 77970 77980 7975 779	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	7800 77800 77800 77800 77910 77920 77935 77945 77955 77945 77955	7800 .7870 .7880 .7890 .7910 .7920 .7920 .7940 .7940 .7870 .7885 .7885 .7805 .7905 .7910 .7920 .7930 .7940 .7955 .7875 .7885 .7805 .7905 .7915 .7925 .7935 .7945 .7955 .7965 .7880 .7000 .7910 .7920 .7930 .7940 .7955 .7965 .7975 .7985 .7800 .7910 .7920 .7930 .7940 .7955 .7965 .7975 .7985 .8005 .8010 .8020 .8030 .8040 .8055 .8055 .8015 .8025 .8035 .8045 .8055 .8055 .8065

				Observ	ed spec	ific grav	VILLES			
Observed emperature in °F	0.800	0.801	0.802	0.803	0.804	0.805	0.806	0.807	0.808	0.809
		(Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30	0.788	0.789	0.790	0.791	0.792	0.793	0.794	0.795	0.793	0.797
32	.788	.789	.790	.791	.792	.793	.795	.796	.797	.798
4	.789	.790	.791	.792	.793	.794	.795	.796	.797	.796
6	.790	.791	.792	.793	.794	.795	.796	.797	.798	.799
8	.791	.792	.793	7.94	.795	.796	.797	.798	.799	.800
0	.7920	.7930	.7940	.7450	.7560	.7970	.7980	.7990	.8000	.801
2	.7930	.7940	.7950	.7960	.7970	.7980	.7990	.8000	.8010	.80
4	.7935	.7945	.7955	.7965	.7975	.7985	.7995	.8005	.8015	.805
ô 8	.7945 .7950	.7955	.7965	.7975 .7980	.7985 .7990	.7995	.8005 .8010	.8015 .8020	.8025 .8030	.80
0	.7930	.7970	.7980	.7990	.8000	.8010	.8020	.8030	.8040	.803.
2	.7970 .7975	.7980 .7985	.7990 .7995	.8000	.8010	.8020	.8030	.8040	.8050 .8055	.800
6	.7985	.7995	.8005	.8015	.8025	.8035	.8045	.8055	.8065	.80
8	.7995	.8005	.8015	.8025	.8035	.8045	.8055	.8065	.8075	.808
0	.8000	.8010	.8020	.8030	.8040	.8050	.8060	.8070	.8080	.80
2	.8005	.8015	.8025	.8035	.8045	.8055	.8065	.8075	.8085	.80
4	.8015	.8025	.8035	.8045	.8055	.8065	.8075	.8085	.8095	.810
J	.8025	.8035	.8045	.8055	.8065	.8075	.8085	.8095	.8105	.811
8	.8030	.8040	.8050	.8060	.8070	.8080	.8090	.8100	.8110	.81:
0	.8040	.8050	.8060	.8070	.8080	.8090	.8100	.8110	.8120	.81
2	.8045	.8055	.8065	.8075	.8085	.8095	.8105	.8115		.81
4	.8055	.8065	.8075	.8085	.8095	.8105	.8115	.8125	.8135	.81
6	.8035 .8070	.8075 .8080	.8085 .8090	.8095 .8100	.8105	.8115 .8120	.8120 .8130	.8130 .8140	.8140 .8150	.81
0	.808	.809	.810	.811	.812	.813	.813	.814	.815	.816
2	.808	.809	.810	.811	.812	.813	.814	.815	.816	.81
4	.809	.810	.811	.812	.813	.814	.815	.816	.817	.818
6	.810	.811	.812	.813	.814	.815	.816	.817	.818	.819
8	.811	.812	.813	.814	.815	.816	.816	.817	.818	.819
0	.812	.813	.814	.815	.816	.817	.817	.818	.819	.820
2	.812	.813	.814	.815	.816	.817	.818	.819	.820	.821
4	.813	.814	.815	.816	.817	.818	.819	.820	.821	.82
6 8	.814 .815	.815 .816	.816 .817	.817 .818	.818 .819	.819 .820	.819 .820	.820 .821	.821 .822	.82
0	.815	.816	.817	.818	.819	.820	.821	.822	.823	.824
2	.813	.817	.818	.819	.819	.821	.822	.823	.824	.82
4	.817	.818	.819	.820	.821	.822	.822	.823	.824	.82
6	.817	.818	.819	.820	.821	.822	.823	.824	.825	.826
8	.818	.819	.820	.821	.822	.823	.824	.825	.826	.827
0	.819	.870	.821	.822	.823	.824	.825	.826	.827	.828
2	.820	.821	.822	.823	.824	.825	.825	.826	.827	.822
4	.820	.821	.822	.823	-824	.825	.826	.827	.828	.829
6	.821	.822	.823	.824	.825	.826	.827	.828	.829	.830
.8	.822	.823	.824	.825	.826	.827	.828	.829	.830	.831
0	.823	.824	.825	.826	.827		.828	.829	.830	.831

				Observ	red spec	eific gra	vities			
Observed temperature in °F	0.810	0.811	0.812	0.813	0.814	0.815	0.816	0.817	0.818	0.819
			Corresp	onding	specific	gravitie	es at 60°	'/60° F		
30	0.798	0.799	0.800	0.801	0 802	0.803	0.804	0.805	0.806	0.807
32	.799	.800	.801	.802	.803	.804	.805	.806	.807	.808
34 36	.799 .800	.800 .801	.801 .802	.802	.803	.804	.806 .807	.807 .808	.808	.809
38	.800	.802	.803	.804	.805	.806	.808	.809	.810	.810
40	.8020	.8030	.8040	.8050	.8030	.8070	9095	2005	9105	
40 42	.8020	.8030	.8040	.8050	.8050	.8070	.8085	.8095	.8105	.8115 .8120
44	.8035	.8045	.8055	.8065	.8075	.8085	.8100	8110	.8120	.8120
46	.8045	.8055	.8065	.8075	.8085	,8095	.3105	.8115	.8125	.8135
48	.8050	.8060	.8070	.8080	.8090	.8100	.8115	.8125	.8135	.8145
50	.8060	.8070	.8080	.8090	.8100	.8110	.8120	.8130	.8146	.8150
52	.8070	.8080	.8090	.8100	.8110	.8120	.8130	.8140	.8150	.8160
54	.8075	.8085	.8095	.8105	.8115	.8125	.8135	.8145	.8155	.8165
56 58	.8085 .8095	.8095 .8105	.8105 .8115	.8115 .8125	.8125 .8135	.8135 .8145	.8145 .8155	.8155 .8165	.8165 .8175	.8175 .8185
60	.8100	.8110	.8120	.8130	.8140	.8150	.8160	.8170	.8180	.8190
62	.8105	.8115	.8125	.8135	.8145	.8155	.8165	.8175	.8185	.8195
64 66	.8115 .8125	.8125 .8135	.8135 .8145	.8145 .8155	.8155	.8165 .8175	.8175 .8180	.8185 .8190	.8195 .8200	.8205 .8210
68	.8120	.8130	.8145	.8155	.8100	.8175	.8180	.8190	.8200	.8210
70	.8140	.8150	.8160	.8170	.8180	.8190	.8200	.8210	.8220	.8230
72	.8145	.8155	.8165	.8175	.8180	.8190	.8200	.8210	.8220	.8230
74	.8155	.8165	.8175	.8185	.8195	.8205	.8215	.8225	.8235	.8245
76	.8160	.8170	.8180	.8190	.8200	.8210	.82:20	.8230	.8240	.8250
78	.8170	.8180	.8190	.8200	.8210	.8220	.8230	.8240	.8250	.8260
80	.817	.818	.819	.820	.821	.822	.823	.824	.825	.826
82	.818	.819	.820	.821	.822	.823	.824	.825	.826	.827
84 86	.819 .820	.820 .821	.821 .822	.822 .823	.823 .824	.824 .825	.825 .826	.826 .827	.827 .828	.828 .829
88	.820	.821	.822	.823	.824	.825	.826	.827	.828	.829
90	.821	.822	.823							
92	.821	.822	.823	.824	.825 .826	.826 .827	.827 .828	.828 .829	.829 .830	.830 .831
94	.823	.824	.825	.826	.827	.828	.828	.829	.830	.831
96	.823	.824	.825	.826	.827	.828	.829	.830	.831	.832
98	.824	.825	.826	.827	.828	.829	.830	.831	.832	.833
100	.825	.826	.827	.828	.829	.830	.831	.832	.833	.834
102	.826	.827	.828	.829	.830	.831	.831	.832	.833	.834
104	.826	.827	.828	.829	.830	.831	.832	.833	.834	.835
106 108	.827 .828	.828 .829	.829 .830	.830 .831	.831 .832	.832	.833 .834	.834	.835	.836
		.010	.000	.001	.892	.833	.804	.835	.833	.837
110	.829	.830	.831	.832	.833	.834	.834	.835	.836	.837
112	.829	.830	.831	.832	.833	.834	.835	.836	.837	.838
114					.834					.839
118										.839 .840
120	.832	.833	.834	.835	.836	.837	.838	.839	.840	.841
112 114 116 118 120	.829 .830 .831 .832 .832	.830 .831 .832 .833 .833	.831 .832 .833 .834 .834	.832 .833 .834 .835 .835	.833 .834 .835 .836 .836	.834 .835 .839 .837 .837	.836 .836 .837	.837 .837 .838	.837 .838 .838 .839 .839	.8. .8. .8

				Observ	red spec	ufic gra	vities			
Observed temperature in °F	0.820	0.821	0.822	0.823	0.824	0.825	0.826	0.827	0.828	0.829
			Corresp	onding	specific	gravitie	es at 60°	°/60° F		
30 32 34 36 38	0.808 .809 .810 .811 .812	0.809 .810 .811 .812 .813	0.810 .811 .812 .813 .814	0.811 .812 .813 .814 .815	0.812 .813 .814 .815 .816	0.813 .814 .815 .816 .817	0.814 .815 .816 .817 .818	0.815 .816 .817 .818 .819	0.816 .817 .818 .819 .820	0.817 .818 .819 .820 .821
40 42 44 45 48	.8125 .8130 .8140 .8145 .8155	.8135 .8140 .8150 .8155 .8165	.8145 .8150 .8160 .8165 .8175	.8155 .8160 .8170 .8175 .8185	.8165 8170 .8180 .8185 .8195	.8175 .8180 .8190 .8195 .8205	.8185 .8190 .8200 .8205 .8215	.8195 .8200 .8210 .8215 .8225	.8205 .8210 .8220 .8225 .8235	.8215 .8220 .8230 .8235 .8240
50525456565885888588588858858858885888588858885888588858885	.8160 .8170 .8175 .8185 .8195	.8170 .8180 .8185 .8195 .8205	.8180 .8190 .8195 .8205 .8215	.8190 .8200 .8205 .8215 .8225	.8700 .8210 .8215 .8225 .8235	.8210 .8220 .8225 .8235 .8245	.8220 .8230 .8240 .8245 .8255	.8230 .8240 .8250 .8255 .8265	.8240 .8250 .8260 .8265 .8275	.8250 .8200 .8270 .8275 .8285
$ \begin{array}{c} 60\\ 62\\ 64\\ 66\\ 68\\ \end{array} $.8200 .8205 .8215 .8220 .8230	.8210 .8215 .8225 .8230 .8240	.8220 .8225 .8235 .8235 .8240 .8250	.8230 .8235 .8245 .8250 .8250 .8260	.8240 .8245 .8255 .8260 .8270	.8250 .8255 .8265 .8270 .8280	.8260 .8265 .8275 .8280 .8290	.8270 .8275 .8285 .8200 .8300	.8280 .8285 .8295 .8300 .8310	.8290 .8295 .8305 .8310 .8320
70 72 74 76 78	.8240 .8245 .8255 .8260 .8270	.8250 .8255 .8265 .8270 .8280	.8260 .8265 .8275 .8280 .8290	.8270 .8275 .8285 .8290 .8300	.8290 .8385 .8295 .8300 .8310	.8290 .8295 .8305 .8310 .8320	.8300 .8305 .8315 .8320 .8330	.8310 .8315 .8325 .8330 .8340	.8320 .8325 .8335 .8340 .8350	.8330 .8335 .8345 .8350 .8360
80 82 84 87 88	.827 .828 .829 .830 .830	.828 .829 .830 .831 .831	.829 .830 .831 .832 .832	.830 .831 .832 .833 .833	.831 .832 .833 .834 .834	.832 .833 .834 .835 .835	.833 .834 .835 .835 .835 .836	.834 .835 .836 .836 .837	.835 .836 .837 .837 .837 .838	.836 .837 .838 .838 .838 .839
90 92 94 96 98	.831 .832 .832 .833 .833 .834	.832 .833 .833 .834 .834	.833 .834 .834 .835 .836	.834 .835 .835 .836 .837	.835 .836 .836 .837 .838	.836 .837 .837 .838 .838 .839	.837 .838 .838 .839 .839 .840	.838 .839 .839 .840 .841	.839 .840 .840 .841 .842	.840 .841 .841 .842 .843
100 102 104 106 108	.835 .835 .836 .837 .838	.836 .836 .837 .838 .839	.837 .837 .838 .839 .840	.838 .838 .839 .840 .841	.839 .839 .840 .841 .842	.840 .840 .841 .842 .843	.840 .841 .842 .843 .843	.841 .842 .843 .844 .844	.842 .843 .814 .845 .845	.843 .844 .845 .846 .846
110 112 114 116 118	.838 .839 .840 .840 .841	.839 .840 .841 .841 .842	.840 .841 .842 .842 .843	.841 .842 .843 .843 .844	.842 .843 .844 .844 .845	.843 .844 .845 .845 .845 .846	.844 .845 .846 .846 .847	.845 .846 .847 .847 .848	.846 .847 .848 .848 .849	.847 .848 .849 .849 .850
120	.842	.843	.844	.845	.846	.847	.848	.849	.850	.851

Observation				Ubserv	eu spec	ific gra	vicies		, <u> </u>	
Observed temperature in °F	0.830	0.831	0.832	0.833	0.834	0.835	0.836	0.837	0.838	0.839
			Corresp	onding	specific	gravitie	es at 60°	°/60° F		
30	0.818	0.819	0.820	0.821	0.822	0,823	0.824	0.825	0.826	0.827
32	.819	.820	.821	.822	.823	.824	.825	.826	.827	.828
34	.820	.821	.822	.823	.824	.825	.826	.827	.828	.829
30 33	.821 .822	.822 .823	.823 .824	.824 .825	.825 .826	.826 .827	.827 .828	.828 .829	.829 .830	.830
40	.8225	.8235	.8 45	.8255	.8265	.8275	.8285	.8295	.8305	.831
42	.8230	.8240	.8250	.8260	.8270	.8280	.8295	.8305	.8315	.832
41	.8240	.8250	.8260	.8270	.8280	.8290	.8300	.8310	.8320	.8330
4 i	.8245	.8255	.8265	.8275	.8285	.8295	.8305	.8315	.8325	.833
43	.8255	.8265	.8275	.8285	.8290	.8305	.8315	.8325	.8335	.8342
50	.8260	.8270	.8280	.8290	.8300	.8310	.8325	.8335	.8345	.8353
52	.8270	.8280	.8290	.8300	8310	.8320	.8330	.8340	.8350	.8360
54 53	.8280 .8285	.8290	.8300	.8310	.8320 .8325	.8330	.8340	.8350	.8360	.8370
58	.8295	.8295 .8305	.8305 .8315	.8315 .8325	.8335	.8335 .8345	.8345 .8355	.8355 .8365	.8335 .8375	.8370
60	.8300	.8310	.8320	.8330	.8340	.8350	.8360	.8370	.8380	.839
62	.8305	.8315	.83 5	.8335	.8345	-8355	.8365	.8375	.8385	.8390
64	.8315	.8325	.8335	.8345	.8355	.8305	.8375	.8385	.8395	.8400
63 68	.8320 .8330	.8330 .8340	.8340 .8350	.8350 .8360	.8\$00 .8370	.8370	.8380 .8390	.8390 .8400	.8400 .8410	.8410
70	.8340	.8350	.8360	.8370	.8380	.8390	.8400	.8410	.8420	.8430
72	.8345	.8355	.8365	.8375	.8385	.8395	.8405	.8415	.8425	.843
74	.8355	.8335	.8375	.8385	.8395	.8405	.8415	.84 5	.8435	.8445
76	.8360 .8370	.8370 .8380	.8380 .8390	.8390 .8400	.8400 .8410	.8410 .8420	.8420 .8430	.8430 .8440	.8140 .8450	.8450 .8460
30	.837	.838	.839	.840	.841	.842	.843	.841	.845	.846
82	.838	.839	.840	.841	.842	.813	.814	.845	.840	.847
84	.839	.840	.841	.842	.843	.844	.845	.846	.847	.848
86	.839 .840	.840 .841	.841 .842	.842 .843	.843 .844	.844 .845	.845 .846	.846 .817	.847 .848	.848 .849
90	.841	.842	.843	.844	.845	.846	.847	.848	.849	.850
92	.842	.843	.844	.845	.846	.847	.848	.849	.850	.851
94	.842	.843	.844	.845	.846	.847	.848	.849	.850	.851
96 98	.843 .844	.844 .845	.845 .846	.846 .847	.847 .848	.848 .849	.849 .850	.850 .851	.851 .852	.852 .853
	.814	.845	.846	.847	.848	.849	.850	.851		.853
02	.845	.840	.847	.848	.849	.840	.851	.851	.852 .853	.803
04	.846	.847	.848	.849	.850	.851	.852	.853	.854	.855
06	.847	.848	.849	.850	.851	.852	.853	.854	.855	.856
08	.847	.848	.849	.850	.851	.852	.853	.854	.855	.853
10	.848	.849	.850	.851	.852	.853	.854	.855	.856	.857
12	.849	.850	.851	.852	.853	.854	.855	.856	.857	.858
14	.850	.851	.852	.853	.854	.855	.855	.856	.857	.858
16 18	.850 .851	.851 .852	.852 .853	.853 .854	.854 .855	.855 .856	.856 .857	.857 .858	.858 .859	.859 .830
20	.852	.853	.854	.855	.856	.857	.858	.859	.860	.861

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.840	0.841	0.842	0.843	0.844	0.845	0.846	0.847	0.848	0.849
			Corresp	onding	specific	gravitie	s at 60°	°/60° F		
30	0.828	0.829	0.830	0.831	0.832	0.833	0,835	0.836	0.837	0.838
32	.829	.830	.831	.832	.833	.834	.835	.836	.837	.838
34	.830 .831	.831	-832	.833	.834	.835	.836	.837	.838	.839
36 38	.832	.832 .833	.833 .834	.834 .835	.835 .836	.836 .837	.837 .838	.838 .839	.839 .840	.840 .841
40	.8325	.8335	.8345	.8355	.8365	.8375	.8385	.8395	.8405	.8415
42	.8335	.8345	.8355	.8365	.8375	.8385	.8395	.8405	.8415	.8425
44 46	.8340	.8350 .8355	.8360 .8305	.8370 .8375	.8380 .8385	.8390 .8395	.8400 .8410	.8410 .8420	.8420 .8430	.8430
48	.8355	.8365	.8375	.8385	8395	.8405	.8415	.8425	.8435	.8445
50	.8365	.8375	.8385	.8395	8405	.8415	.8425	.8435	.8445	.8455
52 54	.8370 .8380	.8380 .8390	.8390 .8400	.8400 .8410	.8410	.8420 .8430	.8430 .8440	.8440 .8450	.8450 .8460	.8460 .8470
56	.8385	.8395	.8405	.8415	8425	.8435	.8145	.8455	.8465	.8475
58	.8395	.8405	.8415	.8425	8435	.8445	.8455	.8465	.8475	.8485
60	.8400	.8410	.8420	.8430	:440	.8450	.8460	.8470	.8480	.8490
62	.8405	.8415	.8425	.8435	.8145	.8455	.8465	.8475	.8485	.8495
64 66	.8415 .8420	.8425 .8430	.8435 .8440	.8445 .8450	.8455 .8460	.8465 .8470	.8475 .8480	.8485 .8490	.8495 .8500	.8505
68	.8430	.8440	.8450	.8460	.8470	.8480	.8490	.8490	.8510	.8510 .8520
70	.8440	.8450	.8460	.8470	.8480	.8490	.8500	.8510	.8520	.8530
72	.8445	.8455	.8465	.8475	.8485	.8495	.8505	.8515	.8525	.8535
74	.8455 .8460	.8465 .8470	.8475 .8480	.8485 .8490	.8495 .8500	.8505 .8510	.8510 .8520	.8520 .8530	.8530 .8540	.8540
78	.8470	.8480	.8490	.8500	.8510	.8520	.8525	.8535	.8545	.8555
80	.847	.848	.849	.850	.851	.852	.853	.854	.855	.856
82	.848	.849 .850	.850 .851	.851 .852	.852 .853	.853 .854	.854 .855	.855 .856	.856 .857	.857 .858
84 83	.849 .849	.850	.851	.852	.853	.854	.855	.856	.857	.858
88	.850	.851	.852	.853	.854	.855	.856	.857	.858	.859
90	.851	.852	.853	.854	.855	.856	.857	.858	.859	.860
92 94	.852 .852	.853	.854 .854	.855 .855	.856 .856	.857 .857	.857 .858	.858 .859	.859 .860	.860 .861
96	.853	.854	.855	.856	.857	.858	.859	.860	.861	.862
98	.854	.855	.856	.857	.858	.859	.860	.861	.862	.863
100	.854	.855	.856	.857	.858	.859	.860	.861	.862	.863
103 104	.855 .856	.856 .857	.857 .858	.858 .859	.859 .860	.860 .861	.861 .862	.862 .863	.863 .864	.864 .865
104	.857	.858	.859	.860	.861	.862	.862	.863	.864	.865
108	.857	.858	.859	.860	.861	.862	.863	.864	.865	.866
110	.858	.859	.860	.861	.862	.863	.864	.865	.866	.867
112	.859	.860	.861	.862	.863	.864	.865	.866	.867	.868
114	.859 .860	.860 .861	.861 .862	.862 .863	.863 .864	.864 .865	.865	.866 .867	.867 .868	.868
116 118	.800	.861	.802	.803	.804	.800	.800	.868	.808	.809
120	.862	.863	.864	.865	.866	.867	.868	.869	.870	.871

				Observ	red spec	ific gra	vities			
Observed temperature in °F	0.850	0.851	0.852	0.853	0.854	0.855	0.856	0.857	0.858	0.859
			Corresp	onding	specific	gravitie	es at 60°	°/60° F		
30	0.839	0.840	0.841	0.842	0.843	0.844	0.845	0.846	0.847	0.848
32	.839 .840	.840	.841 .842	.842	.843	.844 .845	.845 .846	-846	.847	.848
34 36	.840	.842	.843	.844	.845	.840	.840	.847 .848	.848	.849
38	.842	.843	.844	.845	.846	.847	.848	.849	.849 .850	.850 .851
40	.8425	.8435	.8445	.8455	.8465	.8475	.8485	.8495	.8505	.8515
42	.8435	.8445	.8455	.8465	.8475	.8485	.8495	.8505	.8515	.8525
44	.8440	.8450	.8460	.8470	.8480	.8490	.8500	.8510	.8520	.8530
46 48	.8450 .8455	.8460 .8465	.8470 .8475	.8480 .8485	.8490 .8495	.8500 .8505	.8510 .8515	.8520 .8525	.8530 .8535	.8540 .8545
50	.8465	.8475	.8485	.8495	.8505	.8515	.8525	.8535	.8545	.8555
52	.8470	.8480	.8490	.8500	.8510	.8520	.8530	.8540	.8550	.8560
54	.8480	.8490	.8500	.8510	.8520	.8530	.8540	.8550	.8560	.8570
56 58	.8485 .8495	.8495 .8505	.8505 .8515	.8515 .8525	.8525 .8535	.8535 .8545	.8545 .8555	.8555 .8565	.8505 .8575	.8575 .8585
60,	.8500	.8510	.8520	.8530	.8540	.8550	.8560	.8570	.8580	.8590
62	.8505	.8515	.8525	.8535	.8545	.8555	.8565	.8575	.8585	.8595
64	.8515	.8525	.8535	.8545	.8555	.8565	.8575	.8585	.8595	.8605
66 68	.8520 .8530	.8530 .8540	.8540 .8550	.8550 .8560	.8560 .8570	.8570 .8580	.8580 .8590	.8590 .8600	.8600 .8610	.8610 .8620
70	.8540	.8550	.8560	.8570	.8580	.8590	.8595	.8605	.8615	.8625
72	.8545	.8555	.8565	.8575	.8585	.8595	.8605	.8615	.8625	.8635
74	.8550	.8560	.8570	.8580	.8590	.8600	.8610	.8620	.8630	.8640
76 78	.8560 .8565	.8570 .8575	.8580 .8585	.8590 .8595	.8600 .8605	.8610 .8615	.8620 .8625	.8630 .8635	.8640 .8645	.8650 .8655
80	.857	.858	.859	.860	.861 -	.862	.863	.864	.865	.866
82	.858	.859	.860	.861	.862	.863	.864	.865	.866	.867
84 86	.859 .859	.800 .860	.861 .861	.862	.863 .863	.864 .864	.864	.865	.866	.867
88	.860	.861	.862	.862 .863	.864	.865	.865 .866	.866 .867	.867 .868	.868 .869
90	.861	.862	.863	.864	.805	.866	.867	.868	.869	.870
92	.861	.862	.863	.864	.865	.866	.867	.868	.869	.870
94 96	.862 .863	.863 .864	.864 .865	.865	.866 .867	.867	-868	.869	.870	.871
98	.864	.865	.866	.866 .867	.868	.868 .869	.869 .869	.870 .870	.871 .871	.872 .872
00	.864	.865	.866	.867	.868	.869	.870	.871	.872	.873
02	.865	.866	.867	.868	.869	.870	.871	.872	.873	.874
04	.866	.867	.868	.869	.870	.871	.872	.873	.874	.875
08	.866 .867	.867 .868	.868 .869	.869 .870	.870 .871	.871 .872	.872 .873	.873 .874	.874 .875	.875 .876
10	.868	.869	.870	.871	.872	.873	.874	.875	.876	.877
12	.869	.870	.871	.872	.873	.874	.874	.875	.876	.877
14	.869	.870	.871	.872	.873	.874	.875	.876	.877	.878
16	.870	.871	.872	.873	.874	.875	.876	.877	.878	.879
18	.871	.872	.873	.874	.875	.876	.877	.878	.879	.880
20	.872	.873	.874	.875	.876	.877	.877	.878	.879	.880

				Observ	ed spec	ific gra	vities			
Observed temperature in °F	0.860	0.861	0.862	0.863	0.864	0.865	0.866	0.867	0.868	0.869
			Corresp	onding	specific	gravitie	es at 60°	/60° F		
30	0.849	0,850	0.851	0.852	0.853	0.854	0.855	0.856	0.857	0.858
32	.849	.850	.851	.852	.853	.854	.856	.857	.858	.859
34	.850	.851	.852	.853	.854	.855	.856	.857	.858	.859
36 38	.851 .852	.852 .853	.853 .854	.854 .855	.855 .856	.856 .857	.857 .858	.858	.859 .860	.860 .861
40	.8525	.8535	.8545	.8555	.8565	.8575	.8585	.8595	.8605	.8615
42	.8535	.8545	.8555	.8565	.8575	.8585	.8595	.8605	.8615	.8625
44	.8540	.8550	.8560	.8570	.8580	.8590	.8600	.8610	.8620	.8630
46	.8550	.8500	.8570	.8580	.8590	.8600	.8610	.8620	.8630	.8640
48	.8555	.8565	.8575	.8585	.8595	.8605	.8615	.8625	.8635	.8645
50	.8565	.8575	.8585	.8595	.8605	.8615	.8625	.8635	.8645	.8655
52	.8570	.8580	.8590	.8600	.8610	.8620	.8630	.8640	.8650	.8660
54	.8580	.8590	.8600	.8610	.8620	.8630	.8640	.8650	.86.0	.8670
56 58	.8585 .8595	.8595 .8605	.8605 .8615	.8615 .8625	.8625 .8635	.8635 .8645	.8645 .8655	.8655 .8665	.8665 .8675	.8675 .8685
60	.8600	.8610	.8620	.8630	.8340	.8650	.8660	.8670	.8680	.8690
62	.8605	.8615	.8625	.8635	.8645	.8655	.8665	.8675	.8685	.8695
64	.8615	.8625	.8635	.8645	.8055	.8665	.8675	.8685	.8695	.8705
6	.8620	.8630	.8640	.8650	.8660	.8670	.8680	.8690	.8700	.8710
68	.8630	.8640	.8650	.8660	.8670	.8680	.8690	.8700	.8710	.8720
70	.8635	.8645	.8655	.8665	.8675	.8685	.8695	.8705	.8715	.8725
72	.8645	.8655	.8665	.8675	.8685	.8695	.8705	.8715	.8725	.8735
74	.8650 .8600	.8660	.8670	.8580	.8690	.8700 .8710	.8710	.8720	.8730 .8740	.8740
76 78	.8000	.8670 .8675	.8685	.8690 .8695	.8700 .8705	.8715	.8720 .8725	.8730 .8735	.8745	.8750 .8755
80	.867	.868	.869	.870	.871	.872	.873	.874	.875	.876
82	.868	.869	.870	.871	.872	.873	.874	.875	.876	.877
84	.868	.869	.870	.871	.872	.873	.874	.875	.876	.877
801 88	.869 .870	.870 .871	.871 .872	.872 .873	.873 .874	.874 .875	.875 .876	.876 .877	.877 .878	.878 .879
90	.871	.872	.873	.874	.875	.876	.877	.878	.879	.880
92	.871	.872	.873	.874	.875	.876	.877	.878	.879	.880
94	.872	.873	.874	.875	.876	.877	.878	.879	.880	.881
96	.873	.874	.875	.876	.877	.878	.879	.880	.881	.882
98	.873	.874	.875	.876	.877	.878	.879	.880	.881	.882
100	.874	.875	.876	.877	.878	.879	.880	.881	.882	.883
102	.875	.876	.877	.878	.879	.880	.881	.882	.883	.884
104	.876	.877	.878	.879	.880	.881	.882	.883	.884	.885
103	.876 .877	.877	.878 .879	.879 .880	.880 .881	.881 .882	.882 .883	.883 .884	.884 .885	.885 .886
110	.878	.879	.880	.881	.882	.883	.884	.885	.886	.887
112	.878	.879	.880	.881	.882	.883	.884	.885	.886	.887
114	.879	.890	.881	.882	.883	.884	.885	.886	.887	.888
116	.880	.881	.882	.883	.884	.885	.886	.887	.888	.889
118	.881	.882	.883	.884	.885	.886	.886	.887	.888	.889
120	.881	.882	.883	.884	.885	.886	.887	.888	.889	.890

				Observ	red spec	ific gra	vities			
Observed temperature in °F	0.870	0.871	0.872	0.873	0.874	0.875	0.876	0.877	0.878	0.879
			Corresp	onding	specific	gravitie	s at 60°	/60° F		
30	0.859	0.860	0.861	0.862	0.863	0.864	0.865	0.866	0.867	0.868
32 34	.860 .860	.861 .861	.862 .862	.863 .863	.864 .864	.865	.866 .866	.867 .867	.868	.869 .869
36	.861	.862	.863	.864	.865	.866	.867	.868	.869	.870
38	.862	.863	.864	.865	.806	.867	.868	.869	.870	.871
40	.8625	.8635	.8645	.8655	.8665	.8675	.8690	.8700	.8710	.8720
42	.8635	.8645	.8655	.8665	.8675	.8685	.8695	.8705	.8715	.8725
44	.8640	.8650	.8660	.8670	.8680	.8690	.8700	.8710	.8720	.8730
46	.8650	.8660	.8670 .8675	.8680	.8690	.8700 .8705	.8710	.8720	.8730	.8740
48	.8655	.8665	.0010	.0000	.8099	.8109	.8715	.8725	.8735	.8745
50	.8665	.8675	.8685	.8695	.8705	.8715	.8725	.8735	.8745	.8755
52	.8670	.8680	.8690	.8700	.8710	.8720	.8730	.8740	.8750	.8760
. 54	.8680 .8685	.8690	.8700 .8705	.8710	.8720	.8730	.8740	.8750	.8760	.8770
53 58	.8089	.8695 .8705	.8715	.8725	.8725 .8735	.8735 .8745	.8745 .8755	.8755 .8765	.8765 .8775	.8775 .8785
60	.8700	.8710	.8720	.8730	.8740	.8750	.8760	.8770	.8780	.8790
62	.8705	.8715	.8725	.8735	.8745	.8755	.8765	.8775	.8785	.8795
64	.8715	.8725	.8735	.8745	.8755	.8765	.8775	.8785	.8795	.8805
6.1	.8720	.8730	.8740	.8750	.8760	.8770	.8780	.8790	.8800	.8810
68	.8730	.8740	.8750	.8760	.8770	.8780	.8790	.8800	.8810	.8820
70	.8735	.8745	.8755	.8765	.8775	.8785	.8795	.8805	.8815	.8825
72	.8745	.8755	.8765	.8775	.8785	.8795	.8805	.8815	.8825	.8835
74	.8750	.8760	.8770	.8780	.8790	.8800	.8810	.8820	.8830	.8840
76 78	.8760 .8765	.8770 .8775	.8780 .8785	.8790 .8795	.8800	.8810 .8815	.8820 .8825	.8830 .8835	.8840 .8845	.8850 .8855
80	.877	.878	.879	.880	.881	.882	.883	.884	.885	.886
82	.878	.879	.880	.881	.882	.883	.884	.885	.886	.887
84	.878	.879	.880	.881	.882	.883	.884	.885	.886	.887
86	.879	.880	.881	.882	.883	.884	.885	.886	.887	.888
88	.880	.881	.882	.883	.884	.885	.886	.887	.888	.889
90	.881	.882	.883	.884	.885	.886	.887	.888	.889	.890
92	.881	.882	.883	.884	.885	.886	.887	.888	.889	.890
94 96	.882	.883 .884	.884 .885	.885 .886	.886 .887	.887 .888	.888 .889	.889 .890	.890 .891	.891 .892
98	.883	.884	.885	.886	.887	.888	.889	.890	.891	.892
100	.884	.885	.886	.887	.888	.889	.890	.891	.892	.893
102	.885	.886	.887	.888	.889	.890	.891	.892	.893	.894
104	.886	.887	.888	.889	.890	.891	.891	.892	.893	.894
106	.886	.887	.888	.889	.890	.891	.892	.893	.894	.895
108	.887	.888	.889	.890	.891	.892	.893	.894	.895	.896
110	.888	.889	.890	.891	.892	.893	.894	.895	.896	.897
112	.888	.889	.890	.891	.892	.893	.894	.895	.896	.897
114	.889	.890	.891	.892	.803	.894	.895	.896	.897	.898
116	.890	.891	.892	.893	.894	.895	.893	.897	.898	.899
118	.890	.891	.892	.893	.894	.895	.896	.897	.898	.899
120	.891	.892	.893	.894	.895	.896	.897	.898	.899	.900

410

				Observ	ed spec	ific gra-	vities			
Observed temperature in °F	0.880	0.881	0.882	0.883	0.884	0.885	0.886	0.887	0.888	0.889
			Corresp	onding	specific	gravitie	ns at 60°	/60° F		
30	0.869	0.870	0.871	0.872	0.873	0.874	0.875	0.876	0.877	0.878
32	.870	.871	.872	.873	.874	.875	.876	.877	.878	.879
34	.870	.871	.872	.873	.874	.875	.876	.877	.878	.879
3 3 38	.871 .872	.872 .873	.873 .874	.874 .875	.875 .876	.876 .877	.877 .878	.878 .879	.879 .880	.880 .881
40	.8730	.8740	.8750	.8760	.8770	.8780	.8790	.8800	.8810	.8820
42	.8735	.8745	.8755	.8765	.8775	.8785	.8795	.8805	.8815	.8825
41	.8740	.8750	.8760	.8770	.8780	.8790	.8800	.8810	.8820	.8830
4 6 4 8	.8750 .8755	.8760 .8765	.8770 .8775	.8780 .8785	.8790 .8795	.8800 .8805	.8810 .8815	.8820 .8825	.8830 .8835	.8840 .8845
				.8795					1	
50 52	.8765 .8770	.8775 .8780	.8785 .8790	.8795	.8805 .8810	.8815	.8825 .8830	.8835	.8845 .8850	.8855
54	.8780	.8790	.8800	.8810	.8820	.8830	.8840	.8850	.8800	.8870
56	.8785	.8795	.8805	.8815	.8825	.8835	.8845	.8855	.8865	,8875
58	.8795	.8805	.8815	.8825	.8835	.8845	.8855	.8865	.8875	.8885
(0	.8800	.8810	.8820	.8830	.8840	.8850	.8860	.8870	.8880	.8890
62	.8805	.8815	.8825	.8835	.8845	.8855	.8835	.8875	.8885	.8895
64	.8815	.8825 .8830	.8835	.8845	.8855	.8865	.8875	.8885	.8895	.8905
66 68	.8820 .8830	.8840	.8840 .8850	.8850 .8860	.8870	.8870 .8890	.8890	.8890 .8900	.8900 .8910	.8910 .8920
70	.8835	.8845	.8855	.8865	.8875	.8885	.8895	.8905	.8915	.8925
72	.8845	.8855	.8865	.8875	.8885	.8895	.8900	.8910	.8920	.8930
74	.8850	.8800	.8870	.8880	.8890	.8900	.8910	.8020	.8930	.8940
76	.8860	.8870 .8875	.8880 .8885	.8890 .8895	.8900	.8910 .8915	.8915 .8925	.8925 .8935	.8935 .8945	.8945
80	.887	.888	.889	.890	.891	.892	.893	.894	.895	.896
82	.888	.889	.890	.891	.892	.893	.894	.895	.896	.897
84	.888	.889	.890	.891	.892	.893	.894	.895	.896	.897
83 88	.889 .890	.890 .891	.891 .892	.892 .893	.893 .894	.894 .895	.895 .896	.896 .897	.897 .898	-898 .899
					12.					
90 92	.891 .891	.892 .892	.893 .893	.894 .894	.895 .895	.896 .896	.896 .897	.897 .898	.898 .899	.899
94	.892	.893	.894	.895	.896	.897	.898	.899	.900	.901
93	.893	.804	.895	.896	.897	.898	.899	.900	.901	.902
98	.893	.894	.895	.896	.897	.898	.899	.900	.901	.902
	.804	.895	.896	.897	.898	.809	.900	.901	.902	.903
102	.895	.896	.897	.898	.899	.900	.901	.902	.903	.904
06	.895 .896	.896 .897	.897 .898	.898 .899	.899 .900	.900 .901	.901 .902	.902	.903 .904	.904 .905
.08	.897	.898	.899	.900	.900	.902	.903	.904	.905	.906
10	.898	.899	.900	.901	.902	.903	.903	.904	.905	.906
12	.898	.809	.900	.901	.902	.903	.904	.905	.906	.907
14	.899	.900	.901	.902	.903	.904	.905	.906	.907	.908
16	.900	.901	.902	.903	.904	.905	.905	.906	.907	.908
18	.900	.901	.902	.903	.904	.905	.906	.907	.908	.909
20	.901	.902	.903	.904	.905	.906	.907	.908	.909	.910

		Observed specific gravities											
Observed temperature in °F	0.890	0.891	0.892	0.893	0.894	0.895	0.896	ð.897	0.898	0.899			
-581 -		•	Corresp	onding	specific	gravitie	as at 60°	/60° F					
30	0.879	0.880	0.881	0.882	0.883	0.884	0.885	0.886	0.887	0.888			
32	.880	.881	.882	.883	.884	.885	.886	.887	.888	.889			
34	.880	.881	.882	.883	.884	.885	.886	.887	.888	.889			
36 38	.881 .882	.882 .883	.883 .884	.884 .885	.885 .886	.886 .887	.887 .888	.889 .889	.889 .890	.890 .891			
40	.8830	.8840	.8850	.8860	.8370	.8880	.8890	.8900	.8910	.8920			
42	.8840	.8850	.8860	.8870	.8880	.8885	.8895	.8905	.8915	.8925			
44	.8840	.8850	.8860	.8870	.8880	.8890	.8900	.8910	.8920	.8930			
46 48	.8850 .8855	.8860 .8865	.8870 .8875	.8880 .8885	.8890 .8895	.8900 .8905	.8910 .8915	.8920 .8925	.8930 .8935	.8940 .8945			
50	.8865	.8875	.8885	.8895	.8905	.8915	.8925	.8935	.8945	,8955			
52	.8870	.8880	.8890	.8900	.8910	.8920	.8930	.8940	.8950	.8960			
54 53	.8880 .8885	.8890 .8895	.8900 .8905	.8910 .8915	.8920 .8925	.8930 .8935	.8940 .8945	.8950 .8955	.8960 .8965	.8970 .8975			
58	.8895	.8905	.8915	.8925	.8935	.8945	.8955	.8965	.8975	.8985			
60	.8900	.8910	.8920	.8930	.8940	.8950	.8960	.8970	.8980	.8990			
62 64	.8905 .8915	.8915 .8925	.8925 .8935	.8935 .8945	.8945	.8955	.8965 .8975	.8975 .8985	.8985 .8995	.8995			
66	.8919	.8925	.8930	.8945	.8960	.8900	.8975	.8980	.9000	.9010			
68	.8930	.8940	.8950	.8960	.8970	.8980	.8990	.9000	.9010	.9020			
70	.8935	.8945	.8955	.8965	.8975	.8985	.8995	.9005	.9015	.9025			
72 74	.8940 .8950	.8950 .8960	.8960	.8970 .8980	.8980 .8990	.8990	.9000	.9010	.9020 .9030	.9030			
76	.8955	.8965	.8975	.8985	.8995	.9005	.9015	.9025	.9035	.9045			
78	.8965	.8975	.8985	.8995	.9005	.9015	.9025	.9035	.9045	.9055			
80	.897	.898	.899	.900	.901	.902	.903	.904	.905 .905	.906			
82	.898 .898	.899 .899	.900	.901 .901	.902	.903 .903	.903 .904	.904 .905	.905	.906 .907			
83	.899	.900	.901	.902	.903	.904	.905	.906	.907	.908			
88	.900	.901	.902	.903	.904	.905	.906	.907	.908	.909			
90	.900	.901	.902	.903	.904	.905	.906	.907	.908	.909			
92 94	.901 .902	.902 .903	.903 .904	.904 .905	.905 .906	.906 .907	.907 .908	.908 .909	.909 .910	.910 .911			
96	.903	.904	.905	.906	.907	.903	.909	.910	.911	.912			
98	.903	.904	.905	.906	.907	.908	.909	.910	.911	.912			
100	.904	.905	.906	.907	.908	.909	.910	.911	.912	.913			
102	.905	.906	.907	.908	.909	.910	.911	.912	.913	.914			
104	.905	.906	.907	.908	.909	.910	.911	.912	.913	.914			
106	.906	.907	.908	.909	.910	.911	.912	.913	.914	.915			
108	.907	.908	.909	.910	.911	.912	.913	.914	.915	.916			
110	.907	.908	.909	.910	.911	.912	.913	.914	.915	.916			
112 114	.908	.909	.910	.911	.912	.913	.914	.915	.916	.917			
116	.909 .909	.910 .910	.911 .911	.912 .912	.913 .913	.914 .914	.915 .915	.916	.917 .917	.918 .918			
118	.910	.911	.912	.913	.914	.915	.916	.917	.918	.919			
120	.911	.912	.913	.914	.915	.916	.917	.918	.919	.920			

412

Observed	-									
temperature in °F	0.900	0.901	0.902	0.903	0.904	0.905	0.906	0.907	0.908	0.909
			Corresp	onding	specific	gravitie	es at 60°	/60° F		-
30	0.889	0.890	0.891	0.892	0.893	0.894	0.895	0.896	0.897	0.898
32	.890	.891	.892	.893	.894	.895	.896	.897	.898	.899
34	.890	.891	.892	.893	.894	.895	.896	.897	.898	.899
36 38	.891 .892	.892 .893	.893 .894	.894 .895	.895 .896	.896 .897	.897 .898	.898 .899	.899 .900	.900 .901
40	.8930	.8940	.8950	.8960	.8970	.8980	.8990	.9000	.9010	.9020
42	.8935	.8945	.8955	.8965	.8975	.8985	.8995	.9005	.9015	.902
44	.8940	.8950	.8960	.8970	.8.80	.8990	.9005	.9015	.9025	.903
46 48	.8950 .8955	.8960 .8965	.8970 .8975	.8980 .8985	.8990 .8995	.9000 .9005	.9010 .9015	.9020 .9025	.9030 .9035	.9040
50	.8965	.8975	.8985	.8995	.9005	.9015	.9025	.9035	.9045	.9055
52	.8970	.8980	.8990	.9000	.9010	.9020	.9030	.9040	.9050	.9060
54	.8080	.8990	.9000	.9010	.9020	.9030	.9040	.9050	.9060	.9070
56	.8985	.8995	.9005	.9015	.9025 .9035	.9035 .9045	.9045	.9055	.9065	.907
58	.8995	.9005	.9015	.9025			.9055	.9065	.9075	.908
60	.9000	.9010	.9020	.9030	.9040	.9050	.9060	.9070	.9080	.9090
62	.9005	.9015	.9025	.9035	.9045	.9055	.9065	.9075	.9085	.9098
(·4	.9015 .9020	.9025 .9030	.9035 .9040	.9045 .9050	.9055 .9030	.9065	.9075 .9080	.9085	.9095 .9100	.9108
68	.9030	.9040	.9050	.9060	.9070	.9080	.9090	.9100	.9110	.9120
70	.9035	.9045	.9055	.9065	.9075	.9085	.9095	.9105	.9115	.912
72	.9040	.9050	.9060	.9070	.9080	.9090	.9100	.9110	.9120	.9130
74	.9050	.9060	.9070	.9080	.9000	.9100	.9110	.9120	.9130	.9140
76 78	.9055 .9065	.9065 .9075	.9075 .9085	.9085 .9095	.9095 .9105	.9105 .9115	.9115 .9125	.9125 .9135	.9135 .9145	.9143
80	.907	.908	.909	.910	.911	.912	.913	.914	.915	.916
82	.907	.908	.909	.910	.911	.912	.913	.914	.915	.916
84	.908	.909	.910	.911	.912	.913	.914	.915	.916	.917
86 88	.909 .910	.910 .911	.911	.912	.913 .914	.914 .915	.915 .916	.916 .917	.917 .918	.918
90	.910	.911	.912	.913	.914	.915	.916	.917	.918	.919
92	.911	.912	.913	.914	.915	.916	.917	.918	.919	.920
94	.912	.913	.914	.915	.916	.917	.918	.919	.920	.921
96	.913	.914	.915	.916	.917	.918	.918	.919	.920	.921
98	.913	.914	.915	.916	.917	.918	.919	.920	.921	.922
.00	.914	.915	.916	.917	.918	.919	.920	.921	.922	.923
02	.915	.916	.917	.918	.919	.920	.921	.922	.923	.924
.04	.915 .916	.916 .917	.917 .918	.918	.919 .920	.920 .921	.921 .922	.922 .923	.923 .924	.924 .925
08	.917	.918	.918	.919 .920	.920	.921	.922	.924	.925	.926
10	.917	.918	.919	.920	.921	.922	.923	.924	.925	.926
12	.918	.919	.920	.921	.922	.923	.924	.925	.926	.927
14	.919	.920	.921	.922	.923	.924	.925	.926	.927	.928
16	.919 .920	.920 .921	.921 .922	.922 .923	.923 .924	.924 .925	.925 .926	.926 .927	.927 .928	.928 .929
					1.000			1.4		
20	.921	.922	.923	.924	.925	.926	.927	.928	.929	.930

				Observ	ed spec	ific grav	vities			
Observed temperature in °F	0.910	0.911	0.912	0.913	0.914	0.915	0.916	0.917	0.918	0.919
			Corresp	onding	specific	gravitie	s at 60°	/60° F		
30	0.899	0.900	0.901	0.902	0.903	0.904	0.905	0.906	0.907	0.908
32	.900	.901	.902	.903	.904	.905	.906	.907	.908	.909
34	.900	.901	.902	.903	.904	.905	.906 .907	.907	.908	.909
36 38	.901 .902	.902	.903 .904	.904	.906	.900	.908	.908 .909	.909	.910 .911
40	.9030	.9010	.9050	.9060	.9070	.9080	.9090	.9100	.9110	.9120
42	.9035	.9045	.9055	.9065	.9075	.9085	.9095	· .9105	.9115	.9125
44	.9045	.9055	.9065	.9075	.9085	.9095	.9105	.9115	.9125	.9135
46 48	.9050 .9055	.9060 .9065	.9070 .9075	.9080 .9085	.9090 .9095	.9100 .9105	.9110 .9115	.9120 .9125	.9130 .9135	.9140 .9145
50	.9065	.9075	.9085	.9095	.9105	.9115	.9125	.9135	.9145	.9155
52	.9070	.9080	.9090	.9100	.9110	.9120	.9130	.9140	.9150	.9100
54	.9080	.9090	.9100	.9110	.9120	.9130	.9140	.9150	.9160	.9170
53 58	.9085 .9095	.9095 .9105	.9105 .9115	.9115 .9125	.9125 .9135	.9135 .9145	.9145 .9155	.9155 .9165	.9165 .9175	.9175 .9185
60:	.9100	.9110	.9120	.9130	.9140	.9150	.9160	.9170	.9180	.9190
62	.9105	.9115	.9125	.9135	.9145	.9155	.9165	.9175	.9185	.9195
64	.9115	.9125	.9135	.9145	.9155	.9165	.9175	.9185	.9195	.9205
66 68	.9120 .9130	.9130 .9140	.9140 .9150	.9150 .9160	.9160 .9170	.9170 .9180	.9180 .9190	.9190 .9200	.9200 .9210	.9210 .9220
70	.9135	.9145	.9155	.9165	.9175	.9185	.9195	.9205	.9215	.9225
72	.9140	.9150	.9160	.9170	.9180	.9190	.9200	.9210	.9220	.9230
74	.9150	.9160	.9170	.9180	.9190	.9200	.9210	.9220	.9230	.9240
76 78	.9155 .9165	.9165 .9175	.9175 .9185	.9185 .9195	.9195 .9205	.9205 .9215	.9215 .9225	.9225 .9235	.9235 .9245	.9245 .0255
80	.917	.918	.919	.920	.921	.922	.923	.924	.925	.926
82	.917	.918	.919	.920	.921	.922	.923	.924	.925	.926
84	.918	.919	.920	.921	.922	.923	.924	.925	.926	.927
86 88	.919 .920	.920 .921	.921 .922	.922 .923	.923 .924	.924 .925	.925 .926	.926 .927	.927 .928	.928 .929
90	.920	.921	.922	.923	.924	.925	.926	.927	.928	.929
92	.921	.922	.923	.924	.925	.926	.927	.928	.929	.930
94	.922	.923	.924	.925	.926	.927	.928	.929	.930	.931
96 98	.922 .923	.923 .924	.924 .925	.925 .926	.926 .927	.927 .928	.928 .929	.929 .930	.930 .931	.931 .932
100	.924	.925	.926	.927	.928	.929	.930	.931	.932	.933
102	.925	.926	.927	.928	.929	.930	.931	.932	.933	.934
104	.925	.926	.927	.978	.929	.930	.931	.932	.933	.934
106 108	.926 .927	.927 .928	.928 .929	.929 .930	.930 .931	.931 .932	.932 .933	.933 .934	.934 .935	.935 .933
110	.927	.928	.929	.930	.931	.932	.933	.934	.935	.936
112	.928	.929	.930	.931	.932	.933	.934	.935	.935	.937
114	.929	.930	.931	.932	.933	.934	.935	.936	.937	.938
116	.929	.930	.931	.932	.933	.934	.935	.936	.937	.938
118	.930	.931	.932	.933	.934	.935	.936	.937	.938	.939
120	.931	.932	.933	.934	.935	.936	.937	.938	.939	.940

414

			and the	Observ	ed speci	fic grav	rities		1	
Observed temperature in °F	0.920	0.921	0.922	0.923	0.924	0.925	0.926	0.927	0.928	0.929
			Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30	0.909	0.910	0.911	0.912	0.913	0.914	0.915	0.916	0.917	0.918
32	.910	.911	.912	.913	.914	.915	.916	.917	.918	.919
34	.910	.911	.912	.913	.914	.915	.916	.917	.918	.919
36 38	.911 .912	.912 .913	.913 .914	.914 .915	.915 .916	.916 .917	.917 .918	.918 .919	.919 .920	.920 .921
40	.9130	.9140	.9150	.9160	.9170	.9180	.9190	.9200	.9210	.9220
42	.9135	.9145	.9155	.9165	.9175	.9185	.9195	.9205	.9215	.9225
44 49	.9145 .9150	.9155 .9160	.9165 .9170	.9175 .9180	.9185 .9190	.9195 .9200	.9205 .9210	.9215 .9220	.9225 .9230	.9235
48	.9155	.9165	.9175	.9185	.9190	.9205	.9210	.9225	.9235	.9240
50	.9165	.9175	.9185	.9195	.9205	.9215	.9225	.9235	.9245	.9255
52 54	.9170 .9180	.9180 .9190	.9190 .9200	.9200 .9210	.9210	.9220	.9230 .9240	.9240 .9250	.9250 .9260	.9260
56	.9185	.9195	.9205	.9215	.9225	.9235	.9245	.9255	.9265	.9275
58	.9195	.9205	.9215	.9225	.9235	.9245	.9255	.9265	.9275	.9285
60	.9200	.9210	.9220	.9230	.9240	.9250	.9260	.9270	.9280	.9290
62 64	.9205 .9215	.9215	.9225 .9235	.9235	.9245	.9255 .9265	.9265	.9275 .9285	.9285 .9295	.9295
66	.9220	92:0	.9240	.9250	.9260	.9270	.9280	.9290	.9300	.9310
63	.9230	.9240	.9250	.9260	.9270	.9280	.9290	.9300	.9310	.9320
70	.9235	.9245	.9255	.9265	.9275	.9285	.9295	.9305	.9315	.9325
72	.9240	.9250	.9260	.9270	.9280	.9290	.9300	.9310	.9320	.9330
74	.9250 .9255	.9260 .9265	.9270 .9275	.9280 .9285	.9290 .9295	.9300	.9310 .9315	.9320 .9325	.9330 .9335	.9340
78	.9265	.9275	.9285	.9295	.9305	.9315	.9325	.9335	.9345	.9355
80	.927	.928	.929	.930	.931	.932	.933	.934	.935	.936
82 84	.927 .928	.928 .929	.929 .930	.930 .931	.931 .932	.932 .933	.933 .934	.934 .935	.935 .936	.936
86	.929	.929	.931	.932	.033	.934	.935	.936	.937	.938
88	.930	.931	.932	.933	.934	.935	.936	.937	.938	.939
90 92	.930 .931	.931	.932	.933	.904 .935	.935 .936	.936 .937	.937 .938	.938 .939	.939 .940
94	.931	.932	.933 .934	.934	.933	.930	.937	.939	.939	.941
96	.932	.933	.934	.935	.936	.937	.938	.939	.940	.941
98	.933	.934	.935	.936	.937	.938	.939	.940	.941	.942
100	.934	.935	.935	.937	.938	.939	.940	.941	.942	.943
02	.935 .935	.936 .936	.937 .937	.938 .938	.939 .939	.940 .940	.940 .941	.941 .942	.942 .943	.943 .944
06	.936	.930	.937	.938	.939	.940	.941	.943	.944	.945
.08	.937	.938	.939	.940	.941	.942	.943	.944	.945	.946
10	.937	.938	.939	.940	.941	.942	.943	.944	.945	.946
12	.938 .939	.939 .940	.940 .941	.941 .942	.942 .943	.943 .944	.944 9.45	.945 9.46	.946 .947	.947 .948
16	.939	.940	.941	.942	.943	.944	9.45	9.46	.947	.948
18	.940	.941	.942	.943	.944	.945	.946	.947	.948	.949
	.941	.942	.943	.944	.945	.946	.947	.948	.949	.950

				1						
Observed temperature in °F	0.930	0.931	0.932	0.933	0.934	0.935	0.936	0.937	0.938	0.939
		(Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30	0.919	0.920	0.921	0.922	0.923	0.924	0.925	0.926	0.927	0.928
32	.920	.921	.922	.923	.924	.925	.926	.927	.928	.929
34	.920	.921	.922	.923	.924	.925	.926	.927	.928	.929
36 38	.921 .922	.922 .923	.923 .924	.924 .925	.925 .926	.926	.927	.928 .929	.929 .930	.930 .931
	.044	.020	.04%	.040	.020	.041	.940	.040	.000	.901
40	.9230	.9240	.9250	.9260	.9270	.9280	.9290	.9300	.9310	.9320
42	.9235	.9245	.9255	.9265	.9275	.9285	.9295	.9305	.9315	.9325
44	.9245	.9255	.9265	.9275	.9285	.9295	.9305	.9315	.9325	.9335
46. 48.	.9250 .9255	.9260 .9265	.9270 .9275	.9280 .9285	.9290	.9300	.9310 .9320	.9320 .9330	.9330 .9340	.9340
and the second	.0400	.0200	10410	.0200	.0400	.0000	.0020	.0000	.0010	.0000
50	.9265	.9275	.9285	.9295	.9305	.9315	.9325	.9335	.9345	.9355
52	.9270	.9280	.9290	.9300	.9310	.9320	.9330	.9340	.9350	.9360
54	.9280	.9290	.9300	.9310	.9320	.9330	.9340	.9350	.9300	.9370
56 58	.9285 .9295	.9295 .9305	.9305 .9315	.9315	.9325 .9335	.9335 .9345	.9345 .9355	.9355	.9365 .9375	.9375 .9385
	.0400	.9909	.9510	.9325	.9000	.3040	.9000	.9365	.9010	.9569
60	.9300	.9310	.9320	.9330	.9340	.9350	.9360	.9370	.9380	.9390
62	.9305	.9315	.9325	.9335	.9345	.9355	.9365	.9375	.9385	.9395
64	.9315	.9325	.9335	.9345	.9355	.9365	.9375	.9385	.9395	.9405
66 68	.9320 .9330	.9330 .9340	.9340 .9350	.9350	.9360 .9370	.9370 .9380	.9380 .9390	.9390	.9400 .9410	.9410
	.0000	.0040	.0000	.9360	.0010	.0000	.9090	.9400	.9110	.9420
70	.9335	.9345	.9355	.9365	.9375	.9385	.9395	.9405	.9415	.9425
72	.9340	.9350	.9360	.9370	.9380	.9390	.9400	.9410	.9420	.9430
74 76	.9350 .9355	.9360	.9370	.9380	.9390	.9400	.9410	.9420	.9430	.9440
78	.9365	.9365 .9375	.9375 .9385	.9385 .9395	.9395 .9405	.9405 .9415	.9415 .9425	.9425 .9435	.9435 .9445	.9445 .9455
~										
80 82	.937 .937	.938	.939	.940	.941	.942	.943	.944	.945	.946
84	.938	.938 .939	.939 .940	.940 .941	.941 .942	.942 .943	.943 .944	.944	.945 .946	.946 .947
86	.939	.940	.941	.942	.943	.944	.945	.946	.947	.948
88	.940	.941	.942	.943	.944	.945	.946	.947	.948	.949
00	040	0/1	040	040		0.45	010	0.17	040	040
90 92	.940 .941	.941 .942	.942 .943	.943 .944	.944	.945 .946	.946 .947	.947 .948	.948 .949	.949
94	.941	.942	.943	.944 .945	.945 .946	.946	.947	.948	.949	.950
96	.942	.943	.944	.945	.946	.947	.948	.949		
98	.943	.944	.945	.946	.947	.948	.949	.950		
100	.944	.945	.946	047	049	010	050			
02	.944	.945	.940	.947 .947	.948 .948	.949	.950			
.04	.945	.946	.947	.948	.949	.000				
06	.946	.947	.948	.949	.950			I DA		
.08	.947	.948	.949	.950				-		
10	.947	.948	.949					1		
12	.948	.949	.949							
14	.949	.950								
116	.949									
118	.950									

KANSAS CITY TESTING LABORATORY

REDUCTION OF SPECIFIC GRAVITY READINGS TO 60°F-Con.

		Observed specific gravities												
Observed temperature in °F	0.940	0.941	0.942	0.943	0.944	0.945	0.946	0.947	0.948	0.949				
			Corresp	onding a	specific	gravitie	s at 60°	/60° F						
30	0.929	0.930	0.931	0.932	0.933	0.934	0.935	0.936	0.937	0.938				
32	.930	.931	.932	.933	.934	.935	.936	.937	.938	.939				
34	.930	.931	.932	.933	.934	.935	.936	.937	.938	.939				
33	.931	.932	.933	.934	.935	.936	.937	.938	.939	.940				
38	.932	.933	.934	.935	.933	.937	.938	.939	.940	.941				
40	.9330	.9340	.9350	.9360	.9370	.9380	.9390	.9400	.9410	.942				
42	.9335	.9345	.9355	.9365	.9375	.9385	.9395	.9405	.9415	.942				
44	.9345	.9355	.9365	.9375	.9385	.9395	.9405	.9415	.9425	.943				
46	.9350	.9300	.9370	.9380	.9390	.9400	.9410	.9420	.9430	.944				
48	.9300	.9370	.9380	.9390	.9400	.9410	.9420	.9430	.9440	.945				
50	.9365	.9375	.9385	.9395	.9405	.9415	.9425	.9435	.9445	.945				
52	.9370	.9380	.9390	.9400	.9410	.9420	.9430	.9440	.9450	.946				
54	.9380	.9390	.9400	.9410	.9420	.9430	.9440	.9450	.9460	.947				
53	.9385	.9395	.9405	.9415	.9425	.9435	.9445	.9455	.9465	.947				
58	.9395	.9405	.9415	.9425	.9435	.9445	.9455	.9465	.9475	.948				
60	.9400	.9410	.9420	.9430	.9440	.9450	.9460	.9470	.9480	.949				
62	.9405	.9415	.9425	.9435	.9445	.9455	.9465	.9475	.9485	.949				
64	.9415	.9425	.9435	.9445	.9455	.9465	.9475	.9485	.9495	.9500				
6	.9420	.9430	.9440	.9450	.\$460	.9470	.9480	.9490	.9500					
8	.9430	.9440	.9450	.9460	.9470	.9480	.9490	.9500						
70	.9435	.9445	.9455	.9465	.9475	.9485	.9495							
72	.9440	.9450	.9460	.9470	.9480	.9490	.9500							
74	.9450	.9460	.9470	.9480	.9490	.9500								
73	.9455	.9465	.9475	.9485	:9495	.0000								
8	.9465	.9475	.9485	.9495	.9500									
30	.947	.948	.949	.950										
2	.947	.948	.949	.000										
84	.948	.949	.950											
6	.949	.950												
8	.950	.000												

417

BULLETIN NUMBER FIFTEEN OF

				Observ	ed speci	fic grav	vities			
Observed temperature in °F	0.950	0.951	0.952	0.953	0.954	0.955	0.956	0.957	0.958	0.959
			Correspo	onding s	specific	gravitie	s at 60°	/60° F		
30 32	0.939	0.940 .941	0.941	0.942	0.943	0.944 .945	0.945 .946	0.946	0.947	0.948
34	.940	.941	.942	.943	.944	.945	.946	.947	.948	.949
36	.941	.942	.943	.944	.945	.946	.947	.948	.949	.950
38	.942	.943	.944	.945	.946	.947	.948	.949	.950	
40	.9430	.9440	.9450	.9460	.9470	.9480	.9490	.9500		
42	.9435	.9445	.9455	.9465	.9475	.9485	.9495			
44	.9445	.9455	.9465	.9475	.9485	.9495	.9500			
46	.9450	.9450	.9470	.9480	.9490	.9500				
48	.9460	.9470	.9480	.9490	.9500					
50	.9465	.9475	.9485	.9495						
52	.9470	.9480	.9490	.9500	1					
54	.9480	.9490	.9500							
53	.9485	.9495	1. 10.10							
58	.9495	.9500						-		
60	.9500	1.								

Specific Gravity Tables

Equivalent of Degrees Baume' (American Standard) and Specific Gravity at 60°F.

Degrees Baume'=145______ For Liquids Heavier than Water Sp. Gr.

Degrees Baume'	Specific Gravity	Degrees Baume	Specific Gravity	Degrees Baume	Specific Gravity	Degrees Baume	Specific Gravity
0.0	1.0000	.7	1.0262	.4	1.0538	.1	1.0829
.1	1.0007	.8	1.0269	.5	1.0545	.2	1.0837
.2	1.0014	.9	1.0276	.6	1.0553	.3	1.0845
.3	1.0021	4.0	1.0284	.7	1.0561	.4	1.0853
.4	1.0028	.1	1.0291	.8	1.0569	.5	1.0861
.5	1.0035	.2	1.0298	.9	1.0576	.6	1.0870
.6	1.0042	.3	1.0306	8.0	1.0584	.7	1.0878
.7	1.0049	.4	1.0313	.1	1.0592	.8	1.0886
.8	1.0055	.5	1.0320	.2	1.0599	.9	1.0894
.9	1.0062	.6	1.0328	.3	1.0007	12.0	1.0902
1.0	1.0069	.7	1.0335	.4	1.0615	.1	1.0910
.1	1.0076	.8	1.0342	.5	1.0623	.2	1.0919
.2	1.0083	.9	1.0350	.6	1.0630	.2	1.0927
.3	1.0090	5.0	1.0357	.7	1.0638	.4	1.0935
.4	1.0097	.1	1.0365	.8	1.0643	.5	1.0943
.5	1.0105	.2	1.0372	.9	1.0654	.6	1.0952
.6 .7	1.0112	.3	1.0379	9.0	1.0662	.7	1.0960
.7	1.0119	.4	1.0387	.1	1.0670	.8	1.0968
.8	1.0126	.5	1.0394	.2	1.0677	.9	1.0977
.9	1.0133	.6	1.0402	.3	1.0685	13.0	1.0985
2.0	1.0140	.7	1.0409	.4	1.0693	.1	1.0993
.1	1.0147	.8	1.0417	.5	1.0701	.2	1.1002
.2	1.0154	.9	1.0424	.6	1.0709	.3	1.1010
.3	1.0161	6.0	1.0432	.7	1.0717	.4	1.1018
.4	1.0168	.1	1.0433	.8	1.0725	.5	1.1027
.5	1.0175	.2	1.0447	.9	1.0733	.6	1.1035
.6	1.0183	.2 .3	1.0454	10.0	1.0741	.7	1.1043
.7	1.0190	.4	1.0462	.1	1.0749	.8	1.1052
.8	1.0197	.5	1.0469	.2	1.0757	.9	1.1060
.9	1.0204	.6 .7	1.0477	.3	1.0765	14.0	1.1069
3.0	1.0211	.7	1.0484	.4	1.0773	.1	1.1077
.1	1.0218	.8	1.0492	.5	1.0781	.2	1.1083
.2	1.0226	.9	1.0500	.6	1.0789	.3	1.1094
.3	1.0233	7.0	1.0507	.7	1.0797	.4	1.1103
.4	1.0240	.1	1.0515	.8	1.0805	.5	1.1111
.5	1.0247	.2	1.0522	.9	1.0813	.6	1.1120
.6	1.0255	.3	1.0530	11.0	1.0821	.7	1.1128

EQUIVALENT BAUME' DEGREES-Con.

Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume	Specific Gravity
.8	1.1137	.2	1.15%	.6	1.1944	28.0	1.2393
.9	1.1145	.3	1.1535	.0	1.1954	.1	1.2383
15.0	1.1145	.3	1.1545	.8	1.1964	.1	1.2404
.1	1.1162	.4	1.1545	.8	1.1974	.2	1.2414
.2	1.1102	.6	1.1563	24.0	1.1983	.0	1.2425
.2	1.1180	.0	1.1505	.1	1.1993	.4 .5	1.2430
.4	1.1188	.8	1.1572	.1	1.2003	.5	1.2440
.4	1.1197		1.1581	.3	1.2003	.0	1.2454
.5	1.1206	.9			1.2013		
.6 .7	1.1200	20.0	1.1600	.4 .5	1.2023	.8	1.2478
.7		.1	1.1609	.5	1.2033	.9	1.2489
.8	1.1223	.2	1.1619	.6		29.0	1.2500
.9	1.1232	.3	1.1628	.7	1.2053	.1	1.2511
16.0	1.1240	.4 .5	1.1637	.8	1.2063	.2	1.2522
.1	1.1249	.5	1.1647	.9	1.2073	.3	1.2532
.2 .3	1.1258	.6	1.1656	25.0	1.2083	.4	1.2543
.3	1.1267	.7	1.1665	.1	1.2093	.5	1.2554
.4	1.1275	.8	1.1675	.2	1.2104	• .6	1.2565
.5	1.1284	.9	1.1684	.3	1.2114	.7	1.2576
.6	1.1293	21.0	1.1694	.4	1.2124	.8	1.2587
.7	1.1302	.1	1.1703	.5	1.2134	.9	1.2598
.8	1.1310	.2	1.1712	.6	1.2144	30.0	1.2609
.9	1.1319	.3	1.1722	.7	1.2154	.1	1,2620
17.0	1.1328	.4	1.1731	.8	1.2164	.2	1.2631
.1	1.1337	.5	1.1741	.9	1.2175	.3	1.2642
.2	1.1346	.6	1.1750	25.0	1.2185	.4	1.2653
.3	1.1355	.7	1.1760	.1	1.2195	.5	1.2364
.4	1.1364	.8	1.1769	.2	1.2205	.6	1.2675
.5	1.1373	.9	1.1779	.3	1.2216	.7	1.2686
.6	1.1381	22.0	1.1789	.4	1.2226	.8	1.2697
.7	1.1390	.1	1.1798	.5	1.2236	.9	1.2708
.8	1.1399	.2	1.1808	.6	1.2247	31.0	1.2719
.9	1.1408	.3	1.1817	.7	1.2 ± 57	.1	1.2730
18.0	1.1417	.4	1.1827	.8	1.2267	.2	1.2742
.1	1.1426	.5	1.1837	.9	1.2278	.3	1.2753
.2	1.1435	.6	1.1846	27.0	1.2288	.4	1.2764
.3	1.1444		1.1856	.1	1.2299	.5	1.2775
.4	1.1453	.8	1.1866	.2	1.2309	.6	1.2787
.5	1.1462	.9	1.1876	.3	1.2319	.7	1.2798
6	1.1472	23.0	1.1810	.4	1.2330	.8	1.2809
.6 .7	1.1481	.1	1.1895	.5	1.2340	.9	1,2821
.8	1.1490	.1	1.1905	.6	1.2351	32.0	1.2832
.9	1.1499	.2 .3	1.1905	.7	1.2351	.1	1.2843
19.0	1.1499	.0	1.1915	.8	1.2372	.2	1,2855
.1	1.1508	.4 .5		.0	1.2372	.3	1,2866
.1	1.101/	6,	1.1934	.9	1,2000	6,	1,2000

EQUIVALENT BAUME' DEGREES-Con.

Degrees Baume	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume'	Specifi Gravit
.4	1.2877	.8	1.3401	.2	1.3969	.6	1.4588
.5	1.2889	.9	1.3414	.3	1.3983	.7	1.4602
.6	1.2900	37.0	1.3426	.4	1.3996	.8	1.4617
.7	1.2912	.1	1.3438	.5	1.4010	.9	1.4632
.8	1.2923	.2	1.3451	.6	1.4023	46.0	1.4646
.9	1.2935	.3	1.3463	.7	1.4037	.1	1.4651
33,0	1.2946	.4	1.3476	.8	1.4050	.2	1.4676
.1	1.2958	.5	1.3488	.9	1.4064	.3	1.4691
.2	1.2970	.6	1.3501	42.0	1.4078	.0	1.4091
.4	1.2970	.7	1.3514	42.0	1.4078	.4	1.4700
.4		.8	1.3526	.1	1.4105	.6	1.4736
	1.2993	.0	1.3539			.0	
.5	1.3004	.9 38.0	1.3551	.3	1.4119		1.4751
.6	1.3016			.4	1.4133	.8	1.4760
.7	1.3028	.1	1.3564	.5	1.4146	.9	1.4781
.8	1.3040	.2	1.3577	.6	1.4160	47.0	1.4796
.9	1.3051	.8	1.3:53	.7	1.4174	.1	1.4811
34.0	1.3063	.4	1.3602	.8	1.4188	.2	1.483
.1	1.3075	.5	1.3615	.9	1.4202	.3	1.4841
.2	1.3087	.6	1.3328	43.0	1,4216	.4	1.4857
.3	1.3098	.7	1.3641	.1	1.4230	.5	1.4872
.4	1.3110	.8	1.3653	.2	1.4244	.6	1.488
.5	1.3122	.9	1.3666	.3	1.4258	.7	1.490
.6	1.3134	39.0	1.3679	.4	1.4272	.8	1.4918
.7	1.3146	.1	1.3692	.5	1.4283	.9	1.493
.8	1.3158	.2	1.3705	.6	1.4300	48.0	1.4948
9	1.3170	.3	1.3718	.7	1.4314	.1	1.496
35.0	1.3182	.4	1.3731	.8	1.4328	.2	1.4979
.1	1.3194	.5	1.3744	.9	1.4342	.3	1,499
.2	1.3206	.6	1.3757	44.0	1.4356	.4	1.501
.3	1.3218	.7	1.3770	.1	1.4371	.5	1.5020
.4	1.3230	.8	1.3783	.2	1.4385	.6	1.504
.5	1.3242	.9	1.3796	.3	1.4399	.7	1.5057
.6	1.3254	40.0	1.3810	.4	1.4414	.8	1.5073
.7	1.3266	.1	1.3823	.5	1.4428	.9	1.5088
.8	1.3278	.2	1 3836	.6	1.4442	49.0	1.5104
.9	1.3291	.3	1.3849	.7	1.4457	.1	1.5120
33.0	1.3303	.4	1.3862	.8	1.4471	.2	1.5130
.1	1.3315	.5	1,3876	.9	1.4486	.3	1.515
.1	1.3327	.6	1.3889	45.0	1.4500	.4	1.5167
.2	1.3324	.0	1.3889	40.0	1.4515	.5	1.518
.3	1.3329	.8	1.3902	.2	1.4529		1.5199
.5	1.3364	.8	1.3910	.2	1.4544	.7	1.521
	1.3304 1.3376		1.3828			.8	1.5231
.6 .7		41.0		.4	1.4558	.0	1.523
.1	1.3389	.1	1.3956	6,	1.4573	.9	1.024

EQUIVALENT BAUME' DEGREES-Con.

Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity	Degrees Baume'	Specific Gravity
50.0	1.5263	1	1.6129	.1	1.7079	.1	1.8148
.1	1.5279	.2	1.6147	.2	1.7099	.2	1.8140
.1	1.5295	.2	1.6165	.2	1.7119	.2	1.8170
.2	1.5312	.3	1.6183	.3	1.7139	.4	1.8195
.3	1.5328	.5	1.6201	.5	1.7160	.5	1.8239
.5	1.5344	.6	1.6219	.6	1.7180	.6	1.8262
.6	1.5360	.0	1.6237	.0	1.7200	.7	1.8285
.0	1.5376	.8	1.6256		1.7221	.8	1.8200
.7	1.5393	.9	1.6459	.9	1.7241	.9	1.8331
.9	1.5409	56.0	1.6292	61.0	1.7262	66.0	1.8354
51.0	1.5426	.1	1.6310	.1	1.7282	.1	1.8378
.1	1.5442	.2	1.6329	.2	1.7303	.2	1.8401
.2	1.5458	.3	1.6347	.3	1.7324	.3	1.8424
.3	1.5475	.4	1.6366	.4	1.7344	.4	1.8448
.4	1.5491	.5	1.6384	.5	1.7365	.5	1.8471
.5	1.5508	.6	1.6403	.6	1.7386	6	1.8495
.6	1.5525	.7	1.6421	.7	1.7407	.7	1.8519
.1	1.5541	.8	1.6440	.8	1.7428	.8	1.8542
.8	1.5558	.9	1,6459	.9	1.7449	.9	1.8506
.9	1.5575	57.0	1.6477	62.0	1.7470	67.0	1.8590
52.0	1,5591	.1	1.6496	.1	1.7491	.1	1.8614
.1	1.5608	.2	1.6515	.2	1.7512	.2	1.8638
.2	1.5625	.3	1.6534	.3	1.7533	.3	1.8662
.3	1.5642	.4	1.6553	.4	1.7554	.4	1.8686
.4 5	1.5659	.5	1.6571	.5	1.7576	.5	1.8710
5	1.5676	.6	1.6590	.6	1,7597	.6	1.8734
.0	1.5693	.7	1.6609	.7	1.7618	.7	1.8758
.7	1.5710	.8	1.6628	.8	1.7640	.8	1.8782
.8	1.5727	.9	1.6459	.9	1.7661	.9	1.8807
.1)	1.5744	58.0	1.6667	63.0	1.7683	68.0	1.8831
53.0	1.5761	.1	1.6686	.1	1.7705	.1	1.8856
.1	1.5778	.2	1.6705	.2	1.7726	.2	1.8880
.2	1.5795	.3	1.6724	.3	1.7748	.3	1,8905
.3	1.5812	.4	1.6744	.4	1.7770	.4	1.8930
.4	1.5830	.5 -	1.6763	.5	1.7791	.5	1.8954
.5	1.5847	.6	1.6782	.6	1.7813	.6	1.8979
.6	1.5864	.7	1.6802	.7	1.7835	.7	1.9004
.7	1.5882	.8	1.6821	.8	1.7857	.8	1.9029
.8 .9	1.5899	.9	1.6841	.9	1.7879	.9	1.9054
.9	1.5917	59.0	1.6860	64.0	1.7901	69.0	1.9079
	1.5934	.1	1.6880	.1	1.7923	1	1.9104
.1 .2	1.5952	.2	1.6900	.2	1.7946	.2	1.9129
.2	1.5969	.3	1.6919	.3	1.7968	.3	1.9155
.3	1.5987 1.6004	.4	1,6939	.4	1.7990	.4	1.9180
.4	1.6004	.5	1.6959	.5	1.8012	.5	1.9205
.6	1.6040	.6 .7	1.6979	.6 .7	1.8035	.6	1.9231
.0	1.6058	.8	$1.6999 \\ 1.7019$.8	$1.8057 \\ 1.8080$.7 .8	1.9256 1.9282
.8	1.6058 1.6075	.8	1.7019	.8 .9	1.8080 1.8102	.8	
.0	1.6093	60.0	1.7039	.9 65.0	1.8102 1.8125	.9 70.0	1.9308 1.0222
55.0	1.6111	00.0	1.1009	00.0	1.8129	10.0	1.9333

Specific Gravity and Content of Sulphuric Acid

Gravity 15°	correspond to grams		tains	Specific Gravity 15°	we	arts by ight pond to	1 liter contains grams		
4° in vacuo	% SO3	H2SO4	SO3	H ₂ SO ₄	4° in vacuo	% SO3	% H ₂ SO ₄	SO3	H ₂ SO
1.000	0.07	0.09	1	1	1.190	21.26	26.04	253	310
1.005	0.68	0.83	7	8	1.195	21.78	26.68	260	319
1.010	1.28	1.57	13	16	1.200	22.30	27.30	268	328
1.015	1.88	2.30	19	23	1.205	22.82	27.95	275	337
1.020	2.47	3.03	25	31	1.210	23.33	28.58	282	346
1.025	3.07	3.76	32	39	1.215	23.84	29.21	290	355
1.030	3.67	4.40	38	46	1.220	24.36	29.84	297	364
1.035	4.27	5.23	44	54	1.225	24.88	30.48	305	373
1.040	4.87	5.96	51	62	1.230	25.39	31.11	312	382
1.045	5.45	6.67	57	71	1.235	25.88	31.70	320	391
1.050	6.02	7.37	63	77	1.240	26.35	32.28	327	400
1.055	6.59	8.07	70	85	1.245	26.83	32.86	334	409
1.060	7.16	8.77	76	93	1.240	20.83	33.43	341	409
	7.73	9.47	82	102	1.255	27.76	34.00	348	418
1.065				102					
1.070	8.32	10.19	89		1.260	28.22	34.57	356	435
1.075	8.90	10.90	96	117	1.265	28.69	35.14	363	444
1.080	9.47	11.60	103	125	1.270	29.15	35.71	370	454
1.085	10.04	12.30	109	133	1.275	29.62	36.29	377	462
1.090	10.60	12.99	116	142	1.280	30.10	36.87	385	472
1.095	11.16	13.67	122	150	1.285	30.57	37.45	393	481
1.100	11.71	14.35	129	158	1.290	31.04	38.03	400	490
1.105	12.27	15.03	136	166	1.295	31.52	38.61	408	500
1.110	12.82	15.71	143	175	1.300	31.99	39.19	416	510
1.115	13.36	16.36	149	183	1.305	32.46	39.77	424	519
1.120	13.89	17.01	156	191	1.310	32.94	40.35	432	529
1.125	14.42	17.66	162	199	1.315	33.41	40.93	439	538
1.130	14.95	18.31	169	207	1.320	33:88	41.50	447	548
1.135	15.48	18.96	176	215	1.325	34.35	42.08	455	557
1.140	16.01	19.61	183	223	1.330	34.80	42.66	462	507
1.145	16.54	20.26	189	231	1.335	35.27	43.20	471	577
1.150	17.07	20.91	196	239	1.340	35.71	43.74	479	586
1.155	17.59	21.55	203	248	1 345	36.14	44.28	486	596
1.160	18.11	22.19	210	257	1.350	36.58	44.82	494	605
1.165	18.64	22.83	217	266	1.355	37.02	45.35	502	614
1.170	19.16	23.47	224	275	1.360	37.45	45.88	509	624
1.175	19.69	24.12	231	283	1.365	37.89	46.41	517	633
1.180	20.21	24.76	238	292	1.370	38.32	46.94	525	643
1.185	20.73	25.40	246	301	1.375	38.75	47.47	533	653

SPECIFIC GRAVITY AND CONTENT OF SULPHURIC ACID-Con.

Specific Gravity 15°	we	arts by ight spond to	con	iter tains ams	Specific Gravity 15°	we	arts by ight pond to	1 liter contains grams		
4° in vacuo	% 803	[%] H ₂ SO ₄	SO3	H ₂ SO ₄	4° in vacuo	% SO3	[%] H ₂ SO ₄	SO3	H ₂ SO	
1.380	39.18	48.00	541	662	1.675	61.20	74.97	1025	1256	
1.385	39.62	48.53	549	672	1.680	61.57	75.42	1034	1267	
1.390	40.05	49.06	557	682	1.685	61.93	75.86	1043	1278	
1.395	40.48	49.50	564	592	$1.690 \\ 1.695$	62.29	76.30 76.73	1053	1289	
1.400 1.405	40.91 41.33	50.11 50.63	$573 \\ 581$	702 711	1.700	$62.64 \\ 63.00$	77.17	1032 1071	1301 1312	
1.400	41.55	51.15	589	721	1.705	63.35	77.60	1080	1323	
1.415	42.17	51.66	597	730	1.710	63.70	78.04	1089	1334	
1.420	42.57	52.15	604	740	1.715	64.07	78.48	1099	1346	
1.425	42.93	52.63	612	750	1.720	64.43	78.92	1108	1357	
1.430	43.36	53.11	620	759	1.725	64.78	79.36	1118	1369	
1.435	43.75	53.59	628	769	1.730	65.14	79.80	1127	1381	
1.440	44.14	54.07	636	779	1.735	65.50	80.24	1136	1392	
1.445	44.53	54.55	643	789	$1.740 \\ 1.745$	$65.86 \\ 66.22$	80.68 81.12	$1146 \\ 1156$	1404	
$1.450 \\ 1.455$	$44.92 \\ 45.31$	55.03 55.50	651 659	798 808	1.745	00.22 66.58	81.12 81.56	1165	1416 1427	
1.460	45.69	55.97	667	817	1.755	66.94	82.00	1175	1439	
1.465	46.07	56.43	675	827	1.760	67.30	82.44	1185	1451	
1.470	46.45	56.90	683	837	1.765	67.65	82.88	1194	1463	
1.475	46.83	57.37	691	846	1.770	68.02	83.32	1204	1475	
1.480	47.21	57.83	699	856	1.775	68.49	83.90	1216	1489	
1.485	47.57	58.28	707	865	1.780	68.98	84.50	1228	1504	
1.490	47.95	58.74	715	876	1.785	69.47	85.10	1240	1519	
1.495	48.34	59.22	723	885	1.790	69.96	85.70	1252	1534	
$1.500 \\ 1.505$	48.73	59.70	731	896	1.795 1.800	70.46 70.94	86.30	1265	$1549 \\ 1564$	
1.510	49.12 49.51	60.18 60.65	739 748	906 916	1.800	71.50	86.90 87.60	1277 1291	1504	
1.515	49.89	61.12	756	926	1.810	72.08	88.30	1305	1598	
1.520	50.28	61.59	764	936	1.815	72.69	89.05	1319	1621	
1.525	50.66	62.06	773	946	1.820	73.51	90.05	1338	1639	
1.530	51.04	62.53	781	957	1.821	73.63	90.20	1341	1643	
1.535	51.43	63.00	789	967	1.822	73.80	90.40	1345	1647	
1.540	51.78	63.43	797	977	1.823	73.96	90.60	1348	1651	
1.545	52.12	63.85	805	987	1.824	74.12	90.80	1352	1656	
1.550	52.46	64.26	813	996	1.825	74.29	91.00	1356	1661	
1.555 1.560	52.79 53.12	64.67 65.08	821 829	1006 1015	$1.826 \\ 1.827$	$74.49 \\ 74.69$	$91.25 \\ 91.50$	1360 1364	$1666 \\ 1671$	
1.565	53.46	65.49	829	1015	1.828	74.86	91.50	1368	1676	
1.570	53.80	65.90	845	1025	1.829	75.03	91.90	1372	1681	
1.575	54.13	66.30	853	1044	1.830	75.19	92.10	1376	1685	
1.580	54.46	66.71	861	1054	1.831	75.35	92.30	1380	1690	
1.585	54.80	67.13	869	1064	1.832	75.53	92.52	1384	1695	
1.590	55.18	67.59	877	1075	1.833	75.72	92.75	1388	1700	
1.595	55.55	68.05	886	1085	1.834	75.96	93.05	1393	1706	
1.600	55.93	68.51	897	1096	1.835	76.27	93.43	1400	1713	
1.605 1.610	56,30 56,68	68.97	904	1107	1.836 1.837	$76.57 \\ 76.90$	93.80 94.20	$1405 \\ 1412$	1722 1730	
1.615	57.05	69.43 69.89	913 921	$1118 \\ 1128$	1.838	77.23	94.60	1412	1730	
1.620	57.40	70.32	930	1128	1.839	77.55	95.00	1426	1748	
1.625	57.75	70.74	938	1150	1.840	78.04	95.60	1436	1759	
1.630	58.09	71.16	947	1160	1.8405	78.33	95.95	1441	1765	
1.635	58.43	71.57	955	1170	1.8410	79.19	97.00	1458	1786	
1.640	58.77	71.99	964	1181	1,8415	79.76	97.70	1469	1799	
1.645	59.10	72.40	972	1192	1.8410	80.16	98.20	1476	1808	
1.650	59.45	72.82	981	1202	1.8405	80.57	98.70	1483	1816	
$1.655 \\ 1.660$	59.78	73.23	989	1212	1.8400	80.98	99.20	1490	1825	
1.665	$60.11 \\ 60.46$	73.64 74.07	998 1007	$1222 \\ 1233$	1.8395 1.8390	81.18 81.39	99.45 90.70	$1494 \\ 1497$	1830 1834	
1.670	60.40	74.07	1016	1233	1.8390	81.59	99.95	1497	1834	

Percentage of Sulphur Trioxide and Sulphuric Acid in Fuming Sulphuric Acid

Total SO ₃ as found		acid ins %	Total SO3 as found		acid ins %	Total as found	the acid contains			
by titration	H_2SO_4	SO3	by titration	H_2SO_4	803	by titration	H ₂ SO ₄	SO:		
81.8326	100	0	87.8775	66	34	93.9389	33	67		
81.8163	99	1	88.0612	65	35	94.1224	32	68		
82,0000	98		88.2448	64	36	94.3061	31	69		
82,1836	97	23	88.4285	63	37	94.4897	30	70		
82.3374	96	4	88.6122	62	38	94.6734	29	71		
82.5510	95	5	88,7959	61	39	94 8571	28	72		
82,7346	94	6	88.9795	60	40	95.0408	27	73		
82,9183	93	6 7	89.1632	59	41	95.2244	26	74		
83.1020	92	8	89.3469	58	42	95.4081	25	75		
83.2857	91	9	89.5306	57	43	95.5918	24	76		
83.4693	90	10	89.7142	56	44	95.7755	23	77		
83.6530	89	11	89,8979	55	45	95,9591	22	78		
83,8367	88	12	90.0816	54	46	96,1428	21	79		
81.0204	87	13	90,2653	53	47	96.3265	20	80		
84.2040	86	14	90.4489	52	48	96.5102	19	81		
84.3877	85	15	90.6326	51	49	96,6938	18	82		
84 5714	84	16	90.8163	50	50	96.8775	17	83		
84.7551	83	17	91.0000	49	51	97.0612	16	84		
84.9387	82	18	91.1833	48	52	97.2448	15	85		
85,1224	81	19	91.3673	47	53	97.4285	14	83		
85,3061	80	20	91.5510	46	54	97.6122	13	87		
85.4897	79	21	91.7346	45	55	97.7959	12	88		
85.6734	78	22	91,9183	44	56	97.9795	11	89		
85,8571	77	- 23	92,1020	43	57	93.1-532	10	- 90		
86.0408	76	24	92.2857	42	58	98 3469	9	91		
86.2244	75	25	92,4693	41	59	98.5306	8	92		
83.4081	74	26	92.6530	40	60	98.7142	7	93		
86.5918	73	27	92.8367	39	61	98.8979	6	94		
86.7755	72	28	93.0204	38	62	99 0816	5	95		
86.9591	71	29	93.2040	37	63	99.2753	4	96		
87.1428	70	30	93.3877	35	64	99.4489	3	97		
87.3265	69	31	93.5714	35	65	99.6326	2	98		
87.5102	68	32	93.7551	34	66	99.8163	1	99		
87.6938	67	33		10.00						

Sodium Hydroxide Solution at 15°C (Caustic Soda) LUNGE.

Specific	Degrees	Degrees	Per Cent	Per Cent		$\begin{array}{c ccccccccccccccccccccccccccccccccccc$				
Gravity	Baume'	Twaddell	Na2O.	NaOH.	$Na_2O.$	NaOH				
1,007	1.0	1.4	0.47	0.61	4	6				
1.014	2.8	2.9	0.93	1.20	9	12				
1.022	3.1	4.4	1.55	2.00	16	- 21				
1.029	4.1	5.8	2.10	2.70						
1.036	5.1	7.2	2.60	3.35						
1.045	6.2	9.0	3.10	4.00						
1.052	7.2	10.4	3.60	4.64						
1.060	8.2	12.0	4.10	5.29						
1.067	9.1	13.4	4.55	5.87						
1.075	10.1	15.0	5.08	6.55						
1.083 1.091	11.1	$16.6 \\ 18.2$	5.67	$7.31 \\ 8.00$						
1.100	12.1 13.2	20.0	$6.20 \\ 6.73$	8.68						
1.108	13.2	20.0	7.30	9.42						
1.116	15.1	23.2	7.80	10.06						
1.125	16.1	25.0	8.50	10.97						
1.134	17.1	26.8	9.18	11.84						
1.142	18.0	28.4	9.80	12.64						
1.152	19.1	30.4	10.50	13.55						
1.162	20.2	32.4	11.14	14.37	129	167				
1.171	21.2	34.2	11.73	15.13	137	177				
1.180	22.1	36.0	12.33	15.91						
1.190	23.1	38.0	13.00	16.77						
1.200	24.2	40.0	13.70	17.67						
1.210	25.2	42.0	14.40	18.58						
$1.220 \\ 1.231$	26.1	44.0	15.18	19.58						
1.231	27.2 28.2	46.2 48.2	$15.96 \\ 16.76$	20.59 21.42						
1.241	28.2	48.2	17.55	21.42						
1.263	30.2	52.6	18.35	23.67						
1.274	31.2	54.8	19.23	24.81						
1.285	32.2	57.0	20.00	25.80		332				
1.297	33.2	59.4	20.80	26.83						
1.308	34.1	61.6	21.55	27.80						
1.320	35.2	64.0	22.35	28.83						
1.332	36.1	66.4	23.20	29.93						
1.345	37.2	69.0	24.20	31.22						
1.357	38.1	71.4	25.17	32.47						
$1.370 \\ 1.383$	39.2 40.2	74.0 76.6	25.12	33.69						
1.397	40.2	79.4	27.10 28.10	34.96 36.25						
1.410	42.2	82.0	29.05	37.47	410	528				
1.424	43.2	84.8	30.08	38.80	428	553				
1.438	44.2	87.6	31.00	39,99	446	575				
1.453	45.2	90.6	32.10	41.41	466	602				
1.468	46.2	93.6	33.20	42.83	487	629				
1.483	47.2	96.6	34.40	44.38	510	658				
1.498	48.2	99.6	35.70	46.15	535	691				
1.514	49.2	102.8	36.90	47.60	559	721				
1.530	50.2	106.0	38,00	49.02	581	750				

Table of Chloride of Calcium Solution

Specific Fravity at 64 Degrees F.	Degree Beaume at 64 Degrees F.	Degree Sal- ometer at 64 Degrees F.	Per Cent of CaCl ₂	Freezing Point in Degrees F.	Ammonia Gauge Pressure Pounds per Square Inch
1.007	1	4	0.943	+31.20	46
1.014	2	8	1.883	+30.40	45
1.021	3	12	2.89	+29.60	44
1.028	4	16	3.772	+28.80	43
1.035	5	20	4.715	+23.00	42
1.043	6	24	5.658	+26.89	41
1.050	7	28	6,601	+25.78	40
1.058	8	32	7.544	+24.67	38
1.035	9	36	8.487	+23.56	37
1.073	10	40	9.430	+22.09	35.5
1.081	11	44	10.373	+20.62	34
1.089	12	48	11.316	+19.14	32.5
1.097	13	52	12.259	+17.67	30.5
1.105	14	56	13.202	+15.75	29
1.114	15	60	14.145	+13.82	27
1.122	16	64	15.088	+11.80	25
1.131	17	68	16.031	+ 9.96	23.5
1.140	18	72	16.974	+7.68	21.5
1.149	19	76	17.917	+ 5.40	20
1.158	20	80	18.860	+ 3.12	18
1.167	21	84	19.803	- 0.84	15
1.176	- 22	88	20.746	- 4.44	12.5
1,186	23	92	21.689	- 8.03	10.5
1.196	24	96	22.632	-11.63	8
1.205	25	100	23.575	-15.23	6
1.215	26	104	24.518	-19.56	4
1.225	27	108	25.431	-24.43	1.5
1.236	28	112	26.404	-29.29	1" vacuum
1.246	29	116	27.347	35.30	5" vacuum
1.257	30	120	28,290	-41.32	8.5" vacuum
1.268	31		29.233	-47.66	12" vacuum
1.279	32		30.176		15" vacuum
1.290	33		31.119	-44.32	10" vacuum
1.302	34		32,062	-34.66	4" vacuum
1.313	35		33.	-25.00	1.5 pounds

Table of Brine Solution

Per Cent of Salt by Weigh	Degrees on Salometer at (0 Degrees Fahr.	at (0 Degrees Fahr, Specific (0 Degrees Fahr, Specific Fahr, Heat		Weight of One Gallon	Pounds of Salt in One Gallon	Pounds of Water in One Galion	Weight of One Cubic Foot		Pounds of Water in One Cubic Foot	Freezing Point Degrees Fahr.		
0	0	1.	1.	8.35	0.	8.35	12.4	0.	62.4	32.		
1	4	1.007	0.992	8.4	0.084	8.316	62.8	0.628	62.172	31.8		
5	20	1.037	0.96	8.65	0.432	8.218	64.7	3.237	61.465	25.4		
10	40	1.073	0.892	8 95	0.895	8 055	66.95	6.695		18.6		
15	60	1.115	0.855	9.3	1.395	7.905	69.57	10.435		12.2		
20	80	1.150	0.829	9.6	1.92	7.68	71.76	14.352	57.408	6.86		
25	100	1.191	0.783	9.94	2.485	7.455	74.26	18.565	55.695	1.00		
	1				1		1	Jack Street	1	1		

The Metric System, Fundamental Equivalents

The fundamental unit of the metric system is the Meter—the unit of length. From this the units of capacity (Liter) and of weight (Gram) were derived. All other units are the decimal subdivisions or multiples of these. These three units are simply related, e. g., for all practical purposes one Cubic Decimeter equals one Liter and one Liter of water weighs one Kilogram. The metric tables are formed by combining the words "Meter," "Gram," and "Liter" with the six numerical prefixes, as in the following tables:

Prefix	xes. N	leaning.	Units.
centi- = deci- = Unit = deka- = hecto- =	= one thousandth 1/1 = one hundredth 1/1 = one tenth 1/1 = one	$\begin{array}{cccc} 00 & 0.01 \\ 0 & 0.1 \\ & 1. \\ 1 & 10. \\ /1 & 100. \end{array}$	

All lengths, areas, and cubic measures in the following tables are derived from the international meter, the legal equivalent being 1 Meter = 39.37 Inches (law of July 28, 1866). In 1893 the United States Office of Standard Weights and Measures was authorized to derive the yard from the meter, using for the purpose the relation legalized in 1866, 1 Yard = 3600/3937 Meter.

The customary weights derived from the international kilogram are based on the value of 1 avoirdupois pound = 453.5924277 grams. This value is carried out farther than that given in the law, but is in accord with the latter as far as it is there given. The value of the troy pound is based upon the relation just mentioned and also the equivalent 5760/7000 avoirdupois pounds equal 1 troy pound.

In the following tables the metric unit has been selected as the common unit so that conversions may be made through the metric unit.

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A to Cm	$1.0000 \cdot 10^{5}$		•	$1.0000 \cdot 10^{-6}$ 1.0000 $\cdot 10^{-8}$	$1.6093 \cdot 10^{5}$ 5 029 $\cdot 10^{2}$	•	$3.0480 \cdot 10$ 2.5400		$1132 \cdot 10^{\circ}$	•	.85319 . 10 ⁵	$3.6576 \cdot 10^3$		2.01168 · 10 ³	• •	3.7576 · 10 ³	8.4667 $2.2860 \cdot 10$	1.0160 . 10			01 . 21.70.9	
LINEAR DIMENSIONS CONVERSION FACTORS.	$\begin{array}{c} \text{Um. to A.} \\ 1.0000 & 10^{-5}\text{KILOMETER} = 0.62137 - \text{U. S. miles} = 3280.83 \text{ ft}1. \\ 1.0000 & 10^{-2}() & \text{Merryp} - 3 98083 \text{ ft} - 39 37 \text{ inches} (\text{leval}) \end{array}$	10 CENTIMETER = 0.3937 inch = 10 millimeters.	\cdot 10 ⁴ MICRON = 0.00003937 inch = 1000 millimicrons		10^{-6} MILE (Statute) = 5280 feet = 1.609347 kilom	10^{-2} . YARD = 3 feet.	10^{-2}	10^2 MIL = 1/1000 inch	: :	· 10 ⁻⁶	BRITISH NAUTICAL MILE = 6080.4006466 + feet	· 10 ⁻⁹ FURLONG = 660 feet = 10 chains	· 10 ⁻³ U. S. FATHOM = 6 feet.		10^{-2}	10^{-4} BoLT = 40 yards	nch	10^{-2}	10^{-1} PALM = 3 inches	1 LINE = y_1^2 inch.	2.834 · 10 POINT = 7^2 inch	

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SURFACES,	
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A. to so. em.	$1.000 \cdot 10^{8}$	$1.000 \cdot 10^{6}$	$1.000 \cdot 10^{10}$	$1.000 \cdot 10^{4}$	1.000	$1.000 \cdot 10^{-2}$	$9.3239945 \cdot 10^{11}$	2.58999847 . 1010	$. 4.0468726 \cdot 10^7$	$2.52929537 \cdot 10^{5}$	$8.3613 \cdot 10^3$	$9.29034 \cdot 10^{2}$	6.4516	$6.4516 \cdot 10^{-6}$	7.871 . 10^{-6}	4.0468726 . 10°	4.0468726 . 10*	$9.29034 \cdot 10^4$				
	$1.000 \cdot 10^{-8} \dots \text{Hectare} = 2.471043930 \text{ acres} \dots$		$1000 \cdot 10^{-10} \dots square kilometer = 0.386100614 sq. miles \dots 10^{-10} \dots 10^{-10}$.000 . 10 ⁻⁴ SQUARE METER = 10.76386735908 sq. ft1.000 . 10 ⁴	square centimeter = 0.15499968997 sq. in	1.000 · 10 ²	$\cdot 10^{-12}$ 1 TOWNSHIP = 36 square miles	$3.86100614 + 10^{-11}$ 3 QUARE MILE = 640 acres = 2.78784×10^7 sq. in $2.58999847 \cdot 10^{10}$	10^{-8} ACRE = 10 sq. chains = 43560 sq. ft. = 0.0015625 sq. mi	8. 10^{-6} Sq. ROD or POLE = 272.25 sq. ft. = 0.00625 A	1.19598526 · 10 ⁻⁴ . SQUARE YARD = 9 sq. ft. = 1296 sq. in	$59 \cdot 10^{-3}$ square foot = 144 sq. in. = 3.58701×10^{-8} sq. mi	10^{-1} square inch = 0.0069444 sq. ft. = 2.491 \times 10 ⁻¹⁰ sq. mi	· 10 ⁵ square mil = 0.00001 sq. in	$1.2705 \cdot 10^5 \dots \text{CIRCULAR MIL} = (\text{wire}) = 0.00000122 \text{ sq. in} \dots \dots$	10^{-7} square chain = 4356 sq. ft. = 0.1 acres	$2.47104393 \cdot 10^{-4}$. Square Link = 62.7264 so in = 0.4356 sq. ft	L0764 • 10 ⁻⁵ 1 SQUARE (roofs and floors) = 100 sq. ft	$(a)10^{-8} = 1/10^8 = 1/10000000 = 0.0000001$		VOLIME CAPACITY CURIC CONTENTS SDACE	VODUME, CALACITY, COURT, VARATENY IS, STAUE
Sq. em	1.000 . 1	$1.000 \cdot 1$	1.000 . 1(1.000 . 1(1.000	1.000 . 1(1.0725017	3.8610061	2.4710439	3.95367028	1.19598520	1.07638675	1.5499968	1.5499968	1.2705 . 1	2.47104395	2.47104395	1.0764 . 1	(a	-		

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Cubic centimeter to A. 1000. 10^{-8}
$1.308 \cdot 10^{-9} \dots \text{CUBIC VARD} = 20 197 11 \text{S} \text{Gal} = 27 \text{cm} \text{ft}$
200 TA

	meter.
$\begin{array}{c} \cdot & 10^{-2} \\ \cdot & 10^{2} \\ \cdot & 10^{3} \\ \cdot & 10^{5} \\ \cdot & 10^{5}$	c centi c centi 10 ³ 10 ³ 10 ³ 10 ³ 10 ³ 10 ³ 10 ⁵ 10 ⁵
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	A. to cubic centimeter. 5.506 10 ² 1.10123 10 ³ 4.40492 10 ³ 3.523928 10 ³ ft. 1.06180 10 ⁵ ft. 1.16180 10 ⁵
U. S. LIQUID AND APOTHECARY MEASURE. 1623 10^{-1} 8.160 10^{-1} 2.705 10^{-1} 8.10 8.16119 2.705 10^{-1} 8.10 8.16119 2.705 10^{-1} 8.1010 8.16119 2.705 10^{-1} 8.1010 8.16119 2.705 10^{-1} 8.1010 8.16119 2.83815 10^{-3} 6.10158 9.0752 8.454 10^{-3} 6.0116 9.25673 2.113 10^{-3} 6.11189 2.9573 2.113 10^{-3} $6.1118 = 0.0357$ 9.145358 2.113 10^{-3} $6.1118 = 0.1337$ 9.145358 2.641704673 10^{-3} 6.1088 9.145358 2.641704673 10^{-3} 8.10088846 9.463588 2.641704673 10^{-3} 8.10088846 9.463588 2.641704673 10^{-3} 8.10088846 9.463588 2.097 10^{-3} 8.10088864 9.463588 2.97 $10^$	U. S. DRY MEASURE.A. to cubic centU. S. DRY MEASURE.A. to cubic centa. row of the construction of the const

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MIDOLETIANEO US.	$5.085 \cdot 10^{-3}$ BOARD FOOT $(1' \times 1' \times 1'') = 144$ cu. in1.96642	$2.760 + 10^{-7}$ corp $(4' \times 4' \times 8') = 128$ cu. ft3.6246	$8.1034 \cdot 10^{-10} \dots \text{ACRE FOOT} = 362000 \text{ U}$. S. Gal. = $43560 \text{ cu. ft} \dots \dots$	8.88 $\cdot 10^{-7}$ U. S. SHIPPING TON = 40 cu. ft	8.409 · 10^{-7} 1 BRITISH SHIPPING TON = 42 cu. ft1.1893

U. S. liquid measure \times 1.2003 = British liquid and dry of same denomination. U. S. dry measure \times 1.032 = British liquid and dry of same denomination.

1.0000 1.0000 1.0000 	$1000 \cdot 10^{6}$ 10^{-2} 10^{-2} 15552 10^{-2} 15552 10^{-2} 10^{-2} 10^{-2} 10^{-2} 10^{-2} 10^{-2} 10^{-2}		$1.0163 + 10^{6}$ $1.0163 + 10^{6}$ $1.13368 + 10^{4}$ $1.13368 + 10^{4}$ $1.2.91667 + 10^{-1}$
WEIGHTS—CONVERSION FACTORS. Grams to A. 1.000 I.0.4 CRAM (1 cc. water at 4°C.) = 15.432 grains. 1.000 MILLIGRAM = 0.015432 grains. 1.000 10.4°C.) = 15.432 grains. 1.000 MILLIGRAM = 0.015432 grains. 1.000 10.5.432 grains. 1.000 OI 5.679 T. Ibs. 1.000 2.204622341 Av. Ibs.	$\begin{array}{c} 1.000 & 10^{-0} & \text{CRAIN} (T \text{ AP}, \text{ AV}) = 1.1023 \text{ ton (short)} = 2204.6 \text{ lbs.} \\ 1.5432 & 10^{-0} & \text{CRAIN} (T \text{ AP}, \text{ AV}) = 0.0020835 \text{ ounce} (T.) \\ 6.430 & 10^{-1} & \text{FENNYWEGHT} (T.) = 24.0 \text{ grains.} \\ 6.43 & 10^{-2} & \text{CARET} (T.) = 10.0 \text{ pwt.} \\ 7.716 & 10^{-2} & \text{CARET} (T.) = 20.0 \text{ grains.} \\ 7.716 & 10^{-2} & \text{CARET} (Ap.) = 60.0 \text{ grains.} \\ 3.215 & 10^{-2} & \text{OUNCR} (Ap.) = 60.0 \text{ grains.} \\ 3.215 & 10^{-2} & \text{OUNCR} (T.) = 10974 \text{ hos}^{-1} \text{ and} \text{ arealise.} \end{array}$	2.68 $\cdot 10^{-s}$	$\begin{array}{llllllllllllllllllllllllllllllllllll$

KANSAS CITY TESTING LABORATORY

	$\begin{array}{c} \text{6.}\\ \text{H. P. hour.}\\ 3.725 \cdot 10^{-14}\\ 3.725 \cdot 10^{-7}\\ 3.725 \cdot 10^{-4}\\ 1.341 \cdot 10^{-3}\\ 1.3411128\\ 1.0000\\ 5.046 \cdot 10^{7}\\ 1.500 \cdot 10^{-6}\\ 3.931 \cdot 10^{-4}\\ 3.931 \cdot 10^{-4} \end{array}$	$3.153 \cdot 10^{-5}$ $3.653 \cdot 10^{-6}$	ter 10^{-8} 10^{-1} 10^{-1} 10^{-1} 0^{5} 0^{5}
			Kgm. meter Kgm. meter 1.0197 \cdot 10 ⁻⁸ 1.0197 \cdot 10 ⁻⁸ 1.0197 \cdot 10 ² 3.671 \cdot 10 ⁵ 3.671 \cdot 10 ⁵ 3.671 \cdot 10 ⁵ 1.381 \cdot 10 ⁻¹ 1.381 \cdot 10 ⁻¹ 1.076 \cdot 10 ⁹ 8.630 8.630
	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$2.351 \cdot 10^{-5}$ $2.724 \cdot 10^{-6}$	$\begin{array}{c} 0. & 0.1 \ 10^{-2} & 0.1 \ 10^{-2} & 10^{-6} & 0.1 \ 10^{-2} & 10^{-2} & 0.1 \ 10^{-2} & 10^{-2} & 0.1 \ 10^{-2} & 0.1 & 0.1 \ 10^{-2} & 0.1 & 0.1 \ 10^{-1} & 0.1 & 0.1 \ 0.1 & 0.1 \ 0.1 \ 0.1 & 0.1 \ 0.1 \ 0.1 & 0.1 \ 0.1 \ 0.1 \ 0.1 & 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.1 \ 0.$
	$\begin{array}{c} \textbf{Watt hour.}\\ \textbf{Watt hour.}\\ 2.778 \cdot 10^{-11}\\ 2.778 \cdot 10^{-1}\\ 2.778 \cdot 10^{-4}\\ 1000 \\ 145.6494 \\ 745.6494 \\ 745.6494 \\ 1.163 \cdot 10^{-3}\\ 2.931 \cdot 10^{-3} \end{array}$	$2.351 \cdot 10^{-2}$ $2.724 \cdot 10^{-3}$	$\begin{array}{c} 10. \\ \mathrm{Cu.ft.}\mathrm{H_2O1ft.}\\ 1.1812 & 10^{-9}\\ 1.1812 & 10^{-2}\\ 1.1812 & 10^{-2}\\ 1.1812 & 10^{-2}\\ 4.2525 & 10^{4}\\ 3.170 & 10^{-4}\\ 3.170 & 10^{-2}\\ 1.647 & 10^{-2}\\ 1.247 & 10^{-2}\\ 1.247 & 10^{-2}\\ 1.0000\\ 1.000\end{array}$
AUNA PUNA ENDIONS.	$\begin{array}{c} 3.\\ \mathrm{Kilojoule},\\ 1.000 \cdot 10^{-10}\\ 10^{-3}\\ 1.0000\\ 3600 000\\ 3600 000\\ 1.3544 \cdot 10^{-3}\\ 4.189\\ 1.3544 \cdot 10^{-3}\\ 1.3543 \cdot 10^{-3}\\ 1.0553\end{array}$	0.08463 9.8062 . 10 ⁻³ centimeter is an erg.	$\begin{array}{cccccccccccccccccccccccccccccccccccc$
WINDW	$\begin{array}{c} 2.\\ \text{Joule.}\\ 10^{-7}(a)\\ 1.0000\\ 10^3\\ 3600.00\\ 3.6\cdot10^6\\ 2.684\cdot10^6\\ 1.3544\\ 1.3544\\ 1.8543\\ 1055.3\end{array}$	84.63 9.8062 g through one c = 0.0000001.	$\begin{array}{c} 8 \\ \text{Calorie} & (15^\circ\text{C}). \\ 2.387 & 10^{-8} \\ 2.387 & 10^{-1} \\ 2.387 & 10^{-1} \\ 2.387 & 10^{-1} \\ 8.594 & 10^8 \\ 8.594 & 10^8 \\ 8.594 & 10^8 \\ 8.594 & 10^3 \\ 3.233 & 10^{-1} \\ 1.00^5 \\ 2.520 & 10^2 \\ 2.021 & 10 \end{array}$
	Erg. $1.$ $2.$ Erg. 1.0000 1.0000 Joule. 1.0000 $1.0^{-7}(a)$ Kilojoule. 10^{10} 1.0^{10} Kiloyaule. $3.6 \cdot 10^{10}$ $3.6 \cdot 10^{10}$ Kilowatt hour $3.6 \cdot 10^{10}$ $3.6 \cdot 10^{13}$ Horse-power hr. $2.684 \cdot 10^{13}$ $3.6 \cdot 10^{9}$ Foot pound. $1.3544 \cdot 10^{7}$ $1.3544 \cdot 10^{7}$ Brit. Thermal Units $1.0553 \cdot 10^{10}$ 1.3544	T ft. $(4^{\circ}C_{1})_{1}$ 8.463 Kilogram-meter 9.8062 $\cdot 10^{7}$ *The work done by one dyne actin (a) $10^{7} = 1/10^{7} = 1/10,000,000 =$	T_{r} T_{r} Frg.Froot I.b.Joule $7.384 + 10^{-5}$ Joule $7.384 + 10^{-1}$ Kilojoule $7.384 + 10^{-1}$ Watthour $7.384 + 10^{2}$ Watthour $2.658 + 10^{3}$ K. W. hr $2.658 + 10^{3}$ Ft. pound $2.658 + 10^{3}$ Ft. pound $1.982 + 10^{3}$ Ft. pound 3.093 B. T. U $7.794 + 10^{2}$ Cu.ft.H.o 1 ft. $6.250 + 10^{3}$ Kgm-meter 7.241
		11.	110,9,8,7,6,3,4,3,2,1

WORK CONVERSIONS.

434

1. $1.$ $2.$ $3.$ $4.$ $5.$ $6.$ $6.$ 2. 1.0000 0.3937 0.3337 0.3337 0.3337 0.07356 0.073937 0.03937 0.03937 0.03937 0.03281 0.07356 0.07356 0.02896 0.07356 0.02896 0.07256 0.02896 0.02866 0.02866 0.02866 0.02866	110	***		0.		*	-	1		1.		-					211	01
1. $1.$ $2.$ $3.$ $4.$ $5.$ $5.$ 1.0000 0.3937 0.03281 0.7356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.07356 0.0000 0.10000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 0.00000 </td <td>ų</td> <td>In He.</td> <td>0.02896</td> <td>0.07356</td> <td>0.8826</td> <td>0.03937</td> <td>0.39370</td> <td>1.0000</td> <td>0.02896</td> <td>28.96</td> <td>0.12725</td> <td>2.036</td> <td>8.836 . 10-4</td> <td>0.014137</td> <td>2.952 . 10-5</td> <td>29.9212</td> <td></td> <td>and in a latitude</td>	ų	In He.	0.02896	0.07356	0.8826	0.03937	0.39370	1.0000	0.02896	28.96	0.12725	2.036	8.836 . 10-4	0.014137	2.952 . 10-5	29.9212		and in a latitude
$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	LC,	Cm. He.	0.07356	0.18685	2.242	0.10000	1.0000	2.540	0.07356	73.56	0.3232	0.5171	2.245 . 10-3	0.03591	7.500 . 10-*	76.000		
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	4.	Mm. Hg.	0.7356	1.8685	22.42	1.0000	10.00	25.40	0.7356	735.6	3.232	5.171	0.02245	0.3591	7.500 . 10-4	760.00		high
$\begin{array}{cccccccccccccccccccccccccccccccccccc$		Ft. H ₂ O.	0.03281	0.08333	1.0000	0.4461	4.461	1.1330	0.03281	32.81	0.14416	2.307	0.0010012	0.01602	3.3455 . 10-5	33.9005		mercury 76.0
	2.	In H_2O .	0.3937	1.0000	12.00	0.5353	5.353	13.595	0.3937	393.7	1.7300	27.68	0.012012	0.1923	4.0145.10-4	406.806		l by a column of
	1.	Cm. H ₂ O.	1.0000	2.540	30.48	1.3595	13.595	34.54	1.000	000.000	4.394	70.32	0.03052	0.4885	1.0197 . 10-3	033.29	ater at 4° C.	pressure exerted
			water 4° C														0	Atmosphere is the
			1. C.	2. Ir	3. F	4. M	5. C	6. Ir	2.6	8. K	9.0	10. L	11. 0	12. L	13. D	14. A	N	* 150

PRESSURE CONVERSIONS.

of 45° upon the area of one square centimeter.

KANSAS CITY TESTING LABORATORY

435

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	Atmospheres. $9.679 \cdot 10^{-4}$	0.002458 0.02950	0.0013159 0.013159	0.03342 $9.679 \cdot 10^{-4}$	0.9679	0.00425250	0.06805 2.9533.10 ⁻⁵	$4.725 \cdot 10^{-4}$	9.868 · 10 ⁻⁷	1.00000
	^{13.} Dynes/cm ² . 980.62	2492.0 29890.0	1333.3 13333.0	33865.0 980.62	980620.0	6.6064	68950.0 29.93	478.9	1.0	1013295.0
nued.	12. Lbs./ft ² . 2.048	5.205 62.43	2.785 27.85	70.73 2.048	2048.0	2.000	$144.00 \\ 0.06250$	1.000	0^{-2} 2.088 · 10 ⁻³	2116.37
SIUNS-Conti	11. Oz./ft2. 32.77	83.23 998.8	44.56 445.6	1131.7 32.770	32770.0	144.0	2304.2 1.0000	16.000	$3.3410 \cdot 10^{-2}$	33861.9
CONVERS	10.142 Lbs./in ² .	0.036125 0.4335	0.01934 0.1934	0.4912 0.014223	14.223	0.0200.0	1.0000 4.340	0.006944	1.4504	14.697
FRESSURI	$02./in^{2}$. 0.2276	0.5780 6.937	0.3094 3.094	7.860 0.2276	227.6	0000T	16.000 6.946	$.10^{-3}$ 0.11112	2.3208	235.152
	${ m Kgm./gm^2.}0.001000$	0.002540 0.03048	0.0013595 0.013595	0.03454 0.001	1.0000	4.034 . 10^{-3}	0.07032 3.052	4.885 10^{-5} 10^{-4})-3 1.019 10-6	1.03329
	$\frac{7}{1.0000}$	2.540 30.48	1.3595 13.595	34.54 1.000	0.000	4.034	70.32 0.03052	0.4885	1.0197 . 10	033.29
1	1	2	45	6	81		10	12	13	141

PRESSURE CONVERSIONS—Continued.

V COMPARATIVE TEMPERATURE DEGREES.

	Degrees	Degrees	Degrees	Degrees Reaumur.
Degrees Absolute		1.0	%	4/5
Degrees Centigrade	. 1.0	1.0	95	4/5
Degrees Fahrenheit	. 5%	5%	1.0	1/9
Degrees Reaumur	. 5⁄4	5/4	%	1.0

COMPARATIVE TEMPERATURE POINTS. Absolute zero= -273° Centigrade= -459.4° Fahr.= -218.4° Reaum. Freezing water = 0° C. = 273° A. = 32° F. = 0° R. Boiling water = 100° C. = 373° A.= 212° F. = 80° R.

HEAT QUANTITY CONVERSION FACTORS. One British Thermal Unit = $251.995 \times \text{calories}$ (gm.) = $0.251995 \times$ Cal. Large.

One gram caloric = 0.00396832 British Thermal Units.

One B. T. U. per pound = % calorie per gram.

One calorie per gram = 1.8 B. T. U. per pound. TIME CONVERSION FACTORS.

= 365 days, 5 hours, 48 minutes, 48 seconds = 12 calendar One year months.

= 52.1693 + weeks = 8765.8133 + hrs. = 525948.8 minutes= 31556928 seconds.

One week 7 days = 168 hrs. = 10080 minutes = 604800 seconds.One day = 24 hours = 1440 minutes = 86400 seconds. One hour = 60 minutes = 3600 seconds. One minute = 60 seconds.

VELOCITY CONVERSION FACTORS

	Mi./hr.	Ft./sec.	Km./hr.	M/sec.	Mi./da.	Km./da.
	1.	2.	3.	4.	5.	6.
		1.4667	1.6093	0.44704	24.00	38.62
2.	Feet per second0.6819	1.0000	1.0973	0.30480	16.37	26.33
3.	Kilometers/hour .0.6214	0.9114	1.0000	0.2778	14.913	24.00
4.	Meters per second.2.237	3.281	3.600	1.0000	53.69	86.40
5.	Miles per day0.04167	0.06112	0.06706	0.01863	1.0000	1.609
6.	Kilometers/day 0.02589	0.03797	0.04167	0.01157	0.6214	1.0000

CONVERSION FACTORS FOR MONEY.

	\$ to A.	А.		A. to \$.
	1.000	Dollar (U. S.)		1.000
$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	100.000	Cent (U. S.)		0.010
$\begin{array}{ccccccc} (Sovereign) & & & & & & & & & & & & & & & & & & &$				
$ \begin{array}{lllllllllllllllllllllllllllllll$	0.2055		= 20 shillings	4.8665
$ \begin{array}{lllllllllllllllllllllllllllllll$		(Sovereign)		
$ \begin{array}{rrrr} 163.72 & Farthing &= \frac{1}{4} \ penny & 0.00507 \\ 0.822 & Crown &= 5 \ shillings & 1.21660 \\ 4.200 & Mark \ (Germany) &= 100 \ pfennigs & 0.238 \\ 420.0 & Pfennig & 0.00238 \\ 5.182 & Franc \ (France) &= 100 \ centimes & 0.193 \\ \end{array} $	4.11	Shilling (s)	=12 pence	0.24331
0.822 Crown =5 shillings 1.21660 4.200 Mark (Germany) =100 pfennigs 0.238 420.0 Pfennig 0.00238 0.00238 5.182 Franc (France) =100 centimes 0.193	40.93	Penny (d)	=4 farthings	0.02028
4.200 Mark (Germany) = 100 pfennigs 0.238 420.0 Pfennig 0.00238 5.182 Franc (France) = 100 centimes 0.193	163.72	Farthing	$= \frac{1}{4}$ penny	0.00507
420.0 Pfennig 0.00238 5.182 Franc (France) = 100 centimes 0.193	0.822	Crown	=5 shillings	1.21660
5.182 Franc (France) $= 100$ centimes 0.193	4.200	Mark (Germany)	=100 pfennigs	0.238
	420.0	Pfennig		0.00238
518.2 Centime 0.00193	5.182	Franc (France)	=100 centimes	0.193
	518.2	Centime		0.00193

CLASSIFICATION OF U.S. PATENTS ON PETROLEUM RE-FINING.

- A. Water separation, dehydration, de-emulsification, heating and physical purification of oil and bottom settlings.
- B. Cracking, conversion, and decomposition processes.
- Paraffin and wax. C.
- Chemical treatment of petroleum. D
 - Acid or alkali. 1
 - 2. Other than acid or alkali.
- E. Asphalt.
 - 1. Compositions.
 - 2. Production.
 - 3. Refining.
- Simple distillation. F.
 - 1. Fire.
 - 2. Steam.
 - 3 Gas.
 - 4. Air.
 - Vacuum. 5.

Batch. I.

- II Continuous.
- Coal oil, Kerosene and Illuminating oils. G.
- H. Oil-fire prevention, extinction and storage.
- I. Recovery of acid-sludge and alkali-sludge.
- J. Gasoline production and treatment.
- K. Gas.
 - 1. Production.
 - 2. Treatment.
 - Production of carbon black. 3.
- L.
- Chemical products. Patented blends and compounds. M.
- N. Testing apparatus.
- 0. Lubricating oils.
- Ρ. Electrical processes.
- Q. Transporting oil.
- Ř. Methods of removing carbon and coke.
- Mechanical appliances in oil refining, and processes. S. (Not covering any particular operation.)
- Τ. Plastics.
- U. Condensers and condensing.
- V. Desulphurizing and deodorizing.
- W. Oil shales, oil sands and coals.

UNITED STATES PETROLE		
Name Number	Date	Class
Aab., Geo. & S. K. Campbell 369,902	Sept. 13, 1887	C
Adair, James	June 10, 1862	U
Adair, Jas. & Tweddle, H. W. C 56,343	July 17, 1866	F
Adams, Chas	Aug. 4, 1914 Fep. 13, 1866	A C
Adams J H 976 975	Feb. 13, 1866 Nov. 29, 1910	B
Adams, Henry W	Apr. 3, 1855	Ö
Adams, J. H. 976,975 Adams, Henry W. 12,614 Adamson, Wm. 45,007	Nov. 15, 1864	Ď1
Adiassewich, Alexander 629,536	July 25, 1899	F
Adamson, wm. 43,007 Adiassewich, Alexander 629,536 Alberger, J. L. 37,798 Alexander, Clive M. 1,230,975 Alexander, Jas. H. 229,297 Alexander, Jas. H. & Eberhard 156,265 Alexander, Robt. 435,198 Alkemade, J. von R. 1,007,600 Allen Ceorge 186,265	Mar. 3, 1863	G, B
Alexander, Clive M	June 26, 1917	B
Alexander, Jas. H	June 29, 1880 Oct. 27, 1874	F
Alexander Robt 425 108	Oct. 27, 1874 Aug. 26, 1890	F E 3
Alkemade, J. von R	Oct. 14, 1913	č
Allen, George 182,625	Sept. 26, 1876	Ă, O
Allan, D. M., Jr	June 20, 1916	D1
Allen, George. 182,625 Allan, D. M., Jr. 182,625 Alter, David & Hill, S. A 20,026 Alvord, Clark,	Apr. 27, 1858	F
Alvord, Clark	Mar. 11, 1879 Jan. 8, 1918 Dec. 22, 1903 Aug. 9, 1892 Aug. 9, 1892	R
Ambruson, H. J	Jan. 8, 1918	K1
Amend, Otto P. 747,348 Amend, Otto P. 480,312 Amend, Otto P. 480,312 Amend, Otto P. 480,311	Dec. 22, 1903 Aug. 9, 1892	D1,V VD1 P
Amend, Otto P 480,311	Aug. 9, 1892	V, D 1, B V, D 1, B
Amend, Otto P 551,941	Dec. 24, 1895	V.DI
Amend, Otto P 601,331	Mar. 29, 1898	V, DI
Amend, Otto P. 551,941 Amend, Otto P. 601,331 Amend, Otto P. 601,331 Andrews, Saml. 58,197 Andrews, Saml. 69,745 Angus, H. R. 407,274 Anthony C. E. 690,882	Dec. 22, 1903	V, D 1, B V, D 1 V, D 1 V, D 1 V, D 1
Andrews, Saml 58,197	Sept. 25, 1866	P 1, 1
Andrews, Sami 69,740	Oct. 15, 1867 July 16, 1880	SF
Amend, Otto P. 480,311 Amend, Otto P. 551,1941 Amend, Otto P. 601,331 Amend, Otto P. 747,347 Andrews, Saml. 58,197 Andrews, Saml. 69,745 Angus, H. R. 407,274 Anthony, C. E. 620,082 Archebold, Geo. 503,028	July 16, 1880 Feb. 21, 1899	B, T
Archbold, Geo	Aug. 8, 1893	E'1
Archbold, Geo	Sept. 6, 1864	F
Archbold, Geo. 503,028 Archer, Wm. 44,137 Artmann, Carl. 1,031,227 Arvine, Freeling W. 629,059 Arvine, Freeling W. 431,795 Ash, Horace W. 779,197 Ash, Horace W. 757,387 Ash, Horace W. 757,387 Ashworth, A. 1,300,548 Andrews and Averill. 1,312,467 Atwood Luther. 27,767	July 2, 1912	E1
Arvine, Freeling W 629,059	July 18, 1899	A
Arvine, Freeling W 431,795	July 8, 1890	G, N
Ash Horace W	Jan. 3, 1905 Jan. 3, 1905	E 2, F E 2, F
Ash, Horace W	Apr. 12, 1904	F1
Ashworth, A. A	Apr. 15, 1919	ŝ
Andrews and Averill1,312,467	Apr. 10, 1860	S
	Oct. 19, 1858	F 2
Atwood, Luther	Dec. 28, 1858 Dec. 28, 1858	B
Atmood Tathan 00'407	Dec. 28, 1858 Feb. 22, 1859	B
Atwood, Luther. 22,407 Atwood, Luther. 23,337 Atwood, Luther. 28,246 Atwood, Luther. 28,246 Atwood, Luther. 28,448 Atwood, Luther. 27,768 Atwood, Luther. 31,858	Mar. 29, 1859	B
Atwood, Luther 23,337	May 15, 1860	Ĝ
Atwood, Luther 28,246	May 29 1860	G, B
Atwood, Luther 28,448	Apr. 10, 1860	F
Atwood, Luther	Mar. 26, 1861	U
Atwood, Luther. 31,858 Atwood, L. & Atwood, W. 15,506 Atwood, L. & Atwood, W. 15,505 Atwood, William. 226,151 Aubormon, Col. M. 570,888	Aug. 12, 1856 Aug. 12, 1856	B,2D 1 W, F
Atwood, L. & Atwood, W 15,505	Aug. 12, 1856 Apr. 6, 1880	G, P
Atwood, William 226,151	Apr. 20, 1901	Ă
Aukerman, Cal. M 572,882	Aug. 5, 1919	В
Bacon, Brooks & Clark1,131,309	Mar. 9, 1915	J, B
Bacon & Clark	June 23, 1914	B
Backhaus, Arthur A	July 2, 1918 July 2, 1918	M M
Backhaus, A. A.	Mar. 11, 1919	M
Barber, Guy M	Jan. 1, 1918	S
Atwood, William. 226,151 Aukerman, Cal. M. 572,882 Bacon, Brooks & Clark 1,181,809 Backhaus, Arthur A. 1,271,114 Backhaus, Arthur A. 1,271,115 Backhaus, Arthur A. 1,271,115 Backhaus, Arthur A. 1,251,952 Baillard, Chas. L 340,411 Baker, Leslie A. 299,611 Barnes, Wm. T. 24,920 Barnet, Michael. 59,531	Apr. 20, 1886	D1
Baker, Leslie A 299,611	June 3, 1884	A U
Barnes, Wm. T 24,920	Aug. 2, 1859 Aug. 2, 1859	
Barrett Michael 50 531	Aug. 2, 1859 Nov. 6, 1866	GI
Barron Thos J	Mar. 28, 1865	M
Barrett, Michael. 59,531 Barron. Thos. J. 46,987 Barnickel, W. S. 1,093,098	Apr. 14, 1914	A, D1
Barnickel, W. S1,223,659	Apr. 24, 1917	A, D1
Barnickel, W. S. 1,093,098 Barnickel, W. S. 1,223,659 Barnickel, W. S. 1,223,650 Barnickel, W. S. 1,223,650 Barnickel, W. S. 1,223,650 Bartickel, W. S. 1,223,650 Bartels, E. 1,115,887	Apr. 24, 1917	A
Bartels, E	Nov. 3, 1914	н

UNITED STATES F	'ETROLEUN	I PATENTS	S-Con.
Name	Number	Date	Class
Barstow, Frank Q Barthel, Peter. Baskerville, Chas. Bassett, R. D. Bates, H. F. Baum, E. P. Baynes, R. & Fearenside, J Bell, A. F. L. Bell, A. F. L.	181 814	Sept. 5, 187	6 C
Barthel Peter	135,879	Feb. 18, 187	3 E1.8
Baskerville Chas	1.231.985	July 3, 191	7 I
Bassett R D	1,120,669	Dec. 15, 191	4 J
Bassett, R. D.	1.120.670	Dec. 15, 191	
Bates. H. F.	1.046.541	Dec. 10, 191	
Baum, E. P	1.109.103	Sept. 1 191	4 A
Baynes, R. & Fearenside, J	299,324	May 27, 188	4 A 4 D 2
Bell, A. F. L	581,451	Apr. 27, 188	
Bell, A. F. L		Apr. 13, 189	07 E 3
Bell, A. F. L Bell, A. F. L	617,712	Jan. 17, 189	9 E 2, 8
Bell, A. F. L Bell, A. F. L Bell, A. F. L.	1,231,695	July 3, 191	7 B, R 0 E 2, 8
Bell, A. F. L.	655,430	Aug. 7, 190	0 E 2, 8
Bell, A. F. L.	000,410	Sept. 19, 189	3 E 2, 8
Benningrath, Leonaru, Jr	20,400	June 1, 18, July 25, 19 Apr. 9, 19 Mar. 20, 190 May 19, 189	8 F 1, 4
Bonham E B	1 969 576	Apr. 9, 19	1 A 18 K1
Barg Friedrich	645 743	Mar. 20, 19	0 F2,1
Barg Friedrich	560 463	May 19, 18	6 D1
Rerg F	736,479	Aug. 18, 190	NS VDI
Berg. F.	736,480	Aug. 18, 19	3 V, D 1 3 V
Berg, F.	623.066	Apr. 11, 189	0 D1
Berg. H. J.	93,952	Aug. 24, 180	9 F1,1
Bibby, John & Lapham, A	48,896	JULV 25 186	35 F1
Bicknell, John E	313,979	Mar. 17, 188	5 F 2
Bicknell, John E	400,042	Mar. 26, 188	
Bicknell, John E	400,043	Mar. 26, 188	9 C
Birge, Wm. H	175,014	Mar. 21, 187	
Blackmore, H. S.	486,554	Nov. 22, 189	2 U
Blackmore, H. S.	120,020	June 20, 190	5 V, D1
Blair, Joini B	150 007	June 10, 187	
Bell, A. F. L. Bell, A. F. L. Bell, A. F. L. Bellingrath, Leonard, Jr. Bending, Wm. P. Benham, E. B. Berg, Friedrich. Berg, F. Berg, F. Berg, F. Berg, F. Berg, F. Bibby, John & Lapham, A. Bicknell, John E. Bicknell, John E. Bicknell, John E. Bicknell, John E. Biackmore, H. S. Blackmore, H. S. Blair, John B. Blackmore, H. S. Blackmore, H. S. Blair, John B.	219,007	Feb. 16, 187 Feb. 24, 188	5 G
Boleg Friedrich	761 030	Feb. 24, 188 June 7, 190	
Boote, A. J. & Kittredge, H. G.	620.882	Mar. 14, 189	
Bower, Henry,	230.171	July 20, 188	
Boleg, Friedrich. Boleg, Friedrich. Bower, Henry. Beckley, R. E. Bending, Wm. P. Benham, E. B. Putchor, I. A.	1.127.722	Feb. 9, 191	
Bending, Wm. P	1,144,522	June 29, 191	
Benham, E. B. Butcher, J. A. Benhoff, Geo. F., Jr., & Jensen, J Benton, G. L. Berend, Ludwig. Black, J. C. Black, J. C. Black, J. C. Blowski, Jno. & Blowski, A. Borrman, C. H. Bowman, F. Brackebusch, Hans. Bradford, Geo. Braggins, Edw.	1,040,124	Oct. 1. 191	
Butcher, J. A.	1,311,753	July 29, 191	
Bennoff, Geo. F., Jr., & Jensen, J	. 0.1,181,564	May 2, 191	
Benton, G. L.	342,064	May 20, 180	
Borond Ludwig	1 167 979	May 25, 188 Jan. 11, 191	
Blacher L & Sztancel S	856 976	Jan. 11, 191 Apr. 26, 191	
Black J C	968 640	Aug. 30, 191	
Black, J. C.	1.152.478	Sept. 7, 191	5 F 3
Black, J. C	1.164.162	Dec. 14, 191	5 D 2, F 8
Blowski, Jno. & Blowski, A	1,186,373	June 6, 191	6 I
Born, Sidney	1,234,124	July 24, 191	7 F1, II, S
Borrman, C. H	1,220,067	Mar. 20, 191	7 F9 11
Bowman, F.	12,852	May 15, 188	
Brace, H. B. & Swart, Wm. T	54,495	May 8, 186	6 M, G
Brackebusch, Hans.	275,565	Apr. 10, 188	3 D1
Bradiord, Geo.	806,116	Nov. 21, 190	
Bragging Edw	004,010	May 24, 189	18 V, D1
Braggins, Edw Braun, Otto	46,633	Mar. 7, 186 June 28, 188	DO F D
Brainig Revera	206 807	June 28, 188 Oct. 21, 188	
Brooks Esser & Smith	1 101 016	Oct. 21, 188 July 18, 191	
Brooks and Smith.	1,231,123	June 26, 191	
Brickman, Saml,	1.279.506	Sept. 24, 191	8 F
Brown, Arthur L.	1.234.862	July 31, 191	
Brown, Ernest	1,225,569	May 8, 191	
Brown, D. P. & Neeley, J. W	361,671	Apr. 26, 188	7 F1.2
Brown, E. G. Cammann, O. N.	&		
Willcox, O.	510,672	Dec. 12, 189	3 F 1, 2, 4
Brown W A	994,100	May 30, 191	1 A
Braun, Otto. Breinig, Revere Brooks, Essex & Smith. Brooks and Smith. Brokman, Saml. Brown, Arthur L. Brown, D. P. & Neeley, J. W Brown, D. P. & Neeley, J. W Brown, E. G. Cammann, O. N. Willcox, O. Brown, L. W. Brown, W. A. Brown, W. A.	10 055	July 15, 191 Sept. 27, 185	9 A 3 C, W
	10,000	Nopt. 21, 180	0 C, N

UNITED STATES I EIROLEU	M FAIENIS-	-Con.
Name Number	Date	Class
Brownlee, R. H. & Uhlinger1,265,043	May 7, 1918	K 3, B
Brownlee, R. H1,308,161	July 1, 1919	F
Brucke, Otto 963,510	July 5, 1910	A
Brundred, Wm. J 148,806	Mar. 24, 1974	F 2
Bullard, John 34,195	Jan. 21, 1862 Jan. 16, 1872	G
Burcey, Chas. J. T 122,810	Jan. 16, 1872	F
Burch, Ell F	Aug. 28, 1917	0, T
Burdon, J., W. M. & M. M	Sept. 29, 1914	K1
Durgharut, C. A	Dec. 9, 1884	U
Durk, fl. R. Wright S 65 000	Sept. 11, 1883 June 25, 1867	G
Burket D M & Grav I C 57 985	June 25, 1867 Aug. 21, 1866	D 1 0
Brucke, Otto. 963,510 Brundred, Wm. J. 148,806 Bullard, John. 34,195 Burcey, Chas, J. T. 122,810 Burch, Eli F. 1,238,101 Burdon, J., W. M. & M. M. 1,112,051 Burghardt, C. A. 309,027 Burke, H. R. 284,811 Burke, A. M. & Wright, S. 65,999 Burrket, D. M. & Gray, J. C. 57,285 Burrows, H. G. 998,937 Burton, W. M. 1,049,667 Burton, W. M. 1,049,667 Burton, W. M. 1,105,961	July 25, 1911	F 2, II
Burton, W. M.	Mar. 11, 1913	B, E 2
Burton, W. M	Jan. 7, 1913	B, J
Burton, W. M	Aug. 4, 1914	J.B
Burton, W. M	Sept. 29, 1914	J, B C, B B
Burton, W. M1,167,884	Jan. 11, 1916	B'
Burton, W. M	Sept. 8, 1903 Dec. 19, 1882	V. D 1
Bush, Asa A 269,382	Dec. 19, 1882	1P. 1
Busse, Heinrich 376,289	Jan. 10, 1888	Т
Byerley, Francis X 347,288	Aug. 10, 1886	Ĉ, F
Byerley, F. X 524,120	Aug. 7, 194	E 2, 3, F
Byerley, F. X 547,329	Oct. 1, 1895	F 4, 2
Busse, Asa A. 209,382 Busse, Heinrich. 376,289 Byerley, Francis X. 347,288 Byerley, F. X. 524,120 Byerley, F. X. 544,329 Byerley, F. X. 244,431 Byerley, F. X. 132,353 Byerley, F. X. 164,672	Jan. 10, 1882 Jan. 10, 1888 Aug. 10, 1886 Aug. 7, 194 Oct. 1, 1895 July 19, 1881	CCC
Byerley, F. A 152,555	001. 22, 1014	C
Byerley, F. A 104,072	June 22, 1875	C
Diggins, James L	Aug. 6, 1918	B J
Boyle Aley M 1976 866	Aug. 13, 1918 Aug. 27, 1918	W
Buerger C B. 1 302 761	Aug. 27, 1918 May 6, 1919	s
Calkins, A. C	Jan. 3, 1905	B
Calkins, A. C	Sept. 6, 1904	
Campbell, Andrew 999,628	Aug. 1, 1911	ĉ
Cantour, David 552,206	Jan. 14, 1896	F
Carman, F. J 501,988	July 25, 1893	V
Carpenter, Calvin, Jr 82,083	Sept. 15, 1868	D 1 C F V O S
Byerley, F. X. $524,120$ Byerley, F. X. $547,329$ Byerley, F. X. $132,353$ Byerley, F. X. $132,353$ Byerley, F. X. $164,672$ Biggins, James E. $1.274,976$ Black, John C. $1.275,648$ Boyle, Alex. M. $1.276,866$ Buerger, C. B. $1.302,761$ Calkins, A. C. $779,398$ Calkins, A. C. $769,681$ Campbell, Andrew. $999,628$ Carnan, F. J. $501,988$ Carter, G. F. $680,639$ Catlin, Robert M. $1.272,707$ Cazin, Francis F. M. $400,634$ Chambelain, H. P. $1.21,790$ Chemin, Jean C. O. $297,766$ Cherry, C. $15,642$ Cherry, C. $15,642$ Cherry, C. $15,642$ Cherry, C. $12,9886$ Chesebrough, Robt. A. $127,568$ Chesebrough, Robt. A. $127,568$ Chesebrough, Robt. A. $230,239$ Cherery, C. $15,642$ Chery, K. $12,29,886$	Aug. 13, 1901 July 16, 1918	S
Catlin, Robert M		W
Cagin F M F	Apr. 2, 1889	F V, G
Chamberlain H P 1991 700	Apr. 2, 1889 Apr. 2, 1889 Apr. 3, 1917 Apr. 29, 1884	B, G
Chemin Jean C O 297766	Apr. 29, 1884	F, D
Cheney Samuel 230 239	July 20, 1880	F, D F 2
Cherry, Cummings 15.642	Sept. 2, 1856	Ā
Cherry, C 15.643	Sept. 2, 1856	A W
Cherry, L. B	June 12, 1917	B, P
Chesebrough, Robt. A 127,568	June 4, 1872	M
Chesebrough, Robt. A 237,484	Feb. 8, 1881	M
Chesebrough, R. A 49,502	Aug. 22, 1865	G, S
Chesebrough, R. A 48,367	June 27, 1865	S
	Dec. 19, 1865	S
Chesebrough, R. A 51,558	Dec. 19, 1865	S F 2, II
Chesebrough, R. A 542,704	Aug. 21, 1894 Aug. 30, 1870	F 2, II I
Childe Samuel	Aug. 30, 1870 June 13, 1854	F 1, 2, I
Clarke Edward 932685	Sept. 28, 1880	I I, 2, 1
Clark Edward M. 1119496	Dec. 1, 1914	B
Clark, E. M	Dec. 1, 1914 Feb. 16, 1915	B
Clark, E. M	Mar. 16, 1915	В
Clark, C. E1,147,608	July 20, 1915	K 1
Clark, Frank W 547,332	Oct. 1, 1895	F 3, 4
Clark, R. C. & Beecher, W. F 275,589	Apr. 10, 1883	F 1, 4
Clark, R. C. & Warren, M. H 298,825	May 20, 1884 May 26, 1885	F
Chesebrough, R. A. 542,704 Chevrier, Gervais. 106,915 Childs, Samuel. 11,059 Clark, E. Marther and M. 1,119,496 Clark, E. M. 1,129,034 Clark, E. M. 1,129,034 Clark, C. E. 1,147,608 Clark, Grank, C. E. 1,147,608 Clark, Frank W. 547,332 Clark, R. C. & Beecher, W. F. 275,589 Clark, R. C. & Warren, M. H. 298,825 Clark, R. C. & Warren, M. H. 318,698 Clark, R. G. & Warren, M. H. 34,816		F G, F 2, II
Clifford Victor 1966 407	Apr. 1, 1862 May 14, 1918	G, F 2, II H
Coast John W Jr 1950 708	Dec. 18, 1917	B
Clark, S. G	Dec. 18, 1917	В
Coast, J. W., Jr	Dec. 18, 1917	В

UNITED	STATES	PETROLEUM	A PATI	ENTS-	-Con.
Name		Number	Da	te	Class
Name Coast, J. W., Jr Coast, J. W., Jr Coast, J. W., Jr Coast, J. W., Jr Coast, J. W., Jr. Coast, John W., Jr. Coast, John W., Jr. Cobb, J. O. Cobb, E. B. Cochran, A. Cole, Jas., Jr. Coleman, John T. Colin, Theodore F. Colin, T. F.		1 252 401		8, 1918	В
Coast I W Ir		1 253 000	Jan.	8, 1918	B
Coast J W Jr		1,258,190	Mar.	5, 1918	B
Coast J. W. Jr.		1.252.999	Jan.	8, 1918	B
Coast, J. W., Jr.		1.291.414		4, 1919	B
Coast. J. W., Jr.,		1.307.724		4, 1919	ŝ
Coast. John W., Jr.		1,250,799		8, 1917	B
Coast, John W., Jr.		1,258,191	Mar.	5. 1918	B
Cobb, J. O		1,201,558	Oct. 1	7, 1916	A
Cobb, E. B		1,300,816	ADr.	5. 1919	D
Cochran, A		1,296,367	Mar.	4, 1919	В
Cole, Jas., Jr		128,169	Sept. 1	Z, 1876	F 2, 4, II
Coleman, John T		19,406	May 2	9, 1877	F
Colin, Theodore F	• • • • • • • • • • •	607,017	July 1	2, 1898	V, D 1
Colin, T. F Colin, T. F		723,368	Mar. 2	4, 1903	V, D
Colin, T. F Colin, T. F		744,720 685,907	Nov. 2	4, 1903	V, D
Colling Tacob		1 020 420	Nov. June	5, 1901	V, D
Colling John F	• • • • • • • • • • • •	50 224	Oct. 3	4, 1912	A
Colling Jos' G		32 557		0, 1866	F 4, I
Connelly Martin		240.093		8, 1861 2, 1881	S
Connelly, Martin		240,094		2, 1881	D 1, V D 1, V
Cook & Price		1.190,633	July 1	1, 1916	E 3
Cooper, A. S		67.226	Jan.	3. 1899	E 2, 3
Corfield, Wm		54,061	Apr. 1	7 1866	M 2, 0
Corfield, Wm		54,060	Apr. 1	7. 1866	M
Cornell, Sidney		1,202,969	UCL. 3	51.1916	F 2
Cosden, J. S		981,176	Jan. 1	9, 1911	F 2, 11
Cosden, J. S. & Coa	st, J. W., J.	r 258,196	Mar.	5 1918	В
Cosden & Coast		$\dots 1,261,215$	Apr.	2. 1918	В
Cottrell & Wright		987,117	Mar. 2	1. 1911	P
Cottrell & Speed		987,115	Mar. 2	1. 1911	\mathbf{P}, \mathbf{A}
Cottrell & Speed		007 114	Mar. 2 Mar. 2	1, 1911	P
Courtois F A	• • • • • • • • • • • •	788 950	Apr. 2	1, 1911 5, 1905	P
Cowan Wm P	••••••	558 358	Apr. 1	5, 1905 4, 1896	Ñ C
Crane Frederick T)	1 223 153	Apr. 1	7, 1917	M, D
Crane, Gerard		231.280	AUS 1	7 1890	E 1
Crane, Adolphus G.		1.276.879	Aug. 2	7, 1918	Bu t
Crawford, Benjami	in	113,023	Mar. 2	8, 1871	3
Crocker, Saml. H		120,463	July 1	6, 1872	B
Cronemeyer, A. H.		718,318	Jan. 1	3, 1903	1
Cronenberger, W. M	1	1,152,399	Sept.	7, 1915	14
Cronin, C. J.		150,465	may	5, 1874	F
Cross, James P	•••••	1 955 199	Aug. 1	4, 1866	M
Cross, Roy	• • • • • • • • • • • •	1 009 919	Feb. Oct.	5, 1918 3, 1916	F 1, 2
Culmer Geo & Geo	CK	635 490		3, 1916 4, 1899	H
Culmer Geo & Geo	CK	635 430		4, 1899	F. W
Culmer, J. W.		217.995	July 2	9, 1879	G
Cunningham, Chris	topher	158.042	Dec. 2	2. 1874	C
Danckwardt, P		1,141,529	June	$\begin{array}{c} 2, \ 1874 \\ 1, \ 1915 \end{array}$	J. F 1, II
Daul, John		213,395	Mar.	8, 1879	F 2
Daul, Louis		258,284	May 2	23. 1882	\tilde{F} $\tilde{2}$ P
Davidson, J. G. & F	ord, R. W	1,228,042	June	5 1017	Р
Davidson, Samuel .		1,238,644	Aug. 2	18. 1917	J. K 2 F 1, II
Davis, John T		1 150 196	Apr. Nov.	2, 1901	F 1, 11
Davis, John 1		65 884	June 1	2, 1915 2, 1915 8, 1867	F 2, 11
Dav David T		826 089	July	8, 1867 7, 1906	S A, V
Day, D. T.			Oct.	2 1011	A, V B
Day, D. T			Apr.	3. 1917	B, D
Day, D. T		826,089	July 1	17, 1906	B, D V. D W
Day, D. T		1,280,178	Oct.	1. 1918	W
Day, Roland B		1,280,179	Oct.	1, 1918	R
Dayton, W. C			Mar. 1	14, 1916	K 1 K 1
Dayton, W. C	•••••	1,174,970	Mar. 1 Dec. 2	4, 1913	K 1 K 1 F 2, II
Colin, T. F Colin, T. F Colin, T. F Colins, Jacob Colins, John F Collins, Jos, G Collins, Jos, G Collins, Jos, G Collins, Jos, G Connelly, Martin Cook & Price Cooper, A. S Corfield, Wm Corfield, Wm Cosden, J. S. & Coa Cosden, F. C Cottrell & Wright Cottrell & Speed Cottrell & Speed Cottrell & Speed Cottrell & Speed Cottrell, F. G Courois, F. A. Courois, F. A. Courois, F. A. Crane, Frederick I Crane, Gerard Crane, Gerard Crawford, Benjami Crocker, Saml H. Cronemberger, W. M. Cronein, C. J Cross, James P. Cross, Roy Cross, James P. Cross, Valter M. Crons, Samuel M. Cronenberger, W. M. Cronin, C. J Culmer, Geo. & Geo Culmer, J. W Cunningham, Chris Danckwardt, P. Daul, John Davidson, Samuel Davis, John T. Davis, John T. Davis, John T. Davis, Samuel Day, D. T. Day, N. C. Day, O. T. Day, D. T. Day, N. C. Dean, Richard Dean, Richard Dean, Richard		305,056		25, 1883 6, 1884	F 2, II F 1, 2, II
Dean, Richard Dean, Richard			Jan.	6, 1885	F 1, 2, II F
				-, 1000	1

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UNITED STATES FEIRUI		IENIS-	-Con.
Name Nun	nber I	Date	Class
Dean, R 314	,368 Mar.	24, 1885	F 1, 2, 3, II
Dean, R	,500 May	25, 1886	F 2, II
Dehnst, Julius1,112	,602 Oct.	6, 1914	V. D
De Smedt, Edw. J 236	,995 Jan.	25, 1881	V, D E 1, 2
Dean, R. 314 Dean, R. 342 Dehnst, Julius 1,112 De Smedt, Edw. J. 236 De Smedt, E. J. 237 Devericks, F. C. 1,260 Dewar & Redwood, B. 415 Dewar & Redwood, B. 426 Dewar & Hedwood 426	,662 Feb.	8, 1881	E 1, 2
Devericks, F. C1,260	,970 Mar.	26, 1918	J. K 2
Dewar, J. & Redwood, B 419	,931 Jan.	21, 1890	В
Dewar & Redwood 426	,173 Apr.	22, 1890	В
Dewitt, menty Construction of the	1499 Mal.	26, 1867	M
	,096 Aug.	23, 1881	M
Devine, S. R. & Seely, C. A 30	,071 May	29, 1866	F 2
Dickey, Julius C. 160 Dichl, H. A. 469 Deiterichs, E. F. 253 Divine, R. E. 1,300 Divine, R. E. 1,300	,349 Aug. ,777 Mar.	3, 1875 1, 1892	F1
Deitoriche F F 253	.990 Feb.		E 2, 3 F 1, 2
Divine R E	.779 May	$ \begin{array}{c} 21, \ 1882 \\ 13, \ 1919 \end{array} $	I I, 2
Divine B E	.662 Apr.	22, 1919	İ
Divine, R. E	,663 Apr.	22, 1919	Î
Divine, R. E. 1,303 Divine, R. E. 1,303 Divine, R. E. 1,303 Doe, Wm. 174	,789 Mar.	14, 1876	ŝ
Dow. Allan W 688	,073 Dec.	3, 1901	E 1, 2, B
Downard, J. S. & Roloson, B. A 722	,500 Mar.	10, 1903	E 2
Downer, Wm. P 44	,519 Oct.	4, 1864	D 1
Drake, Thos 471	,863 Mar.	29, 1892	L
Draper, Henry V. P 238	,867 Mar.	15, 1881	G, D
Drayton, Thos	,239 July	4, 1854	D
Dubbs, Henry 161	,672 Apr.	6, 1875	D, S
Dubbs, Jesse A 470	,911 Mar.	15, 1892	V
Doe, Wm. 174 Dow, Allan W. 688 Downard, J. S. & Roloson, B. A. 722 Downer, Wm. P. 44 Drake, Thos. 471 Draper, Henry V. P. 238 Drayton, Thos. 11 Dubbs, Jesse A. 470 Dubbs, J. A. 1,005 Dubbs, J. A. 1,005 Dubbs, J. A. 1,005	639 Apr.	3, 1900	F 2. 4
Dubba I A	570 Sept.	5, 1911 23, 1914	A, F B
Dubbs, J. A	.227 Mar.	25, 1913	E 2
Dubbs, J. A	.621 Mar.	4, 1902	F 4, II
		4 1009	F 4
Dubbs, J. A. 407	.182 July	16, 1889	V, D
Dubbs, J. A. 407 Dubbs, J. A. 1,123 Dubbs, J. A. 1,135	,502 Jan.	5, 1915	A
Dubbs, J. A	.506 Apr.	13, 1915	E 2, B
Dubbs, S. P. 1,231 Dubbs, C. P. 1,231 Dubbs, C. P. 1,231 Dubler, John B. 251 Dubreuil, A. 283 Dubreuil, A. 48 Dubreuil, G. H. S. 46 <td>,509 June</td> <td>26, 1917</td> <td>B , _</td>	,509 June	26, 1917	B , _
Dubbs, C. P1,231	,509 June	26, 1917	В
Dubler, John B 251	,770 Jan.	3, 1882	F
Dubler, J. B 283	,471 Aug.	21. 1883	F 1, II
Dubrey II. A. 48 Duffus, G. H. S. 46 Duffus, C. H. S. 46	,265 June	20, 1865	F 2
Duffus, G. H. S 46	,088 Jan.	31, 1865	F. S
Duffus, G. H. S 46	,089 Jan.	31, 1865	F, S
Duffus, G. H. S 46	,090 Jan.	31, 1865	F, S
Dundas, R. C	,980 Mar. .039 Dec.	25, 1913 8, 1914	E 2, B F 1, II
Dundas, R. C	.199 Feb.	19, 1918	г 1, 11 В
Dunham F H 1009	.040 Sept.	12, 1911	Ē
Dunham F H 1013	.283 Jan.	2, 1912	E 2
Dunkle Allen H	.300 Dec.	4, 1894	Ū
Dunscomb. Edward 62	.739 Mar.	12, 1867	ŝ
Dupias, A. C. G. & Fell, W. S 749	.368 Jan.	12, 1904	F. S
Durant, C. W. & Griffith, J 132	,263 Oct.	15, 1872	U
Dvorkovitz, Paul 546	,697 Sept.	24, 1895	F 2
Dyar, N. A. & Augustus, J. F 23	,362 Sept.	6, 1859.	M
Dyer, E. I	,381 Dec.	5, 1916	A
Dyer, E. I),504 Mar.	27. 1917	A
Dyer, E. I. & Heise, A. R1,24	2,784 Oct.	9, 1917	A
Dyer, Frank L	9,360 Mar.	23, 1897	F 2, 5 D
Dyer, Walter	6,535 Feb.	19, 1918 19, 1918	D
Earlo C W 1991	6,536 Feb. .038 Apr.	3, 1917	H
Eaton Richard	.638 Jan.	3, 1871	0
Edeleanu Lazar	.553 Feb.	2, 1909	Ď
Edgerton, Henry H.,	,655 Feb.	9, 1875	K 1
Edwards, E. A	.745 Nov.	4, 1890	K 1 F 2, 4, II F 2
Edwards, Jos. B 100	,874 Mar.	15, 1870	F 2
Edwards, Jos. B	,884 Sept.	3. 1918	B
Duffus, G. H. S. 46 Duffus, G. H. S. 46 Duffus, G. H. S. 46 Dundas, R. C. 1,056 Dundas, R. C. 1,257 Dunham, F. H. 1,003 Dunkle, Allen H. 530 Dunkke, Allen H. 530 Dunkas, A. C. G. & Fell, W. S. 749 Durant, C. W. & Griffith, J. 192 Dvorkovitz, Paul 540 Dyer, E. I. 1,200 Dyer, E. I. 1,200 Dyer, Frank L. 540 Dyer, Frank L. 541 Dyer, Walter 1,250 Dyer, Walter & W. E. 1,251 Dyer, Walter & W. E. 1,251 Dyer, Walter & W. E. 1,251 Eaton, Richard 111 Edeleanu, Lazar 91 Edwards, Jos. B. 100 Edwards, Jos. B. 100 Edwards, Jos. B. 100 Edwards, Jos. B. 1,271 Egeleston, J. E. 1,018 Eldred, B. E. & Mersereau, G. 1,234	,040 Feb.	20, 1912	F, V
Eldred, B. E. & Mersereau, G1,234	,886 July	31, 1917	В

443

UNITED STATES	PETROLEUM	I PATENTS-	-Con.
Name Elliott, W. S Ellis, Carleton Ellis, Carleton Ellis, C. Ellis, C. Ellis, C. Ellis, C. Ellis, C. Ellis, J. Encore, F. F. Engle, Jacob P. Engle, Jacob P. Engle, Jacob P. Engle, J. P. Errekson, Emil T. Erwin, J. B. & O. R. Evans, Edward Everest, H. B. Everest, J. S. Fales, L. S. Everest, L. S.	Number	Date	Class
Elliott, W. S.		Oct. 9, 1917	A, D
Ellis, Carleton	1,089,359	Mar. 3. 1914	Ö , D
Ellis, Carleton	1,191,880	JULY 10, 1910	Ď, L
Ellis, C.	$\dots 1,216,971$	Feb. 20, 1917	B
Eillis, C	1 205 225	Dec. 4, 1917	J, В
Ellis Ino & Kattell E C	63 789	Feb. 25, 1919 Apr. 16, 1867	B
Ellis & Kattell	68.860	Sept. 17, 1867	F 2, 11 F 2, 11
Ellithorp, S. B.	52,277	Jan. 30, 1866	Ū ", "
Emory, F. F	1,148,834	Aug. 3, 1915	S
Engle, Jacob P	481,391	Aug. 23, 1892	A
Engle, J. P	1 281 320	Aug. 23, 1892 Oct. 15, 1918	A W
Erwin, J. B. & O. R.	1.085.805	Feb. 3, 1914	H
Eva. Gray & Christy	1,100,126	June 16, 1914	Ö
Evans, Edward	1,257,829	Feb. 26, 1918	v
Everest, H. B	212,914	Mar. 4, 1879	F
Everest, H. B.	68,426	Sept. 3, 1867	F 2, 5, 11
Ewing, Chas. R	56 852	Jan. 13, 1914 July 31, 1866	$ \begin{array}{c} \mathbf{F} & 2, \ 5, \ 11 \\ \mathbf{F} & 2, \ 5, \ 11 \\ \mathbf{F} & 2, \ 5, \ 11 \end{array} $
Ewing, M. P. & Everest, H. B.	58.021	Sept. 11, 1866	F 2, 5, II F 2, 5, II
Fagan, John G.	1,148,763	Aug. 3 1915	H ^{2, 5, 11}
Fairchild, J. H	53,528	Mar. 21. 1800	$\overline{\overline{U}}_{S}$
Fales, Levi S	49,740	Sept. 5, 1865	S
Fales, L. S.	52,151	Jan. 23, 1866	F, U
Fales, L. S Fales, L. S	97 182	Sept. 5, 1865 Nov. 23, 1869	F 4, I I
Farrar, Alonzo	96.097	Oct. 26, 1869	D
Farrar, A.	100,876	Mar. 15, 1870	T
Faucett, H. W. & McGowan, T	133,426	Nov. 26, 1872	S, F
Faucett & McGowan	133,425	Nov. 26, 1872	S, F S, D U
Fales, L. S. Farrar, Alonzo Farrar, A. Faucett, H. W. & McGowan, T Faucett & McGowan Faucett & McGowan Fazi, Romolo de Felizat, Louis Felton, D. F. Farrar, F. F. & Gill, F. P. Fichet, L. V. Field, John K. Fleming, J. C. Fleury, Huot	1 108 251	Aug. 8, 1871 Aug. 25, 1914	U
Fazi, Romolo de	1 070,435	Aug. 19, 1913	M D
Felton, D. F.		Apr. 11 1916	K 1
Farrar, F. F. & Gill, F. P	206,309	July 23, 1878	I
Fichet, L. V	53,964	Apr. 17, 1866	F 2, 11
Field, John K	408,472	Aug. 6, 1889 Apr. 26, 1910	DÍ
Fleury Huot	50,571	Oct. 24, 1865	S, A F 5, D
Flowers, Geo. W., Happersett	. J. C. &		r 0, D
Fleming, J. C. Fleury, Huot Flowers, Geo. W., Happersett Happersett, D. W. Fordied, John Forward, C. B. Forward, C. B.	74,756	Feb. 25, 1868	м
Fordied, John	54,267	Apr. 24, 1866	W, D 1
Forrest, Chas. N.	1 1 80 082	Dec. 7, 1915	E 1, 3
Forward, C. B.	11 181 301	June 27, 1916 May 2, 1916	B, J
Forward, C. B.		May 2, 1916 Oct. 31, 1916	F 2, 11 B
Forward, C. B	998,569	July 18, 1911	E 2, B
Forward, C. B	1,100,966	June 23, 1914	В
Forward, C. B.	1,088,693	Mar. 3, 1914	В
Forward C B	1 947 808	Mar. 3, 1914 Nov. 27, 1917	E 2, B
Forward, C. B.	1.255.149	Nov. 27, 1917 Feb. 5, 1918	U B
Forward, C. B	1,274,405	Aug. 6, 1918	B
Forward, C. B	1,299,449	Apr. 8, 1919	F
Forward, C. B. & Davidson, J. M	1 611,620	Oct. 4, 1898	E 2, 3, D
Foubert, Andre	71,156	Nov. 19, 1867	F 2
Foubert Andre	60 166	Aug. 29, 1871 Dec. 4, 1866	F F 1
Fowler. David W	75.147	Mar. 3 1868	F 1 M
Forward, C. B. Forward, C. B. Forward, C. B. Forward, C. B. Foubert, Andre Foubert, Andre Foubert, Andre Fowler, David W. Franke, A. H. Frasch, Hans A. Frasch, Hans A.	1,142,512	June 8, 1915	A
Frasch, Hans A	488,628	Dec. 27, 1892	I
Frasch, Hans A	640,292	Jan. 2. 1900	F 2, 11
Frasch Hans A	581.546	Sept. 11, 1894 Apr. 27, 1897	D 1 E 2.3
Frasch, Hans A.	1.212.620	Jan. 16, 1917	E 2, 3 B
Frasch, Hans A. Frasch, Hans A. Frasch, Hans A. Frasch, Hans A.	845,735	Feb. 26, 1907	F 2, II
Frasch. Herman	908.700 .	Aug. 30, 1910	F 1
Frasch, Herman	487,216	Nov. 29, 1892	v

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UNITED STATES	PETROLEUM	PATENTS-	-Con.
Name	Number	Date	Class
Frasch, Herman	564,920	July 28, 1896	v
Frasch, Herman	490,144	Jan. 17, 1893	v
Frasch, Herman		Jan. 14, 1896	S
Frasch, Herman		June 2, 1896	S D 1 V V
Frasch, Herman		JUIV 28, 1896	V
Frasch, Herman		Mar. 17, 1891	V
Frasch, Herman		Feb. 21, 1888	V, D
Flasch, Herman	951,729	Mar. 8, 1910	V, D G, D
Frasch, Herman	951,272	Mar. 8, 1910	G, D
Frasch, Herman		Apr. 11, 1899	V, D G, D G, D V F 2, 4 V V V, D U F
Frasch, Herman	190,483	May 8, 1877	F 2, 4
Frasch, Herman	630,496	Aug. 8, 1899	V
Frasch, Herman	500,252	June 27, 1893	V
Frasch, Herman	572,676	Dec. 8, 1896	V. D
Frasch, Herman	321,420	Aug. 24, 1880	U
Frasch, Herman		Aug. 3, 1878	F
Frasch, Herman	649.047	May 8, 1900	O. V
Frasch, Herman	340,499 487,119 281,045	Apr. 20, 1886	F V
Frasch, Herman	487.119	Apr. 20, 1886 Nov. 29, 1892	v
Frasch, Herman	281.045	July 10, 1883	F 2, 3
Frasch, Herman	564,922	July 28, 1896	v ,
Frasch, Herman		July 28, 1896	V
Frasch, Herman	564,924	July 28, 1896	Ý, F V, D V, D 1
Frasch, Herman	649,048	May 8, 1900	V, D
Freach Hormon		July 16, 1895	V. D 1
Frasch, Herman Fraser, William M. Fraser, Wm. M. Fraderici, C. F.	542 610	July 30, 1895	V, D I
Fraser, William M	1 950 993	Mar. 12, 1918	E 1, 2
Frager Wm M	1 959 109	Mar. 5, 1918	
Fraser, Wm. M Frederici, C. F	10 670		E 1, 2
Freel, John	504,917	July 11, 1865 Sept. 12, 1893	S, F
Cargin Dichard	110.950	Aug 00 1071	F S, F D 2, V
Callementhy Denjamin	1 004 007	Aug. 22, 1871	D 2, V F 2, II
Gallowno T T	1,204,321	July 24, 1917	F 2, II W
Ganoupe, J. H.	1,283,723	Nov. 5, 1918	
Freel, John Gaggin, Richard Gallsworthy, Benjamin Galloupe, J. H. Gardner, J. & Harris, J. F Garner, J. B. & Clayton, H. D. Garner, J. B. Garrity, W. F. & Jarvis, A. Gravey, Benjamin Gathmann, Louis Gathmann, Louis Gay, Cassius M. Gearing, C. M. Gellen, A.	442,802	Dec. 15, 1890	V, F
Garner, J. B. & Clayton, H. D.	1,202,709	Apr. 16, 1918	L
Garner, J. B.	1,299,400	Apr. 8, 1919	J, K
Garrity, W. F. & Jarvis, A	1,190,538	July 11, 1916	0, A
Gravey, Benjamin	29,218	July 17, 1860	G
Gathmann, Louis	768,796	Aug. 30, 1904	F 1, 5
Gainmann, Louis	755,760	Mar. 29, 1904	F
Gay, Cassius M.	1,179,001	Apr. 11, 1916	J
Gearing, C. M	212,084	Feb. 4, 1879	F 1, II
Gellen, A. Gengembre, H. P. Gengembre, H. P.	1,063,025	May 27, 1813	I
Gengembre, H. P	52,283	Jan. 30, 1866	A
Gengembre, H. P	52,284	Jan. 30, 1866	A
Gengembre, H. P	24,454	June 21, 1859	G
Gengembre, H. P	25,109	Aug. 16, 1859	G, B
Gengembre, H. P	27,542	Mar. 20, 1860	G, W
Gengembre, H. P. Gengembre, H. P. Gerbeth, F. L. de Gesner, Abraham Gesner, A	33,699	Nov. 12, 1861	G
Gerbeth, F. L. de	81,071		L, P
Gesner, Abraham	11,205 .	111ne 27 1x54	G
Gesner, A	11,203	June 27, 1854	Ĝ G
Gesner, A	11,204	June 27, 1854	G
Gibbons, Samuel	87,485	Mar. 2. 1869	0
Gesner, A. Gibbons, Samuel Gibbons, S.	87.658	Mar. 9, 1869	F
Gibbons, S	85,810	Jan. 12 1869	F 2, II
Cibbong C	00.074	Sept. 17, 1867	F 1 9 II
Gillespie. Jas	23,362	Mar. 29, 1859	G, F
Gillons, G. H.	1.084.080	Jan. 13, 1914	F 1, 11
Gillespie, Jas. Gillespie, Jas. Goldwater, Henry Goldwater, Henry Goodaire, Wm. & Stead, Geo. Gordon, Thos. Govers, F. X. Gracie, John	366.720	July 19, 1887	F 1. 2. II
Goldwater, Henry	432.525	July 22, 1890	S I
Goodaire, Wm, & Stead, Geo	101.003	Mar. 22, 1870	I
Gordon, Thos.	451.724	May 5, 1891	D
Govers, F. X.	1.297.833	Mar. 18, 1919	W
Gracie, John	114,802	May 16, 1871	F 4
Gracie, John	114,802 114,803	May 16, 1871	F
Gracie, John	117,405	July 25, 1871	F
Gracie, John	117,406	July 25, 1871	F 1, I
Gracie John	99,081	Jan. 25, 1870	F 1, II
Gracie, John Grady, Chas. F	556,412	Mar. 17, 1896	F 1, II F 2, II

UNITED ST.	ATES PET	ROLEUM	PATEN	TS-C	on.
Name		Number	Date		Class
			uly 7,	1903 D	
Grannis C W		36 403 5	Sept. 9,	1862 U	
Grant, Jas. B.			Aug. 21,	1866 F	, 4
Graham, C. B Grannis, C. W Grant, Jas. B Grant, J. B. & Mason, A		220 545	Apr. 6.	1886 F	1, 2, 5, II
Grant & Mason		339,546 653,235	Apr. 6,	1886 F 1900 C 1881 C	1, 2, 5, II 2, 5. 1
Gray, A. McD		653,235	July 10,	1900 C	
Gray, Daniel T.		250,524	Dec. 6,	1881 C	
Grant & Mason Gray, A. McD Gray, Daniel T Gray, D. T.		248,735	Oct. 25, July 17,	1881 C 1883 C	a
Gray, D. I		1 005 495 6	July 17, Oct. 10,	1883 C 1911 I	, S
Gray, D. T. Gray, D. T. Gray, E. B. Gray, G. W. Grant & Mason Grant, H. F. Gray, G. W.		1,193,540	Aug. 8,	101 E	3. J
Grant & Mason		339.545	Apr. 6,	1886 A	
Grant, H. F		1,303,292	May 13,	1919 C)
Gray, G. W		1,193,541	Aug. 8,	1916 E	3
			June 1,	1909 I	
Gray, J. L		923,428 .	fune 1,	1909 I	1.5
Gray, J. L.	• • • • • • • • • • • • •	1,192,889	Aug. 1, June 1.	1916 F 1909 I	
Grav T T		1 158 205 (June 1, Oct. 26,	1909 I 1915 F	
Gregory, Ralph		1.2(1.01]	July 2,	1918 S	
Greene, H. J.		1.252.000	Jan. 1.	1918 F	
Green, Joel		46,794	Mar. 14.	1865 F	2
Greenstreet, Chas. J		1,110,924	Oct. 26,	1915 E	
Greenstreet, Chas. J		1,110,923	Sept. 15,	1914 H	
Greenstreet, C. J.		1,110,925	Sept. 15, Sept. 15,	1914 E	
Gray, J. L. Gray, J. L. Gray T. T. Gregory, Ralph Greene, H. J. Greene, H. J. Greenstreet, Chas. J. Greenstreet, Chas. J. Greenstreet, C. J. Greenstreet, C. J. Greenstreet, C. J. Greenstreet, C. J. Griefin, Jonathan Groble, J. C. Groble, J. C. Grogan, Henry	• • • • • • • • • • • • •	1 900 179	Sept. 15, Apr. 1,	1914 E 1919 E	5
Grieg A & Smith Jas		42.121	Mar. 29.	1864 F	1
Griffin, Jonathan		23,167	Mar. 8,	1869 N	Ĩ
Groble, J. C		1,283,502	Nov. 5,	1918 F	ζ
Grogan, Henry		94,409	Aug. 31,	1869 F	
Grogan, H. & Lape, G.	г	89,988	May 11, Feb. 28.	1869 F	2, 5, II
Grousilleumo, Emilo	• • • • • • • • • • • •	378,774	Feb. 28, June 27,	1888 I 1911 E	,
Gulick W B		1 187 061	June 13.	1916 N	
Groble, J. C. Grogan, Henry Grogan, H. & Lape, G. Grousilliers, Hector de. Gullaume, Emile. Gulick, W. R. Gumpoldt, Emil. Gesner, Abraham Hadley, B. E. Hague, S. L. Hague, S. L. Hall, C. H.		616.838	Dec. 27.	1898 N	
Gesner, Abraham		12,612	Mar. 27,	1855 G	
Hadley, B. E		1,300,230	Apr. 8,	1919 S	
Hague, S. L		775,448	NOV. 22.	1904 V	v, s
Hague, S. L	• • • • • • • • • • • •	759,988	May 17, Feb. 2.	1904 V 1869 F	
Hall, C. H	• • • • • • • • • • • • •	55,855	Feb. 2, June 26,	1866 F	
Hall, C. H. & Ellis, John		58 813 (Det. 16,	1866 F	1, II
Hall, T. G Hall, Wm. A		372,672	Nov. 8.	1887 V	,
Hall, Wm. A			Mar. 14,	1916 B	
Hall, Wm. A		1,105,772	Aug. 4,	1914 E	
Hall, Wm. A Hall, Wm. A	• • • • • • • • • • • •	1,194,289	Aug. 8, Sept. 4,	1916 E 1917 E	
Hall, Wm. A		1 175 010	Sept. 4, Mar. 14,	1917 E	
Hall, Wm. A Hall, Wm. A Hall, Wm. A Hall, Wm. A Hall, Wm. A Hall, Wm. A Hall, Wm. A		1.247.671	Nov. 27,	1917 E	
Hall, Wm. A		1,242,795	Oct 0	1917 E	
Hall, Wm. A		1,242,796	JCL. 9,	1917 E	
Hall, Wm. A		1,239,100	JCDU. 14.	1917 E	
Hall, Wm. A Hall, Wm. A	• • • • • • • • • • • • •	1,261,930	Apr. 9, Oct. 9.	1918 E 1917 E	
Hall Wm A		1 949 795 (Oct. 9, Oct. 9,	1917 E	
Hall, Wm. A		1.285.136	Vov 10		
Hall, Wm. C		266,990 1	Nov. 7.	1918 E 1882 F	2
Hamilton, T. S		1,018,871	CO. 21,	1912 A	
Halvorson, Halvor		305,182	Sept. 16,	1884 S	Contraction of the second
Hand Harry W	• • • • • • • • • • • • •	305,180	Sept. 16, Jan. 4.	1884 F 1898 U	
Handy Jas O	• • • • • • • • • • • • •	1 281 354	Jan. 4, Oct. 15,	1898 U 1918 O	, 13
Handy, Jas. O		1.084.738	Oct. 15,	1918 O	
Hansen, Julius		1,281,355 .	Jan. 20,	1914 C	
Hardy, C. A		51,042	Nov. 21,	1865 F	1.
Hardy, C. A.	• • • • • • • • • • • •	40,168	Oct. 6,	1863 F	
Hall, Wm. A. Hall, Wm. C. Hamilton, T. S. Halvorson, Halvor Handy, Jas. O. Handy, Jas. O. Handy, Jas. O. Hansen, Julius. Hardy, C. A. Hardy, C. A. Hardy, C. A. Hardy, C. A. Hardy, C. M.	• • • • • • • • • • • • •	40,899	Mar. 21, Oct. 15.	1865 F 1918 A	
and the real of the second		1,401,804	. 10,	1010 A	·, ·

UNITED STATES	PETROLEUM	I PAT	TENTS	-Con.
Name	Number	D	ate	Class
Name Harris, John Harris, Milo Hart, Thos. M. Hartshorn, H. M. Hastings, D. & Brink, A. W. Hatch, N. B. Hawes, Benj. N. Hazlett, R. W. & Hobbs, J. H. Hebard, Benj. F. Heckenbleikner & Gilchrist Helbing, H. & Passmire, F. W.	1 983 508	Nov.	5, 1918	K 2
Harris Milo	170 730	Dec.	7 1875	Ū ²
Hart Thos M	1 259 433	Jan.	8 1018	A, E 2, 3
Hartshorn H M	91.843	June	7, 1875 8, 1918 29, 1869	N N
Hastings, D. & Brink, A. W.	867.505	Oct.	1 1007	A, E 2, 3 N K 2, J
Hatch, N. B.	22.798	Feb.	1, 1859	G .
Hawes, Beni, N.	444.833	Jan.	20, 1891	G V
Hazlett, R. W. & Hobbs, J. H.,	24.211	May	31, 1859	G, S
Hebard, Benj. F	31,457	Feb.	19, 1861	M
Heckenbleikner & Gilchrist	1,310,078	July	15, 1919	I
Helbing, H. & Passmire, F. W	666,010	Jan.	15, 1901	D 1
Hempel, H	621,338	Mar.	21, 1899	M
Hempel, H	621,411	Mar.	21, 1899	M
Henderson, Geo. A	1,266,261	May	14, 1918	E 1 C
Henderson, N. M.	490,199	Jan.	17, 1893	C
Henderson, N. M.	340,878	Apr.	27, 1886	F
Hennebutte, H	1,165,878	Dec.	28, 1915	F
Hennebutte, H.	1,165,877	Dec.	28, 1915	F 4, I
Hense, Rudolf	1,073,233	Sept.	16, 1913 22, 1914	M F. D 1
Herber, Samuel M.		Sept. May	22, 1914	F, D 1 F 2, 3, D
Herber, S. M	1 970 750	June	16, 1916 25, 1918	F 2, 3, D
Highia M S & Doughanty A		Aug.	7 1920	B, K 2 C, E, 1
Highie M S & Dougherty A.	001,000	Aug.	7, 1888	C, E, 1. 3
Higging Chas S	300 718	Dec.	7, 1888 23, 1884	Ċ, É, 1 C, E, 1. 3 N
Higham A D	54 157	Apr.	24, 1866	F
Hill B L	1 269 439	June	11, 1918	B
Hill S & Thumm C F	101 364	Mar.	29, 1870	F 1, II
Hill, S. & Thumm, C. F.	101,365	Mar.	29, 1870	F 1, 11
Hill, S. & Thumm, C. F	102.819	May	10, 1870	F 1, 11
Hill, S. & Thumm, C. F	114.293	May	9 1871	F 1, 3, I)
Hirshberg, Leon	1,042,915	Oct.	29, 1912	D
Heckenbleikner & Gilchrist Helbing, H. & Passmire, F. W. Hempel, H. Henderson, Geo. A. Henderson, N. M. Henderson, N. M. Hennebutte, H. Hense, Rudolf Herber, Samuel M. Herber, Samuel M. Herber, S. M. Hibbert, Harold Higbie, M. S. & Dougherty, A. Higbins, Chas. S. Higbians, Chas. S. Higham, A. D. Hill, R. L. Hill, S. & Thumm, C. F. Hill, S. & Thumm, C. F. Hirt, Le. Hornes, F. W. & Blasdell, E. Hodkinson, M. Hofferberth, John Hoffman, Ross J. Holmes, Jos. E.	1,222,402	Apr.	10, 1917 18, 1917	B, P
Hirt, L. E	1,250,879	Dec.	18, 1917	B, P
Hirt, L. E	1,264,796	Apr.	30, 1918	K, 3
Holmes, F. W. & Blasdell, E	1,055,747	Mar.	11, 1913	B
Hodkinson, M	26,326	Nov.	29, 1859	G, W
Hofferberth, John	105,683	July	26, 1870	F 1, I
Hoffman, Bernhard	641,962	Jan.	23, 1900	M
Hoffman, Ross J. Holmes, Jos. E. Holmes, Jos. E. Hod, J. J. & Salamon, A. G. Hopkins, A. S. Hopkins, A. S. Horner, E. N. Houlker, Christopher Howard, F. A. Howard, F. A. Howell, G. G. Howell, C. G. Howell, H. F. Hudson, Chas. R. Hudson, Samuel.	405,738	June	25, 1889	S
Holmes, Jos. E.	23,427	Mar. Oct.	29, 1859	G, W B. J
Holmes, Jos. E	1,241,979		2, 1917	W
Hood T I & Salamon A C	24,212	May June	31, 1859 28, 1910	D 2
Honking A S	1 100 /63	Sept.	26, 1916	B
Honkins A S	1 100 464	Sept.	26, 1916	B
Horner, E. N	92 727	Jan.	25, 1859	W, D
Houlker, Christopher	110,364	Dec.	20, 1870	0
Howard, F. A.	1.284.687	Nov.	12, 1918	F
Howarth, John	42.772	May	17. 1864	W, F
Howe, Ephriam	7,667	Sept.	24, 1850 18, 1919	M
Howell,	1,294,909	Feb.	18, 1919	8
Howell, C. G	66,841	July	16, 1867	F 1, 2
Howell, H. F.	216,518	June	17, 1879	L
Hudson, Chas. R.	681,170	Aug.	20, 1901	A
Hudson, Samuel.	123,907	Feb.	20, 1872	G
Hugio, Victor.	953,952	Apr.	5, 1910	B F4
Hout, F. & Rogers, John	71,619	Nov.	15, 1867 19, 1866	
Humadon C	1 001 000	Sept. Jan.	21, 1919	SS
Hudson, Chas. K. Hudson, Samuel. Huglo, Victor. Hout, F. & Rogers, John Hut, F. & Rogers, John Humphreys, R. E. Humphreys, R. E. Humphreys, R. E. Humphreys, R. E. Humphreys, R. E. Humphreys, R. E.	1 1 2 9 0.09	Dec.	22, 1914	B.S
Humphreys R E	1 199 002	Dec.	22, 1914	B
Humphreys, R. E.	1 119 700	Dec.	1, 1914	В
Humphreys, R. E.	1,286 179	Nov.	$1, 1914 \\ 26, 1918 \\ 1967 \\ 1967 \\ 1967 \\ 1967 \\ 1966 \\ $	I
Huntington, John.	62,750	Mar.	12, 1807	F
Huston, John B	297,603	Apr.	29, 100*	S
Huston, John B	486,406	Nov.	15. 1892	v
Huntington, John. Huston, John B. Huston, John B. Hyde, Burrows. Hall, Wm. A.	281,999	July	24, 1883 9, 1917	T
Hall, Wm. A	1,242,746	Oct.	9, 1917	В

UNITED	STATES	PETROLEUM	I PATENTS-	-Con.
Name		Number	Date	Class
Hall, Wm, A Holmes, Fletcher B Hussey, John S Ihart, J. P Ilges, T. W Isom, Edward W. Jaeger, W. G. W Jaeger, W. G. W Jaeger, W. G. W		1 242 795	Oct. 9, 1917	в
Holmes, Fletcher B		1.276.219	Aug. 20, 1918	Ď
Hussey, John S		1,277,935	Sept. 3, 1918	С
Ihart, J. P		654,258	July 24, 1900	A
Ilges, T. W		968,478	Aug. 23, 1910	
Isom, Edward W.		1,285,200	Nov. 19, 1918	
Jaeger, W. G. W		24,217	May 31, 1859	W
Jaeger, W. G. W.	• • • • • • • • • • • •	54,358	June 28, 1859 Apr. 16, 1866	
		86,232	Apr. 16, 1866 Jan. 26, 1869	F 1, 2, II
James, C. M Jann, John		52.574	Feb. 13, 1866	
Jann, John Jann, John Jenkins, U. S Jennings, Isalah. Jenney, W. P Jenney, W. P Jenney, W. P Jensen, J. O. Johnson, John Johnson & Shodgr Johnston, Jas. J. Johnston, Jas. J.		52,574 57,727	Sept. 4, 1866	M
Jenkins, U. S		1,226,526	May 15, 1917	J, B
Jennings, Isaiah		1,453	Dec. 31, 1839	M
Jenney, W. P		190,762	May 15, 1877	Ĩ
Jenney, W. P		178,061	May 30, 1876	
Jenney, W. P	• • • • • • • • • • • •	1 968 791	May 30, 1876 June 4, 1918	
Johnson John		54 917	June 4, 1918 May 22, 1866	
Johnson & Snodgr	ass	1.283.202	Oct. 29, 1918	S
Johnston, Jas. J.,		117.425	July 25, 1871	F
Johnston, Jas. J		117,426	July 25, 1871	A F 4, 5
Johnston, Jas. J		48,285	June 20, 1865	F 4, 5
Johnston Jas. J.		31.982	Apr. 9, 1861	S F 2
Johnston, Jas. J		50,935	Nov. 14, 1865	F_{2}
Johnston, Jas. J		91,448	June 15, 1869	F 2, II K 2, S
Jones, Philip,	T D	1 080 096	Jan. 29, 1918 Mar. 10, 1914	K 1, 2
Jones & Jones	зъ, ц. р	1 157 995	Mar. 10, 1914 Oct. 19, 1915	
Jones, R. G.		1.166.375	Dec. 28, 1915	
Jones, R. G		1.005,977	Oct 17 1911	A
Johnston, Jas. J. Johnston, Jas. J. Jones, Philip. Jones, E. C. & Jone Jones, R. G. Jones, R. G. Jones, R. G. Jones, R. G. Jones, R. G. Jones, R. G. Jones, R. G. Kaysar, A. Just, John A. Kasson, H. R. & Si Kattell, E. C. Kaysar, Adolf. Kaysar, A. Keen, Morris L. Kelley, E. G. Kelley, E. G. Kelley, E. G.		126,552	May 7, 1872	
Just, John A		658,988	Oct 2 1900	M
Kasson, H. R. & Sa	axton, S. S.	998,691	Apr. 7, 1914	E 1, 2
Kattell, E. C		222,408	Dec. 9, 1879	F 2, 4
Kaysar, Adoll		008,479	Nov. 14, 1893 Jan. 9, 1900	
Keen Morris L		25 552	Jan. 9, 1900 Sept. 20, 1859	F1
Kelley, E. G.		67.988	Aug. 20, 1867	F 1. II
Kelley, E. G. & Ta	it. A. H	32,568	June 19, 1861	F 1, 2, II
Kelley, E. G		84,195	Nov. 17, 1868	
Kells, Edw		298,210	May 6, 1884	F.
Kells, Edw.			Dec. 13, 1887	
Kelsey, S. E	• • • • • • • • • • •	1,092,366	Apr. 7, 1914 May 6, 1919	BS
Kelsey, S. E Kendall, Edw. D		1,302,009	May 6, 1919 Oct. 22, 1889	Ď
Kendall, Edw. D		359 357	Mar. 15, 1887	D1
Kendall, Edw. D.,		284.437	Sept. 4, 1883	D, M
Kendall, Edw. D		451,660	May 5, 1891	D 1, 2
Kendall, Edw. D		1,192,529	July 25, 1916	
Kendall, Edw. D		1,154,517	Sept. 21, 1915	D 1, S
Kendall, Edw. D.	• • • • • • • • • • • •	1,154,516	Sept. 21, 1915	
Kendall, Edw. D Kendall, Edw. D Kendall, Edw. D Kendall, Edw. D Kendall, Edw. D Kennedy, D. McD. Kerr, A. N Keyt M. H	•••••	1 100,002	Oct. 4, 1887 Oct. 3, 1916	
Kert, A. N. Keyt, M. H. Kipper, H. B. Kirchhoffer, G. W. Kirk, Arthur.		1 969 808	Oct. 3, 1916 Apr. 16, 1918	D
Kinner H B		1 253 048	Jan. 8, 1918	D1
Kirchhoffer, G. W.		32.373	May 21, 1861	G, W F 1, II
Kirk, Arthur		78,878	June 16, 1868	F 1, II
Kirk, J. L		215,756	May 27, 1879	F 1, 11
Kirk, Solomon W		267,752	Nov. 21, 1882	C
Kirschbraun, L		1,194,750	Oct. 3, 1916	E 1, 2 F 1, 2, II
Klein John S		1,008,273	Aug. 15, 1916 Oct. 21, 1884	S 1, 2, 11
Kline Geo H		353 369	Nov. 30, 1886	F 1, II, S
Klosterman, Robt.		152,650	June 30, 1874	F
Knottenbelt, H. W		1,194,033	Aug. 8, 1916	W
Knottenbelt, H. W		1,277,605	Sept. 3, 1918	D 1
Koehler, Herman.		507,441	Oct. 24, 1893	V
Kirk, Arthur. Kirk, J. L. Kirk, Solomon W. Kirschbraun, L. Kitchen, J. M. W. Klein, John S. Kline, Geo. H. Klosterman, Robt. Knottenbelt, H. W Knottenbelt, H. W Knottenbelt, H. W Koehler, Herman. Koehler, W. C. & L	ink, L	1,084,016	Jan. 13, 1914	0

UNITED STATES PETROLEI	UM PATENTS-	-Con.
Name Numbe	r Date	Class
Koppers, H	4 June 2, 1914	F 2, II
Kreiser, J. M	8 June 19, 1888	S
Treiser, J. M	7 July 12, 1887	Ŧ
Areusler, A 50,36	⁸ Oct. 10, 1865	S F F
Lacy, B. S	8 Oct. 10, 1865 6 Apr. 23, 1918	L
Lacy, B. S	T Feb. 15, 1910	F 2, II
ng, John 471,29	1 Mar. 22, 1892	B
aind Poht H 50792	7 Dec. 27, 1892 0 Oct. 24, 1893	B F 2. II
aird Roht H. 498.51	0 Oct. 24, 1893 8 May 30, 1893	F 2, II F
ng, John	9 Nov. 3, 1914	Â, P
aird & Raney1,142,76	1 June 8, 1915	A, P
ird & Raney1,142,76	0 June 8, 1915	A.P
1,142,75 & Raney1,142,75	9 June 8, 1915	A.P
_amb, D. M 183,40	1 Oct. 17, 1876	D1 C
Lambe, Frederick 102,13	5 Apr. 19, 1870	C
Lamplough F 1920.00	0 Nov. 6, 1917 8 June 5, 1917	B
Landes Wm. 1100.00	8 June 5, 1917 9 Oct. 3, 1916	B
Landsberg, I	1 Jan. 9, 1917	Ĩ
Lane, Edw 172.13	1 Jan. 11, 1876	F 1.II
Lang, J. S 954,57	5 Apr. 12, 1910	В
Lapham, Allen 59,31	7 Oct. 30, 1866	F
1pp, C. E	1 May 14, 1918	B
_asher, D. F1,075,48	1 Oct. 14, 1913	D1
Jee, A. K 102,39	4 Apr. 20, 1875	E1 D
aman Wm T 797 20	4 Dec. 24, 1918 1 May 5, 1903	Ŭ
ennard F 45012	3 Sept. 8, 1891	F 2, II
Jennard, F	7 June 13, 1893	F2
Lennard, F 659.07	6 Oct. 2, 1900	Ť F
Lepley, Clyde E1,261,41	0 Apr. 2, 1918	F
Leslie, E. H	4 July 15, 1919	Š K 2
essing, Rudolf	7 Oct. 15, 1918	K2
etchford, R. M. & Nation, W 133,04	2 Nov. 12, 1872 8 Jan. 1, 1918	C
evy, L. D. & Jacobs, H. W	8 Jan. 1, 1918 7 June 10, 1862	Q M
ewis S 42.67	1 May 10, 1864	V
ewis, S 43.15	1 May 10, 1864 6 June 14, 1864	M
indenberg, G. & Scott, W. B1,220,65	1 Mar. 27, 1917	K 2, B
_indsey, Wm. J1,256,34	0 Feb. 12, 1918	K1
aird & Raney. 1,142,75 .amb, D. M. 183,40 Lambe, Frederick. 102,13 Lambert, Chas. G. 1,245,93 Lamplough, F. 1,229,09 Landsberg, L. 1,211,72 Lang, J. S. 954,57 Lapham, Allen. 59,31 upp, C. E. 1,266,28 .asher, D. F. 1,026,39 eee, A. K. 162,39 eenard, F. 1,265,90 Leenler, E. H. 1,218,92 eenard, F. 499,55 Leenley, Clyde E. 1,261,41 Leslie, E. H. 1,310,16 essing, Rudolf. 1,281,59 ewis, S. 42,67 ewis, S. 1,226,634 Linne, S. 1,226,634 Linn, S. S. 1,226,634 Linnston, Julius I. 239,266 Livingston, Julius I. 239,266 Livingston, Julius I. 237,66 <td>7 Nov. 5, 1918 8 May 30, 1882</td> <td>M</td>	7 Nov. 5, 1918 8 May 30, 1882	M
Livesay, Jas. & Kidd, Jas 258,77	8 May 30, 1882	FT
Livingston, Julius 1 239,20 Livingston, Max 237,56	0 Mar. 22, 1881 0 Feb. 8, 1881	F
Livingston, Max. 237,56 Livingston, Max. 237,56 Livingston, Max. 728,25 Lockhart, Chas. & Gracie, J. 40,63 Lockhart & Gracie. 80,29	7 May 19, 1903	FII
Lockhart, Chas. & Gracie, J 40,63	2 Nov. 17, 1863	F
Lockhart & Gracie 80,29	4 July 28, 1868	F
Loew, Oscar 101,28	4 Mar. 29, 1870	D1
Lofhjelm, Karl 546,01	8 Sept. 10, 1895	F
Loftus, Robt. G 113,78	2 Apr. 18, 1871	D1 K2
Loftus, Robt. G 81,65	4 Sept. 1, 1868 7 June 14, 1864	I I
Long F P 195614	7 June 14, 1864 6 Feb. 12, 1918	ŝ
Loomig C C 1 280 614	2 Oct. 1, 1918	L
Loomis Wells, Hitchcock & Stryker, 66.36	4 July 2, 1867 9 May 20, 1873 8 Apr. 30, 1918	M
Looney, John J 139,009	9 May 20, 1873	D1
Lorch, H. D	8 Apr. 30, 1918	F 2, 5
Loew, Oscar. 101,28. Lofhjelm, Karl. 546,01 Loftus, Robt. G. 113,78 Loftus, Robt. G. 43,16 Loftus, Robt. G. 43,16 Loomis, Robt. G. 43,16 Loomis, C. C. 1,280,61 Loomis, Wells, Hitchcock & Stryker. 66,36 Loorey, John J. 128,466 Low, Frank S. 1,19,265 Low, Frank S. 1,192,65 Low, W. P. & Bilfinger, C. W. 556,15 Lucas, Owen D. 1,168,40 Lucas, Owen D. 1,183,09	1 Apr. 9, 1895 3 July 25, 1916	V J, B
Low, Frank S	3 July 25, 1916 5 Mar. 10, 1896	B B
Lowe, W. P. & Billinger, C. W 556,154	5 Mar. 10, 1896 4 Jan. 18, 1916	B
Lucas, Owen D	1 May 16, 1916	B
Turno Omorio 51 84	2 Tan 2, 1866	F 3
	7 Jan. 1, 1867	V, D 1 F 3, 4, I
Jugo, Orazio	3 Sept. 18, 1800	F 3, 4, I
Lugo, O. & Schrader, T. O. L 60,390	Dec. In ion	F 3, 4, 1
Lupton, George, 110,054	4 Dec. 13, 1870	D F1,II
Lutz, H. E 240,914	4 May 3, 1881	1, 1, 11

Name Number Duration Madares, A.M., 1,227,022 Aug. 21 MacArlee, A.M., 1,227,022 Aug. 21 MacArlee, A.M., 1,227,023 Aug. 22 MacArlee, A.M., 1,224,012 Aug. 22 MacCarty, F. M., P.M. 1,224,012 Aug. 20 MacCarty, Karl P. 1,259,757 Mar. 11 MacGowan, T. 463,022 Dec. 22 MacGowan, T. 463,023 Duc. 22 MacGowan, T. 1,259,075 Mar. 12 MacGowan, T. 1,259,075 Mar. 12 UNITED STATES PETROLEUM PATENTS-Con. Name Number Date Class 8, 1915 в 27, 1918 C June 2, 1914 Feb. 9, 1915 June 22, 1915 Oct. 24, 1916 ĕ $\mathbf{\bar{B}}$ $\tilde{\mathbf{B}}$ $\mathbf{\bar{B}}$ 24, 1916 27, 1918 27, 1918 31, 1917 8, 1914 5, 1918 D D $\bar{\mathbf{B}}$ \mathbf{B} S Feb. 5, 1916 June 29, 1869 Aug. 6, 1918 Aug. 6, 1918 Aug. 10, 1858 Mar. 19, 1918 F 2, II в BW K 2, B Mar. 19, 1918 Mar. 19, 1918 Feb. 28, 1893 June 16, 1891 Dec. 23, 1890 Oct. 2, 1900 \mathbf{K} F F F 77 16, 1882 F 3, D 1 $\begin{array}{c} 1, 1890 \\ 3, 1875 \\ 28, 1893 \end{array}$ F \overline{F}_2 ŝ 1, 1874 S $\begin{array}{c} 1, 1874\\ 27, 1874\\ 28, 1915\\ 23, 1917\\ 8, 1919 \end{array}$ F 1 B, K 1 L A Apr. 8, 1919 Oct. 6, 1914 Sept. 16, 1884 May 20, 1879 Aug. 7, 1900 Nov. 12, 1901 Dec. 25, 1900 Oct. 18, 1903 June 27, 1916 July 16, 1819 Dec. 7, 1916 July 16, 1819 Dec. 7, 1916 8, 1919 A K1 Ι Ĉ D 1, 2 D 1, 2 F 2, 5, I \mathbf{D} 0, D $\tilde{\mathbf{D}}$ 1 \mathbf{B} 14, 18997, 1915 16, 1916 30, 1917 11, 1917 26, 1918 D \mathbf{L} $\overline{\mathbf{B}}$ в $\tilde{\mathbf{B}}$ May 28, 1878 July 7, 1874 June 26, 1866 Jan. 28, 1879 NNMC Jan. 28, 1879 July 5, 1881 Mar. 14, 1882 June 30, 1908 Jan. 6, 1891 Jan. 6, 1891 Feb. 11, 1919 Nov. 29, 1887 Sept. 1, 1882 Dec. 13, 1882 Feb. 10, 1880 Sept. 4, 1866 F ŕ B, P F 1, 2, II F 1, 2, II M F 2, 5, II F 1, 2, II M M F 1, 4, II 4, 1866 11, 1871 22, 1891 19, 1891 M F F 1, 3 F 3, V $\begin{array}{c} 31, 1855 \\ 22, 1869 \\ 27, 1919 \\ 12, 1867 \end{array}$ w Ö, D J-K C

UNITED STATES PETROLEUR	A PA	TENTS-	-Con.
Name Number	L	Date	Class
Merrill, Francis B 761,315	May	31, 1904	F
Merrill, Joshua 33,955	Dec.	17, 1861	ŝ
Merrill, Joshua,	July	30, 1861	S
Merrill, Joshua 32,706	July	2, 1861	S
Merrill, Joshua	July	2, 1861	D1
Miller, J. R	Aug.	5, 1919	B
Miller, J. R	July	2, 1861	D1
Merrill, Joshua	May	18, 1869	F 1. 2
Merrill Joshua 43 325	June	28, 1864	DI
Merrill, Joshua	Jan.	1 1019	E3
Mertz, Josef	Apr.	$\begin{array}{c} 1, \ 1918 \\ 6, \ 1886 \end{array}$	F 2, 11
Mertz, Josef	Oct.	29 1018	K
Mesereau G 1308 802	July	29, 1918 8, 1919	ĸ
Meucci, Antonio	Sept.	9, 1862	D1
Midgely, T., Jr	Mar.	11, 1919	M
Mijs. Jan	Apr.	11, 1916	Ĉ
Miles George 205407	June	25, 1878	F.S
Miles, George W	Jan.	18, 1916	F, S C
Miles, George W. 1,166,534 Miller, Jas. 77,070 Millochau, Adolph. 38,641 Millochau, A. 37,918	Apr.	21, 1868	F 5, II
Millochau, Adolph	May	19, 1863	D1
Millochau, A	Mar.	17, 1863	Di
Millochau, A 53,167	Mar.	13, 1866	Fi
Millochau, A 46,923	Mar.	21, 1865	Fi
3/211-1-1-1	Jan.	5, 1864	Di .
Millochau A. 49777	Sept.	5, 1865	N
Mills E N	Nov.	7, 1911	Ô
Millspaugh, Pethuel	May	28, 1872	Q N
Mims. John C	Nov.	11, 1902	D1, E3 -
Minshall, F. W	Nov.	26, 1889	F 2 3 V
Mitchell, Willis	May	25, 1915	F 2, 3, V K 1
Montague, H. E	May	22, 1917	B
Millochau, A. 41,085 Millochau, A. 49,777 Mills, E. N. 1,007,788 Millspaugh, Pethuel. 127,259 Mims, John C. 718,475 Minshall, F. W. 514,876 Mitchell, Willis. 1,141,072 Montague, H. E. 1,227,551 Moorey, L. 786,828	Mar.	7, 1916	Ř
Moore E. A	Apr.	11, 1905	A
Moore George H 586 520	July	13, 1897	Ÿ, D1
Moore, E. A. 786,828 Moore, George H. 586,520 Moore, E. S. & Thomas, H. H. 1,281,808 Moore, J. B. 1,130,318 Morehouse, C. L. 55,426 Morehouse, C. L. 174,921 Morehouse, C. L. 62,426	Oct.	15, 1918	S
Moore J B	Mar.	2, 1915	B
Morehouse, C. L. 55 426	June	5, 1866	D1, C
Morehouse, C. L	Mar.	21, 1876	D 1, C G
Morfit, Clarence	July	2, 1867	ŭ
Morris, W. L	Apr.	27, 1915	U C O
Morris, W. L. 1,137,075 Morris, W. L. 1,305,785 Mott, Leander M. 54,192 Mowbray, George M. 25,575 Munson, A. L. 440,830 Murray, Thos. E. 1,273,523 Murray, T. E. and Ricketts, E. B. 1,293,886 Murray, T. E. 1,302,200 Myers, Geo. W. 147,783 Navin, F. 1,312,266 Nealous, Herman. 242,554	June	3, 1919	ŏ
Mott. Leander M	Apr.	24, 1866	ŏ
Mowbray, George M 25,575	Sept.	27, 1859	F 1, 4, II
Munson, A. L	Nov.	18, 1890	D
Murray, Thos. E	July	23, 1918	S
Murray, T. E. and Ricketts, E. B., 1,293,866	Feb.	11, 1919	F
Mueller, C. L. E	Mar.	18, 1919	M
Murray, T. E	Apr.	29, 1919	S
Mvers, Geo, W 147,783	Feb.	24, 1874	K 2, S
Navin, F	Aug.	5, 1919	W
Neahous, Herman. 242,554 Neal, Stephens	June	7, 1881	C
Neal, Stephens,	Aug.	20, 1912	F 2
Neilson Albert,	Apr.	5, 1881	F
Newall, Robert	Apr.	3. 1866	V, D
Newsome, Thos. J 405,047	June	$11, 1889 \\ 6, 1919$	A
Nichols, H. M	May	6, 1919	S
NICHOISON, JOHN	Feb.	15, 1859 3, 1880 16, 1880 22, 1904	W
Nicolai, J. H. & W. F 224,037	Feb.	3, 1880	G, S
Nicolai, Pierre 225,635	Mar.	16, 1880	F2
Nikiforoff, A 755,309	Mar.	22, 1904	В
Noad, James 971,468	Sept.	27, 1910	В
Ncad, Jas. 91,200 Ncad, Jas. 985,053 Nordenson, Carl O. 1,218,575 Norton, J. W. & Rouse, F. H. 313,514 Norton & Rouse. 336,941	Feb.	21, 1911	B, W
Nordenson, Carl O1,218.575	Mar.	6, 1917	K1
Norton, J. W. & Rouse, F. H 313,514	Mar.	10, 1885	S
Norton & Rouse 336,941	Mar.	2, 1886	F 2, 4, D 1
Noteman, Alonzo 512,894	Jan.	16, 1894	D
Noyes, John E 82,151	Sept.	15, 1868	G, M
Noteman, Alonzo. 512,894 Noyes, John E. 82,151 Oglyy, David J. 1,268,142	June	4. 1918	W
O'Hara, Jas 22,573	Jan.	11, 1859	K8

UNITED STATES	PETROLEUM	A PA'	TENTS-	-Con.
Name	Number	E	Date	Class
Olsen, Geo. E O'Neall, J. M	1,199,491	Sept.	26, 1916	J, A F 1, 2, II
O'Neall, J. M.	754,687	Mar.	15, 1904	F 1, 2, II C
Paine. Henry H	9.119	Feb. July	16, 1915 13, 1852	M
Palmer, Chas. S	1,187,380	June	13, 1916	B
Opl, Karl. Paine, Henry H. Palmer, Chas. S. Palmer, Chas. S. Palmer, Chas. S. Darker, J. H.	1 813 009	June Aug.	4, 1918 12, 1919	K1 B
Parker, J. H.	958,820	May	24, 1910	в
Parker, R. B.	1,252,481	Jan.	8, 1918	K 2
Parker, W. C.	1.226,990	Oct. May	26, 1875 22, 1917	0 B
Parsons, Chas. C	88,978	Apr.	13, 1869	F 2, 5
Parsons, C. Chauncey	93,739	Aug. Apr.	17, 1869 29, 1879	CERS
Pease, Francis S	226,187	Apr.	6. 1880	F, K 2 N
Pemberton, Henry	24,952	Aug.	2, 1859 28, 1878 7, 1869	W, I
Penissat, Andre.	204,244	May Oct.	28, 1878 7, 1862	IT
Perkins, George H	399,073	Mar.	5, 1889	T F S F 2, 11
Perkins, Geo. H	240,923	May	5, 1889 3, 1881 4, 1865	S
Perkins, J. & Burnet, Will. H.	731.943	Apr. June	23, 1903	
Perrier, Odilon.	544,516	Aug.	13. 1895	F 1, 2, II V, D J, K 2
Perrine, Robt. M.	419,347	Jan. July	14, 1890 2, 1912	V, D
Petroff, Grigori,		Feb.	17, 1914	J, K 2
Petroff, G	1,233,700	July	17, 1917	D1
Petty, T. K. & Warden, W. G.	37,263	Dec. Dec.	23, 1862 25, 1894	SP
Pfeifer, F.	1,296,115	Mar.	4, 1919	ĸ ĸ
Pfeifer, F.	1,296,116	Mar.	4, 1919	ĸ
Phillip, A Phillippe Joseph	1,286,091	Nov. Jan.	26, 1918 18, 1870	Q G, М
Pictet, Raoul P	1,228,818	June	5. 1917	B
Pielsticker, Carl M	186,951	Feb.	0, 10//	D1
Pielsticker, Carl M	1.070.730	June Aug.	14, 1892 19, 1913	F 2, II C
Pinckney, T. De Witt	221,421	Nov.	11, 1879	N
Pinkham, C. W.	34,772 D 1 057 667	Mar.	$ \begin{array}{r} 25, 1862 \\ 1, 1913 \end{array} $	M,G E3
Pine, J. A. W. & Ruggies, Whi.	379.492	Apr. Mar.	12 1888	F.V
Pitt, Wm. H	411,394	Sept.	17, 1889	F, V F, V F
Place, Chas. T.	243,080	June Feb.	21, 1881	F F 2, II
Pollak, R. R.	1.254.271	Jan.	22. 1918	
Ponton, John.	165,612	July	13. 1875	A N C G
Porges, P, & Neumann, R	1,017,587	Feb. Jan.	13, 1912 27, 1874	C
Porter, Alonzo W	453.386	June	2, 1891	w
Pray, Lyman.	61,098	Jan.	8. 1867	S, F
Prentiss, E. F. & Robertson, R.	A 48,435	June Mar.	27, 1865 8, 1864	U F 2, II
Prentiss & Robertson	1.273.091	July	16, 1918	F 2, 11 F 2, 4
Price, Walter B	548,391	Oct.	22, 1895	D1
Price, W. B	522,628	June	26, 1894	G, D 1 F 2, II
Prichard, Geo. 1	1.290.345	Apr. Jan.	30, 1918 7, 1919	F 2, II I
Propfe, H.	478,265	July		F 1, II
Prutzman, Paul W	1 1 76 004	Aug. Mar.	28, 1917 21, 1916	Aved
Puning, Franz.	1.040.408	Oct.	8, 1912	C 2, 5
Pyzel, Daniel.	1,276,690	Aug.	20, 1918	S
Quinn, Abraham.	31,998	Apr. Sept.	9, 1861 16, 1862	F. F.
Band Alonzo C.	62.362	Feb.	26, 1867	ŝ
Rave, Chas.	425,905	Apr.	15, 1890	I, P
Reese, Jacob.	38,602	May May	19, 1863 5, 1874	220
Paine, Henry H. Palmer, Chas. S. Palmer, Chas. S. Palmer, Chas. S. Parker, J. H. Parker, R. B. Parker, W. C. Parsons, Chas. C. Parsons, C. Chauncey. Parsons, C. Chauncey. Parsons, H. E. Pease, Francis S. Pemberton, Henry. Penissat, Andre. Perkins, A. H Perkins, George H. Perkins, George H. Perkins, George H. Perkins, J. & Burnet, Wm. H. Perkins, George H. Perkins, C. M. Perkins, George H. Perkins, C. M. Perkins, C. M. Perterson, F. P. Petroff, Grigori. Petroff, G. Pettry, T. K. & Warden, W. G. Pettry, T. N. & Warden, W. G. Pettry, T. N. & Warden, W. G. Pettry, T. N. & Warden, W. G. Pettry, T. De Witt. Philips, Joseph. Pictet, Raoul P. Pielsticker, Carl M. Pijzel, Daniel. Pinkham, C. W. Pine, J. A. W. & Ruggles, Wm. Pitt, Wm. H. Pitt, Wm. H. Pitt, Wm. H. Pitt, Wm. H. Porter, Alonzo W. Poterie, George. Pray, Lyman. Protes, F. & Robertson, R. Prentiss, E. F. & Robertson, R. Prentiss, B. A. Prote, M. Prote, M. Prote, C. P. Prote, M. Prote, C. P. Prote, M. Prote, C. P. Prote, M. Prote, C. P. Prote, M. Proter, Joaniel. Pyzel, Daniel. Pyzel,	1,302,090	Apr.	29, 1919	FI, II A 2, S C S F F F S I, P S S T

UNITED STATES PETROLEUM	A PA	TENTS -	-Con.
Name Number	I	Date	Class
Reeves W P	Nov.	5, 1918	S
Reilly, P. C	July	15, 1919	F,S
Rensink, G. C	Apr.	6, 1915	Ā
Regua, Chas. W 77,094	Apr.	21, 1868	F 1, 2, I
Restieux, Thos 63,749	Apr.	9. 1867	V, D 1 F 2
Reynolds, F. R1,119,453	Dec.	$\begin{array}{c} 1, 1914 \\ 25, 1869 \\ 10, 1895 \\ 26, 1918 \end{array}$	F2
Rice, L. M. & Adams, S. E 90,392	May	25, 1869	S
Richardson, Clifford 551,294	Dec.	10, 1895	E 3, A
Richardson, Wm. D1,257,397	Feb.	40, 1010	P C
Richardson, John E 65,275	May	28, 180/	C
Richter, Felix	June	2, 1914	D
Richter, Feilx	June	2, 1914	D D
Ditos E M 1144 798	Jan. June	4, 1916 29, 1915	K1,B
Rites, F. M	June	29, 1915	K 1, B K 1, B
Roake John S 700 373	May	20, 1902	DI
Roberts A E & Emery A L. 1016 958	Feb.	13, 1912	Q
Robertson J. H	Aug.	28, 1917	B, P
Robinson, C. L	Jan.	9, 1912	ī
Robinson, C. I	Feb.	20 1019	F1
Robinson, C. I	Aug.	30 1910	D1
Robinson, C. I	Jan.	26, 1909 26, 1879 19, 1916	V, D F 2, II
Robinson, J. C 218,901	Aug.	26, 1879	F 2, II
Rodman, Hugh	Dec.	19, 1916	В
Rogers, Davenport 211,055	Dec.	17. 1878	F 2, 4, II
Rogers, D 284,331	Sept.	4, 1883	F
Rogers, F. M	Apr.	1. 1919	A F
Rogers, Henry H 120,539	Oct.	31, 1871	F
Rogers, John	Oct.	3, 1865	F
Rogers, Lebbeus H1,269,747	June	18, 1918	w
Rogers, F. M. & Cooke, T. S1,122,220	Dec.	22, 1914	J
Rogers, M. C	Aug.	3, 1915	S
Rogers, Wm. B 60,559	Dec.	18, 1866	M
Roots, James 340,522	Apr.	20, 1886	M, G
Rose, H. C 182,775	Oct.	3, 1876	M, G F 1, 2, II B, K 1
Rose, James R	Jan.	1, 1918 28, 1915	B, K 1 O
Rosen, Jean	Dec.	20, 1915	B
Rosen, Jean	Nov.	30, 1915 14, 1916	B
Russ, S. J. & Scholleld, H	Dec.	12, 1916	B
Roth & Vonturino 1208 914	Dec.	12, 1916	B
Poth & Venturino 1 208 378	Dec.	12, 1916	B
Rowlands \mathbf{P} \mathbf{O} 1 252 955	Jan.	8, 1918	s
Rowsell Tohn 299.167	May	27, 1884	D1
Ryder Henry 142.515	Sept.	2, 1873	F, S
Ryder, Watson 214,199	Apr.	8, 1879	F 1, II
Ryder, W. & Qualey, J. A	Sept.	22, 1903	F
Rosenbaum, R. R	Sept.	3. 1918	C, E 2
Ruff. F. C	Apr.	16, 1918	D1
Sabatier, P. & Mailhe, A1,124,333	Jan.	12, 1915	B, P
Sabatier, P. & Mailhe, A1,152,765	Sept.	7, 1915	B
Salathe, Frederick 452,764	May	19. 1891	T
Salathe, F 564,341	July	21, 1896	T
Sampson, C. E. & Woods, W1,177,816	Apr.	4, 1916	B
Sangster, W. H 54,414	May	1, 1866	S, D
Sangster, W. H. & Spencer, T. C 56,276	July	10, 1866	F
Sargent, Thos. D 20,587	June	15, 1858 24, 1918	W E1
Savage, Wallace	Sept.	24, 1910	E I
Sawyer, G. T., Howland, W., Jr., &	Dec.	10, 1861	S
UNITED STATES PETROLEUT Name Number Reeves, W. P. 1,283,559 Reilly, P. C. 1,310,164 Rensink, G. C. 1,310,164 Regunolds, F. R. 1,114,419 Reynolds, F. R. 1,114,453 Richardson, Clifford 551,294 Richardson, John E. 665,275 Richter, Felix. 1,098,763 Richer, Felix. 1,098,763 Richer, Felix. 1,098,763 Richer, Felix. 1,098,763 Rites, F. M. 1,144,789 Robertson, J. H. 1,283,339 Robbertson, J. H. 1,283,339 Robinson, C. I. 1,018,574 Robinson, C. I. 910,584 Roberts, Davenport. 211,055 Rogers, Davenport. 210,553 Roberts, S. E. 284,331 Roberts, M. E. 284,331 Roberts, S. E. 284,331 Rogers, Davenport. 211,055 Rogers, B. C. 284,331 Rogers, M. C. 1,148,990 Rogers, M. C. <td>Aug.</td> <td>4, 1896</td> <td>DI</td>	Aug.	4, 1896	DI
Saybolt C M 565.040	Aug.	4, 1896	Di
Saybolt C M 020 297	Apr.	18. 1911	J. K 2
Saybolt C M 918.066	July	29, 1879	N
	Aug.	9, 1881	N
	Jan.	13 1874	D1 J,K2 N D
Schalk Emil	Dec.	3, 1872	D.S
Schalk, Emil. 133,598 Schalk, Emil. 133,598 Schesch, H. A. 54,218 Scheuffgen, Robert. 1,118,952	Apr.	3, 1872 24, 1866 1, 1914	F
Scheuffgen, Robert	Dec.	1. 1914	H

UNITED STATES PETROLEUM	I PAT	ENTS-	Con.
Name Number	Da	te	Class
Schildhaus, G. & Condrea, C 950,184	Apr. 2	6, 1910	
Schildhaus, G. & Condrea, C 956,184 Schill, E		6, 1914	F, K 2
Schill, E1,142,275	June	8, 1915	J, K 2 V
Schiller, Max 580,652	Apr. 1	13, 1897	
Schmidt, A. T 164,694	June 9	22, 1875	D
Schmidt, W. A. and Wolcott, E. R. 1.308,161	June :	24, 1919	В
Schubert, Julius, 156.600	Nov.	3, 1874	A
Schwartz, Stephen, 1.247,883	Nov. 2	27, 1917	B
Schill, E. .1,142,275 Schiller, Max. .580,652 Schmidt, A. T. .164,694 Schmidt, W. A. and Wolcott, E. R. 1,308,161 .156,600 Schwartz, Stephen. .1,247,883 Scott, John B. .58,180 Seeger, Robert. .1,259,766 Seely, E. D. .57,390 Seibert, F. M. and Brady. J. D. .12,20,369	Sept. 1	18, 1866	M
Seeger Robert 1 259 786	Mar.	19, 1918	B
Seely E D 57 300	Aug.	21, 1866	M
Sooly, C. A	Feb.	1, 1000	F
Seely, C. A Drody, T. D. 1900.000		23, 1869	
Seely, C. A	Jan.	7, 1919	A
Seldenschur, F. & Dennst, J1,162,729	Nov.	30, 1915	В
Seigle, A 567,751	Sept.	15, 1896	F 1, II
Seigle, A 567,752	Sept.	15, 1896	F
Sellers, H. L. & Conyngton, H. R., 549,399	Nov.	5, 1895	E 3
Setzler, H. B1,292,966	Jan.	28, 1919	В
Sewell, B. F. Brooke 781.045	Jan.	31. 1905	F
Sexton, Wm, A	Dec.	4, 1917 21, 1884	A
Seymour, M. J., 306.965	Oct.	21, 1884	Ā
Shanter I S 61 474	Jan.	22, 1867	F 1, 2, 5
Shaw F D 1008/119	June	9 1014	K1, 2, 0
Shaw, F. D	June	2, 1914	
Shaw, G. E 01,572	Jan.	29, 1867	N
Snaw, G. E 56,107	July	3, 1866	N
Sheets, Earl H1,273,191	July 2	23, 1918	K 2, J
Sherman, L. O 968,088	Aug. 2	23, 1910	В
Sherman, L. O1,260,584	Mar.	26, 1918	B, J
Sherman, L. O	Dec. 2	24, 1918	B
Shiner, O. J		9, 1914	D1
Shively, Martin,		11, 1919	ŝ
Shreves F G	Nov	8 1898	Ă
Shroder Richard 16.255	Dec.	16, 1856	$\frac{A}{W}$
Slater Wm A 1962 050	Apr.	10, 1000	Ĭ
Slidmono C I and Coventur D E 1209.004	Apr.	23, 1918	
Skiumore, C. J. and Coventry, F. F.1,302,094	Apr.	29, 1919	0
Slemmer, Henry T 52,897	Feb.	27, 1866	0
Sloane, W. M 109,772	Nov.	29, 1870	A C
Sloane, W. M. & Potter, R. M 223,549		13, 1880	C
Sloane, W. M. & Bell, Wm 235,057	Nov.	30, 1880	Ĉ
Slocum, F. L. and Stutz, C. C 1,304,211	May 9	20, 1919	В
Slocum, F. L. and Stutz, C. C 1,304,212	May a	30, 1919	В
Small, H. J. & Stillman, H 595,788		21, 1897	D1, F2
Smedley, J. D	Feb.	17, 1863	S
Smith A D 1.239.423	Sept.	4, 1917	J, B
Smith C A 558.747	Apr.	21, 1896	TT D
Smith H C 300.811	June 9	24, 1884	F'II
Smith Hamilton T	Dec.	10 1004	E TT
	Dec.	18, 1866	S F 2, 4, I
Smith, H. L	Nov.	27, 1866	F 2; 4, 1
Smith, H. J. & Jones, W 35,184	wav	0. 1892	N C
Smith, Rollin H 306,653	Oct.	14, 1884	C
Smith, Wm	Apr.	19. 1890	G, S
Smith, Wm. A 596,437	Dec.	28, 1897	v
Smothers, H. F. & Norquist, E. E1,263,337	May	14, 1918	Q
Snee, J. A	Dec.	28, 1915	K_2
Snelling, W. O	Mar.	25, 1913	J. K 2. B
Snelling W 0	June	13, 1916	F 1
Shelling W O 1 215 732	Feb.	13, 1917	v T
Snow Wm B 120.668	Aug.	20, 1872	g
Show, Will. D 120,000	Ann	1 1072	g
Seely, C. A. 87,207 Seibert, F. M. and Brady, J. D. 1,290,369 Seidenschur, F. & Dehnst, J. 1,162,729 Seigle, A. 567,752 Sellers, H. L. & Conyngton, H. R. 549,899 Setzler, H. B. 1,249,760 Seymour, M. J. 306,965 Shapter, J. S. 61,474 Shaw, G. E. 61,572 Shaw, G. E. 61,677 Sherman, L. O. 968,088 Sherman, L. O. 1,268,088 Sherman, L. O. 1,268,088 Sherman, L. O. 1,268,050 Shidmore, C. J. 1,099,6622 Shively, Martin. 613,728 Shreves, F. G. 1,297,022 Shively, Martin. 163,728 Shroves, F. G. 1,297,022 Stidmore, C. J. and Coventry, P. F. 1,302,094 Siloane, W. M. & Potter, R. M. 225,657 Slocum, F. L. and Stutz, C. 1,304,212 Smail, H. J. & Stilliman, H. 599,788 <	Apr. Jan.	$\begin{array}{c} 1, \ 1873 \\ 18, \ 1918 \end{array}$	V Q K 2 J, K 2, B F 1 V S S F 2 V V
Souchulu and Doberg	Jall.	11 1004	TT 4
Sommer, Adolph 525,969	Sept.	11, 1894	v
Sommer, A	July	31, 1894	V
Southey, A. W	Dec.	15, 1914	K1
Spangle, George W 58,905	Oct.	16, 1866	D
Spears, Wm 107,734	Sept.	27, 1870	\mathbf{F}, \mathbf{G}
Spier, Robert & Mather, J 168,060	Sept.	21. 1875	U
Speller, F. N	Nov.	8, 1904	Ň
Snelling, W. O. 1,186,855 Snelling, W. O. 1,215,732 Snow, Wm. B. 130,668 Snow, Wm. B. 137,496 Soderlund and Boberg. 1,252,962 Sommer, Adolph. 525,969 Sommer, A. 523,716 Southey, A. W. 1,120,857 Spangle, George W. 58,905 Spears, Wm. 107,734 Spier, Robert & Mather, J. 168,060 Speller, F. N. 774,341 Squires, Frederick. 1,249,232 Squire, F. B. 197,197	1)00	1 1017	J, K 2 N
Squire, F. B	Nov.	13, 1877	N
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UNITED STATES	PETROLEU	M PATENTS-	-Con.
Name	Number	Date	Class
Stafford, Jas. R		Ann 05 1954	
Starke, Eric A	597,920	Jan. 25, 1898	U.
Starles E A	701 010	Jan. 31, 1905	D1 De D
Stanley, A. M		Apr. 4, 1916	E 3, B K 1
Starke, E. A.	913.780	Mar. 2, 1909	D, F 2
Starke, E. A	1.109.187	Sept. 1, 1914	D, 1 2
Stearns, H. A	103,385	May 24, 1870	F 2, 11
Starke, E. A. Starke, E. A. Starke, E. A. Stearns, H. A. Steenbergh, B. van.	1,124,364	Jan. 12, 1915	K1, B
Steinschneider, ——	1,302,988	May 6, 1919 Jan. 17, 1911 July 25, 1916	S
Steinschneider, Leo	981,953	Jan. 17, 1911	F 5
Steinschneider, Leo.	1,192,581	July 25, 1916	F 5
Stelwagon, W. H	503,996	Aug. 29, 1893	S
Stevens, Levi.	363,432	May 24, 1887	F 2
Stevens, Levi.	414,601	Nov. 5. 1889	B
Stevens, will. n	1,100,402	Dec. 28, 1915	M
Stowart, Julii L	169 065		W F 2. II
Stewart J L & Logan J P	112 811	May 4, 1875 Apr. 18, 1871	F 2, II F
Stewart, J. L. & Dubler, J. B.	136 557	Mar. 4, 1873	S
Stewart, Lyman,	1.163.570	Dec. 7, 1915	B
Still Carl	1.080.177	Dec. 2, 1913	S
Stombs, D. S. & Brace, J	27.842	Apr. 10, 1860	SG
Stone, C. W	1.070.555	AUS. 10, 1010	Ă
Stott, Chas	68,257	Aug. 19, 1867	F 1, 2
Strache, H. & Porges, P	1,205,578	Nov. 21, 1916	B
Strain, E. W	311,543	Feb. 3. 1883	F 1, 2, II
Street, G. E. J	70,715	Mar. 11. 1902	M
Stringfellow, John H. W	454,777	June 23, 1891	D
Stuber, John, Stuber, Jacob &	300 511		
Mager, John W	123,741	Feb. 13, 1872	F 1, 2, II
Suckert, Junus J	1 1 9 1 6 0	Feb. 19, 1895	V
Stearns, H. A. Steenbergh, B. van Steinschneider, <u>Leo</u> . Steinschneider, Leo. Steinschneider, Leo. Steinschneider, Leo. Stevens, Levi. Stevens, Levi. Stevens, Levi. Stewart, John L. Stewart, J. L. & Logan, J. P. Stewart, Lyman. Still, Carl. Stonbe, D. S. & Brace, J. Stone, C. W. Stott, Chas. Strache, H. & Porges, P. Strain, E. W. Strache, H. & Porges, P. Straingfellow, John H. W. Suber, John, Stuber, Jacob & Mager, John W. Suckert, Julius J. Subr, C. L. Swan, O. C.	1 950 596	Dec. 22, 1914 Dec. 18, 1917	F 2, II
Swan O C	1 983 045	Nov. 12, 1918	A S B
Swaton I A	1 260 731	Mor 96 1018	B
Svlvester F	68,669	Sept. 10, 1867	A
Suhr, C. L. Swan, O. C. Swan, O. C. Swaton, J. A. Swaton, J. A. Sylvester, F. Symonds, D. Tagliabue, Chas. J. Tagliabue, Chas. J. Tagliabue, Giuseppe. Tagliabue, Giuseppe. Tagliabue, Giuseppe. Tagliabue, Giuseppe. Tagliabue, John. Tait, A. H. Tait, E. W. Tait, G. M. S. Tait, A. H. & Avis, J. W. Tait & Avis. Tait & Avis.	26.000	Mar. 20, 1918 Sept. 10, 1867 Nov. 1, 1859 May 28, 1867 May 28, 1867 Oct 3, 1882	A G V V
Symonds, D	65,136	May 28, 1867	v
Symonds, D	65,137	May 28, 1867	
Tagliabue, Chas. J.	265,462	000 0, 2002	F 1, 2, 3, 4, II
Tagliabue, Chas. J.	254,176	Feb. 28, 1882	F 1, 2, 11
Tagliabue, Giuseppe.	36,826	Oct. 28, 1862	N
Tagliabue, Chas. J.	1,263,145	Apr. 16, 1918	NNNS
Tagliabue, Gluseppe	30,421	May 5, 1863 Sept. 16, 1862	N
Tagliabue, Joill	06 007	Sept. 16, 1862 Nov. 16, 1869	C C
Tait E W	1 069 908	Nov. 16, 1869 Aug. 12, 1913	
Tait G M S	1 128 549	Feb. 16, 1915	J, K 2 K 1, B
Tait, A. H. & Avis, J. W.	53.359	Mar. 20, 1866	F 2, 3, II
Tait & Avis.	63,115	Mar. 19, 1867	F1
Tait & Avis.	135,673	Feb. 11, 1873	F 2, II
		Feb. 8, 1870	D1
Tatro, Jos. A. Tatro, Jos. A. Taveau, Rene de M. Taylor, H. K. & Graham, D. M. Taylor & Graham	106,233	Aug. 9, 1870	D1
Taveau, Rene de M	1,271,387	July 2, 1918	I,E1
Taylor, H. K. & Graham, D. M.	54,978	May 22, 1866	D1
Taylor & Graham	59,751	Nov. 20, 1866	D1
Tempere, Albert J Testelin, A. & Rehard, G	557,291	Mar. 31, 1896	V, D
Testellin, A. & Renard, G	1,138,200	May 4, 1915 Dec. 31, 1895	BF
Theisen, Eduard.	002,400	Dec. 31, 1895	F
Thiele Felix C	683 354	Sept. 24, 1901	D1
Thiele, Felix Carl	1 254 866	Jan. 29, 1918	Ĩ.
Theisen, Eduard. Theisen, Eduard. Thiele, Felix C. Thiele, Felix Carl. Thirault, A.	61,120	Jan. 8, 1867	F 4, II
Thirault, A.	41.871	Mar. 8, 1864	
Thirault, A.	63,963	Apr. 16, 1867	F2
Thomas, John J	178,889	June 20, 1876	S
Thomas, Joshua	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	July 31, 1883	FII
Thomas, Joshua	314,490	Mar. 24, 1885	F2
Thirault, A. Thomas, John J. Thomas, Joshua. Thomas, Joshua. Thomas, Richard. Thompson, W. P.	781,854	Feb. 7, 1905	Sp
Thompson, W. P	1,160,670	Nov. 16, 1915	В

UNITED STATES PETROLEUM PATENTS—Con.NameNumberDateClassThursby, John, F $\$0.067$ May 2, 1888M.Thursby, John, F $\$0.067$ May 2, 1885D1Temann, J. H $\$0.060$, 644JulyT. 1885D1Tienen, W. O, Th. van. $1.000, 646$ Aug. 15, 1911F1Timmons, J. R. $1.000, 646$ Aug. 15, 1911F1, HTimmons, J. R. $1.000, 646$ Aug. 15, 1917F2, 4Timmons, J. R. $1.248, 651$ Dec. 4, 1917AToopan, Chass $498, 588$ May 30, 1893DTravers, W. J. $1.000, 610$ Sept. 26, 1911AToppan, Chas, M. J. $1.000, 610$ May 30, 1893DTravers, W. J. $1.000, 610$ May 30, 1893DTravers, W. J. $1.000, 610$ May 6, 1911ATravers, W. J. $1.000, 610$ May 6, 1911F2, 111Trumble, M. J. $1.250, 652$ Dec. 11, 1917F3, 111Trumble, M. J. $1.250, 652$ Dec. 11, 1917F3, 111Trumble, M. J. $1.250, 652$ Dec. 10, 1912BTrumble, M. J. $1.250, 652$ Dec. 10, 1918F2, 511<						
Name		Number	Date		Class	
Thumm, Chas. F		389,988	Sept. 25,	1888	F 2, 4	
Thursby, John		3,067	May 2	$1843 \\ 1885$	M	
Tiemann, Julius H	• • • • • • • • •	321,465	July 7, Nov. 17,	$1885 \\ 1885$	D1 D1	
Tienen, W. O. Th. va	n	1.000.646	Aug. 15,	1911	Ĩ	
Timmons, J. R		1,105,383	July 28	$\begin{array}{r}1914\\1916\end{array}$	F 1, II	
Timmons & Swain, O.		1,179,243	Apr. 11	1916	F 1	
Timmons I B	• • • • • • • • • •	1 204,943	Nov. 12 Sept. 24	1018	F 2, 4 A	
Tokheim, J. J		1,248,951	Dec. 4	1878 1918 1917 1893	A	
Toppan, Chas		498,588	May 30	1893	D	
Travers, W. J.		1,004,219	Sept. 26	1911	A F 2	
Trumble Milton J	ег, н . w	996.736	Jan. 31 July 4	1.911	S, F	
Trumble, M. J.		1,002,474	Sept. 5	1911	F 1, 11	
Trumble, M. J		1,070,361			F2 II	
Trumble, M. J.	•••••	1,182,601	May 9 Dec. 11	$\begin{array}{r}1916\\1917\end{array}$	E 2, F 2, II F, S	
Trumble, M. J.		1,259,052	Mar. 12	1918	F 2, A	
Trumble, M. J		1,260,598	Mar. 26	1918 1918	F	
Trumble, M. J		1,262,875	Apr. 16	, 1918	F	
Trumble, M. J.	•••••	1 046 682	June 11 Dec. 10	1918	K 2, S B	
Turner, C. W.		1.151.422	Aug. 24	, 1912 , 1915	B	
Turner, R. D		194,275	Aug. 14	1877	Ā, V S	
Thompson, N. W	•••••	1,298,602	Mar. 25	1877 1919	S	
Trumble, M. J.	••••	1 204 124	May 20 May 20	1919 1919	B A	
Trumble, M. J.		1.281.884	Oct. 15	. 1918	B	
Turner, R. D		154,430	Aug. 25.	1874	Ã	
Turner, R. D		156,899	Nov. 17	, 1874	A S, F D	
Tweddle, Herbert W. (3	120,349	Oct. 24 Apr. 10	1871 1877	D T	
Tweddle, H. W. C.		189,402	Apr. 10	1877	T T	
Tweddle, H. W. C		45,363	Dec. 6	$ \begin{array}{r} 1864 \\ 1867 \\ 1867 \end{array} $	K 2	
Tweddle, H. W. C		72,125	Dec. 10	, 1867	F 2, 5, II F 2, 5, II G, F 2, 5, II	
Tweddle, H. W. C		72,126	Dec. 10 Feb. 4	, 1867	G. F 2, 5, 11	
Tyler, Chas. N.		38.015	Mar. 24	1863	M 1 2, 0, 1	
Ujhely, Heinrich		289,788	Dec. 4	. 1883	D	
Ujhely, H. & Buerle, C	3	131,137	Sept. 3	, 1872	C	
Van Devort C & Van	Fleet C	168 549	Jan. 9 Oct. 5	1894, 1875	E 3 F 2	
Van Dyke, J. & Irish,	Wm	1,095,438	May 5	, 1914	B	
Van Dyke & Irish		1,073,548	Sept. 16	, 1913	В	
Van Dyke & Irish		1,143,466	June 15 Mar. 9	, 1915	B B	
Van Vliet L & O'Neil	я	1 094 762	Apr. 28	, 1915 , 1914	K 1	
Vander Weyde, Peter	H	104,798	June 28	. 1870	Ň	
Vander Weyde, P. H.		61,125	Jan. 8	. 1867	A	
Vander Weyde, P. H.		58,005	Sept. 11	, 1866	F 2, 4, 5, II F	
Vander Weyde, P. H.		53 062	Mar. 6	, 1866 . 1866	F	
Van Syckel, Samuel.		191,203	May 22	. 1877	F, II F 2	
Van Syckel, S		140,801	July 15	. 1873	F 2	
Van Syckel, S		152,440	June 23 May 7	, 1874	F, II S	
Van Syckel, S.		154,772	Sept. 8	, 1872 , 1874	F, II	
Van Syckel, S		154,771	Sept. 8	. 1874	U	
Van Syckel, S		143,945	Oct. 21	. 1873	K 2	
Van Syckel, S		101 204	Dec. 27 May 22	, 1870 , 1877	F 2, I F, II	
Van Tine, Henry C.		60,290	Dec. 4	. 1866	D .	
Van Wyck, C. I		27,603	Mar. 20	, 1860 , 1867	W	
Van Wyck, William. ,		65,313	May 28	, 1867	S	
Vaughan, Aaron C		49 680	Apr. 3	, 1866 , 1865	G F 1, 2, II	
Von Boyen, Edgar.		689.381	Dec. 24	. 1901	C C	
Von Boyen, Edgar		690,693	Jan. 7 Sept. 17	, 1902		
vuilleumier, Rudolph.		1,038,691	Sept. 17	, 1912	К 1, В	

UNITED STATES PETROLEUM	M PA	TENTS-	-Con.
Name Number	D	ate	Class
Von Groeling, A. F. G. P. J	Feb.		В
Waddell Alexander 1940.644		18, 1919	
Waddell, Alexander	Dec.	11, 1917	K1
Waltz, J. W	Aug.	4, 1914	J, K 2
Walker, Henry V	Oct.	18, 1910	D 1
Walker, H. V	Apr.	19, 1910	V
Walker, H. V. 955,372 Walker, W. E. 1,307,280 Wallace, Geo. W. 1,283,000 Wallace, John Stewart. 716,132 Warden, Henry. 266,929 Warden, Wm. G. 240,937 Warden, Wm. G. 250,936 Warden, Wm. G. 110,806	June	17, 1919	L, K
Wallace, Geo. W	Oct.	29, 1918	Ŵ
Wallace, John Stewart	Dec.	16, 1902	D
Warden, Henry	Oct.	31, 1882	C S S, D F 1, II
Warden, Wm. G	May	3, 1881	S
Warden, Wm. G	May	3, 1881	S, D
	Jan.	3, 1871	F 1, II
	Mar.	14, 1871	F 1
Warfield, R. N. 40,068 Warfield, R. N. 40,068 Waring, Richard S. 284,098 Waring, Wilson. 643,578 Warren, Cyrus M. 248,074 Warren, Cyrus M. 47,225 Warren Lohn 97,008	Sept.	22. 1863	F 1 V
Waring, Richard S 284,098	Aug.	28, 1883	Ť
Waring, Wilson	Feb.	13, 1900	I
Warren, Cyrus M	Oct.	11, 1881	Î
Warren, Cyrus M	Apr.	11, 1865	ĪĪ
Warren, Cyrus M	Dec.	14, 1869	Ŧ
Warren John 102186	Apr.	19, 1870	F S V
Warren John W 705 168	July	22, 1902	v
Warren John W 666 446	Jan.	22, 1901	v
Warren M H 1110 261	Sept.	15 1014	
Worth (U 1121 900	Mar.	15, 1914 16, 1915	B F 2, II, G
Walth, C. H	May	10, 1915	B 2, 11, 0
Weigenhorgen D		4, 1915	D 1
Weisenberger, F	May Feb.	22, 1866	
Weiser, Josef		9, 1915	S
Weilman, F. E	Aug.	13, 1918	B
Wells, A. A	July	3, 1917	B
Warren, John. 97,998 Warren, John N. 102,186 Warren, John W. 705,168 Warren, John W. 666,446 Warren, M. H. 1,110,361 Warth, C. H. 1,131,880 Washburn, C. H. 1,138,266 Weisenberger, P. 54,984 Weiser, Josef. 1,127,951 Wells, A. A. 1,232,454 Wells, A. A. 1,248,225	June	20, 1916	B
Wells, A. A	Nov.	27, 1917	B, J
Wells, Willet C. & Wells, F. E 877,620	Jan.	28, 1908	F 1, 3, II
Wells, W. C. & F. E	Mar.	4, 1919	F
Wells, A. A. 1,187,874 Wells, A. A. 1,248,225 Wells, Willet C. & Wells, F. E. 877,620 Wells, W. C. & F. E. 1,296,224 Wells, W. C. & F. E. 61,291 Wellman, F. E. 1,245,291 Welsh, M. J. 1,159,450 Wemple, H. R. 1,262,886 Wendtland, August. 618,307 Weston, Elijah. 219,546 Wetmore, I. W. 39,978	Jan.	15, 1867	S
Wellman, F. E	Nov.	6, 1917	B, S
Welsh, M. J	Nov.	9, 1915	C
Wemple, H. R	Apr.	16, 1918	K 1
Wendtland, August 618,307	Jan.	24, 1899	C
Weston, Elijah	Sept.	9, 1879	S
Wetmore, I. W. 39,978 Wheeler, Norman W. 52,477 Whitall, Frank M. 768,101 Whitall, Grane D. 768,101	Sept.	15, 1863	U
Wheeler, Norman W	Feb.	6, 1866	S T
Whitall, Frank M	Aug.	24, 1904	T
Whitall, Samuel R	July -	21, 1903	T
wnitall, Samuel R. 734,452 White, Carter. 1,226,041 Whiting, Jas. R. 622,936 Whiting, J. R. & Lawrence, W. A. 583,779 Whitmore, Samuel W. 1,125,422 Wiegand, S. Lloyd. 39,607 Wiegins, Isaac B. 63,777 Wilber, William 63,777	May	15, 1917	B
Whiting Jas R 622,936	Apr.	11, 1890	S
Whiting J. B. & Lawrence W. A. 583,779	June	1 1907	s v
Whitmore Samuel W 1125422	Jan.	19, 1915	F 1, II
Wiegand S Lloyd 39 607	Aug.	18, 1863	F 2
Wiegand S Lloyd 62 583	Mar.	5, 1867	Ĉ -
Wiggins Isaac B	Apr.	9, 1867	M
Wilher William 23 210	Mar.	8, 1859	M
Wilcox L N 49.020	July	25, 1865	F
Wilkinson Asa W 145 707	Dec.	19, 1873	F 3
Wilkinson Walter S 519 249	Jan.	9, 1894	E 3
Wilkinson, Walter S	Jan.	25, 1898	E 3
Willard Franklin W 96 720	Jan.	3, 1880	G, S
Willard Franklin W	Mar.	13, 1860	F
Willard Enonlylin W.		28, 1860	G. S
Williams D A & Drogg T 204 200	Feb. Sept.	22, 1884	S, S
Willia Coo M 010 200		20, 1909	E 3
Wilson D T	Apr.		F4
Wingett John M	Mar.		P *
Wingert, John N	June	5, 1917	Ď
Wintz, Jas. P	Dec.	19, 1905	D1
wirkner, George von	Feb.	28, 1905	D
Wolff, Albert	Sept.	28, 1917	D1
Wolf, Hermann	May	17, 1898	K1
Wolf, Linus	May	7, 1918	DI
Wonie, Salo	Dec.	16, 1913 20, 1908	D
Wiggins, Isaac B. 63,777 Wilegins, Isaac B. 63,777 Wilex, L. N. 23,210 Wilcox, L. N. 49,020 Wilkinson, Asa W. 145,707 Wilkinson, Walter S. 512,348 Wilkinson, Walter S. 597,882 Wilkinson, Walter S. 597,882 Willard, Franklin W. 26,739 Willard, Franklin W. 27,327 Willard, Franklin W. 27,327 Willison, R. J. 379,090 Wingett, John N. 1,229,189 Wintr, Jas. P. 807,983 Wirkner, George von. 783,916 Wolff, Albert. 1,240,523 Wolf, Hermann. 604,280 Wolf, Linus. 1,265,573 Wohne, Edward W. 901,411	Oct.	20, 1908	-

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Name	Number	Date	Class
Wallace, Geo. W		Oct. 29, 1918	
Wells, Raymond	.1.267.611	May 28, 1918	A
Wright, E. H. & Atwood, E. H		Sept. 10, 1918	F
Whitman, J. C	.1,312,375	Aug. 5, 1919	
Yaryan, Homer T.	. 300,185	June 10, 1884	F 2, 5, II
Young, Jas	. 127,446	June 4, 1872	
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INDEX.

	Page
Abel tester	318
Abel-Pensky tester.	318
Absorption	
Gasoline produced by absorption process	23
Specifications for straw oil for benzol absorption	155
Absorption process for casinghead gasoline	155
Relation of absorption test, specific gravity and vield of	448
gasoline by compression	248
Absorption method of testing gas for gasoline content	354
Literature on absorption processes	61-462
Absorption Gasoline produced by absorption process. Specifications for straw oil for benzol absorption. Specifications for absorption oil. Absorption process for casinghead gasoline. Relation of absorption test, specific gravity and yield of gasoline by compression. Absorption method of testing gas for gasoline content. Literature on absorption processes. Accumulation of oil and gas in anticlines, synclines and faults	7
Uset of combustion of	90-254
Acid (see sludge acid) Treatment of benzine with acid and caustic soda Method of removing sulphonic acid Acid in lubricating oil	254
Explosions of	252
Acta (see sludge acta)	101
Treatment of sludge acid	131
Method of removing sulphonic acid	138
Acid in lubricating oil Determination of free acid	161
Determination of free acid	338
Adhesive tests of rope lubricants Aeroplanes	173
Fighting grade of aeroplane gasoline	141
Navy specifications for aeroplane gasoline	141
U. S. Navy specifications for Liberty aero oil	168
General specifications for aviation gasoline	139
Specifications for aeroplane machine gun oil	169
Aeroplane oil	175
Aggregates In asphalt pavements.	192-5
For asphaltic concrete	192-5
For asphaltic concrete For Topeka specifications	92-5-7
For bituminous concrete	92-5-7
For sheet asphalt	192-5
For hinder course	192-5
Of rock asphalt For binder course. Tables for calculating the voids in aggregates from weight	101
per cubic foot	201
Methods of grading aggregates of asphaltic surface mixture	345-9
Agitation and agitators	131 184
Air required for burning fuel oil Ultimate composition of air blown residuum	128
Air blowing of asphalt	191-3
Air blowing of asphalt Specific heat of air	254
Alabama	
Summary of kerosene inspection laws	156
Alcohol Heat of combustion of grain alcohol and wood alcohol	190
Use of alcohol with benzol	146
Allen County, Kansas,	-
Properties of crude oil from Fractional gravity distillation analysis of crude oil from	120
Fractional gravity distillation analysis of crude oil from	127
Aluminum Chloride Use of in removing color and odor	136
Use of in cracking oil.	215
Use of in cracking oil	
Distillation apparatus of for mineral spirits	306
Flash Dollit test	313
Softening point of hituminous materials	301-4 297-8
Carbon test	322
	39-40
Ammonia	
Lifting power of ammonia gas compared with other gases in	0.10
balloons	249
Ammonium sulphate in oil shale	234

Pa	ge
Analysis Outline of methods of analysis of petroleum products 269 Index to applications of methods of analysis	9-70 271
Mathods of analysis	
Of crude petroleum, gasoline, benzine, naphtha, kerosene, il- iuminating oil, straw oil, absorption oil, lubricants, paraf- fin oil fuel oil Dissel parine oil road oil fur oil sanhalt	
pitch, wax, grease, asphaltic surface mix	271)
Viscosity	297
Cold tests	2-4
Distillation tests	312
Melting point and softening point	325
Heat of combustion and calorimeter tests	332
Ultimate analysis	34-5 335
refining	6-8
Free acid Free acid Floc test Corrosion and gumming tests. Penetration and consistency tests. Sast	
Loss on heating. Amount of asphalt. Asphaltenes and solubility in petroleum ether.	341 341 342
Loss on heating. Amount of asphalt. Asphaltenes and solubility in petroleum ether	342 844
Bitumen and grading of asphaltic mineral mixtures	8-9
Specific gravity of gas	2-3
Complete chemical analysis of gas by the Burrell-Orsat apparatus	8-9
Determination of heat of combustion of gas	5-6
Anticlines and domes. Relation of anticlines to surface topography. Illustration showing conditions for accumulation of oil, gas	45
and water in anticlines	7
Geologic occurrence of oil in the Appalachian field Outline map of oil fields and pipe lines of Appalachian district	6 21
	223 430
Argentina	-11
Arizona Kerosene regulations of	156
	156 143
A pomption (see also have and haveal)	
Chemical properties of aromatic hydrocarbons Aromatic compounds in gasoline Graphic representation of relation of gravity to distilling temperature of aromatics	138
Method of determining aromatics in petroleum	6-8

		Page
st	Determination of in bituminous materials	324
st	halt Significance of asphalt outcrops Refinery operations on crude oil in 1918 with amount of oil treated and amount of asphalt, etc., produced	5 32
	Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in	128
	Trinidad asphalt and paraffin base oil. Definition of asphalt and paraffin base oil. Definition of asphalt. Heating value of hard asphalt. Oil suitable for asphalt manufacture. Definition of asphalt pavements. Of asphaltic concrete. Of sheet asphalt.	129
	Definition of asphalt	132-192
	Oil suitable for asphalt manufacture	192
	Definition of asphalt pavements	192
	Of sheet asphalt.	192
		193 193
	Composition of natural asphalt	193
	Composition of Cholin. Mexicall, California, Stanoniu appliate	193 193
	Composition of oil asphalt	193
	Of pressure still asphalt Of pressure still asphalt	193 194
	Of asphaltic sandstone, of asphaltic limestone Of asphaltic limestone from Siciliy, France, Missouri and	194
	Of asphaltic limestone from Siciliy, France, Missouri and Oklahoma	194
	Of asphaltic sandstone from Kentucky, Oklahoma, Missouri	194
	Composition of asphalt pavement	195 195
	Of Topeka specification asphalt Of sheet asphalt	195
	Of sheet asphalt. Of binder course for sheet asphalt. Composition of top course for sheet asphalt.	195 195
	Standard composition of sneet asphalt	196
	Material required for asphaltic concrete Relations of defects of an asphalt pavement to its physical	196
	properties	196
	Cracking of asphalt pavement Disintegration and hole formation of asphalt pavements	196 196
	Scaling of asphalt surface mixture	196
	Scaling of asphalt surface mixture	196
	Functions of various constituents of asphaltic surface mixture	196
	Filler in asphalt pavement	196 197
	Filler in asphalt pavement. Specifications for asphaltic cement Specifications for asphaltic surface mixture. Epitome of the purposes of the specifications for asphaltic	197
	Epitome of the purposes of the specifications for asphaltic	198
	Ductility of asphaltic cement.	198
	Ductility of asphaltic cement Viscosity of asphaltic cement Effect of mineral matter on penetration of asphaltic cement.	199
	Fluxing of hard asphalt	199 199
	Fluxing of hard asphalt Typical specifications for the wearing surface of asphaltic	202
	concrete	202
	Mathada of applying asphaltic surface mixture	202
	Asphalt filler for brick, Mexican and Texaco type Asphalt filler for brick, Sarco type Asphalt produced in cracking oil Method of determining the specific gravity of asphaltic	207 208
	Asphalt produced in cracking oil.	210
	cement.	274
	Method of determining the specific gravity of asphaltic	0.5.4
	surface mixtures and solid asphaltic materials Illustration of methods of determining the specific gravity of	274
	asphaltic cement	277
	Viscosity of asphalt	294
	Float test of asphalt. Melting point of asphaltic cement by the ring and ball, the	294
	cube and the general electric methods	297-300

	Decre
Flash point determination of asphalt	13-316
Carbon and ash in asphaltic cement	324-5
Determination of sulphur in asphalt	332
Determination of carbon, hydrogen and nitrogen in asphalt.	333
Method of determining the penetration of asphalt Method of determining loss on heating and of penetration of	339
Method of determining loss on heating and of penetration of	341
oil and asphalt compounds Determination of amount of asphalt in oil	341
Determination of ductility of asphalt	342
Determination of asphaltenes and solubility	342
Determination of asphaltenes and solubility Apparatus for determination of loss on heating tests and per-	
centage of asphalt in bituminous materials	343
Method of determining the resistance of asphaltic cement to	
oxidation Determination of bitumen and grading in asphaltic surface	344
Determination of bitumen and grading in asphaltic surface	345
mixtures by burning and by extraction Paraffin scale in asphalt Method of determining tensile strength of asphaltic surface	347
Method of determining tensile strength of asphaltic surface	0
mixtures	348
Illustration of apparatus required for determination of bitu-	
men in asphaltic surface mixtures	349
Illustration of apparatus for determination of grading of	010
asphaltic surface mixtures	349
Asphaltnes-determination of.	342
Association of salt water, oil and gas	4
Atomization of fuel oil with steam	184
Atwood process for cracking	217
Automobiles	
Demand for petroleum for automobiles	2
Gasoline consumed by automobiles Number of automobiles in the United States Properties of automobile oil	2
Number of automobiles in the United States	200 1 7
Axle oil	175
Baku, Russia	1
Graphic illustration of composition of crude oil from	121
Ball and ring	
Determination of melting point or softening point of bitumin- ous materials by the ring and ball method	007 0
	297-8
Barbour, E. H	466
Barometric pressure	
Correction tables on flash points for barometric pressure	317
Barrel	
Capacity of petroleum barrel	99
Barrett, E	466
Barton, J, E. Base of petroleum. Baume' (see specific gravity and gravity)	466
Base of petroleum	129
Banton experience gravity and gravity)	216
Benton apparatus for cracking oil Benton process for cracking oil	217
Fractional gravity distillation of gasoline made by the Benton	
cracking process	231
Benzene (see Benzol)	
Benzine	
Definition of	131
Treatment of benzine with acids and caustic	131
Stream gravity of benzine from various crude oils	131
Definition of Treatment of benzine with acids and caustic. Stream gravity of benzine from various crude oils. Specifications for benzine or petroleum ether of U. S. P	145
Renzol manufactured in U.S. and Canada	80-1
Ultimate composition of Chemical properties of benzene hydrocarbons	128
Properties of benzol	129
Properties of benzol Heat of combustion of	138
Heat of combustion of Comparison of gasoline and benzol as motor fuel	146
	146
Rate of evaporation of benzol	146
Solubility of wax in benzol. Fractional gravity distillation of benzol.	180
Benzol from light oils of gog works	232
Benzol from light oils of gas works Penzol in manufactured gases	241

	Page
Bermudez asphalt, composition of By-product coke plants and benzol manufacturers in U. S. and	193
By-product coke plants and benzol manufacturers in U. S. and Canada	80-1
Billings, Okla., crude oil, fractional gravity distillation analysis of	125
Binder course for sheet asphalt	195
Bitumen (see asphalt) Determination of bitumen and grading in asphaltic surface mixture	345-8
Bituminous materials (see asphalt)	007
Determination of fixed carbon and ash	297 324
Determination of fixed carbon and ash Determination of fixed carbon and ash	
in	343 340
Determination of ductility Determination of wax	344
Bituminous concrete	92-195
Bituminous surface mixture1	92 - 195
Bituminous earth	123
Blocks, asphalt filler for Bloom or fluorescence in petroleum products and methods of re-	207-8
Bloom of huorescence in petroleum products and methods of re- moval	136
Blown oil	175
Blown asphalt Boiling point (see temperature and distilling) of gaseous, liquid and solid paraffin hydrocarbons giving boiling point	193
and solid parafin hydrocarbons giving boiling point	130
Bolt oil	175
Books relating to petroleum, asphalt and natural gas	459-60 305
Bownocker, J. A	466
Brick filler	207-8
Brick plants, consumption of natural gas by Brine	250
Gravity, strength and freezing temperature of brine and cal-	
cium chloride solutions	427 466
Brown, C. W Buehler, H. A. Bunker oil, specifications for	466
Bunker oil, specifications for	154
Bureau of Mines Methods of distilling gasoline	306
Eulletins of	462 461
Bureau of Standards	401
Scale for converting Baume to specific gravity	272
Hydrometers	272 287-9
Hydrometers Tables for conversion of viscosity readings Publications of, on petroleum, asphalt and natural gas	462
Burkburnett, Texas, crude oil, fractional gravity distillation analy-	100 0
sis of	120-2
Specifications for long time burning oil	150
Burning tests for kerosene	151
For mineral seal oil For signal oil	152 153
Method of determining flash and burning points of lubricants	
and other heavy petroleum products	315
Burning tests by the Cleveland open cup method	316
Burrell-Orsat apparatus for analysis of gasBurton	358
Composition of gas obtained from Burton cracking stills	210
Burton apparatus for cracking oil Description of Burton process with modifications	220 221
Rutane, properties of	130
Butane, properties of In natural gas Heat of combustion and ignition temperatures of	245
Heat of combustion and ignition temperatures of Byerlite, ultimate analysis of Byerlite pitch	254 128

	Page
Cabin Creek petroleum, properties of	120 132 427
Calcium Chloride, gravity, strength and freezing temperature of	441
California	6
Geologic occurrence of oil in California fields Outline map of oil fields and pipe lines of Daily production of crude oil by pools in	6 17 25
Duling map of on a foude of ly nols in	25
Oil oushers in	30
Oil gushers in Prices of crude oil in.	35
Large producers of oil in Important oil companies of.	40
Important oil companies of	41
Properties of California crude oll Ultimate composition of California crude oils	120-1
Ultimate composition of California crude olls	128
Color of cracked gasoline from California crude oil	$136 \\ 143-4$
Kerosene inspection laws of California	156
Tiddact of tomponeture on viscosity of lubricants made from	100
California crude oil	163
California fuel oil.	181
Gravity and flash point of California crude oil	187
Composition of California asphalt	193
Calorific value (see heat of combustion)	
Of petroleum	330
Method of determining calorific value of gas by the gas	0.04
calorimeter	361
Calorimeter	0.01
Illustration of bomb calorimeters	331 331
Determination of sulphur by the bomb calorimeter	331
Parr calorimeter	334
Gas calorimeter	361
Cambrian Series	901
Stratigraphic section of rocks in	10-11
Canada, production of petroleum in	12
Canada, production of petroleum in By-product coke plants and benzol manufacturers	80-1
Cannel coal	3-235
Capacity	
Formula for cylindrical horizontal tanks Capacity tables for cylindrical horizontal tanks	100
Capacity tables for cylindrical horizontal tanks	101
Capacity tables and gauging tables for tank cars	104
Car oil	175
Cars (see tank cars)	342
Carbons in asphalt Carbon	344
Amount of in asphalt	128
Amount of in asphalt In various crude petroleums and their products	131
In lubricating oil	161
In lubricating oil. Produced by cracking. Composition of still carbon from cracking stills	209
Composition of still carbon from cracking stills	223
Method of determining carbon residue in lubricating oil by the	
Conradson method	322
Conradson method Apparatus for Conradson carbon test	323
Determination of carbon in petroleum products	333-4
Carbon black from natural gas	249
Properties of, uses of. Theoretical and actual amount obtained.	249
Theoretical and actual amount obtained	249
Carbon bisulphide, determination of the solubility of bituminous	342
materials Carbon dioxide in stack gases from fuel oil furnaces	184
In gas	246
Carbon monoxide in gas	246
Carbon monoxide in gas	254
Explosibility of	252
Relation of oil to carboniferous period Character of oil from carboniferous formation	4
Character of oil from carboniferous formation	4
Carbon tetrachloride, solubility of asphalt in	342
Carbon tetracher of off from carbon ferous formation	0.0
Production of, by absorption and compression	23
List of essinghead geoline plants	24
Casinghead gesoling in solution to other manufactor	61-64 138
Gasoline carried by natural cas	247
Cost of agginghood gagoling plants	241

	Page
Absorption process for Properties of hydrocarbons found in Testing the capacity of casinghead gas wells with the orifice	248 251
Method of determining the vapor pressure of	253 321
Castor oil Viscosity of lubricating oil made from Effect of temperature on viscosity of	163 163
Catalysis in cracking of oil	215 360 360
In treatment of benzine Cement (see asphaltic cement) Cement plants, natural gas consumed by	131
Chain lubricants.	171
Chemical properties (see specifications and special subject) of naphthene hydrocarbons Of various crude oils	$128 \\ -8, 131 \\ 128 - 30 \\ 251 - 2$
Of aromatic hydrocarbons. Chemical nature of the cracking of oil	129 209
Chlorine for removal of color and odor Chromometer, Sayboit Universal	136 278
For color of refined oil. Comparison of chromometer color with potassium bichromate solution. Claroline oil.	280 155
Cleveland flashtester	313-6 304
For signal oil For mineral seal oil Coal oil (see kerosene)	153 152
Coal	
Comparison of heat of combustion of coal with petroleum and natural gas. Equivalents of coal and fuel oil.	181
Equivalents of coal and fuel oil. Equivalents of coal and fuel oil. Heating value of bituminous coal and cannel coal. Heat of combustion of anthracite coal. Resemblance of cannel coal to oil shale. Properties of cannel coal.	190 190 233
Coal gas, heat of combustion of	235 181-6
Explosions of coal gas Tables for calculating the orifice capacities of coal gas26	252 6-268
Coke produced in refinery operations on crude oil	32 80-1
Coke produced in refinery operations on crude oil. By-product coke plants. Definition of coke. Coking stills. Cold test.	132 176
Color of refined petroleum	136
Effect of cracking on color	136 136
Color of cracked gasoline Shrinkage in removing color	136 136 161
Shrinkage in removing color. In lubricants. Method of determining color with the Saybolt Chromometer With the Lovibond Tintometer. By the potassium bichromate solution.	$278 \\ 279$
By the loume solution	280 282-3
Colorado— Gasoline inspection laws of Kerosene inspection laws of	143 156
Oil shale operations in Compressor oil.	234 164
Concrete storage tanks	23 87
Asphaltic concrete.	192 175
Oil shale operations in Compressor oil. Compression, gasoline by compression of natural gas Concrete storage tanks Asphaltic concrete. Condenser oil. Condenser oil. Condenser and condensing Effect of sulphur on condensers. Water required for condensers.	137
Water required for condensers	224

	Page 224
Surface required for condensers	224
Surface required for condensers. Cross section area of vapor lines. Heat absorbed in condensing gasoline and kerosene. Connecticut—Kerosene inspection laws of.	224
Heat absorbed in condensing gasonne and kerosene	224 156
Conradson carbon test	322-3
	022-0
Conversion factors	99
For units of measure of water and petroleum For temperature degrees on Fahrenheit and Centrigrade	00
scales	367-9
For viscosities	-7-8-9
For Baume' and specific gravity for liquids lighter than	
water, Bureau of Standards scale	372
water, Bureau of Standards scale For Baume' and specific gravity for liquids lighter than water, Tagliabue scale For Baume' and specific gravity for liquids heavier than	
water, Tagliabue scale	373-5
For Baume' and specific gravity for liquids heavier than	419-20
	415-20
For metric units For units of linear dimension. For units of square measure, surfaces and areas For units of U.S. liquid and apothecary and U.S. dry measure For units of U.S. liquid and apothecary and U.S. dry measure	429
For units of square measure surfaces and areas	430
For units of II S liquid and anotherary and II S dry measure	431
For units of capacity. British liquid and dry measure	432
For units of weight.	433
For units of work.	434
For units of pressure. •	435-6
For units of C. S. liquid and apothecary and U.S. ary measure For units of capacity, British liquid and dry measure For units of work. For units of pressure. For units of pressure. For units of temperature, heat, time, velocity and money	437
Copper chloride for gas analysis Copper oxide for removing color and odor	360
Copper oxide for removing color and odor	136
Corrosion tests	73-338
Cottonseed oil, heat of combustion of	190
Cove oil	175
Oracking Braduction of analysis and analysis	82
Production of cracked gasoline 1913-8	120
Gasoline obtainable by cracking	124
Pyrometry in cracking Effect of cracking on odor and color Color of cracked gasoline from California and Mexican oil	136
Color of cracked gasoline from California and Mexican oil	136
Amount of cracked gasoline from crude oil	138
Color of cracked gasoline from California and Mexican oil Amount of cracked gasoline from crude oil Effect of cracking on the lubricating qualities of oil and the viscosity of oil Effect of cracking on the visocity and gravity of fuel oil Fuel oil required for cracking Chemical nature of the cracking of oil. Carbon produced by cracking Yield from cracking parafilm wax	
viscosity of oil	166
Effect of cracking on the visocity and gravity of fuel oil	181
Fuel on required for cracking.	185
Chemical nature of the cracking of oil	209
Chamical formulas illustrating cracking	209 209
Vield from cracking paraffin way	209
Yield from cracking paraffin wax Polymerization and side reactions involved in cracking oil	210
Asphalt produced in cracking oil	210 210
Composition of gas obtained from Burton cracking stills	210
Olefins produced by various systems of cracking	210
Graphic illustration of the relation of specific gravity to dis-	
tilling temperature of water white distillate before crack-	
ing and after cracking and of heavy fuel oil distillate be-	
fore cracking and after the first, second and third cycles	
Polymerization and side reactions involved in cracking oil Asphalt produced in cracking oil Composition of gas obtained from Burton cracking stills Olefins produced by various systems of cracking Graphic illustration of the relation of specific gravity to dis- tilling temperature of water white distillate before crack- ing and after cracking and of heavy fuel oil distillate be- fore cracking and after the first, second and third cycles of cracking, illustrating polymerization Graphic representation of the relation of the percentage dis- tilled to specific gravity and distilling temperature of water white distillate before and after cracking Graphic representation of relation of Baume' gravity to per- centage distilled of water white distillate before and after cracking.	211
tilled to specific gravity and distilling tomporature of	
water white distillate before and after cracking	212
Graphic representation of relation of Baume' gravity to per-	
centage distilled of water white distillate before and after	
Graphic representation of relation of distilling temperature and percentage of olefins to specific gravity of light hydrocarbons produced from heavy hydrocarbons by several well known cracking processes.	213
Graphic representation of relation of distilling temperature	
and percentage of olefins to specific gravity of light	
hydrocarbons produced from heavy hydrocarbons by	
Several well known cracking processes	214
Constring of all in the more shares it is	215
Classification of oil cracking processes Cracking of oil in the vapor phase at atmospheric pressure	015
Cracking oil in the liquid phase with and without distillation	215
Cracking of oil in the liquid phase with and without distillation at and above atmospheric pressure with and without vapor space	
vapor space	215
vapor space Catalytic processes for cracking of oil Effect of certain chemicals on cracking of oil	215
Effect of certain chemicals on cracking of oil.	215
Benton apparatus for cracking oil	216

	Page
Development of commercial practice in the cracking of oil Atwood process for cracking	217 217
Young process for cracking	217
Young process for cracking. Denton process for cracking. Specifications and claims of Dewar & Redwood on the crack-	217 217
specifications and claims of Dewar & Redwood on the crack-	217
ing of oil Dewar & Redwood apparatus for cracking oil	218
Burton apparatus for cracking oil	220
Advantages of liquid phase cracking Refinery engineering data on distilling and cracking petro-	222
leum	223
Fuel consumed in cracking crude oil to produce one gallon	
of gasoline. Gas produced in cracking oil. Carbon produced in cracking	223-4
oil	223
Composition of so-called carbon residue in cracking	223
Typical run of a typical cracking process Graphic representation of effect of oil cracking temperatures	223
on strength of steel. Graphic representation of vapor cracking and slow cracking.	225
Graphic representation of vapor cracking and slow cracking.	226
Graphic representation of relation of specific gravity to dis- tilling temperature of gasoline, by cracking, of natural	
gasoline from various sources, of naphthenes, of aromatics	
and of olefins. Equilibrium cracking tests on different petroleum hydro- carbons giving viscosity before and after cracking, the	227
carbons giving viscosity before and after cracking, the	
pressure, the gas produced, the shrinkage loss, the gaso-	
line yield and the character of residue	228
Effect of varying pressures on the products of cracking kero- sene and fuel oil.	229
sene and fuel oil. Detailed properties of water white distillate before and	
after cracking including fractional gravity distillation	230
and stream gravity. Fractional gravity distillation of gasoline made by Benton	400
cracking process. Fractional gravity distillation of shale oil before and after	231
cracking.	237-8
Graphic representation of relation of gravity to nercontage	201-0
distilled of shale oil before and after cracking Graphic representation of relation of gravity to distilling	240
temperature of shale oil before and after cracking	239
Pressure cracking test for heavy petroleum hydrocarbons	319
Illustration of apparatus for determining the vapor pressure and for cracking tests of petroleum	320
	164
Crank case oil Cream separator oil	175
Cream separator oil Cretaceous—relating to oil of Cretaceous formation	4
Cube method—of determining melting point Cup grease.	297-8 175
Cushing, Okla., crude oil, fractional gravity distillation analysis of	124
Properties of. Graphic representation of.	120
Graphic representation of	121
Cycle-result of first, second and third cycle of equilibrium crack-	
ing of fuel oil distillate	211
Cylinder oil Properties of	162-4
Properties of	174-5
Cylinder stock	132
From Cabin Creek crude oil	132
Cylindrical tanks, capacity and gauging of1	.00-119
Decane Specific gravity formula melting point boiling molecular	
Specific gravity, formula, melting point, boiling, molecular weight. etc., of	130
weight, etc., of Decline and production of individual oil wells	38
Degrees Fahr. and Cent. (see temperature and conversion factors)	
Dehydration Befining of oil by and method of	101
Refining of oil by and method of	191 156
Delaware—State kerosene inspection laws of Demand for petroleum for automobiles and for fuel oil	100
For petroleum products.	2-3

	Page
Depth of oil wells	F
Effect of depth of wells on quality of oil Effect of depth on temperature.	6
Devonian-Stratigraphic section of rocks of Devonian series	10-11
Dewar Specifications and claims of patent of Dewar & Redwood on	
cracking of oil Dewar & Redwood apparatus and process for cracking oil	217 218-9
Diesel engine—specifications for fuel oil for Consumption of fuel oil by	$\begin{array}{c} 183 \\ 184 \end{array}$
Displacement method of determining specific gravity	277
Disposition of crude oil in the U. S. in 1913 Distillate	13
Uses for heavy distillate	3 131
Water white distillate. Distillate oil.	132
Pressed distillate	2-175
Distillation and distilling	110
Relation of specific gravity to distilling temperature of water	
Find oil required for distillation of crude oil	$211-3 \\ 185$
Fractional gravity distillation of various crude oil12	0-128
Fractional gravity distillation of various crude oil12 Effect of steam on distillation Typical refinery practice of fire distillation	131
Pyrometry applied to petroleum distillation Diagram of pyrometric control Effect of distillation on sulphur content of distillate Distillation curves for market gasoline Effect of vacuum and fire distillation on sulphur content and viscosity of distillate.	134
Diagram of pyrometric control	$ \begin{array}{r} 135 \\ 137 \end{array} $
Distillation curves for market gasoline	148
Effect of vacuum and fire distillation on sulphur content	1.07
Refining of oil by fractional distillation and steam distilla-	167
Cas produced by fire distillation	191 209
District	
Petroleum products by districts	$13 \\ 32$
Refinery operation by districts Prices of crude oil by districts	33
Doctor test for gasoline Driling costs for oil and gas wells in various fields	335
Drinnig costs for on and gas wents in various helds	$^{31}_{247}$
Dry gas. Ductility of asphaltic cement. Method of determining ductility	199
Duocosane.	340-2 128
Duodecane	218
Dust laying oil (see road oil) Dust East Indies, production of petroleum in	12
Dutch Shell group of oil companies	60
Dynamo oil. ,	$175 \\ 192-5$
Edwards gas balance, specific gravity by	51-3 350
Effusion method of determining specific gravity of gas	$350 \\ 12$
Earth, bituminous. Edwards gas balance, specific gravity by. Effusion method of determining specific gravity of gas. EgyptWorld's production of petroleum from 1857 to 1918 Electricity-costs of electrical heat compared with petroleum.	181
Clacking with electricity	215
Elliott flash and fire closed tests	313
Society for Testing materials	306
End point—apparatus of U. S. Bureau of Mines and American Society for Testing materials. Method of making end point distillations	1-8-9
Energy-Conversion of factors for units of	434
Conversion of Sayholt Universal viscosity readings to Engler	284
Description of Engler viscosimeter Factors for converting viscosities on the Engler, Saybolt and Bedwood machines to each other the Engler, Saybolt and	284
Redwood machines to each other	1-8-9
Engine oil-properties of various lubricants including engine oil.164	-175
Engineering—refinery engineering data on distilling and cracking	
oil. Equivalents (see Tables, Convesion factors, temperature, etc.)	223
Of units of measurement of water and petroleum at 60°F	99

	Page
Eschka-determination of sulphur by the Eschka method	331
Ether—petroleum, U. S. P Ethane—explosiveness of.	145 252
Ethane—explosiveness of. Heat of combustion and ignition temperatures of Gravity, formula, melting point, boiling point and molecular weight of athene	254
Gravity, formula, melting point, boiling point and molecular	
Weight of othere	130
Ethane in natural gas. Specific gravity, liquefaction pressure, heating value and ex-	245
plosibility of ethane	251
plosibility of ethane	129
Evaporation	
Losses in the storage of crude petroleum due to	87
Effect of temperature, of storage, gasoline content, surface exposed, roof and color on evaporation	87
Amount of oil lost by evaporation	87
Rate of evaporation of gasoline and benzol	146
Amount of oil lost by evaporation. Rate of evaporation of gasoline and benzol. Expansion of petroleum by heat-tables. Coefficient of expansion of fuel oil.	189
Explosions of natural gas and gasoline	109
Definition of explosion, limits of explosibility of mixtures of	
compustible gases and air, explosions of ethane, rates	050
of propagation of explosions Express—rules governing the shipment of oil samples by express.	252 91
Extraction effect of viscosity norosity and pressure on ex-	91
Extraction—effect of viscosity, porosity and pressure on ex- traction of oil from oil sands	4
Factors (see conversion factors)	
Faults-illustration showing conditions for accumulation of oil	7
and gas	
Coological accumumance of all in the Annalashian field Ohio	
Indiana field, Illinois, Mid-Continent, Wyoming, Gulf Coast, California, and Mexican fields Outline map of oil fields and pipe lines of Mexico, California, Wyoming, Toyag, and Louisiana, Oklahoma and Kansas	0
Coast, California, and Mexican neids	0
Wyoming, Texas and Louisiana, Oklahoma and Kansas,	
Appalachian.	16-21
Appalachian. Outline map of oil fields of the United States and the world. Total oil wells drilled in the Mid-Continent field Relative activity of oil fields in 1918 showing number of	22
Total oil wells drilled in the Mid-Continent field	29
rigs and wells drilled	29
Table showing price per foot for drilling oil and gas wells	
in various fields.	31
rigs and wells drilled Table showing price per foot for drilling oil and gas wells in various fields Filler in asphalt pavement Asphalt filler for brick, Mexican, Texaco and Sarco type Filtration of petroleum through Fuller's earth.	196 207-8
Filtration of petroleum through Fuller's earth	136
	87-8
Fire—losses of petroleum due to Causes of and conditions for	88
Static discharge as cause of	88 88
Static discharge as cause of. Methods of prevention and eradication of. Fixed carbon-method of determining fixed carbon and ash in	00
bituminous materials.	324
ADDARATUS TOP.	325
Flash and fire tests of lubricants. Illustration of various types of flash and fire test apparatus. Method of determining flash point of kerosene and lubricants Determination of flash point of lubricants by Cleveland open cup.	161
Method of determining flash point of korosono and lubricants	313 314-5
Determination of flash point of lubricants by Cleveland open	011-0
cup	316
Correction tables of flash point for barometric pressures Various types of flash and fire testers	317
	318
Flask for end point distillation of gasoline	306
Float test, determination of viscosity of petroleum residue by Floc test, determination of in petroleum products	294-6
Floor oil.	175
Floor oil. Florida—State Kerosene laws of Flour mills—natural gas consumption by Flow more formula for the flow of oil in pipe lines Flow sheet—of complete refinery Fluorescence or bloom in petroleum and method of removing Fluxing of hard asplait.	156
Flow formula for the flow of oil in size lines	250 82-258
Flow sheet—of complete refinery.	82-258
Fluorescence or bloom in petroleum and method of removing.	136
Fluxing of hard asphalt Foam for fire extinction	199
Foam for fire extinction	88
Formolit reaction	128 130
Fractional gravity distillation of shale oil before and after crack-	
ing	237-8

		Page
Fra	sch method of removing sulphur. nce—asphaltic limestone from. ezing temperatures of gasoline and benzine. ezing test—apparatus for freezing test of natural gas ler's earth—filtration of petroleum through el consumed in distilling crude oil. Consumed in cracking crude oil to produce one gallon of gasoline.	137
Fra	nce-asphaltic limestone from	197
Fre	ezing temperatures of gasoline and benzine	146
Fre	ezing test-apparatus for freezing test of natural gas	356 136
Ful	lers earth—Intration of petroleum through	223
T. ere	Consumed in cracking crude oil to produce one gallon of	220
125	gasoline	223 - 4
Fue	d oil Demand for fuel oil of the U.S. Navy Refinery operation on crude oil in 1918 with amount of oil	181
	Demand for fuel oil of the U.S. Navy	2
	Refinery operation on crude oil in 1918 with amount of oil	
14	treated and amount of fuel oil produced	$32 \\ 89 - 90$
	Specifications for fuel oil storage tanks	132
3	Definition of. Gravity and viscosity of fuel oil from various sources	101
	Effect of cracking on viscosity and gravity of fuel oil	181
	Equivalents of coal and fuel oil	182
	Gravity and viscosity of fuel oil from various sources Effect of cracking on viscosity and gravity of fuel oil Equivalents of coal and fuel oil Heat of combustion, gravity, flash point and sulphur in Mid- Continent and Mexican fuel oil	1 109
	Specifications for fuel oil of the U.S. Nevy	182
125	Specifications for fuel oil for Diesel engines	183
	Miscellaneous information concerning fuel oil	184
	Steam required for atomizing fuel oil	
	Air required for burning fuel oil	184
	Temperature of fuel ail and gas flames	184 184
	Air required for burning fuel oil. Carbon dioxide in stack gases from fuel oil furnaces Femperature of fuel oil and gas flames. Gases produced by combustion of fuel oil	184
	Fuel oil for melting iron and steel Advantages of fuel oil for metallurgical purposes	184
	Advantages of fuel oil for metallurgical purposes	184
	Consumption of fuel oil by Diesel engines Stack area for fuel oil furnaces	$ 184 \\ 184 $
	Efficiency of fuel oil	184
	Efficiency of fuel oil	184
1 - 5	Fuel oil required for distilling crude oil and in refining	185
	Fuel oil required for distilling crude oil and in refining Fuel oil required for cracking Calculation of heat of combustion of fuel oil from gravity18	185
	Calculation of heat of combustion of fuel oil from gravity18	185
1.10	Method of sampling fuel oil	180
	Heating value of fuel oil	50, 100
	of various fuel oils	188
	of various fuel oils. Relation of gravity to heating value of fuel oils in B. T. U. per pound and per gallon	101
	Effect of varying pressures on the products of cracking kero-	189
	sene and fuel oil	229
	sene and fuel oil Proximate distillation of petroleum giving amount of fuel oil,	
-		305
Gal	icia-World's production of petroleum from 1857-1918	12
Gar	ber, Okla., crude oil Fuel oil from.	181
	Graphic illustration of composition of crude oil from	121
1.18	Fractional gravity distillation analysis of	126
Gas		
	Geologic occurrence of petroleum and natural gas	4
	Association of salt water with oil and gas	4
	Conditions for occurrence of oil and gas Association of salt water with oil and gas Relation of salt water, oil and gas Diagram showing conditions for accumulation of oil and gas in anticlines, synclines and faults	5
	Diagram showing conditions for accumulation of oil and gas	
	in anticlines, synclines and faults	7
		65-79
	names of companies	130
	Comparison of the heat of combustion of petroleum. coal.	
	natural gas and coal gas	181
	Relative cost and heating value of natural gas, producer gas,	0 100
	Formula for calculating cost of producer gas	190
	Products of refining of light oil from gas works	241
	Benzol from light oil of gas works	241
	Gas manufacturing process in use in U. S	242-3
	Gas oil for gas manufacture	242-3
~ ~	Comparison of the heat of combustion of petroleum, coal, natural gas and coal gas	242-3
	Content of light oil from manufactured gases	244

		Page
	Yield of crude light oil from manufactured gases Benzol, toluol and naphtha in manufactured gases Paraffin hydrocarbons separated from manufactured gases	244 244 244
	Natural gas, occurrence of	245
	Composition and heating value of natural gas of Oklahoma	245-8
	and Kansas. Comparative properties of natural and manufactured gases Carbon dioxide, carbon monoxide, hydrogen and hydrogen	246 246
	sulphide in Dry gas, wet gas. Gasoline carried by natural gas Formula for calculating the probable yield of gasoline from	246 247 247
		248
	gas. Properties of incombustible gases in natural gas. Method and cost of extracting helium from natural gas Lifting power of helium, hydrogen, ammonia and methane in balloons.	248 248 249
	Carbon black from natural gas Theoretical and actual amount of carbon black from natural	249 249
1	gas. Commercial uses of natural gas Comparison of natural gas with coal and oil Natural gas consumed for gas engines. brick plants. etc	250 250 250
	Natural gas consumed for gas engines, brick plants, etc Composition of gas from Burton cracking stills Detailed properties of hydrocarbons found in natural and casinghead gas.	210
	Natural gas, coal gas, water gas and illuminating gas ex- plosions. Limits of explosibility of mixtures of combustible gases and	252
	Limits of explosibility of mixtures of combustible gases and air.	252
	Testing the capacity of casinghead gas wells with orifice meter.	253
	Production of natural gas in United States Pitot tube for testing open flow of gas wells Tables for determining flow of gas wells by means of the Pitot tube	253 255-6 257
1 14	Formula for calculating the flow of gas in pipes Method of measuring the flow of natural gas by the orifice meter	258 263
	Tables for calculating the capacity of orifices for testing the flow of natural gas from gas wells	266-7-8
	Tables for calculating orifice capacity for natural gas, water gas and coal gas	
	Method of determining specific gravity of gas by effusion	350
	Absorption and freezing method for testing natural and cas- inghead gas for gasoline content	354-5-6
	apparatus	359
	data Reagents for gas analysis Method of determining heating value of gas by the gas calorimeter.	360
	calorimeter. Approximate heating value of natural gas by calculation from oxygen consumed on combustion	362
	Heating value of gas by calculation from chemical analysis Tables for reduction of volume of gas to 60°F and 30 inches of mercury	362
Gas	balance-illustration of Edwards gas balance for specific gravity.	
Gas	black (see carbon black)	001-4-0
Gas	Compressors	
Gas	Natural gas consumed for gas compressors	250
Gas	Natural gas consumed by gas engines Engine Oil	250
5 5	Properties of	164

		Page
Gas		2
	Uses of Definition of, properties of, and specifications for Heat of combustion, gravity, flash point and sulphur in Mid- Continent and Kansas gas oil Gas oil for gas manufacture	154
	Heat of combustion, gravity, flash point and sulphur in Mid-	
	Continent and Kansas gas oil	$182 \\ 242 - 3$
		444-3
Gas	oline Increase in gasoline from crude oil	2
	Consumption by automobiles	2 2 3
	Uses for gasoline Production of natural gas gasoline in 1917 by states and from	3
	Production of natural gas gasoline in 1917 by states and from	92
	absorption and compression Total gasoline from natural gas marketed in 1917	23 23
	Refinery operations on crude oil in 1918, with amount of oil treated and amount of gasoline produced. Production of cracked gasoline from 1913-8. Kind of gasoline produced in 1917. Production of casinghead gasoline plants by states. Rules governing the location of loading racks and gasoline storage	24
	treated and amount of gasoline produced	32
	Production of cracked gasoline from 1913-8	32 32
	Kind of gasoline produced in 1917	32
	Production of casinghead gasoline plants by states	61-4
	storage	96
	storage Content of crude oil giving gravity, percentage of gasoline and kerosene and total obtainable gasoline by cracking Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in commercial gasoline	
	and kerosene and total obtainable gasoline by cracking	120
	Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in	128
	Detailed properties of hydrocarbons constituting gasoline	130
	Definition of Effect of various types of crude oil on gravity of gasoline Color of cracked gasoline from California and Mexican oils	131 - 8
1	Effect of various types of crude oil on gravity of gasoline	$ 131 \\ 136 $
	Properties of hydrocarbons constituting gasoline	130
	Color of cracked gasoline from California and Mexican oils. Properties of hydrocarbons constituting gasoline. Aromatic compounds in gasoline. Amount of natural gasoline, casinghead gasoline, and cracked gasoline from crude oil. Desirable properties for gasoline. U. S. specifications for avlation gasoline. Soecifications for avlation gasoline. Corrosion and gumming tests of gasoline. Export grade, fighting grade and domestic grade of gasoline. Summary of gasoline inspection laws. Navy specifications for gasoline, aero gasoline, export and domestic	138
	Amount of natural gasoline, casinghead gasoline, and cracked	100
	gasoline from crude oil	138
	U. S. specifications for gasoline	139
	Specifications for aviation gasoline	139
1	Corrosion and gumming tests of gasoline	140
	Specifications for ILS motor gasoline	141
	Summary of gasoline inspection laws	143-4
	Navy specifications for gasoline, aero gasoline, export and	140
		$146 \\ 146$
	Freezing temperature of gasoline and benzol. Boiling temperature of gasoline and benzol. Rate of evaporation of gasoline and benzol. Losses and possible savings in gasoline.	146
	Freezing temperature of gasoline and benzol	146
	Bolling temperature of gasoline and benzol	146 146
	Losses and possible savings in gasoline	147
	Suggestions for saving gasoline Distillation curves for market gasoline	147
	Distillation curves for market gasoline	148 190
	Heating value of gasoline Relations of heating area to amount of gasoline distilled	223
	Fuel consumed in cracking crude oil to produce one gallon of	
	gasoline	223
	Heat absorbed in condensing gasoline	224
	kerosene etc undergoing cracking	226
	kerosene, etc., undergoing cracking Graphic representation of the relation of specific gravity to	
	distilling temperature of gasoline by cracking, of natural gasoline from various sources	0.07
	gasonne from various sources	227
	bons giving gasoline vield, etc	228
	bons giving gasoline yield, etc Fractional gravity distillation of gasoline made by the Ben-	
	ton cracking process. Gasoline from oil shale. Gasoline carried by natural gas and casinghead gas.	$\begin{array}{c} 231 \\ 233 \end{array}$
	Naptha or gasoline from oil shale	235
	Gasoline carried by natural gas and casinghead gas	247
		247
	Absorption process for casinghead gasoline	$\begin{array}{r} 247 \\ 248 \end{array}$
	Cost of casinghead gasoline plants Absorption process for casinghead gasoline Yield of gasoline from casinghead gas by compression com- pared with absorption and specific gravity tests.	
	pared with absorption and specific gravity tests	248

		Page
	a for calculating the probable yield of gasoline from	
0		248
Gasoline	e and natural gas explosions	252 252
Method	bility of gasoline vapors	90-291
Proxima	te distillation of petroleum, giving amount and grav-	BOF
ity	of us of the U.S. Bureau of Mines and American Society	305
for	Testing Materials end point distillation of	306
Method	of making end point distillation of gasoline	307-8-9
Method	of making end point distillation of gasoline	321
Absorpt	ion method for testing material and casinghead gas	354
Freezing	gasoline content g method for testing natural gas for gasoline content.	355
	oles for tank cars and horizontal cylinder tanks1	100-119
Gear case o	oil	175
	ints, specifications for	172
General elec	tric method	
For mel	lting point or softening points of bituminous ma-	297
	tion of apparatus for melting point by	299
Caslann		3
Geologic	occurrence of petroleum and natural gas	4
Relation	of petroleum concurrence to volcanic action	4 4
Conditio	ns for occurrence of oil and gas	4
Associat	c occurrence of petroleum and natural gas of petroleum concurrence to volcanic action of carboniferous period to oil ns for occurrence of oil and gas ion of salt water with oil and gas er of oil from tertiary, cretaceous and carboniferous lods	4
Characte	of oil from tertiary, cretaceous and carboniferous	4
Physical	condition of oil bearing rocks	
Voids in	1 oil sand	4
Relation	n oil sand f water on migration of oil of oil, salt water and gas	4 4 5 4 4 4
Domes a	and anticlines	4
Amount	of oil in one cubic foot of oil sand	4
Effect o	f viscosity, porosity and pressure on extraction of oil	4
from	a oil sand	4
Significa	nce of asphalt outcrops	5
Depth of	f oil wells, effect of depth of wells on quality of oil.	5
Deepest	of oil discharged from oil sand f viscosity, porosity and pressure on extraction of oil a oil sand of asphalt outcrops of anticlines to surface topography f oil wells, effect of depth of wells on quality of oil. well in the world f wells at Ranger, Texas s as to origin of petroleum, demonstration of the in	4.0.0.000
Depth of Theories	f wells at Ranger, Texas demonstration of the	• 5
orig	in	5
Tempera	ture in wells	5666
Tempera	ture of oil in Texas wells West Virginia wells	6
Summar	ized table of oil occurrences	6
Geologic	occurrence of oil in Appalachian, Ohio-Indiana field,	
and	the second secon	6
Typical	composition of "Mississinni lime" from Kansas	6
Diagram	a showing conditions for accumulation of oil and gas nticlines, synclines and faults. aphic section of rocks in oil bearing regions of Kan- showing Permian and Pennsylvanian series.	7
Stratigra	aphic section of rocks in oil bearing regions of Kan-	
sas,	showing Permian and Pennsylvanian series	9
Stratigra	and rection of rocks in oll bearing regions of North- Oklahoma, showing Permian and Pennsylvanian	
serie	es, Mississippian, Devonian, Ordovician and Cambrian	
		10-11
gas		465
	tate Geologists and their addresses	466
Georgia		12.3.4
Gasoline	f essential points in State Kerosene Laws	156
Gasonne	**************************************	143
Producti	on of petroleum from 1857-1918 in	12
lisonite		
Composi	tion of	193

e - 17	Page
Grahamite	11.2
Composition of Amount of carbon, hydrogen, oxygen, nitrogen and sulphur i	n 193 n 128
Amount of carbon, hydrogen, oxygen, nitrogen and sulphur i	n 128
Wram,	. 428
Gravity	100
Content of crude oil giving gravity, percentage gasoline, etc	. 120
Detailed properties of hydrocarbons, giving gravity, etc Stream gravity for benzine Effect of various types of crude oil on gravity of gasoline. Gravity of kerosene from various crude oils Sneeffe gravity of hubricents	· 130 · 131
Effect of verious types of avude oil on gravity of gravita	: 131
Gravity of kerosene from various crude oils	. 131
Gravity of kerosene from Wyoming	: 132
Gravity of Mexican fuel oil	. 181
Gravity of Mexican fuel oil Viscosity and gravity of fuel oil Effect of cracking on the viscosity and gravity of fuel of	. 181
Effect of cracking on the viscosity and gravity of fuel of	11
and heavy crude oils	. 181
Heat of combustion, gravity, etc., of Mid-Continent and Mex	-
ican fuel and gas on	• TOT
Calculation of heat of combustion of fuel oil from gravity.	. 185
Gravity and flash point of various crude oils in Mexico, Call fornia, Ohio, Illinois, Indiana and Texas Graphic representation of relation of gravity to heating valu of fue oils in B. T. U. per pound and per gallon Graphic illustration of relations of specific gravity, Baume gravity, distilling temperature and percent distilled of water white distillets before ond after creating.	. 187
Grantic representation of relation of gravity to heating valu	A 101
of fuel oils in B. T. U. per pound and per gallon	. 189
Graphic illustration of relations of specific gravity. Baume	e'
gravity, distilling temperature and percent distilled o	f
water white distillate before and after cracking2	211-2-3-4
water white distillate before and after cracking2 Graphic representation of the relation of specific gravity t distilling temperature of gasoline made by cracking Detailed properties of water white distillate before and after anothing including tractional ensuity distillate before and after	0
distilling temperature of gasoline made by cracking	. 227
Detailed properties of water white distillate before and afte	r
cracking, including fractional gravity distillation tem	. 230
perature and stream gravity Fractional gravity distillation of gasoline made by Bento	. 430
cracking process.	. 231
Fractional gravity distillation of coal tar benzol	. 232
Fractional gravity distillation of coal tar benzol Fractional gravity distillation of shale oil before and afte	r
cracking	. 237-8
Detailed properties of hydrocarbons found in natural and cas	-
inghead gas giving specific gravity, etc	. 251
Specific gravity and Baume' gravity with the hydrometer.	
Formula for converting specific gravity to Baume' gravity fo	
liquids lighter than water and heavier than water on th Eureau of Standards and Tagliabue scales	. 272
Method of determining the specific gravity with the picno	-
meter, the Westphal balance	. 273
Method of determining the specific gravity of asphaltic ce	-
ment	. 274
Method of determining the specific gravity of asphaltic sur face mixtures and solid asphaltic material	274
Illustration of methods of determining the specific gravity o	e f
agning the compart by displocoment	977
Method of determining the specific gravity of gas by effusion Method of determining the specific gravity by Edwards ga	n 350
Method of determining the specific gravity by Edwards ga	S
balance U S. Bureau of Standards tables for equivalents of Baume	; 351-2
Stantarus tables for equivalents of Baume	270 1 2
gravity, specific gravity and pounds per gallon Tagliabue tables for equivalents of Baume' gravity, specific	
gravity and pounds per gallon	.373-4-5
Baume' gravity correction tables for temperature to 60°F.	.376-383
Specific gravity correction tables for reduction of specific	С.
gravity readings to 60°F	.384-418
specific gravity tables giving equivalents of Baume' and spe	410 400
Specific gravity at bor for inquitis neavier than water	419-422
 Fightable tables for equivalents of Baume' gravity, specific gravity and pounds per gallon Faume' gravity correction tables for temperature to 60°F. Specific gravity readings to 60°F Specific gravity tables giving equivalents of Baume' and spe clifc gravity at 60°F for liquids heavier than water Specific gravity tables for strength of sulphuric acid Specific gravity Baume' and Twaddell tables for strength o sodium hydroxide solutions 	. 440-4-0 f
sodium hydroxide solutions	426
sodium hydroxide solutions Gravity, strength and freezing temperature of calcium chlo- ide and brine solutions	
ride and brine solutions	427
Grense	
Cup grease	. 175
Gun oil specifications	. 172
Gushers	
Oil gushers of Russia, Mexico, Louisiana, Texas, California	
Roumania and Kansas	. 30

KANSAS CITY TESTING LABORATORY

en l'i	Page
Hammer oll	175
Harness oll	175
Healdton, Okla, crude oil, properties of	120
Hammer oil Harness oil Healdton. Okla., crude oil, properties of. Graphic illustration of composition of	121
Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in	128
	100
Heat	407
Conversion factors for units of	437
Heat of combustion	1 0 000
Of fuel oil and gas oil and other petroleum products18	
Of various substances Of grain alcohol, wood alcohol, hard asphalt, benzol, coke, acetylene, coal gas, methane, water gas, hydrogen, iron, anthracite coal, bituminous coal, lignite, cannel coal, pe- troleum coal natural gas neat contonseed oil gasoline	190
of grain alconol, wood alconol, nard aspnalt, benzol, coke,	
acetylene, coal gas, methane, water gas, hydrogen, iron,	
anthracite coal, bituminous coal, lignite, cannel coal, pe-	
troleum, coal, natural gas, peat, cottonseed oil, gasoline,	
troleum, coal, natural gas, peat, cottonseed oil, gasoline, fuel oil, shale oil, paraffin wax, wood, sulphur, naptha- lene and Gilsonite	
lene and Gilsonite	181-190
Determination of heat of compustion of petroleum products.	
Determination of heat of combustion of gas Calculation of heat of combustion of fuel oil from gravity	362-6
Calculation of heat of combustion of fuel oil from gravity	189-185
Composition and heat of combustion of natural gas found in	
Composition and heat of combustion of natural gas found in various sections of Oklahoma and Kansas Method of determining heating value of gas by the gas	246
Method of determining heating value of gas by the gas	
calorimeter Approximate heating value of natural gas by calculation	361-2
Approximate heating value of natural gas by calculation	-
from oxygen consumed by compustion	362
Heating value of gas by calculation from chemical analysis	362
Heat absorbed in condensing gasoline and kerosene in distilling	
crude oil and in coking oil	224
Heating area of stills	223
Heat exchanges—calculation of, in refinery condensers	224
Heat of fusion of solid petroleum hydrocarbons	185
Heat of vaporization of petroleum	185
Heavy distillate, uses for	3
Helium	
In natural gas	245-248
Properties of	248
Method and cost of extracting from natural gas	248-9
Lifting power of, in balloons	249
Heptane	
	130
Properties of	251
	401
Hexane	
Specific gravity, liquefaction pressure, heating value, etc., of. Heat of combustion and ignition temperatures of	130
Heat of combustion and ignition temperatures of	254
Horizontal cylindrical tanks	
	100
Formula for calculating content of Tables for capacity of, showing relation of diameter in per-	
centage, to content in percentage of capacity	101
Contents of, at different shell depths	105
Contents of, at different shell depths. Contents of, with various depths of liquids and with diameter	
of from 36 to 120 inches	105-119
Hydrocarbons (see special hydrocarbons, paraffin, aromatics,	
napthenes, olefins, etc.) .	
Hydrogen	
Explosibility of	252
Explosibility of	
of	254
of Lifting power of Determination of, in petroleum products. Hydrogen chloride for removing color and odor in petroleum Hydrogeter-(see gravity)	249
Determination of, in petroleum products	333-4
Hydrogen chloride for removing color and odor in petroleum	136
my arometer (see gravity)	070
Specific gravity and Baume' gravity with the hydrometer	272
Method of reading the hydrometer Illustration of method of reading the hydrometer Illustration of various types of picnometers, hydrometers and Weathed belence	272
illustration of method of reading the hydrometer	275
illustration of various types of picnometers, hydrometers and	
	275
Ice machine oil Ice plants, natural gas consumption	164
Ice plants, natural gas consumption	250
Idaho	
State kerosene inspection laws State gasoline inspection laws	156
State gasoline inspection laws	143-4

	Page
Ignition temperatures	252
Of natural gas Of hydrogen, carbon monoxide, methane, ethane, propane,	
butane, pentane, etc	254
Gravity and flash point of crude oil from	187
Gravity and flash point of crude oil from Graphic illustration of composition of crude oil from	$121 \\ 143-4$
State kerosene inspection laws of	156
Geologic occurrence of oil in the Illinois field	252
Illuminating gas, explosibility of	3-149
State gasoline inspection laws of State kerosene inspection laws of Geologic occurrence of oil in the Illinois field Illuminating gas, explosibility of India—production of petroleum in.	12
Indiana Gravity and plant point of crude oil from	187
State gasoline inspection laws. State kerosene inspection laws.	156
State kerosene inspection laws	143-4
Insulated storage tanks	87
	282-3
Iowa State gasoline inspection laws	156
State kerosene inspection laws	143-4
Iron Heating value of	190
Heating value of Fuel oil for melting iron and steel	184
Italy—Production of petroleum in Japan—Production of petroleum in Journals—List of petroleum trade journals and technical publica-	$12 \\ 12$
Journals—List of petroleum trade journals and technical publica-	14
tions on petroleum and gas	464
Kansas Typical composition of "Mississippi lime" from	6
Stratigraphic section of rocks in oil bearing regions of,.	8-9
Stratigraphic section of rocks in oil bearing regions of Outline map of oil fields and pipe lines of Okla. and Kansas Daily production of crude oil by individual pools in	20 27
Oil gushers of	30
Large producers of crude oil in	40
Properties of crude oil from Allen County Kansas	41 120
Graphic illustration of composition of Kansas crude oil	121
Oil gushers of. Large producers of crude oil in. Important oil companies operating in. Properties of crude oil from Allen County, Kansas. Graphic illustration of composition of Kansas crude oil Fractional gravity distillation analysis of Allen County, Kansas, oil Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in Kansas air blown residuum	127
Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in	
Kansas air blown residuum Summary of gasoline inspection laws of Kansas	$128 \\ 143-4$
State kerosene laws of. Composition and heating value of natural gas found in Okla-	156
Composition and heating value of natural gas found in Okla-	246
home and Kansas	157
Kentucky	101
Properties of crude oil from Fractional gravity distillation analysis of crude oil from	120
Daily production of crude oil by individual pools in	$127 \\ 25$
Daily production of crude oil by individual pools in State kerosene inspection laws of	156
Asphaltic sandstone from	194
Kerosene Uses for	3
Uses for Refinery operations on crude oil in 1918 with amount of gaso- line, kerosene, etc., produced	32
Hydrocarbons constituting kerosene	130
Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in	
kerosene	$128 \\ 31 - 149$
Definition of	131
Gravity of kerosene from Wyoming, Oklahoma and Pine	132
Island crude oils Kerosene content of crude oil. Kerosene, coal oil and illuminating oil	120
Kerosene, coal oil and illuminating oil	149 149
Sulphur in kerosene. Requirements for good kerosene.	149
U. S. Specifications for water while kerosene and U. S. Navy	10 150

	Page
Burning test for kerosene	151 156
Burning test for kerosene Kerosene regulations in effect in 1919. State kerosene regulations. Effect of varying pressures on the products of cracking kero-	156
Effect of varying pressures on the products of cracking kero-	229
sene Illustration of Ubbelohde viscosimeter for kerosene and gaso-	449
line	290
Method of determining viscosity of kerosene Determination of kerosene in crude oil	291 305
Method of determining flash point of kerosene.	314
Graphic representation of vapor pressure of kerosene and heavy oil undergoing rapid cracking and slow cracking.	226
Knitting machine all	175
Knitting machine oil Leather oil	175
Liberty aero oil, U. S. specifications for Lifting power of gases in balloons	168 249
Lighthouse oil, specifications for	151
Limestone, asphaltic ···	194
Composition of From Sicily, France, Missouri and Oklahoma	194
Linear dimensions, tables of equivalents of units of	429
Liquid petrolatum, specifications for Liquid phase	177
Cracking oil in the liquid phase	215
Advantages of liquid phase cracking	222
Lording racks, rules governing	96-8
Locomotives-advantages of fuel oil installations in	184
Liter	40-949
seepage	87-88
Louisiana	10
Outline map of oil fields and pipe lines of Daily production of crude oil by individual pools in	19
Oil gushers of State kerosene inspection laws	25 30
	156
Lovibond tintometer	
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281 32
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281 32 160 160
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281 32 160 160 161
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281 32 160 160 161 161 161
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants	79-281 32 160 160 161 161 161 161
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants Refinery operations on crude oil, with amount of kerosene produced Lubricating oil, sources and refining of Emonomy and theory of lubrication. Selection of proper lubricants. Meaning of physical test for lubricants. Acid in lubricating oil Effect of temperature on viscosity of lubricating oil Properties of various lubricants. Effect of high temperature motors on the lubricating proper-	79-281 32 160 160 161 161 161
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants Refinery operations on crude oil, with amount of kerosene produced Lubricating oil, sources and refining of Emonomy and theory of lubrication. Selection of proper lubricants. Meaning of physical test for lubricants. Acid in lubricating oil Effect of temperature on viscosity of lubricating oil Properties of various lubricants. Effect of high temperature motors on the lubricating proper-	79-281 32 160 160 161 161 161 163 164 165
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants Refinery operations on crude oil, with amount of kerosene produced Lubricating oil, sources and refining of Emonomy and theory of lubrication. Selection of proper lubricants. Meaning of physical test for lubricants. Acid In lubricating oil. Froperties of various lubricants. Effect of temperature motors on the lubricating proper- ties of oil. Effect of oil.	79-281 32 160 161 161 161 163 164 165
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants Refinery operations on crude oil, with amount of kerosene produced Lubricating oil, sources and refining of Emonomy and theory of lubrication. Selection of proper lubricants. Meaning of physical test for lubricants. Acid In lubricating oil. Froperties of various lubricants. Effect of temperature motors on the lubricating proper- ties of oil. Effect of oil.	79-281 32 160 161 161 161 163 164 165
Lovibond tintometer Method of determining color by and illustration of2 Lubricating oils and lubricants Refinery operations on crude oil, with amount of kerosene produced Lubricating oil, sources and refining of Emonomy and theory of lubrication. Selection of proper lubricants Meaning of physical test for lubricants Acid in lubricating oil Effect of temperature on viscosity of lubricating oil Froperties of various lubricants. Effect of racking on the lubricating qualities of oil Effect of cracking on the lubricating qualities of oil U. S. specifications for lubricating oil Emulsion tests for	79-281 32 160 160 161 161 161 163 164 165 166 68-326
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 161 161 161 163 164 165
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 161 161 163 164 165 168 68-326 68-326 170 172
Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 163 164 165 166 166 166 166 166 166 166 172 172 173
Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 164 168 68-326 170 172 173 175
Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 163 164 165 166 166 166 166 166 166 166 172 172 173
Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 164 168 68-326 170 172 173 175
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 164 165 168 68-326 170 172 173 175 315-6 322
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 163 164 165 166 68-326 170 172 173 175 315-6 322 -7-8-9 338
Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 161 165 168 68-326 68-326 170 172 173 315-6 322 -7-8-99 -8-38 128
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 163 164 165 166 68-326 170 172 173 175 315-6 322 5-7-8-9 338
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 161 163 164 165 166 68-326 170 172 173 175 315-6 322 -7-8-9 338 128 169
 Lovibond tintometer Method of determining color by and illustration of	79-281 32 160 160 161 161 161 161 163 164 165 166 68-326 170 172 173 175 315-6 322 -7-8-9 338 128 169

	20.000
	Page
MacMichael viscosimeter Illustration of and determination of viscosity on	292-3
Conversion of Saybolt Universal viscosity readings to	284
Magruder viscosimeter	
Conversion of Saybolt Universal viscosity readings to	284
Maine-State kerosene inspection laws	156
Maine-State kerosene inspection laws Manganese Dioxide in removing color and odor in oil	136
Mans-Outline mans of oil fields of U.S. and the world	16-22
Of principal oil shale areas of U. S Marketed—Petroleum in U. S. from 1859 to 1915 by states Marketers—Refiners and producers of the Standard Oil group	236
Marketed-Petroleum in U. S. from 1859 to 1915 by states	14-15
Marketers-Refiners and producers of the Standard Oil group	59
Maryland-State kerosene inspection laws for	156
Maryland—State kerosene inspection laws for Massachusetts—State kerosene inspection laws for Measure—(see conversion factors)	156
U. S. liquid and apothecary and U. S. dry measure tables	430
British liquid and dry, tables of equivalents of units of	432
	102
Measurement-Equivalents of units of measurement of water and	99
petroleum	39
Melting point	
Detailed properties of hydrocarbons giving melting point	100
gravity, etc Determination of melting point of bituminous material by ring	130
Determination of melting point of bituminous material by ring	
and ball, cube and general electric method and apparatus	7-8-9
for Comparison of melting point determination by various methods	1-0-0
methods	300
methods Method of determining melting point of paraffin wax and ap-	
paratus for	1-2-3
Metallurgical purposes-advantages of fuel oil for	184
Meter	428
Methane	Sec. 1
Specific gravity, liquefaction pressure, heating value and ex-	
plosive mixture of	251
Explosions of	252
Specific neat of, heating of combustion and ignition tempera-	25.4
In potymol cos	245
Lifting nower of in balloons	249
Explosions of Specific heat of, heating of combustion and ignition tempera-, tures of. In natural gas. Lifting power of in balloons. Heating value of.	190
Methods of analysis—(see analysis)	100
- Index to application of	271
Index to application of Outlines of methods of analysis of petroleum products 2	69-70
Metric System—Fundamental equivalents of	428
Mexico	
Geologic occurrence of oil in Properties of crude oil from Production of crude oil in Mexico from 1901-1919 Production of petroleum from 1857 to 1918 in Outline map of oil fields and pipe lines of Daily production of crude oil by individual pools in Oil wells in Mexico by companies. Oil unshers in	6
Properties of crude oil from	-120
Production of crude oil in Mexico from 1901-1919	40 12
Outline man of all falled and pine lines of	16
Daily production of crude oil by individual pools in	26
Oil wells in Mexico by companies	0.0
Oil gushers in	30
Oil weils in Mexico by companies. Oil gushers in	7-8-9
Graphic illustration of composition of Mexican crude oil in	121
Color of cracked gasoline from Mexican oil	136
Gravity of fuel oil from Mexico	181
Heat of combustion, gravity, flash point and sulphur in Mex-	100
Crevity and floor point of versions and a sile in Maria	182
Composition of Mexican asphalt	193
Asphalt filler for brick, Mexican type	207
Michigan	201
State kerosene inspection laws for	156
State gasoline inspection laws for	143-4
Mid-Continent	
Geologic occurrence of oil in Mid-Continent field	6
Map of oil pools and pipe lines of	20
Fuel oll	181
Total oil wells drilled in Mid-Continent field	28
Heat of combustion, gravity, flash point and sulphur in fuel	182
and gas oil Price changes in crude oil in Mid-Continent field from 1905-	104
1 1010 The changes in crude on in Mid-Continent heid from 1905-	

1	Page
Migration-Effect of water on migration of oil	4-
Mineral Value of petroleum as a mineral product Marketed mineral products in U. S. 1918 compared with pe-	2
troleum	- 32
Mineral oil or liquid petrolatum of the U.S.P. specifications of	199 177
Specifications for Cloud test and burning test for	152-175
Cloud test and burning test for	152-142
Minnesota	
- State kerosene inspection laws for	156
Mississinni-State kerosene inspection laws for	143-4
State gasoline inspection laws for Mississippi—State kerosene inspection laws for Mississippi lime—Typical composition of, from Kansas	6
Mississippian series-Stratigraphic section of rocks in Northern	10-11
Okla. showing	10-11
Missouri State gasoline inspection laws for	143-4
State kerosene inspection laws for	156
Asphaltic limestone and asphaltic sandstone from	194 437
	491
Montana State gasoline inspection laws for	143-4
State gasoline inspection laws for. State kerosene inspection laws for. Molecular weight-Detailed properties of hydrocarbons giving Motor fuel-Comparison of gasoline and benzol as.	156
Molecular weight-Detailed properties of hydrocarbons giving	130
Motor fuel-Comparison of gasoline and benzol as	146 169
Motor oil—Specifications for gasoline motor oil Motors—Effect of high temperature motors on lubricating prop- erties of oil from Pennsylvania and Texas	
erties of oil from Pennsylvania and Texas	165
Naphtha (see gasoline, benzine, mineral spirits)	3
Uses for Definition of In manufactured gases. Method of making end point distillation of	121
In manufactured gases	244
Napthalene, heating value of	190
Nanthene hydrocarbons	
Chemical properties of. Graphic representation of relation to specific gravity to dis-	128-9
tilling temperature of	227
Natural gas	245
Geologic occurrence of petroleum and	130
Hydrocarbons constituting	186
Occurrence of	245
Helium, methane, ethane, propane, butane and nitrogen in Composition of as delivered to various cities	245-8
Composition and heating value of as found in various sections	
of Kansas and Oklahoma Comparative properties of natural and manufactured gases	246
Comparative properties of natural and manufactured gases	246 247
Gasoline carried by natural and casinghead gas Amount of gasoline obtained from Method of extracting helium from and cost of extraction	- 247
Method of extracting helium from and cost of extraction	249
Theoretical and actual amount of carbon black from	249 250
Commercial uses of Comparison of, with coal and oil	250
Freezing method for testing natural gas for gasoline content.	355
Illustration of apparatus for freezing test of natural gas Approximate heating value of natural gas by calculation from	356
oxygen consumed on combustion,	362
Nebraska Gasoline inspection laws for	143-4
State kerosene inspection laws for	
Nentral oil	175
Nonviscous and viscous neutral oils	175
Nonviscous and viscous neutral oils. Nevada—State kerosene inspection laws for. New Hampshire—State kerosene inspection laws for	156
New Jersey-State kerosene inspection laws for Newkirk, Okla-Properties of crude oil from	156
New Mexico	120
State kerosene inspection laws for	156
State gasoline inspection laws for	143-4

	Page
New York	156
State kerosene inspection laws for. Nitrogen in petroleum. In natural gas. Properties of Specific heat of. Determination of, in petroleum products. Nonsne properties of	128
Nonane, properties of North America—petroleum refineries of, by states North Carolina	42-52
State kerosene inspection laws for State gasoline inspection laws for North Dakota—State gasoline inspection laws for	156 143-4 143-4
Occurrence Summarized table of oil occurrence	6 4 4 6 130
	250
Odor Cause of odor and color in refined petroleum Relation of sulphur to color and Effect of cracking on Shrinkage in removing color and Chemicals for removing color and	135 136 136 136 136
Ohio State kerosene inspection laws for State gasoline inspection laws for Gravity and flash point of various crude oils in Geologic occurrence of oil in the Ohio-Indiana field Oildag	156 143-4 187 6 175
Oil fields Outline map of and pipe lines of Mexico, California, Wy- oming, Texas and Louisiana, Oklahoma and Kansas, and the Appalachian field Relative activity of, in 1918, showing number of rigs and wells drilled	16-31 29
Oklahoma Important oil companies operating in. Dally production of crude oil by individual pools in Stratigraphic section of rocks in Northern Oklahoma. Outline map of oil field and pipe lines of. Properties of crude oil from Fractional gravity distillation analysis of crude oil from12 Amount of carbon, hydrogen, oxygen, nitrogen and sulphur in crude oil from	$ \begin{array}{r} 41\\ 26\\ 10-11\\ 20\\ 120\\ 3-4-5-6\\ 128 \end{array} $
crude oil from. Composition and heating value of natural gas found in various sections of Gravity of kerosene from crude oil from. State kerosene inspection laws for. State gasoline inspection laws for. Asphaltic limestone and asphaltic sandstone from. Chemical properties of.	246 132 156 143-4 194 129
Olefins Chemical properties of List of individual olefins Olefins produced by various systems of cracking Graphic representation of relation of distilling temperature and percent olefins to specific gravity of light hydrocar- bons produced from heavy hydrocarbons by several well known processes	129 129 210
Specific heat of. Determination of Graphic representation of relation of specific gravity to dia	254 335
centrifugal method for determining olefins in oil Illustration of apparatus for determination of olefins in pe-	227 336
ordovician-Stratigraphic section of rocks in Northern Okla-	337
homa showing Ordovician series	10-11

	Page
Oregon	
State gasoline inspection; laws of	143-4
State kerosene inspection laws of Orifice meter for testing the capacity of casinghead gas wells	156 253
	200
Origin Theories as to origin of netroleum	- 5
Theories as to origin of petroleum Demonstration of the origin of petroleum Ostwald viscosimeter, conversion of Saybolt Universal viscosity	5
Ostwald viscosimeter, conversion of Saybolt Universal viscosity	
readings to Oxidation—Method of determining the resistance of asphaltic ce-	284
Oxidation-Method of determining the resistance of asphaltic ce-	244
ment toOxygen	344
Amount of in petroleum	128
Specific heat of	254
Ozokerite—paraffin wax from	180
Paint thinner (see naphtha and mineral spirits)	9.00
Pailadium Chloride Paraffin (see wax)	360
Chemical properties of paraffin hydrocarbons1	28-132
Paraffin base oil	29-181
Paraffin base oil1 Detailed properties of individual paraffin hydrocarbons	130
Paraffin oll	175
Paraffin wax in Pennsylvania petroleum Method of valuating paraffin wax	180 180
Parafin wax solubility.	180
Paraffin from shale oil	180
Heat of combustion of	190
Heat of fusion of	185
Yield from cracking of paraffin wax	209
Yield from cracking of paraffin wax Method of determining paraffin wax or scale	24-341
Panuco crude oil Parr calorimeter, sulphur with	120-1 331
Patents, U S., on petroleum refining4	
Pavements (see asphalt)	
Paving (see asphalt)	
Peat, heating value of	190
Penetration of rope lubricants	173
Effect of mineral matter on the penetration of asphaltic ce- ment	199
ment	339
Penetrometers	340
Loss of penetration on heating	343
Pennsylvania	
Prices of crude oil in Pennsylvania from 1859 to 1919	34
Lubricating oil from Pennsylvania petroleum	165
State kerosene inspection laws of	128
Lubricating oil from Pennsylvania petroleum. Ultimate analysis of petroleum from. State kerosene inspection laws of	180
Pennsylvania Railroad specifications for petroleum products	158-9
Viscosity pipet of	284
Pennsylvanian series of rocks	9-11
Pentane, properties of	130
Pentadecane	130
Pentacosane	130 129
Pentylene Permian series of rock	9-11
Peru, production of petroleum in. Petrolatum of the U.S.P. Petrolene, determination of.	12
Petrolatum of the U.S.P	177-8
Petrolene, determination of the U S P	342
Petroleum ether of the U. S. P. Petroleum (see special products or subject)	145
Picnometer, determination of specific gravity with	273
Pine Island Crude oil from Pine Island, Louisiana. Fractional gravity distillation of crude oil from.	
Crude oil from Pine Island, Louisiana	120
I work a granty and that of the day of the month of the	. 125
Pipe lines and pipes	
Map of pipe lines in Mexico, California, Wyoming, Texas, Louisiana, Oklahoma, Kansas and the Appalachian field.	
Formula for flow of oil in pipe lines Principal pipe lines of U. S	22 82

BULLETIN NUMBER FIFTEEN OF

6

1327	Pa	ge
Formula fo	or flow of gas in pipes	258
Canacity of	f pipe lines	262
Ditch Tiltimot	a producing of Duranita nitab	128
Fitch-Olumate	e analysis of Byernite pitch	140
Pitot tube, met.	hod for determining the open now of gas wells 2	55-7
Polymonigation	and side reactions involved in the creeking of oil 210)-11
Polymethylenes	sub traction of oil pools of U. S. and Mexico	129
Doole doily pre	dustion of oil pools of U.S. and Meyico.	5-27
Pools, daily pro	Subction of on pools of C. S. and Mexico 22	
Porosity of oil	i sand	4
Potassium bich	romate, method for color of refined petroleum	280
Dotocolum bydy	avide for gas analysis	360
Fotassium nyui	Toxine for gas analysis	
Potassium pyrc	ogailate for gas analysis	360
Pour test, meth	hod of making)1-4
Pressure		
Fressure		
Effect of p	ressure on extraction of oil from sand	4
Effect of	various pressures on production of gasoline by	
erackin	10"	229
Clackin	ng or liquefaction of gaseous hydrocarbons	
Pressure 10	or inquefaction of gaseous hydrocarbons	251
Pressure ci	racking test for heavy hydrocarbons	9-20
Vanor pres	sure test of casinghead gasoline	321
Capiti pres	for the for units of programs	0
Conversion	factors for units of pressure 43	35-6
Pressure distill	late (see gasoline)	
	ite	176
Prices		
Cost of dri	lling oil wells	31
Duice of an	Illing oil wells ude oil at wells in various districts	33
Price of cru	aue on at wens in various districts	
Prices of r	efinery products	33
Price of cr	ude oil in Pennsylvania since 1859	34
Duisag of a	and all in Collifornia and Mid Continent folds	35
Frices of .C.	ude oil in California and Mid-Continent fields patents, refining and cracking)	20
Processes (see	patents, refining and cracking)	
Producer gas .	producers	186
Draduction and	nyodytoone	
riounction and	producers have been been been been been been been be	10
world's pro	oduction of petroleum by countries	12
By states a	and districts	3-15
Production	of assinghead gasoline	24
Delladouton	the state of the s	
Daily produ	uction of crude oil by pools in U. S. and Mexico 25	5-27
Production	and decline of average oil wells	28
Production	producers oduction of petroleum by countries	32
Manhathal's	divided in a divide company of with petrology	32
Marketed n	mineral products compared with petroleum	32
Production	of crude oil by companies in Mexico 36	6-39
Production	of crude oil in Kansas, California, Wyoming,	
Torog	Oklahema)-41
1 exas,	Oklaholila 40	
Producing	companies of the Standard Oil group	594
Production	of natural gas in the United States	$\begin{array}{c} 252\\ 252 \end{array}$
Dronggation of	or natural gas in the United States f explosions erties of tables, specifications and chemical properties) of oll from various sources	259
Tropagation of		404
propane, prope	srues of	130
Properties. (see	e tables, specifications and chemical properties) of	
crude (oil from various sources	20-1
Chamicaln	reportion of petroloum	100
Chemical p	toperties of perforeum	148
Or parainin	is, napinenes, oleiins, polymethylene, aromatics 12	28-9
Of various	types of lubricating oil	175
Protonoraffing		180
n i i i i i i i i i i i i i i i i i i i		100
Publications		
Patents		458
Books on n	petroleum, asphalt and natural gas 450	1-60
TI C Dunoc	of Minor bulleting and technical papers	1 0
U. S. Burea	433- betroleum, asphalt and natural gas	1-4
U. S. Burea	au of Standards publications	462
U. S. Geold	ogic Survey publications	463
II S Denau	rtment of Agriculture nublications	169
O. S. Depai	The first and the state of the	400
Smithsonia	in institute publications	463
Petroleum	trade journals	464
Technical	publications	464
State goolo	rie reports	ACT
Blate geolo	BIC LEDULD	400
Fyrometry app	lied to petroleum distillation 13	34-5
Pyroparaffins		180
Rooks rilles or	overning location of loading racks	6-98
Deartha, Iules ge		
nanger, Texas	india ali franci di ali ali ali ali ali ali ali ali ali al	1
Depth of v	vells at	5
Properties	of crude oil from	120
Fractional	gravity distillation of any do all from	100
Fractional	vells at. of crude oil from. gravity distillation of crude oil from.	120
Temperatu	re of oil from	5
Dewar & F	Redwood process for cracking oil	7-0
- Wigoogita- L	Redwood process for cracking oil	-1-3
viscosity .	Jy neuwoou viscosimeter	-289

12.7	Page
Refinery and refining	
Refinery and refining Refinery operation on crude oil. Refineries of North America classified by states and citles. Refineries of the Standard Oil group. Typical refinery practice.	42-52
Refineries of the Standard Oil group	59
Pyrometry in refining. Lubricating refinery terminology	134
Lubricating refinery terminology	175
Refined wax Fuel oil required in refining	176
Refining oil for road and paving purposes by sedimentation,	
Fuel oil required in refining. Refining oil for road and paving purposes by sedimentation, by dehydration, by fractional gravity distillation, by stear and inert gas distillation, and by air blowing.	190
troleum Determination of loss on refining of gasoline. Classification of patents on refining. Products of refining light oil from gas works.	335
Classification of patents on refining	138-458
Residues	241
Residues Uses for liquid residua Ultimate composition of regiduum	3-132
Ultimate composition of residuum	128
Carbon residue from cracking	156
Road oil	
Refined petroleum for:, Amount of road oil required per mile of road Specifications for Rocks (see geology) Rock Asphalt	191
Amount of road oil required per mile of road	200 203-6
Rocks (see geology)	203-0
Rock Asphalt :::::	194 176
Roll Oil. Rope lubricants, specifications for	172
Rope lubricants, penetration tests of	173
Production of petroleum in Oil gushers of	12
	30
Russia Production of petroleum in	12
Oil gushers of Properties of crude oil from	- 30
Properties of crude oil from	120-1
Salt plants, natural gas consumed by	250
Samples and sampling	
Preparation and apparatus for sample distillation of pe-	311-2
Method of shipping samples by express.	185
Sand and sandstone	91
Voids in oil sand oil in sand oil discharged from sand effect	
of viscosity, porosity and pressure on extraction of oil from oil sand	
of viscosity, porosity and pressure on extraction of oil from oil sand. Amount of oil extracted per acre from oil sand.	29
Tables for calculating volds in sand from weight per cubic foot Grading of Composition of asphaltic sand	201
Grading of	42-349
Composition of asphaltic sand	194
Sarco filler	208
Saybolt Universal chromometer	278-81
Scaling of asphaltic surface mixture	-196
Scaling of asphaltic surface mixture	284
Sedimentation, refining by	191
Sedimentation, refining by. Sewing machine oll. Shale oll and oll shale	176
Heating value of shale oil	190
Shale oil and oil shale Heating value of shale oil. Wax from shale oil. Occurrence of, commercial operation on, gasoline from, and description of shale oil. Resemblance of oil shale to cannel coal. Ammionium sulphate from. Map of oil shale areas.	180
description of shale oil	233-5
Ammonium subpate from	233-4
Map of oil shale areas.	233-4

ŝ

	Page
Fractional gravity distillation of shale oil before and after	237-8
cracking Graphic representation of relation of gravity distillation, temperature and percentage distilled of shale oil before and after cracking	
temperature and percentage distilled of shale oil before and after cracking	939-40
Sheet asphalt (see asphalt)	192-5
Shell group (see Dutch Shell)	
Shipping regulations by express Shrinkage of oil on cracking	91 228
Sicily, asphaltic limestone from	194
Sicily, asphaltic limestone from Signal oil, specifications and tests of Sludge acid, treatment of	153
Snudge acid, treatment of	138 250
Smelters, gas used by Sodium hydroxide as reagent Gravity and strength solutions of	360
Gravity and strength solutions of	425 131
Sodium hypohypomite as respont	360
Sodium hypobromite as reagent	297
Solar oil	132
and carbon tetrachloride	42-346
Sources of crude oil in the U. S	
South Carolina	15
State kerosene inspection laws for	156
State gasoline inspection laws for	143-4
State kerosene inspection laws for	156
State kerosene inspection laws for State gasoline inspection laws for	143-4
Space, units for measuring space and capacity	431-2
Specific gravity (see gravity)	
Specific heat of petroleum	185
Specific heat of petroleum Of air, carbon dioxide, hydrogen, olefins, methane, nitrogen, oxygen, and water vapor	254
Specifications (see also methods of analysis)	
Specifications (see also methods of analysis) U. S. gasoline specifications	139
Motor gasoline	141
Mineral spirits	142
Aviation gasoline	143-4
Navy specifications for gasoline, aero gasoline, export gaso-	1000
Navy specifications for gasoline, aero gasoline, export gaso- line, domestic gasoline U. S. specifications for water white kerosene Navy specifications for kerosene Navy specifications for long time burning oil	146 149
Navy specifications for kerosene	149
Navy specifications for long time burning oil	150
Lighthouse oil 300° mineral seal oil	151
Signal oil	152 153
Gas oil	154
Bunker oilStraw oil	$154 \\ 155$
	155
Claroline oil	155
Petroleum products of the Kansas City Southern Ry, Co	$ \begin{array}{r} 156 \\ 157 \end{array} $
Petroleum products of Pennsylvania railroad	158-9
U. S. specifications for lubricating oil, aero oil	$\begin{array}{r} 168 \\ 169 \end{array}$
Aeroplane machine gun oil	169
Absorption oil Claroline oil State kerosene inspection laws Petroleum products of the Kansas City Southern Ry. Co Petroleum products of Pennsylvania railroad U. S. specifications for lubricating oil, aero oil. U. S. specifications for motor oil. Aeroplane machine gun oil. Transmission lubricants Non-fluid transmission lubricants. Medium cup grease. Gun oil	170
Medium cup grease	$170 \\ 171$
Gun oil	172 173
	173
Mineral cylinder oil. Compound cylinder oil. Petroleum jelly of the U. S. P. Paraffin wax of the U. S. P. U. S. Navy fuel oil.	174
Petroleum jelly of the U. S. P.	178
Paraffin wax of the U.S. P.	179
Diesel engine oil	182 183

KANSAS CITY TESTING LABORATORY

	Page
Asphaltic cement	197
Topeka specifications Purpose of asphalt specifications. Typical specifications for asphaltic surface mixtures. Road oil Specifications for Mexican, Texaco and Sarco brick and block	195
Purpose of asphalt specifications.	198 202
Typical specifications for asphaltic surface mixtures	203-6
Road oil	203-0
filler	207-8
	201-0
Spindle oil	-4-176
Stack area for fuel oil furnaces	184
Stanolind asphalt	193
standard off group, refiners, marketers, producers, pipe fines and carriers	59
	and the second s
Steam required for atomizing fuel oil	184
Steam refining	191
Steel	
Effect of oil cracking temperatures on strength of	225
Fuel oil required to melt	184
Stills	
Calculation of capacity of horizontal stills	00-224
Heating area of	224
Advantage of continuous stills	224
Dewar & Redwood stills	218
Burton stills	220
Benton stills	216
Stitching oil	176
Storage	
Losses of petroleum in storage from evaporation, fire and	
seebage	87
seepage	
Submerged storage	87 87
Specifications for fuel oil storage tanks	89-90
Submerged storage Specifications for fuel oil storage tanks Rules governing location of gasoline storage	96-98
Stratigraphic section of rocks in Oklahoma and Kansas	8-11
Straw oil	32-155
Straw oil	131
Structures, domes and anticlines	4
Sulphonic acid	137
Sulphuric acid as reagent	360
For treatment of refined petroleum	131
Gravity tables for	423-5
Sulphur, amount of in petroleum	128
Belation to odor and color	136
Effect on refining and on condensers	137
Distribution in neuroleum distillation	37-167
In fuel oil Heat of combustion of. Removal of, from gas.	182
Heat of combustion of	190
Removal of, from gas	242
Determination of	332-4
Summer oil	176
Surfaces, tables of equivalents of units of surface and square	
measure	430
Surface mixture (see asphalt)	
Sweeting of Way 1	80-176
Sweating of Wax1 Synclines, illustration of accumulation of oil in	7
Synthetic gasoline (see cracked gasoline)	-
Tables	
Of units of measurement of petroleum and water For capacity of horizontal cylindrical tanks and tank cars1	
For canadity of honizontal aviandaical tanks and tanks and	99
For capacity of norizontal cymunical tanks and tank cars	99 00-119
Outage tables of tank cars	99 00-119 00-119
Outage tables of tank cars	99 00-119 00-119 120
Outage tables of tank cars	99 00-119 00-119 120 62-164
Outage tables of tank cars	99 00-119 00-119 120 62-164 187
Of properties of crude oils	99 00-119 00-119 120 62-164 187 190
Outage tables of tank cars	00-119 120 62-164 187 190
Outage tables of tank cars	00-119 120 62-164 187 190 201
Outage tables of tank cars	00-119 120 62-164 187 190 201 251
Outage tables of tank cars	00-119 120 62-164 187 190 201 251 257
Outage tables of tank cars	00-119 120 62-164 187 190 201 251

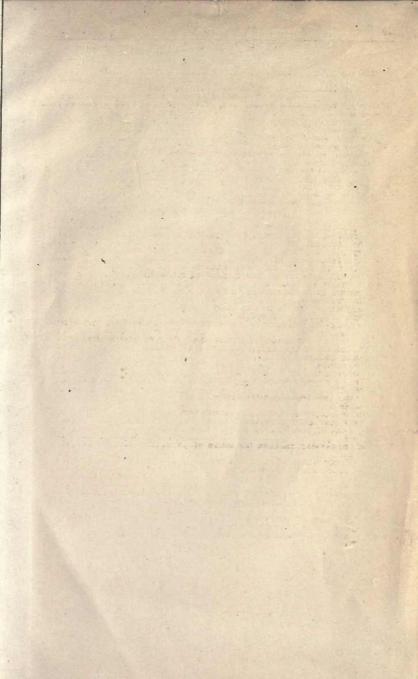
	Page
For correcting Saybolt, Engler and Redwood viscosities to	1 007
each other Of equivalents of Fahr. and Cent. degrees Of Bureau of Standards for equivalents of specific gravity, Baume' gravity and weight per gailon	367-369
Of Bureau of Standards for equivalents of specific gravity,	001 000
Baume' gravity and weight per gallon,	370-1-2
Of Detroloum Association for equivalents of specific gravity	
Baume' gravity and weight per gallon	373-5
C correction of Baume' gravity to 60°F	376-383
Of equivalents of Baume' specific gravity and weight per	004-418
gallon for liquid heavier and liquids lighter than water.	419-422
Of specific gravity and strength of sulphuric acid Of specific gravity and strength of sodium hydroxide	423-5
Of specific gravity and strength of sodium hydroxide	426
Of specific gravity and strength of calcium chloride and brine	427
solutions	428
Of equivalents of units of linear dimension	429
square measure, surfaces, areas	430
of fundamental equivalents of linear dimension	431
British liquid and dry	432
weight	434
work	434 435-6
Tarliabue hydrometer scale	499-0
money	437
	272
Viscosimeter Flash tester	284
	313
 Tank and tank cars (see storage) Specifications for fuel oil tanks. Ownership of tank cars. Formula for calculating content of horizontal tanks. Gauging tables and outage tables for capacity of horizontal cylindrical tanks containing varying amount of oil 	00 00
Ownership of tank cars	92-5
Formula for calculating content of horizontal tanks	100
Gauging tables and outage tables for capacity of horizontal	
cylindrical tanks containing varying amount of oil	100-119
Tar (see asphalt)	132
Tar stills	176
Tempering oil	
Tempering oil	176 176
Tempering oil Temperature Effect of temperature on viscosity	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 376-383 \\ 384-418 \\ 437 \\ 376-383 \\ 384-418 \\ 437 \\ 376-383 \\ 384-418 \\ 437 \\ 384-418 \\ 384-$
Tempering oil Temperature Effect of temperature on viscosity	176 176 167-188 165 254 364-5 267-8-9 376-383 384-418 437 143-4
Tempering oil Temperature Effect of temperature on viscosity	176 176 167-188 165 254 364-5 267-8-9 376-383 384-418 437 143-4 156
Tempering oil Temperature Effect of temperature on viscosity. Effect of high temperature motors on lubricating oils. Temperature of ignition of gases. Conversion tables for gas volumes. Conversion of Cent. and Fahr. scales. Tables for correction of Baume' gravity readings to 60°F Tables for correction of specific gravity readings to 60°F Conversion factors for units of temperature. Tennessee State gasoline inspection laws for State kerosene inspection laws for Tertiary, character of oil from tertiary formation,	176 176 167-188 165 254 364-5 267-8-9 376-383 384-418 437 143-4
Tempering oil Temperature Effect of temperature on viscosity. Effect of high temperature motors on lubricating oils. Temperature of ignition of gases. Conversion tables for gas volumes. Conversion of Cent. and Fahr. scales. Tables for correction of Baume' gravity readings to 60°F Tables for correction of specific gravity readings to 60°F Conversion factors for units of temperature. Tennessee State gasoline inspection laws for State kerosene inspection laws for Tertiary, character of oil from tertiary formation,	$176 \\ 176 \\ 167 - 188 \\ 165 \\ 254 \\ 364 - 5 \\ 267 - 8 - 9 \\ 376 - 383 \\ 88 - 418 \\ 437 \\ 143 - 4 \\ 156 \\ 4$
Tempering oil Temperature Effect of temperature on viscosity. Effect of high temperature motors on lubricating oils. Temperature of ignition of gases. Conversion of Cent. and Fahr. scales. Tables for correction of Baume' gravity readings to 60°F. Tables for correction of specific gravity readings to 60°F. Conversion factors for units of temperature. Tennessee State gasoline inspection laws for. State kerosene inspection laws for. Tertiary, character of oil from tertiary formation, Tests and testing (see analysis)	176 176 167-188 165 254 364-5 267-8-9 376-383 184-418 437 143-4 156 4 130
Tempering oil	$\begin{array}{c} 176\\ 176\\ 167-188\\ 165\\ 254\\ 364-5\\ 267-8-9\\ 376-383\\ 884-418\\ 437\\ 143-4\\ 156\\ 4\\ 130\\ 130\\ \end{array}$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$176 \\ 176 \\ 167-188 \\ 165 \\ 254 \\ 364-5 \\ 267-8-9 \\ 976-383 \\ 884-418 \\ 437 \\ 143-4 \\ 156 \\ 4 \\ 130 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 130 \\ 207 \\ 143-4 \\ 150 \\ 10$
Tempering oil	$\begin{array}{c} 176\\ 176\\ 176\\ 167-188\\ 254\\ 364-5\\ 254\\ 364-5\\ 254\\ 437\\ 165\\ 4\\ 130\\ 130\\ 207\\ 6\\ 19\\ 30\\ 207\\ 6\\ 19\\ 30\\ 207\\ 122-6\\ 156\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 12$
Tempering oil	$\begin{array}{c} 176\\ 176\\ 176\\ 167-188\\ 254\\ 364-5\\ 254\\ 364-5\\ 254\\ 437\\ 165\\ 4\\ 130\\ 130\\ 207\\ 6\\ 19\\ 30\\ 207\\ 6\\ 19\\ 30\\ 207\\ 122-6\\ 156\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 12$
Tempering oil	$\begin{array}{c} 176\\ 176\\ 176\\ 167-188\\ 254\\ 364-5\\ 254\\ 364-5\\ 254\\ 437\\ 165\\ 4\\ 130\\ 130\\ 207\\ 6\\ 19\\ 30\\ 207\\ 6\\ 19\\ 30\\ 207\\ 122-6\\ 156\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 12$
Tempering oil	$\begin{array}{c} 176\\ 176\\ 176\\ 167-188\\ 254\\ 364-5\\ 254\\ 364-5\\ 254\\ 437\\ 165\\ 4\\ 130\\ 130\\ 207\\ 6\\ 19\\ 30\\ 207\\ 6\\ 19\\ 30\\ 207\\ 122-6\\ 156\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 126\\ 12$

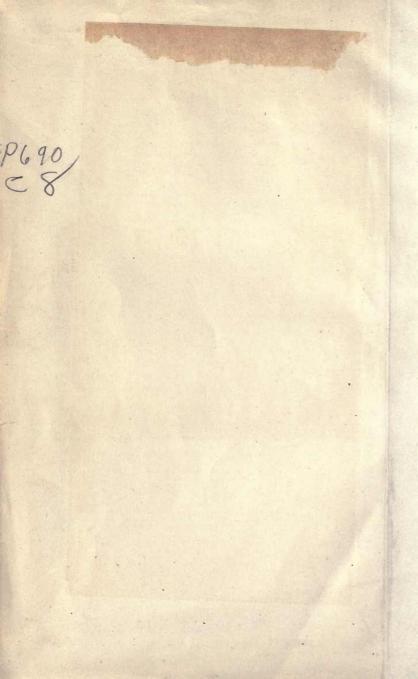
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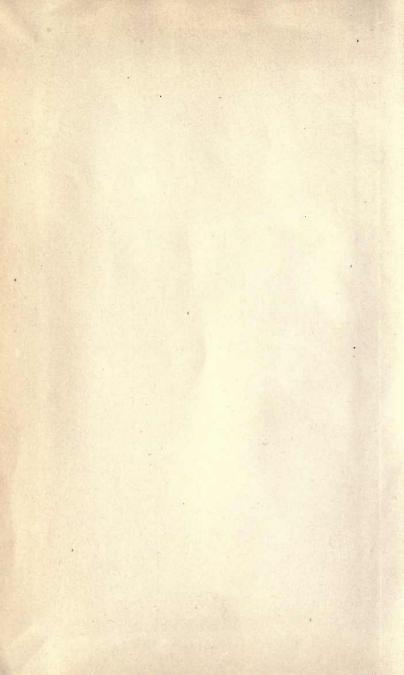
	Page
Tintometer, Lovibond Toluol	138
Topueda specifications Topping crude oil (see dehydration)	192
Topping crude oil (see dehydration)	176
Tower stills	-1-128
Transformer oll	127
Transmission lubricants	170-177
Trinidad Bronerties of petroleum in	12
Properties of petroleum in Ultimate analysis of Trinidad asphalt Composition of Trinidad asphalt	128
Composition of Trinidad asphalt	193
Turbine oil	290-1
Ubbelohde viscosimeter Unsaturated hydrocarbons (see olefins and ethylenes)	290-1
United States	and the second
Demand for fuel of the U.S. Navy	2
Number of outomobiles in the U.S.	12-15 ²
Production of petroleum in. Sources of crude oil in. Disposition of crude oil in.	13 13
Disposition of crude oil in Oil fields of	13
Refineries of	22 42-52
Refineries of Benzol, manufacturers of Marketed crude oil in.	80-1 14-15
Pipe lines of	83-86
Pipe lines of	139-146 461-2
U. S. Bureau of Mines publications.	462
U. S. Geologic Survey publications	463
U. S. Department of Agriculture publications	463 145
U. S. Bureau of Mines publications. U. S. Bureau of Standards publications. U. S. Geologic Survey publications. U. S. Department of Agriculture publications. U. S. Pharmacopoeia, benzine. Petrolatum	177-8
Parallin wax	179
Utah State gasoline inspection laws of	144
State kerosene inspection laws of	156
Vacuum, effect of vacuum distillation on viscosity and sulphur content	167
Value of Petroleum as a mineral product	2
Vapor phase cracking of oil	215
Vapor space, effect of on cracking of oll	215 185
Vaseline	177
Velocity, conversion factors for units of	437 156
Velocity, conversion factors for units of Vermont, state kerosene inspection laws of Virginia, state kerosene inspection laws of	156
Viscosity	
Effect of cracking of oil on viscosity	81-228
Effect of character of distillation on viscosity	167
Relation of viscosity to temperature of oil.	181
Effect of an extraction of oil from sand	199
Determination of vigoaity with the Carbolt Hadron air	187-8-9
cosimeter	204
With the MacMichael viscosimeter	285 293
Of asphalt by the float test	294
With the Engler viscosimeter. With the Engler viscosimeter. Of asphalt by the float test. Of kerosene and gasoline by the Ubbelohde viscosimeter Of semisolid asphalt by the zero viscosity method.	290-1 295
Voids in oil sand	495
Voids in oil sand	201
Volcanic action, relation of to petroleum	4
Warrenite	195
State kerosene inspection laws for	156
State gasoline inspection laws for	143-4

		and the second second	Page
Watch oil			177
Water			
Association of salt water with oil	and gas		4-5
Equivalent of units of measurement			99
Water white distilate before and afte:	r cracking (se	e kerosene)2	11-2-3
Water gas			
Heat of combustion of			190
Explosion of			252
Orifice capacities for			266-8
Wax (see paraffin wax)			200 0
Amount of wax produced by refin	eries		32
Refined wax			76
Specifications for U. S. P. wax			179
Solubility and valuation of			180
Wax distillate			177
Waviness of asphalt			196
Wearing surface (see asphalt)			100
Weight, conversion factors for units	of weight		433
Wells	or weight		100
Depth of oil wells			
Effect of depth of wells on qualit	v of oil		
The deepest well			
Depth of wells at Ranger, Texas.	••••••••••••••••		5
Temperature in oil wells			6
Effect of depth on temperature.			6
Average production and decline of	oil wells		28
Oil wells in Mexico			28
Number of wells drilled in 1918	•••••••••••••••	•••••	29
Oil gushers			30
Largest oil wells in the world			30
Cost of drilling oil wells		•••••	31
Testing the capacity of casinghea	d gog molle mi	th the orifice	91
			253
Pitot tube for measuring the ope	n flow of gog	walla	255-6
Tables for determining flow of ga	I now of gas	wens	
Westphal balance			273-6
West Virginia			210-0
Temperature in deep wells of			6
State kerosene inspection laws of.			156
Winter oil			177
Wire rope lubricant specification			172
Wisconsin	•••••		114
State gasoline inspection laws for	r		144
State kerosene inspection laws for	r		156
Wood, heat of combustion of			190
Wool oil			177
Work, conversion factors for units of.			434
Wyoming			101
Geologic occurrence of oil in			6
Oil fields and pipe lines of			18
Daily production by pools			25
Oil companies of			41
Oil companies of Properties of crude oil from			120
State gasoline inspection laws for			144
Xyloi, properties of			138
Young process of cracking			217
Zero viscosity of asphaltic cement			295
Zinc smelters, natural gas consumed b	v		250
bane same ters, natural Bab consumed p			200











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