## TRANSMISSION OF HEAT THROUGH TILE & CONCRETE FIREPROOFING BY C. A. SNOW

# ARMOUR INSTITUTE OF TECHNOLOGY 1912

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## A Study

of

THE TRANSMISSION OF HEAT

THROUGH TILE AND CONCRETE FIREPROOFING

A THESIS

Presented by C. A. Snow

to the

PRESIDENT and FACULTY

of

ARMOUR INSTITUTE OF TECHNOLOGY

For the Degree of

BACHELOR OF SCIENCE IN FIRE PROTECTION ENGINEERING

Having Completed the Prescribed Course of Study in Fire Protection Engineering.

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May 29, 1912.

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#### PREFACE

Although considerable thought and attention has been given to the value of various materials as fire retardents, and thorough tests have been made to ascertain their ability to resist destruction by fire, only a small amount of work has been done to determine the amount of heat actually passing through the materials at the high temperatures which are found in burning buildings. It is the purpose of this thesis to establish, through experiment, a basis from which the transmission of heat through tile or concrete fireproofing may be determined.

#### THE TRANSMISSION OF HEAT

## THROUGH TILE AND CONCRETE FIREPROOFING.

#### EQUIPMENT.

### Construction of Tile Blocks.

The blocks were made from tile slabs. so placed that there was an air chamber between. The slabs were about 17-1/2" long and 12-1/4" wide, none varying more than 1/8" from the above measurements in either length or width, and were medium burned and rather porous. The blocks were built up by placing shellacked wedge forms between each slab 1-3/4" in from each side on three sides, and filling in up to them on the three sides of the block.with concrete. This concrete was a 1 - 3 mixture, having one part portland cement to three parts 1/8" screened torpedo sand. It was re-enforced by two wire rods on the rear side, and one wire rod on each end of the block. The concrete extended from the wedge forms out a distance of 3-1/4", making a wall of concrete around the tile 1-1/2" thick, not including the concrete between the slabs. This can readily be seen in figure No. 3. which is block No. 2 cut through the center after the test.

The block was allowed to set for fortyeight hours before removing the wedge forms. The cells were then measured, and the block was buttered up with concrete on the fourth, or front, side, leaving a space of about 1-1/2" for the introduction of the thermocouples, as shown in Fig. 2. The block was then allowed to set for one week, at the end of which time it was placed in a drying kiln and kept at a temperature of from 212° F.to 220° F.for sixty-six hours. The block was then ready for the test.

All measurements for each tile block are given on plate No. 4. The slab thicknesses given on plate No. 4 are the average of eight measurements taken on each slab just before being built into blocks.



The depth of cell measurements given are the average of three measurements taken with internal calipers just before the block was buttered up, on the front side, and a measurement taken after the test when the block was parted centrally. This latter measurement was taken on a line 6" from the right side of the cells, and 6-1/8" in from the buttered up face of the block. In all cases the latter measurement checked within 1/32" of the average of the first three readings taken. The length of the cells was found to be 13-1/2" and the width 9-1/2", making an area of tile section 128.25 square inches, straight through the block. The concrete in the buttered up side, as shown in figure No. 3, appears to enter the cells to a considerable depth but on close examination it was found that it only adhered to the tile slabs for an inch in, or less, the remainder merely projecting into the cell but not touching the tile.

## Construction of Concrete Blocks.

The concrete blocks were made in a mold or form, constructed as follows :- The sides are each 8" in height and 18" long. The ends are 8" in height and 12" long. The sides and ends overlap each other in such a way that a double joint is formed, making a water tight corner. They both have guide strips at their lower edges, which overlap the bottom of the mold and serve not only to keep the sides and ends in place, but they are staunch enough to prevent any warping of those parts. The bottom of the mold is made of heavy 2" planking through which six holes are drilled to allow for the placing of six 5/16" drill rods. These holes are located on the circumference of a circle, 6" in diameter, having its center in the center of the 12" x 18" face, and are each 3" apart. The drill rods project up through the bottom of the mold to heights of 7", 6", 5", 4", 3" and 2" from the bottom. This allows the thermocouples, which are placed in the holes thus made in the concrete. to penetrate into the block at distances of 1", 2", 3", 4". 5" and 6" from the exposed face. The mold is

ŧ • \* held together by four boards hinged at the corners and brought up tightly by means of two wedges. The entire inside of the mold is heavily shellacked and the rods are thoroughly greased.

A 1-3-5 mixture was used in making the blocks. The sand used was torpedo bank sand, passing a 1/8" screen, while the aggregate consisted of pebbles less than 1/2" in diameter, together with some very coarse gravel. Enough water was added to make the mixture quite workable, then it was shoveled into the mold, tamped, and the top struck off even with the top of the mold. By following the above method in each cease, each block resulted in being exactly the same in all respects, as the others.

The concrete blocks were allowed to set for three weeks before they were dried out. During the first week of setting, they were wetted once each day. At the end of three weeks the blocks were placed in the drying kiln and kept at a temperature of from 2120 F. to 2200 F. It was the intention of the writer to dry each block for sixty-six hours, but the burner under the kiln was tampered with while blocks No. 1 and No. 2 were being dried, and it was found out later on that they were only dried for a period of fifty-two hours. Block No. 3, having been dried out at a different time, was given the full sixty-six hours of drying. The purpose of the drying was to drive out all of the uncombined water, and thus enable the test conditions to be accurately duplicated, which could not be done if varying quantities of water were contained within the concrete.

A view of one of the concrete blocks, showing the 12" x 18" face opposite the 12" X 18" face which is to be exposed, is shown in Fig. 4.

#### Construction of Furnace.

A rear view of the furnace used for these tests is shown in Fig. 1. The front or open face of the furnace contains an opening 33" wide and 35" high, measuring from the keystone of the erch to the base of e de la companya de

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the opening. This base is 17" from the floor. The inner walls, base and arch are made of the best quality fire-brick, cemented with a portland cement mortar. The outer walls are made of pressed brick and an air space separates them from the inner walls.

Four burners, which are in principal, larg Bunsen burners, enter each side of the furnace and all eight burners are directed against the back wall of the furnace. In this way an even, uniformly distributed heat reaches the front face of the opening. The arch which forms the top of the furnace contains five rather large openings which allows the gaseous products of combustion to escape. A blower, which is operated by a motor that can be run on a llo V. or 220 V. circuit, supplies the air to the burners at a point just beyond the gas control valves. A small hole in the right side of the furnace allows for the placing of the furnace thermocouple.

#### Construction of Thermocouples.

Four of the seven thermocouples used were made of platinum and iridium, while the other three were made of copper and constantan. The platinum iridium couples were made of platinum and iridium wires, the terminals of which were welded in an electric arc. This comprised the hot junction which was always placed where the temperature measurement was desired. The wires passed from the hot junction through a porcelain tube, having two separate chambers. From the porcelain tubes they ran through concentric asbestos tubes, one wire passing through the center and the other through the space between the tubes. The wires were then soldered to copper leads, these joints being the cold junction which was kept in ice during the test. The copper constantan couples were made in the same way except that they could be welded with a blow pipe instead of the electric arc.

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### METHODS

## Calibration of Thermocouples.

The melting points of copper, zinc, tin and ice were utilized in calibrating the platinum iridium couples. The material that is to be used for calibrating is placed in a crucible and placed in a small electric furnace and melted. The hot junction of the thermocouple is placed in the center of the molten metal, the cold junction is placed in tubes which extend into ice, while the ends of the copper leads are connected to a galwanometer. The electric furnace is then turned off and the metal is allowed to cool while readings are taken on the galvanometer. The point or temperature at which the metal begins to solidify is indicated by a succession of constant readings on the galvanometer. The taking of readings is taken a little beyond this point to make certain that the true freezing or melting point The same method is used with each has been obtained. metal. Knowing the melting points of each metal used and getting the corresponding galvanometer readings a calibration curve may be drawn for each thermocouple, plotting temperatures as ordinates and miltivolts or galvanometer deflections as abscissa.

The copper constantan couples are calibrated in exactly the same way as the platinum iridium couples, except that the melting point of copper is not utilized for obvious reasons. Copper constantan couples are not considered reliable for measuring temperatures above 1000° F. so for all temperatures above that the platinum iridium couples must be used.

#### Position of the Blocks.

Each block that was tested was placed in the movable partition shown in Fig. 1. The opening in the brick partition was so located that where the partition was rolled into place in front of furnace the block was in the center of the furnace opening. In the tests on tile blocks the partition opening was made 16" high and 21" wide. This allowed about 3/8" on the sides and bottom for asbestos packing

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and about 3/4" at the top for the same. A rack was made on the frame-work back of the partition to support that part of the tile block extending beyond the brick-work of the partition.

The tile block, after having been dried out, was placed in the partition with the side containing the openings uppermost. The space between the brick-work and the block was then thoroughly plugged up with asbestos fibre. In Fig. 5 one of the tile blocks is shown in the partition placed as described above. Although this photograph was taken after the test, it shows the way the tile blocks were laid into the partition. Care was taken to have the exposed face of the block in the same plane as the face of the partition.

In the case of the concrete blocks the same method of placing the block in the partition was used, but the brick-work had to be rebuilt to accomodate the smaller exposed face of these blocks. In figure 1 a concrete block is shown in place in the partition, and ready for the test.

## Position of Thermocouples.

The furnace thermocouple was so placed that when the movable partition was rolled into the position for testing its hot junction was within 1/4"of the face of the block, being tested and at the center of the exposed face.

In the tile blocks the thermocouples were placed as follows:- Each couple was lowered to a depth of 6" in the cells and was placed against the face of the slab farthest away from the source of het. This located the point of the couple 1/8" away from the rear face of each cell. For example: the couple in cell No. 1 is at a distance of the thickness of the first slab, plus the depth of the cell minus 1/8" from the exposed face of the block. The distance of each couple from the exposed face can be figured in a similar menner. After the couples were so

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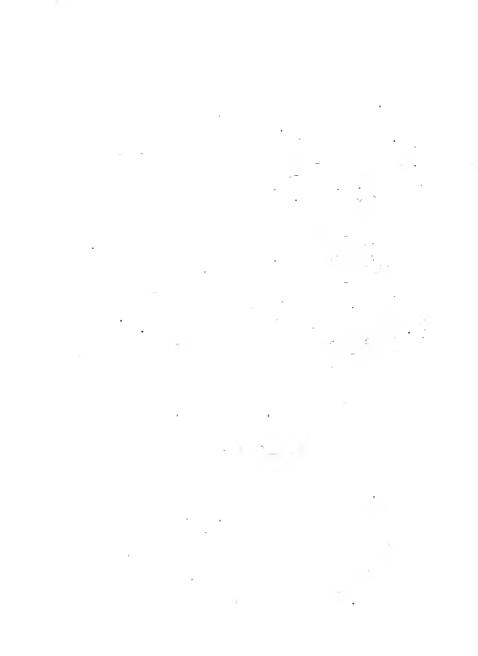
placed, the openings through which they were suspended were plugged up with asbestos fibre, thus holding the coupled in place, and completely closing the cell. Due to the high temperatures which were anticipated and which did occur in cells Nos. 1, 2 and 3, the platinum-iridium couples were used in these cells. The copper-constantan couples were used in cells Nos. 4, 5 and 6, where the temperatures never were above 1000° F.

The holes made in the concrete blocks by the drill rods were  $1/32^{n}$  larger in diameter than the porcelain tubes containing the couple wires. This gave a very close fit, making it unnecessary to do any filling with asbestos fibre. The depth of each hole was re-measured before inserting the couples in them, and they were all found to be accurate to  $1/64^{n}$ of an inch. As previously stated, the holes were so made that the points of the couples were  $1^{m}$ ,  $2^{m}$  and  $3^{m}$  $4^{m}$ ,  $5^{m}$  and  $6^{m}$  away from the exposed face. Platinumiridium couples were used at the  $1^{m}$ ,  $2^{m}$  and  $3^{m}$  depths and copper-constantan couples were used at the  $4^{m}$ ,  $5^{m}$ and  $6^{m}$  depths.

In all tests each couple was so connected that it could be thrown into series with a galvanometer by a double throw, double pole, knife blade switch.

## Procedure During Test.

The partition was first lined up so that it could be rolled within 1" of the open face of the furnace. A shield made of sheet iron backed up with asbestos paper was hooked on to the furnace so that it covered the entire front face. The pilot lights were then lit and the blower started. The gas to the burners was then turned on and the furnace brought up to the required temperature of the test. Cracked ice was placed in the junction bottles and a zero, or the starting reading was taken. The furnace shield was then unhocked and removed and the partition rolled into place. The partition was jammed tight against



the furnace face by means of long bars, and was kept there by means of wedge blocking. The furnace temperature was then regulated and maintained constant for four hours. The ice in the junction bottles was replenished from time to time and all necessary observations were made.

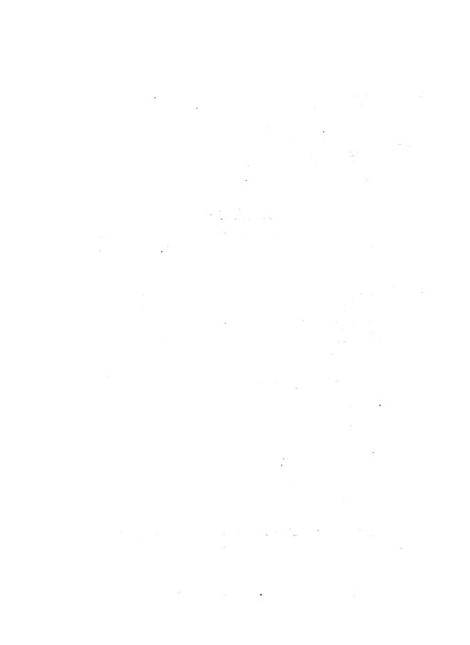
> The blocks were allowed to cool normally after each four hour test, no water being thrown on them.

## Readings.

Temperature readings were taken every three minutes for the first half hour, and every five minutes from then on to the end of the test. The readings for the first half hour were taken closer together than those of the last three and a half hours in order to get more accurately the effect that any retained moisture might have on that part of the blocks near the exposed face, for the temperature here rises very rapidly and the true shape of the curve, which is ploted from these readings, could not be obtained unless this were done.

The readings taken were in milli-volts as represented by the deflection of the galvanometer needle. The deflection is due to the electro-motive force generated by the thermocouples. This electromotive force is in proportion to the difference in temperature between the hot and cold junctions of the couples. If the cold junction is maintained at a constant temperature, it is obvious that the galvanometer reading depends on the temperature of the hot junction at the point where it is desired to measure the temperature.

The temperature in degrees Fahrenheit, as shown on the data sheets opposite the milli-volt reading, was obtained in each case by referring to the calibration curve of that particular thermocouple. It was found that one and one-half minutes was needed to obtain all seven thermocouple readings at each three or five minute interval. This cannot be considered as



instantaneous, but since the readings are relative this does not affect the value of the data in any way.

#### DISCUSSION OF RESULTS

Condition of Blocks After Test.

Although this part of the subject is not directly connected with the work at hand, it may be of interest to know the condition of each block after the test, therefore this date is tabulated as follows:

Tile Block No. 1. Test No. 1. 1300 ° F.

The concrete around slab No. 1 was heavily cracked and came off upon removal of block from partition. That part of the concrete which fell off could be quite readily broken with the hands.

Slab Nolheavily cracked and slightly separated down through the center. This slab could be broken up with the hands. It was a much lighter color than before the test.

Slabs Nos.2.3 and 4, lightly cracked across center.

Slabs Nos. 5.6 and 7, all sound.

Tile Block No. 2. Test No. 2. 1600 ° F.

This block was affected in the same manner as block No. 1, only the cracks in slabs Nos. 1, 2, and 3 were more severe.

This block is shown in Fig. 5, as it was just after the test. The dark spots noticeable on the exposed face are due to rust from the iron re-enforcing wir which were looped under the block in the process of building and were afterwards cut off even with the surface. The heavy oracking referred to is plainly noticeable here.

The cracking of the concrete in this block is notice able in Fig. 3.

Tile Block No. 3. Test No. 3. 1900 ° F. All slabs were cracked, slabs Nos. 1 and 2 being

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Concrete Block No. 1. Test No. 4. 1300 OF.

The exposed face was covered with a network of very fine surface cracks, and was a bluish grey in color. The concrete was discolored to a depth of 4-1/8" from exposed face. The color of the stones in the aggregate varied from a dark salmon red at the face (hav ing some red spots in the center due to iron oxide) to a light salmon color, having no bright red spot in center to a depth of about 3" from exposed face, and at 4" from exposed face, the stones were a dull brownish red.

The general coloring of the concrete was: Brownish grey to a depth of 1/2"; Selmon red from 1/2" to 1-3/4" in from exposed face Muddy grey from 1-3/4" to 3-3/4" " " " Yellowish grey from 3-3/4"to 4'8" " "

A screw driver could be forced in to a depth of 1/4" without pounding.

Concrete Block No. 2. Test No. 5. 1600 ° F.

Discolored to a depth of 4-7/8". Exposed face blue grey and covered with surface cracks.

Coloring:

Blue grey - 3/8" in from exposed surface, Light pink tinge - 3/8" to 1-3/8" in from exposed face Muddy brown - 1-3/8" to 3-5/8" " " " A Pink tinged with purple, 3-5/8" to 4-7/8" " The remainder of the block was a little lighter

grey than normal concrete.

A screw driver could be forced into a depth of  $1/2^{n}$  without pounding.

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Concrete Block No. 3 Test No. 6. 1900 OF.

About 1" of exposed face came off near lower corners when block was removed from partition. Exposed face blue grey in color and covered with fine surface cracks. Discolored to a depth of 5-3/4" Coloration: Very dark grey - 1" in from surface. Purplish brown - 1" to 2-1/4" in from surface Grey - 2-1/4" to 4-1/2" " " "

A screw driver could be forced in to a depth of 1" without pounding.

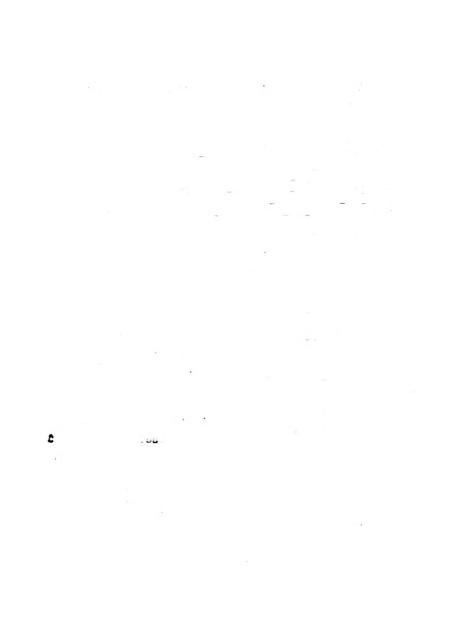
The foregoing coloring described was in every case very faint and the colors were in all cases only tints and not as vivid as the description might imply.

#### Discussion of Temperature Curves.

The curves plotted from the reading of the tests on tile will be considered first, and their significance discussed before dealing with those plotted from the readings of the tests on concrete.

The readings of tests Nos. 1, 2 and 3 are plotted on plates Nos. 10, 11 and 12, respectively. Since these curves are ploted with time as the abscissae, the slope of the curves shows the rate of change of temperature in the cells throughout the tests. The points as plotted were so close to each other that it seemed advisable not to strike a smooth curve through them, so they were connected to each other by straight lines, and in this way each curve is accurately defined.

In each of these three sets of curves the effect of replenishing the supply of ice in the junction bottles is quite noticeable, especially where a considerable length of time elapses between the periods where



fresh ice was added. It is obvious that this rise is due to the greater difference in temperature between the cold and hot junctions, which occurs when fresh ice is added.

It will be noticed that the furnace temperature curves rise and fall very steeply for about the first half hour. This could not be avoided because the placing of the partitions in position at the start, tended to make the furnace temperature drop and in bringing it up again it always overshot the desired temperature, so the resulting curve is very irregular. This, however, has practically no affect on the block curves, as the block is only at room temperature when placed in the test position, while the furnace temperature is very much higher.

The irregularities of the curves having been explained, their general form may now be considered:

All curves on plates Nos. 10, 11 and 12 continue to rise throughout the entire four hours. This indicates that equilibrium of the block temperatures is never reached. If equilibrium were reached, the quantity of heat passing through the part of the block near the exposed face would be exactly equal to the heat going through another part of the block farther out from the furnace, otherwise equilibrium is impossible. This is evidence showing that most of the heat entering the blocks goes to raise the temperature of the latter, only a small percentage passing through and being lost by radiation. If. however, tests were made for a longer period of time, it is fair to assume that equilibrium would be established in a few more hours. The curves indicate that this would be so, for they tend to become horizontal, which means that a constant temperature is being approached. If such a condition did obtain, all of the heat entering the blocks would be lost in radiation.

The significance of the above will be shown farther on in this report.

The readings of tests Nos. 4, 5 and 6 are

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plotted on plates Nos. 13, 14 and 15 respectively. The same statement in regard to the furnace temperature curves of the tile block applies to these concrete block curves.

The effect of adding new supplies of ice to the junction bottles is noticeable on these curves to about the same extent that it is on the tile block curves, and the explanation for it is the same. The concrete block curves, however, differ from the tile block curves in two respects, namely, the effect of moisture in the concrete blocks shows up in all three sets, and these curves are at a greater angle to the horizontal time axis during the last hour of each run

This latter fact indicates that it would take a longer time for concrete to reach equilibrium temperatures than it would in the case of tile blocks It will be noticed that the effect of moisture in the curves of test No. 6 on plate No. 15 is less pronounced than in the curves of tests Nos. 5 and 6. Probably the reason for this is that concrete block No. 3 was dried for fourteen hours longer than either of the other two concrete blocks. This change of direction of the curves in the vicinity of 212 oF may not be due entirely to uncombined retained water, but it may be influenced somewhat by the water of crystalization in the concrete itself. It is safe to say that fully as much, and probably in all cases, more moisture will be found in the concrete used in building construction than was contained in any of the concrete blocks tested.

By further comparison of these curves with the tile block curves, it will be noted that the first point curves are lower in concrete than the first cell curves in the tile tests, but that the curves of points Nos. 2, 3, 4, 5 and 6 are higher than the curves of cells Nos. 2, 3, 4, 5 and 6. It is not fair, however, to compare these curves in this way since the tile couples in the tile cells are all considerably farther away from the exposed face than those corresponding thereto in concrete blocks.



A further basis of comparison may be shown by the curves on plates Nos. 16 and 17. These curves have distances from the exposed face as abscissae and temperature as the ordinates. The readings at the end of two and three-quarter hours were plotted on plate No. 16 for all six tests. This time was chosen because all conditions of furnace temperature were practically constant for a considerable period before this time, and the concrete block curves had risen above the 212 °F region. The curves on plate No. 17 were made in the same manner as those on plate No. 16, except that the time was the end of four hours, instead of two and three-quarter hours. The reason for not plotting the curves of test No. 6 was that the readings beyond two hours and fifty minutes are not considered reliable as noted in the data on data sheet No. 12.

In both of these sets it will be noted that for the same distances out from the exposed surfaces, the concrete block curves show a lower temperature than the tile block curves. However, the difference in construction of the two types of blocks and the detaining effect of moisture should be considered before drawing definite conclusions. This will be considered later in the report.

The quantity of heat flowing by conduction from one plane to another, through any portion of material, depends on the difference of temperature between these two planes and upon the resistance to the heat flow. With the same temperature difference, if the resistance is high, only a small quantity of heat flows through; whereas if the resistance is low a large quantity of heat flows through. If the quantity of heat remains constant the temperature difference must be large, if the resistance is low. When the quantity of heat passing between any two planes parallel to the exposed face is the same, the temperature difference between any two planes the resistance which the material or space between the two planes offers to the flow of heat.



For example, if the difference in temperature between, say 2" of concrete is high, it may be said that the resistance of concrete to the flow of heat is high. Thus it is possible to rely on the temperature differences, being a true indicator of high or low resistance to heat flow between any two planes which are parallel to the exposed face. This condition of equal quantities of heat passing through all parts of the block is only obtained at equilibrium, therefore, the results obtained from the tests made at this time can be used to indicate the relative heat resistance of the two kinds of fireproofing and not to obtain the absolute values of this heat resistance.

In the preceding discussion the flow of heat by conduction only has been considered, but in the case of tile blocks the flow of heat by radiation is of prime importance, and is discussed in detail under "Physical Laws". It has generally been assumed that an air space was a very good heat insulator. This is only true at the lower temperatures. While heat does travel very slowly through the air by conduction, it leaps over the air space readily by radiation. Although this latter mode of heat propogation is common in nature, the laws governing it are not generally known and taken into consideration.

### DISCUSSION OF PHYSICAL LAWS.

The quantity of heat passing through a portion of solid block or partition by conduction depends on the difference between the temperatures of the two planes limiting the portion of partition or block, but the quantity of heat that passes across the air spaces in tile blocks depends on the difference of the fourth powers of the absolute temperatured of the surfaces enclosing the air spaces. It follows that in case the heat passes by conduction through a solid block, the amount of heat passing will remain the same so long as the difference in temperature of the two limiting planes remains constant, no matter .

Page No. 17.

what that temperature may be. On the other hand, the heat passing across an air space by radiation increases very rapidly with the rising temperature of the enclosing surfaces, although the difference in temperature may remain constant.

The old law of raidation given by Isaac Newton, which stated that the heat radiated from a hot body to a cold surrounding body was proportional to the difference of their temperatures, has been proven faulty by Boltzmann and Stefan, who about twenty-five years ago demonstrated mathematically that from the principles of thermodynamics the fourth power law should hold exactly for an ideal black body.

This law is expressed by the following equation:  $H = C (T_1^4 - T_2^4)$ (1)Where H = the net heat exchanged between the hot and cold surface per unit of the hot surface per unit of time.  $T_1$  = the absolute temperature of the hot surface. To: the absolute temperature of the colder surface. C = A constant depending on the units used. If H. is expressed in B.T.U per sq. ft. of the hot surface per minute and T1 and T2 are expressed in degrees Fahrenheit on the absolute scale, then  $C = 2.66 \times 10^{-11} = \frac{2.66}{100,000,000,000}$ The above constant is only good for black surfaces and the hot surface must not "see" anything

but the colder surface.

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For surfaces not blackened this law must be modified. As a brick surface does not radiate so much heat as a blackened surface, the net heat exchange between two such surfaces is less than that exchanged between blackened surfaces at the same temperatures. Therefore in the case of tile faces, (which may be considered as radiating and absorbing heat in the same manner as brick) a co-efficient must be used in formula No. 1. This co-efficient has been found to be about .5 at 700 °C absolute temperature.

(2) 
$$H = .5 \times C (T_1^4 - T_2^4)$$

If in the tests made on the tile blocks the temperatures of the faces enclosing each cell has been taken, the actual conductivity of the tile/slabs and the air spaces could be accurately figured and the amount of heat transmitted by each could be determined. That the air space is less effective at high temperature than at low ones is known by makers of "thermos" bottles, who claim that such bottles keep liquids cold seventy-two hours and keep liquids hot only twenty-four.

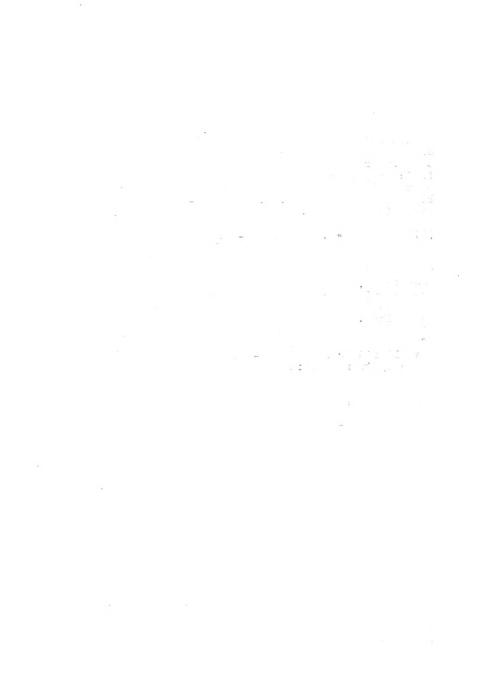
The law of heat by conduction is quite simple and may be expressed as follows:

(3)  $H = -\frac{c}{a} (T_{1} - T_{2})$ 

Where H = the quantity of heat conducted per unit of area per unit of time.

- c = the conductivity of the material, which varies somewhat with the temperature.
- d = the distance between the two parts of the body
- T1 = the temperature of the hotter part
- $T_2$  = the temperature of the colder part.

The curves on plates Nos. 16 and No. 17 seem to indicate that the conductivity of the material varies somewhat with the temperature, but since these temperatures are taken before equilibrium is established

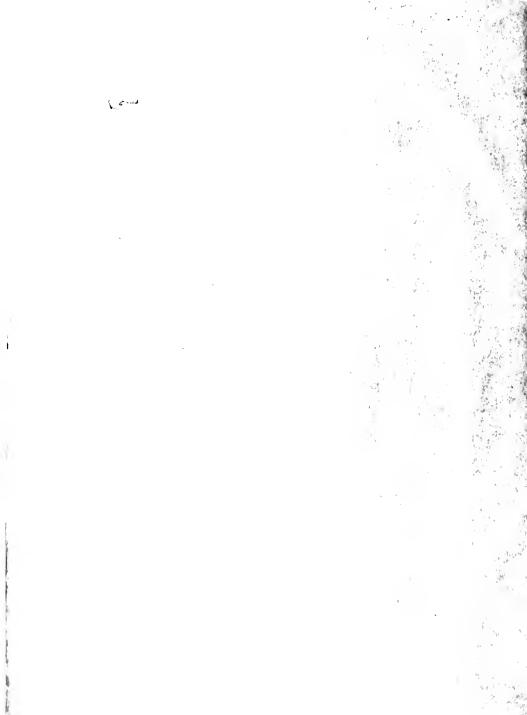


it would not be proper to say that this is proven by those curves.

Knowing the surface area, the time, the temperatures, distances d, and c, H can be easily found or knowing H, c can be readily found. All but H and c can be obtained from the data taken. A method of determining this "H", or quantity of heat transmitted would be to simply place an enclosed tank containing water at the unexposed face of the block, having an insulating covering on all sides, except that adjoining the block and measuring the quantity and rise in temperature of the water. This is, of course, only true after equilibrium has been established. Having "H" as stated "c", or conductivity of the black material, could be readily calculated.

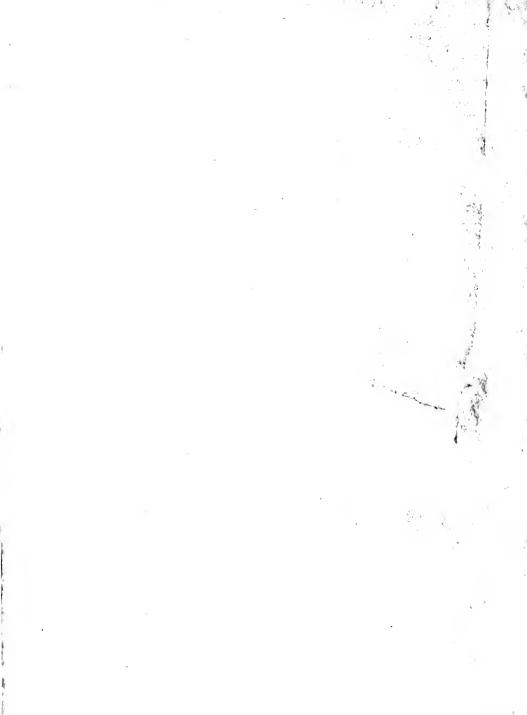
The tile fireproofing as constructed, always contains air spaces, so the conditions as they are in the experimental work should be considered comparable to those met with in actual construction. Therefore, the curves on plates Nos. 16 and 17 may be considered as giving some evidence that concrete fireproofing, when of the same thickness as tile fireproofing, will transmit less heat through it. It is the belief of the writer that, knowing the fore going physical laws, and using the results of the tests made as a basis, the transmission of heat through all types of tile and concrete fireproofing can be determined.

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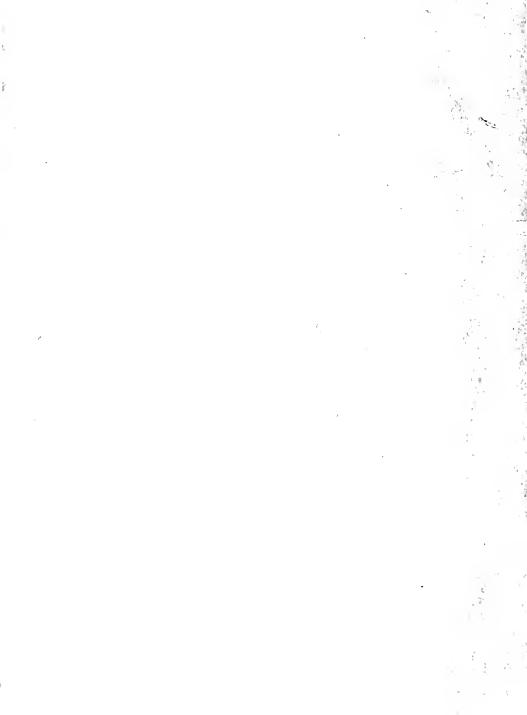
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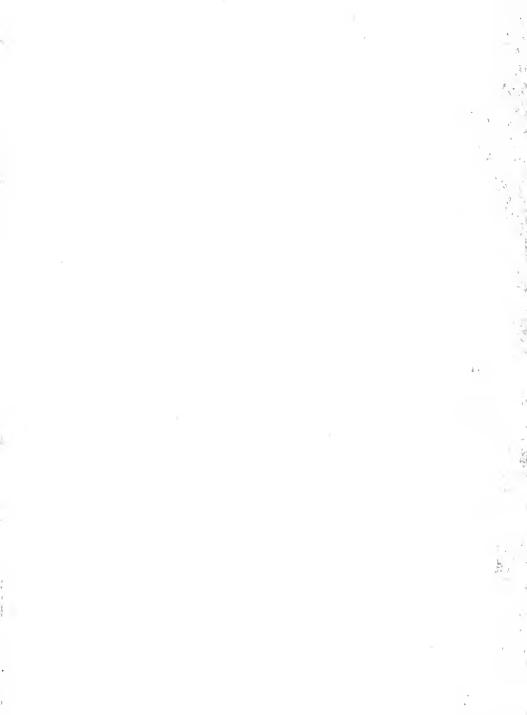
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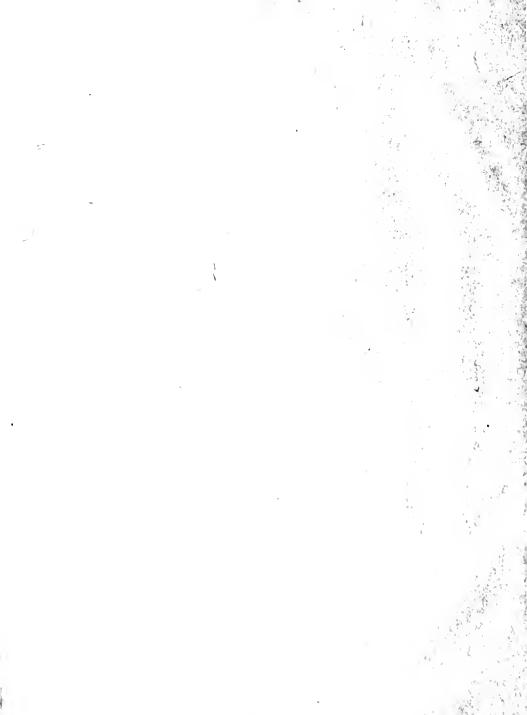
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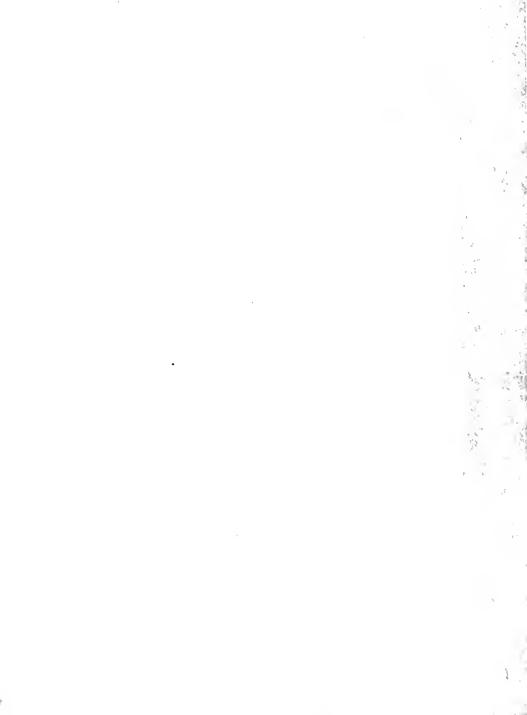


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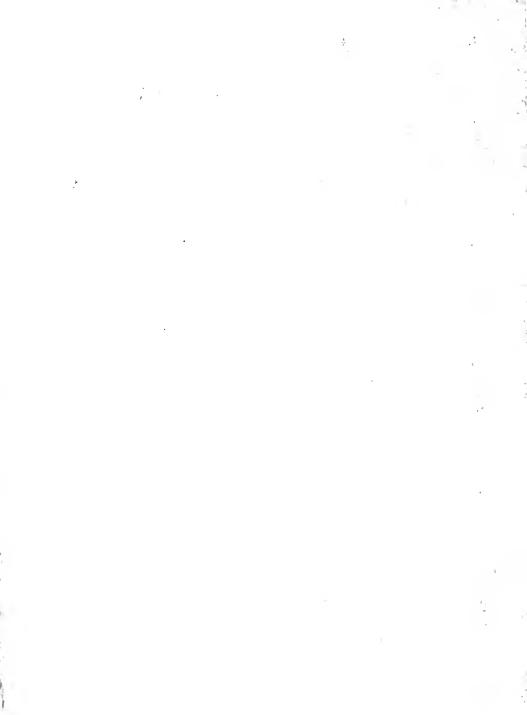
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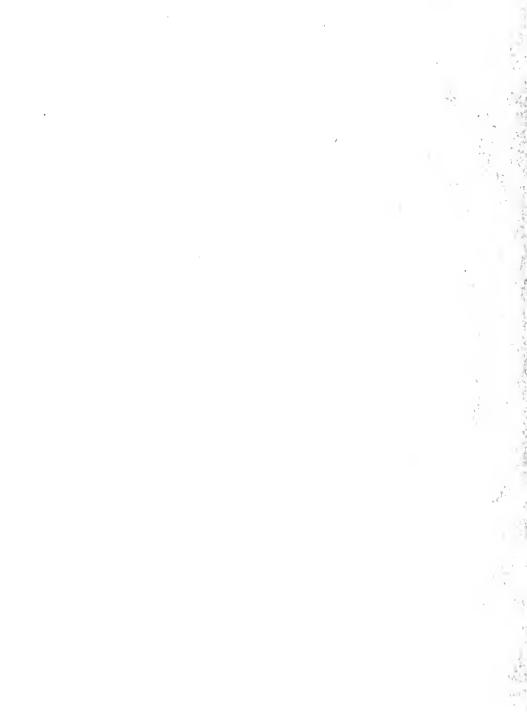
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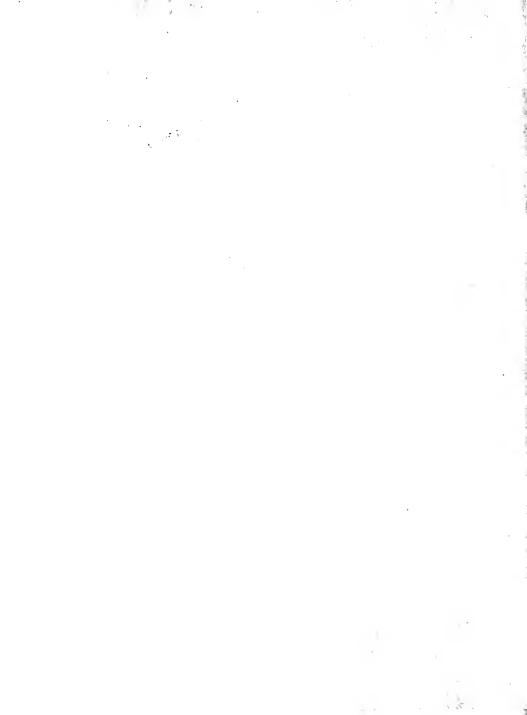
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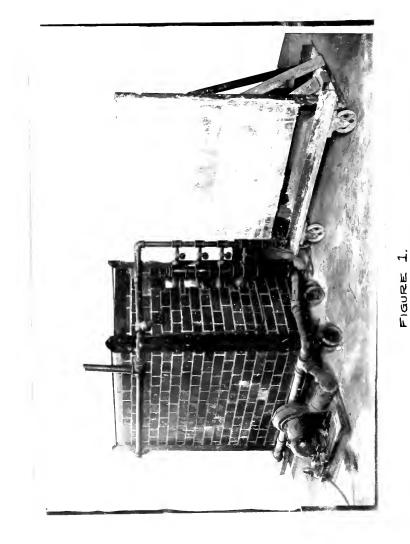
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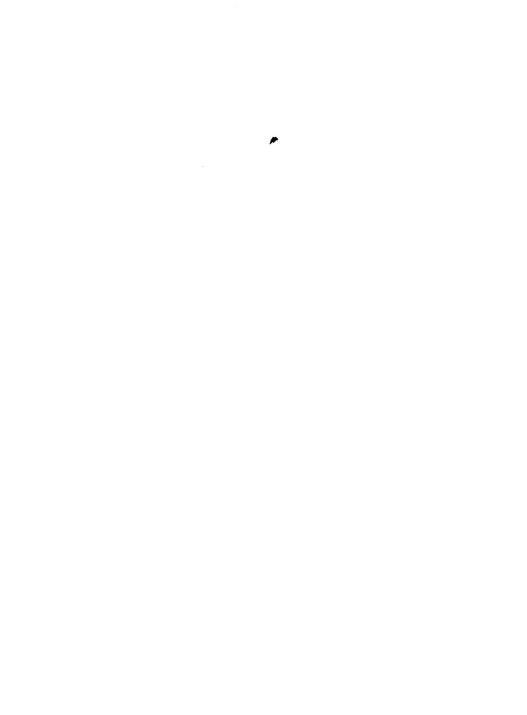
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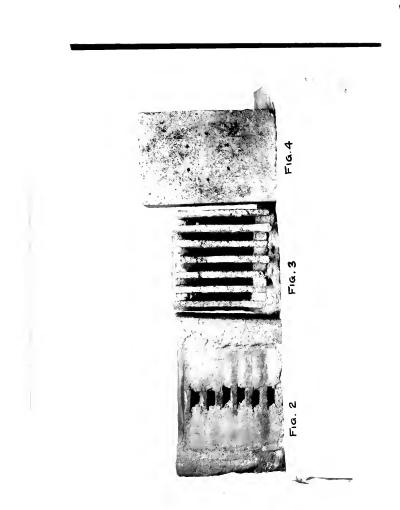
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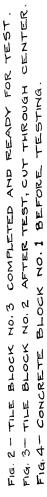
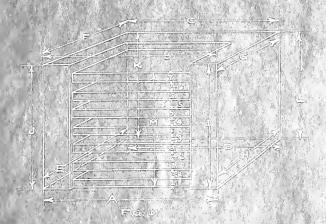


FIG. 5

TILE BLOCK NO. 2 IN PARTITION AFTER TEST,



## TILE BLOCK DIMENSIONS



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1		AQE E OLA DUREN <sup>9</sup> /64-	217 3 1978 TTG 19664
10 2	ヘンER マドリー ハモハ (ディレー (55)(55)	Aqe = 31_4 30.Re.k <sup>-5</sup> /04- <sup>1</sup> <sup>-5</sup> 2/04-	95 83 15.0 TO 36/64 <sup>-00</sup> 59/64
10 2	ヘンER マドリー ハモハ (ディレー (55)(55)	AQE E OLA DUREN <sup>9</sup> /64-	95 83 15.0 TO 36/64 <sup>-00</sup> 59/64
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1 1 10	AVER TRILL MEAS 55/04: 29/04:	Aqe = 31_4 30.REN <sup>3/</sup> 04- <sup>1</sup> <sup>32</sup> /04-	27 3 12, 17 3 5 1/2 4 5 1/2 4 5 1/2 4 5 1/2 4
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